

# Water-Air Exchanges In The Lower Estuary Of The Patos Lagoon: Seasonal Variability, Drivers, And Sources Of CO<sub>2</sub>

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## Research Article

**Keywords:** estuaries, carbon dioxide, CO<sub>2</sub> partial pressure, CO<sub>2</sub> fluxes

**Posted Date:** February 10th, 2022

**DOI:** <https://doi.org/10.21203/rs.3.rs-1336400/v1>

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# Water-air exchanges in the lower estuary of the Patos Lagoon: seasonal variability, drivers, and sources of CO<sub>2</sub>

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## Highlights

- The lower zone of the Patos Lagoon Estuary (PLE) is an annual net CO<sub>2</sub> sink zone.
- The CO<sub>2</sub> uptake in the PLE surface waters during the summer/autumn seasons is nearly twofold higher than the winter/spring CO<sub>2</sub> emissions.
- Autochthonous CO<sub>2</sub> production dominates the carbon dynamics in the PLE.
- The high surface water dynamics in the mouth of the PLE prevent accumulation of CO<sub>2</sub> and enable carbon export to the adjacent coast.

Submitted to Biogeochemistry

## 50 **Acknowledgments**

51 This study contributes to the activities of the Brazilian Ocean Acidification Network  
52 (BrOA; [www.broa.furg.br](http://www.broa.furg.br)) and the CARBON-FLUX project (National Council for  
53 Scientific and Technological Development – CNPq grant no. 420118/2018–7). C.A.W.C.  
54 received PhD. grant no. 88887.478990/2020–00 from the Coordination for the  
55 Improvement of Higher Education Personnel (CAPES). R.K. and C.R.B. received  
56 researcher grant nos. 304937/2018–5 and 306899/2018–3 from CNPq, respectively. We  
57 acknowledge the availability of the monitoring database provided by the Patos Lagoon  
58 Brazilian Long-Term Ecological Research Program (database available at  
59 <https://doi.org/10.15468/xmlvxn>), which has received grants from CNPq no.  
60 441492/2016–9, FAPERGS no. 16/2551–0000102–2, and CAPES no. 16/2551–  
61 0000102–1. The BrOA Network dataset is available by request  
62 (<https://carbonteam.furg.br/2-uncategorized/47-monitoramento>). We also acknowledge  
63 the resources provided by CAPES to support the Graduate Program in Oceanology and  
64 free access to many scientific journals through *Periódicos* CAPES. We thank all the  
65 researchers, technicians and students who have been involved with and contributed to  
66 fostering this valuable database on the PLE.

67

## 68 **Author contributions**

69 C.A. conducted the data analysis and main interpretations of this study as part of her Ph.D  
70 thesis. R.K. lead the BrOA activities in the Patos Lagoon, proposed the study and  
71 supervised C.A. in conducting the planning and data collection in this study. E.M., T.M.,  
72 and A.C.O.C contributed as experts in carbonate system and biogeochemical processes.  
73 C.R.B.M. contributed as an expert on phytoplankton and ecology. All authors contributed  
74 to the interpretation of results and writing the manuscript as experts on CO<sub>2</sub> systems.

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79

80 **Abstract**

81 We investigated the primary drivers of changes in the partial pressure of carbon dioxide  
82 ( $p\text{CO}_2$ ) together with the seasonal and interannual variability in the water-air net carbon  
83 dioxide flux ( $\text{FCO}_2$ ) in the lower estuarine zone surface waters of Patos Lagoon, the  
84 largest choked lagoon worldwide. Sampling occurred monthly during May 2017-June  
85 2021 at the estuary's inner inlet and mouth, which are contrasting hydrodynamic zones  
86 in the Patos Lagoon Estuary (PLE). The water  $p\text{CO}_2$  was mainly controlled by seasonal  
87 changes in total alkalinity and total dissolved inorganic carbon. The lower zone  
88 experienced periods of  $\text{CO}_2$  ingassing (austral summer/autumn) and  $\text{CO}_2$  outgassing  
89 (austral winter/spring). During summer/autumn, both protected and sea-exposed areas  
90 uptake an average of  $-15 \text{ mmol m}^{-2} \text{ d}^{-1} \text{ CO}_2$ , whereas during winter/spring,  $\text{CO}_2$   
91 emissions prevail, reaching an average of  $22 \text{ mmol m}^{-2} \text{ d}^{-1}$  at the inner estuary.  
92 Additionally, while much of the  $\text{CO}_2$  absorbed in summer/autumn is released to the  
93 atmosphere in the inner estuarine zone, the summer/autumn  $\text{CO}_2$  uptake in the estuary  
94 mouth is 4-fold higher than the winter/spring  $\text{CO}_2$  released. Unlike most estuarine  
95 systems, the PLE acted as a net  $\text{CO}_2$  sink of  $-2 \text{ mmol m}^{-2} \text{ d}^{-1}$  during the period  
96 investigated. The balance between  $\text{CO}_2$  uptake and emissions in the PLE was modulated  
97 by the combination of wind speed, freshwater discharge, water temperature, and  
98 outflow/inflow currents. Furthermore, phytoplankton blooms and strong wind-induced  
99 vertical mixing lead to highly variable  $\text{CO}_2$  exchanges. The highest estuarine  $\text{CO}_2$   
100 concentration by autochthonous production indicates heterotrophy in estuarine waters.  
101 Part of this carbon produced in the estuary is exported to the coast, as evidenced by the  
102 high  $\text{CO}_2$  concentration in the estuary mouth. Therefore, the lower estuarine zone resists  
103 increased  $\text{CO}_2$  concentrations and has overcome regional anthropogenic emissions. The  
104 regional  $\text{FCO}_2$  range and complex PLE biogeochemistry dynamics need ongoing  
105 investigation to improve knowledge of regional  $\text{CO}_2$  exchanges and elucidate the role of  
106 large estuaries and coastal bays in the global carbon budget.

107

108 **Keywords:** estuaries; carbon dioxide;  $\text{CO}_2$  partial pressure;  $\text{CO}_2$  fluxes

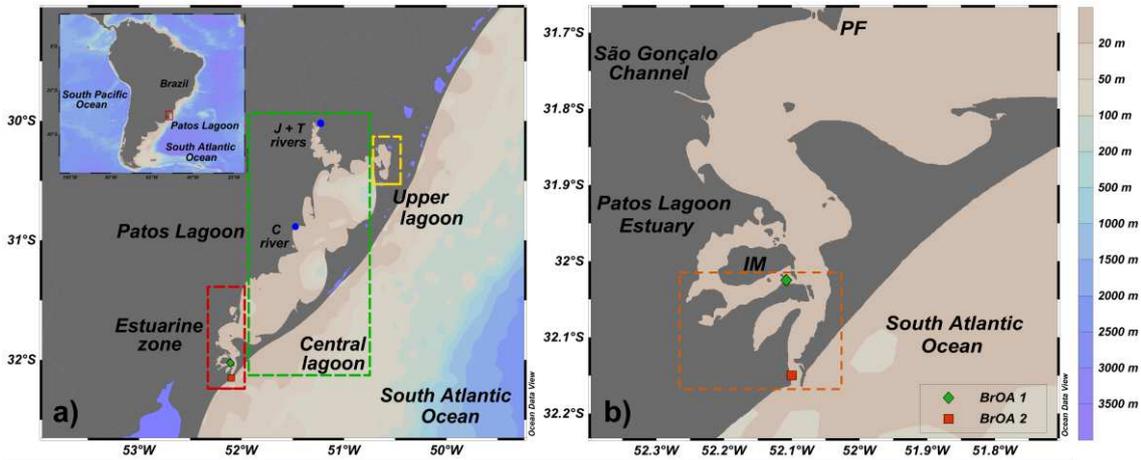
## 109 **1. Introduction**

110 Estuaries are known to be large sources of carbon dioxide (CO<sub>2</sub>) to the atmosphere (e.g.,  
111 Cai 2011; Bauer et al. 2013; Evans et al. 2013; Dinauer and Mucci 2017; Yao and Hu  
112 2017; Yao et al., 2020), with the surface water partial pressure of CO<sub>2</sub> (*p*CO<sub>2</sub>) ranging  
113 from 350 to 10,000 μatm and the water-air CO<sub>2</sub> net flux (FCO<sub>2</sub>) ranging from -5 to 80  
114 mol C m<sup>-2</sup> year<sup>-1</sup> (Cai, 2011). This CO<sub>2</sub> outgassing behavior is attributed to intense carbon  
115 fixation and respiration in estuarine waters, owing to the higher primary production and  
116 leading to high rates of organic matter decomposition. The high amounts and fast cycling  
117 of autochthone and allochthone inputs of organic matter are degraded by microbial action,  
118 which causes supersaturation of CO<sub>2</sub> in estuarine surface waters (Sunda and Cai 2012).  
119 The carbon inputs and their reactions support prominent heterotrophy and contribute to  
120 the widespread supersaturation of CO<sub>2</sub>, decreasing pH (e.g., Feely et al., 2010; Borges  
121 and Abril 2011; Cloern et al. 2014). The changes in estuarine *p*CO<sub>2</sub> regulate the water-air  
122 CO<sub>2</sub> gradients that determine the direction of CO<sub>2</sub> exchanges (Sarma et al. 2001). In  
123 addition, estuaries are commonly neglected in global FCO<sub>2</sub> inventories, and estuarine  
124 processes are not factored into the global carbon budget (e.g., Cotovicz et al. 2020).  
125 Furthermore, estuaries are highly productive regions where natural biogeochemical  
126 reaction rates are elevated; thus, understanding CO<sub>2</sub> dynamics in these environments is  
127 essential (Sarma et al., 2001).

128 Patos Lagoon (Figure 1) is the largest choked lagoon in the world (Kjerfve, 1986), and it  
129 is connected to the sea by a narrow channel (Castelão and Möller 2003; Marques and  
130 Möller 2008; Marques et al. 2009) in extreme southern Brazil (~32°S). The  
131 hydrodynamics of the Patos Lagoon Estuary (PLE) are dominated by wind at time scales  
132 associated with the passage of frontal systems and the strength of freshwater discharge in  
133 the microtidal region (Möller et al. 2001). The seawater intrusions are greater during the  
134 autumn due to the more frequent passage of frontal systems. In this case, southerly winds  
135 combined with low freshwater input allow flooding and salinization of the water.  
136 Northeastern winds occur at the end of winter and during spring, with periods of high  
137 freshwater discharge that favor ebb flow. The north and south quadrant winds form  
138 vertical salinity structures that can range from a salt wedge to a well-mixed gradient  
139 (Möller et al. 2001; Möller and Fernandes 2010). The complex balance between  
140 freshwater outflow and oceanic inflow leads to large variations in salinity, which  
141 significantly affects physical (e.g., Möller et al. 2001), chemical (e.g., Niencheski et al.  
142 2006; Albuquerque et al. under review) and biological properties (e.g., Haraguchi et al.

143 2015). Although the PLE is a well-studied environment in terms of estuarine  
144 hydrodynamics (e.g., Möller et al. 2001; Möller and Fernandes 2010), biology and  
145 physiology of dominant estuarine species, and ecosystem ecology (e.g., Haraguchi et al.  
146 2015; Abreu and Odebrecht 2016; Mendes et al. 2016; Islabão et al. 2017; Odebrecht et  
147 al. 2017), very little is known about estuarine carbon biogeochemistry (Albuquerque et  
148 al. under review). Recently, the first assessment of the estuarine carbonate system in the  
149 region indicated that the surface waters in the lower zone of the PLE have natural alkaline  
150 conditions, with an average  $p\text{CO}_2$  of  $\sim 380 \mu\text{atm}$  and a supersaturated calcium carbonate  
151 environment with respect to both calcite and aragonite (Albuquerque et al. under review).  
152 Additionally, the predominant estuarine processes governing changes in the carbonate  
153 system in the region were dilution and concentration of salts due to freshwater input and  
154 seawater intrusions, respectively. Notwithstanding its socioeconomic and environmental  
155 significance (Odebrecht et al. 2017), only a few studies have assessed the chemical  
156 changes in estuarine waters in this complex environment (e.g., Niencheski et al. 2006;  
157 Baumgarten and Niencheski 2010; Wallner-Kersanach et al. 2016).

158 Studies on water-air  $\text{CO}_2$  exchanges in estuarine regions are mostly located along the  
159 European, Asian, Indian, and eastern North American coasts (e.g., Bauer et al. 2013;  
160 Evans et al. 2013), while knowledge of  $\text{CO}_2$  coastal dynamics in the Southern Hemisphere  
161 remains limited. Few studies have addressed the water-air  $\text{CO}_2$  exchange systems that  
162 border the Brazilian coast (e.g., Noriega et al. 2013; Noriega and Araujo 2014; Cotovicz  
163 et al. 2015; Cotovicz et al. 2020; Abril et al. 2021), where highly diversified environments  
164 in terms of  $\text{CO}_2$  saturation with the atmosphere were found. As a larger body of water in  
165 the Southern Hemisphere, a better understanding of the regional behavior of the  $\text{CO}_2$   
166 fluxes in the PLE is mandatory for its inclusion in a global  $\text{CO}_2$  analysis. Therefore, in  
167 this study, we present the first overview of the behavior of the water-air  $\text{CO}_2$  exchanges  
168 in the lower estuarine zone of the Patos Lagoon (Figure 1) by investigating the  $p\text{CO}_2$   
169 drivers, sources of  $\text{CO}_2$  in the region, and the temporal variability in the water-air  $\text{CO}_2$   
170 fluxes in this environment. This assessment of the intrinsic variability in the  $\text{CO}_2$  system  
171 in the PLE represents the first step toward understanding both anthropogenic and climatic  
172 impacts that may affect the carbon dynamics in the region.



173

174 **Figure 1:** Map of the study region: Patos Lagoon. (a) Location of Patos Lagoon and geomorphological  
 175 divisions for the estuarine zone (red rectangle), central lagoon (green rectangle) and upper lagoon (yellow  
 176 rectangle). The inset in (a) shows a South American map with the location of Patos Lagoon (red rectangle).  
 177 Blue dots indicate the regions of freshwater discharge from the Jacuí and Taquari (J + T) and Camaquã (C)  
 178 Rivers. (b) Location of the pier-fixed monitoring BrOA #1 (green diamond) and #2 (red square) stations in  
 179 the Patos Lagoon Estuary with the southern region highlighted (orange rectangle). Marinheiros Island (IM)  
 180 and Ponta da Feitoria (PF) are indicated.

181

## 182 2. Data and Methods

### 183 2.1 Database from the Brazilian monitoring programs

184 The databases of the Brazilian Long-Term Ecological Research (BR-LTER; Odebrecht  
 185 and Abreu 2019; Lemos et al. 2021; dataset available at  
 186 <https://doi.org/10.15468/xmlvxn>) and the Brazilian Ocean Acidification Network (BrOA  
 187 Network; Kerr et al. 2016; dataset available by request at [https://carbonteam.furg.br/2-  
 188 uncategorised/47-monitoramento](https://carbonteam.furg.br/2-uncategorised/47-monitoramento)) were compiled for the study of a four-year monthly  
 189 time series (May 2017-June 2021) in two zones of the PLE. The pier-fixed station BrOA  
 190 #1 is in the inner mesomixohaline region, while BrOA #2 is in a more exposed area of  
 191 the lower euhaline region. The stations are marked by differences in water salinity and  
 192 hydrodynamics. The physical, biological, and chemical parameters were sampled by BR-  
 193 LTER and included surface water (~1 m) temperature, salinity, chlorophyll-a (Chl-*a*) and  
 194 dissolved nutrients (i.e., nitrate, nitrite, silicic acid, and phosphate). The surface water (~1  
 195 m) total alkalinity ( $A_T$ ), total dissolved inorganic carbon ( $C_T$ ) and pH were sampled by  
 196 the BrOA Network monitoring program. For more details about the datasets and sampling  
 197 methodology, the reader is referred to Albuquerque et al. (under review) and Lemos et al.  
 198 (2021).

199

## 200 **2.2 Sampling and determination of physical, chemical, and biogeochemical** 201 **properties**

202 Surface water temperature and salinity were measured *in situ* by a digital thermometer ( $\pm$   
203  $1^\circ\text{C}$ ) and portable refractometer ( $\pm 1$ ), respectively. The water sample was collected, and  
204 the salinity was further verified in the laboratory by a conductivity meter. Water for  $A_T$   
205 analysis was sampled in 500 mL borosilicate glass bottles and fixed with 100  $\mu\text{L}$  of a  
206 supersaturated mercury chloride solution to prevent biological activity, following the  
207 procedure described by Dickson et al. (2007). The samples were refrigerated to prevent  
208 evaporation, and  $A_T$  was measured by potentiometric titration in a closed cell (Dickson  
209 et al. 2007) with an automated titrator (Metrohm® Titrand 808) and a combined glass-  
210 reference electrode (Metrohm® 6.0262.100) at a controlled temperature of  $25 \pm 0.1^\circ\text{C}$   
211 sustained by a thermostatic bath (Tamson® TLC 15). The analytical precision of the  $A_T$   
212 analyses was  $\pm 4.0 \mu\text{mol kg}^{-1}$  (ranging from  $2.3 < A_T < 5.0 \mu\text{mol kg}^{-1}$  considering the  
213 analyzed sample batches). Water for pH analysis was sampled in 125 mL borosilicate  
214 amber flasks, and the pH was potentiometrically determined before two hours after  
215 sampling. We used a Metrohm® 913 or 914 pH meter coupled with a glass-reference  
216 electrode cell and a temperature sensor. The uncertainty for pH was  $\leq 0.05$  pH NBS units.  
217 The total scale was chosen and further used for pH at *in situ* temperature.

218 The  $p\text{CO}_2$  and other  $\text{CO}_2$ -carbonate variables not directly measured were estimated  
219 through the software  $\text{CO}_2\text{Sys v.2.1}$  developed by Lewis et al. (1998) and modified by  
220 Pierrot et al. (2006). Surface water temperature, salinity,  $A_T$ , pH, silicic acid, and  
221 phosphate concentrations were used as input parameters. Since the study was performed  
222 in an estuarine environment with a broad salinity range, we applied the following set of  
223 constants: the  $K_1$  and  $K_2$  dissociation constants of Millero et al. (2006), such as those used  
224 in PLE (Albuquerque et al. under review) and other estuary and coastal environments  
225 (Liu et al. 2017, Carstensen et al. 2018, Chen et al. 2020); and the sulfate and borate  
226 constants of Dickson (1990) and Uppström (1974), respectively. The  $p\text{CO}_2$  uncertainty  
227 was determined to be  $\pm 46 \mu\text{atm}$  according to Orr et al. (2018). A more detailed description  
228 of the sampling procedure and laboratory analysis of  $A_T$  and pH and reconstruction of  
229  $p\text{CO}_2$  is fully described in Albuquerque et al. (under review).

230

## 231 **2.3 Drivers of changes in the estuarine water partial pressure of $\text{CO}_2$**

232 The  $p\text{CO}_2$  drivers were calculated based on the seasonal differences in parameters and  
 233 their corresponding partial derivatives. The differences in  $p\text{CO}_2$  were separated into  
 234 contributions representing the roles of differences in temperature (Temp), salinity (Sal),  
 235  $A_T$ , and  $C_T$ . The relative contributions of the drivers changing  $p\text{CO}_2$  (i.e.,  $\Delta p\text{CO}_2^{drv}$ ) were  
 236 assessed by converting their relative changes into  $p\text{CO}_2$  units ( $\mu\text{atm}$ ) following Lenton et  
 237 al. (2012) and Equation 1:

$$238 \quad \Delta p\text{CO}_2^{drv} = \frac{\partial p\text{CO}_2}{\partial \text{Temp}} \Delta \text{Temp} + \frac{\partial p\text{CO}_2}{\partial \text{Sal}} \Delta \text{Sal} + \frac{\partial p\text{CO}_2}{\partial A_T} \Delta A_T + \frac{\partial p\text{CO}_2}{\partial C_T} \Delta C_T, \quad \text{Eq. 1}$$

239 where  $\Delta \text{Temp}$ ,  $\Delta \text{Sal}$ ,  $\Delta A_T$  and  $\Delta C_T$  are the respective differences in the water surface  
 240 property averages between each season and the previous season in the lower zone of the  
 241 PLE, considering the sampling period from May 2017 to June 2021. The partial  
 242 derivatives ( $\delta$ ) were calculated using Equations 2, 3 and 4 (see details in Sarmiento and  
 243 Gruber 2006), and the term involving temperature was calculated using Equation 5  
 244 (Takahashi et al. 2014):

$$245 \quad \frac{\partial p\text{CO}_2}{\partial C_T} = \frac{p\text{CO}_2}{C_T} \times \text{Revelle Factor}, \quad \text{Eq. 2}$$

$$246 \quad \frac{\partial p\text{CO}_2}{\partial A_T} = \frac{p\text{CO}_2}{A_T} \times \text{Alkalinity Factor}, \quad \text{Eq. 3}$$

$$247 \quad \frac{\partial p\text{CO}_2}{\partial \text{Sal}} \approx 0.026 \times p\text{CO}_2, \quad \text{Eq. 4}$$

$$248 \quad \frac{\partial p\text{CO}_2}{\partial \text{Temp}} \Delta \text{Temp} \approx 2p\text{CO}_2 \times \left[ \text{Exp} \left( 0.0423 \times \frac{\Delta \text{Temp}}{2} \right) - 1 \right]. \quad \text{Eq. 5}$$

249 where the Revelle and Alkalinity factors are 14.2 and  $-26.5$ , respectively.

250

## 251 **2.4 Water-air $\text{CO}_2$ net flux**

252 The water-air  $\text{CO}_2$  fluxes ( $\text{FCO}_2$ ) were obtained by Equation 6:

$$253 \quad \text{FCO}_2 = K_t K_s \Delta p\text{CO}_2, \quad \text{Eq. 6}$$

254 where  $K_t$  is the coefficient for  $\text{CO}_2$  transfer velocity as a function of wind speed ( $U$ ),  $K_s$   
 255 is the solubility coefficient of  $\text{CO}_2$  calculated as a function of both temperature and  
 256 salinity (Weiss 1974), and  $\Delta p\text{CO}_2$  is the difference between surface water  $p\text{CO}_2$  and  
 257 atmospheric  $p\text{CO}_2$  ( $p\text{CO}_2^{\text{air}}$ ).  $\text{CO}_2$  is taken up by estuarine water when the  $\text{FCO}_2$  value is  
 258 negative (ingassing), while it is released to the atmosphere when the  $\text{FCO}_2$  value is  
 259 positive (outgassing).

260 The  $p\text{CO}_2^{\text{air}}$  was calculated following Equation 7:

$$261 \quad p\text{CO}_2^{\text{air}} = x\text{CO}_2^{\text{air}} (p\text{Air} - \left(\frac{1.5}{101.325}\right) - p\text{H}_2\text{O}), \quad \text{Eq. 7}$$

262 where  $x\text{CO}_2^{\text{air}}$  (ppm) is the mole fraction of atmospheric  $\text{CO}_2$  in dry air, obtained from the  
263 Mauna Loa Observatory (NOAA ESRL Global Monitoring Laboratory, 2019; Thoning  
264 et al. 2021), with data 6 months before the corresponding period due to the atmospheric  
265 response between the Northern and Southern Hemispheres (Millero, 2013).  $p\text{Air}$  is the  
266 barometric pressure from the Rio Grande city (Brazil) meteorologic station, and  $p\text{H}_2\text{O}$   
267 (atm) is the water vapor pressure calculated using salinity and temperature (Weiss and  
268 Price, 1980).

269 The main challenge in calculating  $\text{FCO}_2$  in estuarine waters is the determination of  $K_t$  due  
270 to its complex hydrodynamics and varied geomorphology (e.g., Dinauer and Mucci,  
271 2017; Yao et al., 2020). Several different predictive relationships between wind speed  
272 and gas transfer velocity of  $\text{CO}_2$  have been proposed based on laboratory and field studies  
273 (e.g., Jiang et al. 2008; Raymond and Cole 2001; Takahashi et al. 2009). Here, the gas  
274 transfer velocity  $K_t$  was parameterized using wind speed and the equation from Jiang et  
275 al. (2008), which was derived from Raymond and Cole (2001). The parametrization of  
276 Jiang et al. (2008) is mostly used in estuarine environments (e.g., Evans et al. 2013; Van  
277 Dam et al. 2018; Yao et al. 2020). The following equation is the  $K_t$  equation (Equation 8)  
278 of Jiang et al. (2008):

$$279 \quad K_t = (0.314 U^2 - 0.436 U + 3.99) x (Sc/600)^{-0.5}, \quad \text{Eq. 8}$$

280 where  $U$  is the wind speed at 10 m height and  $Sc$  is the Schmidt number of  $\text{CO}_2$  at *in situ*  
281 temperature (Wanninkhof, 2014). Average monthly wind speed data were available from  
282 the 8<sup>th</sup> Meteorology District of the National Institute of Meteorology (8<sup>th</sup> DISME/INMET)  
283 for Rio Grande do Sul State.

284 The average standard error of the calculated  $\text{FCO}_2$  was  $\pm 0.34 \text{ mmol m}^{-2} \text{ d}^{-1}$ . We  
285 recalculated  $K_t$  from the equation of Raymond and Cole (2001) to verify the sensitivity  
286 of  $\text{FCO}_2$  based on the equations used. The average differences between the estimations  
287 derived from the original application used (Jiang et al. 2008) and the Raymond and Cole  
288 (2001) approach were  $-45.3 \pm 44.8 \text{ mmol m}^{-2} \text{ d}^{-1}$ .

289

## 290 **2.5 $\text{CO}_2$ estuarine concentration estimates**

291 We followed the approach described in Jiang et al. (2008) to determine the  $C_T$  change  
 292 caused by river–ocean mixing, using  $C_{Tmr}$  (Equation 9) to estimate the riverine water input  
 293 and  $C_{Tmix}$  (Equation 10) to estimate ocean mixing at each pier-fixed station ( $i$ ):

$$294 \quad C_{Tmr} = \left(\frac{S_i}{S_{oc}}\right) \times C_{Toc} + \left(\frac{1-S_i}{S_{oc}}\right) \times C_{Tr}, \quad \text{Eq. 9}$$

$$295 \quad C_{Tmix} = \frac{(S_{oc} - S_i) \times C_{Tr} + (S_i - S_r) \times C_{Toc}}{S_{oc} - S_r}, \quad \text{Eq. 10}$$

296 where  $C_{Tr}$ ,  $S_r$ ,  $C_{Toc}$  and  $S_{oc}$  are the  $C_T$  and salinity river and ocean end-members,  
 297 respectively, and  $S_i$  is the salinity at station  $i$ .

298 As there was no continuous sampling during the study period at the mouth of the river or  
 299 at a fixed point in the ocean, the average values of  $C_{Tr}$ ,  $S_r$ ,  $C_{Toc}$  and  $S_{oc}$  representing the  
 300 river and ocean most pure conditions were determined from the station closest to the river  
 301 outlet (BrOA #1; salinity < 5) and from the nearest ocean station (BrOA #2; salinity <  
 302 32), respectively. Thus, the  $C_T$  and salinity river and ocean end-members were  $S_r = 2.57$ ,  
 303  $C_{Tr} = 703.37 \mu\text{mol kg}^{-1}$ ,  $S_{oc} = 31.36$ , and  $C_{Toc} = 1648.38 \mu\text{mol kg}^{-1}$ .

304 When there is no river influence, the  $C_T$  at station  $i$  can be calculated as follows:

$$305 \quad C_{Tm0} = \frac{S_i}{S_{oc}} \times C_{Toc}, \quad \text{Eq.}$$

306 where  $C_{Tm0}$  is  $C_T$  due to water mixing;  $C_{Toc}$  and  $S_{oc}$  are  $C_T$  and salinity at the ocean end-  
 307 member, respectively; and  $S_i$  is the salinity at station  $i$ .

308 Then, produced/consumed  $C_T$  due to estuarine-biogeochemical processes ( $C_T^{est}$ ) can be  
 309 calculated as follows:

$$310 \quad C_T^{est} = C_{Ti} - C_{Tm}, \quad \text{Eq. 12}$$

311 where  $C_{Ti}$  is  $C_T$  at station  $i$  and  $C_{Tm}$  is  $C_T$  due the mixing of river and ocean and can be  
 312 calculated from Equations 9-11. Following the same approach,  $A_{Tm}$  and  $A_T^{est}$  can be  
 313 estimated by simply replacing  $C_T$  with  $A_T$ . For  $A_T$  end-members, averages were also  
 314 defined.  $A_{Tr}$  and  $A_{Toc}$  were  $699.47$  and  $2087.04 \mu\text{mol kg}^{-1}$ , respectively.

315 Finally, the  $\text{CO}_2$  estuarine concentration was calculated through the software  $\text{CO}_2\text{Sys}$   
 316 v.2.1 (Lewis et al. 1998; Pierrot et al. 2006) using  $C_T$ ,  $A_T$ , salinity, and temperature as  
 317 input parameters. We used  $[\text{CO}_2]_{\text{ocean}}$ ,  $[\text{CO}_2]_{\text{river}}$ , and  $[\text{CO}_2]_{\text{est}}$  to represent ocean-borne  
 318  $[\text{CO}_2]$ , river-borne  $[\text{CO}_2]$ , and estuarine-produced  $[\text{CO}_2]$ , respectively. Aqueous  $\text{CO}_2$   
 319 ( $[\text{CO}_2]$ ) does not mix conservatively, so  $[\text{CO}_2]_{\text{ocean}}$  is the aqueous  $\text{CO}_2$  concentration of

320 the ocean end-member if these were diluted by freshwater with zero  $C_T$ , calculated by  
321  $C_{Tmo}$  and  $A_{Tmo}$  (Equation 12).  $[CO_2]_{river}$  is the difference between  $[CO_2]$  due to mixing  
322 and  $[CO_2]_{ocean}$  (Jiang et al. 2008).

323

$$324 \quad [CO_2]_{river} = [CO_2]_m - [CO_2]_{ocean}, \quad \text{Eq. 13}$$

325

$$326 \quad [CO_2]_{est} = [CO_2]_i - [CO_2]_m, \quad \text{Eq. 14}$$

327

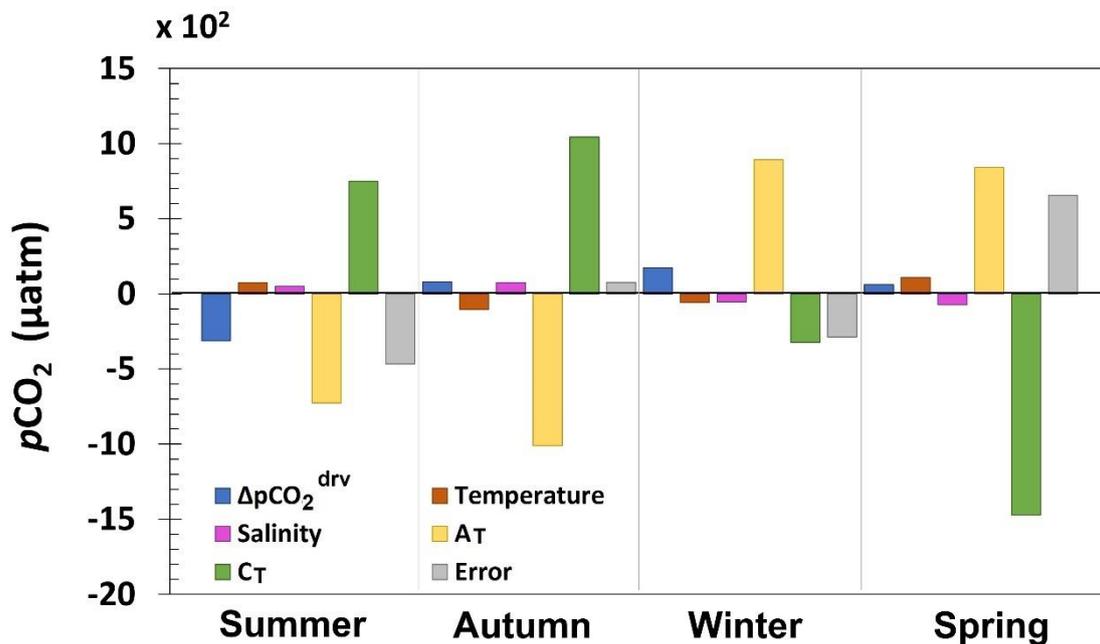
328 where  $[CO_2]_m$  is  $[CO_2]$  if conservative mixing occurred between river and ocean end-  
329 members, calculated using  $C_{Tmr}$  or  $C_{Tmix}$  and  $A_{Tmr}$  or  $A_{Tmix}$ ;  $[CO_2]_{ocean}$  is the aqueous  $CO_2$   
330 concentration of the ocean; and  $[CO_2]_i$  is the aqueous  $CO_2$  concentration at station  $i$ . The  
331  $[CO_2]$  is consumed when the  $[CO_2]$  value is negative, while the  $[CO_2]$  is produced in the  
332 estuary once the  $[CO_2]$  value is positive. When  $[CO_2]$  is calculated from  $C_T$  and  $A_T$ , the  
333 annual average temperature of  $21.03^\circ C$  was used since the dissolved  $[CO_2]$  is subject to  
334 changes in water temperature.

335

### 336 **3. Results**

#### 337 **3.1 Drivers of seasonal changes in the partial pressure of $CO_2$**

338  $A_T$  and  $C_T$  had the dominant effect on changes in  $pCO_2$ , while salinity and temperature  
339 had a minor influence on surface  $pCO_2$  (Figure 2). In summer, there was a considerable  
340 decrease in  $pCO_2$ , driven mainly by an increase in  $A_T$ . In autumn, winter and spring, there  
341 was an increase in  $pCO_2$ ; however, its driver was different in each season. The increase  
342 in  $pCO_2$  was partially counteracted by the temperature drawdown in autumn. In winter  
343 and spring,  $C_T$  and  $A_T$  had opposite effects on  $pCO_2$  compared with summer and autumn.  
344 In winter, the decrease in  $A_T$  led to an increase in  $pCO_2$ . On the other hand, although  $C_T$   
345 decreased considerably,  $pCO_2$  increased in spring.



346

347 **Figure 2:** Effects of surface water temperature, salinity, total alkalinity ( $A_T$ ) and total dissolved inorganic  
 348 carbon ( $C_T$ ) on the partial pressure of  $CO_2$  ( $pCO_2$ ) for each season for the stations located in the lower zone  
 349 of the Patos Lagoon Estuary region. The variation in each parameter is calculated as the difference between  
 350 the values of each parameter and their respective averages in previous seasons. The unit of all drivers is the  
 351 same as that for  $pCO_2$  ( $\mu atm$ ), and their magnitudes represent their influence on  $pCO_2^{sw}$  changes. The error  
 352 bars (gray) show the difference between the sum of all drivers and the actual variation in  $pCO_2$  ( $\Delta pCO_2^{drv}$ ),  
 353 indicating the extent to which the decomposition of  $pCO_2$  into its drivers differs from  $\Delta pCO_2^{drv}$ . More  
 354 details are given in the methods section.

355

### 356 3.2 Seasonal and interannual variability in air-water $CO_2$ net flux

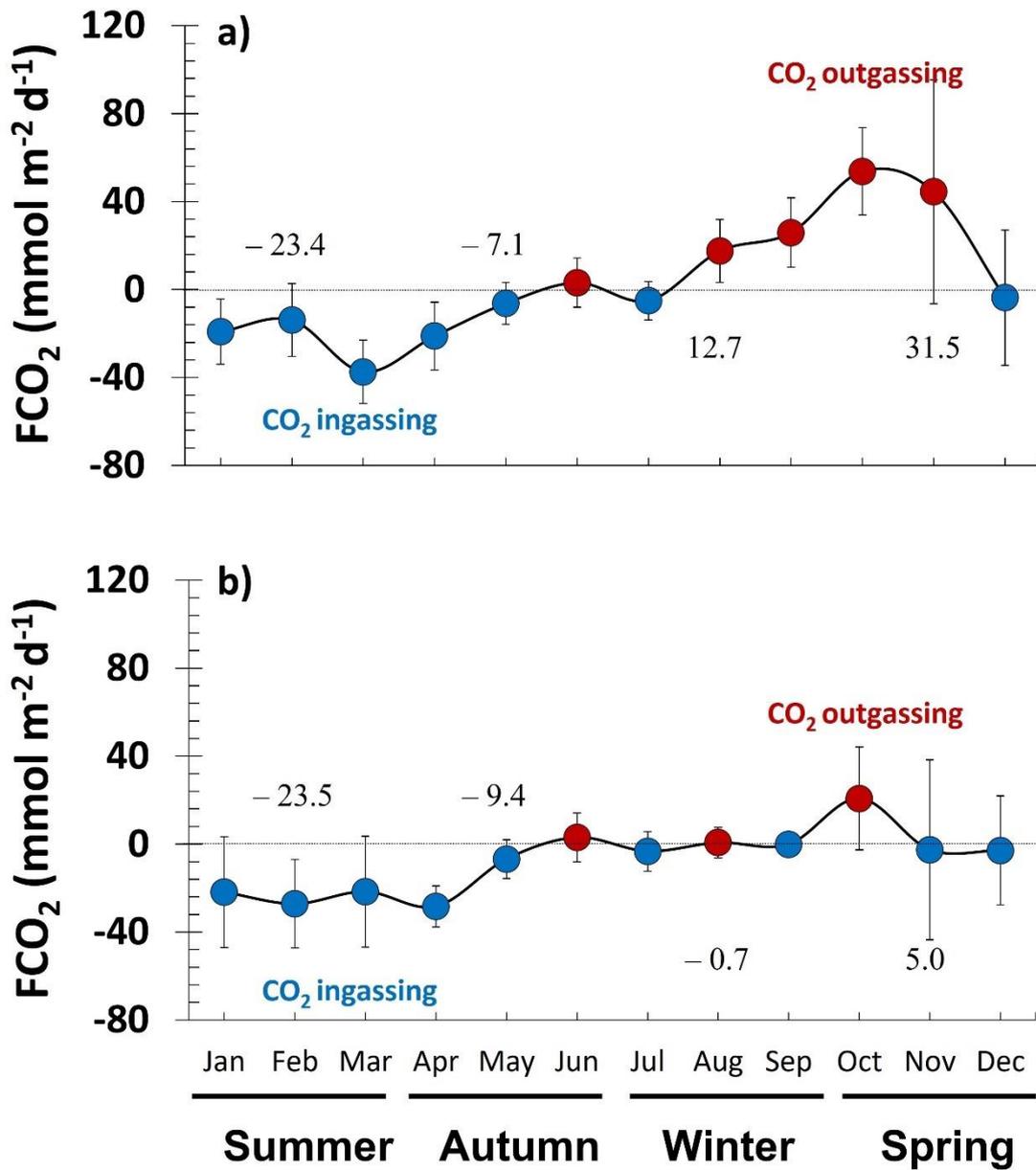
357 The seasonal cycle of water-air  $CO_2$  net fluxes showed a similar variability pattern in the  
 358 areas investigated but with different amplitudes at stations BrOA #1 and #2 (Figure 3a  
 359 and b). At the inner inlet station (BrOA #1), the seasonal amplitude of  $CO_2$  exchanges  
 360 varied from  $-38$  to  $54$   $mmol\ m^{-2}\ d^{-1}$  and was higher than that found at the mouth of the  
 361 estuary (BrOA #2), which ranged from  $-29$  to  $21$   $mmol\ m^{-2}\ d^{-1}$ . In addition, although the  
 362 net  $CO_2$  ingassing during summer/autumn seasons are very close to each other in both  
 363 areas (average of  $\sim -16 \pm 23$  and  $\sim -18 \pm 28$   $mmol\ m^{-2}\ d^{-1}$ ), the same is not true for the  
 364 behavior of the  $CO_2$  exchange during winter/spring seasons. The winter/spring net  $CO_2$   
 365 outgassing is 10-fold higher in the inner inlet zone (average of  $22 \pm 40$   $mmol\ m^{-2}\ d^{-1}$ )  
 366 than that found near the more sea-exposed zone (average of  $2 \pm 31$   $mmol\ m^{-2}\ d^{-1}$ ). In  
 367 general, the net  $CO_2$  outgassing was highest in spring (October and November) at the  
 368 BrOA #1 station, although this behavior was also observed during winter. The lower  
 369 estuarine zone of the Patos Lagoon acts as net  $CO_2$  ingassing during half the year, from

370 December to May, while a behavior of water-air CO<sub>2</sub> quasi-equilibrium is observed from  
371 June to September in the mouth of the estuary (Figure 3a and b).

372 Despite the high monthly variability and the marked seasonal cycle of the CO<sub>2</sub> exchanges  
373 observed in the PLE (Figures 3 and 4), during the 4-year period analyzed, the inner inlet  
374 zone behaved as net CO<sub>2</sub> outgassing to the atmosphere ( $2.9 \pm 41.4 \text{ mmol m}^{-2} \text{ d}^{-1}$ ), in  
375 contrast to the behavior as an estuarine net CO<sub>2</sub> sink zone ( $-7.2 \pm 33.3 \text{ mmol m}^{-2} \text{ d}^{-1}$ )  
376 observed in the area close to the ocean. Thus, the net behavior during the entire period  
377 revealed that the lower zone of the PLE behaved as an area of CO<sub>2</sub> uptake ( $-2.1 \pm 27.2$   
378  $\text{mmol m}^{-2} \text{ d}^{-1}$ ; Figure 4).

379 In 2019, there was intense CO<sub>2</sub> uptake ( $-23.9 \pm 33.8 \text{ mmol m}^{-2} \text{ d}^{-1}$ ) by the PLE surface  
380 waters, while in 2018, the region often released CO<sub>2</sub> ( $7.5 \pm 41.3 \text{ mmol m}^{-2} \text{ d}^{-1}$ ) to the  
381 atmosphere. The CO<sub>2</sub> outgassing was high in 2017. However, we did not evaluate the  
382 summer data when most of the CO<sub>2</sub> was absorbed. For 2021, we did not consider the  
383 winter/spring season, when CO<sub>2</sub> emissions are higher (Figure 4). According to our  
384 analysis, the water-air net CO<sub>2</sub> fluxes evolve at periods of 3 and 6 months and one year,  
385 mainly at the BrOA #1 station (Figure S1).

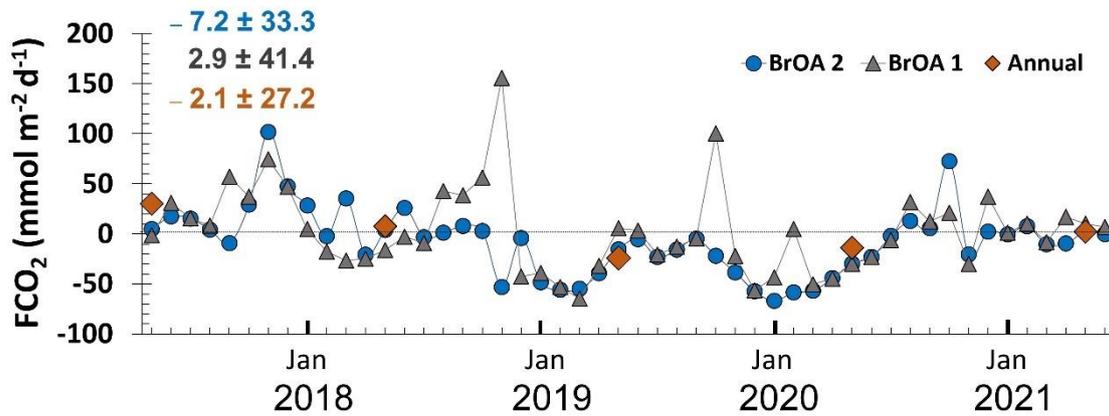
386



387

388 **Figure 3:** Seasonal cycle of surface water-air CO<sub>2</sub> net flux (FCO<sub>2</sub>) of the lower zone of the Patos Lagoon  
 389 Estuary obtained from May 2017 to June 2021 at the (a) pier-fixed station BrOA #1 (inner inlet) and (b)  
 390 BrOA #2 (sea-exposed area). The error bars indicate the standard error of each month. The values indicate  
 391 the average FCO<sub>2</sub> for each season. Blue dots indicate months of CO<sub>2</sub> ingassing, while red dots indicate  
 392 months of CO<sub>2</sub> outgassing.

393



394

395 **Figure 4:** Monthly and interannual variability in water-air surface CO<sub>2</sub> net flux (FCO<sub>2</sub>) for the BrOA #1  
 396 (gray triangles) and BrOA #2 (blue dots) stations at the Patos Lagoon Estuary from May 2017 to June 2021.  
 397 The orange diamonds depict the annual averages (note that 2017 and 2021 do not consider all the seasons).  
 398 The FCO<sub>2</sub> annual average and standard deviation for each estuarine station (color indicated by the legend)  
 399 and joined regions (orange) are indicated in the top left.

400

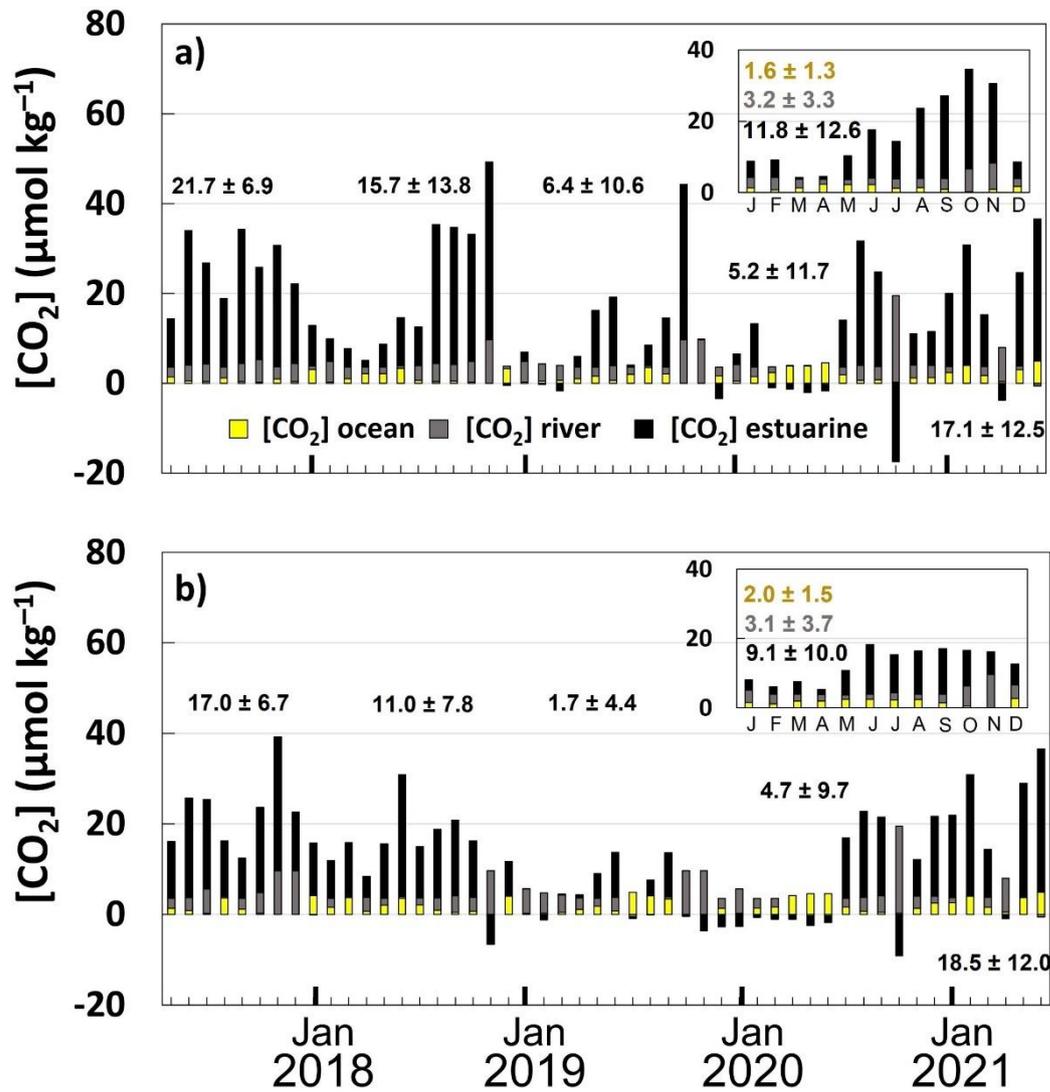
### 401 **3.3 River-borne, ocean-borne, and estuarine-generated CO<sub>2</sub>**

402 The seasonality of CO<sub>2</sub> concentrations is relatively different in each region analyzed. At  
 403 the inner inlet station, the seasonal cycle of the CO<sub>2</sub> estuarine is marked and varies from  
 404 -17 to 39 μmol kg<sup>-1</sup>, while at the mouth of the estuary, the CO<sub>2</sub> estuarine varies from -9  
 405 to 31 μmol kg<sup>-1</sup> (Figure 5a and b). Despite the difference in estuarine, ocean and river  
 406 CO<sub>2</sub> concentrations, the same variation was observed at both the inner inlet and estuarine  
 407 mouth stations. The CO<sub>2</sub> concentrations averaged  $1.8 \pm 0.4$  μmol kg<sup>-1</sup> for the ocean,  $2.2$   
 408  $\pm 0.5$  μmol kg<sup>-1</sup> for the river and  $5.2 \pm 3.2$  μmol kg<sup>-1</sup> for all lower zones of the estuary  
 409 during the summer/autumn seasons, while in the winter/spring seasons, the CO<sub>2</sub> estuarine  
 410 concentration was more than 2-fold higher (average of  $\sim 13 \pm 4.5$  μmol kg<sup>-1</sup>) than that  
 411 in other seasons. Thus, the lower estuarine zone of the Patos Lagoon acts as a region of  
 412 production of estuarine CO<sub>2</sub>, mainly during months of high freshwater discharge into the  
 413 lagoon (i.e., end of winter and spring; Figures 5 and S2), with great magnitudes of CO<sub>2</sub>  
 414 in the more protected embayment and less hydrodynamic zones. Regarding river-borne  
 415 CO<sub>2</sub>, the behavior of the PLE was different from that observed for estuarine CO<sub>2</sub>. The  
 416 river-borne CO<sub>2</sub> was higher in periods of high freshwater discharge, while in months  
 417 where the freshwater discharge was low, the input of CO<sub>2</sub> with marine sources was  
 418 intense.

419 Despite the small difference, the inner inlet zone produced more CO<sub>2</sub> ( $11.8 \pm 12.6$  μmol  
 420 kg<sup>-1</sup>) than the exposed area ( $9.1 \pm 10.0$  μmol kg<sup>-1</sup>). At the end of 2018, there was a

421 decrease in CO<sub>2</sub> estuarine production and an increase in CO<sub>2</sub> estuarine consumption in  
 422 the region, with more CO<sub>2</sub> estuarine consumption at the mouth of the estuary (Figure 5b)  
 423 than in the sheltered area. From July 2019, the ocean-borne CO<sub>2</sub> increased at both stations.

424



425

426 **Figure 5:** Monthly variability in aqueous surface CO<sub>2</sub> concentration ( $[\text{CO}_2]_{\text{ocean}}$ ; yellow bars,  $[\text{CO}_2]_{\text{river}}$ ;  
 427 gray bars, and  $[\text{CO}_2]_{\text{estuarine}}$ ; black bars) for the stations located in the lower zone of the Patos Lagoon Estuary  
 428 from May 2017 to June 2021 in the (a) BrOA #1 and (b) BrOA #2 pier-fixed stations with annual average  
 429 and standard deviation for CO<sub>2</sub> estuarine concentration. The seasonal cycle is inserted in the top right with  
 430 the average and standard deviation for each CO<sub>2</sub> concentration.

431

## 432 4. Discussion

### 433 4.1 Seasonal drivers of $p\text{CO}_2$

434 The  $A_T$  and  $C_T$  variations were the main causes of changes in  $pCO_2$  variability. The  
435 concentration of salts and dilution are the main processes identified in the  $pCO_2$  variation,  
436 indicated by increases (summer/autumn) and decreases (winter/spring) in  $A_T$  and  $C_T$   
437 (Figure 2). The lower estuarine zone of the Patos Lagoon shows a high amplitude of  $A_T$   
438 (479 to 2,245  $\mu\text{mol kg}^{-1}$ ) and  $C_T$  (500 to 2,040  $\mu\text{mol kg}^{-1}$ ; Albuquerque et al. under  
439 review), related to periods of increased (winter/spring) freshwater discharge and seawater  
440 inflows (summer/autumn). Furthermore, the PLE is a region with high organic matter  
441 concentrations due to large continental inputs, abundant macrophyte communities and  
442 local anthropogenic sources (Baumgarten and Niencheski 2010) that lead to heterotrophy  
443 and macroalgal blooms in spring, which directly influence the carbon concentration  
444 (Haraguchi et al. 2015; Lanari and Copertino 2017; Lanari et al. 2018). Another factor is  
445 the hydrodynamic processes that lead to large variations in salinity (Haraguchi et al.  
446 2015). These features aid in the final behavior of  $A_T$ ,  $C_T$  and salinity and indirectly affect  
447 the  $pCO_2$  variation.

448 The estimated deviations of the analysis (Figure 2) are an indication that some estuarine  
449 processes were not yet accounted for in the  $pCO_2$  changes, such as the influence of  
450 phytoplankton biomass (e.g., Lee et al. 2006; Cai 2011) and the property changes related  
451 to groundwater input (e.g., Sadat-Noori et al. 2016; Jeffrey et al. 2018), which configures  
452 a current limitation and a challenge for further studies. The higher freshwater discharge  
453 in late winter and during spring brings a higher concentration of organic matter,  
454 supporting phytoplankton blooms in the PLE (Albuquerque et al. under review). The  
455 deviations observed in summer and spring are probably due to photosynthesis rates  
456 (Albuquerque et al. under review), which leads to a decrease in  $pCO_2$  levels. Groundwater  
457 may be enriched in dissolved carbon species; therefore, groundwater input affects the  
458 carbon concentration in water (Sadat-Noori et al. 2016), increasing the  $pCO_2$   
459 concentration. Although the approach used in this study is mostly used to infer the main  
460 variables and processes affecting the  $pCO_2$  distribution for oceanic waters (e.g.,  
461 Takahashi et al. 2014; Moreau et al. 2017; Monteiro et al. 2020), it is also a useful  
462 approach in estuarine systems that have periods of waters with high salinity signals.  
463 However, as previously indicated, further investigation of coupled estuarine-  
464 biogeochemical processes must be performed to reduce uncertainties, mainly during  
465 freshwater dominance periods in the estuarine system.

466

## 467 **4.2 Water-air $CO_2$ fluxes and $CO_2$ concentrations**

488 Estuarine systems are significant sources of CO<sub>2</sub> to the atmosphere with recognized  
489 relevance for regional and global carbon budgets (e.g., Guo et al. 2009; Gupta et al. 2009;  
490 Jeffrey et al. 2018; Joesoef et al. 2015), and estuarine lower zones realize approximately  
491 23 mmol m<sup>-2</sup> d<sup>-1</sup> (Chen et al. 2013). The magnitude of the CO<sub>2</sub> exchanges in the lower  
492 estuarine zone of the PLE is variable, with values ranging from -66 to 155 mmol m<sup>-2</sup> d<sup>-1</sup>  
493 and a net CO<sub>2</sub> ingassing average of -2 mmol m<sup>-2</sup> d<sup>-1</sup> for the 4-year period investigated  
494 here (Figure 4; Table 1). The variations observed in the lower zone of the PLE display a  
495 typical distribution pattern seen in other estuarine environments (Table 1), mainly in  
496 shallow estuaries dominated by freshwater discharge, such as the PLE (e.g., Evans et al.  
497 2013; Koné et al. 2009; Table 1). Moreover, the same range of water-air CO<sub>2</sub> exchanges  
498 observed in the subtropical PLE can be found in other tropical estuarine environments in  
499 Brazil (e.g., Cotovicz et al. 2020; Noriega and Araujo 2014), shifting between periods of  
500 CO<sub>2</sub> ingassing and CO<sub>2</sub> outgassing to the atmosphere over the year.

501 **Table 1:** Comparison between the range of water-air CO<sub>2</sub> flux (FCO<sub>2</sub>) from previous estuarine studies around the world and the current study. The  
 502 studies are organized first by country and second by climate.

Reference	Estuary location	Country	Climate	Type	Period	Water-air FCO <sub>2</sub> (range in mmol m <sup>-2</sup> d <sup>-1</sup> )
<b>This study</b>	Lower zone of PLE South Brazil	Brazil	Subtropical	Freshwater discharge dominated	May 2017 to June 2021	-66 to 155
<b>Cotovicz et al. 2020</b>	Paraíba do Sul River Southwestern Brazil	Brazil	Tropical	Freshwater discharge dominated	February 2017 October 2017 March 2018	-37.8 to 216.1
<b>Noriega and Araujo 2014</b>	North and Northeast Brazilian estuaries	Brazil	Tropical and semiarid	Various	July 2012 to June 2013	2.4 to 175.2
<b>Chen et al. 2020</b>	Chesapeake Bay	USA	Subtropical	Partially mixed and microtidal	March to December 2016 February 2019	-11.2 to 92.4
<b>Yao and Hu 2017</b>	Mission-Aransas	USA	Subtropical semiarid	Freshwater discharge dominated	May 2014 to April 2015	-13.5 to 380.3
<b>Ho et al. 2014</b>	Shark River	USA	Tropical	Mangrove	November 2010 November 2011	20 to 118
<b>Evans et al. 2013</b>	Columbia River	USA	Temperate	Freshwater discharge dominated	August 2007 November 2007 April 2008 June 2008 September 2008	-53.0 to 193.2
<b>Joesoef et al. 2015</b>	Delaware	USA	Temperate	Freshwater discharge dominated	June 2013 August 2013 October 2013 November 2013 March 2014 July 2014 August to September 2014 October to November 2014 December 2014	-21.0 to 129.1

<b>Crosswell et al. 2012</b>	Neuse River	USA	Temperate	Macrotidal	June 2009 to July 2010	−38 to 271
<b>Dinauer and Mucci 2017</b>	St. Lawrence	Canada	Temperate	Freshwater discharge dominated	July 2003 June 2006 May 2007 July 2007 June 2009 July 2009 July 2010 May 2011 June 2013 May 2016	−21.9 to 28.4
<b>Koné et al. 2009</b>	Lagoon's system	Ivory Coast	Tropical	Microtidal	June to July 2006 September 2006 November to December 2006 March 2007	−20.0 to 186.2
<b>Gupta et al. 2009</b>	Cochin	India	Tropical	Microtidal	February 2005 April 2005 September 2005	64 to 274
<b>Guo et al. 2009</b>	Pearl River	China	Subtropical	Freshwater discharge dominated	November 2002 February 2004 January 2005 August 2005 April 2007	−25.8 to 907.7
<b>Oliveira et al. 2017</b>	Tagus	Portugal	Temperate	Tide dominated	1999 to 2007	16.6 to 347.1
<b>Borges et al. 2004</b>	Scheldt	Netherlands	Temperate	Macrotidal	November 2002 April 2003	31 to 2,189
<b>Bozec et al. 2012</b>	Loire	France	Temperate	Tide dominated	April 2009 July 2009 October 2009 February to March 2010	−9 to 140
<b>Flecha et al. 2015</b>	Guadalquivir	Spain	Mediterranean	Mesotidal	November 2007 to August 2009	−0.7 to 83.9

**Maher and Eyre 2012**

Hastings River	Australia	Temperate	Wave dominated	June 2006	-8.7 to 25.3
Camden Haven	Australia	Temperate	Wave dominated	October 2006	-16.5 to 12
Wallis Lake	Australia	Temperate	Wave dominated	February 2007	-30.4 to 26.2
				April 2007	

503

504 Significant seasonal variation in the water-air CO<sub>2</sub> fluxes between summer/autumn and  
505 winter/spring was observed in the lower zone of the PLE. The results presented here  
506 demonstrate that the region experienced periods of both CO<sub>2</sub> ingassing (between  
507 December and May) and CO<sub>2</sub> outgassing (between June and November), mainly in  
508 protected areas along the estuary (Figure 3b). CO<sub>2</sub> exchanges are modulated by winds  
509 (Wanninkhof 2014), different hydrodynamic conditions and/or extreme events (Sims et  
510 al. 2021). The higher the wind speed is, the greater the influence on the gas exchange  
511 between the surface water and the atmosphere; thus, wind speed can play a crucial role in  
512 estuarine CO<sub>2</sub> fluxes (e.g., Yao et al. 2020) by exchanging momentum with the  
513 atmosphere. In addition, a shallower environment responds faster to wind action (Moller  
514 et al. 2001; Lanari and Copertino 2017), which seems to favor fast air-water gas exchange  
515 (Yao et al. 2020) and increased CO<sub>2</sub> gradient magnitudes. The water-air CO<sub>2</sub> fluxes in the  
516 lower zone of the PLE follow the seasonal pattern of freshwater discharge, which is what  
517 occurred with the variability in the carbonate system parameters (Albuquerque et al. under  
518 review). In late winter and spring, the freshwater discharge is higher (Marques 2012), and  
519 consequently, the entry of organic matter promoting significant heterotrophy in relation  
520 to CO<sub>2</sub> is also higher, which is what occurs in the tropical coastal systems in northern and  
521 northeastern Brazil (Noriega and Araújo 2014). In the same period, the higher wind speed  
522 (~ 4.0 m/s; Figure S2) leads to higher gas transfer velocity in this period with respect to  
523 that of summer and autumn. The inner inlet station receives more influence from the  
524 freshwater discharge and has a high residence time, leading to an increase in CO<sub>2</sub>, while  
525 the sea-exposed station is close to the ocean and directly influenced by seawater dynamics  
526 and outflow/inflow currents (Möller et al. 2001; Lisboa 2015). Thereafter, CO<sub>2</sub>  
527 outgassing in the inner inlet zone should be mainly related to heterotrophic respiration  
528 and degradation of organic matter generated on a larger scale by the influence of  
529 freshwater discharge, even though this area is also influenced by resuspension of fine  
530 bottom sediments (Moller et al. 2001; Lanari and Copertino 2017).

531 The phytoplanktonic community composition exerts a fundamental role in CO<sub>2</sub> uptake  
532 via photosynthesis. In the PLE, diatoms are the dominant phytoplankton group, followed  
533 by cyanobacteria, flagellates, dinoflagellates, and chlorophytes along the coastal offshore  
534 gradient (Islabão et al. 2017). Microphytoplanktonic diatoms tend to show a higher  
535 capacity for capturing CO<sub>2</sub> in water (e.g., Hopkinson et al. 2011) than nano- and  
536 picoplanktonic species, which helps to balance the carbon concentration. The production  
537 of CO<sub>2</sub> was lower in the more sea-exposed area than in the inner inlet station, suggesting

538 that the abundance of diatoms in the region closest to the coast is higher due to local  
539 hydrodynamics, mainly at the end of winter and spring. Moreover, in spring (October and  
540 November), the blooms caused by biological activity combined with vertical mixing due  
541 to the higher wind intensity lead to highly variable water-air CO<sub>2</sub> fluxes in the lower zone  
542 of the PLE (Figure 3). Despite the different tide regimes, these seasonal conditions are  
543 similar to those previously observed in the macrotidal Neuse River estuary (Crosswell et  
544 al. 2012) and microtidal Aby Lagoon (Koné et al. 2009). The water temperature  
545 influences the CO<sub>2</sub> fluxes because water-air CO<sub>2</sub> exchanges depend on the gas transfer  
546 velocity, which is further induced by temperature that changes the solubility of gas in  
547 water (Wanninkhof 2014). The higher summer temperatures combined with the CO<sub>2</sub>-  
548 unsaturated waters from the ocean turn the region into a CO<sub>2</sub> sink in summer and autumn.  
549 In general, freshwater input, wind and water temperature are important in the exchange  
550 of CO<sub>2</sub>; nevertheless, residence time and phytoplanktonic composition also modulate the  
551 CO<sub>2</sub> dynamics in different areas of the lower zone of the PLE over the seasons.

552 The production or consumption of CO<sub>2</sub> is an indicator of estuarine-biogeochemical  
553 processes (Yao et al. 2020) and plays a crucial role in determining the water-air CO<sub>2</sub>  
554 fluxes of the lower zone of the PLE. Most of the CO<sub>2</sub> released into the atmosphere is  
555 produced in the PLE itself. The high freshwater discharge between the end of winter and  
556 spring carries organic matter (Niencheski and Windom 1994; Niencheski et al. 2006) that  
557 favors photosynthetic respiration and degradation of organic carbon. These processes  
558 intensify the production of carbon for the estuary. Although the source is the river,  
559 heterotrophy occurs in the estuary because there is more time for the decomposition of  
560 organic matter. During this same period, CO<sub>2</sub> inputs from the river are also significant.  
561 These two CO<sub>2</sub> sources contribute to CO<sub>2</sub> supersaturation and outgassing to the  
562 atmosphere. Therefore, the observed seasonal pattern of CO<sub>2</sub> fluxes and different CO<sub>2</sub>  
563 sources suggest that heterotrophy controls the metabolic status of the estuarine waters,  
564 mainly in the inner areas of the PLE. In addition, the produced carbon is exported to the  
565 coast, which contributes to the high concentration of CO<sub>2</sub> in the mouth of the estuary.  
566 This region is a channel with high hydrodynamics (Möller et al. 2001), which allows  
567 faster currents and more intense water exchanges. Thus, this area prevents the surface  
568 water from CO<sub>2</sub> supersaturation by releasing the gas into the atmosphere.

569 The interannual variability in estuarine properties and processes in the PLE is influenced  
570 by El Niño-Southern Oscillation (ENSO) events (e.g., Odebrecht et al. 2017;  
571 Albuquerque et al. under review). The warm phase of ENSO, El Niño, is characterized

572 by abnormal heating of surface waters in the tropical Pacific Ocean. ENSO affects the  
573 regional and global climate, changing the wind patterns worldwide and thus affecting the  
574 rainfall patterns in tropical and midlatitude regions (e.g., Cai et al. 2020). Between May  
575 2017 and June 2021, two ENSO events were identified. These events included one El  
576 Niño event (in 2017/2018; ENSO warm phase; INPE 2021) and one La Niña event (in  
577 2020/2021; ENSO cold phase; INPE 2021). Under El Niño conditions, freshwater  
578 discharge is above the average level ( $\sim 1770 \text{ m}^3 \text{ s}^{-1}$ ), and the opposite is true under La  
579 Niña conditions (Vaz et al. 2006; Seeliger and Odebrecht 2010). The decreased water-air  
580  $\text{CO}_2$  fluxes from the beginning of monitoring, the increased  $\text{CO}_2$  exchanges in the last  
581 two years (Figure 4) and the composition of  $\text{CO}_2$  (Figure 6) are likely associated with the  
582 change from moderate El Niño (2017–2018) to the beginning of La Niña conditions  
583 (2019–2020). However, since the time series is relatively short, with some years (2017  
584 and 2021) having missing months, further discussion on this matter is inhibited. Future  
585 research focusing on longer time series is still needed to better characterize the role of  
586 climate mode teleconnections with changes in the water-air  $\text{CO}_2$  exchanges in the PLE.

587 Finally, it is worth mentioning that at the end of 2018 (October to December) and between  
588 November 2019 and January 2020, there was a dredging event in the port access channel  
589 located in the estuary (Mirlean et al. 2020). These dredging activities resuspend the  
590 sediment, allowing nutrients and carbon to return to the water column and producing  
591 some impacts (e.g., water chemical alteration, mud deposition, and changes in the benthos  
592 composition), and a monitoring program should always be considered to evaluate their  
593 environmental consequences (Torres and Philomena 2013; Mirlean et al. 2020). The  
594 environmental impacts associated with the dredging process and spoil disposal can be  
595 characterized by direct effects on organisms and habitats and indirect effects attributed to  
596 alterations in water quality. Thus, in addition to likely ENSO effects on the region, the  
597 behavior of the studied area as a  $\text{CO}_2$  ingassing zone (Figure 4) and the increase in  
598 estuarine  $\text{CO}_2$  consumption from that period (Figure 5) onward may also be influenced  
599 by changes in the water properties and water-sediment processes caused by dredging,  
600 despite the controversial about the causes and consequences of mud deposition events  
601 along the coast (Calliari et al. 2022; Mirlean et al. 2020, 2021; Garcia et al. 2021).

602

## 603 **5. Conclusion**

604 Overall, the estuarine ecosystem of the Patos Lagoon behaved as summer/autumn CO<sub>2</sub>  
605 ingassing and winter/autumn CO<sub>2</sub> outgassing to the atmosphere. The combined effect of  
606 wind speed, continental freshwater discharge, inflow/outflow currents, water  
607 temperature, biological activity, and water residence time is responsible for the  
608 modulation of the CO<sub>2</sub> exchange in the PLE. However, comparing the different areas of  
609 the PLE lower zone, we noticed non-similar forcings acting on the variation in the CO<sub>2</sub>  
610 concentration. The water-air CO<sub>2</sub> flux variations at the mouth of the estuary were mainly  
611 driven by the balance between the seawater dynamics and freshwater discharge  
612 input/output, which influences the scenario of a CO<sub>2</sub> sink. In the inner estuary,  
613 heterotrophic respiration and degradation of organic matter contributed to CO<sub>2</sub> outgassing  
614 to the atmosphere between the end of winter and spring. Autochthonous production was  
615 responsible for the highest concentration of CO<sub>2</sub>, indicating heterotrophy in the estuarine  
616 surface water. Part of the carbon produced is exported to the coast, contributing to a high  
617 concentration of CO<sub>2</sub> in the mouth of the estuary. Therefore, this study shows that the  
618 lower zone of the PLE is resilient to high CO<sub>2</sub> concentrations and has been able to  
619 overcome anthropogenic emissions in the region. Hence, long-term monitoring programs  
620 must continue to improve the understanding of CO<sub>2</sub> exchange variability and help shed  
621 light on the role played by the PLE on the global carbon budget.

622

## 623 **References**

- 624 Abreu PC, Marangoni J, Odebrecht C (2016) So close, so far: differences in long-term  
625 chlorophyll a variability in three nearby estuarine-coastal stations. *Marine Biology*  
626 *Research*. <https://doi.org/10.1080/17451000.2016.1189081>
- 627 Abril G, Libardoni BG, Brandini N et al (2021) Thermodynamic uptake of atmospheric  
628 CO<sub>2</sub> in the oligotrophic and semiarid São Francisco estuary (NE Brazil). *Marine Chemistry*.  
629 <https://doi.org/10.1016/j.marchem.2021.103983>
- 630 Albuquerque C, Kerr R, Monteiro T et al (Under review) Seasonal and interannual in the  
631 carbonate system parameters of the Patos Lagoon Estuary.
- 632 Bauer JE, Cai WJ, Raymond PA et al (2013) The Changing Carbon Cycle of the Coastal  
633 Ocean. *Nature*. <https://doi.org/10.1038/nature12857>
- 634 Baumgarten MGZ, Niencheski LFH (2010) A coluna sedimentar como reservatório e  
635 fonte de nutrientes em enseadas estuarinas. *Tropical Oceanography Online*. 38:88–104
- 636 Borges AV, Abril G (2011) Carbon Dioxide and Methane Dynamics in Estuaries.  
637 *Estuarine and Coastal Science*. <https://doi.org/10.1016/B978-0-12-374711-2.00504-0>

638 Borges AV, Delille B, Schiettecatte L et al (2004) Gas transfer velocities of CO<sub>2</sub> in three  
639 European estuaries (Randers Fjord, Scheldt, and Thames). *Limnology and*  
640 *Oceanography*. 49:1630–1641

641 Bozec Y, Cariou T, Macé E et al (2012) Seasonal dynamics of air-sea CO<sub>2</sub> fluxes in the  
642 inner and outer Loire estuary (NW Europe). *Estuarine, Coastal and Shelf Science*.  
643 <https://doi.org/10.1016/j.ecss.2011.05.015>

644 Cai W (2011) Estuarine and coastal ocean carbon paradox: CO<sub>2</sub> sinks or sites of terrestrial  
645 carbon incineration? *Annual Review of Marine Science*. [https://doi.org/10.1146/annurev-](https://doi.org/10.1146/annurev-marine-120709-142723)  
646 [marine-120709-142723](https://doi.org/10.1146/annurev-marine-120709-142723)

647 Cai W, McPhaden MJ, Grimm AM et al (2020) Climate impacts of the El Niño-Southern  
648 Oscillation on the South America. *Nature Reviews Earth & Environment*.  
649 <https://doi.org/10.1038/s43017-020-0040-3>

650 Calliari LJ, Machado AA, Marroig P et al (2020) Mud deposits at Cassino beach: role of  
651 dredging. *Geo-Marine Letters*. <https://doi.org/10.1007/s00367-019-00619-6>

652 Carstensen J, Chierici M, Gustafsson BG, Gustafsson, E (2018) Long-term and seasonal  
653 trends in estuarine and coastal carbonate systems. *Global Biogeochemical Cycles*.  
654 <https://doi.org/10.1002/2017GB005781>

655 Castelão RM, Möller OO (2003) Sobre a circulação tridimensional forçada por ventos na  
656 Lagoa dos Patos. *Atlântica*, 25:91–106

657 Chen B, Cai W, Brodeur JR et al (2020) Seasonal and spatial variability in surface pCO<sub>2</sub>  
658 and air-water CO<sub>2</sub> flux in the Chesapeake Bay. *Limnology and Oceanography*.  
659 <https://doi.org/10.1002/lno.11573>

660 Chen C-TA., Huang T-H, Chen, Y-C et al (2013) Air-sea exchanges of CO<sub>2</sub> in the world's  
661 coastal seas. *Biogeosciences*. <https://doi.org/10.5194/bg-10-6509-2013>

662 Cloern JE, Foster SQ, Kleckner AE (2014) Phytoplankton primary production in the  
663 world's estuarine coastal ecosystems. *Biogeosciences*. 11:2477–2501

664 Cotovicz LC, Knoppers BA, Brandini N, et al (2015) A strong CO<sub>2</sub> sink enhanced by  
665 eutrophication in a tropical coastal embayment (Guanabara Bay, Rio de Janeiro, Brazil).  
666 *Biogeosciences*. <https://doi.org/10.5194/bg-12-6125-2015>

667 Cotovicz LC, Vidal LO, Rezende CE et al (2020) Carbon dioxide sources and sinks in  
668 the delta of the Paraíba do Sul River (Southeastern Brazil) modulated by carbonate  
669 thermodynamics, gas exchange and ecosystem metabolism during estuarine mixing.  
670 *Marine Chemistry*. <https://doi.org/10.1016/j.marchem.2020.103869>

671 Crosswell JR, Wetz MS, Hales B, Paerl HW (2012) Air-water CO<sub>2</sub> fluxes in the  
672 microtidal Neuse River estuary, North Carolina. *Journal of Geophysical Research*.  
673 <https://doi.org/10.1029/2012JC007925>

674 Dickson AG, Sabine CL, Christian JR (2007) *Guide to Best Practices for Ocean CO<sub>2</sub>*  
675 *Measurements*. North Pacific Marine Science Organization, Sidney.

676 Dickson AG (1990) Thermodynamics of the dissociation of boric acid in synthetic  
677 seawater from 273.15 to 318.15 K. *Deep-Sea Res.* 37:755–766

678 Dinauer A, Mucci A (2017) Spatial variability in surface-water pCO<sub>2</sub> and gas exchange  
679 in the world's largest semi-enclosed estuarine system: St. Lawrence Estuary (Canada).  
680 *Biogeoscience*. <https://doi.org/10.5194/bg-14-3221-2017>

681 Evans W, Hales B, Strutton PG (2013) pCO<sub>2</sub> distributions and air-water CO<sub>2</sub> fluxes in the  
682 Columbia River estuary. *Estuarine, coastal and shelf science*. 117:260–272

683 Feely RA, Doney SC, Cooley SR (2010) Ocean acidification: Present conditions and  
684 future changes in a high CO<sub>2</sub> world. *Oceanography*. 22:36–47

685 Flecha S, Huertas IE, Navarro G, et al (2015) Air–Water CO<sub>2</sub> Fluxes in a Highly  
686 Heterotrophic Estuary. *Estuaries and Coasts*. <https://doi.org/10.1007/s12237-014-9923-1>

687 Garcia CAE, Evangelista H, Möller, OO (2021) Comments on -Dredging in an estuary  
688 causes contamination by fluid mud on a tourist ocean beach. Evidence via REE ratios- by  
689 N. Mirlean, L. Calliari, and K. Johannesson in *Marine Pollution Bulletin* 159 (2020)  
690 111495. *Marine Pollution Bulletin*. <https://doi.org/10.1016/j.marpolbul.2021.112115>

691 Gupta GVM, Thottathil SD, Balachandran KK et al (2009) CO<sub>2</sub> supersaturation and net  
692 heterotrophy in a tropical estuary (Cochin, India): influence of anthropogenic effect.  
693 *Ecosystem*. <https://doi.org/10.1007/s10021-009-9280-2>

694 Guo X, Dai M, Zhai W et al (2009) CO<sub>2</sub> flux and seasonal variability in a large subtropical  
695 estuarine system, the Pearl River Estuary, China. *Journal of Geophysical Research*.  
696 <https://doi.org/10.1029/2008JG000905>

697 Haraguchi L, Carstensen J, Abreu PC, Odebrecht C (2015) Long-term changes of the  
698 phytoplankton community and biomass in the subtropical shallow Patos Lagoon Estuary.  
699 *Estuarine, Coastal and Shelf Science*. 162:76–87

700 Ho DT, Ferrón S, Engel VC et al (2014) Air-water gas exchange and CO<sub>2</sub> flux in a  
701 mangrove-dominated estuary, *Geophysical Research Letter*.  
702 <https://doi.org/10.1002/2013GL058785>

703 Hopkinson BM, Dupont CL, Allen AE, Morel FMM (2011) Efficiency of the CO<sub>2</sub>-  
704 concentrating mechanism of diatoms. *Proceeding of the National Academy of Sciences*  
705 of the United States of America. <https://doi.org/10.1073/pnas.1018062108>

706 INPE. 2021. <http://enos.cptec.inpe.br/>. Accessed on 17 August 2021.

707 Islabão CA, Mendes CRB, Detoni AMS, Odebrecht C (2017) Phytoplankton community  
708 structure in relation to hydrographic features along a coast–to–offshore transect on the  
709 SW Atlantic continental shelf. *Continental shelf research*.  
710 <https://doi.org/10.1016/j.csr.2017.10.003>

711 Jeffrey LC, Maher DT, Santos IR et al (2018) The spatial and temporal drivers of pCO<sub>2</sub>,  
712 pCH<sub>4</sub> and gas transfer velocity within a subtropical estuary. *Estuarine, Coastal and Shelf*  
713 *Science*. <https://doi.org/10.1016/j.ecss.2018.04.022>

- 714 Jiang LQ, Cai W, Wang Y (2008) A comparative study of carbon dioxide degassing in  
715 river- and marine dominated estuaries. *Limnology and Oceanography*.  
716 <https://doi.org/10.4319/lo.2008.53.6.2603>
- 717 Joesoef A, Huang W-J, Gao Y, Cai, W-J (2015) Air-water fluxes and sources of carbon  
718 dioxide in Delaware Estuary: spatial and seasonal variability. *Biogeosciences*.  
719 <https://doi.org/10.5194/bg-12-6085-2015>
- 720 Kerr R, Da Cunha LC, Kikuchi RKP et al (2016) The Western South Atlantic Ocean in a  
721 high-CO<sub>2</sub> world: current measurement capabilities and perspectives. *Environmental*  
722 *Management*. <https://doi.org/10.1007/s00267-015-0630-x>
- 723 Koné YJM, Abril G, Kouadio KN et al (2009) Seasonal variability of carbon dioxide in  
724 the rivers and lagoons of ivory coast (West Africa). *Estuaries and Coasts*.  
725 <https://doi.org/10.1007/s12237-008-9121-0>.
- 726 Lanari M, Copertino M (2017) Drift macroalgae in the Patos Lagoon Estuary (southern  
727 Brazil): effects of climate, hydrology and wind action on the onset and magnitude of  
728 blooms. *Marine Biology Research* (online).  
729 <https://doi.org/10.1080/17451000.2016.1225957>
- 730 Lanari M, Copertino M, Cooling LA, Bom FC (2018) The impact of short-term  
731 depositions of macroalgal blooms on widgeon-grass meadows in a river-dominated  
732 estuary. *Harmful algae*. <https://doi.org/10.1016/j.hal.2018.07.006>
- 733 Lee K, Tong LT, Millero FJ et al (2006) Global relationships of total alkalinity with  
734 salinity and temperature in surface waters of the world's oceans. *Geophysical Research*  
735 *Letters*. <https://doi.org/10.1029/2006GL027207>
- 736 Lemos VM, Lanari M, Copertino M et al (2021) Patos Lagoon Estuary and adjacent  
737 marine coastal biodiversity long-term data. *Earth Syst. Sci. Data Discuss.* [preprint].  
738 <https://doi.org/10.5194/essd-2021-353>
- 739 Lenton A, Metzl N, Takahashi T (2012) The observed evolution of oceanic pCO<sub>2</sub> and its  
740 drivers over the last two decades *Global Biogeochemical Cycles*.  
741 <https://doi.org/10.1029/2011GB004095>
- 742 Lewis E, Wallace D, Allison LJ (1998) Program developed for CO<sub>2</sub> system calculations.  
743 Carbon Dioxide Information Analysis Center, USA.
- 744 Lisboa PV (2015) Escalas temporais de transporte do estuário da Lagoa dos Patos, Rio  
745 Grande, RS. Dissertation, University of Rio Grande, Rio Grande do Sul, Brasil
- 746 Liu Q, Charette MA, Breier CF (2017) *Geochimica et Cosmochimica Acta*.  
747 <https://doi.org/10.1016/j.gca.2017.01.041>
- 748 Maher DT, Eyre BD (2012) Carbon budgets for three autotrophic Australian estuaries:  
749 Implications for global estimates of the coastal air-water CO<sub>2</sub> flux. *Global*  
750 *biogeochemical cycles*. <https://doi.org/10.1029/2011GB004075>

751 Marques, WC (2012) The temporal variability of the freshwater discharge and water  
752 levels at the Patos Lagoon, Brazil. *International Journal of Geosciences*.  
753 <https://doi.org/10.4236/ijg.2012.34076>

754 Marques WC, Fernandes EH, Monteiro IO, Möller OO (2009) Numerical modeling of  
755 the Patos Lagoon coastal plume, Brazil. *Continental Shelf Research*. 29:556–571

756 Marques WC, Möller OO (2008) Variabilidade temporal em longo período da descarga  
757 fluvial e níveis de água da Lagoa dos Patos, Rio Grande do Sul, Brasil. *Revista Brasileira*  
758 *de Recursos Hídricos*. 13: 155–163

759 Mendes CRB, Odebrecht C, Tavano VM, Abreu PC (2016) Pigment-based  
760 chemotaxonomy of phytoplankton in the Patos Lagoon estuary (Brazil) and adjacent  
761 coast. *Marine Biology Research*. <https://doi.org/10.1080/17451000.2016.1189082>

762 Millero FJ (2013) *Chemical Oceanography*. CRC Press.

763 Millero FJ, Graham TB, Huang F et al (2006) Dissociation Constants of Carbonic Acid  
764 in Seawater as a Function of Salinity and Temperature. *Marine Chemistry*.  
765 <https://doi.org/10.1016/j.marchem.2005.12.001>

766 Mirlean N, Calliari L, Johannesson K (2020) Dredging in an estuary causes contamination  
767 by fluid mud on a tourist ocean beach. Evidence via REE ratios. *Marine Pollution*  
768 *Bulletin*. <https://doi.org/10.1016/j.marpolbul.2020.111495>

769 Mirlean N, Calliari L, Johannesson K (2021) Reply to MPB-D-20-01629. Carlos Alberto  
770 Eiras Garcia Heitor Evangelista Osmar Olinto Möller Jr. Comments on -Dredging in an  
771 estuary causes contamination by fluid mud on a tourist ocean beach. Evidence via REE  
772 ratios- by N. Mirlean, L. Calliari, and K. Johannesson in *Marine Pollution Bulletin* 159  
773 (2020) 111495. *Marine Pollution Bulletin*.  
774 <https://doi.org/10.1016/j.marpolbul.2021.112161>

775 Möller OO, Castaing P, Salomon JC, Lazure P (2001) The Influence of Local and Non-  
776 Local Forcing Effects on the Subtidal Circulation of Patos Lagoon. *Estuaries*. 24:297–  
777 311

778 Möller OO, Fernandes EH (2010) Hidrologia e Hidrodinâmica. In: Seeliger, U. and  
779 Odebrecht, C. (Eds.) *O Estuário da Lagoa dos Patos: Um século de transformações*. Rio  
780 Grande, FURG, pp: 17–30.

781 Monteiro T, Kerr R, Orselli IBM, Lencina-Avila JM (2020) Towards an intensified  
782 summer CO<sub>2</sub> sink behaviour in the Southern Ocean coastal regions. *Progress in*  
783 *Oceanography*. <https://doi.org/10.1016/j.pocean.2020.102267>

784 Moreau S, Penna AD, Llort J et al (2017) Eddy-induced carbon transport across the  
785 Antarctic circumpolar current. *Global Biogeochemical Cycles*.  
786 <https://doi.org/10.1002/2017GB005669>

787 Niencheski LF, Baumgarten MGZ, Cabrera L, Juliano SK (2006) Patos Lagoon:  
788 indicators of organic pollution. *Journal of Coastal Research*. 39:1357–1359

789 Niencheski LF, Windom, HL (1994) Nutrient flux and budget in Patos Lagoon estuary.  
790 Science of the Total Environment. [https://doi.org/10.1016/0048-9697\(94\)90004-3](https://doi.org/10.1016/0048-9697(94)90004-3)

791 Noriega C, Araujo M (2014) Carbon dioxide emissions from estuaries of northern and  
792 northeastern Brazil. Scientific Reports. <https://doi.org/10.1038/srep06164>

793 Noriega CED, Araujo M, Lefèvre N (2013) Spatial and temporal variability of the CO<sub>2</sub>  
794 fluxes in a tropical, highly urbanized estuary. Estuaries and coast.  
795 <https://doi.org/10.1007/s12237-013-9608-1>

796 Odebrecht C, Abreu PCOV (2019) Phytoplankton and water quality parameters in the  
797 Patos Lagoon estuary and adjacent marine coast. Sistema de Informação sobre a  
798 Biodiversidade Brasileira - SiBBr. Sampling event  
799 dataset <https://doi.org/10.15468/xmlvxn> accessed via GBIF.org on 2020-06-09

800 Odebrecht C, Secchi ER, Abreu PC, Muelbert JH, Uiblein F (2017) Biota of the Patos  
801 Lagoon estuary and adjacent marine coast: long-term changes induced by natural and  
802 human-related factors. Marine Biology Research.  
803 <https://doi.org/10.1080/17451000.2016.1258714>

804 Oliveira AP, Cabeçadas G, Mateus MD (2017) Inorganic carbon distribution and  
805 CO<sub>2</sub> fluxes in a large European estuary (Tagus, Portugal). Scientific  
806 Reports. <https://doi.org/10.1038/s41598-017-06758-z>

807 Orr JC, Epitalon J, Dickson AG, Gattuso J (2018) Routine uncertainty propagation for  
808 the marine carbon dioxide system. Marine Chemistry.  
809 <https://doi.org/10.1016/j.marchem.2018.10.006>

810 Pierrot D, Lewis E, Wallace DWR (2006) MS Excel Program Developed for CO<sub>2</sub> System  
811 Calculations, ORNL/CDIAC-105a. Carbon Dioxide Information Analysis Center. Oak  
812 Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, Tennessee.  
813 [https://doi.org/10.3334/CDIAC/otg.CO2SYS\\_XLS\\_CDIAC105a](https://doi.org/10.3334/CDIAC/otg.CO2SYS_XLS_CDIAC105a)

814 Raymond PA, Cole JJ (2001) Gas exchange in rivers and estuaries: Choosing a gas  
815 transfer velocity. Estuaries. 24:312–317

816 Sadat-Noori M, Maher DT, Santos IR (2016) Groundwater discharge as a source of  
817 dissolved carbon and greenhouse gases in a subtropical estuary. Estuaries and coasts.  
818 <https://doi.org/10.1007/s12237-015-0042-4>

819 Sarma VVSS, Kumar MD, Manerikar M (2001) Emission of carbon dioxide from a  
820 Tropical Estuarine System, Goa, India. Hydrology and Biogeochemical Cycles.  
821 <https://doi.org/10.1029/2000GL006114>

822 Sarmiento J, Gruber N (2006) Ocean Biogeochemical Dynamics Princeton Univ. Press,  
823 New Jersey.

824 Seeliger U, Odebrecht C (2010) O Estuário da Lagoa dos Patos: Um Século de  
825 Transformações. Rio Grande, FURG.

826 Sims RP, Bedington M, Schuster U et al (2021) Tidal mixing of estuarine and coastal  
827 waters in the Western English Channel controls spatial and temporal variability in  
828 seawater CO<sub>2</sub>, Biogeosciences Discuss. [preprint]. <https://doi.org/10.5194/bg-2021-166>

829 Sunda WG, Cai W (2012) Eutrophication induced CO<sub>2</sub>-acidification of subsurface  
830 coastal waters: interactive of temperature, salinity and atmospheric pCO<sub>2</sub>. Environmental  
831 Science & Technology. <https://doi.org/10.1021/es300626f>

832 Takahashi T, Sutherland SC, Chipman DW (2014) Climatological distributions of pH,  
833 pCO<sub>2</sub>, total CO<sub>2</sub>, alkalinity, and CaCO<sub>3</sub> saturation in the global surface ocean, and  
834 temporal changes at selected locations Marine Chemistry.  
835 <https://doi.org/10.1016/j.marchem.2014.06.004>

836 Takahashi T, Sutherland SC, Wanninkhof R et al (2009) Climatological mean and decadal  
837 change in surface ocean pCO<sub>2</sub>, and net sea-air CO<sub>2</sub> flux over the global oceans. Deep sea  
838 research part II: Topical studies in oceanography.  
839 <https://doi.org/10.1016/j.dsr2.2008.12.009>

840 Thoning KW, Crotwell AM, Mund JW (2021) Atmospheric carbon dioxide dry air mole  
841 fractions from continuous measurements at Mauna Loa, Hawaii, Barrow, Alaska,  
842 American Samoa and South Pole. 1973-2020, Version 2021-08-09 National Oceanic and  
843 Atmospheric Administration (NOAA), Global Monitoring Laboratory (GML), Boulder,  
844 Colorado, USA <https://doi.org/10.15138/yaf1-bk21> FTP path:  
845 [ftp://aftp.cmdl.noaa.gov/data/greenhouse\\_gases/co2/in-situ/surface/](ftp://aftp.cmdl.noaa.gov/data/greenhouse_gases/co2/in-situ/surface/)

846 Torres JR, Philomena LA (2013) Environmental Port Management: Conceptual Model  
847 Development and Use of Tools to Evaluate and Monitor Dredging Activities in the Port  
848 of Rio Grande, Brazil. African Journal of Environmental Economic and Management.  
849 1:22–27

850 Uppström LR (1974) Boron/chlorinity ratio of deep-sea water from the Pacific Ocean.  
851 Deep-Sea Res. 21:161–162

852 Van Dam BR, Crosswell JR, Paerl HW (2018) Flood-driven CO<sub>2</sub> emissions from  
853 adjacent North Carolina estuaries during Hurricane Joaquim (2015). Marine Chemistry.  
854 <https://doi.org/10.1016/j.marchem.2018.10.001>

855 Vaz AC, Möller OO, Almeida TL (2006) Análise quantitativa da descarga dos rios  
856 afluentes da Lagoa dos Patos. Atlântica. 28:13–23

857 Wallner-Kersanach M, Mirlean N, Baumgarten MGZ et al (2016) Temporal evolution of  
858 the contamination in the southern area of the Patos Lagoon estuary, RS, Brazil. Journal  
859 of Integrated Coastal Zone Management. <https://doi.org/10.5894/rgci596>

860 Wanninkhof R (2014) Relationship between wind speed and gas exchange over the ocean  
861 revisited. Limnology and oceanography. <https://doi.org/10.4319/lom.2014.12.351>

862 Weiss RF (1974) Carbon dioxide in water and seawater: the solubility of a non-ideal gas.  
863 Marine Chemistry. [https://doi.org/10.1016/0304-4203\(74\)90015-2](https://doi.org/10.1016/0304-4203(74)90015-2)

864 Weiss RF, Price BA (1980) Nitrous oxide solubility in water and seawater. *Marine*  
865 *Chemistry*. [https://doi.org/10.1016/0304-4203\(80\)90024-9](https://doi.org/10.1016/0304-4203(80)90024-9)

866 Yao H, Hu X (2017) Responses of carbonate system and CO<sub>2</sub> flux to extended drought  
867 and intense flooding in a semiarid subtropical estuary. *Limnology and Oceanography*.  
868 <https://doi.org/10.1002/lno.10646>

869 Yao H, McCutcheon MR, Staryk CJ, Hu X (2020) Hydrologic controls on CO<sub>2</sub> chemistry  
870 and flux in subtropical lagoonal estuaries of the northwestern Gulf of Mexico. *Limnology*  
871 *and Oceanography*. <https://doi.org/10.1002/lno.11394>

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### 873 **Statements & Declarations**

874 The authors declare that they have no known competing financial interests or personal  
875 relationships that could have appeared to influence the work reported in this paper.

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