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Formation mechanism of β'' -Mg₅Si₆ and its PFZ in an Al-Mg-Si-Mn alloy: experiment and first-principles calculations

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Abstract

Formation mechanism of β'' -Mg₅Si₆ and its PFZ in an Al-Mg-Si-Mn alloy was studied by the means of experiment and first-principles calculations. Results show that at 270°C during the 100 °C/h heating period, β'' -Mg₅Si₆ precipitated inside the dendrites, whereas precipitation free zones (PFZs) of β'' -Mg₅Si₆ presented near the dendrite arm regions. The formation of the β'' -Mg₅Si₆ and its PFZ were related to the concentration of vacancy. Low-concentration zones of vacancy formed near the eutectics during the solidification due to the constitutional supercooling, no β'' -Mg₅Si₆ precipitated in the low-concentration zones of vacancy due to the vacancy-dependence of β'' -Mg₅Si₆, the Si vacancy-containing β'' -Mg₅Si₆ was extremely unstable and Si vacancies prefer to stay away from distribution.

Key words: precipitation free zone (PFZ), β'' -Mg₅Si₆, vacancy, first-principles calculations

1. Introduction

Dispersoids-strengthened alloys refer to the resistance to motion of dislocations is increased by introducing finely divided hard particles of second phase in the metal matrix. In our previous study [1], 6082 alloys strengthened via α -Al(Fe, Mn)Si dispersoids were developed by adding Mn in 6082 alloys and by performing an ultra-low-temperature homogenization during which a number of α -Al(Fe, Mn)Si dispersoids precipitated, these dispersoids exerted a significant increase in flow stresses and hot deformation activation energy of the alloys. In our further investigation on the α -Al(Fe, Mn)Si dispersoids [2], uniformly distribution of the dispersoids was found, precipitation-free zones (PFZs) emerged near the interdendritic regions. These PFZs of dispersoid brought a consequence that local recrystallization occurred in the PFZs during the subsequent annealing due to the absence of dispersoids and the weakened recrystallization resistance.

Generally, PFZs are not preferable in the structure material, they were thought to be “weak zones” in alloys since they are often softer than other regions in the matrix [3, 4]. It is reported that the mechanical behavior, fracture resistance, and stress corrosion cracking resistance of alloys had a close relationship with PFZ features [5-7]. Two mechanisms were put forward to account for PFZ formation: vacancy depletion and solute depletion. The vacancy depletion mechanism takes the effects of vacancy sinks during quenching into account, which results in retarded precipitation around grain boundaries compared with that in the inner grain [8]. Under the solute depletion mechanism, the suppressed precipitation around grain boundaries is attributed to the decreased supersaturation of solute atoms, which is frequently induced in conjunction with the formation of more stable, coarsened grain boundary precipitates [9].

So far, the formation mechanism of PFZ about α -Al(Fe, Mn)Si dispersoids was not able to found in the literatures, however, the formation of α -Al(Fe, Mn)Si dispersoid itself has been studied by researchers. Li [10] suggested that during the heating stage of homogenization, Mg- and Si- bearing β' phase precipitated along the $\langle 100 \rangle_{Al}$ direction at 275°C and then dissolved when the temperature reached 375°C. The dissolution of β' phase created a lath-like “Si”-rich region at the former position. These lath-like “Si”-rich region thus provided favorable sites for the nucleation of α -Al(Fe, Mn)Si and resulted the α -Al(Fe, Mn)Si dispersoids formed and grow along the $\langle 001 \rangle_{Al}$ direction in the previous Si-rich sites. Hu [11] suggested that the needle-like β' phase is strongly affected by the heating rate of the homogenization, a slower heating rate of homogenization in Al-Mg-Si-Mn alloy is favorable than the fast-heating rate in aspect of generating fine and dense of β' phase. Lars Lodgaard et. al [12] provided some evidence that a “u-phase” was formed during homogenization, after continuous heating to 400°C during homogenization with a heating rate of 3K/min, a series of intermediate “u-phase” was formed, the “u-phase” has a hexagonal unit cell and was located lined up in the [100] directions of in the Al lattice since they were nucleated on the pre-existed β' needles. The meta-stable Mg-and Si-bearing particle played significant roles on the α -Al(Fe, Mn)Si dispersoids in dispersoid-strengthened Al-Mg-Si-Mn alloys.

Since the development of the α -Al(Fe, Mn)Si dispersoid-strengthened alloys several years ago, investigation focusing on promoting the dispersoid number density and improving the mechanical properties have been carried out [10, 13,14], however, the PFZ of the dispersoids had not been drawn enough attention. To date, no detailed study about the formation of its PFZ has been carried out by the means of first-principles calculations. Actually, the PFZ was one of the key factors relating to the mechanical behavior, fracture resistance, and stress corrosion cracking resistance of alloys [15, 16]. Therefore, the formation mechanism of the β'' -Mg₅Si₆ phase and its PFZ should be better understood.

In this study, a systematic investigation on the microstructure evolution refers to the Mg- and Si-containing particles and its PFZ during homogenization was performed. First-principles calculation involving the β'' -Mg₅Si₆ was performed. The focus was particularly placed on the relationship between vacancy, β'' -Mg₅Si₆ and its PFZ.

2. Experimental and calculation methods

Experiments were carried out on an Al-0.8Mg-0.75Si-0.75Fe-1Mn (in wt.%) alloy. A rectangular plate measuring 200 mm × 100 mm × 7 mm was cast using a sand mold, the cooling rate of the solidification was tested to be 1.4 °C/s. An ultra-low-temperature homogenization at 430 °C for 6 h was conducted to promote the precipitation of dispersoids. A heating rate of 100 °C/h was set to further enhance dispersoid precipitation [1, 2, 11]. Quenching at 250, 270 and 300°C during heating stage was performed to verify the microstructural evolution. An optical microscope (Leica DMI5000M), a scanning electron microscope (Shimadzu, SSX-550) equipped with an energy-dispersive X-ray spectrometer and a transmission electron microscope (JEM-2100) were used for the microstructural observations. To reveal the precipitation particles, the surfaces of some polished specimens were etched with 0.5% HF solution for 30 s. Emission electron probe microanalysis (EPMA) was used to measure the alloying element distributions.

The first-principles calculations performed in the present work were based on the density functional theory (DFT) by the use of the Vienna ab initio Simulation Package (VASP) [17]. The interaction between ions and electrons was described by the projector-augmented wave method [18], and the exchange-correlation potential was treated by the Perdew–Burke–Ernzerhof (PBE) implementation of generalized gradient approximation (GGA) [19]. The $3s^2$ and $3s^2 3p^2$ were treated as valence states for Mg and Si. The energy of the cut-off was 319 eV. A Monkhorst–Pack grid was employed to sample the Brillouin zone [20], and the $4 \times 16 \times 10$ k-point grids was used to calculate the β'' -Mg₅Si₆ and its vacancy structure. During the structural optimization, the criteria for the total energies and the forces on individual atoms were set to 1 meV and 0.02 eV/Å, respectively.

3. Results and discussions

3.1 Microstructure evolution during homogenization

Homogenization treatment was conducted with a heating rate of 100 °C/h. Fig. 1 shows the microstructural evolution. Fig. 1a shows the microstructure of the sample when the temperature reached 250°C during heating, the microstructure is consistent with the Al matrix and eutectics distributed along the dendrite boundaries. With the increasing of temperature to 270°C (Fig. 1b), large number of fine precipitates was observed inside the dendrites, whereas precipitation free zones (PFZs) presented near the dendrite arm regions. The inset TEM image shows the morphology of the precipitation, the particles are lath-like and are approximately 25 nm in width and about 300 to 500 nm in length which located along with the $\langle 100 \rangle$ Al directions, the black dots represent the cross sections. These precipitates were identified to be β'' -Mg₅Si₆ phase according to the corresponding selected area diffraction pattern (SADP, Fig inset in Fig. 1b, as indicated by the red circles) [10, 21]. When the temperature reached 300°C during the heating stage of homogenization (Fig. 1c), the OM microstructure shows

limit variation with that of 270°C condition, however, in the TEM image (as inset of Fig. 1c), partially dissolution of the β'' phase was observed. It is noticeable that some dot-like particles emerged on the position of the pre-existed β'' - Mg_5Si_6 . After 6h hold at 430°C (Fig. 1d), PFZ could be still observed in the OM image. In the inset TEM image, it is found that the particles appeared in various shapes, mostly short-plates and rods, It is distinguishable that these particles mainly located along with the $\langle 001 \rangle$ Al direction. The particles were confirmed to α -Al(Fe, Mn)Si dispersoids according to previous studies [1, 2].

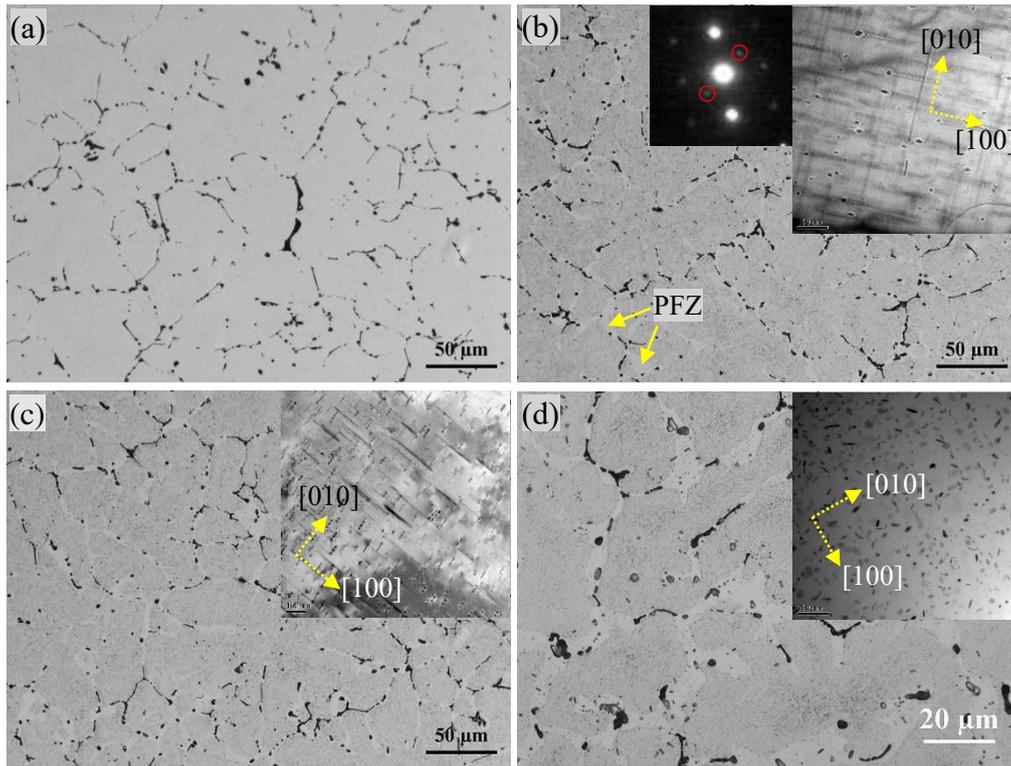


Fig. 1 OM image of the sample when the temperature reached (a) 250°C, (b) 270°C, (c) 300°C during heating; (d) after 6h hold at 430°C. TEM image were inset, the selected area diffraction pattern was from the particles in the TEM image of (b), β'' phase was indicated by the red circles.

3.2 β'' - Mg_5Si_6 and its PFZ

During the heating stage of homogenization, the supersaturated solid solution of Mg and Si created during casting decompose and β'' - Mg_5Si_6 precipitated (Fig.1). Inhomogeneous distributions of β'' - Mg_5Si_6 was a result of the performance differences between the region inside the dendrites and the region near dendrite arms and intermetallics. In general, the formation of PFZ could be attributed to lack of particle former elements and/or heterogeneous nucleation sites [8, 9]. Emission electron probe analysis (EPMA) was carried out to check the alloying element concentrations of the as-cast alloy. The chosen positions for the test are shown in Fig. 2. Four points were set as the test positions: points 1 and 4 were set inside the dendrites, while points 2 and 3

were located near the intermetallics. The results of EPMA are shown in Table 1. As indicated, uneven distribution of alloying elements of Mg, Si, Fe and Mn was found, all the alloying element exhibited higher concentrations at the regions near dendrite arms than the positions inside the dendrites. For example, concentrations of Mg and Si at the dendrites arm region (position 2 and 3) were 0.522-0.594, while it decreased to 0.215-0.217 at the region inside dendrites (position 4 and 1). This result, the element of Mg and Si was not deficient at the position of β'' PFZ formed, but even in a higher concentration, excluded the possibility that the β'' PFZ was attributed to the lack of alloying elements.

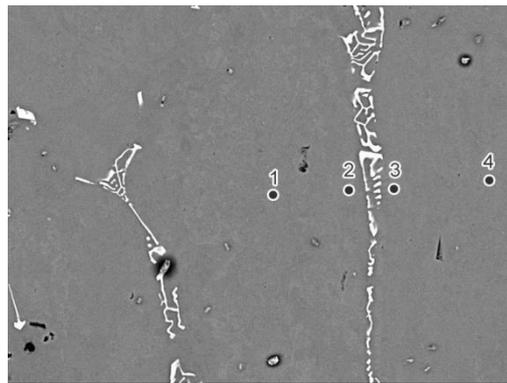


Fig. 2 EPMA positions of the as-cast sample.

Table. 1 Alloying element content (wt.%) at different positions in Fig. 2

Position	1	2	3	4
Mg	0.217	0.522	0.594	0.215
Si	0.268	0.572	0.578	0.273
Fe	0.010	0.056	0.068	0.012
Mn	0.481	0.656	0.587	0.476

The higher concentration of alloying elements at the dendrites arm regions than that inside the dendrites could be explained by the classic solidification theory. As illustrated in Fig. 3, during the eutectic reaction of solidification, the alloying elements are released to the liquid from the interface of the solid and liquid. Along with the ongoing solidification, more alloying elements were accumulated at grooves where the boundaries of dendrites arms were formed. With the finishing of solidification, the high concentration of solutes was squeezed into the boundaries dendrites arms, thus formed the high concentration region of alloying elements at the dendrite are regions [22].

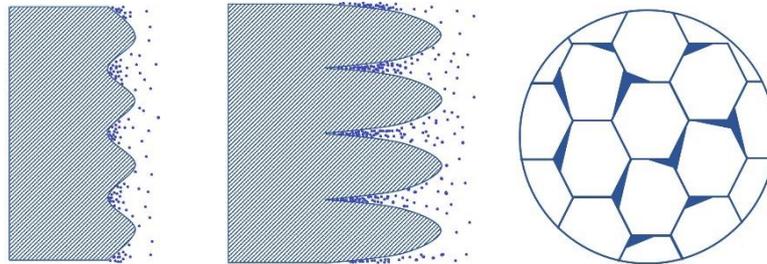


Fig. 3 Schematic of the process of solidification

The nonuniformly distribution of the solutions in the Al dendrites was accompanied with the vary of solidification and the concentration of vacancy. Fig. 4 shows a phase diagram illustrating a typical eutectic solidification. The primary Al crystal formed at the temperature of liquid line. Along with the solidification, as the solutes were pushed to the liquid phase, the solution concentration increased and the T_s reduced accordingly, which is referred as “constitutional supercooling” [23, 24].

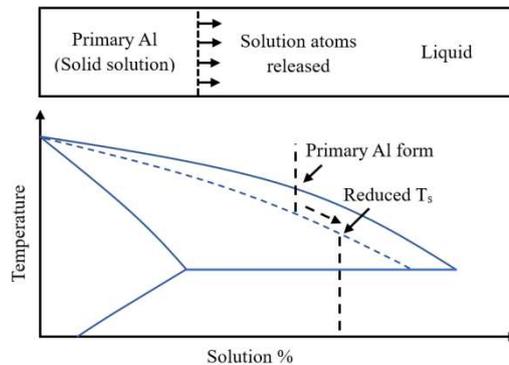


Fig. 4 Phase diagram illustrating typical eutectic solidification

Based on the relationship between the solidification temperature and vacancy, the concentration of vacancy declined with the decreasing solidification temperatures, it was inferred that the concentrations of vacancy at the region near eutectic were lower than those at the region inside the dendrites, that was, the vacancy concentration positions 2 and 3 in Fig. 2 were lower than those of positions 1 and 4.

The demonstration above suggested that the at the region near dendrite arms where PFZ of β'' and α -Al(Fe, Mn)Si dispersoid emerged, the alloying elements was

accumulated during the eutectic solidification, which led a consequence that the overall lowered solidification temperature and concentration of vacancy at the region near dendrite arms than that inside the dendrites.

3.3 First-principles calculations: vacancy and β'' -Mg₅Si₆

In order to verify the relationship between the vacancy and β'' , the first-principles calculations were carried out. Here we use the defect formation energy to express the stability of the structure. Defect formation can be calculated by the following equation [25]:

$$E_f = E_{def} - E_{id} + n_{Mg}\mu_{Mg} + n_{Si}\mu_{Si}$$

where E_f is the defect formation energy; E_{def} and E_{id} are the total energies of the defective and ideal unit cell, respectively; n_{Mg} and n_{Si} are the number of Mg and Si atoms which is transferred to or from a chemical reservoir; μ_{Mg} and μ_{Si} are the corresponding chemical potential of the pure element.

The optimized lattice constants of the β'' -Mg₅Si₆ structure are a=15.01, b=4.07, c=6.87, and β =107.3°. This is very close to the experimental values (a=15.16, b=4.05, c=6.74, and β =105.3°) [26, 27]. The relative errors for a, b, and c are 0.98%, 0.49% and 1.89% respectively, and that of the monoclinic angle β is 1.86%. The above results prove the accuracy of the calculations.

Fig. 5 shows the formation energies of the β'' -Mg₅Si₆ and its vacancy structures. As can be seen from the figure, β'' -Mg₅Si₆ structure has the lowest formation energy, indicating that the structure is the most stable at this point. With the formation energy gradually increasing as the vacancy concentration increases, leading to a substable state for the vacancy-containing β'' -Mg₅Si₆ structure. The Si vacancy-containing β'' -Mg₅Si₆ structure is the most unstable, implying that the Si vacancy-containing precipitates are extremely unstable and thus it can be preferentially precipitated, which is consistent with the experimental observations (Fig. 1).

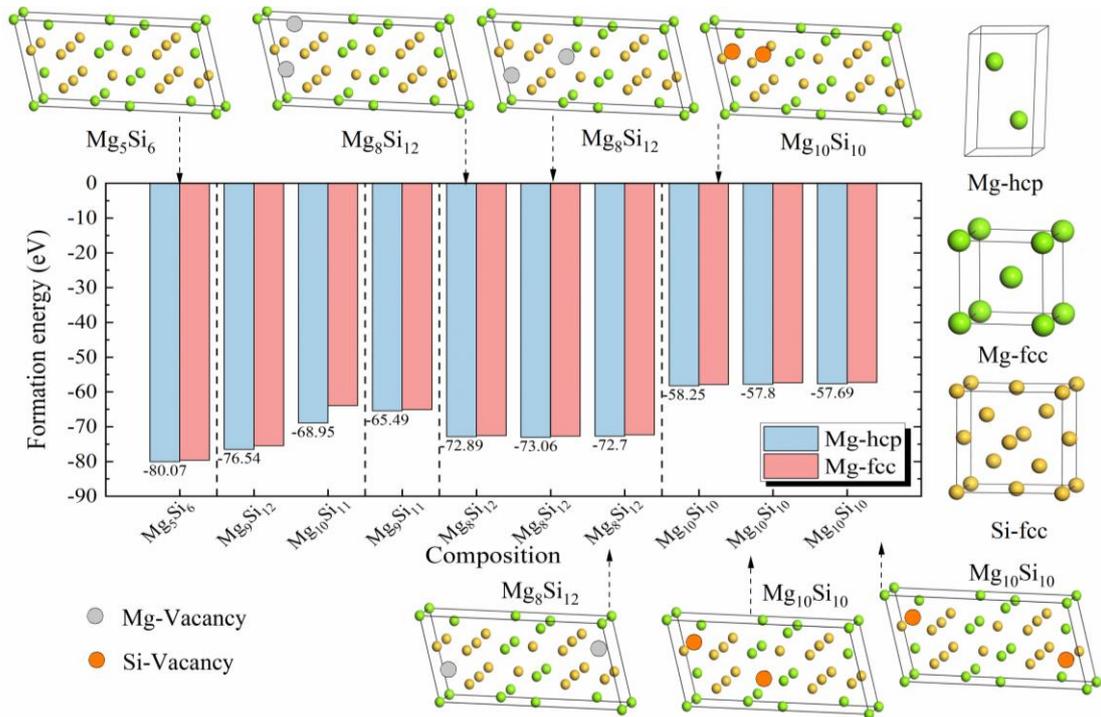


Fig. 5 Illustrations of the Mg₅Si₆, the Mg-vacancy, and Si-vacancy. Formation energies of Mg₅Si₆ structure and containing-vacancy in Mg₅Si₆ structure.

Subsequently, the variation of the formation energy with the degree of aggregation of Mg or Si atoms was calculated. The degree of atomic aggregation is divided into three main cases: aggregated, uniform, and distant distributions. The results show that both Mg vacancies and Si vacancies are most unstable when distant distributed. These are extremely easy to observe experimentally (Fig. 1).

4. Conclusions

An Al-Mg-Si-Mn alloy was prepared and an ultra-low-temperature homogenization at 430 °C for 6 h was conducted. Formation mechanism of β''-Mg₅Si₆ and its PFZ in an Al-Mg-Si-Mn alloy was studied by means of experiment and first-principles calculations. The following conclusions were drawn:

(1) At 270°C during the 100 °C/h heating, β''-Mg₅Si₆ precipitated inside the dendrites, whereas precipitation free zones (PFZs) of β''-Mg₅Si₆ presented near the dendrite arm regions. After homogenization at 430°C for 6h, α-Al(Fe, Mn)Si dispersoids formed inside the dendrites while the PFZs of α-Al(Fe, Mn)Si were located near the dendrite arm regions.

(2) The element of Mg and Si at the regions near the dendrite arms were higher than the regions inside dendrites, which excluded the possibility that the β''-Mg₅Si₆ PFZ was attributed to the lack of alloying elements.

(3) Low-concentration zones of vacancy formed near the eutectics during the solidification due to the constitutional supercooling, no β''-Mg₅Si₆ precipitated in the low concentration zones of vacancy owing to the vacancy acted as the nucleation sites for the precipitation of β''-Mg₅Si₆.

(4) The results of first-principles calculations show that the Si vacancy-containing precipitates are extremely unstable and Si vacancies prefer to stay away from distribution, which further confirmed the vacancy-dependence of β'' -Mg₅Si₆ precipitation and explained the formation of PFZs of β'' -Mg₅Si₆ and α -Al(Fe, Mn)Si dispersoids.

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