

The Combinatory Effects of Zinc Oxide Nanoparticles (ZnO NPs) and Thiamine on Skin of Alloxan-Induced Diabetic Mice; A Stereological and Biochemical Study

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Abstract

Various studies have shown that ZnO NPs as an anti-diabetic agent can be effective in the experimental model of diabetes. This study aimed to investigate treatment effects of ZnO NPs plus thiamine on histomorphological and biochemical parameters in the diabetic mouse skin. In total 49 BALB/C mice were used and divided into seven groups. Fourteen mice were equally considered as the control and thiamine groups (groups I and II). The rest of the mice were injected with alloxan (220 mg/kg) to induce diabetes; one untreated diabetes group (group III). Other diabetic mice were treated with ZnO NPs (0.1 and 0.5 mg/kg) alone (groups IV and V, respectively) and along with thiamine (groups VI and VII, respectively). Body weights of mice at days 0, 10, and 20 were calculated to appraise weight gain. Stereological and biochemical parameters were measured to evaluate diabetes effects in all groups. An increase in weight was observed in the diabetic group compared to the control group. Diabetic skin showed decreasing in volume density of collagen bundles and decreasing in the epidermis and dermis thickness, as well as an increase in the hypodermis's thickness. Administration of ZnO NPs (0.1 and 0.5 mg/kg) alone and along with thiamine in the diabetic animals resulted in anti-hyperglycemic activity, reducing GGT, BUN, Cr, MDA, and NO levels in treated diabetic mice. In conclusion, the concomitant use of ZnO NPs along with thiamine presents the potential as a combination therapy for the treatment of alloxan-induced diabetic mice skin changes.

Introduction

Diabetes is a chronic metabolic disease in which some disturbances occur in the metabolism of lipids, proteins and carbohydrates, besides an increase in blood glucose level [1, 2]. Complications followed by this disease include nephropathy, retinopathy and neuropathy. It can trigger cardiovascular disorders like cardiac infarction and hypertension and also cataract in the eyes [3]. Since some researchers have found changes in the skin epidermis in diabetic patients, so this has led to increased attention to the study of non-injured diabetic skin [4]. Skin diseases caused by diabetes have serious side effects; therefore, recognizing the most effective treatment measures results from a better understanding of the pathology of diabetic skin disorders [5]. Although several factors may be involved in chronic skin complications in diabetics, these side effects become more complicated because of changes in the skin's structure. Besides, fibroblast growth failure decreased collagen synthesis, and, in particular, accumulation of advanced glycation end products (AGEs) occurs in diabetic skin complications [6].

In recent years, many studies have been conducted on zinc oxide (ZnO) and various forms of its nanostructures. It has also a vast application in sensors due to its high sensitivity to chemicals and high contact surface. ZnO NPs trap light due to their high contact surface [7], therefore, these nanoparticles are frequently used in the manufacture of solar cells. Other applications of these compounds are iron galvanization [8], pigments, corrosion preventing agents, production of toothpaste, sunscreens, etc. [9]. ZnO NPs have antibacterial and antifungal properties that are used in prophylactic drugs against bacterial disease [10]. ZnO NPs, as a new agent for zinc delivery, in addition to its biotechnological applications, have many implications for the treatment of numerous diseases, including diabetes[11].

The first vitamin discovered in the group of B vitamins was thiamine (B1). In the human body, this vitamin is converted to free thiamine and phosphorylated to different forms, including mono-, di-, tri- and Pyrophosphate forms, which is the active form of thiamine [12]. One of the most important co-factors in many crucial metabolic processes, including the production of acetylsalicylic acid (COA), Calvin cycle and Pentose phosphate pathway, is thiamine pyrophosphate [13]. In the body, high concentrations of thiamine are found in skeletal muscle, heart, liver, kidney and brain (Martin 2001). The most important functions of thiamine are carbohydrate catabolism and nucleic acid and NADPH production [14]. most important functions of thiamine are carbohydrate catabolism and nucleic acid and NADPH production [15]. It has been also shown that thiamine can stabilize the membrane ion channels and change their activity [16]. The application of thiamine in the treatment of neurological disorders and seizures has been revealed and proven for healing of diabetic complications. Studies on laboratory animals showed that there is a logical relationship between thiamine and pre-synaptic acetylcholine release [17].

The present study aimed to evaluate the preventive effects of two different concentrations from ZnO NPs (0.1 and 0.5) alone and along with thiamine against diabetes-induced skin changes in a mouse model.

Materials And Methods

Animals

In this experimental study, 49 healthy male BALB/C mice, weighing about 22 ± 1 g and 5-weeks-old (each group 8 samples) were selected. As an adaptation period, animals were kept for a week at a temperature of $19 \pm 2^\circ\text{C}$ with a 12-hour dark-light cycle. In order to provide a convenient environment during the experiment period, cage floors were covered with fresh and soft sawdust, and every three days the cages were cleansed to maintain the hygiene. The animals were prepared with commercial mouse pellet food containing essential vitamins required for the body besides free access to water. The commercial mouse pellet diet was purchased from Experimental Animal Center in Tehran, Iran. All ethical regulations and protocols on laboratory animal's research were considered and approved by the Animal Rights Committee of the Shahrekord Veterinary Faculty.

Preparation of ZnO Nanoparticles

ZnO nanoparticles gained from US Research Nanomaterials, Inc., (Houston, TX, 77084, USA). This nanoparticle (ZnO) harvest itself was a white powder with amount of purity $\geq 99 + \%$. The actual characteristics of ZnO nanoparticles refer to its basic physical properties. Nanoparticles of ZnO had dimensions of 10-30 nm in size (average 20 nm), as listed in Table 1, and used in concentrations of 0.1 and 0.5 mg/kg (figure 1) [2].

Grouping and Sampling

After a week of adaptation to the laboratory environment, the mice were randomly divided into 7 groups of 7 individuals each. The diabetes was induced in the experimental mice (groups 3 to 7) by using a

single intraperitoneal injection (IP) at a dose of 220 mg/kg body weight of alloxan monohydrate (Sigma-Aldrich, St. Louis, MO, USA) in PBS soluble (pH=7). Seventy-two hours later, a few mice were randomly selected from diabetic mice and their fasting blood glucose was measured. Mice with a blood glucose level of above 200 mg/dl [2] were considered diabetic mice. Also, the effects of diabetes were assessed by biopsy of the pancreas with the approval of a pathologist in some induced diabetes specimens as definitive confirmation.

With diabetes confirmed, ZnO nanoparticles were injected into the diabetic mice at two doses of 0.1 and 0.5 mg/kg [2] and thiamine injected at a dose of 5 mg/100 g. Finally, all mice were divided into the following groups: All the injections were intraperitoneal (IP).

group I: Control group; distilled water administered.

group II: Thiamine (5 mg/100 g)

group III: Diabetic group (injection of alloxan; 220 mg/kg)

group IV: Diabetes + ZnO nanoparticles 0.1 mg/kg

group V: Diabetes + ZnO nanoparticles 0.5 mg/kg

group VI: Diabetes + ZnO nanoparticles 0.1 mg/kg along with thiamine (5 mg/100 g)

group VII: Diabetes + ZnO nanoparticles 0.5 mg/kg along with thiamine (5 mg/100 g)

On the 20th day after the last injection, the mice were anesthetized (with fasting) by chloroform (Merck KGaA, Index No: 2806-18-9), and the blood samples were withdrawn directly from the ventricle of the heart. Blood was centrifuged (Hettich-Germany) for 10 min at 3000 r/min and 4 °C, the supernatant solution was reserve at -70 °C until biochemical analysis.

The skin samples at the three areas mid-dorsal, mid-ventral and lateral were considered. After shaving the long hair, the skin from the mentioned regions was excised and trimmed into small-size pieces, fixed in a 10% buffered formalin (Merck) solution for histo-morphometric examinations. Tissue sections of about 5µm were serially obtained, stained with H&E and, and evaluated under the light microscope.

Stereological evaluation

According to the table of random numbers, in the interval between the 50 sections the first section, or primary section (e.g., 10th section), and the second section that is reference section were selected. Sampling in the distance between sections was so that each 50th section was sampled (the primary sections). Therefore, for the primary and reference sections, e.g. 10, 12, 60, 62 were sampled in the entire skin; as detailed in our previous research (figure 1) [18].

To evaluate the volume densities of tissue collagen bundles and fibroblast, a network of uniformly and permanently spaced points (point grid) was superimposed at random over the image of each predicted section. Collagen bundles that were hit by a point were sampled and subjected to volume densities estimation according to the number of points that hitting with the bundles. The total volume of collagen and fibroblasts were calculated using the following formula [19]:

$$Vv = \sum_{1}^{n} A_{sec} \times (t)$$

Where, A_{sec} is the area of cut surfaces (1st-nth slice) of systematic-random sections through the structure, and t , the thickness of the sections.

Measurement of serum glucose

Serum glucose levels were measured using a glucose analysis kit (Pars Azmoon Co, Tehran, Iran). The principle was based on the bi-enzymatic assay comprise enzymatic oxidation of glucose to gluconic acid, yielding hydrogen peroxide and then the reaction of hydrogen peroxide with 4-aminoantipyrine and phenol to yield the colorimetric product which is measured at 500 nm. The results were reported as mg/dl.

Estimation of MDA Level

Malondialdehyde (MDA) as the end product of lipid peroxidation reacts with Thiobarbituric acid (TBA) and yields colored complex which can be measured spectrophotometrically [20]. In this study, 2 ml of TBA reagent containing 0.375% TBA, 15% TCA and 0.25 mol/L HCl was added to 1 ml of serum from all groups. The mixture was placed in boiling water for 50 min, cooled to room temperature and centrifuged at 1000 rpm for 10 min. Thereafter, the absorbance of the supernatant was read at the wavelength of 535 nm against the blank reference. The results were calculated by using a molar extinction coefficient for MDA of 1.56×10^5 M/cm. The concentration of MDA was expressed as nmol/ml [21].

Nitric oxide assay

Briefly, 1 ml of serum was deproteinized by adding ten microliters of 1.5 g/mL ZnSO₄ solution. The mixture was vortexed and centrifuged at 10000 × g for 10 min at 4 °C. Then, 100 ul of the supernatant was added to a 96-well ELISA plate, and each well was submitted to 100 μL of vanadium (III) chloride (8mg/ml) which converts nitrate to nitrite. After the addition of 100 ul of Griess reagent (equal mixture of 1% sulphanilamide in 5% phosphoric acid and 0.1% N-(1-naphthyl) ethylenediamine hydrochloride in distilled water), the plates were maintained at *room temperatures* for 30 minutes and the nitrite production was quantified calorimetrically at 540 nm. The NO levels were calculated using various concentrations of sodium nitrite (0.1-100 μm) as a standard and expressed as μmol/ml [22].

Measurement of serum BUN and Cr levels

Serum BUN was measured using a commercial kit (Pars Azmoon Co., Tehran, Iran) based on the enzymatic hydrolysis of urea to ammonia by urease. Then, in a parallel reaction, the ammonia is converted to glutamate by glutamate dehydrogenase and this reaction was monitored at 340 nm. The Cr level was measured by the Jaffe method (Pars. Azmoon Kit, Tehran, Iran) in which creatinine reacts with picric acid to form a reddish complex [23].

GGT assay

Gamma-glutamyl transferase (GGT) activity was measured by using the Pars Azmoon kit (Tehran, Iran, according to the manufacturer). The reaction is evaluated based on transferring of γ -glutamyl group from γ -glutamyl-p-nitroanilide to acceptor glycylglycine and production of para-nitroaniline which can be determined calorimetrically at 405 nm; the results were reported as U/l.

Statistical Analysis

Statistical analyses of volume, histo-morphometric and biochemical data were carried out using the SPSS statistical software package version 23.0 (SPSS Inc., Chicago, IL) for Windows. Data are declared as mean \pm standard deviation (SD) and statistical variations were tested by One-way ANOVA. LSD post-hoc test was applied and the values confirming $p < 0.05$ were considered significant compared to the rest.

Results

Body weight

There were no deaths during and after investigating due to drug injection.

To assess the effect of ZnO NPs and thiamine on the weight gain, the mice body weight was determined in the beginning treatment (day 0) and no significant differences were seen between the controls, untreated, and treated diabetic mice (groups I to VII) ($P > 0.05$). In the untreated diabetic group (group III) on days 10 and 20 body weight was significantly increased compared to the control and thiamine groups (groups I and II) and also day 0 in all groups (figure 2) ($P < 0.05$). Although, groups IV and V (diabetes+ZnO NPs, 0.1 and 0.5) on day 20, showed a significant increase than to controls ($P < 0.05$), but, a non-significant decrease was observed between groups IV and V when compared to the diabetic group ($P > 0.05$). A marked reduction in body mass was observed in the ZnO NPs groups 0.1 and 0.5 + thiamine (groups VI and VII) than in the diabetic group ($P > 0.05$). Therefore, a gradual reduction was observed in the body weight in groups VI and VII (diabetes treated with ZnO NPs, 0.1 and 0.5) along with thiamine compared to the untreated diabetes group; near to baseline ($P > 0.05$).

Volume of collagen bundles

The volume density of collagen bundles (Table 2) in the skin samples of the untreated diabetic mice was significantly lower than those of the control and thiamine individuals ($P < 0.05$). However, twenty-day treatment with ZnO NPs (0.1 and 0.5 at 10 mg/kg) and thiamine (groups 4 to 7) caused a significant increase in volume density of collagen when compared to the untreated diabetic group ($P < 0.05$).

Morphometric criteria

Figure 3 presents findings from the morphometrical analysis of dermis, epidermis, and hypodermis of skin in the control, untreated and treated diabetic groups. By evaluating the morphometric data, evident variations were obtained among untreated and treated diabetic groups (groups 3 to 7) than to control groups (groups 1 and 2), with mention to the various factors studied. Histo-morphometric criteria of the epidermis thickness exhibited a significant decrease in the untreated diabetic group (group 3) compared to the control group ($P < 0.05$). In addition, diabetes treated with ZnO NPs (0.1 and 0.5) and thiamine showed a significant increase in epidermal thickness compared to untreated diabetic mice ($P > 0.05$), near to normal conditions.

The dermal criteria included the thickness of the dermis showed a similar histomorphometry pattern with the epidermis, as there statistically was a significant decrease in the thickness of the dermis layer in the diabetic group compared to the control group ($P < 0.05$). On the other hand, the thickness of the dermis layer displayed a sensible decrease in group 3 (diabetes alone) than in groups 4 to 7 (ZnO NPs, 0.1 and 0.5, alone and along with thiamine) ($P > 0.05$).

The mean thickness (μm) of the hypodermis layer in the diabetic and diabetic + ZnO NPs (0.1 mg/kg) groups increased significantly compared to the control group ($P < 0.05$). Treatment with ZnO NPs and thiamine showed a marked reduction in thickening hypodermis in groups 5 to 7 compared to the diabetes group (figures 2 and 3) ($P > 0.05$).

Fasting blood glucose

Table 3 summarizes the level of blood glucose (mg/dl), Cr (mg/dl), BUN (mg/dl) and GGT (u/l) of controls and experimental animals. When all the mentioned biochemical parameters were evaluated in the control group, it was found that there was no significant difference with the thiamine group ($P > 0.05$). The blood glucose level of the untreated diabetic group was significantly ($p < 0.05$) higher than the control group. Also, the concentrations of the fasting glucose in the diabetic group (351.32 ± 4.17) revealed a larger significant increase than those in the other four groups (groups 4 to 7), when all groups were compared with the control group ($P < 0.05$).

Cr, BUN, GGT values

A similar pattern was observed among Cr, BUN, and GGT levels when all groups in each parameter were compared. Plasma Cr, BUN, and GGT values were elevated significantly in untreated diabetic mice compared with controls ($P < 0.05$). However, the doses of 0.1 and 0.5 mg/kg ZnO NPs alone and along with thiamine (groups 4 to 7) produced a significant decrease ($P < 0.05$) in the serum levels of Cr, BUN, and

GGT than to control and untreated diabetic groups. However, the biochemical experiments were not significantly changed between treated diabetic groups 4 to 7, ($P>0.05$) (Table 3).

MDA and NO levels

Table 4 displayed that in the untreated diabetic mice (group 3), the MDA value was significantly higher than the control mice (5.97 ± 0.62 and 3.29 ± 0.38 nmol/ml, respectively) ($P<0.05$). Treatment with ZnO NPs (0.1 and 0.5 mg/kg) alone and along with thiamine decreased the MDA value of the serum compared with the untreated diabetic mice, however, these changes were significant. ($p<0.05$). A significant increase in the skin amount of NO was seen in the untreated diabetic mice compared with the control group ($P<0.05$). In diabetic mice treated with ZnO NPs (0.1 and 0.5) alone and in combination with thiamine (groups 4 to 7), serum NO concentration was lower than (0.80 ± 0.05 nmol/ml) untreated diabetic mice and higher than in the control group ($P>0.05$).

Discussion

Cutaneous complications of diabetes have attracted a lot of attention because of the clinically remarkable problems, to the extent that it leads to a reduction in skin thickness and increased blood glucose levels [5, 24]. From a long time ago, the balance between the disadvantages against the benefits of metal-based nanoparticles has been debated their effectiveness in counteracting the destructive effects of some diseases including diabetes, has been more pronounced. These investigations led to the fact that nanoparticles, such as zinc oxide, can show anti-hyperglycemic activity with a managed treatment for various scheduled periods [24–26]. Our study showed that alloxan-induced diabetes in mice significantly altered the skin of treated animals compared with non-diabetic animals. Diabetic mice, against non-diabetic animals, exhibit clinical symptoms of diabetes like polydipsia and polyuria (by comparing their body weight) and high fasting blood sugar (351.32 ± 4.17 versus 82.33 ± 4.10) ($p<0.05$).

A specific hallmark that facilitates molecular diagnosis in the evaluation of diabetes is the glycation formation; glycation is a non-enzymatic chemical reaction of free reducing sugars, in which reducing sugar linkage with a free amino group of proteins [27, 28]. These irreversible reactions produce advanced glycation end products (AGEs) with the help of a chain of various reactions [27, 29]. Some AGEs that may use as a glycation stress indicator, and a marker for diabetes, included CML (N ϵ -(carboxymethyl) lysine) [30]. CML, a product of proteins modification by glyoxal, as a result, can be observed in high oxidative stress states and/or diabetic situations [31]. These investigations led to the fact that CML can alter skin collagen by reducing the elasticity of the skin, and shrinkage form on the skin [27].

The results of the present study showed that diabetes decreased the collagen volume density (%) in mice skin compared to healthy mice. Besides, in groups of the ZnO NPs (0.1 and 0.5) alone and along with thiamine, the collagen volume decreased than those in the control group. Nevertheless, the effect of untreated diabetes was more severe than the effect of ZnO NPs (0.1 and 0.5) alone and along with thiamine on the collagen volume. ZnO NPs and thiamine had a controlling role in further reducing collagen volume in the diabetic group. Concerning the ZnO NPs, our results pointed out that its

administration in combination with thiamine has more therapeutic effects than using alone. Angela et al., (2016) in the research, showed that the level of matrix metalloproteinases (MMPs) in the skin of diabetics is greatly increased [32]. Nagase et al., (2006) demonstrated MMPs are a group of proteinases, because of the nature of their substrate, contain six subclasses, including collagenase [33]. Studies that by researchers conducted over the past few years have shown that MMP-1 (collagenase) is significantly increased in naturally aged skin and or diabetes [34]. These investigations led to the conclusion that MMP-1 is the major protease capable of initiating the fragmentation of native fibrillar collagen [35].

Collagen assists the molecular constructions in developed and organized tissues, such as the skin, to maintain their morphological and mechanical properties. Since the consistency, integrity, and longevity of the dermis depend on collagen, therefore, the quality characteristics of collagen can affect these indicators [36]. Chithra et al., (1998) in their research, showed that collagen catabolism occurs in the dermis of diabetic mice, so they concluded diabetes increases collagen catabolism in skin samples. Other results of this study included thickened cutaneous artery sheaths, rupture and multilayered basement membranes with destroyed collagen fibers [37]. The present study agrees with previous research, in that the thickness per μm of the epidermis and dermis in untreated diabetic skin showed marked changes and was significantly reduced compared to the control group. In the groups of ZnO NPs (0.1 and 0.5) alone and in combination with thiamine, thickness reduction was partially compensated and showed a sensible increase compared to the untreated diabetic group.

Zinc has been shown to improve glucose metabolism by increasing hepatic glycogenesis through insulin signal transductions [38]. When zinc complexes are administered orally, a tendency to lower blood glucose is also observed [39]. In the present study, intraperitoneally administration of ZnO NPs alone and along with thiamine displayed that values of serum parameters of fasting blood glucose, Cr, BUN, GGT, MDA, and NO improved compared to the untreated diabetes group. However, hyperglycemia manifested its effects by increasing Cr, BUN, GGT, MDA, and NO levels, as a destructive agent. Some studies have been presented about the acute effects of high-dose ZnONPs on circulating blood glucose levels and have been shown that high doses of ZnONPs (e.g. 10 mg/kg) can limit the therapeutic utilizations at diabetic patients [25]; but the anti-diabetic effects of ZnONPs have been affirmed in low doses [2, 40]. As the present study showed, the use of ZnO NPs in low doses (0.1 and 0.5) was useful as an anti-hyperglycemic agent and be effective in controlling the destructive effects of diabetes.

The available data about the mechanism of diabetes effect on thiamine processes are not complete and the precise mechanism is unknown; however, some research has suggested some mechanisms. Among the several enzymes involved in carbohydrate metabolism, thiamine (vitamin B1) is an essential factor that plays a vital role in carbohydrate metabolism [41]. Patrini et al., (1996) showed that diabetes causes insulin deficiency, and following, the rate of thiamine transfer among the intestine is reduced. They found that rats with insulin deficiency showed noticeably reduced transfer of free thiamine and monophosphate [42]. Inversely, when the amount of the thiamine decreases (both intake and absorption), insulin synthesis and secretion are significantly disrupted [43].

Conclusion

Hyperglycemia as a definitive marker in the challenge with normal conditions altered serum biochemical parameters and stereological factors in the skin. According to the stereological and biochemical findings, from one-dimension ZnO NPs and another dimension thiamine, was shown to act as anti-diabetic agents in combination together.

Abbreviations

GGT, Gamma-glutamyl Transferase; BUN, Blood Urea Nitrogen; Cr, Creatinine; MDA, Malondialdehyde; ZnO NPs, Zinc Oxide Nanoparticles; ZnO, Zinc Oxide; COA, Acetylsalicylic Acid; NADPH, Nicotinamide Adenine Dinucleotide Phosphate; IP, Intra-Peritoneal; H&E, Hematoxylin and Eosin; TBA, Thiobarbituric Acid

Declarations

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Conflict of interests

The authors do not declare any conflict of interest and confirm the content of this article. This research was funded by the research findings of the veterinary faculty. All the procedures are approved by the committee of animal's ethical rights of the faculty.

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Tables

Table 1: The principal physical characteristics of the ZnO NPs (production reported)

ASP	10-30 nm
Crystal phase	Single crystal
Color	White
Morphology	Nearly spherical
True density	5.606g/cm ³
SSA	20-60 m ² /g

*Table 2: Quantitative parameter of volume density of collagen bundles (V_v ; %), (Mean±SD)

groups		Collagen Volume Density, %	<i>P Value</i>
g I	Control	37.5 ^a	-
g II	Thiamine	36.8 ^a	-
g III	Diabetes	33.5 ^b	<i>P</i> <0.05
g IV	D + ZnO NPs (0.1 mg/kg)	36.4 ^a	<i>p</i> >0.05
g V	D + ZnO NPs (0.5 mg/kg)	36.2 ^a	<i>p</i> >0.05
g VI	D + ZnO NPs (0.1 mg/kg) + Thiamine	35.9 ^a	<i>p</i> >0.05
g VII	D + ZnO NPs (0.5 mg/kg) + Thiamine	36.2 ^a	<i>p</i> >0.05

^{a-b} Means in the same column with different letters are significantly different (*P*<0.05); *The table above shows the values measured 20 days after diabetes induction on mice; D=Diabetes.

*Table 3: The amounts of blood factors (means±SD) in all the groups during the study period

Groups		Glucose (mg/dl)	Cr (mg/dl)	BUN (mg/dl)	GGT (U/L)
g I	Control	82.33±4.10 ^a	0.23±0.02 _a	43.83±5.89 ^a	6.31±0.65 ^a
g II	Thiamine	80.80±3.61 ^a	0.21±0.03 _a	44.40±5.47 ^a	6.42±0.57 ^a
g III	Diabetes	351.32±4.17 ^b	0.89±0.12 _b	86.74±7.10 ^b	12.15±1.56 _b
g IV	D + ZnO NPs (0.1 mg/kg)	150.87±5.51 ^c	0.52±0.02 _c	63.62±5.82 ^c	9.91±1.09 ^c
g V	D + ZnO NPs (0.5 mg/kg)	162.66±3.67 ^c	0.47±0.03 _c	67.50±6.93 ^c	8.89±0.91 ^c
g VI	D + ZnO NPs (0.1 mg/kg) + Thiamine	123.16±7.38 ^c	0.36±0.02 _c	55.23±5.36 ^c	8.06±0.87 ^c
g VII	D + ZnO NPs (0.5 mg/kg) + Thiamine	114.2±4.63 ^c	0.32±0.02 _c	54.42±5.34 ^c	7.57±0.86 ^c

^{a-c} Means in the same column with different letters are significantly different (*P*<0.05); *The above table shows values measured 20 days after induction of diabetes in mice; D: diabetes.

*Table 4: The concentrations of MDA and NO in all the groups (means±SD)

groups		MDA (nmol/ml)	NO (μmol/ml)	<i>P Value</i>
g I	Control	3.29±0.38 ^a	0.21±0.04 ^a	-
g II	Thiamine	3.31±0.47 ^a	0.22±0.03 ^a	-
g III	Diabetes	5.97±0.62 ^b	0.80±0.05 ^b	<i>p</i> <0.05
g IV	D + ZnO NPs (0.1 mg/kg)	4.86±0.44 ^a	0.45±0.04 ^a	<i>p</i> >0.05
g V	D + ZnO NPs (0.5 mg/kg)	4.72±0.35 ^a	0.42±0.03 ^a	<i>p</i> >0.05
g VI	D + ZnO NPs (0.1 mg/kg) + Thiamine	4.01±0.33 ^a	0.40±0.05 ^a	<i>p</i> >0.05
g VII	D + ZnO NPs (0.5 mg/kg) + Thiamine	3.85±0.32 ^a	0.38±0.04 ^a	<i>p</i> >0.05

^{a-b} Means in the same column with different letters are significantly different (*P*<0.05); *The table above shows the values measured 20 days after diabetes induction on mice; D=Diabetes.

Figures

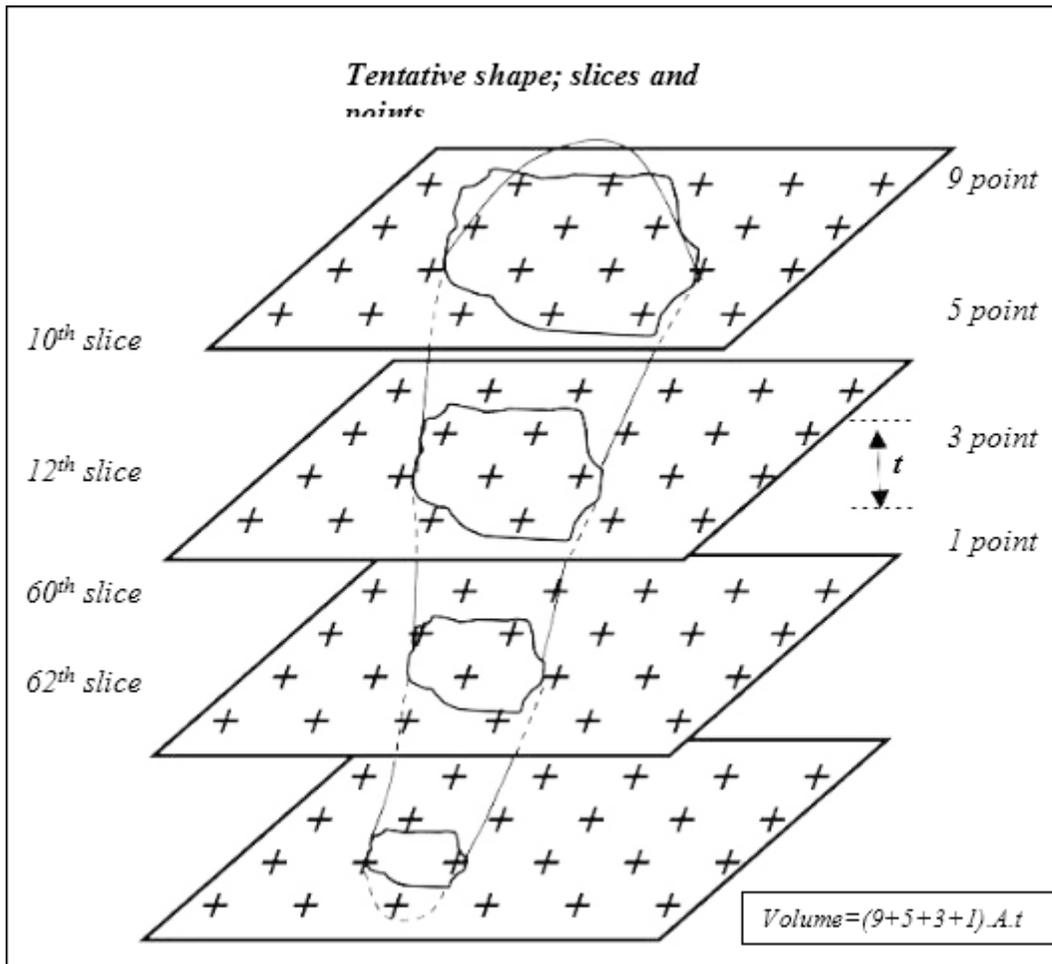


Figure 1

Illustration of the sample scheme of skin tissue sections. Systematic random samples from skin slices superimposed by a grid of the point square to obtain volume of uniform random samples. Samples were prepared by a series of parallel cutting plates (serially sections) with slice thickness t distance. The number of serial sections on which the point grid is superimposed, includes four slices.

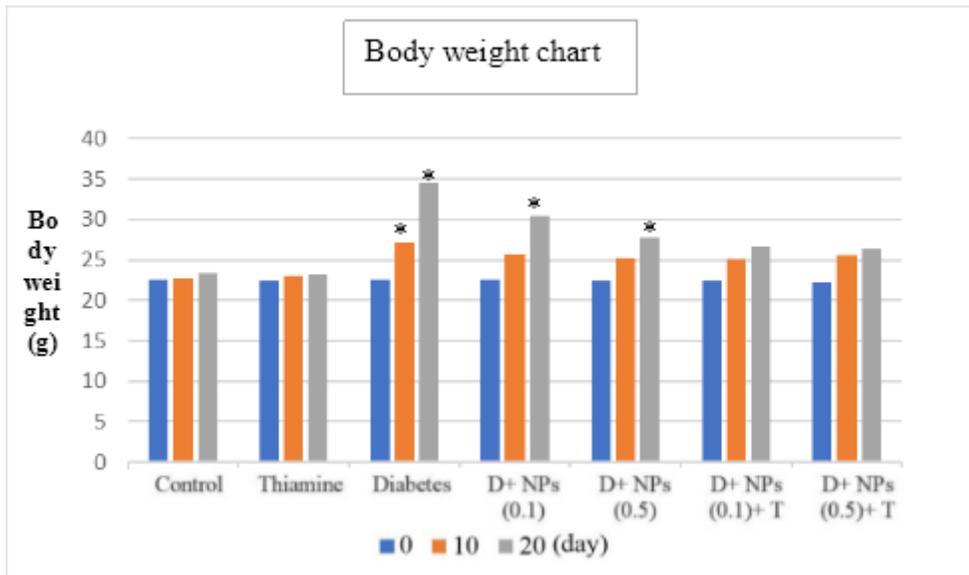


Figure 2

The mean changes from day 0 in body weight (g), for all controls plus diabetes (untreated and treated); D: Diabetes; T: Thiamine, NPs: ZnO nanoparticles; mean \pm SD (n=7); *significant difference in each group with day 0 and control, $p < 0.05$.

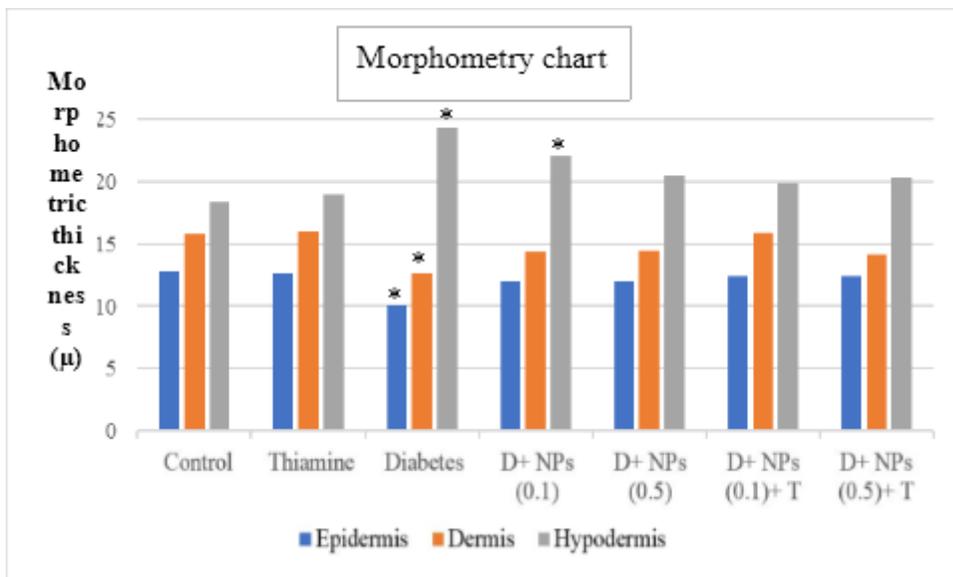


Figure 3

The morphometrical parameters of measured in different layers of cutaneous tissue in experimental groups; mean \pm SD (n=7); * significant difference in each parameter with the control group; $p < 0.05$.