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Effect of different preparation conditions on the properties of nano-hydroxyapatite/bamboo fiber composite membrane

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Effect of different preparation conditions on the properties of nano-hydroxyapatite/bamboo fiber composite membrane

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16 Abstract

3

17 A novel nano-hydroxyapatite/bamboo fiber (n-HA/BF) composite membrane was obtained by a simple casting technique. The membrane forming mechanism and the effects of different 18 forming membrane methods, drying methods and n-HA amounts on the properties of n-HA/BF 19 composite membrane were investigated by Fourier Transform infrared spectroscopy (FT-IR), 20 X-ray diffraction (XRD), scanning electron microscopy (SEM), contact angle, electromechanical 21 universal tester, in vitro soaking in simulated body fluid (SBF) and in vitro cell culture 22 experiment. The results demonstrated that the dispersity of n-HA nanoparticles in BF matix was 23 not affected by the prepartion condition, however, the morphologies of membrane was 24 determined by the different preparation conditions owing to different hydrogen bond shrinkage. 25 Moreover, the hydrophilicity of the composite membrane was improved under the condition of 26 the membrane formation in oven, freeze drying and high addition content of n-HA, and the 27 mechanical properties of composite membrane depended on n-HA content. In vitro soaking 28 behavior indicated that the degradability and bone-like apatite deposition could be controlled by 29 different preparation conditions. And the cell proliferation experiment showed that the n-HA/BF 30 composite membranes obtained under different preparation conditions were all non-toxic. The 31 above results indicated that the casting technique could be used to prepare n-HA/BF composite 32 membrane, and the properties of the composite membrane could be controlled by adopting 33 different preparation conditions, which would have a great promising as guide bone tissue 34 regeneration (GBR) membrane, and the study would provide a new application for BF in 35 biomedical field. 36 37

38 Keywords: Bamboo fiber; nano-hydroxyapatite; composite membrane; degradation

39 Introduction

40 Guided bone regeneration (GBR) membrane is commonly used in bone defect, which was placed

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43 on the bone defect area as barrier membrane to create a singular space, so as to prevent epithelial cells from growing into the defect, and permit osteoblasts to proliferate and new bone to form 44 (Jiang et al., 2015; Niu et al., 2021; Li et al., 2020; 2015). Ideally, GBR membrane should 45 possess appropriate mechanical properties, space-retention ability, biocompatibility and 46 biodegradability (Lee et al., 2016; Yu et al., 2020; Hoornaert et al., 2016). To obtain a satisfactory 47 GBR membrane, natural biodegradable polymers have become a research hotspot owing to 48 better biocompatibility and biodegradability, compared with the synthetic polymers (Prajatelistia 49 et al., 2021; Ma et al., 2019; Mora-Boza et al., 2020; Bierhalz and Moraes, 2017; Pappu et al., 50 2019; Gurunathan et al., 2015; Weng et al., 2020). 51 Bamboo fiber (BF) is a natural fiber extracted from bamboo, which has high strength, 52 biodegradability and low cost. Therefore, BF is usually used as a reinforcing agent for polymers 53 (Khalil et al., 2012; Liu et al., 2012; Phuong et al., 2019; Long et al., 2019; Zuo et al. 2019). In 54 our group, we have systematically studied and concluded that BF had remarkable reinforce effect 55 on the n-HA/poly (lactide-co-glycolide) (n-HA/PLGA) composite (Li et al., 2015; Jiang et al., 56 2017; 2018; 2019). Moreover, BF has been used to reinforce electrospunn membrane 57 (Chingakham et al., 2020; Cai et al., 2018). However, as we know, casting membrane has denser 58 structure, which would display better mechanical strength and slower degradation than 59 electrospinning membrane (Oksana et al., 2020). Moreover, casting membrane could be 60 prepared by a simple casting technology method. Therefore, we studied the reinforce effect of 61 62 carboxylated BF on chitosan-based casting membrane via ionic crosslinking (Tang et al., 2020). However, the preparation procession of carboxylated BF was too tedious. In addition, in the 63 above-mentioned study, BF was usually added in the primeval state fiber form as reinforcement, 64

which would be adverse for dispersion, so it would be expected to explore a novel BF-based
 polymer membrane by a simple and green processing.

BF could be dictrectly dissolved into homogeneous solution and casting membrane would be 67 formed, whileas, pure BF membrane lacks osteoconductivity, which would be detrimental to 68 guide bone tissue regeneration, while nano-hydroxyapatite (n-HA) has good osteogenic activity 69 becausee of its similarity with inorganic component of natural bone, so it was normally added 70 into polymeric matrix as nanofiller to endow polymers with better biological performance (Zhao 71 et al., 2018; Muhammad et al., 2017). In addition, in our previous study, we found BF could 72 replace other polymers to develop n-HA/BF nanocomposite by co-precipitation method, and it 73 had a promising to be used as bone materials (Ma et al., 2020). However, whether BF could be 74 replayed other polymers to obtain n-HA/BF composite membrane by solution blending method, 75 and what effects would different prepartion conditions produce on the properties of the n-HA/BF 76 membrane, including different forming membrane methods, drying methods and n-HA amounts, 77 and whether the n-HA/BF membrane would be used as GBR membrane, which were all not be 78 reported and it was worth exploring. 79

Based on these, in this work, we attempt to study the fabrication of the n-HA/BF composite membrane by casting method, and the effects of different preparation conditions including forming film methods (in air and in oven), different drying methods (in air, in oven, in freeze dryer) and different n-HA content (10%, 20%, 30% and 40%) on the properties of the n-HA/BF composite membranes were studied by Fourier Transform infrared spectroscopy (FT-IR), X-ray diffraction (XRD), scanning electron microscopy (SEM), contact angle, electromechanical universal teste. Moreover, the degradation behavior of the n-HA/BF composite membrane was

investigated by soaking in simulated body fluid (SBF). Finally, the cell culture experiment was
carried out. The main purpose of the work is to demonstrate the feasibility of the fabrication of
the n-HA/BF composite membrane by a simple casting method, so as to provide a new GBR
membrane by making full use of natural biomass resources.

91 **Experiment Section**

92 Materials

Bamboo fiber (BF) was provided by Zhejiang A&F University, whose size was Φ (0.03-0.2) mm

94 × (6-10) cm. Dimethylacetamide (DMAc, AR) and LiCl (AR) were purchased from Aladdin. Ca

95 (NO₃)₂.4H₂O(AR), Na₃PO₄.12H₂O(AR), P₂O₅, and NaOH (AR) were all from Tianjin Fengchuan

96 Chemical Reagent Technologies Co. Ltd.. Other agents were analytical grade.

97 Preparation of the n-HA/BF composite membranes

Bamboo fiber was dissolved in DMAc/LiCl system with the concentration of 1.3 wt%. A certain 98 amount of n-HA was dispersed in DMAc by ultrasonic treatment, which was slowly added into 99 bamboo fiber solution with the magnetic stirring, and the evenly dispersed mixture solution was 100 101 poured on the clean and dry glass plate, and the thickness of the membrane was adjusted by the glass rod with two copper rings. Then, the glass plate covered with n-HA/BF mixture solution 102 103 was put in air or in oven (70 °C) to form film, afterword, it was dried in air, in oven (70 °C) or freeze drying, thus, n-HA/BF composite membrane with 20% n-HA were obtained by six 104 105 methods, and the membranes were noted as air-air, air-oven, air-freeze drying, oven-air, 106 oven-oven and oven-freeze drying, respectively. Moreover, n-HA/BF composite membrane with different n-HA contents of 10 %, 20 %, 30 % and 40 % n-HA were prepared by air-air method in 107 108 the similar procedure.

109 Characterization of the composite membranes

110	The appearance of the membranes acquired by the two different preparation methods of air-oven		
111	and oven-oven were given as examples, which was taken by normal digital camera.		
112	Thermo Niclet 670 spectrometer was used to analysis the Fourier transformation infrare		
113	(FTIR) of samples, and the collected spectrum range was $600 \sim 4000 \text{ cm}^{-1}$.		
114	The phase analyses of samples were carried out by X-ray diffraction (XRD) (a Rigaku		
115	Corporation X-ray diffractometer) with Cu-K α radiation, the scanning speed of 5°/min at 40 kV		
116	and 45 mA, the range of $2\theta = 10 \sim 70^{\circ}$.		
117	The morphologies of samples treated with the gold sputtering were observed by scanning		
118	electron microscopic (SEM, S-520, Hitachi, Japan).		
119	The contact angles of samples were measured with rotating drop interfacial tensiometer		
120	(TX500TM, Kono, USA). The sample was put on the slide, and the distilled water was dropped		
121	onto the membrane surface by stop drop method, then the water drop on the sample was		
122	observed.		
123	The tensile strengths of samples with the size of 0.2 mm \times 4 mm \times 60 mm were measured by		
124	electromechanical universal testing machine (WDW-20, China) with the speed of 5 mm/min		
125	under 60 % relative humidity at room temperature, and the mean value was calculated based on		
126	the five parallel samples of each specimen.		
127	<i>In vitro</i> degradation of the composite membranes		
128	The degradation of n-HA/BF composite membrane in vitro was studied by soaking in SBF,		
129	whose ion concentration was very similar to that of human plasma, and it was obtained according		

130 to the following procedure, that is, NaHCO₃ (0.350 g), NaCl (7.996 g), KCl (0.224 g), K₂HPO₄ \cdot

131	$3H_2O$ (0.228 g), CaCl ₂ (0.278 g), Na ₂ SO4 (0.071 g), MgCl ₂ · $6H_2O$ (0.305 g) were dissolved in		
132	deionized water in order, and buffer with trimethylolmethylaminomethane (6.057 g) and		
133	hydrochloric acid to adjust the solution to physiological pH=7.40 at 37 °C (± 0.5 °C). The samples		
134	were taken out from SBF at 2, 4, 6 and 8 weeks, and filter paper was used to absorb surface		
135	residual washing water. The weight loss rate and water absorption were calculated as follows.		
136 137	Weight loss rate $/\% = (w_1 - w_3)/w_1 \times 100\%$		
138 139	Water absorption $/\% = (w_2 - w_3)/w_3 \times 100\%$		
140	The original weight of the sample was noted as W ₁ , and the wet weight and the dry weight		
141	was noted as W_2 and W_3 , respectively, after being entirely dried after soaking.		
142	In vitro cell biocompatibility of the composite membranes		
143	Bone mesenchymsal stem cells (BMSCs) were used to primarily assess in vitro cells viability,		
144	which was isolated according to the related literature (Hoseini et al., 2015; Ye et al., 2019). The		
145	samples with the thickness of 0.2 mm and diameter of 6.0 mm were cleaned with 75 % ethanol		
146	solution, sterilized under ultraviolet lamp. The treated samples were placed in a 96-well plate		
147	with the density of 4000 cells/well without disturbed in an incubator for 3 hours, then an		
148	additional 1 mL culture medium was added into each well.		
149	The cell proliferation was evaluated by MTT (3-(4,5-dimethylthiazol-2yl)-2,5-diphenyl-2H-		
150	tetrazolium bromide) assay (Priyadarshini et al., 2018; Shakeri et al., 2014; Hivechi et al., 2021).		
151	At designated time of 1, 2 and 3 days, the medium for the cell-seeded materials were discarded,		
152	and 100 μL MTT solution with 3 mg/mL was added, incubated at 37 $^{\circ}C$ in an air atmosphere		
153	containing 5 % CO ₂ for at least 4 hours, and 100 μL DMSO was added to dissolve the formazan		
154	crystals. Then the 200 μ L purple solution was absorbed and transferred into a new 96-well plate,		

and the optical density (OD) values of the solution were measured in microplate reader (Synergy
HTX, BIOTEK, USA) at 495 nm.

157 **Results and discussion**

158 Characterization of surface-modified n-HA

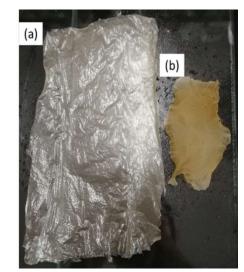
159 Appearance of the membranes

Fig. 1 displays the n-HA/BF composite membrane appearance, it can be seen that the membrane 160 161 formed in air had smaller area and thicker than that formed in oven, which was mainly caused from the different changes of the hydrogen bonds between the bamboo fiber molecules. In air, 162 the bamboo fiber was easy to absorb the water, which would make the hydrogen bond shrinkage 163 in the bamboo fiber, thus the bamboo fiber was gelled and DMAC/LiCl solution precipitated 164 from the edge, resulting in thicker film and smaller area. While in oven, DMAC solvent was easy 165 to volatilize and the hydrogen bond between the bamboo fiber molecules was destroyed, and the 166 bamboo fiber has been formed before shrinkage, resulting in thinner film and larger area. 167

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- Fig.1. Membrane appearances formed by different conditions. (a) in oven, (b) in air.
- 175 IR analysis



methods and different n-HA contents. The characteristic peaks of 2920 cm⁻¹ and 2846 cm⁻¹ 177 corresponded to the tensile vibration of C-H bond on methylene of BF existed in n-HA/BF 178 composite membrane. In addition, the characteristics peaks at 3567.2 cm⁻¹ attributed to the 179 tensile vibration of OH⁻ and the peaks at 1095 cm⁻¹, 604 cm⁻¹ and 565 cm⁻¹ of PO₄³ were related 180 to n-HA (Chesley et al., 2020). The peak position did not transfer obviously when n-HA was 181 182 added into BF, which indicated that n-HA was simply blended with BF without chemical change. Compared with the membrane formed in oven and in air, the peak intensity of the film formed in 183 oven was markedly weakened, for example, the sharp peak at 1043 cm⁻¹, which was caused by 184 less substance in the unit area for the thinner film formed in oven. Additionally, for the different 185 n-HA/BF composite membrane with different n-HA contents, there was no obvious difference 186 for the characteristics peak. 187

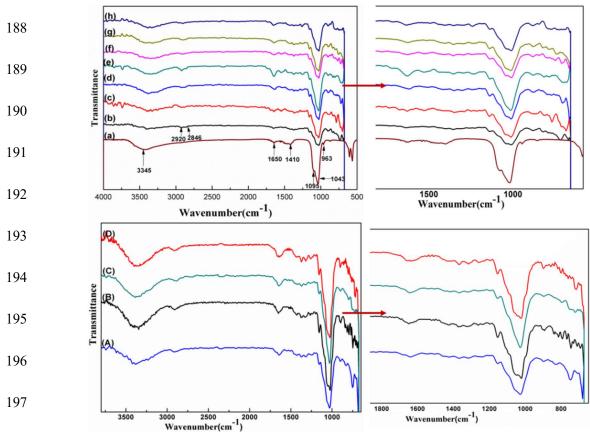


Fig.2. FTIR spectra of samples. (a)n-HA, (b)BF, (c)Air-Air, (d)Air-oven, (e)Air-freeze drying,
(f) Oven-air, (g)Oven-oven, (h)Oven-freeze drying, (A)10% n-HA/BF, (B)20% n-HA/BF, (C)
30% n-HA/BF, (D)40% n-HA/BF.

199 XRD investigation

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To further understand the phase structure of n-HA/BF membrane, XRD pattern is given in Fig.3. 200 Obviously, the amorphous peak at 20.5 ° was the peak of bamboo fiber (Guimaraes et al., 2015), 201 marked with "♦", and the peaks at 25 ° and 31 ° of n-HA were found, noted as "♣" (Ma et al., 202 2020). The crystallization peak position of n-HA in the composite membrane did not change, 203 204 indicating that the two components of n-HA and BF were only blended. Similarly, the crystallization peaks of n-HA in Fig.3 (f), (g), (h) were also obviously weaker than those in Fig.3 205 (c), (d) and (e). This was because the content of n-HA in the membrane formed in oven was less 206 than that in the membrane formed in air, which led to weaker crystal peak in the spectrum. With 207 the increase of n-HA content, the characteristic peak and its crystallinity of n-HA in different 208 membrane increased, which also confirmed that n-HA and BF were only blended without 209 chemical reaction. 210

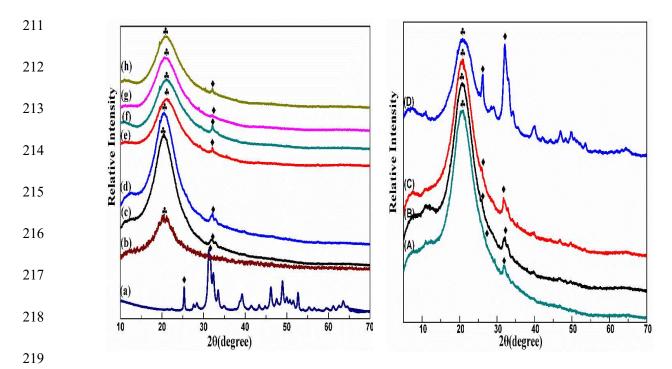
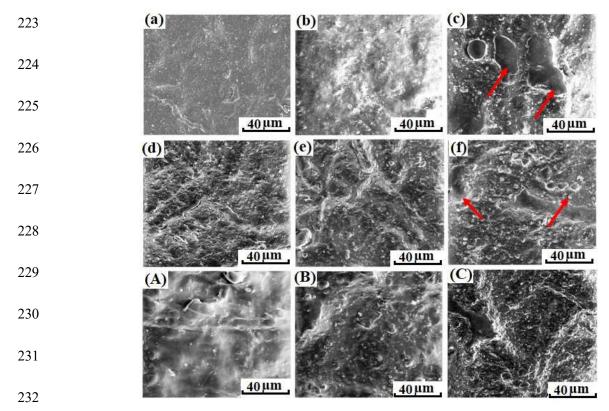


Fig. 3 XRD of samples. (a) n-HA, (b)BF, (c) Air-Air, (d) Air-oven, (e) Air-freeze drying, (f) Oven-air, (g) Oven-oven, (h) Oven-freeze drying, (A) 10% n-HA/BF, (B) 20% n-HA/BF, (C) 30% n-HA/BF, (D) 40% n-HA/BF.

222 SEM observation



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Fig. 4 SEM of samples. (a)Air-Air, (b)Air-oven, (c)Air-freeze drying, (d)Oven-air, (e)Oven-oven, (f)Oven-freeze drying, (A)10%HA/BF, (B) 30%HA/BF, (C)40%HA/BF.

Fig. 4 shows the morphologies of n-HA/BF composite membranes achieved by different 234 235 methods. According to SEM micrographs, it can be seen that white particles existed in the 236 composite membranes, which proved the existence of n-HA. However, there was subtle difference for the dispersion of n-HA particles in the membrane, and the n-HA was relatively 237 238 more uniform in oven (Fig.4(d), (e), (f)) than in air (Fig.4(a), (b), (c)), which was caused by the different changes of hydrogen bond between molecules during the forming membrane of bamboo 239 fiber. In oven, the solvent of DMAC was evaporated at high temperature, and the fiber 240 morphology had been fixed before the hydrogen bond shrinking between bamboo fiber 241 molecules, which would bring out better dispersion of n-HA. However, for the three different dry 242 methods, the surfaces were compact without porous structures, and there was no obvious 243 244 difference for the surface in air and in oven, which indicated that drying method in air or in oven

245 had little effect on the membrane morphology after bamboo fiber molding, and the compact structure could effectively prevent the invasion of connective tissue. While for the freeze-drying 246 method, there was some closed pores on the surface of freeze-dried membrane (Fig.4(c), (f)), 247 which was originated from the pores left by the sublimation of water molecules, and the porous 248 structure was conducive for cell adhesion for guided bone regeneration. Additionally, for the 249 composite membranes with different n-HA contents, the white particles gradually increased on 250 the surface of the membrane, but there was no obvious agglomeration phenomenon, and there 251 was no cavity between n-HA and BF, which indicated that the n-HA/BF composite membrane 252 had good compositional compatibility of hydrophilic n-HA and BF. 253

254 Contact angle measurement

To further make clear the hydrophilicity of the n-HA/BF composite membrane, the contact 255 angle of the n-HA/BF composite membranes were tested with rotating drop interfacial 256 tensiometer by dropping distilled water onto the liquid surface, and the results are shown in 257 Fig.5. As expected, the contact angle of all n-HA/BF casting membranes were less than 90°, 258 which proved that n-HA/BF casting membranes were hydrophilic membrane. Comparing with 259 the membranes obtained by different methods, it could be found that the same forming 260 membrane method had little effect on the contact angle (Fig.5(a) \approx 5(b), 5(d) \approx 5(e)), while the 261 freeze-dried membrane possessed lower contact angle because of the rough membrane surface 262 with the porous structure (Fig. 5(c) and 5(f)), and the stronger hydrophilicity would be more 263 264 useful for cell adhesion and proliferation (Dhinasekaran et al., 2021). However, for the same dried method, the membrane formed in oven had lower contact angle than that of the membrane 265 formed in air (Fig.5(d) \leq Fig.5(a), Fig.5(e) \leq Fig.5(b)), and the reason was that the hydrogen 266

bonds between the bamboo fibers molecules was destroyed and the surface tension was reduced when the membrane was formed in oven. In addition, for the n-HA/BF composite membranes with different n-HA contents, the contact angle gradually decreased with the increasing of n-HA, which suggested that the higher n-HA content brought about better hydrophilicity of the membrane.

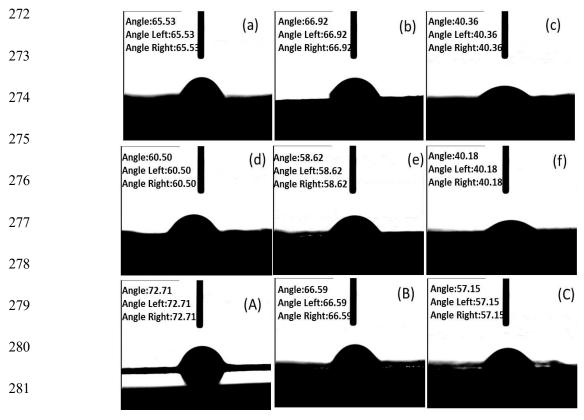


Fig.5. Contact angle of the n-HA/BF composite membranes obtained by different methods. (a)
 Air-Air, (b) Air-oven, (c) Air-freeze drying, (d) Oven-air, (e) Oven-oven, (f) Oven-freeze
 drying, (A) 10%-n-HA/BF, (B) 30% n-HA/BF, (C) 40% n-HA/BF.

Fig. 6 is the tensile strengths of all the n-HA/BF composite membranes. We found that membrane formed in air had higher mechanical strength than that in oven. Comparing to the membrane formed in air (Fig.6 (a), (b), (c)), the tensile strength of membrane dried by freeze-drying was the worst, and the tensile strength of the membrane dried in air was improved

²⁸⁴ Tensile strength test

289 by 69.64%, and dried in oven was increased by 51.7% than dried by freeze-drying, respectively. Likely, for the three membranes formed in oven (Fig.6(d), (e) and (f)), the tensile strengths of the 290 membranes dried in air and in oven were improved by 23.89% and 5%, respectively. The reason 291 was that the freeze-drying membrane was porous structure, and the forming membrane or dry 292 membrane in oven caused the destruction of intermolecular hydrogen bond during molding, 293 resulting in thinner membrane and larger area, so as to possessed lower tensile strength. For the 294 n-HA/BF composite membranes with different n-HA contents, the tensile strength increased at 295 first but decreased with the increasing of n-HA content, and the 20% n-HA/BF composite 296 membrane represented the highest tensile strength, which was accord with the principle of 297 nanoparticles filler reinforce polymer (Yadav et al., 2020), and the mechanical strength could 298 meet the application requirement of GBR membrane (Castro et al., 2018). 299

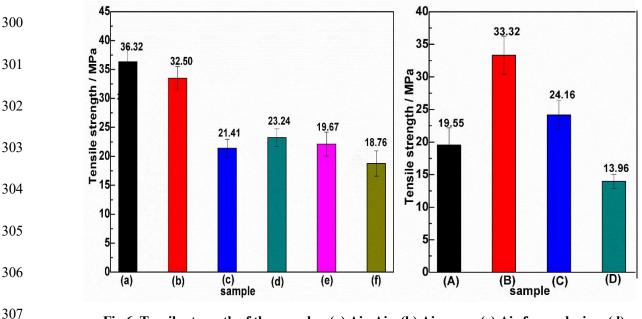


Fig.6. Tensile strength of the samples. (a) Air-Air, (b) Air-oven, (c) Air freeze drying, (d) Oven-air, (e) Oven-oven, (f) Oven-freeze drying, (A) 10% n-HA/BF, (B) 20% n-HA/BF, (C) 30% n-HA/BF, (D) 40% n-HA/BF.

309 In vitro degradation and cell culture

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310 Weight loss of composite membrane after degradation

311 Fig.7 gives the weight loss rate of the n-HA/BF composite membranes. It can be seen that the weight loss rate of the composite membrane first decreased and then increased within 8 weeks. 312 During the first 4 weeks, the weight loss rate was negative continuously, which indicated that the 313 mass of composite membrane did not decrease but increased after immersion, meaning that the 314 amount of apatite deposited on the surface of the membrane was greater than the mass of 315 degradation, so that the total mass of composite membrane was greater than that before 316 317 immersion. Similarly, the weight loss rate of 4-8 weeks was negative, but exhibited an upward trend, suggesting that the membrane had more visible degradation trend. Comparing to the 318 different composite membranes, the weight loss rate of the membrane obtained by oven-freeze 319 drying method changed the most among the membrane different methods, which indicated that 320 bone like apatite adhered the most owing to the porous structure, and the more apatite deposition 321 would have better biological activity. Moreover, for the composite membranes with different 322 n-HA contents, the higher the content of n-HA, the greater the negative value of weight loss rate, 323 which revealed that the higher content of n-HA was more favorable for the bone-like apatite 324 deposition (Kumar et al., 2014). 325

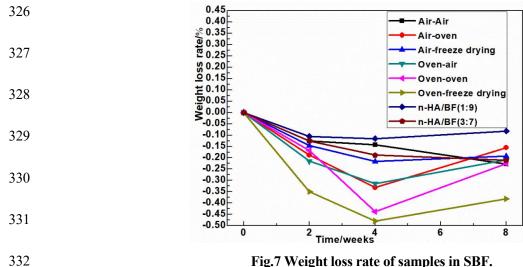


Fig.7 Weight loss rate of samples in SBF.

333 Water absorption of composite membrane after degradation

Fig.8 is the water absorption of n-HA/BF composite membrane. The results indicated that the 334 water absorption of n-HA/BF composite membrane had a similar trend during the immersion 335 process, that is, the water absorption increased slightly at the initial degradation, and then 336 decreased a little from 4 weeks to 8 weeks, which was caused by the bone-like apatite deposition 337 with the extension of soaking time. Moreover, the freeze-dried membrane had the lowest water 338 absorption, and the main reason was that the porous structure had the fastest degradation, which 339 would produce much micropore and bring more apatite deposition, and it could be proved by the 340 weight loss result. Similarly, for the composite membrane with different n-HA contents, the 341 higher BF content, the higher water absorption had, which was related to the water absorption of 342 BF. In a word, the water absorption results indicated that the n-HA/BF composite membrane had 343 good degradation and water absorption performance. 344

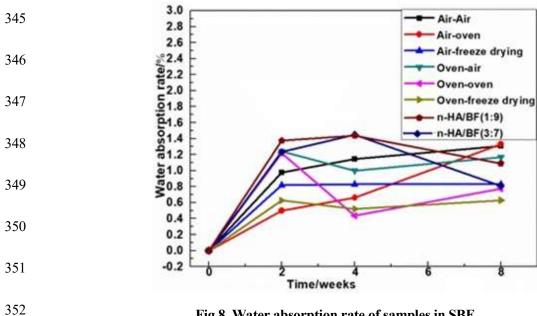


Fig.8. Water absorption rate of samples in SBF.

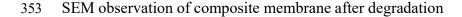
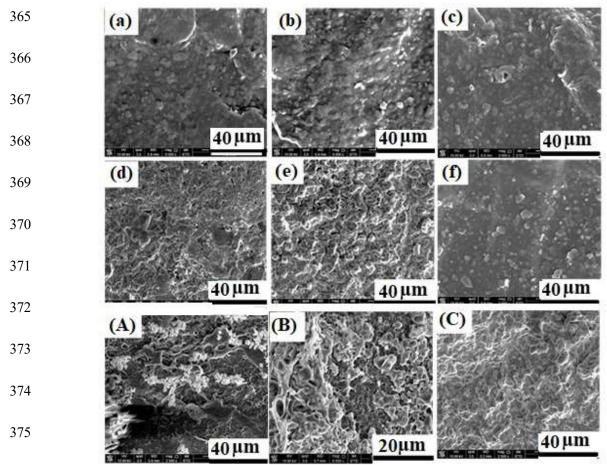


Fig. 9 shows the surface morphology of n-HA/BF composite membrane after soaking in SBF for 354

355 8 weeks. It can be found that new bone like apatite was disposited on the surface of the n-HA/BF composite membrane. However, comparing to the different composite membranes, the composite 356 membrane formed in air had less bone like apatite (seen Fig.9 (a), (b), (c)) than the membrane 357 formed in oven (seen Fig.9 (d), (e), (f)). The reason was that the hydrogen bond shrank between 358 bamboo fiber molecules, and n-HA particles were wrapped in the inner part of the membrane 359 360 when the membrane was formed in air. While the composite membrane was formed in the oven, 361 the intermolecular hydrogen bond was broken, and n-HA particles were dispersed on the surface 362 of the membrane, which was more conducive to the deposition of bone-like apatite on the surface of the membrane, and the bone-like apatite would help to improve interface between the 363 tissue-implant and its surrounding tissues (Zhu et al., 2020). 364



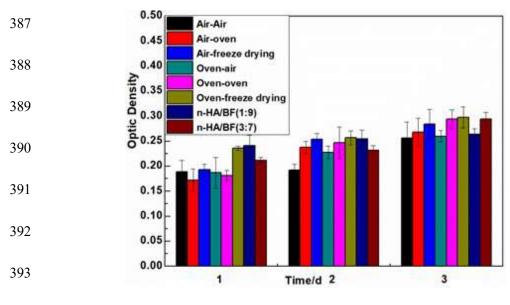
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Fig. 9. SEM micrographs of samples after soaking for 8 weeks. (a) Air-Air, (b) Air-oven, (c)Air-freeze drying, (d) Oven-air, (e) Oven-oven, (f) Oven-freeze drying, (A)10% n-HA/BF, (B) 30% n-HA/BF, (C)40% n-HA/BF.

For the composite membrane with different n-HA contents of 10%, 30% and 40% n-HA (seen Fig.9(A), (B), (C)), respectively. From the 10% n-HA/BF composite membrane, it can be seen that there were pores on the surface of bamboo fiber (Fig.9(A)), which indicated that bamboo fiber could be degraded during soaking. With the increase of n-HA content, the more bone-like apatite was deposited, which further demonstrated that the n-HA in composite membrane could induce bone-like apatite deposition, and the results were consistent with the previous analysis.

384 MTT test of cell culture

The cell proliferation results of BMSC cultured on the surface of n-HA/BF composite membrane for 1, 2 and 3 days are given in Fig. 10.



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Fig. 10. MTT value of cell culture on sample surface.

It can be seen that the OD value of each sample increased significantly with the extension of culture time, indicating that cells could normally grow and proliferate on different membrane surfaces. In addition, the OD values of freeze-dried composite membrane were significantly higher than those of the membrane dried in air or in oven, which further indicated that the porous structure of freeze-dried composite membrane was more conducive for cell proliferation and possessed better biocompatibility (Ai et al., 2021). Comparing to the n-HA/BF membranes with 10%, 20% and 30%, the OD value enhanced significantly with the increase of n-HA content and the extension of culture time. The results illustrated that the cells proliferated rapidly on the surface of the membrane, suggesting that the n-HA/BF composite membrane had good biological properties (Bee et al., 2019). This was consistent with the previous *in vitro* immersion analysis.

405 **Conclusions**

In this study, the n-HA/BF composite membrane was successfully prepared by a simple casting 406 technology, and the effects of different forming membrane, drying methods, and different n-HA 407 408 contents on the properties of the composite membranes were studied. The results confirmed that there was no chemical bonding between n-HA and BF components, however, the membrane 409 formed in air was thicker than that in oven, because the hydrogen bond shrinkage made the BF 410 411 gel and DMAC/LiCl solution precipitated from the edge in air. Thus, the tensile strength of the membrane formed in air were higher than that of in oven, especially, the 20 % n-HA/BF 412 composite membrane was the highest, but the tensile strength of freeze-dried composite 413 membrane was the worst because of the porous structure. The contact angle test of the composite 414 membrane confirmed that the composite membrane was hydrophilic, and the freeze-dried 415 membrane exhibited better hydrophilicity owing to the rough surface and the small surface 416 417 tension. SBF soaking results indicated that n-HA/BF composite membrane displayed different 418 degradability behaviours, and the higher n-HA content in composite membrane possessed better bone-like apatite deposition. The cell proliferation results proved that the composite membranes 419 had no cytotoxicity. This study would provide a new way for developing a novel GBR membrane 420

421 based on the utilization of natural BF.

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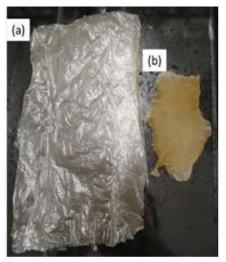
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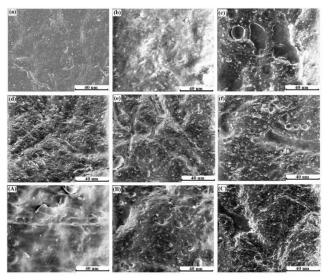
Graphical Abstract

568 569

In this manuscript, nano-hydroxyapatite/BF (n-HA/BF) composite membrane was prepared by 570 casting method. and the membrane forming mechanism and the effects of different forming 571 572 membrane methods, drying methods and n-HA amounts on the corresponding n-HA/BF membrane were investigated. Results demonstrated that the morphologies of membrane was 573 determined by the different preparation conditions owing to different hydrogen bond shrinkage. 574 Moreover, the hydrophilicity, the mechanical properties, the degradability and bone-like apatite 575 deposition could be controlled by different preparation conditions, and all the n-HA/BF 576 composite membranes were all non-toxic. The above results indicated that the n-HA/BF 577 composite membrane have a great promising as guide bone tissue regeneration (GBR) 578 membrane, which would provide a new application for BF in biomedical field. 579



Membrane appearances formed by different conditions. (a) in oven, (b) in air.



SEM micrographs of samples. (a)Air-Air, (b)Air-oven, (c)Air-freeze drying, (d)Oven-air, (e)Oven-oven, (f)Oven-freeze drying, (A)10%HA/BF, (B) 30%HA/BF, (C)40%HA/BF.