

Sources and trends of contaminants in a shallow lake in the Mediterranean area from sediment archives of the Anthropocene

Gravina Paola (✉ paolagravi@gmail.com)

Università degli Studi di Perugia: Università degli Studi di Perugia

Bartolomeo Sebastiani

University of Perugia: Università degli Studi di Perugia

Federica Bruschi

University of Perugia: Università degli Studi di Perugia

Chiara Petroselli

University of Perugia: Università degli Studi di Perugia

Beatrice Moroni

University of Perugia: Università degli Studi di Perugia

Roberta Selvaggi

University of Perugia: Università degli Studi di Perugia

Enzo Goretti

University of Perugia: Università degli Studi di Perugia

Matteo Pallottini

University of Perugia: Università degli Studi di Perugia

Alessandro Ludovisi

University of Perugia: Università degli Studi di Perugia

David Cappelletti

University of Perugia: Università degli Studi di Perugia

Research Article

Keywords: Shallow lake, sediment cores, geochemical baseline, trace elements, Pb isotopes

Posted Date: April 22nd, 2022

DOI: <https://doi.org/10.21203/rs.3.rs-1314354/v1>

License:   This work is licensed under a Creative Commons Attribution 4.0 International License.

[Read Full License](#)

1 Sources and trends of contaminants in a
2 shallow lake in the Mediterranean area from
3 sediment archives of the Anthropocene

4 Paola Gravina^{1*}, Bartolomeo Sebastiani¹, Federica
5 Bruschi¹, Chiara Petroselli¹, Beatrice Moroni¹, Roberta
6 Selvaggi¹, Enzo Goretti¹, Matteo Pallottini¹, Alessandro
7 Ludovisi¹ and David Cappelletti¹

8 ^{1*}Department of Chemistry Biology and Biotechnology, University
9 of Perugia, via Elce di Sotto 8, Perugia, 60123, Umbria, Italy.

10 *Corresponding author(s). E-mail(s): paolagravi@gmail.com;
11 Contributing authors: bartolomeo.sebastiani@unipg.it;
12 federica.bruschi@studenti.unipg.it; petrosellichiaara@gmail.com;
13 b.moroni@tiscali.it; roberta.selvaggi@unipg.it;
14 enzo.goretti@unipg.it; matteo.pallottini@unipg.it;
15 alessandro.ludovisi@unipg.it; david.cappelletti@unipg.it;

16 **Abstract**

17 In this study, the anthropogenic contamination in Trasimeno lake (Cen-
18 tral Italy) was investigated using three sediment cores of the last 150
19 years (Anthropocene) to identify the primary sources of pollutions and
20 quantify the level of contaminant enrichment in the basin. To this aim,
21 trace elements, polycyclic aromatic hydrocarbons, and lead isotope ratios
22 have been determined at biannual resolution. The deeper parts of the
23 sediment cores were used to estimate the local geochemical baseline,
24 based on known statistical methods, and to determine the values of
25 enrichment factors. Principal component analysis and factor analysis
26 were exploited to associate chemical proxies to human-driven contamina-
27 tion processes. On these grounds it was possible to identify the timings

28 and impacts of pre-world war industrial productions and more recent
29 activities characteristic of the basin's surroundings, such as agriculture.

30 **Keywords:** Shallow lake, sediment cores, geochemical baseline, trace
31 elements, Pb isotopes

32 1 Introduction

33 Freshwater ecosystems such as lakes and rivers are critical to understanding
34 the effects of the environmental change driven by human activities, which
35 have been particularly relevant in the Anthropocene [Dubois et al \(2018\)](#). To
36 assess the impact of human activities, an undisturbed reference, such as pris-
37 tine natural background sediments, is necessary to reflect the situation of the
38 individual lake on a local or regional scale. A background concentration is
39 defined as the concentration of trace elements before industrialisation, so its
40 level should reflect natural processes uninfluenced by human activities ([Tapia](#)
41 [et al \(2012\)](#); [Tylmann \(2005\)](#)). However, this reference is not easy to obtain
42 because of the rapid population growth, industrialization, and urbanization
43 processes over the past two centuries.

44 In the present work, we focus on Trasimeno, a shallow lake in Central Italy,
45 which, similar to other basins in the Mediterranean area, has been severely
46 impacted by human activities, which have altered its sensitive hydrological
47 status since the beginning of the twenty century and affected the ecosystem
48 and biota ([Goretti et al \(2016\)](#)). Recently, exploiting high-resolution strati-
49 graphic archives ([Gravina et al \(accepted 2022\)](#)), we characterized the three
50 major hydrological regimes of the lake, focussing on the variations of precipita-
51 tion and sedimentation processes of endogenic carbonates, calcite, and calcium
52 carbonate phosphate and relating the presence of these compounds to natural

53 and human-driven processes. This study allowed us to identify the range of
54 the sedimentary archives less impacted by anthropogenic processes.

55 Herein, we started from this piece of knowledge to define a geochemi-
56 cal baseline which allowed us to identify and date the primary sources of
57 pollution affecting the Trasimeno lake and quantify the degree of contami-
58 nant enrichment. Three short sediment cores have been exploited to this aim
59 and characterized in terms of major and trace elements, polycyclic aromatic
60 hydrocarbons, and lead isotope ratios. Data have been further processed by
61 multivariate analysis and discussed. The significant human impacts have been
62 classified into three periods covering the Anthropocene and encompass arti-
63 ficial water level manipulation, industrial activities in the pre-second World
64 War period, and more recent eutrophication related to agricultural practices.

65 2 Materials and Methods

66 2.1 Study area, sampling and processing

67 The Trasimeno Lake is a shallow and closed basin located in Central Italy.
68 Despite its large extension (124 km²), which makes it the largest lake of central
69 and southern Italy, the lake is very shallow, and its bathymetry is very smooth
70 (Figure *SM1 – a*). In the last two centuries, the hydrometric trend of the
71 basin passed through 3 distinct phases: (i) an old phase (OP) (from 1860 to
72 1900), during which the water level exceeded the overflow level (257,5 m a.s.l.),
73 allowing the discharge of suspended materials to the outside; (ii) a middle
74 phase (MP) (from 1900 to 1960), during which the water level remained below
75 the overflow level and also suffered a substantial lowering, causing saturation
76 of the components in the water column; (iii) a young phase (YP) (from 1960
77 to 2010), during which the water level rose but not above the overflow level
78 (Figure *SM1 – b*) The present study has characterized three sediment cores

79 covering the 1860-2010 period and superficial sediments sampled during 2018
80 (Figure *SM1 – a* and table in Figure *SM2*). The cores have been cut at
81 biennial resolution, dated, and finally a portion of each section has been acid
82 digestion or extracted in a proper solvent for the subsequent chemical analyses.
83 All the details of the sampling site and sediment cores processing are reported
84 in ([Gravina et al \(accepted 2022\)](#)).

85 **2.2 Chemical analysis**

86 Major and trace elements (Al, Fe, Ti, V, P and Mn, Co, Ni, Pb, Cr,
87 Cu, Zn) concentrations were determined by Inductively Coupled Plasma an
88 Optical Emission Spectrometry (ICP-OES, Ultima 2, HORIBA Scientific)
89 equipped with an ultrasonic nebulizer (CETAC Technologies, U-5000AT).
90 Analytical wavelengths are reported in Table *SM1*. PAHs have been analysed
91 using a gas chromatography-mass spectrometry with a triple-axis detector
92 (HP7890A/5975CVL - Agilent Technologies, USA), equipped with a low bleed
93 Select-PAH capillary column (Agilent J&W, CP 7461). The determination of
94 $^{208}\text{Pb}/^{207}\text{Pb}$ and $^{206}\text{Pb}/^{207}\text{Pb}$ ratios was performed by an Agilent 8900 ICP-
95 MS/MS (100 version, Agilent Technologies, Japan), operated in the MS/MS
96 mode. The collision reaction cell was pressurized with a NH_3/He 15 % mix-
97 ture, used as a damping gas to lower the RSD. The instrumental mass bias
98 was corrected with the standard sample bracketing method using the lead iso-
99 topic standard SRM 981 from NIST (Gaithersburg, MD, USA) ([Vanhaecke](#)
100 [et al \(2009\)](#); [Bazzano and Grotti \(2014\)](#); [Bazzano et al \(2021\)](#); [Bertinetti et al](#)
101 [\(2022\)](#)). The Supplementary Material (Section 1 - Chemical analysis) reports
102 more details on the analytical methodologies.

103 A standard principal component analysis (PCA) and a factor analysis (FA)
104 have been performed on the chemical dataset with R - Version 1.2.5033 using
105 the packages *tidyverse* and *factoextra*.

106 2.3 Geochemical baseline and enrichment factors

107 In this study, the geochemical baseline (GBL) was obtained using the bottom
108 portions of the 3 sediment cores (approximately 25 cm), considering that these
109 portions represent the least contaminated sediments due to their deposition
110 before 1900. Then, two different GBLs were calculated using passive methods,
111 the relative cumulative frequency technique (RCF) and the linear regression
112 method (LRM).

113 The RCF technique is commonly used to obtain GBL (Matschullat et al
114 (2000); Teng et al (2009)) of an individual element and is based on different
115 slopes of the relative cumulative frequency content fitting curves for the natural
116 origin and abnormal concentration. A bend of the slope in the upper part of
117 the curve (higher values) can be used to distinguish between anthropogenically
118 non-influenced samples (low values) and anthropogenically influenced samples
119 (high values). The baseline is obtained from the data below the first inflection
120 value. The data in the cumulative frequency curve are tested with a linear
121 regression method, until achievement of the condition of $p < 0,05$ and $R^2 >$
122 $0,9$. Otherwise the maximum value is removed and the procedure is repeated
123 until the two criteria are both met.

124 The LRM allows obtaining the regional GBL from deep sediments by plot-
125 ting the element towards the normalizer (Selvaggi et al (2020)) According to
126 the principle of normalization, the relative proportions of metals within mate-
127 rials from a particular region tend to be constant, even if the absolute metal
128 concentration varies between crustal elements from one region to another. In

129 the present study, normalization was done with Al or Fe according to the best
 130 correlation with the element considered. The samples lying beyond the confi-
 131 dence interval of 95% have been labeled as anthropogenically influenced. In the
 132 linear relationship between each element of concern and the reference elements
 133 (e.g. Al or Fe), data lying out of the 95% confidence band are eliminated, a
 134 new linear equation is built with the updated data set and, the process goes
 135 on until all the data are within the 95% confident band.

136 The local GBL was used in the present study to improve the assessment
 137 of anthropogenic impact through the enrichment factor (EF). The EFs are
 138 calculated as:

$$EF = \frac{\left(\frac{[Me]}{[X]}\right)_{sample}}{\left(\frac{[Me]}{[X]}\right)_{reference}}$$

139 where [Me] is the concentration of the element, [X] is the concentration
 140 of the reference element (Al, Fe, V, Li or Sc); at the numerator, there is the
 141 sample, while at the denominator there is the reference baseline (Schropp et al
 142 (1990)). Al was chosen as the reference for the present study.

143 **3 Results and Discussions**

144 **3.0.1 Geochemical Baseline**

145 Two different methods were screened to identify the better GBL as represen-
 146 tative of the maximum pristine state of the sampled Trasimeno sediments, in
 147 order to quantify the enrichment of the trace elements through the EF factors.

148 An example of the RCF values provided by cumulative frequency curves
 149 method are plotted in the Figure 1 for the cases of Co and Cr (other ele-
 150 ments are in Supplementary Material in Figure SM4). Figure 2, instead, shows
 151 the linear regression method for calculating GBL, where Al or Fe are chosen

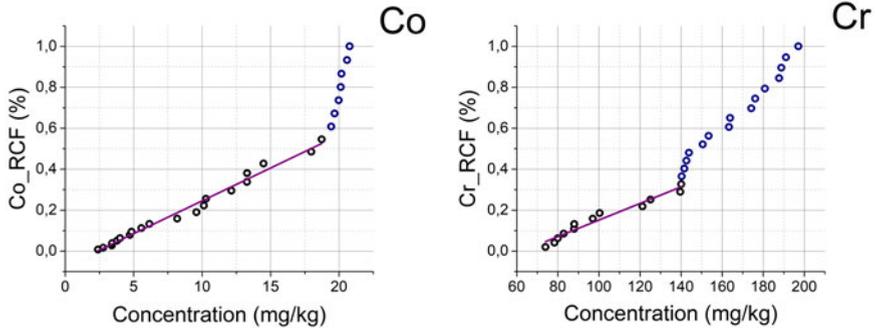


Fig. 1 Cumulative frequency curves (scatters) of Co and Cr. Co represents the curve with only 1 inflection point, while Cr represents the curve with 2 inflection points. Linear regressions were performed on the cumulative frequency curve with $p < 0.01$ and $R^2 > 0.9$ (purple line).

152 according to the highest correlation value with the element considered. Both
 153 GBLs obtained through RCF and linear regression methods are shown in the
 154 Table 1. Eventually, Background (Bkg) values, also reported in the Table 1,
 155 were calculated by arithmetic average of the bottom part of the cores (portion
 156 before 1900s corresponding to the pre-industrial period).

157 As shown in Table 1, the RCF is the closest lacustrine sediment baseline to
 158 other baseline values from different areas of the world (e.g. Callender (2014)).
 159 As regards Ni, Cd and Cr the RCF baselines are higher than those reported by
 160 Callender (2014). While the RCF baselines for Pb, Cu and Zn are lower than
 161 Callender (2014) values, being 11, 16 and 70 ppm compared to 22, 34 and 97
 162 ppm, respectively.

163 3.0.2 Enrichment factors and trends of contaminants

164 EF values were calculated using the RCF baseline, considering that, even if
 165 widely used as a sediment metal-enrichment assessment tool, the calculated
 166 EF is highly dependent on the choice of an appropriate background or base-
 167 line as reference level (Rubio et al (2000)). Interpretation of EF values allows
 168 the assessment of contaminant enrichment. According to Birch et al (1996),

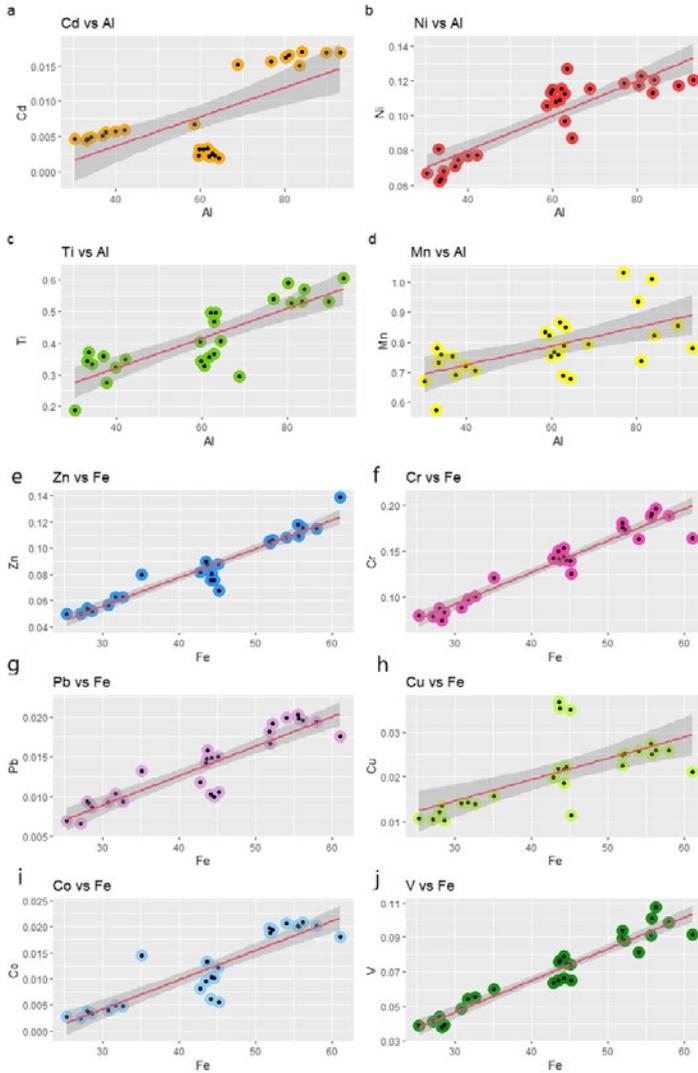


Fig. 2 Normalisation of heavy metals on the reference element: Al (from *a* to *d*) and Fe (from *e* to *j*) of bottom dataset. Concentration are in mg/kg. The linear regression was obtain with $p < 0.01$ and R^2 greater. The dark grey area represents the 95% confidence.

169 there were several scenarios : $EF < 1$ "no enrichment", $1 < EF < 3$ "minor
 170 enrichment", $3 < EF < 5$ "moderately enrichment", $5 < EF < 10$ " moder-
 171 ately enrichment", $10 < EF < 25$ "severe enrichment" $25 < EF < 50$ "very

172 severe enrichment" and $EF > 50$ "extremely severe enrichment". The EFs val-
173 ues were also interpreted by [Zhang and Liu \(2002\)](#): when EF is between 0.5
174 and 1, the metal could be mainly from the weathering process; while if EF is
175 greater than 1.5, the metal is from anthropogenic sources or grater percentage
176 of the metal is from no-natural process.

177 The boxplot, reported in Figure SM5 in Supplementary Material, shows
178 means, high and low quartiles and outliers for the EF values in the 3 sediment
179 cores (C1,C2,C3).

180 The enriched elements in C2 (Figure 3), listed in descending order, are
181 Co, Pb, Zn, Mn, P, Cu, Ti, Cr, V, and Ni. The enriched elements, listed in
182 descending order for C1 core are Cu, Pb, Co, Ni, Zn, Cr, V, Mn, P, and Ti;
183 while for C3 core are Pb, Zn, Cu, Ni, Cr, V, Mn, Co, Ti, and P. The global
184 anthropogenic enrichment in the basin seems to be high, especially for Pb, Cu,
185 Co, and Zn.

186 **3.0.3 Trends of polycyclic aromatic hydrocarbons**

187 Specific organic proxies such as some polycyclic aromatic hydrocarbons (PAHs)
188 allow to identify their possible source, which can be petrogenic, pyrogenic, bio-
189 genic, and diagenetic ([Abdel-Shafy and Mansour \(2016\)](#)). The pyrogenic kind
190 of PAHs is linked to exposure to high temperatures under low oxygen or no-
191 oxygen conditions, such as wood and coal combustion. The petrogenic origin
192 is due to transportation, storage, and spills of crude oil and crude oil products.
193 In contrast, the biological origin of some PAHs is due to their synthesis by
194 certain plants and bacteria or to their formation during degradation of vege-
195 tative matter. For the estimation of anthropogenic sources, Perylene is usually
196 removed from the sum of PAHs, because it is formed after deposition transfor-
197 mation during diagenesis and is derived from natural precursors ([Jiang et al](#)

Methods	GBL / Bkg		Co	Ni	P	Pb	Cr	Cu	Ti	Zn	Mn	V
Relative cumulative frequency methods (RCF)	Core sed	Mean	8.46	72.82	1381,04	10.59	101.23	16.14	325.09	69.69	758.04	64.10
		dev sta	5.14	7.99	157,59	2.53	24.09	4.62	49.73	14.70	69.26	17.23
Linear regression technique - Al	Core sed	Mean		85.83	1798,06				447.61		823.45	
		dev staa		23.87	387,72				143.59		124.12	
Linear regression technique - Fe	Core sed	Mean	13.00			14.91	134.89	19.19		86.20		66.13
		dev sta	7.50			4.75	43.59	5.95		23.85		23.15
Baseline mean 3 bottom cores	Core sed	Mean	11.48	99.11	1680,05	13.73	137.42	20.82	415.41	84.19	784.26	70.48
		dev sta	6.81	21.79	441,74	4.44	39.39	7.74	111.02	24.77	100.77	20.79
Bottom core C1	Core sed	Mean	10.66	108.98	1497,81	13.48	144.11	24.41	406.82	87.17	780.99	73.207
		dev sta	3.70	10.86	77,26	2.63	10.20	8.43	65.242	19.36	64.43	8.595
Bottom core C2	Core sed	Mean	19.93	118.13	2272,85	19.09	182.41	25.36	523.20	110.03	871.76	93.86
		dev sta	0.65	3.14	97,92	1.16	10.92	1.37	97.07	5.37	110.25	8.32
Bottom core C3	Core sed	Mean	4.87	71.23	1292,26	9.23	89.98	12.79	317.27	57.91	710.12	46.68
		dev sta	3.69	6.58	190,90	1.90	14.51	2.00	59.95	9.54	61.97	8.21
Current state	Core sed	Mean	10,7	76,6	418,6	18,8	100,0	18,2	292,5	78,5	1063,9	53,3
		dev sta	3,7	16,2	79,1	3,0	27,6	4,1	104,4	13,1	303,9	17,6
Förstner and Wittmann (1981)	Lake sed	Mean		44.5		27	42	30		65.5		
Pre-industrial Callender (2014)	Lake sed	Mean		40		22	48	34		97		
Recent lacustrine Callender (2014)	Lake sed	Mean		39		102	63	60		207		

Table 1 Concentration of geochemical baseline and background values calculated with different methods. All data are presented as mg/Kg or ppm.

198 (2000)). Therefore, the total PAH ($=\Sigma\text{PAHs}$) is calculated by subtracting the
199 perylene concentration in the following.

200 As shown in Fig. 4 - a), ΣPAHs) was relatively constant until 1920 with an
201 average value of 26.5 ng/g. This value is consistent with those found in surface
202 sediments of relatively pristine environments in the Italian alps (Poma et al
203 (2017)). In the MP hydrological phase we observed a significant increase of
204 ΣPAHs , starting in 1930-40, and reaching values a factor 4 higher in the YP,
205 after WW II with a first maximum in 1965-70 and a second maximum in the
206 late 1990s (Figure 4 - a). On the other side, Perylene remained constant and low
207 during the most recent YP phase but increased rapidly with the depth of burial
208 in sediments due to diagenesis, unlike other PAHs molecules (Figure 4 - b). We
209 exploited specific diagnostic ratios (typically based on isomers) to identify the
210 contamination sources (Roszko et al (2020); Yunker et al (2002)), assuming
211 that different PAH molecules transform and degrade at the same rate during
212 their lifetime in the environment, so that the characteristic concentration ratios
213 of the sources are preserved. A scatterplot of the two PAHs diagnostic ratios
214 obtained for the Trasimeno core sediments, is shown in Figure 4 - c. The
215 Flu/(Flu + Pyr) ratio distinguishes the origin of samples belonging to different
216 historical moments (Guo et al (2011)). The Trasimeno sediment samples from
217 the period of most significant industrial activity, i.e. around the Second World
218 War, indicate oil and liquid fossil fuel combustion sources. Most YP samples
219 represented anthropogenic origin related to biomass combustion and generated
220 by pyrolysis, which can be associated with characteristic agricultural activities
221 of the surroundings of Trasimeno lake in the last part of the 20th century.

3.1 Identification of anthropogenic sources by PCA and FA analysis

A PCA analysis was performed on the C2 central core, which has the largest number of samples and which is more undisturbed and representative of the entire basin. The PCA performed on the C2 included trace elements and Σ PAHs (Figure 5 and Table SM4). The analysis identified 5 groups of tracers: one natural group (A) including Al, Ti, V and Fe and 4 distinct anthropogenic groups (B, C; D and Pb). Group B included Co, Ni, Zn and Cr, group C included Cu and P, and D included Mn and Σ PAHs. In addition, the Mn variable is explained in part also by PC1. Pb is an isolated element that is separated from the groups mentioned above and is explained mainly by PC2.

The score legend of Figure 5 effectively separates the three different hydro-metric phases that have occurred in the lake over the past 150 years (OP, MP and YP), and allows to determine the sources of elements and organic compounds. In particular, the YP scores and PC2 distinguished the anthropogenic element groups (B, C and D groups). The distinct groups of anthropogenic elements have different origins; in fact, the anthropogenic group B is well separated from the natural elements (groups A) by PC2, but is also well isolated from Pb, which is likely associated with atmospheric deposition due to human use of leaded gasoline since the 1960s.

Finally, the results of the PCA analysis individuated a specific type of contamination throughout the lake, influenced by the intensification of agriculture during the eutrophic phase (YP, after 1970). The use of fertilizers, which can contain Mn, Cu, P, Zn, and Cr, Pb, is the main impacting factor. Manganese can be emitted as Mn sulfates in fertilisers or from pesticides, animal feed, or water treatment products. On the other hand, copper is contained in pesticides and fungicides (used to control plant diseases) or is emitted as Cu sulfate

(treatment used to control algal growth). Actually, Cu sulfate is also added to pig feed to suppress bacterial action and therefore it is poured in the soil when pig slurry is used as fertilizer (Panagos et al (2018)). The use of fungicides based on the mixture of Cu sulfate, lime and water started in 1880s and became widely used to control downy mildew in grapevines, while on the global scale, the use of Cu in agriculture has increased since 1980 (Panagos et al (2018)). Ultimately, the use of both Cu and P as fertilizers has been a common procedure for much of the last century (confirming their appearance in group *C*), in contrast, the use of Mn as a fertilizer was not approved in Europe until the early 2000s, and that's why Mn does not belong to group *C* (EFSA Panel (NDA) (2013); (FEEDAP); Christensen (2015)). The increasing trend in PAHs concentrations also correlates with Mn, given the increase of both especially in recent times (confirming the affiliation of PAHs to the group *D*).

Factor analysis (in supplementary material Figure SM7) confirms the source-dependent separation of elements obtained by the PCA. It is observed that Al and Fe are separated from other elements, representing the group of natural origin, while among the elements of anthropogenic origin, there is the subgroup P-Mn.

3.2 Industrial contamination in the middle hydrological phase

The average concentration of the trace elements (V, Co, Ni, P, Pb, Cr, Cu, Ti, Mn and Zn) measured for the three sediment cores (top part, bottom part - used for GBL calculation - and whole core) together with maximum and minimum values are reported in supplementary material in Table SM3. The trends are qualitatively similar for the three cores (Figure SM3). Nevertheless the highest concentrations have been recorded in C2 (central lake) and the

275 lowest in C3 (Southern lake). These differences can be associated with the dif-
276 ferent sedimentation rates (Gaino et al (2012)), the distance from the shore
277 or the centre of the basin, and the different composition of washed-out mate-
278 rial from the Eastern, Western, and Southwestern parts (Figure SM1) of the
279 area surrounding the lake Yang and Rose (2005). Observing data reported in
280 Table SM3 for the central C2 core Ni, Cr and Ti have higher concentrations
281 in the bottom with respect to the top of the core, and the current (2018) con-
282 centration values drop even further. On the other side V, Co, Cu, Zn have
283 comparable values along with the sediment core, but also show lower values in
284 the most superficial and current sample; unlike previous trends, Pb and Mn
285 present lower mean concentrations in the bottom part of the core with respect
286 to the top part and also with respect to the current (2018) conditions. The sit-
287 uation is slightly different for the north-eastern C1 core: Cu, Pb, Co, Mn and
288 Zn show lower mean concentration values in the bottom part than in the top
289 part and whole core while V presents stable values for the whole core record. In
290 the case of the southern C3 core, V, Co, Ni, Pb, Cr, Cu and Zn present lower
291 mean concentration values in the bottom part of the core than in both the top
292 and the whole sediment core; Ti and Mn present relatively stable values for
293 the entire core record.

294 A closer view of the concentration profiles of some specific trace elements
295 (Pb, Ni, Co, Zn, Cr, and V) of the C2 core is shown in Figure 6. The concentra-
296 tion trends are characterised by a clear peak dated between 1930-1945 for all
297 the elements. The peak corresponded to a simultaneous increase of the element
298 concentrations and coincided with a relative rise in the water level (see Figure
299 SM1 - b), even if in the general context of the lowering phase. Overall, these
300 pieces of evidence point to a significant contamination process lasting more or
301 less 20 years. The nature of the trace elements involved suggests an industrial

302 source. Indeed, in this period, a military air-plane industry was operative and
303 very active in Passignano, the main village on the lake's northern coast. This
304 air-plane industry moved from Milan to the surroundings of the Trasimeno's
305 shores in 1916, and one of the most important Italian aeronautical schools was
306 also established close to the basin. Technological development began in 1922
307 when the "*Societa' Aeronautica Italiana*" (SAI) was created and the combat
308 aircraft production increased significantly with the outbreak of World War II
309 (Bellaveglia (2015) ; SAI (2018)).

310 After the peak, most elements lowered in concentration while, in the case
311 of Pb, we notice a growth from the 1960s to the present day. This trend must
312 be associated with a different source and reasonably connected with the use of
313 leaded gasoline, containing Pb as an anti-knockout substance (Resongles et al
314 (2021)). A further investigation for the Pb case has been made by measuring
315 the $^{208}\text{Pb}/^{207}\text{Pb}$ and $^{206}\text{Pb}/^{207}\text{Pb}$ isotope ratios which are proxies of the emis-
316 sion source. Results are shown in Figure 7 for the $^{206}\text{Pb}/^{207}\text{Pb}$ together with
317 the trend of the total Pb concentration. A $^{208}\text{Pb}/^{207}\text{Pb}$ versus $^{206}\text{Pb}/^{207}\text{Pb}$
318 plot is presented in the Supplementary Material (Figure SM8). The results
319 in Figure 7 show a clear lowering of the $^{206}\text{Pb}/^{207}\text{Pb}$ isotope ratio from val-
320 ues larger than 1,230 (1,233 the maximum) previous of 1920, which can be
321 considered a natural situation, towards values less than 1,210 (1,207 the min-
322 imum) in the most recent years. Isotope ratio values can vary easily between
323 different areas depending on the mineralogy of the area and on the level and
324 type of contamination from the surrounding. In general, the values obtained
325 in the Trasimeno Lake are consistent with the literature (Odigie et al (2014);
326 Gobeil et al (1995); Townsend and Seen (2012); Chiaradia et al (1997); Brän-
327 vall et al (2001)). The lowering of the $^{206}\text{Pb}/^{207}\text{Pb}$ ratio plotted in Figure 7
328 started clearly in coincidence with the pollution peak of the 1930-1940 and

329 carried on further with a minimum in the 1980s and a slight recover later on.
330 According to the literature this particular ratio is higher in natural soils, while
331 it decreased because of anthropogenic pollution (Renberg et al (2002); Sakata
332 et al (2018); Chiaradia et al (1997); Bränvall et al (2001)). Moreover, a three-
333 isotope plot can often help identify and differentiate between anthropogenic
334 and natural/geogenic Pb sources, particularly when the naturally occurring
335 and anthropogenically introduced Pb has significantly different isotope ratios
336 (Townsend and Seen (2012)). We reported in the supplementary material
337 (Figure SM8) the correlation between the $^{208}\text{Pb}/^{207}\text{Pb}$ and the $^{206}\text{Pb}/^{207}\text{Pb}$
338 isotope ratios which, consistently, suggest the contribution of at least two end-
339 members to the observed Pb isotope ratio. In conclusion, the isotope ratio
340 results confirm the anthropogenic nature of the contamination in the 1930-
341 1940 event and the successive anthropogenic origin of Pb in the more recent
342 years.

343 **3.3 Impacts of vehicular traffic and agricultural** 344 **contaminations in the young hydrological phase**

345 Pb shows a characteristic enrichment trend, which is in good agreement
346 between the different sampling sites. In particular, the trend shows an increase
347 in concentration during the WWII period that persisted until the early 2000s.
348 As discussed above, Pb enrichment is primarily due to anthropogenic activ-
349 ities such as the use of leaded gasoline and the disposal of sewage sludge to
350 the ground: only core C1 showed a lowering of enrichment after 2000, indi-
351 cating the termination of leaded gasoline use, which came into effect in 2002
352 in Europe. The C1 core is located in the area of the lake most impacted by
353 vehicular traffic due to the presence of a highway, built in the 1965, at 200m
354 from the shore.

355 The intensification of the agricultural activities in the YP is responsible for
356 the release of various elements and pollutants in the environment, such as the
357 essential elements (Cu and Zn) contained in fertilisers, and the heavy metals
358 (Pb and Cr) used in the synthesis of fertilisers as catalysts (Nacke et al (2013)).
359 Moreover, the use of leaded fuel in agricultural machinery for the development
360 of farming operations is another reason for enrichment of Pb (Nicholson et al
361 (2003); Zan et al (2012)).

362 Cu is consistently enriched in the sampled sediments, with maximum val-
363 ues recorded between 1970 and 1980. One of the main sources of Cu in the
364 Trasimeno basin is the use of Cu sulphate pentahydrate in the production of
365 fungicides, introduced in agriculture as early as the 19th century and widely
366 distributed on fruit plants (olives and vines) and vegetables. Other sources of
367 land pollution are sewage sludge, municipal compost and animal waste (Cal-
368 lender (2014)). Mn enrichment began only later in 1965, with a maximum
369 enrichment around the 1990s, confirming the delay in using this element for
370 agricultural purposes towards the end of the century.

371 The current state of the sediment showed that pollution caused by some
372 elements and heavy metals, such as Cu, Co, Ni, Cr, Zn, P and V has returned
373 to baseline levels after increasing during the industrialization, urbanization
374 and agricultural intensification. On the other hand, elements like Pb, Mn, Cu
375 and Zn maintain enriched values relative to the baseline but lower than the
376 period between 1940-1990. Mn and Cu are currently used in the predominant
377 agricultural activities around the Trasimeno lake. The Pb enrichment may
378 be due to the persistence of lead contamination even after the cessation of
379 leaded petrol use in the early 2000s (Resongles et al (2021)). The process of
380 remobilization of historical Pb deposited in soil from the atmosphere can be
381 considered a critical current secondary source (MacKinnon et al (2011)).

4 Conclusion

A set of short sediment cores have been used to identify the natural baseline and quantify the anthropogenic contamination in the context of a fragile shallow lake in Central Italy. The challenge was to disentangle the pollution effects from the human alteration of the hydrological regime of the lake, which during 150 years produced three distinct hydrometric phases. The main proxies identified to investigate the lake pollution have been industrial metals (Cr and Zn), elements used in agriculture (Cu, Mn and P) and some specific organic compounds of anthropogenic and natural origin (PAHs). Pb isotope ratios have also been exploited to attribute the nature of the intense industrial event. In addition, the determination of the local baseline by using deep sediment cores by the RCF statistical method ensured the quantification of sediment enrichment by pollutants. The impact of industrial pollution at the regional level starting before the 2nd World War mainly tied up to the increase in the airplane industry activities in Passignano. Over time, the main causes responsible for the input of pollutants into the lake become intensification of agriculture, livestock, urban and transportation, which have increased since the 1960s. The assumption made to obtain the local baseline seemed adequate, according to the comparison with global baseline values in the literature and the variability of the enrichments describing the human activities around Trasimeno. The assessment of the current condition of the lake demonstrated that some elements such as Cu, Co, Ni, Cr and V, mainly related with industrialization, urbanization and agriculture has returned to GBL levels. On the other side, Mn and Pb, used in intensive agricultural activities and highly persistent in the environment even after being banned at the end of the last century, respectively, remain enriched compared to GBL values.

408 **Acknowledgments.** We thank MIUR and University of Perugia for financial
409 support through AMIS project (“Dipartimenti di Eccellenza–2018–2022”).

References

- Abdel-Shafy HI, Mansour MS (2016) A review on polycyclic aromatic hydrocarbons: Source, environmental impact, effect on human health and remediation. *Egyptian Journal of Petroleum* 25(1):107–123. doi:<https://doi.org/10.1016/j.ejpe.2015.03.011>
- Bazzano A, Grotti M (2014) Determination of lead isotope ratios in environmental matrices by quadrupole ICP-MS working at low sample consumption rates. *Journal of Analytical Atomic Spectrometry* 29(5):926. doi:<https://doi.org/10.1039/c3ja50388g>
- Bazzano A, Bertinetti S, Ardini F, et al (2021) Potential Source Areas for Atmospheric Lead Reaching Ny-Ålesund from 2010 to 2018. *Atmosphere* 12(3):388. doi:<https://doi.org/10.3390/atmos12030388>
- Bellaveglia C (2015) *Aeronautica Sul Trasimeno: Storia Della "SAI Ambrosini" Di Passignano*. Murena editrice, Perugia
- Bertinetti S, Bolea-Fernandez E, Malandrino M, et al (2022) Strontium isotopic analysis of environmental microsamples by inductively coupled plasma – tandem mass spectrometry. *Journal of Analytical Atomic Spectrometry* 37(1):103–113. doi:<https://doi.org/10.1039/D1JA00329A>
- Birch L, Hanselmann KW, Bachofen R (1996) Heavy metal conservation in Lake Cadagno sediments: Historical records of anthropogenic emissions in a meromictic alpine lake. *Water Research* 30(3):679–687. doi:[https://doi.org/10.1016/0043-1354\(95\)00231-6](https://doi.org/10.1016/0043-1354(95)00231-6)
- Bränvall ML, Bindler R, Emteryd O, et al (2001) [No title found]. *Journal of Paleolimnology* 25(4):421–435. doi:<https://doi.org/10.1023/A:1011186100081>
- Callender E (2014) Heavy Metals in the Environment – Historical Trends. In: *Treatise on Geochemistry*. Elsevier, p 59–89, doi:<https://doi.org/10.1016/>

437 [B978-0-08-095975-7.00903-7](https://doi.org/10.1016/S0048-9697(97)00199-X)

438 Chiaradia M, Chenhall B, Depers A, et al (1997) Identification of historical
439 lead sources in roof dusts and recent lake sediments from an industrialized
440 area: Indications from lead isotopes. *Science of The Total Environment*
441 205(2-3):107–128. doi:[https://doi.org/10.1016/S0048-9697\(97\)00199-X](https://doi.org/10.1016/S0048-9697(97)00199-X)

442 Christensen F (2015) Survey of manganese(II) sulphate: part of the LOUS
443 review. Danish Environmental Protection Agency

444 Dubois N, Saulnier-Talbot É, Mills K, et al (2018) First human impacts and
445 responses of aquatic systems: A review of palaeolimnological records from
446 around the world. *The Anthropocene Review* 5(1):28–68. doi:<https://doi.org/10.1177/2053019617740365>

448 EFSA Panel (NDA) (2013) Scientific Opinion on Dietary Reference Values
449 for manganese. *EFSA Journal* 11(11). doi:[https://doi.org/10.2903/j.efsa.](https://doi.org/10.2903/j.efsa.2013.3419)
450 [2013.3419](https://doi.org/10.2903/j.efsa.2013.3419)

451 (FEEDAP) E (2016) Safety and efficacy of manganese compounds (E5) as feed
452 additives for all animal species: Manganous carbonate; manganous chlo-
453 ride, tetrahydrate; manganous oxide; manganous sulphate, monohydrate;
454 manganese chelate of amino acids, hydrate; manganese chelate of glycine,
455 hydrate, based on a dossier submitted by FEFANA asbl. *EFSA Journal*
456 14(2). doi:<https://doi.org/10.2903/j.efsa.2016.4395>

457 Förstner U, Wittmann GTW (1981) *Metal Pollution in the Aquatic Environ-*
458 *ment*. Springer Berlin Heidelberg, Berlin, Heidelberg

459 Gaino E, Scoccia F, Piersanti S, et al (2012) Spicule records of *Ephydatia*
460 *fluviatilis* as a proxy for hydrological and environmental changes in the
461 shallow Lake Trasimeno (Umbria, Italy). *Hydrobiologia* 679(1):139–153.
462 doi:<https://doi.org/10.1007/s10750-011-0861-7>

- 463 Gobeil C, Johnson WK, Macdonald RW, et al (1995) Sources and Burden
464 of Lead in St. Lawrence Estuary Sediments: Isotopic Evidence. *Environmental Science & Technology* 29(1):193–201. doi:[https://doi.org/10.1021/
465 es00001a025](https://doi.org/10.1021/es00001a025)
466
- 467 Goretti E, Pallottini M, Ricciarini M, et al (2016) Heavy metals bioaccumula-
468 tion in selected tissues of red swamp crayfish: An easy tool for monitoring
469 environmental contamination levels. *Science of The Total Environment*
470 559:339–346. doi:<https://doi.org/10.1016/j.scitotenv.2016.03.169>
- 471 Gravina P, Ludovisi A, Moroni B, et al (accepted 2022) Geochemical proxies
472 and mineralogical fingerprints of sedimentary processes in a closed shallow
473 lake basin since 1850. *Aquatic Geochemistry*
- 474 Guo JY, Wu FC, Zhang L, et al (2011) Screening Level of PAHs in Sediment
475 Core from Lake Hongfeng, Southwest China. *Archives of Environmental
476 Contamination and Toxicology* 60(4):590–596. doi:[https://doi.org/10.
477 1007/s00244-010-9568-4](https://doi.org/10.1007/s00244-010-9568-4)
- 478 Jiang C, Alexander R, Kagi RI, et al (2000) Origin of perylene in ancient sedi-
479 ments and its geological significance. *Organic Geochemistry* 31(12):1545–
480 1559. doi:[https://doi.org/10.1016/S0146-6380\(00\)00074-7](https://doi.org/10.1016/S0146-6380(00)00074-7)
- 481 MacKinnon G, MacKenzie A, Cook G, et al (2011) Spatial and temporal varia-
482 tions in Pb concentrations and isotopic composition in road dust, farmland
483 soil and vegetation in proximity to roads since cessation of use of leaded
484 petrol in the UK. *Science of The Total Environment* 409(23):5010–5019.
485 doi:<https://doi.org/10.1016/j.scitotenv.2011.08.010>
- 486 Matschullat J, Ottenstein R, Reimann C (2000) Geochemical background -
487 can we calculate it? *Environmental Geology* 39(9):990–1000. doi:<https://doi.org/10.1007/s002549900084>
488

- 489 Nacke H, Gonçalves AC, Schwantes D, et al (2013) Availability of Heavy Metals
490 (Cd, Pb, and Cr) in Agriculture from Commercial Fertilizers. Archives of
491 Environmental Contamination and Toxicology 64(4):537–544. doi:<https://doi.org/10.1007/s00244-012-9867-z>
492
- 493 Nicholson F, Smith S, Alloway B, et al (2003) An inventory of heavy
494 metals inputs to agricultural soils in England and Wales. Science of
495 The Total Environment 311(1-3):205–219. doi:[https://doi.org/10.1016/S0048-9697\(03\)00139-6](https://doi.org/10.1016/S0048-9697(03)00139-6)
496
- 497 Odigie KO, Cohen AS, Swarzenski PW, et al (2014) Using lead isotopes and
498 trace element records from two contrasting Lake Tanganyika sediment
499 cores to assess watershed – Lake exchange. Applied Geochemistry 51:184–
500 190. doi:<https://doi.org/10.1016/j.apgeochem.2014.10.007>
- 501 Panagos P, Ballabio C, Lugato E, et al (2018) Potential Sources of Anthro-
502 pogenic Copper Inputs to European Agricultural Soils. Sustainability
503 10(7):2380. doi:<https://doi.org/10.3390/su10072380>
- 504 Poma G, Salerno F, Roscioli C, et al (2017) Persistent organic pol-
505 lutants in sediments of high-altitude alpine ponds within stelvio
506 national park, italian alps. Inland Waters 7(1):34–44. doi:<https://doi.org/10.1080/20442041.2017.1294345>, URL <https://doi.org/10.1080/20442041.2017.1294345>, <https://arxiv.org/abs/https://doi.org/10.1080/20442041.2017.1294345>
507
508
509
- 510 Renberg I, Brännvall ML, Bindler R, et al (2002) Stable lead isotopes and lake
511 sediments—a useful combination for the study of atmospheric lead pollu-
512 tion history. Science of The Total Environment 292(1-2):45–54. doi:[https://doi.org/10.1016/S0048-9697\(02\)00032-3](https://doi.org/10.1016/S0048-9697(02)00032-3)
513
- 514 Resongles E, Dietze V, Green DC, et al (2021) Strong evidence for the con-
515 tinued contribution of lead deposited during the 20th century to the

- 516 atmospheric environment in London of today. Proceedings of the National
517 Academy of Sciences 118(26):e2102791,118. doi:[https://doi.org/10.1073/
518 pnas.2102791118](https://doi.org/10.1073/pnas.2102791118)
- 519 Roszko MŁ, Juszczak K, Szczepańska M, et al (2020) Background levels
520 of polycyclic aromatic hydrocarbons and legacy organochlorine pesti-
521 cides in wheat sampled in 2017 and 2018 in Poland. Environmen-
522 tal Monitoring and Assessment 192(2):142. doi:[https://doi.org/10.1007/
523 s10661-020-8097-5](https://doi.org/10.1007/s10661-020-8097-5)
- 524 Rubio B, Nombela M, Vilas F (2000) Geochemistry of Major and Trace Ele-
525 ments in Sediments of the Ria de Vigo (NW Spain): An Assessment
526 of Metal Pollution. Marine Pollution Bulletin 40(11):968–980. doi:[https://doi.org/10.1016/S0025-326X\(00\)00039-4](https://doi.org/10.1016/S0025-326X(00)00039-4)
- 527
- 528 SAI (2018) SAI Società Aeronautica Italiana
- 529 Sakata M, Xu H, Mashio AS (2018) Analysis of historical trend of pol-
530 lution sources of lead in Tokyo Bay based on lead isotope ratios in
531 sediment core. Journal of Oceanography 74(2):187–196. doi:[https://doi.
532 org/10.1007/s10872-017-0448-7](https://doi.org/10.1007/s10872-017-0448-7)
- 533 Schropp SJ, Lewis FG, Windom HL, et al (1990) Interpretation of Metal
534 Concentrations in Estuarine Sediments of Florida Using Aluminum as
535 a Reference Element. Estuaries 13(3):227. doi:[https://doi.org/10.2307/
536 1351913](https://doi.org/10.2307/1351913)
- 537 Selvaggi R, Damianić B, Goretti E, et al (2020) Evaluation of geochemical base-
538 lines and metal enrichment factor values through high ecological quality
539 reference points: A novel methodological approach. Environmental Sci-
540 ence and Pollution Research 27(1):930–940. doi:[https://doi.org/10.1007/
541 s11356-019-07036-3](https://doi.org/10.1007/s11356-019-07036-3)

- 542 Tapia J, Audry S, Townley B, et al (2012) Geochemical background, base-
543 line and origin of contaminants from sediments in the mining-impacted
544 Altiplano and Eastern Cordillera of Oruro, Bolivia. *Geochemistry: Explo-*
545 *ration, Environment, Analysis* 12(1):3–20. doi:[https://doi.org/10.1144/](https://doi.org/10.1144/1467-7873/10-RA-049)
546 [1467-7873/10-RA-049](https://doi.org/10.1144/1467-7873/10-RA-049)
- 547 Teng Y, Ni S, Wang J, et al (2009) Geochemical baseline of trace elements
548 in the sediment in Dexing area, South China. *Environmental Geology*
549 57(7):1649–1660. doi:<https://doi.org/10.1007/s00254-008-1446-2>
- 550 Townsend AT, Seen AJ (2012) Historical lead isotope record of a sediment
551 core from the Derwent River (Tasmania, Australia): A multiple source
552 environment. *Science of The Total Environment* 424:153–161. doi:<https://doi.org/10.1016/j.scitotenv.2012.02.011>
553 [//doi.org/10.1016/j.scitotenv.2012.02.011](https://doi.org/10.1016/j.scitotenv.2012.02.011)
- 554 Tylmann W (2005) Lithological and geochemical record of anthropogenic
555 changes in recent sediments of a small and shallow lake (Lake Pusty Staw,
556 northern Poland). *Journal of Paleolimnology* 33(3):313–325. doi:<https://doi.org/10.1007/s10933-004-5506-7>
557 [//doi.org/10.1007/s10933-004-5506-7](https://doi.org/10.1007/s10933-004-5506-7)
- 558 Vanhaecke F, Balcaen L, Malinovsky D (2009) Use of single-collector and
559 multi-collector ICP-mass spectrometry for isotopic analysis. *Journal of*
560 *Analytical Atomic Spectrometry* 24(7):863. doi:[https://doi.org/10.1039/](https://doi.org/10.1039/b903887f)
561 [b903887f](https://doi.org/10.1039/b903887f)
- 562 Yang H, Rose N (2005) Trace element pollution records in some UK lake
563 sediments, their history, influence factors and regional differences. *Envi-*
564 *ronment International* 31(1):63–75. doi:[https://doi.org/10.1016/j.envint.](https://doi.org/10.1016/j.envint.2004.06.010)
565 [2004.06.010](https://doi.org/10.1016/j.envint.2004.06.010)
- 566 Yunker MB, Macdonald RW, Vingarzan R, et al (2002) PAHs in the Fraser
567 River basin: A critical appraisal of PAH ratios as indicators of PAH source
568 and composition. *Organic Geochemistry* 33(4):489–515. doi:[https://doi.org/10.1016/S0167-6369\(02\)00001-0](https://doi.org/10.1016/S0167-6369(02)00001-0)

569 [org/10.1016/S0146-6380\(02\)00002-5](https://doi.org/10.1016/S0146-6380(02)00002-5)

570 Zan F, Huo S, Xi B, et al (2012) A 100-year sedimentary record of natural and
571 anthropogenic impacts on a shallow eutrophic lake, Lake Chaohu, China.

572 *Journal of Environmental Monitoring* 14(3):804. doi:[https://doi.org/10.](https://doi.org/10.1039/c1em10760g)
573 [1039/c1em10760g](https://doi.org/10.1039/c1em10760g)

574 Zhang J, Liu C (2002) Riverine Composition and Estuarine Geochemistry
575 of Particulate Metals in China—Weathering Features, Anthropogenic
576 Impact and Chemical Fluxes. *Estuarine, Coastal and Shelf Science*
577 54(6):1051–1070. doi:<https://doi.org/10.1006/ecss.2001.0879>

578 **Declarations**

- 579 • Funding -This work was supported by MIUR and University of Peru-
580 gia for financial support through AMIS project (“Dipartimenti di Eccel-
581 lenza–2018–2022”).
- 582 • Conflict of interest/Competing interests - The authors declare that they
583 have no known competing financial interests or personal relationships that
584 could have appeared to influence the work reported in this paper.
- 585 • Ethics approval - Not applicable
- 586 • Consent to participate -Yes
- 587 • Consent for publication - Yes
- 588 • Availability of data and materials -Yes
- 589 • Authors’ contributions - All authors contributed to the study. Material
590 preparation, data collection and analysis were performed by [Paola Gravina],
591 [Federica Bruschi], [David Cappelletti]. The first draft of the manuscript
592 was written by [Paola Gravina] and all authors [Bartolomeo Sebastiani],
593 [Federica Bruschi], [Chiara Petroselli], [Beatrice Moroni], [Roberta Sel-
594 vaggi], [Enzo Goretti], [Matteo Pallottini], [Alessandro Ludovisi], [David

595 Cappelletti] commented on previous versions of the manuscript. All authors
596 [Bartolomeo Sebastiani], [Federica Bruschi], [Chiara Petroselli], [Beatrice
597 Moroni], [Roberta Selvaggi], [Enzo Goretti], [Matteo Pallottini], [Alessandro
598 Ludovisi], [David Cappelletti] read and approved the final manuscript.

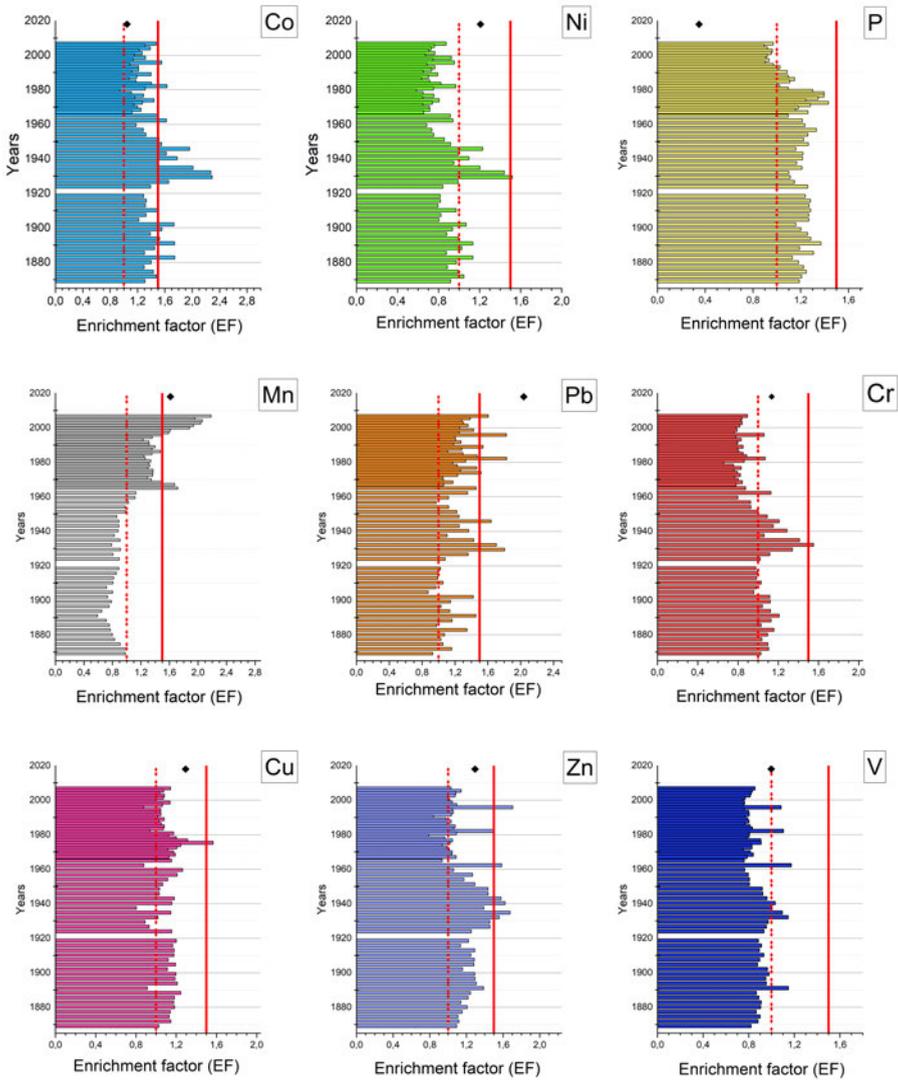


Fig. 3 C2 core trend of enrichment factors for trace elements (Co, Ni, P, Mn, Pb, Cr, Cu, Zn and V) represented by the histogram graph. The black rumber represents the current state of enrichment in the surface sediments of the lake (corresponding to 2018). The dotted red line indicates limit at 1 (according to [Birch et al \(1996\)](#)), while continuous red line indicates limit at 1.5 (according to [Zhang and Liu \(2002\)](#)).

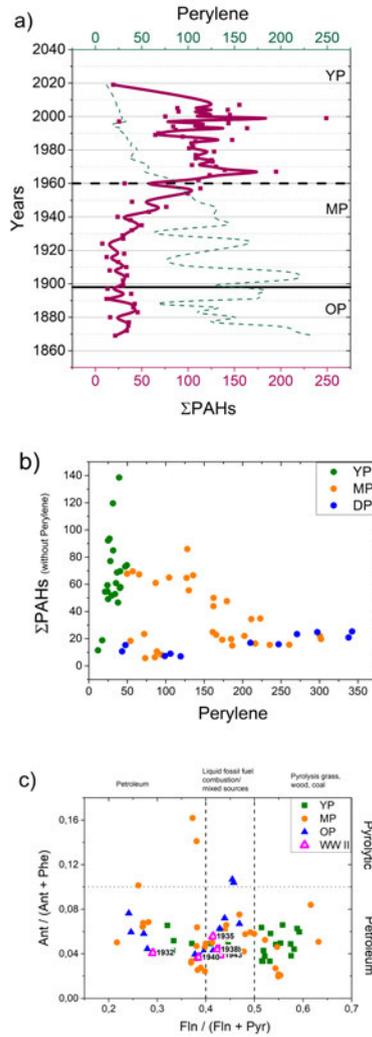


Fig. 4 a) Σ PAHs (continuous red curve) and perylene (dashed green line) concentration (ng/g) in core C2; b) Σ PAHs versus perylene concentration (diagenetic origin); c) scatterplot of $\text{Flu}/(\text{Flu} + \text{Pyr})$ and $\text{Ant}/(\text{Ant} + \text{Phe})$ diagnostic ratios with marked regions corresponding to different sources. Square represent samples in YP, circles represent sample in MP, triangles represent sample in OP and empty triangles represent samples during the 2nd World War (WW II).

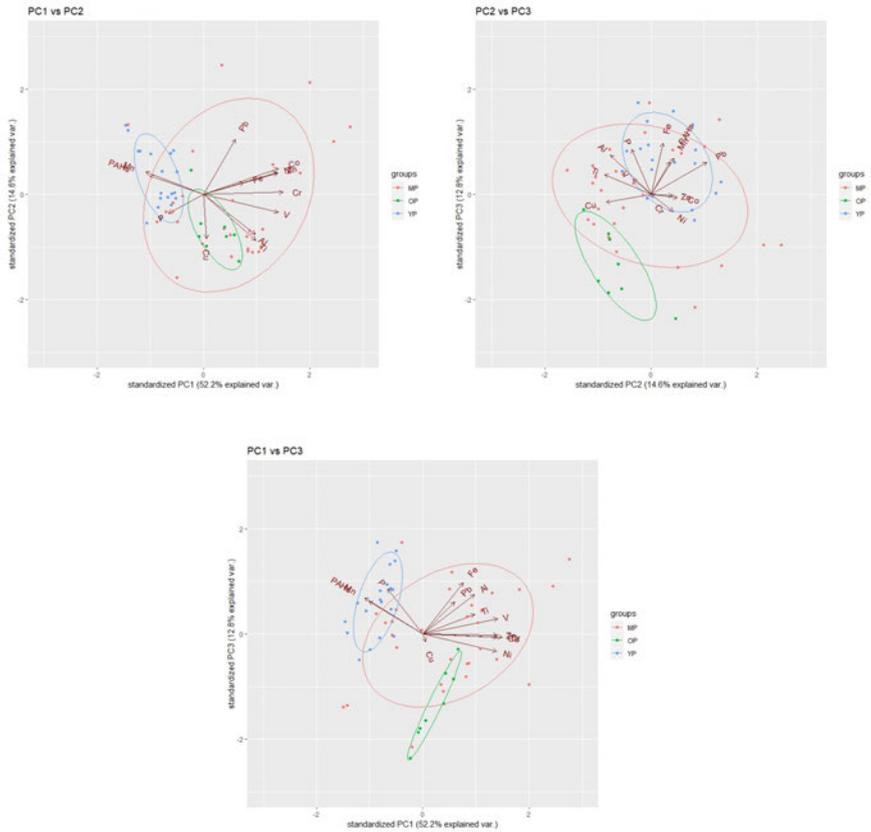


Fig. 5 C2 core PCA: the different major principal components against each other are reported in each graph plot, comparing trace elements, heavy metals, phosphorus and ΣPAHs. The legend shows the scores according to specific hydrometric phase: phase OP in green, phase MP in red and phase YP in blue.

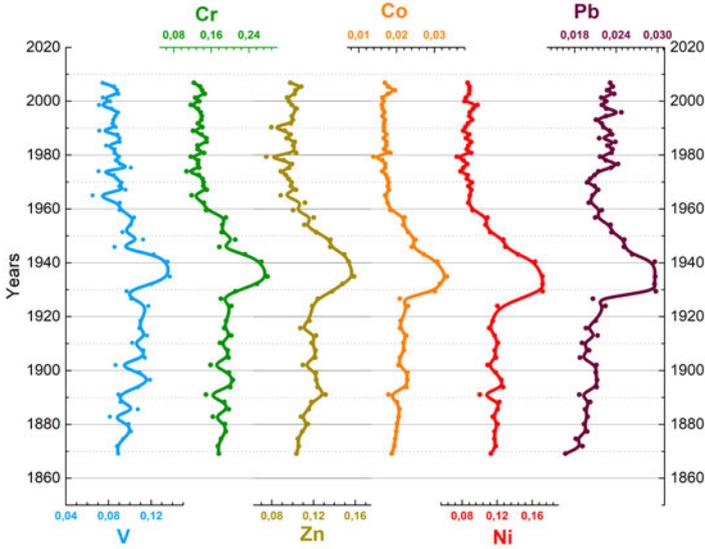


Fig. 6 Trace metal concentration (V, Cr, Zn, Co, Ni, Pb) temporal profiles of the C2 core (adjacent-averaging method, with weighted average options was used on 5 points to obtain the trends line)

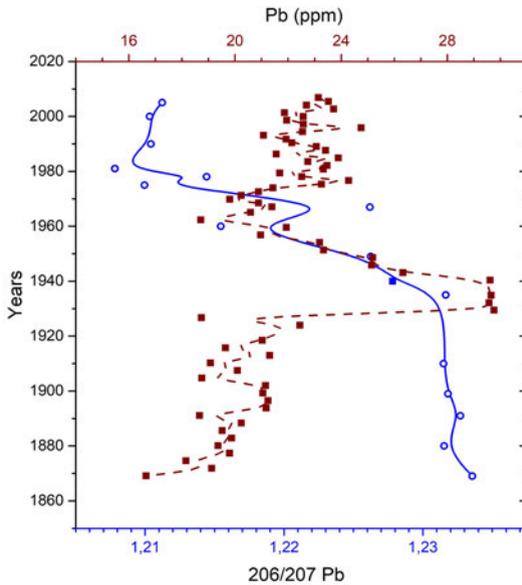


Fig. 7 Trends of total Pb concentration and $^{206}\text{Pb}/^{207}\text{Pb}$ isotope ratio, for the C2 core. The solid blue line indicates the $^{206}\text{Pb}/^{207}\text{Pb}$ isotope ratio, while the dashed brown line indicates the Pb concentration (ppm), and adjacent-averaging method, with weighted average options was used on 5 points to obtain the trends line.

Supplementary Files

This is a list of supplementary files associated with this preprint. Click to download.

- [SMarticlePG1.pdf](#)