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In Situ Observation of the Atomic-scale Hightemperature Phase Transition Enhancement Mechanism at the ODS-W Interface

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Rare earth oxide doping Metal is a common strengthening method for metal matrices. Recent studies show that the ODS-W alloys exhibit a more superior mechanical properties at higher temperatures, but the underlying strengthening mechanism is obscure due to the experimental technical limitations. This study performed in situ tensile deformations at temperatures up to 1273 K inside a transmission electron microscope with atomic-scale resolution. The results uncovered that the crack source caused by tension is at the W grain boundary, rather than the expected weak interface between W and oxide. We showed that the W near the ODS-W interface occurs BCC-FCC phase transition and breaks away from the W matrix. By combining molecular dynamics simulation with firstprinciple theory calculations, we confirmed that the WFCC matrix after the BCC-FCC phase transition possesses excellent adhesion work, which remarkably enhances the W and the oxide bonding force. Based on these observations, we proposed a high-temperature phase transformation strengthening method, which provides essential guidance for oxide-strengthen alloys at high temperatures and adds new ideas to the second-phase strengthening mechanism.

he strengthening mechanisms of metal materials can roughly divided into two types. One is to increase the alloying atoms' interatomic bonding force to increase their academic strength to obtain a perfect crystal without defects. The other is to introduce flaws in the crystal, such as dislocations, point defects, heterogeneous atoms, grain boundaries, and highly dispersed particles or heterogeneous structures (e.g., segregation). The movement of dislocations is hindered by these defects, which results in a significant improvement in strength. The main strengthening methods include solid solution strengthening¹⁻³, second-phase particle strengthening^{4,5}, fine grain strengthening⁶⁻⁸, and work hardening⁹⁻¹¹. As a standard strengthening method, the second-phase particle strengthening is widely used in structural materials like tungsten and molybdenum. For tungsten, the doping of the second phase can improve its mechanical properties and ductile-brittle transition temperature. Because the oxide particles have a high melting point and stable physical and chemical properties, they usually doped in the W matrix as the second phase; ODS-W usually alloy has good mechanical properties, thermal shock resistance, and radiation damage resistance¹²⁻¹⁴. Therefore, doping nano-oxide particles in the W matrix is the primary strengthening means of W as a nuclear fusion-oriented plasmonic material.

Explanations for the strengthening mechanism of ODS alloys are twofold. One is that the second-phase particles inhibit the growth of W matrix grains during the sintering process, and they refine the grain size to achieve the effect of fine-grain strengthening for improving the toughness of pure W^{15-16} . The other is that the pinning

effect of uniformly dispersed nanoparticles on dislocation slip and GB migration significantly enhances the strength at high temperature and thermal stability of W alloy¹⁷. Given that oxide particles are mostly located at grain boundaries, the increase in heterogeneous phases often acts as a source of cracks. Thus, many researchers currently attempt to dissolve particles completely in the crystal, the W–X alloy that is dissolved in the crystal improves the toughness of pure W at room temperature; however, the ODS alloy exhibits an excellent tensile strength at high temperatures¹⁸, and the author could not clearly explain this phenomenon; the superb performance at high temperatures is believed to be due to fine grain strengthening. Owing to the exponential increase of the tensile strength of ODS alloy at high temperatures, the strengthening mechanism of ODS alloy should include fine-grain strengthening and traditional second-phase strengthening, and other strengthening methods.

The microscopic mechanism of metal materials under stress becomes more apparent with the development of electron microscopy techniques, such as the behavior of phase transitions under pressure. The phase transition under the stress of body-centered cubic (BCC) structure metals has been observed by TEM and in situ TEM, such as Nb¹⁹, Mo²⁰, and Ta²¹. W with the BCC structure is stable at room temperature. The DAC experiment has determined that the W material will not induce a room-temperature phase change even if the stress of up to 423 GPa is applied to it ²². Some interesting phenomena can be produced in small-size W nanowires, such as twins and inverse twins²³⁻²⁵, but they cannot meet the phase change conditions and form a new phase. Compared with the phase transition behavior of other BCC structural metals at room temperature, W

exhibits a strong temperature dependence. The phase transition process near the crack tip under high-temperature tensile conditions is captured and reported by in situ TEM²⁶, and the corresponding molecular dynamics (MD) simulations also confirmed that a phase transition occurs in W at high temperature²⁷. Therefore, we predicted that the ODS-W alloy has a phase transition under high-temperature tensile stress, and this phase transition effect increases the tensile strength of the alloy. To observe the microscopic process of ODS-W alloy tensile fracture, A high-temperature real-time mechanical testing and a field emission high-resolution transmission electron microscopy (HRTEM) detection were applied to observe the microscopic process of ODS-W alloy tensile fracture.

This study successfully conducted high-temperature in situ tensile experiments on ODS alloys. Advanced technology from Beijing Bestron Science and Technology and Beijing University of Technology was used²⁸, which allows in situ stretch deformation at up to 1273 K in a transmission electron microscope with atomic resolution. Using this unique technology, we found for the first time that the oxide particles at the grain boundary interacted with the W matrix at high temperatures, and the W matrix was torn off at the oxide grain boundary; the phase transition of the W matrix induced by the oxide interface under high-temperature tensile conditions was also revealed. Using MD combined with first-principle calculations, we concluded that the W matrix of the FCC phase has a more robust interface binding ability than the W matrix of the BCC phase. In combination with the action mechanism of the second phase observed in situ during high-temperature deformation of the matrix, the high-temperature phase transformation

strengthening theory of the ODS-W matrix was proposed. Finally, the reason for the improved strength and toughness of the alloy at high temperatures was explained.

Results

In situ ODS-W TEM sample preparation and high-temperature tensile characterization system: Nano-oxide dispersion strengthened W-Y₂O₃ alloy with uniform second-phase particles was successfully prepared by wet chemical combined with spark plasma sintering (SPS). Figs. 1(a–c) shows the process of preparing samples through the focused ion beam (FIB) technology. FIB was first used to extract rectangular block material from W-Y₂O₃ block through the FIB milling method, and then the sample was loaded onto the heating chip. Using Pt nanocomposite, the rectangular sample was welded to the tensile beam by ion beam-induced deposition, clearly showing dense W matrix particles and dispersed Y₂O₃ particles, as shown in Fig. 1(b). The ion beam of 30 kV and 80 pA (Fischione Model 1040) was used to reduce the width of the sample from 2 µm to 100 nm. The 30 kV and 1.1 pA ion beam was used to mill the sample into a dog-bone shape. The width of the final stretched sample is 700 nm, and the sample contains two Y₂O₃ particles, as shown in Fig. 1(c). The stretching process is shown in Fig. 1(d). Applying pressure on both sides of the beam causes the sample to be subjected to indirect load and deform, and the temperature loading accuracy controlled by voltage reaches $\pm 0.1 \text{ K}$; at a heating rate of 20 K/min, the sample can be stably heated to 1273 K²⁹. Fig. 1(e)shows the schematic of a highresolution transmission electron microscope for high-temperature in situ analysis²⁶.



Fig. 1. Preparation and characterization system of in situ W–Y₂O₃TEM sample (a) FIB sample preparation diagram, (b) sample section after welding, (c) shaped in situ W–Y₂O₃TEM tensile sample, (d) working principle diagram of in situ high-temperature stretch rod, (e) high-resolution transmission electron microscope for high-temperature in situ analysis

In situ TEM observation on ODS-W sample of tensile deformation: Fig. 2 shows the in situ TEM observation of W–Y₂O₃ sample of tensile deformation. Fig. 2(a) shows the TEM image of a dog bone-shaped sample installed on a high-temperature mechanical test device. The measurement section size of the sample is 700 nm×600 nm×100 nm, and the sample contains two Y₂O₃ particles, which are located at the grain boundary and the three-fork grain boundary, respectively, as shown in Fig. 2(b). The red particles are Y₂O₃, which are marked as Y₁ and Y₂. Gray line indicates W crystal boundary, and W matrices in the three areas are marked as W₁, W₂, and W₃. At a heating rate of 20 K/min, the sample was first heated to 1273 K and then kept for 10 min to ensure uniform and stable overall temperature. The inset of Fig. 2(a) shows the electron diffraction (SAED) pattern of selected area of the sample. The selected area is W1, which is the focus of our subsequent analysis. Figs. 2(c-k) show the sample images at different stages of stretching deformation. A stable tensile force was applied to the sample continuously such that the tensile stress actually acts on the sample, as shown in Fig. 2(c). When the boundary between W₁ and W₂ shows stress concentration area, it is considered a possible source of cracks and defined as 0 s. As the tensile stress increases, the dislocation density continues to increase, and the stress on the crack source gradually increases, as shown in Figs. 2(d-e). At 5 s, the sample cracks along the grain boundary of W1 and W3, and the local stress relaxes immediately, as shown in Fig. 2(f). In Fig. 2(g), the yellow dotted circle marks the obvious malposition at the two grain boundaries after the fracture. Then, tension was applied. Figs. 2(h-j) show the sample image after 11, 54, and 73 s of stretching, and the crack gradually expands and slowly extends to Y₂. We will follow up on the in situ TEM analysis of the fracture process at high magnification at Y₂. The sample was further stretched to complete fracture. Fig. 2(k) is the final fracture morphology of the sample, which shows that the sample fractures along the W₁ grain boundary through Y₁ and Y₂. The in situ stretching process reveals no fracture source on the interface of Y₂O₃ and W matrix. The stress concentration at the grain boundary of the W matrix causes fracture, which finally passes through two Y₂O₃ particles along the grain boundary.



Fig. 2. In situ TEM observation of $W-Y_2O_3$ sample of tensile deformation (a) TEM image of dog bone-shaped $W-Y_2O_3$ sample made by FIB, (b) schematic of $W-Y_2O_3$ sample, (c–i) continuous TEM image at different stages of tensile deformation of the sample at 1273 K, (k) final fracture morphology of $W-Y_2O_3$ sample

TEM observation of oxide interface after the fracturing of ODS-W sample: Fig. 3 shows the TEM image of the Y₂O₃ interface after fracturing of W–Y₂O₃ sample. Given that the crack source is not at the interface between W matrix and Y₂O₃, a specific abnormal phenomenon is observed at the two interfaces, leading to the enhancement in the binding force of the heterogeneous interface. As shown in Fig. 3(a), two Y₂O₃(Y₁, Y₂) particles were analyzed by HRTEM when the sample was fractured. As observed, the W matrix at the interface of Y₁ and Y₂ is torn. A few W matrices are torn at Y₁(Fig. 3b), and they hang like a drop of water on the interface of Y₁. Multiple drop-like W particles are also found on the surface of Y₂(Fig. 3c). HRTEM analysis was conducted

for the torn W at Y₁ (Fig. 3d). Fast Fourier transform (FFT) was performed on W in the white box, and the resulting image is e1. Considering that the torn W is relatively small (5 nm) and the corresponding diffraction spot has only one point, the obtained $d^*=4.16$ nm^{-1} (Fig. e3), and the d^{*} value does not match the BCC structure W(JCPDS#04-0806) but matches perfectly with the FCC structure W³⁰. The inverse FFT corresponding to the diffraction spot is shown in Fig. e2, where the bright area encircled by a yellow dotted line corresponds to the location of torn W in the white box of Fig. 3(d). This point is indeed the diffraction spots corresponding to torn down W. Thus, the substance on the Y₁ interface is not Y₂O₃ or BCC structure W, but it is most likely the FCC structure W. Similarly, HRTEM analysis was conducted at the Y₂ interface. Two areas with torn W matrices were selected, as shown in Fig. 3(f), and FFT was performed on the white box (Fig. 3g). Given that the Y₂O₃ area is large, the diffraction spot is prominent, and the crystal ribbon axis corresponding to the white parallelogram is $[00\overline{4}]$. The two-crystal plane is the orientation relationship of (200) $(0\overline{2}0)$ Y₂O₃(JCPDS#43-0661). The inverse Fourier of the diffraction spot penetrated by the red line is shown in Fig. 3h1, where the bright area enclosed by the red dotted line corresponds to the red dotted frame position in Fig. 3(f). It proves that the diffraction spot corresponds to the W matrix torn down in the upper half, and $d^*=4.18 \text{ nm}^{-1}$ (Fig. 3h₂). Similarly, the diffraction spot penetrated by the green line corresponds to the W matrix torn down in the lower half, and $d^*=4.19 \text{ nm}^{-1}$ (Fig. 3l₂). The two W matrices at Y₂ interface do not match the BCC structure W but fit perfectly with the FCC structure W, and the d^{*} value corresponds to the $(11\overline{1})$ surface of FCC structure W in Fig. 4.

Combining the HRTEM analysis results of W torn at Y₁–Y₂ interface shows that W leaving the matrix presents an FCC phase. This result is discovered for the first time.



Fig. 3. TEM observation at Y_2O_3 interface after fracture of W-Y₂O₃ sample (a) TEM image of W-Y₂O₃ sample after fracture, (b-c) TEM and partially enlarged view of Y₁-Y₂ interface, (d) HRTEM of Y₁ interface, (e₁-e₃) d^{*} value of the FFT, inverse Fourier transform, and diffraction spots in the white box of Fig. d, (f) HRTEM of Y₂ interface, (g) FFT in the white box of Fig. f, (h₁-h₂) Inverse Fourier transform corresponding to the red diffraction spot and d^{*}value in Fig. g, (l₁-l₂) Inverse Fourier transform corresponding to the green diffraction spot and d^{*} value in Fig. g

BCC–FCC phase transition near the interface of oxide interface and W matrices under tensile load at 1273 K: Fig. 4 shows the TEM image with atomic resolution. Fig. 4(a) is the overall topography of Y_2 after a fracture. Y_2 and the matrix fracture process are shown in the supplementary video 1, which shows that the matrix is deformed by the tensile stress given by the interface when the W–Y₂O₃ interface is separated. The white box area (W₁) was used for analysis, as shown in Fig. 4(b). This area includes the red box area W_{BCC} and the blue box area W_{FCC}. The high-resolution image of the red box area is Fig. 4(c), the illustration is the FFT pattern in the area, the d spacings of the two basic lattice vectors are $d_1=0.2277 \text{ nm}(011) \text{ BCC}$ and $d_2=0.1573$ nm(200) BCC, the basic lattice vector distance is 90°, which conforms to the orientation relationship of crystal ribbon axis $[01\overline{1}]_{BCC}$. The high-resolution image of the blue box closer to the strain area is Fig. 4(d), the spacings of the two basic lattice vectors in the FFT pattern are $d_1=0.1461 \text{ nm}(220)_{FCC}$, and $d_2=0.2417 \text{ nm}(11\overline{1})_{FCC}$. The basic lattice vector distance is 90°, which conforms to the orientation relationship of crystal ribbon axis[112]_{FCC}. According to the comparison of the two sets of diffraction spots, the phase transition process has noticeable lattice distortion, and the phase transition is the primary stress release method in the deformation process. The selected area of Fig. 4(e) is the location of the orange box in Fig. 4(b), and the parallelogram corresponding to the fast Fourier diagram is shown in the white box in the illustration, which is located between W_{bcc} and W_{FCC}. D spacings of the two basic lattice vectors are 0.1613 and 0.2315 nm, and the basic lattice vector distance is 86°. The structure cannot be indexed to any area axis of W_{BCC} or W_{FCC}, and thus, it is regarded as an intermediate state. Given that the stress ends at this time and the FCC phase in the W matrix is unstable, this intermediate state is likely to be an instantaneous capture of the FCC \rightarrow BCC phase transition process. The BCC \rightarrow FCC phase transition caused by deformation can only occur in the environment around the interface and not in the main body of the matrix. The crystallographic mechanism transformation of BCC→FCC is considered to be the lattice shear in $[01\overline{1}]_{BCC}$, as shown in Fig. 4(f). According to the position of the middle phase in Fig. 4(e), the crystal lattice is distorted clockwise, and finally, the phase change of BCC \rightarrow FCC is completed. The lattice correspondence between the two phases is defined as(011)_{BCC}//(112)_{FCC} and (011)_{BCC}//(220)_{FCC}, and this phase transition process corresponds to the Pitsch model³¹. After phase transition, (111)_{FCC} surface (d2=0.2417 nm) in [112]_{FCC} corresponds to the diffraction spot value (d1=0.2386 nm, d2=0.2392 nm) of the torn W in Fig. 3(f), which further proves that the torn W is in FCC phase, and the torn part is the phase transited part of W1. Certain stress exists on the Y2O3–W interface during the high-temperature tensile fracture process. Unlike previous W–Y2O3 semi-coherent interface that broke directly at room temperature³², phase transition at the interface at high temperature is an effective way to relieve stress. The bonding force at W–Y2O3 interface can lead to BCC \rightarrow FCC phase transition of W matrix, which provides strong support for tearing off the FCC phase W matrix at the Y2O3 interface.



Fig. 4. HRTEM observation of BCC–FCC phase transition near the interface under tensile load at 1273 K (a) TEM bright-field image of the sample after tensile fracture of W–Y₂O₃ at 1273 K, (b) HRTEM after fracture of W₁ and Y₂, (c) HRTEM of the red frame in Fig. b, the illustration shows the FFT pattern in the red frame, (d) HRTEM of the blue frame in Fig. b, the illustration shows the FFT pattern in the blue

frame, (e) HRTEM of the orange frame in Fig. b, the illustration shows the FFT pattern in the orange frame, (f) crystallographic diagram of BCC–FCC phase transition

MD stretching simulation at room temperature and 1273 K: We used MD to simulate the tensile phase transition process of the W1 matrix to determine further the effect of temperature on the phase transition process. As shown in Fig. 5(a), the model sample contains approximately 108000 tungsten atoms, and the coordinates are x axis[211], y axis[01 $\overline{1}$], and z[$\overline{1}$ 11]axis. Tension was applied along the x-axis [211] at a constant strain rate until it was broken. The model corresponds to $[01\overline{1}]_{BCC}$ in Fig. 4(c). Common neighbor analysis command was used in Open Visualization Tool (OVITO), where the balls represent the W atoms in the BCC phase, green balls indicate the W atoms in the FCC phase, red and white indicate W atoms in the HCP phase, and other structures. The model was stretched after applying room temperature (293 K) and 1273 K. The model profile and overall FCC phase atom diagram are shown in Figs. 5(b-c). At room temperature, only a tiny amount of FCC phase atoms are generated at room temperature when the strain rate reaches 12% (Fig. 5b). When the strain rate goes 14%, FCC phase atoms disappear, indicating nearly no phase transition at room temperature. Studies have shown that plastic deformation under the temperature of BCC metal DBTT depends on the proliferation and cross-slip of dislocations; the mobility of screw dislocations also increases with temperature, which makes BCC metal plasticity strongly dependent on temperature³³. At 1273 K, the strain rate of 12% causes many W atoms in the FCC phase, which accounts for 1.6% of the total atoms. As the strain rate reaches 14%, the number of W atoms in the FCC phase reaches a

peak, and it accounts for nearly 2.8% of the total, which shows a clear contrast with the stretching at room temperature. Compared with the MD analysis²⁷ of W single-crystal stretching at other orientations, the temperature is an essential factor that affects whether the phase transition occurs in the plastic deformation process of W. Fig. 5(d) is the enlarged view of the red circle in Fig. 5(c). According to the position of the coordinate axis of the atoms, part of $[01\overline{1}]_{BCC}$ structure is transited to $[112]_{FCC}$ structure, which causes lattice correspondence in $(01\overline{1})_{BCC}//(112)_{FCC}$. The process conforms to the Pitsch³¹ model and fully corresponds to the W₁ phase transition at W–Y₂O₃ interface in Fig. 4. The crystallographic transition mechanism corresponding to this process is shown in Fig. 4(e). The phase transition process is expressed as

Pitsch $[01\overline{1}]_{BCC} \rightarrow [112]_{FCC}$



Fig. 5. MD stretching simulation at room temperature and 1273 K (a) W singlecrystal stretching along [211] at the initial state, (b–c) screenshot when the strain rates are 12% and 14% at room temperature and 1273 K, (d) partially enlarged view at the red dotted frame in Fig. c, (e) crystallographic mechanism of the MD phase transition process, and corresponding Pitsch phase transition process

First-principle calculation of interface energy of ODS-W alloy phases: The abovementioned experimental phenomena all point to an important piece of information. The comparison of the interface bonding strength between Y_2O_3 and W with different crystal structures can predict the preferential breakage position, from which the experimental phenomenon can be reasonably explained. The ideal work of adhesion (W_{ad}) was used to describe the interface bonding strength³⁴⁻³⁶. This parameter

refers to the reversible work required to separate the interface into two free surfaces while ignoring the plasticity and diffusion degrees of freedom. It depends on the type of interface atoms and the electronic structure. The adhesion work W_{ad}, as one of the key parameters describing the bonding strength of interface atoms, can be used to measure the stability of the interface. The calculation formula of W_{ad} is as follows:

$$W_{ad} = (E_A^{slab} + E_B^{slab} - E_{AB}^{total})/S$$
(1)

where W_{ad} is the interface adhesion work; E_A^{slab} , E_B^{slab} , and E_{AB}^{total} refers to the total energy of surfaces A, B, and A/B interface system, respectively; S is the interface area.

According to experimental results of in situ stretching TEM, the W matrix at the interface undergoes a phase transition from BCC to FCC, and the FCC phase W matrix is torn off by Y₂O₃. On this basis, the W_{ad} of Y₂O₃/FCC-W interface, Y₂O₃/BCC-W interface, and BCC-W/FCC-W interface, as well as that the interface in BCC-W matrix, were explored to elaborate the abovementioned experimental results. Fig. 6(a) shows the isolated Y₂O₃(020) and FCC-W(220) surface, and the corresponding optimized Y₂O₃(020)/FCC-W(220) interface structure is shown in Fig. 6(b). At the interface, W binds well to O atoms on Y₂O₃ surface, with the bond length in the range of 1.85–2.05 Å and the interfacial distance of 1.801 Å. The corresponding work of adhesion W_{ad} is 0.368 eV/Å²(Fig. 6b). Fig. 6(c) shows the isolated Y₂O₃(020) interface with lower W_{ad} of 0.208 eV/Å². The longer bond length of 1.95–2.25 Å is consistent with the lower W_{ad}. As observed, different interface atomic arrangements between FCC-W(220) and BCC-W(220) and BCC-W(220) interface atomic arrangements between FCC-W(220) and BCC-W(220) and BCC-W(200) cause different relative atoms positions when forming interface with

Y₂O₃(020). Thus, the bonding strength of W–O differs, and the interface energy is different eventually. In situ experiment proves that BCC-W/FCC-W interface exists in the form of BCC-W(211)/FCC-W($3\overline{1}\overline{1}$)interface. Notably, in the process of structural optimization of the interface, FCC-W tends to transit to BCC-W, which is consistent with the phenomenon observed by Dr. Zhang Jianfei²⁶ and further proves that FCC phase is unstable. Herein, UBER curve (universal binding energy relation) was adopted to estimate the adhesion work as shown in Supplementary Fig. 2. Instead of structure optimization, the electronic energy of fixed BCC-W/FCC-W interface structure with the interface spacing in the range of $1.70 \sim 3.10$ Å was calculated. The maximum interface adhesion work in this interval is taken, and the corresponding spacing of this point is the best interface spacing. According to the curve, the optimal interface spacing of 1.90 Å corresponds to the maximum interface adhesion work of 0.243 eVÅ² (Fig. 6f). For the W_{BCC} matrix, its internal crystal plane has ultra-high adhesion energy (0.577 eV/Å², Fig. 6h). So far, we have obtained the inequality of interface bonding strength as follows:

$$W_{BCC}-W_{BCC} > W_{FCC}-Y_2O_3 > W_{BCC}-W_{FCC} > W_{BCC}-Y_2O_3$$
(2)

Accordingly, the ultra-high adhesion energy of W_{BCC} matrix indicates that tearing it off is nearly impossible, but the phase transition becomes an effective way to release the stress of the W matrix at high temperature. After the phase transition, the interface bonding strength of W_{BCC} – W_{FCC} is significantly lower than that of W_{BCC} – W_{BCC} , and even lower than that of W_{FCC} and Y_2O_3 interface. Therefore, the fracture occurs at the W_{BCC} – W_{FCC} interface, which enables the Y_2O_3 interface to tear off the W_{FCC} matrix. This condition explains why all the torn W exhibits FCC phase. Given that $W_{BCC}-Y_2O_3$ shows the lowest interface bonding strength, if no phase transformation occurs, then the pure W_{BCC} would not be torn off by Y_2O_3 and the fracture would occur at the $W_{BCC}-Y_2O_3$ interface. At room temperature, none phase transition occurs to promote tearing phenomenon. The crack will extend all the way to the matrix along Y_2O_3 . Therefore, $W-Y_2O_3$ alloy only shows ordinary second phase strengthening effect at room temperature, but the strengthening effect of second-phase transformation at high temperature is reflected.



Fig. 6. First-principle simulation of interface energy of W–Y₂O₃ alloy phases (a) FCC-W($2\overline{2}0$)/Y₂O₃(020) bonding, (b) work of adhesion at W_{FCC}–Y₂O₃ interface, (c) BCC-W(200)/Y₂O₃(020) bonding, (d) work of adhesion at W_{BCC}–Y₂O₃ interface, (e) BCC-W(211)/FCC-W($3\overline{1}\overline{1}$) bonding, (f) work of adhesion at W_{BCC}–W_{FCC} interface, (g) BCC-W(200)/BCC-W(200) bonding, (h) work of adhesion at W_{BCC}–W_{BCC} interface

In summary, HRTEM analysis during in situ high-temperature stretching shows that W transformed from BCC phase to FCC phase has higher activity. Combining MD

simulation and first-principle calculations confirms that W in the FCC phase has a stronger interfacial bonding force with the rare-earth oxide interface. In the stretching process, the crack preferentially selects the phase transition area. Thus, W in the FCC phase is separated from the matrix and sticks to the rare-earth oxide interface, similar to a water drop. This process helps absorb deformation energy and reduce local stress concentration, which prevents crack propagation and causes a ductile fracture. These observations supplement the new second-phase strengthening mechanism, fill the gap in the second-phase strengthening mechanism at high temperatures, and provide a new idea for strengthening BCC structural metals used at high temperatures.

Methods

Preparation of W–Y₂O₃ alloy block: W–Y₂O₃ alloy was prepared by wet chemical method combined with SPS. W–Y₂O₃ alloy precursor powder was obtained by combining W source [(NH₄)₆H₂W₁₂O₄₀·xH₂O)] and Yttrium source [(Y(NO₃)₃·6H₂O)] and mixed in deionized water solution, and oxalic acid (C₂H₂O₄·2H₂O) and dispersant triethanolamine (C₆H₁₅NO₃) were added after stirring, which obtained W–Y₂O₃ precursor. The obtained precursor was placed in a tube furnace and reduced in a hydrogen atmosphere to obtain WY₂O₃ composite powder. W–Y₂O₃ alloy block with a density of 99% was finally obtained by SPS sintering.

MD calculation method: MD simulations were conducted by adopting the classical open-source code large-scale atomic/molecular massively parallel simulator³⁷. The simulated single-crystal W nanowire was modeled as a cuboid with dimensions of 28.49

nm × 94.95 nm × 94.95 nm, where coordinate axes of X, Y, and Z were along the crystal orientations of [211], $[01\overline{1}]$, and $[\overline{1}11]$, respectively. The sample contained a total of 108000 atoms. The interaction between W atoms was described by the Finnis–Sinclair-type potential proposed by Ackland and Thetford³⁸, the single-crystal W nanowire was uniaxially stretched along the X direction at a constant strain rate of 108 s⁻¹ until failure. Periodic boundary condition was applied along the tensile direction of the nanowire, and the side surfaces were under the traction-free boundary conditions. During tensile deformation, the temperature was kept constant and the time step was taken as 1 fs. The simulations were performed at 1273 and 293 K for comparison. The atoms were colored according to their local crystal structure by using the common neighbor analysis method³⁹, and the dislocations were extracted based on the dislocation extraction algorithm⁴⁰. Snapshot visualization was conducted using the OVITO software⁴¹.

First-principle calculation method: The density functional theory calculations were performed with the Vienna Ab initio Simulation Package (VASP 5.4) based on projector augmented wave method⁴²⁻⁴³. Perdew–Burke–Ernzerhof functional within generalized gradient approximation was adopted to treat electronic exchange– correlation energy⁴⁴. For W and Y₂O₃ surfaces, integration in Brillouin zone was performed to optimize geometric configuration using the k-point mesh of 1×1×1 on the basis of the Monkhorst–Pack scheme⁴⁵. The plane-wave cutoff energy was set to 400 eV. The convergence criteria for structural optimization were set as 0.1E-04 eV and 0.03 eV/A for energy and force, respectively. The vacuum layer of 20 Å was used to

avoid interaction between adjacent layers.

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Author Contributions

G.Q.W., L.M.L. and Y.C.W. conceived the project. G.Q.W., Q.Z., X.P.L. and Y.F.Z. planned and performed the experiments and collected and analyzed the data. H.X.L., Z.P.L. assisted with the experiments and characterizations. G.Q.W., L.M.L., X.P.L., Q.Z. and Y.C.W. co-wrote the manuscript. All authors discussed the results and commented on the manuscript.

Ethics declarations

Competing interests

The authors declare no competing interests.

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