Effect of Ventilation Quantity on Electron Transfer Capacity and Spectral Characteristics of Humic Substances During Sludge Composting

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Abstract

Humic substances (HSs) plays an important role in the degradation of organic pollutants and the passivation of heavy metals in soil, and their environmental effects are closely related to their electron transfer capacity (ETC). However, the effect of ventilation quantity on ETC of compost-derived HA is unclear. In this study, HS was extracted from compost with different ventilation quantities (0.1, 0.2, and 0.3 L kg⁻¹ dry matter (DM) min⁻¹, denoted as VQ1, VQ2, and VQ3, respectively) and divided into humic acids (HAs) and fulvic acids (FAs). Their ETC and electron recycling rate (ERR) were measured by electrochemical method. The structure and composition of HS and its evolution in composting process were analyzed by excitation–emission matrix spectroscopy, Fourier transform infrared spectroscopy, and ultraviolet and visible spectrophotometry. The influence of HS composition characteristics on ETC and ERR was investigated by correlation analysis. Results showed that the increase in the ventilation quantity did not change the composition structure of HS, but it affected the formation of HS. With the composting process, the electron donating capacity (EDC), electron accepting capacity (EAC), ETC, and ERR of the three piles VQ1, VQ2, and VQ3 showed an increasing trend. By contrast, the pile VQ2 with ventilation quantity of 0.2 L kg⁻¹ DM min⁻¹ enhanced EAC, ETC, and ERR of HA and FA mainly because of the accelerating consumption of lignin, protein-like, and aliphatic substances and the generation of aromatic and fulvic-like substances. Such transformation improved the humification degree. Correlation analysis showed that EDC, ETC, and ERR were correlated with the generation of FA, and EAC was correlated with the generation of HA.

1. Introduction

Composting is an effective method to treat organic solid waste (Chen et al., 2020). Compost products not only play an important role in promoting plant growth, improving soil properties, and enhancing soil carbon sequestration (Jurado et al., 2015; Piccolo et al., 2004; Said-Pullicino et al., 2007) but also have great potential in remediation of soil pollution due to humic substances (HSs) derived from the composting process (Madejón et al., 2016; Yuan et al., 2016).

HS is a complex macromolecular organic matter, including humic acid (HA), fulvic acid (FA), and humin (Guo et al., 2019). HS has electron transfer capacity (ETC) because it has a large number of functional groups with redox activity. HS can transfer electrons between microorganisms and terminal electron receptors as electron shuttles, which mediates the passivation of heavy metals and the degradation of refractory organic pollutants; thus, HS plays an important role in environmental pollution remediation (Wu et al., 2017). Examples of this method are reduction of Cr(VI) to Cr(III) (Yang et al., 2020), reduction of Fe(III) to Fe(II) (Stern et al., 2018), reduction of nitrobenzene to aniline (Yuan et al., 2017), and reduction and dechlorination of pentachlorophenol (Zhao et al., 2020). HS also has the characteristic of transferring electrons repeatedly and can further accelerate the transformation rate of pollutants (Xu et al., 2009). Electron recycling rate (ERR) is often used to measure the durative of electron transfer (Yuan et al., 2012). The application of compost in soil remediation actually depends on the ETC and ERR of HS in compost. The value of compost in soil remediation is higher when the ETC and ERR values are stronger (Zhang et

al., 2021). Therefore, ETC and ERR of HS are important indicators to determine its environmental effect (Azim et al., 2018).

Ventilation quantity is an important parameter that affects composting process (Gao et al., 2010). Insufficient ventilation can cause anaerobic phenomenon, but excessive ventilation can decrease the compost temperature and may not be conducive to the degradation of organic matter (Ahn et al., 2007). The heat and temperature of the reactor can be adjusted by regulating the ventilation volume; thus, the degradation of organic matter and the formation of HS are affected (Chen et al., 2020). Most current studies on ventilation quantity focus on the effect of ventilation quantity on physicochemical properties, such as maintaining the temperature of compost, accelerating the degradation of organic matter, and improving germination index (Zang et al., 2016). However, the effect of ventilation quantity on the ETC of HS during composting is unclear.

The main objectives of this study were as follows. (1) The evolution of ETC during composting with different ventilation quantities was investigated by electrochemical method. (2) The changes in organic composition and structure during composting with different ventilation quantity were analyzed using spectral analysis method to clarify the key affecting factors of ETC of HS. The results provide scientific basis for the application of compost products and the improvement in pollution remediation efficiency of compost products.

2. Materials And Methods

2.1 Composting process and sample collection

Municipal sludges were from a sewage treatment plant in Guilin, China. The sludge and rice bran were evenly mixed at 3:1 (m/m) ratio and then placed into three identical reactors (total effective volume of approximately 50 L, $\Phi 37 \times 57$ cm) for the continuous aeration composting. The aeration quantities of the three piles were set as 0.1, 0.2, and 0.3 L kg⁻¹ DM min⁻¹, which were denoted as VQ1, VQ2, and VQ3, respectively. Each pile was turned over once a week. Samples were collected on days 0, 3, 6, 9, 12, 15, 21, 30, and 40. One part of the samples was stored directly in a -20 °C freezer, and the other part was freeze-dried and passed through a 60-mesh sieve. After screening, the samples were stored in the -20 °C freezer for further analysis.

2.2 Extraction and separation of the HA and FA

The extraction and evaluation methods of HS refer to the methods of Song (Song et al., 2015a) and Zhou (Zhou et al., 2014). The initial extraction of HS was by lye extraction. The extracted lye was 0.1 mol/L Na₂P₂O₇ and 0.1 mol/L NaOH mixed in 1:1 solution, and then, it was placed into the horizontal constant temperature oscillator to oscillate for 24 h. The mixture was centrifuged at 10,000 rpm for 20 min, and the centrifuged supernatant was filtered through a 0.45 µm filtration membrane. The precipitate obtained after centrifugation (4,000 rpm for 15 min) was crude HA, and the supernatant was crude FA after the pH of the filtrate was adjusted to 1–2.

The HA was redissolved with 0.1 mol/L NaOH, supernatant was collected after centrifugation (10,000 rpm for 20 min), and pH of supernatant was adjusted to 1–2 with 6 mol/L HCl, followed by static settlement for 6 h. It was centrifugated again, and the supernatant was discarded. Pure HA was obtained after four times washing with distilled water and 0.1 mol/L HCl. The crude FA was passed through the XAD-8 resin column, and the FA was trapped in the resin and then eluted in the resin column with 0.1 mol/L NaOH. The solution was passed into IR120 hydrogen cation exchange resin column to eliminate the interference of sodium ion. Purified FA was obtained after being freeze-dried. HA and FA were stored in a 4 °C refrigerator for further analysis (Ma et al., 2001).

2.3 ETC and ERR measurements

The ETC includes electron accepting capacity (EAC) and electron donating capacity (EDC), both of which can be measured by electrochemical methods (Bi et al., 2013). Electrochemical measurements were performed on an electrochemistry workstation CHI660 (Chenhua Co., Ltd., Shanghai, China). The electrochemical detection method uses a three-electrode electrolytic cell device with working electrode (glassy carbon electrode) opposite electrode (platinum column electrode) and reference electrode (Ag/AgCl electrode) (Tang et al., 2019). The purified HA and FA were determined by voltammetry, and 2 g/L ABTS and 2 g/L DQ were used as mediators for oxidation and reduction, respectively (Bi et al., 2013). During the detection, the electrolytic device should be installed first, and 2 mL neutral phosphate buffer and 2 mL KCl solution should be added to the electrolytic cell, and nitrogen should be used for 5 min to drain oxygen. Nitrogen should also be used during the detection to ensure the stable reading of the detection data. The oxidation potential and reduction potential were set at 0.61 and –0.49 V, respectively, for baseline detection. When the baseline was stable, 1 mL of mediating agent was added into the measuring device for further detection (ABTS was added for EDC detection, and DQ was added for EAC detection). After the reaction was stable, diluted HA and FA samples were added to keep TOC at 50 mg/L. The ETC was calculated using the following formula:

$$\text{ETC(EDC or EAC)} = \frac{A_p}{(e \times N_A \times M_C)}$$

The unit of EAC and EDC was $\mu\text{mol e}^- \cdot (\text{g C})^{-1}$, A_p was the Coulomb integral of compost organic matter oxidation curve or compost organic matter reduction curve, and the unit of Coulomb was (C), N_A was Avogadro constant ($6.02 \times 10^{23} \text{ mol}^{-1}$), e was the charge per unit electron ($1.6 \times 10^{-19} \text{ C}$), M_C was the content of carbon in the reaction system ($3.5 \times 10^{-4} \text{ g}$), and ETC of compost sample was the sum of EDC and EAC.

ERR refers to the number of electrons that HS can steadily accept and supply as an electron shuttle. ERR is commonly used to represent its detection device. The potential-step method was used to detect HA and FA, and 0.5–1 mL of purified HA and FA were diluted and added into 2 mL DMSO and 1 mL in the electrolyte solution of 1 mmol/L sodium perchlorate. The total TOC was kept at 50 mg/L, the oxidation

and reduction potentials were set at 0.61 and −0.49 V, respectively, and the number of cycles was set to 3. The ERR was calculated using the following formula:

$$\text{ERR} = \frac{Q_{\text{cycle 2}} + Q_{\text{cycle 3}}}{2Q_{\text{cycle 1}}} \times 100\%.$$

2.4 Structure and component characterization of organic compounds

2.4.1 Fluorescence spectra

The composition and evolution of HS in composting process are often detected by excitation–emission matrix (EEM) spectroscopy (Marhuenda-Egea et al., 2007). In this study, the EEM fluorescence spectra were recorded by an F98 fluorescence spectrophotometer (Linguang, Shanghai, China). The excitation wavelength (Ex) was set at 200–450 nm, and the emission wavelength (Em) was set at 280–550 nm. Slit broadband was 5 nm. The scanning speed was set at 3000 nm/min.

The fluorescence spectra image of HS was transformed into a matrix, and MATLAB R2018 was used for PARAFAC analysis. DOMFluor and EEMscat toolkit (www.models.life.ku.dk) were used to separate the components and calculate the content of each component in the sample. Before separation, the blank sample was deducted to reduce the influence of Raman scattering (Stedmon et al., 2003). Humification index (HIX) and fluorescence index (FI) were calculated by pickpeaks function. HIX was an important index to reflect the humification degree of compost, while FI could reflect the contribution degree of aromatic substances and nonaromatic substances to compost (He et al., 2013; Wang et al., 2016).

2.4.2 Ultraviolet and visible spectrophotometry

Ultraviolet and visible (UV–Vis) spectrophotometry was performed with UV-2820 UV–Vis spectrophotometer (Unique, USA). TOC was diluted to 20 mg/L before scanning, and pure water was used as blank control group. The scanning wavelength ranged from 190 nm to 700 nm, and the scanning interval was 1 nm.

2.4.3 Fourier transform infrared spectroscopy

Fourier transform infrared spectroscopy (FT-IR) using an infrared spectrophotometer (Is10, ThermoFisher, USA) for detection. Freeze-dried sample and KBr after grinding were mixed with a mass ratio of 1:100 and then ground with agate mortar. The mixed samples after grinding were placed into a tablet pressing mold for pressing. The pressed glass slides were uniformly transparent and placed into infrared spectrograph to determine the scanning range of 450–4000 cm^{-1} , the number of scanning was set to 14, and the resolution was 2 cm^{-1} .

2.5 Data analysis

Data statistical analysis was conducted using SPSS 16.0 and Origin software.

3. Results And Discussion

3.1 ETC and ERR of HA and FA

The ETC of HA and FA samples are shown in Fig. 1. The EDC and EAC of HA show an overall upward trend. The EDC of HA of VQ1, VQ2, and VQ3 gradually increases from day 0 (41.52, 39.41, and 38.06 $\mu\text{mol e}^-/\text{g C}$, respectively) to the maximum on day 40 (44.15, 57.12, and 57.66 $\mu\text{mol e}^-/\text{g C}$, respectively); the EAC gradually increases from day 0 (40.45, 39.76, and 39.91 $\mu\text{mol e}^-/\text{g C}$, respectively) to the maximum on day 40 (48.02, 55.96, and 52.90 $\mu\text{mol e}^-/\text{g C}$, respectively). The increase in EDC and EAC may be due to the oxidation of lignin-like matter and the increases in phenols, quinones, and carboxyl groups in HA during composting (He et al., 2019). The increasing trend of the ETC of HA in the three piles is similar to that of EDC and EAC; it reaches the maximum value on the 40th day, and it reaches 92.17, 113.08, and 110.56 $\mu\text{mol e}^-/\text{g C}$ for the three piles, respectively. The ETC of HA in VQ2 and VQ3 is stronger than that in VQ1, while the difference between VQ2 and VQ3 is not obvious.

The EDC of FA fluctuates from day 0 (24.22, 26.92, and 27.37 $\mu\text{mol e}^-/\text{g C}$) to day 40 (31.95, 35.34, and 34.56 $\mu\text{mol e}^-/\text{g C}$), which may be due to the degradation of organic matter that increases the reducing group content in FA. EAC increases from day 0 (40.45, 24.83, and 26.68 $\mu\text{mol e}^-/\text{g C}$) to day 40 (48.02, 33.84, and 36.54 $\mu\text{mol e}^-/\text{g C}$), which is similar to EDC. The EAC of FA increases gradually, which is possibly due to the decomposition of protein-like substances and aromatic polycondensation during the composting process (He et al., 2019). The ETC of FA in the three piles shows an increasing trend, and it reaches the maximum value at day 40 (65.3, 72.45, and 71.86 $\mu\text{mol e}^-/\text{g C}$).

In summary, the ETC of FA and HA in VQ2 is stronger than that in VQ1 and VQ3, which may be due to that VQ2 improves the degradation and humification process and promotes the oxidation of organic matter and the formation of HSs. In addition, a slight difference is observed in the contribution of EAC and EDC to ETC in HA and FA, which indicates that the oxidation and reduction capacities of HS are the same. This result is different from those of the study of He et al. (2014).

The ERR of HA and FA in the three piles is shown in Fig. 2. HA and FA show a slightly declining trend first and then a rising trend. The ERR of HA is in the range of 45.82–70.59%, and that of FA is in the range of 51.88–72.37%. Only a slight difference is observed between FA and HA in ERR. However, FA has a slightly higher ERR than HA. This result may be due to the fact that FA contains a large number of unstable electron groups and has a stronger cyclic reaction capacity.

3.2 Composition and structure evolution of HS and its influence on ETC

3.2.1 EEM fluorescence spectra

Four components were identified through PARAFAC analysis (Fig. 3). Component 1 (C1, Ex=345 nm, Em=430 nm) shows one excitation peak and one emission peak ascribe to HA peak (Ex=300–370 nm, Em=420–510 nm). The result is similar to the humic-like substance peak discovered by Zhou et al. (2019). Component 2 (C2, Ex=400 nm, Em=485 nm) also shows a single excitation peak and an emission peak located in the humic-like substance peak region, but its Ex has a blue shift. Component 3 (C3, Ex=320 nm, Em=389 nm) peak is located in the traditional fulvic-like region (Ex=310–369 nm, Em=370–450 nm), which corresponds to fulvic-like substances. The peak of component 4 (C4, Ex=240 nm, Em=351 nm) is located in the T-peak region (Ex=270–280 nm, Em=320–350 nm) with blue shift in Ex and red shift in Em, which mainly reflects protein-like substances such as tryptophan (Zhang et al., 2021).

HA and FA of the three piles can be divided into four components by PARAFAC analysis, and content changes of different components are shown in Fig. 4. The C4 of the HA decreases gradually with the composting process, which implies that protein-like substances are degraded by microorganisms, and the C4 content in VQ1, VQ2, and VQ3 decreases to 4%, 1%, and 4%, respectively. By contrast, VQ2 has a higher utilization efficiency of organic matter. The content of C4 in FA increases first and then decreases, but the content of C4 in FA of the three piles decreases to the minimum at the end of composting.

The content of C3 in HA increases first and then decreases, and it increases slightly at the end of composting compared with the initial value, which may be due to the formation of FA from small molecular substances in composting process. Then, the FA gradually transforms into humic-like substances or HS precursors. However, a slight change is observed in C3 content of FA in the three piles, which may be due to the continuous conversion and degradation of FA (Huang et al., 2021). The content of C2 of HA and FA in the three piles increases, which means the gradual formation of humic-like substances during composting. The C1 of HA at the end of composting is insignificantly different from the initial value, but the C1 of FA shows a significant increase, which is probably due to C1 being closer to the fulvic-like substance in the fluorescence composition. The formation rates of C1, C2, and C3 in VQ2 are the highest, which implies that the degradation of protein and the production of humic-like substances are enhanced in pile VQ2.

The HIX of HA and FA in piles VQ1, VQ2, and VQ3 increases steadily (Fig. 5), and the humification degree of HA is higher than that of FA. Therefore, the composting process increases the humification degree of HA and FA (Jacquin et al., 2017). As shown in Fig. 5, the FI of all three piles gradually increases (McKnight et al., 2001). This result may be due to the microbial degradation and polymerization of organic substances into aromatic structure, such as aromatic groups and benzene ring, which improves the aromatic properties of HA and FA. The FI value of FA is higher than that of HA, which means that aromatic substances in FA contribute more to compost maturity.

3.2.2 UV–Vis spectra

The UV–Vis spectra of HA and FA of the three piles are shown in Fig. 6. The UV–Vis absorption values of the three piles have similar trend, and all of them tend to decrease with the increase in wavelength. HA and FA have an absorption peak at approximately 260–280 band, which is caused by the light absorption

of lignosulfonate and its derivatives in the sample, and its absorbance value will increase with the continuous generation of unsaturated conjugated bond $C=C/C=O$ (Li et al., 2010). Except for FA of VQ3, the absorption values of HA and FA at the absorbance of 280 nm increase, which implies that the aromaticity and unsaturation of HS increases, and the degree of humification increases with the progress of composting.

The $SUVA_{254}$ and the $SUVA_{280}$ are generally used to represent the content change of unsaturated $C=C$ bond and the change in aromatic molecular weight (Albrecht et al., 2011). During composting, the condensation of lignin and amino acid will increase the HS content, and the absorption value of the $SUVA_{254}$ will also increase as well (Song et al., 2015b). The absorption value of the $SUVA_{280}$ is related to the molecular weight of organic matter (Ren et al., 2019). Both indicators can represent degree of humification. As shown in Fig. 6, $SUVA_{254}$ and $SUVA_{280}$ of HA and FA for VQ1 and VQ2 show an overall increasing trend, which implies that their HS content is increased. The HA of VQ3 shows a fluctuating rise, but its FA shows a fluctuating decline. According to EEM spectrum analysis, this phenomenon may be due to that excessive aeration in the early stage of compost results in a large amount of FA of VQ3, but the unstable structure of FA leads to a low conversion rate in the stable period.

The E_{250}/E_{365} is the ratio of lignin to carboxylic acid, and it is often used to indicate the molecular weight of organic matter (Strobel et al., 2001). The E_{253}/E_{203} can be used to characterize the types of substituents on the aromatic ring. A lower E_{253}/E_{203} value indicates that the substituents are stable groups such as aliphatic chains, while a higher E_{253}/E_{203} value indicates that the substituents are active groups such as hydroxyl, carboxyl, and carbonyl groups (He et al., 2014). As shown in Fig. 7, the E_{250}/E_{365} value of HA and FA of the three piles decreases with the composting process. The E_{250}/E_{365} value of HA (2.54–2.56) at the end of composting is lower than that of FA (3.14–3.66), and the decrease in VQ2 is larger, which indicates that more macromolecular HAs are generated in the HS of VQ2 (Said-Pullicino et al., 2007). As shown in Fig. 6, the E_{253}/E_{203} value of HA fluctuates during the composting process, and its increase is most significant in VQ2. Unlike HA, the E_{253}/E_{203} value of FA in all the three piles shows a continuous upward trend. The interaction between HS and heavy metals in soil mainly depends on the transfer of electrons of active groups, and remediation of soil pollution can be achieved through chelating and redox effects of heavy metals (Chen et al., 2019). The change in the E_{253}/E_{203} indicates that soil heavy metals can be better passivated by compost products derived from VQ2.

The ratio of the integral of the 275–295 nm absorption band to the integral of the 350–400 nm absorption band is denoted as S_r , which is negatively correlated with the molecular weight of HS and can be used to characterize the degree of aromatization in HS (Spencer et al., 2012). Fig. 9 shows that the S_r value of HA and FA in the three heaps decreases continuously with the composting process, which implies that macromolecular aromatic substances increase continuously during composting.

3.2.3 FT-IR results

The FT-IR results of HA and FA of the three piles are shown in Fig. 9. The absorption peaks of HA and FA in the three piles are mainly $3439\text{--}3280\text{ cm}^{-1}$, $2957\text{--}2917\text{ cm}^{-1}$, $2856\text{--}2842\text{ cm}^{-1}$, $1657\text{--}1646\text{ cm}^{-1}$, $1547\text{--}1513\text{ cm}^{-1}$, 1402 cm^{-1} , $1255\text{--}1241\text{ cm}^{-1}$, and $1043\text{--}1026\text{ cm}^{-1}$. The wave peaks of $3439\text{--}3280\text{ cm}^{-1}$ in the three piles gradually weaken, which means that O–H functional groups such as hydroxyl group on HA and FA are decreasing during the composting process (Peltre et al., 2017; Soobhany et al., 2017; Sun et al., 2014). The absorption peak of $2957\text{--}2917\text{ cm}^{-1}$ may be caused by the C–H antisymmetric vibration of lignin and C–H stretching vibration of aliphatic group. The peak of HA and FA in the three piles weakens and disappears on day 12. Therefore, the fat substances are completely consumed during the cooling period of compost. The wave peaks of $1657\text{--}1646\text{ cm}^{-1}$ is the C=C stretching vibration of aromatic group, and the growth degree of VQ2 and VQ3 in this section is higher than that of VQ1, which proves a higher degree for the humification (Masmoudi et al., 2013). The absorption peak of $1547\text{--}1513\text{ cm}^{-1}$ may be generated by N–H and C=N stretching vibration, FA in the VQ2 and VQ3 performs more significant in this section, and HA performs more significant in VQ2 (Merlo et al., 2020). The peak of 1402 cm^{-1} may be caused by the stretching vibration of carboxylic acids, and VQ1 and VQ3 are stronger than VQ2 at $1255\text{--}1241\text{ cm}^{-1}$, which may be caused by the stretching of lignin-related functional groups. HA and FA of the three piles have little difference at $1043\text{--}1026\text{ cm}^{-1}$, which may be generated by C–O stretching vibration of polysaccharides or alcohols, where VQ2 has the best effect. The analysis of the wave peak above shows that VQ2 has a higher degree of humification.

3.4 Relationship among ETC, fluorescence characteristics, and maturity degree of HA and FA in compost with different ventilation quantities

Correlation analysis was conducted to elucidate the relationship among ETC, ERR, maturity degree, and fluorescence characteristics of HA and FA. As shown in Fig. 10, components in HA of VQ1 are insignificantly correlated with changes in EDC, EAC, and ETC. Meanwhile, ERR and C4 are significantly negatively correlated. This result may be due to that the decomposition of protein-like substances will make HS generate more active reversible groups (Amir et al., 2010). HIX is significantly positively correlated with ERR and negatively correlated with C4. Therefore, the decomposition of small molecules such as proteins during the composting process would increase the degree of humification and improve ERR of HS. FI, Sr, and HIX for FA in VQ1 are significantly positively correlated with EDC, ETC, and ERR, respectively.

Sr is positively correlated EAC with HA in VQ2 ($P<0.01$), and HIX is negatively correlated with EAC in FA ($P<0.01$). Therefore, the formation of HA will promote the enhancement in EAC, while the formation of FA will increase EDC and ETC. The correlation heat map of VQ3 is similar to that of VQ2, which also proves that EDC and ETC are related to FA generation in compost from different composting conditions and EAC is related to HA generation (Li et al., 2021).

4. Conclusion

The EDC, EAC, ETC, and ERR of three piles showed an upward trend with the progress of composting. By contrast, VQ2 enhanced EAC, ETC, and ERR of HA and FA. This result was mainly due to that VQ2 promoted the consumption of lignin protein, the generation of aliphatic and aromatic substances, and the maturity of compost. Correlation analysis showed that EDC, ETC, and ERR were correlated with the generation of FA, and EAC was correlated with the generation of HA. Therefore, sludge compost with ventilation quantity of $0.2 \text{ L kg}^{-1} \text{ DM min}^{-1}$ could enhance the ETC of HSs.

Declarations

Ethics approval Not applicable.

Consent to participate All authors agree to participate in the editing of the paper.

Consent for publication All authors agree to publish this manuscript in the journal.

Author contribution This manuscript was contributed by all the authors. Zhihan Tan: Conceptualization, Xiaojie Sun and Hongxiang Zhu: Funding acquisition, Xiaojie Sun, Beidou Xi: Project administration, Quanyi Ouche and Yuxin Tian: Data curation, software and methodology, Zhihan Tan, Xiaojie Sun, Hongxia Zhang and Xiaosong He: Writing–review & editing. All authors read and approved the final manuscript.

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Availability of data and materials The datasets used or analysed during the current study are available from the corresponding author on reasonable request.

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Figures

Figure 1

Changes in EDC(a, b), EAC(c, d) and ETC(e, f) of HA and FA at different stages of compost

Figure 2

Changes in ERR of HA (a) and FA (b) at different stages of compost

Figure 3

Overview of each component and related fluorescence load obtained after three-dimensional fluorescence parallel factor analysis

Figure 4

Changes in the proportion of HA and FA in each stage of compost humic substances :VQ1 (a, b); VQ2 (c, d); VQ3(e, f)

Figure 5

Changes of HIX index of HA(a) ,FA (b) and FI index of HA (c), FA (d) in compost humic substances

Figure 6

UV-Vis spectrum of HA and FA: VQ1 (a, b); VQ2 (c, d); VQ3(e, f) in sludge compost humic substances

Figure 7

SUVA₂₅₄ of HA(a), FA (b) and SUVA₂₈₀ of HA(c), FA (d) in compost humic substances

Figure 8

E_{250}/E_{365} of HA (a), Fa (b) and E_{253}/E_{203} of HA (c) ,Fa (d) in compost humic substances

Figure 9

Sr variation of HA(a) and FA (b) in compost humic substances

Figure 10

Infrared spectra of HA and FA of VQ1 (a), VQ2(b), VQ3(c) in compost humic substances

Figure 11

Heat map of correlation between HA and FA components and fluorescence electrochemical indices in VQ1, VQ2, VQ3