

Stratifying by Vegetation and Hydrology Improves Tidal Marsh Methane Emission Accounting

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Abstract

Methane emissions must be directly measured or estimated using methods such as proxies when managing wetlands for greenhouse gas offset activities. Salinity is a useful proxy for tidal marsh CH₄ emissions when comparing across a wide range of salinity regimes but does not adequately explain variation in brackish and freshwater regimes where variation in emissions is large. We sought to improve upon the salinity proxy in a marsh complex on Deal Island Peninsula, Maryland, USA by identifying four strata based on hydrology and plant community composition. Mean CH₄ chamber-collected emissions measured as mg CH₄ m⁻² hr⁻¹ ranked as *S. alterniflora* (1.2 ± 0.3) >> High-elevation *J. roemerianus* (0.4 ± 0.06) > Low-elevation *J. roemerianus* (0.3 ± 0.07) = *S. patens* (0.1 ± 0.01). Sulfate depletion generally reflected the same pattern with significantly greater in the *S. alterniflora* stratum (61 ± 4%) than in the *S. patens* stratum (1 ± 9%) with the *J. roemerianus* strata falling in between. We attribute the high CH₄ emissions in the *S. alterniflora* stratum to sulfate depletion likely driven by limited connectivity to tidal waters. Low CH₄ emissions in the *S. patens* stratum are attributed to lower water levels, higher levels of ferric iron, and shallow rooting depth. Moderate CH₄ emissions from the *J. roemerianus* strata were likely due to plant traits that favor CH₄ oxidation over CH₄ production. We concluded that stratification by hydrology and plant community composition can be an effective proxy to estimate CH₄ emissions at the site scale.

Introduction

Methane is a potent greenhouse gas produced under the dominantly anaerobic conditions found in wetland soils. The global warming potential of methane (CH₄) gas is 32-45 times greater than an equivalent amount of carbon dioxide (CO₂) over a 100-year period (Neubauer and Megonigal 2015). While the majority of CH₄ emissions come from anthropogenic sources, wetlands produce most of the naturally emitted CH₄ (Wang et al. 1996; Solomon et al. 2007) and are the most important source of uncertainty in current global CH₄ budgets (Saunio et al. 2020). Coastal wetland CH₄ emissions were recently estimated at 5.3-6.2 Tg CH₄ yr⁻¹, amounting to 60% of the global marine CH₄ budget (Al-Haj and Fulweiler 2020), < 7% of global wetland CH₄ budget (Saunio et al. 2020), and the largest source of uncertainty in the coastal wetland greenhouse gas budget (Holmquist et al. 2018). There is emerging interest in using tidal marsh restoration and conservation to mitigate greenhouse gases in the atmosphere and as a source of carbon credits (Crooks et al. 2011; Emmer et al. 2015, Needelman et al. 2018, Emmer et al. 2020a, Emmer et al. 2020b), but the high carbon sequestration rates characteristic of tidal wetlands soils (Nahlik and Fennessy, 2016; Chmura et al. 2003) can be partly or completely offset by CH₄ emissions (Poffenbarger et al. 2011). The uncertainty introduced to greenhouse gas offset activities by CH₄ emissions is especially large for coastal wetlands ecosystems with freshwater-to-brackish salinity <18 ppt (Poffenbarger et al. 2011). The sources of this variability remain elusive as there has been relatively little research designed to partition variation. A better understanding of the factors that regulate coastal wetland CH₄ emissions is needed to improve global CH₄ budgets and to support the implementation of carbon credit methodologies in freshwater and brackish coastal wetlands.

Methane is produced in wetlands by methanogenic archaea and bacteria. The production of CH₄ occurs when there is an excess of electron donors over electron acceptors, depleting the availability of alternative electron acceptors such as ferric iron (Fe(III)) and sulfate (SO₄²⁻) (Megonigal et al. 2004). Electron donors are produced from labile organic materials that undergo fermentation to low molecular weight carbon compounds and H₂. Electron donors can be present in the soil (e.g. Fe(III)), supplied from external sources such as floodwater (e.g. SO₄²⁻), or provided by plants (e.g. molecular oxygen or oxidized compounds generated by radial oxygen loss). The availability of SO₄²⁻

from seawater suppresses CH₄ emissions from polyhaline (salinity > 18 ppt) marshes to consistently low rates (0.2 to 5.7 g CH₄ m⁻² yr⁻¹) (Poffenbarger et al. 2011). Methane emissions from mesohaline brackish systems (5-18 ppt salinity) are greater and more variable (3.3 to 32.0 g CH₄ m⁻² yr⁻¹). The process is also regulated by such physiochemical factors as pH (Walker et al. 1998; Garcia et al. 2000) and temperature (Meronigal and Schlesinger 2002; Whalen 2005).

Plant species composition affects CH₄ emissions through several mechanisms (Koebsch et al. 2013; Moor et al. 2017; Mueller et al. 2020). The availability of electron donors is largely determined by primary productivity which varies with species composition (Meronigal et al. 2004). Species composition also regulates electron acceptor availability through rhizosphere processes such as root oxygen loss (Calhoun and King, 1997; Colmer, 2003; Jespersen et al. 1998) and rhizosphere regeneration of ferric iron (Neubauer et al, 2005; Sutton-Grier and Meronigal, 2011). Methane can be transported to the atmosphere via aerenchyma tissue, bypassing the emission barriers caused by slow CH₄ diffusion rates through soils and soil-surface CH₄ oxidation zones (Ding et al. 2005; Le Mer and Roger, 2001; Sorrell et al. 2013; Villa et al. 2020). This is important because CH₄ oxidation can consume 70% or more of the CH₄ produced in tidal wetland soils (Meronigal and Schlesinger 2002).

In tidal marshes, water table position and periods of soil inundation are controlled by hydrologic factors such as soil hydraulic conductivity, distance from open water, and soil surface elevation relative sea surface elevation. Elevation zonation subdivides tidal marshes into low marsh areas that are frequently inundated by the tides, and high marsh areas that are infrequently influenced by the tides. Water table depth influences soil oxygen availability (Epp and Chanton, 1993; Gilbert and Frenzel, 1995), and hence the potential for aerobic processes such as CH₄ oxidation (Grünfield and Brix, 1999; Meronigal and Schlesinger 2002). Soils in low marsh areas that are permanently or frequently inundated experience low rates of O₂ diffusion and sustain anaerobic environments where methanogenesis can occur (Ding et al. 2010). Tidal wetland studies have documented correlations between elevation, water level, and CH₄ emissions (Altor and Mitsch, 2006; Audet et al. 2013; Ding et al, 2010; Grünfield and Brix, 1999), implicating hydroperiod as a dominant influence on wetland CH₄ emissions.

Both plant community composition and water table depth have proven to be effective proxies for predicting CH₄ emissions in wetland ecosystems such as peatlands (Audet et al. 2013; Bubier et al. 1995; Couwenberg et al. 2001; Dias et al. 2010). Wetland plant species exhibit different tolerances to inundation (Sorrell et al. 2000; Vann and Meronigal, 2003), leading to varying plant community composition across elevation gradients (Perry and Hershner 1999). Wetland vegetation is well suited for serving as a proxy to predict CH₄ fluxes due to direct and indirect influences of plant species on labile soil organic carbon (i.e. root exudates), soil moisture, and CH₄ gas transport via plant aerenchyma tissue (Couwenberg et al. 2001). Previous studies have established direct links between plant species composition and CH₄ fluxes (Audet et al. 2013; Bhullar et al. 2014; Shäefer et al. 2011) and have used plant species composition to accurately predict CH₄ fluxes from peatlands (Couwenberg et al. 2001; Dias et al. 2010). Water table depth, as influenced by relative elevation, has also proven to be a good proxy for predicting CH₄ emissions in peatlands as water table level determines aerobic/anaerobic zones and redox states in the soil profile (Ding et al. 2010).

The objective of this study was to advance our understanding of the effects of water level and plant species composition on CH₄ emissions in brackish marshes at a site scale. We measured CH₄ fluxes in two brackish marshes on the Deal Island Peninsula on the Eastern Shore of Maryland, USA across four different strata defined by water level and plant community composition (Needelman et al. 2018). We collected field data on elevation, water

level, soil temperature, and soil pore water SO_4^{2-} , sulfides, pH and salinity, and laboratory soil incubations using field-collected soil cores to assess potential CH_4 production.

Methods

Study Area

Our study area was located on the Deal Island Peninsula in Somerset County, Maryland, USA (38.185172N, 75.906279W) (Fig. 1). It consisted of two brackish tidal marshes, one unditched (Unditched) and one that had been ditched then restored (Ditched) located in the same marsh complex (referred to as Unditched-2 and Ditched-2 in Needelman et al. 2015). Ditch plugs were installed at the Ditched site in April of 2014 by inserting a plastic polyethylene sheet vertically across the ditch approximately 50 m upstream from the tidal source and securing the plug using sediment sourced from the ditch upstream of the plug. The Ditched site had an overall lower elevation than the unditched site, and was primarily composed of *Juncus roemerianus* (black needlerush). The Unditched site had a more diverse species community including *J. roemerianus*, *Spartina patens* (salt marsh hay), *Spartina alterniflora* (smooth cordgrass), *Phragmites australis* (common reed), and *Iva frutescens* (marsh elder). Plant productivity in tidal marshes in this region include a period of senescence during the late fall through the early spring, with peak plant productivity occurring in late July through August. Soils on site consist of thick moderately to highly decomposed organic horizons overlying loamy mineral horizons; within Soil Taxonomy they classify as the Mispillion series, Loamy, mixed, euic, mesic, Terric Sulphhemists, which are common estuarine marsh soil in this area. This microtidal marsh had a diurnal tidal range of approximately 0.6 meters as measured in the adjacent tidal creek.

Design

Four strata that differed in their plant community composition and elevation, both of which are closely associated with water levels, were identified prior to the study from onsite observations and overhead satellite imagery. The strata corresponded to geographic units that may be used to estimate CH_4 emissions when engaged in site-specific carbon crediting accounting methodologies (Emmer et al., 2015). Water level variability was primarily controlled by elevation in these marshes, with lower elevations having higher water levels. Two of the strata had a plant community composition dominated by *J. roemerianus*, but differed in elevation; with one site at a “High” elevation and the other at a “Low” elevation. The High *J. roemerianus* stratum was located at the unditched site and had a mean elevation of 0.334 m relative to NAVD88, while the Low *J. roemerianus* stratum was located at the ditched site with a mean elevation 0.305 m. The two additional strata consisted of one dominated by *S. alterniflora* at a relatively low mean elevation of 0.299 m, and one dominated by *S. patens* at a relatively high mean elevation of 0.409 m. Both the Low *S. alterniflora* and High *S. patens* strata were located at the Unditched site. A representative area was selected within each stratum that included a range of elevation and plant diversity. Five sampling plots were randomly established within the representative area in each stratum, for a total of 20 plots. It should be noted that our flux measurements covered a small inference space, since they were only in representative areas of each strata and not randomly distributed across the entire marsh. Three of the strata were located within 25-50m of a tidal creek, while one (*S. alterniflora*) was located in a more central location in the marsh complex at approximately 100 m from a tidal creek. The 20 plots occupied an approximate area of 0.06 km², with each 5-plot strata encompassing an approximate area of 1,000 m².

Field Methods

We sampled monthly from April to December 2015; samples were not taken from January until March under the assumption that CH₄ production would be negligible due to low temperatures (Marsh et al. 2005). Methane flux, air temperature, and pore water concentrations of pH, SO₄²⁻, hydrogen sulfide, and CH₄ were measured at each plot. Soil temperature at 10 cm was recorded at two plots per stratum hourly during the sampling season using HOBO 8k Pendant sensors (Onset Corp., Bourne, MA). Soil temperature and water level data were not collected during the month of April because loggers were not ready for deployment until May.

Each of the 20 sample plots received a custom-fabricated square aluminum metal collar that was permanently inserted into the marsh to a depth of 10 cm nine months prior to the first sample. Flux chambers were constructed of an aluminum frame made of 2.5-cm wide angle stock covered with transparent polycarbonate plastic film. Chambers were placed on top of the collar about 10 minutes prior to sampling. Chambers were equipped with a closed-cell neoprene strip on the top and bottom, which when clamped to the collar assured an airtight seal (Yu et al. 2013). The taller plants in the *J. roemerianus* strata were accommodated without damaging plant stems by stacking chambers. Opaque chamber lids with a sampling port were clamped to the top of the chamber to complete the seal. Chambers had a height of 69.5 cm and an interior length and width of 49.5 cm, yielding a total volume of 0.17 m³ for single chambers and 0.34 m³ for double chambers. In order to prevent the weight of the observer from causing ebullition due to soil compression (Sorrell et al. 2013), each plot had a 3m wooden boardwalk suspended above the soil surface by PVC legs for approaching the flux collar.

Methane flux samples were collected over a 1-hr period from the 5 replicate flux plots in a given strata. An initial sample was taken immediately after each chamber was sealed with four subsequent samples taken at approximately 15-min intervals for a total of 25 samples (5 per plot) over the 1-hr period. Using a 30 mL syringe, the sampling port was opened and then expelled back into the chamber three to five times before each sample was taken. Each 18 mL air sample was withdrawn from the chamber and injected into a N₂-flushed 12-mL Exetainer vial with rubber septum until analysis (Yu et al. 2013). Air temperature within the sampling chamber was recorded upon the collection of each flux sample from thermometers affixed to the interior of each chamber.

Porewater samples were taken at 10 cm depth using a porewater sipper and syringe (Fisher et al. 2013) and analyzed for pore water CH₄, hydrogen sulfide (unfiltered), pH (unfiltered), salinity (unfiltered), and SO₄²⁻ (filtered through a 0.45- μ m filter) as described by Keller et al. 2009. Porewater CH₄ was collected by withdrawing 15 mL of pore water, after which 15 mL of ambient air was drawn into the syringe and the syringe capped. The sample was then agitated for 1-2 minutes for the CH₄ to be stripped into the drawn air, the stripped water was expelled, and the gas sample was stored in N₂-flushed Exetainers for analysis (Keller et al. 2009). Hydrogen sulfide samples were diluted in a 1:1 ratio of sample to sulfide antioxidant buffer in the field to prevent sulfide volatilization and oxidation (Koch et al. 1990). Hydrogen sulfide and pH samples were analyzed the same day as sample collection; salinity was analyzed within two weeks in the laboratory using a YSI Model 3100 conductivity meter; and all other pore water samples were frozen and analyzed during the winter of 2016.

Additional data were collected during the July 2015 sampling event, which was predicted to be during a peak CH₄ emission period. We collected porewater at 20 cm depth in addition to 10 cm and analyzed it for the same analytes excluding CH₄ but including ferrous iron (Fe²⁺).

Water level was measured at each stratum in order to determine water levels at the time of sampling and antecedent water level conditions during the two-week period leading up to the sampling period. Water level recorders (HOBO U20-L, Onset Corp, Bourne, MA) were installed adjacent to the chamber transects to continuously record water levels

in the marsh; one was also installed in the tidal creek adjacent to the field site during the field season. We deployed two water level loggers in each stratum, except for the low water table *J. roemerianus* stratum, which had one water level logger due to its small area relative to the other strata. Barometric pressure was collected onsite to correct the unvented loggers (HOBO U20-L, Onset Corp, Bourne, MA). We surveyed the elevation of all 20 plots and water level logger locations using a Real-time Networking Global Positioning System (RTN GPS) unit, which provides elevation data with approximately 2 cm accuracy (<http://www.keynetgps.com>).

Soil cores were collected during the July sampling event and analyzed for potential anaerobic CH₄ and CO₂ production. Cores were collected from approximately 0-40 cm depth using a circular metal gouge corer. The corer was inserted into the marsh, with careful attention paid to minimize compaction of the soft peat. The core was removed and cut at a depth of 20 cm, yielding two depth increments per plot. Cores were placed into sample bags in which as much air as possible was removed. The cores were then placed in a cooler with ice and transported back to the laboratory, where each bag was flushed three times with nitrogen gas to remove oxygen, stored on ice during transport, and placed in a 4 °C cold room until processing. Water for these incubations was collected from the bore hole from which the core was removed, stored on ice for transport back to the lab, purged with nitrogen gas to remove oxygen before being sealed and placed in a cold room at 4 °C. Soil cores and water samples were stored in the cold room within 8 hours of their collection and incubated within 5 days.

Laboratory Analyses

Flux chamber headspace samples were measured on a Varian 450 gas chromatograph using a Combi-Pal autosampler and corrected for dilution of 18 ml of sample into 12 ml of N₂ in the Exetainer. Flux rate was calculated as the linear increase in headspace [CH₄] over time based on measurements of chamber temperature and volume and assuming atmospheric pressure (n=147 fluxes). The linear slope was calculated in Excel using the Regression function. Data points were excluded from the regression if they indicated an ebullition event (large spike in [CH₄]) or an Exetainer leak (large drop in [CH₄]). Most fluxes were calculated from five points, but never from fewer than three points. No flux measurements were excluded based on arbitrary regression R² or p-value limits but fluxes were excluded in several cases where an ebullition event or leak was large compared to the CH₄ flux rate rate (n=17). In cases where there was no significant trend in headspace [CH₄] and no evidence of ebullition or leaks the flux was assigned a value of zero. Because most of the excluded fluxes were collected during periods of low CH₄ flux they had relatively little influence on the annual flux calculation.

To estimate annual emissions averaged rates from each measurement campaign. We assumed that the fluxes in the unsampled months of January, February, and March were equal to our observed values from April. The twelve monthly values were averaged and converted to annual units. While this method likely overestimated CH₄ emissions, overestimation is the conservative and therefore preferable approach for carbon credit accounting (Needelman et al. 2018).

Hydrogen sulfide was determined with a Lazar Laboratory model 146S sulfide electrode. Sulfide antioxidant buffer was prepared the day before sample collection with deoxygenated (N₂-stripped) distilled water, sodium salicylate, sodium hydroxide, and ascorbic acid according to Koch et al. (1990). A standard curve created from a serial dilution of a Na₂S/buffer solution prepared on the day of each sampling event and readings were complete within 4 hours of sample collection. The pH was measured with a YSI Pro Plus (<https://www.ysi.com>) pH meter calibrated with standards at pH 7.0 and 10.0. Salinity was measured with a calibrated conductivity/salinity electrode. The remaining analytes SO₄²⁻ and Fe²⁺ were quantified at the Chesapeake Biological Laboratory, Solomons Island, MD. Reduced

iron was analyzed according to EPA method 200.1 and SO_4^{2-} was analyzed according to the National Environmental Methods Index Standard Methods: 4110B for ions in water by ion chromatography (www.nemi.gov).

Sulfate depletion was calculated by assuming the SO_4^{2-} concentration in the absence of sulfate reduction was that of full-strength marine seawater (Canfield 2004) diluted to the observed salinity of our sample. We then divided our observed SO_4^{2-} concentration by this expected concentration of SO_4^{2-} to estimate the consumption of SO_4^{2-} (i.e. depletion) in the porewater.

Soil cores collected for incubations were removed from cold storage within 5 days of collection and placed into an anaerobic hood containing a N_2/H_2 mixture (Magonigal and Schlesinger 2002). Two sections were removed from each core, yielding a 8-12 cm depth sample and a 28-32 cm depth sample. The outer 10 mm (approximately) of the resulting disks were removed to expose the center of the core, which was assumed to have had minimal O_2 exposure from collection to processing. We then removed as many live roots as feasible. Five grams of wet soil material was placed in a 35-mL serum bottle with 5 mL of the degassed water from the core hole. Headspace samples of 0.5 mL were injected directly into a Shimadzu gas chromatograph with a flame ionization detector for CH_4 or a LI-COR LI-7000 (LiCor, Lincoln, NE) for CO_2 . Methanogenesis generally slowed dramatically after 5 days, so our calculations of potential anaerobic CH_4 production rates are based on incubation days 1-5.

Statistical Analysis

Data were analyzed using SAS 9.3 (SAS Institute Inc. Cary, NC). Regression analyses were performed on flux data using Proc Reg to determine the slope of CH_4 or CO_2 concentration change over time. All variables were evaluated for normality using PROC UNIVARIATE and those that required transformation were log transformed to improve normality. Parameters transformed were: CH_4 flux, pore water hydrogen sulfide concentration, pore water SO_4^{2-} concentration, and pore water CH_4 concentrations. All parameters were analyzed using PROC MIXED with strata and month in the model statement with repeated measures. Post-hoc Tukey mean comparisons were used with $\alpha = 0.05$ used to indicate significance.

Results

Antecedent Water Levels

Water level data collected during the 2 weeks prior to and during sampling events varied significantly between strata ($p < 0.0001$) and month ($p < 0.0001$). Water levels of the *S. patens* stratum was significantly lower than all other strata, with a mean of 9 cm below the soil surface, while the other strata had similar mean water levels approximately 1 cm below the soil surface (Table 1). We were unable to test for a strata by month interactive effect because only two wells were deployed in each strata (and only one in the High *J. roemerianus* stratum); however, *S. patens* had a lower mean water level in all months (Derby 2016). Mean water levels were highest in July, August, and October and lowest in May, June, September, and December.

Methane Emissions and Porewater Chemistry by Strata

Average CH_4 flux over the study varied significantly between strata ($p < 0.0001$) and month ($p < 0.0001$), and had a significant interactive effect ($p = 0.018$). Methane emissions from the four strata ranked *S. alterniflora* >> High *J. roemerianus* > Low *J. roemerianus* = *S. patens* (Table 1). Mean CH_4 emissions from the *S. alterniflora* stratum was

2.72 times greater than the next highest CH₄ emitter (Table 1) despite similar inundation regimes among the three highest emitting strata. High CH₄ emissions in the *S. alterniflora* stratum coincided with significantly higher porewater CH₄ concentrations and lower SO₄²⁻ concentrations than the other three strata (Table 1). Porewater salinity in the *S. alterniflora* stratum was similar to other strata (Table 1) suggesting that relatively low SO₄²⁻ concentrations were due to high rates of SO₄²⁻ consumption. Indeed, SO₄²⁻ was depleted by 61% in the *S. alterniflora* stratum. Sulfate depletion was significantly different between strata ($p < 0.0001$) and month ($p < 0.0001$) and ranked *S. alterniflora* = High *J. roemerianus* > Low *J. roemerianus* > *S. patens*. Sulfate depletion rates in the *S. alterniflora* stratum were over two times greater than those seen in low *J. roemerianus* (Table 1). Low SO₄²⁻ concentrations in the *S. alterniflora* stratum were accompanied by significantly higher amounts of hydrogen sulfide (72 mg L⁻¹) as compared to the other three strata (all < 20 mg L⁻¹).

The *S. patens* stratum had the lowest mean porewater CH₄ concentrations and SO₄²⁻ depletion rates, with only 1.4% SO₄²⁻ depleted. *S. patens* also exhibited the highest mean concentrations of reduced iron (i.e. ferrous iron, Fe²⁺) during the single campaign when it was measured, with 72 mg L⁻¹ of ferrous iron in porewater collected 10 cm below the soil surface (Table 1). None of the other three strata had reduced iron porewater concentrations exceeding 0.8 mg L⁻¹. *S. patens* also had a significantly lower mean porewater salinity (12.3 ppt) than the other three strata, which ranged in mean salinity from 14.2 to 14.8 ppt. The High and Low *J. roemerianus* and *S. alterniflora* strata were not significantly different in porewater iron concentrations or salinity (Table 1).

The difference in elevation between the two *J. roemerianus* strata was not highly apparent in the water level and CH₄-related attributes we measured. The two strata were not significantly different from one another in, salinity, SO₄²⁻ concentrations, sulfide concentrations, percent SO₄²⁻ depleted, porewater CH₄ concentrations, and reduced iron concentrations (Table 1).

The highest CH₄ emissions were observed in July, August, and September; the lowest were observed in April, November, and December (Fig. 2). Significant strata by month interactions were observed in May, June, and September. In May and June, *S. alterniflora* was not significantly different from any strata other than Low *J. roemerianus*; all other strata were not significantly different from one another. In September, *S. alterniflora* was not significantly different from any strata other than *S. patens*; all other strata were not different from one another. No significant within-month differences were observed in April, July, August, October, November and December (Derby 2016). Porewater CH₄ exhibited a similar seasonal trend as CH₄ emissions, with the highest concentrations in the months July through November (Derby, 2016).

Anaerobic incubations

Surficial soils (8-12 cm) from the *S. patens* stratum had the lowest CH₄ production, highest CO₂ production and a significantly higher ratio of CO₂:CH₄ production (ratio=993) than the other strata. At the other extreme was the *S. alterniflora* stratum which produced substantially more CH₄ and less CO₂ than the other strata, and therefore had the lowest CO₂:CH₄ ratio (ratio=40) among the four sites (Table 2). The two *J. roemerianus* strata fell in between these extremes with a CO₂:CH₄ ratio of about 200 (Table 2), though there were no significant differences in CO₂:CH₄ ratio between these strata and *S. alterniflora*. The 10 cm incubations produced significantly more CH₄ and CO₂ than those from 30 cm depth ($p = 0.04$, data not shown, Derby 2016).

Discussion

Methane emissions varied by plant community type and hydrogeomorphic setting, suggesting that these variables are useful for dividing tidal marshes into strata to optimize the costs of sampling effort with the need to reduce parameter uncertainty when estimating CH₄ emissions. Mean emissions across strata ranked as *S. alterniflora* > High-elevation *J. roemerianus* > Low-elevation *J. roemerianus* > *S. patens* (Table 1). There is a need to understand the mechanisms that lead to such differences among strata in order to advance proxies and models of CH₄ emissions from tidal wetlands.

We attribute the low emissions from the *S. patens* stratum to a combination of relatively low water levels, shallow rooting depth, and higher mineral inputs, all of which have the capacity to suppress CH₄ production and promote CH₄ oxidation. It is well established that CH₄ production is suppressed by alternative electron acceptors such as O₂, Fe(III) and SO₄²⁻ (Holm et al. 2016; Neubauer et al. 2005; Poffenbarger et al, 2011; Roden and Wetzel, 2003). Relatively high O₂ flux into the upper soil surface (0-10 cm depth) would be favored by both the relatively thick aerobic zone (i.e. deeper water-table) and shallow distribution of root biomass that is characteristic of *S. patens* communities (Bernal et al. 2016). Mean antecedent water depth was 9 cm deep in this stratum compared to other strata with water levels near the soil surface.

Because the majority of *S. patens* roots are in the top 10 cm of the soil profile (Windham 2001), it is likely that root oxygen loss was also an O₂ source in this community. O₂ suppresses methanogenesis via electron donor competition by two mechanisms, directly as an electron acceptor for aerobic bacteria and indirectly by regenerating poorly crystalline iron oxides on the root surface (i.e. iron plaque). Root-deposited iron plaque is rapidly consumed by iron-reducing bacteria (Weiss et al. 2003; Weiss et al. 2004), suppressing both SO₄²⁻ reduction and methanogenesis (Neubauer et al. 2005). This mechanism is consistent with our observation that the *S. patens* community had dramatically higher concentrations of reduced iron (measured only in July) and SO₄²⁻ than the other strata, and the lowest SO₄²⁻ depletion (Table 1). The close proximity of this site to the tidal creek may have also allowed for greater mineral inputs during flooding events to support iron cycling. Finally, if rates of methanotrophy are O₂-limited as studies suggest (King, 1996; Lombardi et al, 1997; Megonigal and Schlesinger 2002), then relatively high rates of CH₄ oxidation would be expected to further decrease the amount of CH₄ emitted through passive diffusion through plants, as documented in other tidal wetlands (Megonigal and Schlesinger 2002).

The *S. alterniflora* stratum had the highest average CH₄ emissions and porewater chemistry that differed from the *S. patens* community in several respects (Table 1). The *S. alterniflora* stratum was in the center of the marsh complex (Fig. 1). Because hydrologic fluxes generally decrease with increasing distance from the tidal creek (Jordan et al. 1985), it is likely that soils in the *S. alterniflora* stratum had relatively slow rates of advection compared to the other strata. Indeed, water table depths decreased relatively slowly after floods in the *S. alterniflora* stratum (Derby 2016). This hydrologic difference likely decreased rates of advective transport of O₂ and SO₄²⁻ to the soil profile to replenish these electron acceptors. We propose that the relatively low inputs of SO₄²⁻ to the *S. alterniflora* stratum led to SO₄²⁻ limitation of SO₄²⁻ reduction rates, allowing methanogens to compete more effectively with sulfate-reducing bacteria for electron donors. Porewater evidence supporting this interpretation includes low concentrations of Fe²⁺, high SO₄²⁻ depletion, and high concentrations of both hydrogen sulfide and CH₄. This interpretation is also consistent with the results of the July anaerobic incubations showing that the CO₂:CH₄ ratio was lowest in *S. alterniflora* soils and highest in *S. patens* soils. Because aerobic respiration, sulfate reduction, and iron reduction generate CO₂ rather than

CH₄, these data suggest that methanogens had relatively little competition for organic carbon and H₂ in the *S. alterniflora* stratum.

Water table depths in the High and Low *J. roemerianus* strata were similar to the *S. alterniflora* stratum but metrics related to CH₄ emissions were consistently lower than the *S. alterniflora* and higher than the *S. patens* strata, namely CH₄ emissions, porewater SO₄²⁻ concentrations, and anaerobic incubation CO₂:CH₄ ratio. The difference in CH₄ emissions between *S. alterniflora* and the *J. roemerianus* strata cannot be explained by water levels, salinity, pH, or reduced iron because they were not significantly different between these strata. For example, mean water table depth in the *S. alterniflora* stratum and the two *J. roemerianus* strata were within 1 cm of each other, while emissions were 2.5 times greater in the *S. alterniflora* stratum. The *J. roemerianus* strata were closer to the tidal creek and presumably CH₄ production was not limited by tidal inputs of SO₄²⁻ supply as we suspect was the case in the *S. alterniflora* stratum. However, the relatively low CH₄ emissions in the *J. roemerianus* strata compared to *S. alterniflora* may also have been related to plant traits that regulate CH₄ emissions by influencing the balance between CH₄ production and oxidation (Sutton-Grier and Megonigal 2011, Mueller et al. 2020), which itself is influenced by traits that affect CH₄ transport through plants (Komiya et al. 2020). Mueller et al. (2020) proposed that plant traits vary across species such that some push the balance between these opposing processes toward net CH₄ production while others favor net CH₄ oxidation. We hypothesize that among the dominant species present at the site, *S. alterniflora* favors net CH₄ production while *J. roemerianus* favors net CH₄ oxidation, and that the lower CH₄ emissions rates in the *J. roemerianus* strata may have been due to relatively high rates of root oxygen loss by *J. roemerianus*. This interpretation is supported in part by data on porewater [SO₄²⁻], which mediates the outcome of competition between sulfate-reducing bacteria and methanogens. Porewater [SO₄²⁻] was 4.5 mM in the *S. alterniflora* stratum but exceeded 6 mM in the *J. roemerianus* strata where CH₄ emissions rates were relatively low. Although this difference in porewater [SO₄²⁻] seems small, the relationship between these variables is non-linear and displays a threshold value above which sulfate reduction dominates and below which methanogenesis dominates (Megonigal et al. 2004). There is a general lack of CH₄-relevant porewater data for coastal wetlands, but a robust record from a brackish marsh located 100 km from the study site in Chesapeake Bay observed that porewater [CH₄] declined abruptly when SO₄²⁻ concentrations exceeded approximately 4 to 6 mM (Keller et al. 2009). We propose that the *S. alterniflora* and *J. roemerianus* strata fell on opposite sides of this threshold.

Although floodwater salinity can be an effective proxy for porewater SO₄²⁻ concentrations when comparing sites across large spatial scales (Poffenbarger et al. 2011), our data demonstrate the limitations of using the salinity proxy at local scales. Variation among strata in SO₄²⁻ concentration and SO₄²⁻ depletion may have been caused by variation in rates of SO₄²⁻ transport from tidal floodwater into soils, sulfide oxidation related to O₂ diffusion from water table depth or root O₂ loss, or primary production (i.e. carbon availability). We cannot distinguish among these mechanisms with the present dataset. Sulfate depletion was a better indicator of CH₄ flux than salinity or SO₄²⁻ concentration alone and may prove to be a superior proxy for CH₄ emissions in tidal brackish marshes.

Within our strata, CH₄ production did not strictly follow the conventional interpretation that differences in the free energy yield among competing microbial respiration processes means that just one process dominates microbial respiration at a time, with methanogenesis expected to occur only when all other electron acceptors are fully (or nearly) depleted. Our data suggest that peak sulfate reduction activity was occurring concurrently with peak CH₄ production in the *S. alterniflora* stratum which had both the highest mean pore water hydrogen sulfide concentrations

and the lowest SO_4^{2-} concentrations. We attribute this to spatial variation in electron donors and acceptors that creates microsites of high SO_4^{2-} depletion and methanogenesis. Microsites have been shown to produce small amounts of CH_4 in upland forested systems, originally thought to be too dry and too aerobic to produce this greenhouse gas (Meronigal and Guenther 2008). Microsites also exist in tidal marsh soils due to local (i.e. rhizosphere) consumption of electron acceptors at rates faster than they can be replenished (e.g. Rabenhorst et al. 2010).

Carbon offset implications

Salinity is a useful proxy for CH_4 emissions from tidal marshes with salinity regimes > 18 ppt because CH_4 emissions are low compared to their soil carbon sequestration rates and variation among marshes is small. However, tidal brackish marshes at lower salinities may emit enough CH_4 to offset a significant fraction of their radiative cooling and variation in CH_4 emissions among marshes within a given salinity regime is large. Such uncertainty is accommodated in carbon offset programs such as the Verified Carbon Standard by requiring the project to estimate CH_4 emissions by direct monitoring or by using published data, a model, or a proxy that can be demonstrated to be valid for the project site (Needelman et al. 2018). Stratification by hydrology and vegetation characteristics may provide a more effective proxy than salinity to estimate CH_4 emissions. Stratification also offers a means to constrain variability within direct monitoring schemes.

We compared the CH_4 offset estimates produced through our direct measurements to those predicted by the salinity-based proxy equation of Poffenbarger et al. (2011). For this comparison we assumed a single soil carbon sequestration rate of $1.46 \text{ Mg C ha}^{-1} \text{ yr}^{-1}$ for all strata in the marsh complex, which is the default rate in the carbon crediting methodology of Emmer et al. (2015), derived as the median value from Chmura et al. (2003). In three of the four strata the salinity proxy overestimated emissions by about 20%, 40%, and 80% (Table 3). Overestimation is preferable to underestimation to avoid awarding carbon credits that exceed actual greenhouse gas benefits, but overestimation decreases the financial viability of carbon offset projects. These errors caused the positive radiative balance to be underestimated by just 4-8% in the *J. roemerianus* strata, suggesting that incorporating vegetation and hydrology proxies would not be a substantial improvement over the salinity proxy alone. In addition, the cost of *in situ* emission measurements would not be rewarded by a meaningful increase in carbon credits in the *J. roemerianus* strata. However, the positive radiative balance of the *S. patens* stratum was underestimated by $>20\%$, suggesting that a proxy based on vegetation and hydrology would improve CH_4 emission estimates, and that the expense of collecting *in situ* data may be worthwhile. The salinity proxy overestimated CH_4 emissions in the *S. alterniflora* stratum where the positive radiative balance was overestimated by about 25% (Table 3), indicating that a salinity-based proxy would award carbon credits exceeding actual climate benefits in this stratum. Improved proxies are needed to incentivise carbon offsets projects by reducing monitoring costs while ensuring that projects achieve estimated climate benefits.

Our results suggest that proxies for CH_4 emissions from tidal wetlands with salinity regimes < 18 ppt can be improved by incorporating metrics related to hydrology such as flooding frequency and duration or the position of the soil surface relative to the tidal frame, and metrics related to plant traits such as species identity, plant functional type, or biomass. Ideally these metrics would be identifiable at high spatial resolution for low cost, such as through analysis of freely available remote sensing data. Currently there is no widely accepted method to remotely sense surface water salinity, but robust methods exist for remote sensing of plant cover, biomass, primary production, and elevation (Buffington et al. 2016, Byrd et al. 2018, Feagin et al. 2020).

Conclusions

Tidal wetland restoration and conservation projects have the potential to mitigate greenhouse gas emissions to the atmosphere and generate carbon credits, but a better understanding of the factors influencing wetland CH₄ emissions in brackish and freshwater systems (salinity < 18 ppt) is required to lower carbon crediting project costs and estimation uncertainty at site-specific scales. We found significantly different methane emission rates across four strata defined by hydrology and plant community composition that otherwise had similar salinity regimes. We inferred that they deviated from the rates predicted by a salinity proxy due to processes that regulate the availability of competing terminal electron acceptors such as O₂, Fe(III), and SO₄²⁻ and due to plant traits that regulate CH₄ emissions. Low CH₄ emission rates in the high-elevation *S. patens* stratum was attributed to relatively high inputs of Fe(III) through tidal inputs and oxidation of reduced iron and O₂ through diffusion across the soil surface and root O₂ loss, both of which maintain high [SO₄²⁻] by suppressing microbial reduction of SO₄. By contrast, the *S. alterniflora* stratum was relatively isolated from tidal inputs of Fe(III) and SO₄, so it produced large amounts of CH₄ once SO₄²⁻ had been sufficiently depleted. The greater CH₄ emissions from the low-elevation *S. alterniflora* stratum than the low and high elevation *J. roemerianus* strata could not be explained by water table depth, salinity, pH, and [Fe²⁺], suggesting an important role for plant traits such as root O₂ loss regulating CH₄ emissions at local scales. The mechanisms driving these patterns were not measured directly but likely involve variations in rates of SO₄²⁻ diffusion from tidal floodwater based on site hydrologic connectivity; rates of sulfide oxidation as influenced by O₂ diffusion or root O₂ loss; and differing plant primary productivity. Our results suggest that stratification by vegetation and hydrologic setting improves estimates of CH₄ emissions from tidal marshes. Our findings illustrate the need to better understand controls over CH₄ emissions at site-specific scales to improve carbon sequestration offset estimates in brackish and freshwater coastal wetland ecosystems.

Declarations

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Conflicts of interest

None.

Availability of data and material (data transparency)

All data from this manuscript will be made available through the Coastal Carbon Research Coordination Network.

Code availability (software application or custom code)

Not applicable

References

- Al-Haj, AN, Fulweiler, RW (2020) A synthesis of methane emissions from shallow vegetated coastal ecosystems. *Glob Change Biol* 26: 2988– 3005. <https://doi.org/10.1111/gcb.15046>.
- Altor, AE, Mitsch, WJ (2006) Methane flux from created riparian marshes: Relationship to intermittent versus continuous inundation and emergent macrophytes. *Ecological Engineering* 28(3):224-234. doi:10.1016/j.ecoleng.2006.06.006.
- Audet, J, Johansen, JR, Andersen, PM, Baattrup-Pedersen, A, Brask-Jensen, KM, Elsgaard, L, Hoffmann, CC (2013) Methane emissions in Danish riparian wetlands: Ecosystem comparison and pursuit of vegetation indexes as predictive tools. *Ecological Indicators* 34:548-559. doi:10.1016/j.ecolind.2013.06.016.
- Bernal, B, Megonigal, JP, Mozdzer, TJ (2016) An invasive wetland grass primes deep soil carbon pools. *Global Change Biology* 23(5):2104-2116. doi:10.1111/gcb.13539.
- Bhullar, GS, Edwards, PJ, & Venterink, HO (2014) Influence of Different Plant Species on Methane Emissions from Soil in a Restored Swiss Wetland. *PLoS ONE*, 9(2). doi:10.1371/journal.pone.0089588.
- Bubier, JL, Moore, TR, Bellisario, L, Comer, NT, Crill, PM (1995) Ecological controls on methane emissions from a Northern Peatland Complex in the zone of discontinuous permafrost, Manitoba, Canada. *Global Biogeochemical Cycles* 9(4):455-470. doi:10.1029/95gb02379.
- Buffington KJ, Dugger BD, Thorne KM, Takekawa JY (2016) Statistical correction of lidar-derived digital elevation models with multispectral airborne imagery in tidal marshes. *Remote Sens Environ* 186:616–625.
- Byrd KB, Ballanti L, Thomas N, Nguyen D, Holmquist JR, Simard M, Windham-Myers L (2018) A remote sensing-based model of tidal marsh aboveground carbon stocks for the conterminous United States. *ISPRS J Photogramm Remote Sens* 139:255–271.
- Calhoun, A, King, GM (1997) Regulation of root-associated methanotrophy by oxygen availability in the rhizosphere of two aquatic macrophytes. *Applied and Environmental Microbiology* 63(8):3051-3058. doi:10.1128/aem.63.8.3051-3058.1997.
- Canfield DE (2004) The evolution of the Earth surface sulfur reservoir. *Am J Sci* 304:839–861. doi: 10.2475/ajs.304.10.839.
- Chmura GL, Anisfeld SC, Cahoon DR, Lynch JC (2003) Global carbon sequestration in tidal, saline wetland soils. *Glob Biogeochem Cycles* 17:1111. doi: 10.1029/2002GB001917.
- Colmer, TD (2003) Long-distance transport of gases in plants: A perspective on internal aeration and radial oxygen loss from roots. *Plant, Cell & Environment* 26(1):17-36. doi:10.1046/j.1365-3040.2003.00846.x.
- Couwenberg, J, Thiele, A, Tanneberger, F, Augustin, J, Bärtsch, S, Dubovik, D, Liashchynskaya, N, Michaelis, D, Minke, M, Skuratovich, A, Joosten, H (2011) Assessing greenhouse gas emissions from peatlands using vegetation as a proxy. *Hydrobiologia* 674(1):67-89. doi:10.1007/s10750-011-0729-x.
- Crooks S, Herr D, Tamelander J, et al (2011) Mitigating climate change through restoration and management of coastal wetlands and near-shore marine ecosystems: challenges and opportunities.

Derby, K (2016) Methane emissions from a tidal brackish marsh on Maryland's eastern shore and the factors impacting them. Master's Thesis, University of Maryland.

Drum DOI: <https://doi.org/10.13016/M2PV6R>

Dias, AT, Hoorens, B, Logtestijn, RS, Vermaat, JE, Aerts, R (2010) Plant Species Composition Can Be Used as a Proxy to Predict Methane Emissions in Peatland Ecosystems After Land-Use Changes. *Ecosystems* 13(4):526-538. doi:10.1007/s10021-010-9338-1.

Ding, W, Cai, Z, Tsuruta, H (2005) Plant species effects on methane emissions from freshwater marshes. *Atmospheric Environment* 39(18):3199-3207. doi:10.1016/j.atmosenv.2005.02.022.

Ding, W, Zhang, Y, Cai, Z (2010) Impact of permanent inundation on methane emissions from a *Spartina alterniflora* coastal salt marsh. *Atmospheric Environment* 44(32):3894-3900. doi:10.1016/j.atmosenv.2010.07.025.

Emmer IM, Unger M von, Needelman B, Crooks, S, Emmett-Mattox, S (2015) Coastal Blue Carbon in Practice: A Manual for Using the VCS Methodology for Tidal Wetland and Seagrass Restoration. *Restore America's Estuaries*, Arlington, VA., v. 1, 82 p. Available at https://www.estuaries.org/images/rae_coastal_blue_carbon_methodology_web.pdf

Emmer, I.M., M. von Unger, B.A, Needelman, S. Cooks, and S. Emmett-Mattox. 2015. Coastal Blue Carbon in Practice: A Manual for Using the VCS Methodology for Tidal Wetland and Seagrass Restoration. .

Emmer IM, Needelman BA, Emmett-Mattox S, Crooks S, Megonigal JP, Myers D, Oreska MPJ, McGlathery KJ (2020a) Estimation of Baseline Carbon Stock Changes and Greenhouse Gas Emissions in Tidal Wetland Restoration and Conservation Project Activities (BL-TW). VCS Module VMD0050, v 1.0. Verra (Verified Carbon Standard): Washington, D.C.

Emmer IM, Needelman BA, Emmett-Mattox S, Crooks S, Megonigal JP, Myers D, Oreska MPJ, McGlathery KJ (2020b) Methods for Monitoring of Carbon Stock Changes and Greenhouse Gas Emissions and Removals in Tidal Wetland Restoration and Conservation Project Activities (M-TW). VCS Module VMD0051, v 1.0, Verra (Verified Carbon Standard): Washington, D.C.

Epp, MA, Chanton, JP (1993) Rhizospheric methane oxidation determined via the methyl fluoride inhibition technique. *Journal of Geophysical Research: Atmospheres* 98(D10): 18413-18422. doi:10.1029/93jd01667.

Feagin RA, Forbrich I, Huff TP, Barr JG, Ruiz-Plancarte J, Fuentes JD, Najjar RG, Vargas R, Vázquez-Lule A, Windham-Myers L, Kroeger KD, Ward EJ, Moore GW, Leclerc M, Krauss KW, Stagg CL, Alber M, Knox SH, Schäfer KVR, Bianchi TS, Hutchings JA, Nahrawi H, Noormets A, Mitra B, Jaimes A, Hinson AL, Bergamaschi B, King JS, Miao G (2020) Tidal Wetland Gross Primary Production Across the Continental United States, 2000–2019. *Global Biogeochemical Cycles* 34.

Garcia J-L, Patel BKC, Ollivier B (2000) Taxonomic, Phylogenetic, and Ecological Diversity of Methanogenic Archaea. *Anaerobe* 6:205–226. doi: 10.1006/anae.2000.0345.

Gilbert, B, Frenzel, P (1995) Methanotrophic bacteria in the rhizosphere of rice microcosms and their effect on porewater methane concentration and methane emission. *Biology and Fertility of Soils* 20(2):93-100. doi:10.1007/bf00336586.

- Grünfeld, S, Brix, H (1999) Methanogenesis and methane emissions: Effects of water table, substrate type and presence of *Phragmites australis*. *Aquatic Botany* 64(1):63-75. doi:10.1016/s0304-3770(99)00010-8.
- Holm GO, Perez BC, McWhorter DE, et al (2016) Ecosystem Level Methane Fluxes from Tidal Freshwater and Brackish Marshes of the Mississippi River Delta: Implications for Coastal Wetland Carbon Projects. *Wetlands*. doi: 10.1007/s13157-016-0746-7.
- Holmquist, J, Windham-Myers, L, Bernal, B, Byrd, KB, Crooks, S, Gonnee, ME, Herold, N, Knox, SH, Kroeger, K, McCombs, J, Megonigal, JP, Meng, L, Morris, JT, Sutton-Grier, AE, Troxler, TG, Weller, D (2018) Uncertainty in United States coastal wetland greenhouse gas inventorying. *Environmental Research Letters* 13: 115005. <https://doi.org/10.1088/1748-9326/aae157>.
- Jespersen, DN, Sorrell, BK, Brix, H (1998) Growth and root oxygen release by *Typha latifolia* and its effects on sediment methanogenesis. *Aquatic Botany* 61(13):165-180. doi:10.1016/s0304-3770(98)00071-0.
- Jordan, T.E. and D.L. Correll (1985). Nutrient chemistry and hydrology of interstitial water in brackish tidal marshes of Chesapeake Bay. *Estuarine, Coastal and Shelf Science*, 21(1): 45-55.
- Keller JK, Wolf AA, Weisenborn PB, et al (2009) Elevated CO₂ affects porewater chemistry in a brackish marsh. *Biogeochemistry* 96:101–117. doi: 10.1007/s10533-009-9347-3.
- King, GM (1996) In Situ Analyses of Methane Oxidation Associated with the Roots and Rhizomes of a Bur Reed, *Sparganium eurycarpum*, in a Maine Wetland. *Applied and Environmental Microbiology*, 62:4548-4555. doi:10.1128/aem.62.12.4548-4555.1996.
- Koch MS, Mendelssohn IA, McKee KL (1990) Mechanism for the hydrogen sulfide- induced growth limitation in wetland macrophytes. *Limnol Oceanogr* 35:399– 408. doi: 10.4319/lo.1990.35.2.0399.
- Koebisch F, Glatzel S, Jurasinski G (2013) Vegetation controls methane emissions in a coastal brackish fen. *Wetl Ecol Manag* 21:323–337. doi: 10.1007/s11273-013- 9304-8.
- Komiya, S, Yazaki, T, Kondo, F, Katano, K, Lavric, JV, Mctaggart, I, Pakoktom, T, Siangliw, M, Toojinda, T, Noborio, K (2020) Stable Carbon Isotope Studies of CH₄ Dynamics Via Water and Plant Pathways in a Tropical Thai Paddy: Insights Into Diel CH₄ Transportation. *Journal of Geophysical Research: Biogeosciences*, 125:9. doi:10.1029/2019jg005112.
- Le Mer, J, Roger, P (2001) Production, oxidation, emission and consumption of methane by soils: A review. *European Journal of Soil Biology* 37(1):25-50. doi:10.1016/s1164-5563(01)01067-6.
- Lombardi, AE, Epp, MA, Chanton, JP (1997) Investigation of the methyl fluoride technique for determining rhizospheric methane oxidation. *Biogeochemistry*. 36:153-172.
- Marsh, AS, DP Rasse, BG Drake, and JP Megonigal (2005) Effect of elevated CO₂ on carbon pools and fluxes in a brackish marsh. *Estuaries* 28:694-704.
- Megonigal JP, Guenther AB (2008) Methane emissions from upland forest soils and vegetation. *Tree Physiol* 28:491–498. doi: 10.1093/treephys/28.4.491.

- Megonigal, JP, ME Hines, and PT Visscher (2004) Anaerobic Metabolism: Linkages to Trace Gases and Aerobic Processes. Pages 317-424 in Schlesinger, W.H. (Editor). *Biogeochemistry*. Elsevier-Pergamon, Oxford, UK.
- Megonigal JP, Schlesinger WH (2002) Methane-limited methanotrophy in tidal freshwater swamps. *Glob Biogeochem Cycles* 16:1088. doi: 10.1029/2001GB001594.
- Moor, H, Rydin, H, Hylander, K, Nilsson, MB, Lindborg, R, Norberg, J (2017) Towards a trait-based ecology of wetland vegetation. *Journal of Ecology* 105:1623-1635. doi:10.1111/1365-2745.1273.
- Mueller, P, Mozdzer, TJ, Langley, JA, Aoki, LR, Noyce, GL, Megonigal, JP (2020) Plant species determine tidal wetland CH₄ response to sea level rise. *Nature Communications*. <https://doi.org/10.1038/s41467-020-18763-4>.
- Nahlik, A., Fennessy, M. Carbon storage in US wetlands (2016) *Nature Communications* 7, 13835. <https://doi.org/10.1038/ncomms13835>.
- Needelman, BA, IM Emmer, S Emmett-Mattox, S Crooks, JP Megonigal, D Myers, MPJ Oreska, K McGlathery (2018). The Science and Policy of the Verified Carbon Standard Methodology for Tidal Wetland and Seagrass Restoration. *Estuaries and Coasts* 41(8): 2159-2171. doi:10.1007/s12237-018-0429-0.
- Neubauer, SC, Givler, K, Valentine, S, & Megonigal, JP (2005). Seasonal Patterns And Plant-Mediated Controls Of Subsurface Wetland Biogeochemistry. *Ecology*, 86(12):3334-3344. doi:10.1890/04-1951.
- Neubauer SC, Megonigal JP (2015) Moving Beyond Global Warming Potentials to Quantify the Climatic Role of. *Ecosystems* 1–14. doi: 10.1007/s10021-015-9879- 4.
- Perry, JE, Hershner, CH (1999) Temporal changes in the vegetation pattern in a tidal freshwater marsh. *Wetlands* 19(1):90-99. doi:10.1007/bf03161737.
- Poffenbarger HJ, Needelman BA, Megonigal JP (2011) Salinity Influence on Methane Emissions from Tidal Marshes. *Wetlands* 31:831–842. doi: 10.1007/s13157-011- 0197-0.
- Rabenhorst, MC, JP Megonigal and JK Keller (2010) Synthetic iron oxides for documenting sulfide in marsh pore water. *Soil Science Society of America Journal*. 74(4): 1383-1388. <https://doi.org/10.2136/sssaj2009.0435>.
- Roden, E, Wetzel, R (2003) Competition between Fe(III)-Reducing and Methanogenic Bacteria for Acetate in Iron-Rich Freshwater Sediments. *Microbial Ecology* 45(3):252-258. doi:10.1007/s00248-002-1037-9.
- Saunois et al (2020) The Global Methane Budget 2000-2017. *Earth Systems Science Data* 12:1561-1623. <https://doi.org/10.5194/essd-12-1561-2020>.
- Solomon S, Qin D, Manning M, et al (2007) IPCC Fourth Assessment Report (AR4). Retrieved 4:2011.
- Sorrell BK, Brix H, DeLaune RD, et al (2013) Gas Transport and Exchange through Wetland Plant Aerenchyma. In: SSSA Book Series. Soil Science Society of America.
- Sorrell, B, Mendelssohn, IA, McKee, KL, Woods, RA (2000) Ecophysiology of Wetland Plant Roots: A Modelling Comparison of Aeration in Relation to Species Distribution. *Annals of Botany*, 86(3), 675-685. doi:10.1006/anbo.2000.1173.

Sutton-Grier AE, Megonigal JP (2011) Plant species traits regulate methane production in freshwater wetland soils. *Soil Biol Biochem* 43:413–420. doi: 10.1016/j.soilbio.2010.11.009.

Vann, CD, Megonigal, JP (2003) Elevated CO₂ and water depth regulation of methane emissions: Comparison of woody and non-woody wetland plant species. *Biogeochemistry*, 63:117-134, <https://doi.org/10.1023/A:1023397032331>.

Villa, JA, Ju, Y, Stephen, T, Rey-Sanchez, C, Wrighton, KC, Bohrer, G (2020) Plant-mediated methane transport in emergent and floating-leaved species of a temperate freshwater mineral-soil wetland. *Limnology and Oceanography* 65(7):1635-1650. doi:10.1002/lno.11467.

Walker DA, Auerbach NA, Bockheim JG, et al (1998) Energy and trace-gas fluxes across a soil pH boundary in the Arctic. *Nature* 394:469–472. doi: 10.1038/28839.

Wang Z, Zeng D, Jr WHP (1996) Methane emissions from natural wetlands. *Environ Monit Assess* 42:143–161. doi: 10.1007/BF00394047.

Weiss, JV, Emerson, D, Backer, SM, Megonigal, JP (2003) Enumeration of Fe(II)-Oxidizing and Fe(III)-Reducing Bacteria in the Root Zone of Wetland Plants: Implications for a Rhizosphere Iron Cycle. *Biogeochemistry* 64:77-96.

Weiss, JV, Emerson, D, Megonigal, JP (2004) Geochemical control of microbial Fe(III) reduction potential in wetlands: Comparison of the rhizosphere to non-rhizosphere soil. *FEMS Microbiology Ecology* 48(1):89-100. doi:10.1016/j.femsec.2003.12.014.

Whalen SC (2005) Biogeochemistry of methane exchange between natural wetlands and the atmosphere. *Environ Eng Sci* 22:73–94.

Windham L (2001) Comparison of biomass production and decomposition between *Phragmites australis* (common reed) and *Spartina patens* (salt hay grass) in brackish tidal marshes of New Jersey, USA. *Wetlands* 21:179–188. doi: 10.1672/0277-5212(2001)021[0179:COBPAD]2.0.CO;2.

Yu K, Hiscox A, DeLaune RD, et al (2013) Greenhouse Gas Emission by Static Chamber and Eddy Flux Methods. In: SSSA Book Series. Soil Science Society of America.

Tables

Table 1. Mean values by strata of Methane Flux, Porewater Salinity, Porewater Sulfate concentration, Porewater Sulfate Depletion, Porewater Hydrogen Sulfide, Porewater Methane, Porewater Reduced Iron (*Reduced Iron was collected in July only), and Antecedent Water Levels (defined as mean water level below soil surface for the two weeks prior to sampling) in a brackish marsh complex. Superscript letters denote statistical differences between strata. Means with the different letters have statistically significant differences. Standard error is presented below each mean.

Strata	Methane Flux (mg CH ₄ m ⁻² hr ⁻¹)	Salinity (ppt)	Sulfate (mg/L)	Sulfate Depletion (%)	Hydrogen Sulfide (mg/L)	Porewater Methane (mg/L)	Reduced Iron* (mg/L)	Antecedent Water Level (cm)
<i>S. alterniflora</i>	1.2 ^A ±0.3	14.2 ^A ±0.2	434.3 ^C ±33.7	60.6% ^C ±3.5	72.2 ^A ±8.8	4795.6 ^A ±781.2	0.02 ^B ±0.009	1cm ^B
<i>S. patens</i>	0.1 ^C ±0.01	12.3 ^B ±0.4	852.2 ^A ±63.0	1.4% ^A ±8.5	18.9 ^B ±3.5	263.7 ^C ±29.7	72.0 ^A ±40.1	9cm ^A
<i>J. roemerianus</i> (Low Elevation)	0.3 ^C ±0.07	14.8 ^A ±0.3	693.0 ^{AB} ±45.7	38.3 ^B ±4.0	14.9 ^B ±2.7	1174.7 ^B ±217.4	0.78 ^B ±0.4	1.1cm ^B
<i>J. roemerianus</i> (High Elevation)	0.4 ^B ±0.06	14.8 ^A ±0.3	601.8 ^B ±35.8	54.3 ^{BC} ±3.3	19.9 ^B ±3.3	1243.0 ^B ±247.8	0.30 ^B ±0.2	0.7cm ^B

Table 2. Mean Values of Mol CO₂ and CH₄ Produced on Day Five of Anaerobic Soil Incubations and the Ratio of CO₂ to CH₄ produced from soils collected from a brackish marsh complex. Superscript Letters Denote Statistical Differences Between strata. Means with the different letters have statistically significant differences. Standard error is presented below each mean.

Strata	Mol of CH ₄ Produced at 10cm in Anaerobic Incubations	Mol of CO ₂ Produced at 10cm in Anaerobic Incubations	Ratio of CO ₂ :CH ₄ Produced at 10cm in Anaerobic Incubations
<i>S. alterniflora</i>	13.8 ^A ±5.8	101.5 ^A ±19.0	39.8 ^A ±22.1
<i>S. patens</i>	0.4 ^B ±0.1	229.5 ^B ±16.6	933.1 ^B ±324.2
<i>J. roemerianus</i> (Low Elevation)	0.9 ^B ±0.1	171.8 ^{AB} ±29.7	199.1 ^A ±31.1
<i>J. roemerianus</i> (High Elevation)	0.8 ^B ±0.1	146.5 ^{AB} ±19.5	209.2 ^A ±43.5

Table 3. Field-measured salinity and methane flux mean values from four strata in a tidal marsh complex as compared to predicted flux (based on observed salinity) from Poffenbarger et al. 2011, with percent differences of actual and predicted carbon sequestration offsets.

Strata	Salinity (ppt)	Methane Flux (Mg C ha ⁻¹ yr ⁻¹)	Predicted Methane Flux from Poffenbarger et.al (Mg C ha ⁻¹ yr ⁻¹)	Percent Difference from Poffenbarger et. al	Percent of Carbon Sequestration Offset (actual)	Percent of Carbon Sequestration Offset (predicted)
<i>S. alterniflora</i>	14.2	0.67	0.29	131.0% More than Predicted	45.9%	19.9%
<i>S. patens</i>	12.3	0.07	0.37	81.0% Less than Predicted	4.8%	25.3%
<i>J. roemerianus</i> (Low Elevation)	14.8	0.16	0.27	40.7% Less than Predicted	11.0%	18.5%
<i>J. roemerianus</i> (High Elevation)	14.8	0.22	0.27	18.5% Less than Predicted	15.1%	18.5%

Figures

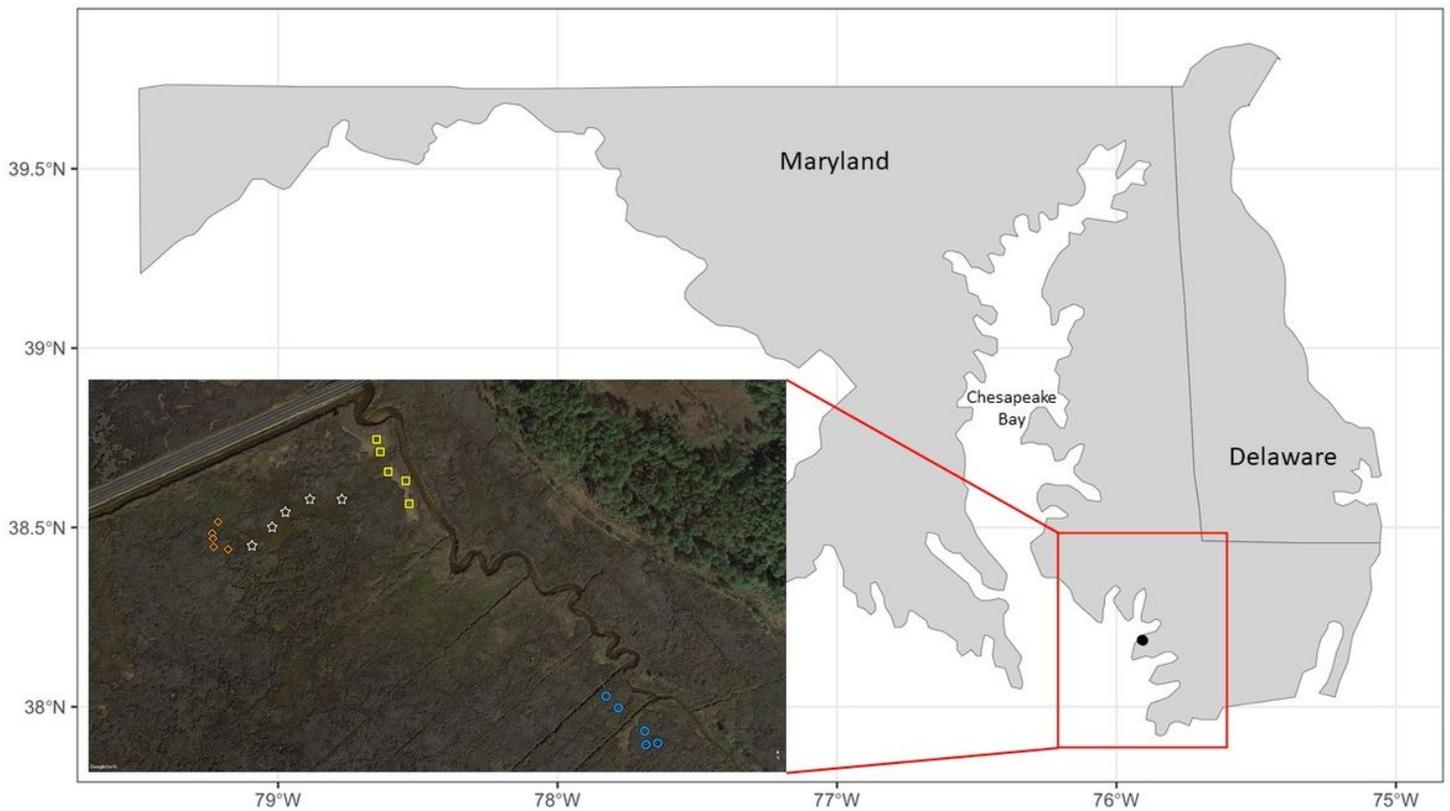


Figure 1

Study site and plot locations (inset) located on the Eastern Shore of Maryland, USA. *S. alterniflora* plots are marked with star icons, *S. patens* plots are marked with squares, High-elevation *J. roemerianus* plots are marked with diamonds, and Low-Elevation *J. roemerianus* plots are marked with circles. Note: The designations employed and the presentation of the material on this map do not imply the expression of any opinion whatsoever on the part of Research Square concerning the legal status of any country, territory, city or area or of its authorities, or concerning the delimitation of its frontiers or boundaries. This map has been provided by the authors.

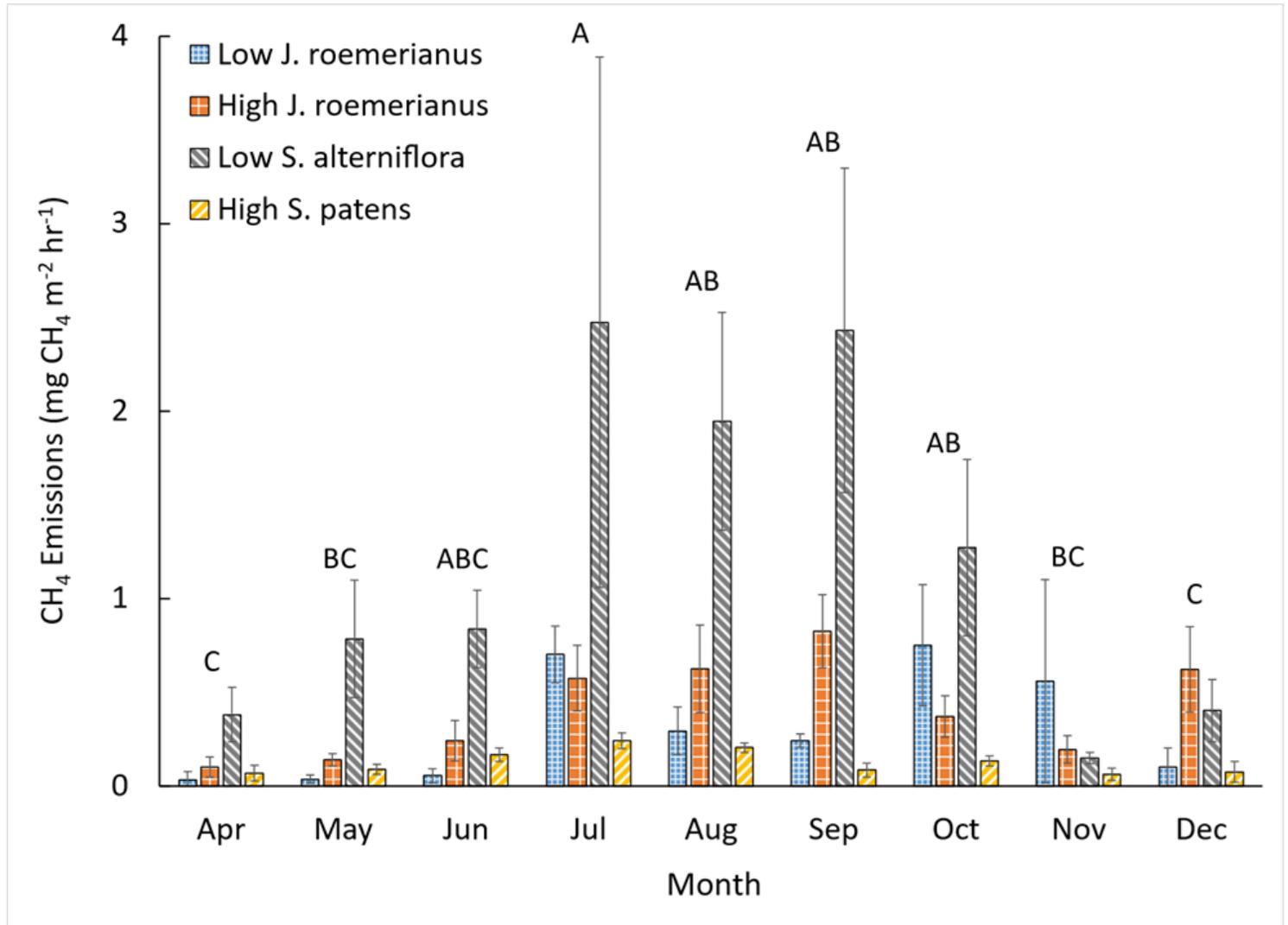


Figure 2

Mean hourly methane flux during the sampling period in a tidal brackish marsh on the Deal Island Peninsula on Maryland's Eastern Shore. "Low" vs "High" refers to comparative elevations within the site. Error bars signify standard error; significant differences are shown with letters. ANOVA results of the log transformed data showed significant differences between strata ($p < 0.0001$), month ($p < 0.0001$) and strata*month ($p = 0.0182$).