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1 **Expansion of C₄ vegetation and wildfire intensification during the late Cenozoic caused**
2 **atmospheric CO₂ drawdown**

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8 **Abstract:**

9 The late Cenozoic [6.8 to 0.5 Ma] vegetation ecosystem preferred to hatch slack photosynthetic
10 C₄ pathways^{1,2} and witnessed intense wildfire activity³⁻⁵. Modern-day such ecosystems are
11 considered as potential atmospheric CO₂ pumps due to their ability to enhance erosion rate and
12 rapid burial of charcoal of terrestrial origin⁶⁻⁸. However, their role has never been considered
13 in the context of the Late Cenozoic CO₂ drawdown. Here, we showed evidence of C₄ vegetation
14 expansion, intense wildfire activities, and consequent shift in the soil erosion rate during Late
15 Cenozoic based on terrestrial organic carbon (TOC) stable isotope, clay fixed ammonium stable
16 nitrogen isotope, and its mean accumulation rate (MAR) from the sediments of International
17 Ocean Discovery Program (IODP) expedition 362 site U1480, Nicobar fan. Our result suggests
18 that the Plio-Pleistocene boundary ecosystem recorded 22% more C₄ vegetation and 2.5 times
19 more wildfire-associated erosion rate than the Late Miocene. We exhibit the yield of both
20 processes on ~150ppm drop-in pCO₂ during the Plio-Pleistocene boundary through an 11%
21 rise in silicate mineral reactivity and 2.3 times terrestrial organic carbon burial. These estimates
22 are based on a reappraisal of global seawater ⁸⁷Sr/⁸⁶Sr ratio and TOC MAR at the present study
23 site. Overall, our results demonstrate that C₄ vegetational expansion and wildfire are likely to
24 function in silicate mineral reactivity and terrestrial organic burial during Late Cenozoic. We

25 anticipate our results serve as a premise for further investigations into the linkage between
26 global wildfire occurrences in driving C₄ vegetational abundances and climate change.

27

28 **Main text:**

29 Late Cenozoic drop in the partial pressure of atmospheric CO₂ (pCO₂) is attributed to the
30 increased erosion rate⁹⁻¹¹ and burial of terrestrial derived organic matter¹². However, the
31 driving mechanism for these processes is still poorly understood. Concurrently, the preference
32 of hatch slack photosynthetic C₄ vegetation at the expense of the C₃ photosynthetic pathway
33 happened worldwide^{1,2}. C₄ vegetation is supported in a fine crumb clod texture mollisols
34 whose internal surface area is larger than the coarse blocky textured soils hosting C₃
35 vegetation^{6,13}, which facilitate ten times excess carbon fixation by C₄ vegetation-associated
36 soil than forestland in the comparable climate regime¹⁴. The C₄ vegetation is susceptible to
37 wildfire and can recover quickly after each fire event^{15,16}. The recurrent wildfire activities are
38 attributed as a significant contributor to atmospheric CO₂ globally at an interannual time scale
39 but compensated by the drawdown of CO₂ due to plant regrowth⁸. Further, Wildfires
40 sequester atmospheric CO₂ by producing 256 Tg C of charcoal annually⁸ and promoting soil
41 erosion⁷. Charcoal can be stockpiled in a terrestrial and marine system for centuries to millennia
42 and can be contemplated as a long-term effective mechanism for carbon sequestration. We
43 proposed here an effective method for locking atmospheric carbon by expansion of C₄
44 vegetation and high intense event of wildfire activities in the flood plain region of Himalayan
45 river basin during the Late Cenozoic.

46 Modern-day Himalayan rivers, especially Ganges-Brahmaputra, supply 24% of sediments¹⁷,
47 which includes 10-20% of organic carbon of terrestrial origin to the global ocean¹⁸ and
48 participate in regulation of the atmospheric CO₂ levels. These sediments are brought by rivers
49 to the Bengal-Nicobar fan and develop a continuous record of sedimentation documenting shift

50 in climate and ecology over flood plain region of South Asia. The previous record of
51 vegetational shift^{2,19}, wildfire intensity⁵ and erosion rate⁹ over South Asia during Late Cenozoic
52 were reported from the region of Bengal fan. A recent expedition of the International Ocean
53 Discovery Program (IODP) 362 sampled and logged the sedimentary record from the Nicobar
54 fan. The detrital zircon age spectra²⁰ and radiogenic Sr and Nd isotope composition²¹ of
55 turbidite deposits from IODP 362 site U1480 suggest that the Eastern Himalaya and Indo-
56 Burma range are major sediment sources. The biostratigraphy and magnetostratigraphy datums
57 based age-depth model suggests the Nicobar fan received nearly 400% excess sediment from
58 river draining South East Asia than the proportion accumulated at the Bengal fan during late
59 Cenozoic^{20,22}. Together with this, the presence of wood remains from the terrestrial plants, and
60 the low calcium carbonate content of sediments confirms the exclusive dominance of land-
61 derived sediments with insignificant input from primary marine productivity at the site of
62 deposition^{23,24}. These features make the Nicobar fan an archetypal for examining the factors
63 responsible for C₄ vegetation and ascertaining the link with intense wildfire events, driving
64 rapid CO₂ sequestration trends detected in the Late Cenozoic sedimentary record.

65 Here we reconstructed proportional shift in C₄ vegetation (%), wildfire intensity, and coupled
66 erosion rate over the region of Eastern Himalaya and Indo-Burma using stable carbon
67 $\{\delta^{13}\text{C}_{\text{TOC}} = [({}^{13}\text{C}/{}^{12}\text{C}_{\text{TOC}})/({}^{13}\text{C}/{}^{12}\text{C}_{\text{VPDB}}) - 1] * 1000\}$ of Total Organic Carbon (TOC), and
68 nitrogen isotope $\{\delta^{15}\text{N}_{\text{f. NH}_4^+} = [({}^{15}\text{N}/{}^{14}\text{N}_{\text{f. NH}_4^+})/({}^{15}\text{N}/{}^{14}\text{N}_{\text{N}_2 \text{ air}}) - 1] * 1000\}$ and Mean
69 Accumulation Rate (MAR) of fixed ammonium in clay from Late Cenozoic [6.8 to 0.5 Ma]
70 sedimentary succession of IODP 362 site U1480. Furthermore, we showed coupling of C₄
71 vegetational shift and increased erosion rate on the regulation of mineral reactivity and
72 consequent burial of terrestrial organic matter through a reappraisal of global seawater ⁸⁷Sr/⁸⁶Sr
73 ratio and TOC MAR at the site of the present study.

74 **Late Cenozoic C₄ expansion on Eastern Himalaya and Indo-Burma Range:**

75 Isotopic composition of organic matter ($\delta^{13}\text{C}_{\text{TOC}}$ value) ranges between -32.1‰ VPDB and -
76 15.8‰ VPDB (Fig. 1a & **Error! Reference source not found.**), which implies variable
77 contribution from C_3 and C_4 vegetations present in the catchment. These isotope values are
78 converted in terms of fraction of C_4 vegetation-derived carbon by assuming a binary mixing
79 model with a well-defined $\delta^{13}\text{C}$ values for the modern-day C_3 and C_4 plant present in South
80 Asia. The prescribed region specific $\delta^{13}\text{C}$ values for C_3 and C_4 vegetation are -32.6‰ and -
81 10.4‰, respectively²⁵. Our estimates show C_4 vegetation was present in the catchment with
82 fractional percentage of $\sim 25 \pm 3\%$ (n=17) during Late Miocene (6.8 to 5.2 Ma) (Fig. 1a). This
83 is substantiated with reconstructed C_4 vegetation record over the region of Indo-Burma based
84 on $\delta^{13}\text{C}$ of mammals tooth enamel²⁶ and dominance of deciduous forest in the expense of
85 wetland evergreen vegetation over Eastern Himalaya²⁷. Afterward, the percentage dominance
86 of C_4 vegetation started growing and reached maximum value of $\sim 47 \pm 2\%$ (n=31) at Plio-
87 Pleistocene boundary (3.0 to 1.8 Ma) (Fig. 1a). These two stages registering expansion of C_4
88 vegetation during Late Cenozoic are global in nature and found in several regions across many
89 continents². However, factors responsible for the vegetational shift is highly debatable with
90 argument supporting appropriate pCO_2 level²⁸, increased aridity^{29,30} and intense wildfire
91 activity^{4,5}. While the earlier two factors are respectively contested by fossilised C_4 grass pollen
92 record¹ and stable hydrogen isotope of leaf wax³¹, sufficient evidence favoured wildfire
93 intensity as the only option promoting C_4 vegetation³⁻⁶.

94 **Proxy for wildfire intensity and associated erosion rate:**

95 Wildfire is adjudged as a possible mechanism to explain the late Cenozoic proliferation of C_4
96 vegetation based on the modern-day analogy^{3,5,15}. Wildfire generates tropospheric ozone,
97 which is toxic to plant productivity³². The waning of productivity by Wildfire increases the soil
98 temperature and moisture, thereby significantly accelerating the soil ammonification rate and
99 reducing plant NH_4^+ uptake, making the soil environment excess in NH_4^+ radical³³. This

100 process enriches lighter isotope of nitrogen(^{14}N) in soil ammonium³⁴. Due to similar ionic radii
101 and low hydration energy, NH_4^+ in the soil can replace K^+ in the interlayer structure of clay
102 minerals³⁵. The penetration of NH_4^+ into the clay mineral interlayer reduces the interlayer
103 distance by 1nm and traps NH_4^+ in silicate structure as fixed NH_4^+ .

104 On the note of clay-fixed NH_4^+ concentration, the competition between soil erosion and
105 leaching plays a significant role. The leaching process can reduce soil NH_4^+ content via the
106 process of dissolution with water from runoff and groundwater interaction. This will allow
107 diffusion of clay-fixed NH_4^+ to attain chemical equilibrium with vicinity soil ammonium³³.
108 Such diffusion can be avoided by quick transport of clay from the place of origin to the
109 adjoining lowland area. These characteristics make the stable nitrogen isotope ($\delta^{15}\text{N}_{\text{f. NH}_4^+}$) and
110 fixed NH_4^+ in the clays as a unique proxy for wildfire intensity and soil erosion rate,
111 respectively.

112 **Wildfire intensity and triggering mechanism:**

113 The absence of fixed NH_4^+ in the clays deposited during the time window of 6.8 to 6.3Ma
114 revealed a period without wildfire activity in the river's catchment. With progressive younging
115 up in the strata, $\delta^{15}\text{N}_{\text{f. NH}_4^+}$ values varied between -6.5‰ and +4.5 (Fig. 1c & **Error! Reference**
116 **source not found.**). The $\delta^{15}\text{N}_{\text{f. NH}_4^+}$ denotes a net effect of ammonium mineralization and
117 ammonium assimilation by the plants. The organic nitrogen pool in the soil (N_s) is enormous
118 and undergoes infinitesimal change due to the mineralized NH_4^+ pool generated after the
119 Wildfire. We considered the ammonium mineralization process a zero-order reaction based on
120 modern-day observation from the Wildfire experiencing chaparral soil of California³⁶, which
121 contains 10 to 5mg/g of organic nitrogen compared to about 2.37 to 2.48 $\mu\text{g/g}$ of NH_4^+ . Then,
122 the post-wildfire mineralized NH_4^+ ($\delta^{15}\text{N}_{\text{NH}_4^+(\text{initial})}$) is mostly consumed by the process of
123 assimilation rather than nitrification³⁷⁻³⁹. Therefore, the plants' ammonium assimilation

124 governs the soil $\delta^{15}\text{N}_{\text{NH}_4^+}$. This ammonium assimilation imparts large isotopic fractionation
 125 ($\epsilon_{\text{plants-soil.NH}_4^+}^{15} = \delta^{15}\text{N}_{\text{plants}} - \delta^{15}\text{N}_{\text{soil.NH}_4^+}$) of $-10.2 \pm 7.9\%$ ³⁴. We converted our
 126 $\delta^{15}\text{N}_{\text{f. NH}_4^+}$ record to the fractional percentage of soil NH_4^+ assimilated by plants using the
 127 steady-state isotope model.

$$128 \quad f_{\text{Plants NH}_4^+ \text{ assimilation}} = \frac{\delta^{15}\text{N}_{\text{soil.NH}_4^+(\text{initial})} - \delta^{15}\text{N}_{\text{soil.NH}_4^+(\text{remaining})}}{\epsilon_{\text{plants-soil.NH}_4^+}^{15}}$$

$$129 \quad \delta^{15}\text{N}_{\text{soil.NH}_4^+(\text{remaining})} = \delta^{15}\text{N}_{\text{f. NH}_4^+}$$

130

131 We assigned lightest $\delta^{15}\text{N}_{\text{f. NH}_4^+}$ value as $\delta^{15}\text{N}_{\text{soil.NH}_4^+(\text{initial})}$ and estimated fraction of soil
 132 NH_4^+ assimilated by the plant population. This is with the assumption that the lightest
 133 $\delta^{15}\text{N}_{\text{f. NH}_4^+}$ reflects absence of soil NH_4^+ assimilation by plant ecosystem. The low nutrient
 134 uptake value was prescribed over the time interval 6 to 4.9 Ma with a mean $\pm 1\text{SE}$ value of 42
 135 $\pm 13\%$ (n=6) and similarly, in the time interval 2.6 to 1.8 Ma with the mean $\pm 1\text{SE}$ value of 43
 136 $\pm 5\%$ (n=14) (Fig. 1c). These intervals are unequivocally registered intense wildfire
 137 activity and contemporaneous with evident from the independent record of Wildfire across the
 138 globe^{4,5,40,41}.

139 The most widely accepted Late Cenozoic wildfire triggering mechanism is increased
 140 monsoonal seasonality^{3,5}. More recently, an increase in cosmic ray flux due to nearby
 141 supernova explosion is proposed as a most possible mechanism to induce anomalous lightning
 142 and thunderstorm activities during the Late Cenozoic, attributed as a cause for widespread
 143 wildfire⁴². The seasonal monsoon precipitation drives the terrestrial productivity during the wet
 144 season and converted into highly combustible fuel biomass. The cosmic ray (high energy
 145 proton and other nuclei) cause frequent lightning activities; which spark Wildfire in a
 146 deciduous and evergreen forest⁴³. Monsoonal seasonality in the Late Cenozoic as a factor for
 147 wildfire activity during the dry time in the tropics⁶, but fails to explain events of wildfire

148 activity in the Arctic during the nearly same period ⁴¹. A global scenario of nearby supernova
149 explosion inducing widespread wildfire occurrences due to frequent lighting activities can
150 explain the C₄ expansion and climatic transformation during the late Cenozoic. The observation
151 of the decay corrected abundances of Iron-60 (⁶⁰Fe) incorporation in Ferromanganese crust⁴⁴
152 suggests intense cosmic ray shower due to the explosion of nearby supernova coincides with
153 the late Cenozoic wildfire events(Fig. 1b).

154 **Wildfire linked Erosion rate:**

155 The fixed clay NH₄⁺MAR estimated here serves as a proxy for the erosion rate and varies
156 between 1 and 49.5 g/cm²/Myr during the Late Cenozoic (Fig. 1d & **Error! Reference source**
157 **not found.**). The fixed NH₄⁺MAR (Mean ± 1SE) estimates during Late Miocene (6.3 to 6 Ma)
158 and Plio-Pleistocene (2.3 Ma to 1.9 Ma) time was defined using multiple samples with the
159 mean value of 10.7±1.7 g/cm²/Myr (n=5) and 26.3±4.2 g/cm²/Myr (n=11) respectively. The
160 increment of erosion rate by 2.5 times is observed during the Plio-Pleistocene compared to Late
161 Miocene. Such growth in fixed clay NH₄⁺MAR during Plio-Pleistocene was also reported over
162 the Pacific ocean ⁴⁵. In addition, independent validation of erosion rate shift is available from
163 other proxies, such as the magnetic susceptibility index of the sedimentary package from our
164 study location and decay corrected ¹⁰Be concentration-based erosion rate evaluation from
165 central Asia (Extended Data Fig 1). Observed amplification of erosion rate can also be
166 addressed by incorporating rapid tectonics^{9,10} and glacial reworking¹¹ processes. However, the
167 low-temperature thermochronometers from continental sedimentary archives suggests
168 tectonism and glacial retreat caused insignificant change in the erosion rate during Late
169 Cenozoic⁴⁶. We proposed here that intensifying wildfire activity during Late Miocene and Plio-
170 Pleistocene boundary is a potential driver for enhanced erosion rate and allowing the rapid
171 burial of organic matter from terrains dominated by the C₄ vegetation.

172 **Erosion implication on atmospheric CO₂ sequestration:**

173 Erosion usually subjects reactive/fresh minerals to the weathering process and transports
174 organic matter from the continent to the adjacent basin for burial^{13,18}. These mechanisms allow
175 rapid sequestration of atmospheric CO₂ and are registered in the sedimentary proxy record (Fig.
176 1g). The seawater ⁸⁷Sr/⁸⁶Sr record displays two stages of increase in radiogenic Sr in the time
177 interval 7Ma to 4Ma and 2.5Ma to a modern-day and intermittent state of nearly constant values
178 (Extended Data Fig. 3). The seawater ⁸⁷Sr/⁸⁶Sr is recorder of ⁸⁷Sr/⁸⁶Sr ratio and Sr flux from
179 the river runoff and hydrothermal input from fluids generated in the Mid Oceanic Ridges. The
180 hydrothermal Sr flux and ⁸⁷Sr/⁸⁶Sr ratio are governed by the seafloor spreading rate and
181 ⁸⁷Sr/⁸⁶Sr of magmatic fluid at Mid Oceanic Ridges. These ratios were constant during Late
182 Cenozoic^{47,48}; encompassing the period of the present study. The ¹⁰Be/⁹Be ratio of seawater
183 was constant and indicative of constant weathering flux⁴⁹. Therefore the ⁸⁷Sr/⁸⁶Sr ratio recorded
184 in Late Cenozoic seawater is indication of riverine ⁸⁷Sr/⁸⁶Sr ratio. This is also substantiated by
185 the ⁸⁷Sr/⁸⁶Sr record in the pedogenic clay from Bengal fan⁵⁰ and pedogenic carbonate from
186 lowland sediments of Himalayan foreland⁵¹.

187 Modern-day, the Himalayan-Tibetan Plateau (HTP) river solely contribute ~23.3% of Sr flux
188 to the Ocean⁴⁸. If we held the rest of river ⁸⁷Sr/⁸⁶Sr ratio at modern value, the reconstructed
189 ⁸⁷Sr/⁸⁶Sr ratio of HTP river suggests 0.0014 rise during Late Cenozoic (Fig. 1f). This rise can
190 be accounted by evolution of ⁸⁷Rb decay and increased reactivity of mineral with high ⁸⁷Sr/⁸⁶Sr
191 ratio in the parent rock. The Sr and Nd isotopic composition of sediments from Bengal and
192 Nicobar fun suggests major contribution from Higher Himalayan Crystalline Series (HHCS)
193 ^{21,50}. The HHCS consists of meta-sedimentary rocks (⁸⁷Sr/⁸⁶Sr = 0.7646 ± 0.0391) and Anatectic
194 granites (⁸⁷Sr/⁸⁶Sr = 0.7600 ± 0.0142) that extends from Assam to Pakistan⁵². The estimated
195 change in ⁸⁷Sr/⁸⁶Sr ratio due to excess radioactive decay of ⁸⁷Rb during Late Cenozoic, is
196 inferred using modern-day value of ⁸⁷Sr/⁸⁶Sr and ⁸⁷Rb/⁸⁶Sr, for meta-sedimentary rock and

217 Anatectic granite are 0.0001 and 0.0004, respectively. This variation solely cannot explain the
218 increment in the riverine $^{87}\text{Sr}/^{86}\text{Sr}$ without involving increased reactivity of minerals with a
219 high $^{87}\text{Sr}/^{86}\text{Sr}$ ratio. We estimated mineral reactivity in terms of reaction rate (k) of ^{87}Sr
220 dissolution (& hence $^{87}\text{Sr}/^{86}\text{Sr}$), assuming first-order kinetics for the time interval 7 Ma to 4 Ma
221 and 2.5 Ma to modern-day. The estimated k (Fig. 1f) value for the time interval 7Ma to 4Ma is
222 $3.17 \times 10^{-4} \text{ Myr}^{-1}$ (~31.5 m.y. is required to increase the riverine ^{87}Sr by 1%) and rise to $3.51 \times$
223 10^{-4} Myr^{-1} for the time interval 2.5 Ma to modern-day (accordingly ~28.5 m.y. time will be
224 required to increase the riverine ^{87}Sr by 1%).

225 Another consequence of increased erosion is the rapid organic carbon burial in proximal basins.
226 Further, the C₄ vegetational expansion in the Late Cenozoic can also contribute high organic
227 carbon, given that C₄ vegetation is supported by mollisols, which has the potential to hold ten
228 times more organic carbon than coarse-grained soil found for C₃ vegetation in a comparable
229 climate zone ⁶. We estimated the burial of organic carbon using TOC MAR values. The high
230 value of TOC MAR (Mean \pm 1SE) was noted during the Late Miocene time interval (6.5 to 6
231 Ma) and Plio-Pleistocene time interval (2.4 to 1.8 Ma) with $51 \pm 6 \text{ g/cm}^2/\text{Ma}$ (n=11) and $117 \pm$
232 $12 \text{ g/cm}^2/\text{Ma}$ (n=20) respectively (Fig. 1e & **Error! Reference source not found.**) defining
233 the abundances. The shift in the TOC MAR values was in-phase with erosion rate proxy and
234 C₄ expansion trend, providing for the first time a new mechanism of rapid CO₂ suction from
235 the atmosphere and the role of nature-based terrestrial vegetation management scenario in late
236 Cenozoic settings over the region of Eastern Himalaya and Indo-Burma.

237 **Conclusion:**

238 Overall, the present study elucidates the role of C₄ vegetation and associated Wildfire in the
239 increment of erosion rate and rapid drawdown of CO₂ from the global atmosphere during the
240 Late Cenozoic. Compared to Late Miocene, the Plio-Pleistocene boundary recorded 22% more
241 C₄ vegetation and 2.5 times more wildfire-associated erosion rate. Both processes yielded a

222 ~150ppm drop-in pCO₂ through an 11% rise in silicate mineral reactivity and 2.3 times organic
223 carbon burial. Given that the 24% global terrestrial sediment flux and 10 to 20% of terrestrial
224 organic carbon are depositing in Bengal and Nicobar fan^{17,18}, our study provides insight into
225 the C₃ to C₄ vegetational shift associated with wildfire activity is effective in global
226 environmental transformation during Late Cenozoic. The results may serve as a premise for
227 further investigations into the linkage between C₄ vegetational abundance, Wildfire, and
228 climate change.

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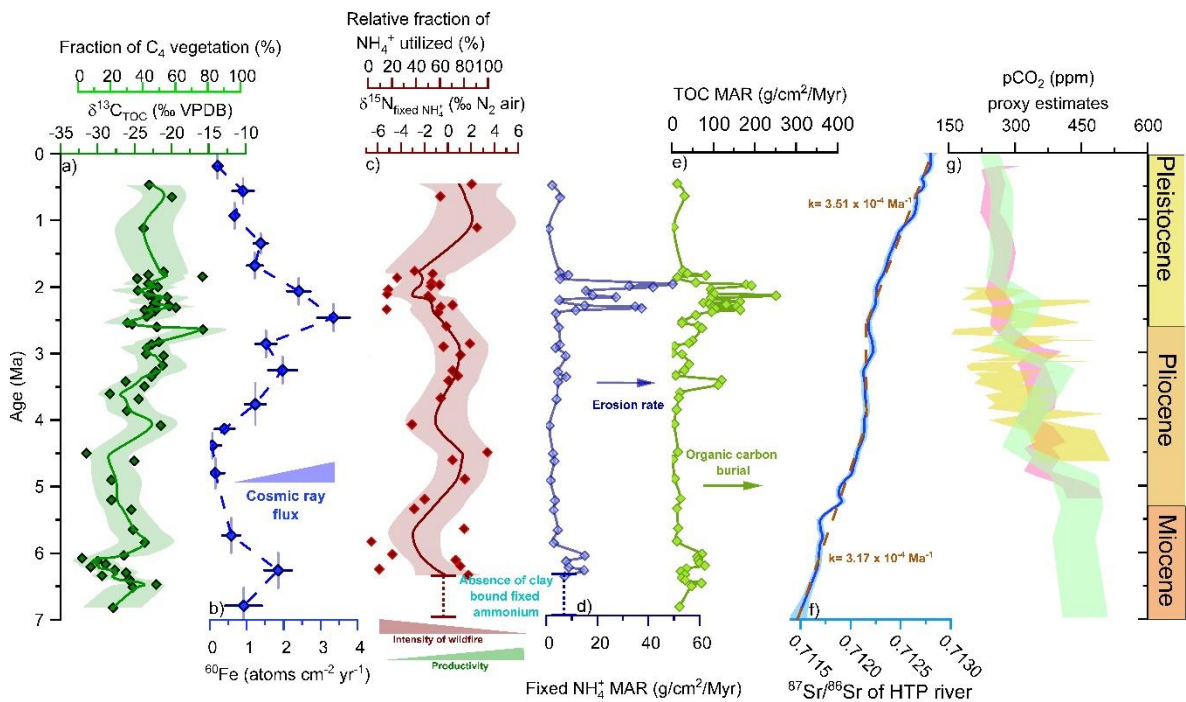
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396

397 **Fig. 1: C₄ vegetation and Wildfire impact on CO₂ drawdown during Late Cenozoic.** **a,**
398 $\delta^{13}\text{C}_{\text{TOC}}$ and estimated fraction of C₄ vegetation (%). **b,** Decay corrected ^{60}Fe incorporation rate
399 as a proxy of cosmic ray flux from nearby supernova explosion⁴⁴. **c,** $\delta^{15}\text{N}_{\text{fixedNH}_4^+}$ and estimated
400 relative fraction of nutrient (NH_4^+) utilization as a proxy of wildfire intensity. **d,** Fixed NH_4^+
401 MAR as a proxy for erosion rate. **e,** TOC MAR as a proxy of the burial of organic carbon. **f,**
402 Reconstructed $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of Himalayan Tibetan Plateau(HTP) river and reaction rate (k)
403 between 7 Ma to 4 Ma and 2.5 Ma to the modern-day. **g,** Alkenone (Pink)⁵³ and Boron isotope
404 of foraminifera (Yellow⁵⁴ & Green⁵⁵) based pCO₂ reconstruction. The solid line and shaded
405 area in panels a,c, and f indicate the LOESS fit line and 95% confidence interval, respectively,
406 computed using the R language SpaitalEco package.

407

408 **Methods:**

409 **Stable isotope of Total Organic Carbon and TOC MAR:**

410 Total 64 samples were selected from Unit I and Unit II of IODP-362 site U1480 based on the
411 yield of organic matter. Prior to isotope analysis, the samples were decarbonated with 1N HCl
412 at room temperature. After ensuring the complete removal of carbonate, the samples were
413 rinsed with Milli-Q water until pH attain neutral. These samples were dried in a hot air oven at
414 70°C for a minimum duration of 24 hours and homogenized using agate mortar and pestle.
415 Around 5mg of these samples were packed in tin capsule and combusted at 850°C in the Flash
416 2000 Elemental Analyser that was coupled with Delta V advantage IRMS (by Thermo Electron
417 Corporation, Bremen) interfaced via CONFLO IV. Carbon isotopic compositions are reported
418 in the conventional delta notation (δ) in per mill (‰) relative to Vienna Pee Dee Belemnite
419 (VPDB). The lab working reference materials, which includes OASIS_Glucose ($\delta^{13}\text{C} = -10.99$
420 ‰VPDB) and OASIS_Rice ($\delta^{13}\text{C} = -27.67$ ‰VPDB) were assigned isotope value analyzing
421 with IAEA secondary reference material; IAEA-CH-6⁵⁶. Further any drift in analytical results
422 was ascertained with analysis of reference material for multiple times in a batch at regular
423 intervals after analysis of every 3-4 samples. This allows assignment of δ -values to the samples
424 using standard-sample bracketing methodology. The standard deviation of $\delta^{13}\text{C}$ for
425 OASIS_Glucose (n=17) and OASIS_Rice (n=13) are 0.07‰ and 0.1‰, respectively.

426 For estimating the weight percentage of organic carbon, we calibrated our analysis with
427 OASIS_Glucose varying the amount ranging between 100 to 500 μ g in a batch. A linear
428 regression relationship between amount of carbon in OASIS Glucose (from the mass of
429 glucose* 0.4) and total peak area (peak area of ^{12}C and peak area of ^{13}C) to determine the TOC
430 % in the sample [(mass of carbon in the sample/ total mass of sample) * 100%]. Further, TOC
431 Mass Accumulation Rate (TOC MAR) was obtained from dry bulk density of sediment,
432 sedimentation rate and TOC content (DBD* Sed Rate* TOC %/100) where Dry Bulk Density
433 (DBD) followed previously in other studies^{20,57}.

434 **Clay bound ammonium stable isotope and fixed NH_4^+ MAR:**

435 For stable isotope and weight percentage analysis of fixed NH_4^+ in clay, we treated the 500-
436 1000mg clay fraction of 38 samples from Unit I and Unit II of IODP-362 site U1480 with 20
437 ml of KBr-KOH solution to eliminate the organic nitrogen fraction⁵⁸. The KBr-KOH
438 solution was concocted by adding 6 ml bromine solution at 0.5 ml/min to 200 ml of 2 M KOH
439 cooled in the ice bath. After adding the KBr-KOH solution with clay, the beaker was swirled
440 and allowed to stand for 2 hours with the beaker mouth covered by watch glass. Then, 60ml of
441 milli-Q water was added in the beaker and the mixture was boiled strenuously for 5 minutes.
442 After that the mixture cooled overnight and decanted and discarded the supernatant liquid. To
443 remove exchangeable ammonium, the residue was treated with 0.5M KCl. The high
444 concentration of potassium in the solution during the whole process prevents the adsorption by
445 clay minerals of ammonium formed during the oxidation process. Finally, the sample was
446 brought to neutral pH by rinsing with milli-Q water and dried in hot-air over at 70°C for a
447 duration of 24 hours.

448
449 Around 50 to 100 mg of dried clay was packed in tin capsule and combusted at 850°C in the
450 Flash 2000 Elemental Analyser that was coupled with Delta V advantage IRMS (by Thermo
451 Electron Corporation, Bremen) interfaced via CONFLO IV. Nitrogen isotopic compositions
452 are reported in the conventional delta notation (δ) in per mill (‰) relative to N_2 air. IAEA_NO3
453 ($\delta^{15}\text{N} = 4.7\text{‰}$ N_2 air) standard was run between 4-5 samples to correct for isotopic drift during
454 a batch and to assign δ -value for samples using standard-sample bracketing method. The
455 analytical precision $\delta^{15}\text{N}$ based on replicate analysis of Merck Sodium nitrate is 0.2‰ (n=10).

456

457 For estimating the weight percentage of fixed NH_4^+ in clay, we loaded the IAEA_NO3 with
458 amount ranges between 200 to 500 μg in each batch of analysis. We then used the linear
459 regression relationship between amount of nitrogen in IAEA_NO3 (mass of glucose* 0.139)
460 and total peak area (peak area of ^{14}N and peak area of ^{15}N) to estimate sample fixed NH_4^+ %
461 [(mass of nitrogen in the sample/ total mass of sample) * 1.29* 100%]. Then, We calculated
462 fixed NH_4^+ Mass Accumulation Rate (fixed NH_4^+ MAR) as (DBD* Sed Rate* fixed NH_4^+
463 %/100) where Dry Bulk Density (DBD) and Sedimentation Rate were obtained from McNeill
464 et al.,⁵⁷ and ²⁰, respectively.

465

466 As shown by Silva and Bremner⁵⁸, the efficiency of the method for removing the organic
467 nitrogen fraction (hydrolyzable and nonhydrolyzable organic nitrogen) is higher than 98%.
468 Additionally, we checked the efficiency of the oxidation of proteinaceous material (Peptone)
469 by determining the total N content of the peptone before and after the KOBBr-KOH treatment.
470 The N content value for KOBBr-KOH treated peptone is below the detection limit.

471 **FT-IR analysis of clay fraction of sample:**

472 FT-IR spectra for the clay fraction sample after organic matter removal by KOBBr-KCl
473 treatment are shown in Extended Data Fig 2. All spectrums were measured under the
474 reflectance mode using Spectragryph ver.1.2.6 software and converted into absorbance mode
475 to validate with library. The weak absorption of NH_4^+ was detected at around 1280-1600 cm^{-1}
476 within NH_4^+ bend region⁵⁹. Also, the very tiny to weak N-H stretching absorption was detected
477 within 3100-3430 cm^{-1} ⁵⁹⁻⁶¹. The trace levels of absorption within the NH_4^+ region suggest that
478 the mineralized nitrogen in the clay fraction is possibly ammonium fixed in the interlayers of
479 clay mineral.

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490 **Author contributions:** PG and TS conceptualize the problem; PG and NN generated fund for
491 the project. TS analysed the samples and wrote the paper along with PG. NN provided the
492 samples and involved in the discussion.

493 **Declaration of Competing Interest:**

494 Authors declare that they have no competing interests.

495 **Data and materials availability:**

496 All data needed to evaluate the conclusions in the paper are present in the paper and/or the
497 Supplementary Materials. Additional data related to this paper may be requested from the
498 authors.

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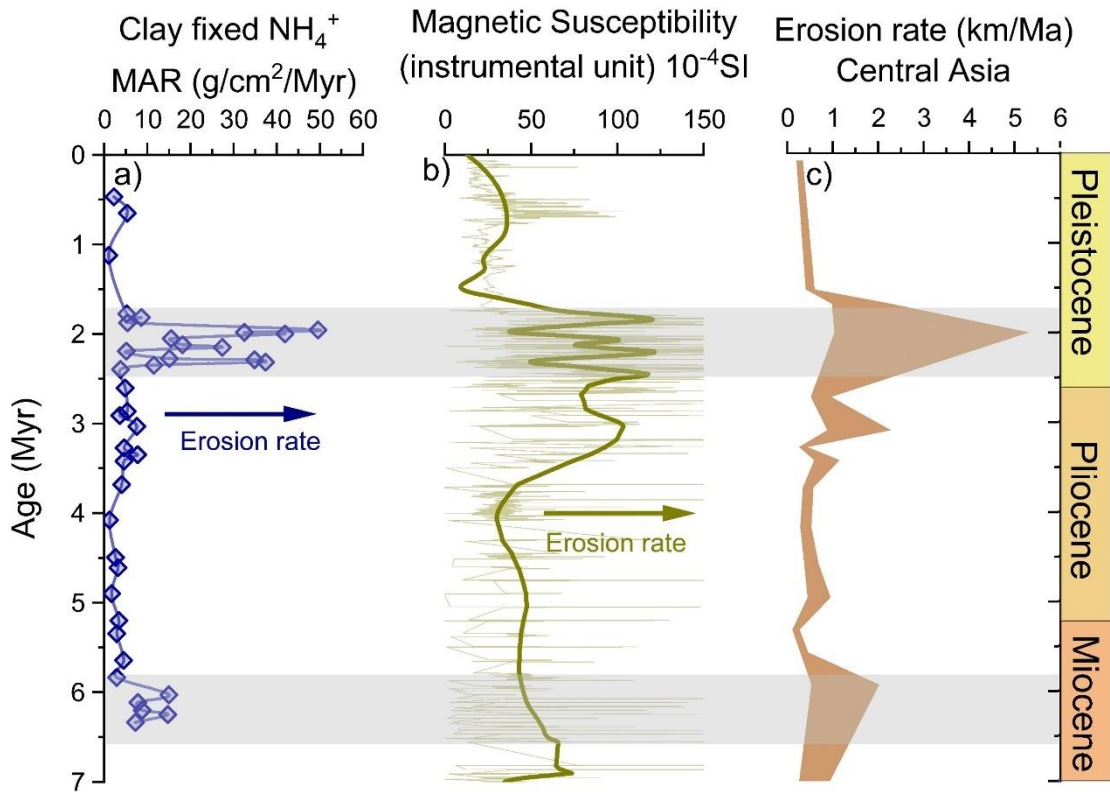
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507 **Extended Data Fig 1: Multiple evidence of increased erosion rate during late Miocene**
 508 **and Plio-Pleistocene boundary**
 509



510

511 a) Fixed NH₄⁺ MAR from IODP site U1480 (present study). b) Magnetic susceptibility from
 512 IODP site U1480⁶². c) cosmogenic ¹⁰Be concentrations based erosion rate from central Asia⁶³.

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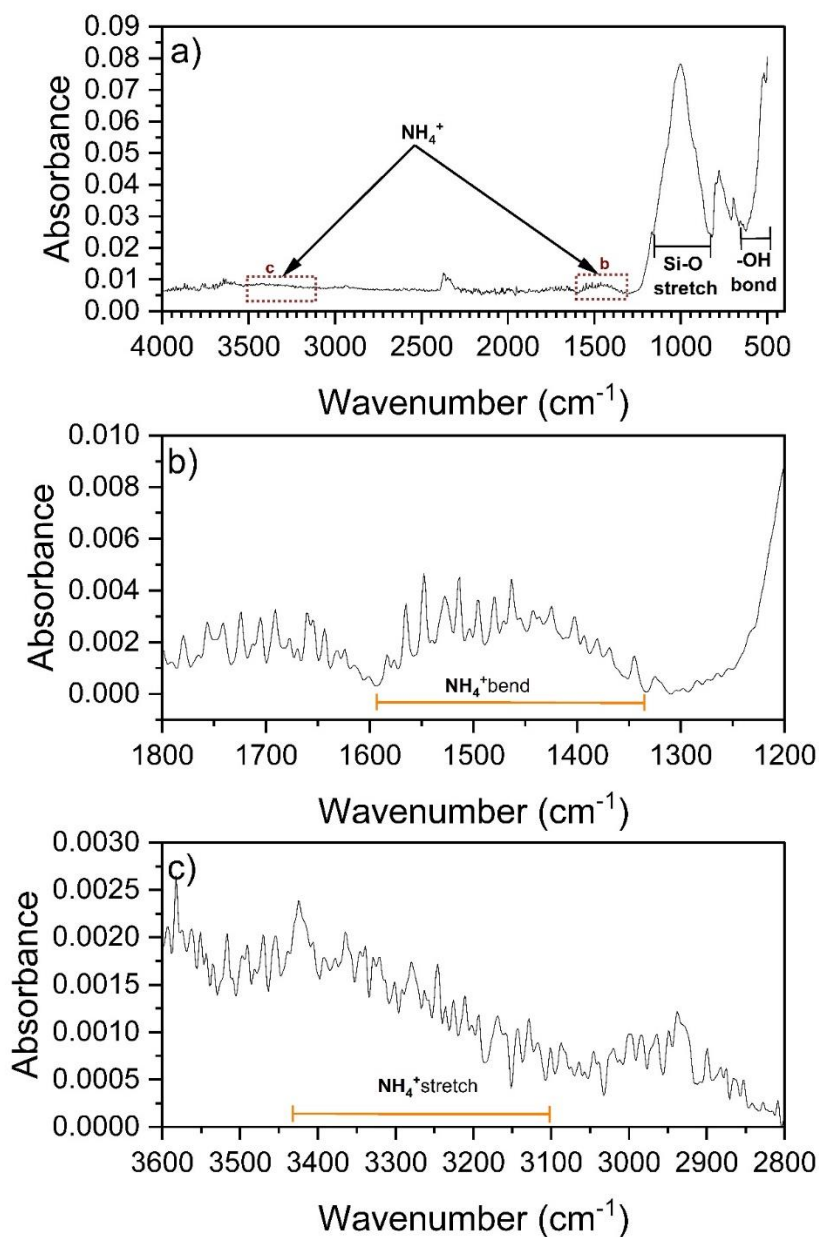
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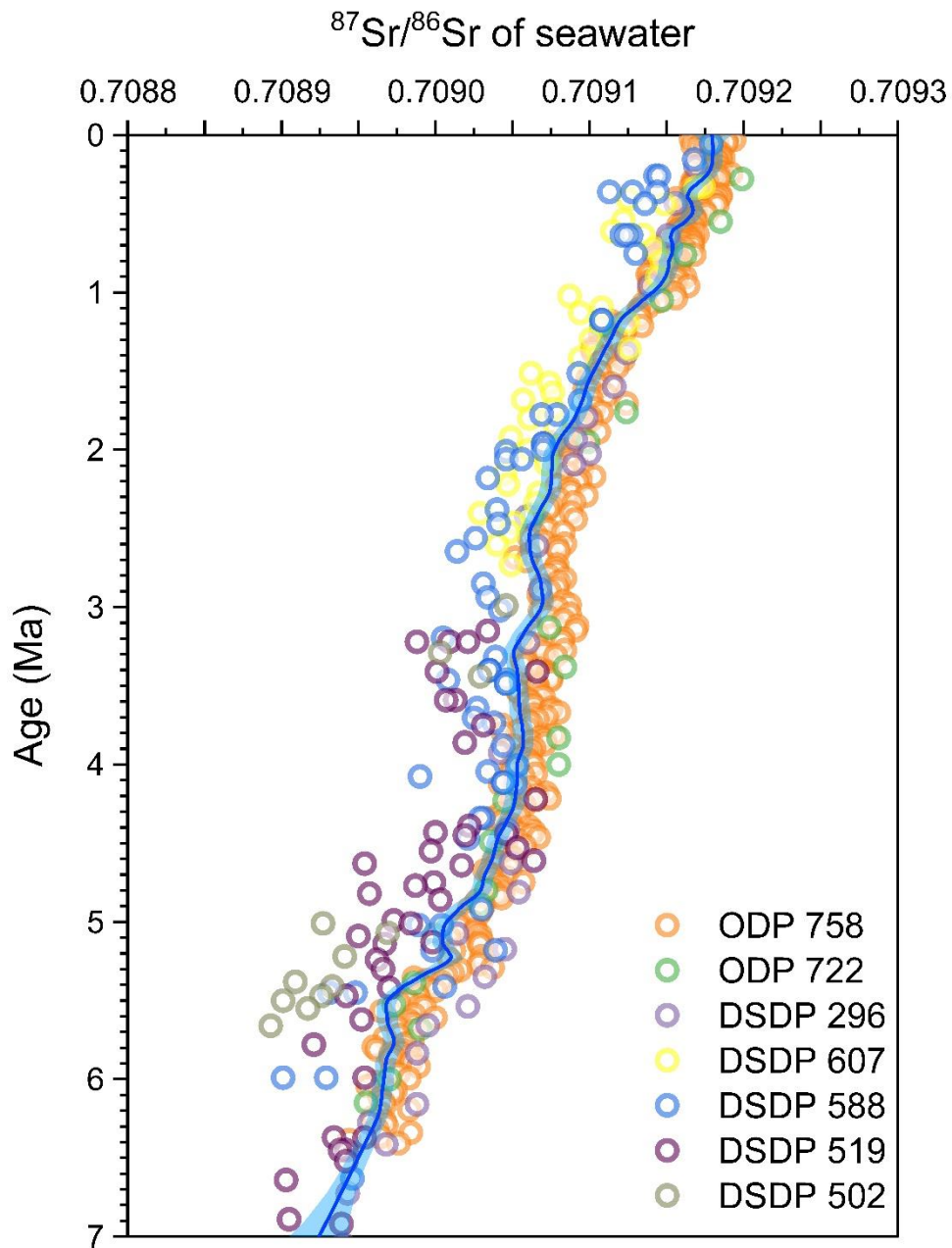
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Extended Data Fig 2: Representative FT-IR spectra of the clay fraction of sample from Late Pliocene section of IODP 362 after KOBBr-KCl treatment.



525

526 a) OH- bond, Si-O stretch, NH₄⁺ absorption peak was shown in the absorbance band from 400
527 to 4000 cm⁻¹ wavenumber; b) Absorbance band of NH₄⁺ bend; c) Absorbance band of NH₄⁺
528 stretch.



530

531 Compilation of $^{87}\text{Sr}/^{86}\text{Sr}$ marine authigenic carbonate from ODP^{64,65} and DSDP⁶⁶⁻⁶⁸ sites. The
 532 solid blue line and shaded area indicate the LOESS fit line and 95% confidence interval
 533 computed using the R language SpaitalEco package.

534

Supplementary Files

This is a list of supplementary files associated with this preprint. Click to download.

- [Extendeddatatable1.xlsx](#)
- [Extendeddatatable2.xlsx](#)