

# Hydrochemical Formation Mechanism of Mineral Springs in Changbai Mountain, China

**Jialin Li**

Jilin University

**min jian Bian** (✉ [bianjianmin@126.com](mailto:bianjianmin@126.com))

Jilin university

**Yihan Li**

Jilin University

**Yuxi Ma**

Jilin University

**Yanmei Li**

University of Guanajuato: Universidad de Guanajuato

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## Research Article

**Keywords:** Hydrochemical formation mechanism, Cluster analysis, PCA , Metasilicic-acid mineral springs, Changbai Mountain

**Posted Date:** February 18th, 2021

**DOI:** <https://doi.org/10.21203/rs.3.rs-208397/v1>

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1 **Hydrochemical formation mechanism of mineral springs in**  
2 **Changbai Mountain, China**

3 Jialin Li<sup>1,2</sup>, Jianmin Bian<sup>1,2\*</sup>, Yihan Li<sup>1,2</sup>, Yuxi Ma<sup>1,2</sup>, Yanmei Li<sup>3</sup>

4 <sup>1</sup>Key Laboratory of Groundwater Resources and Environment, Ministry of Education,  
5 Jilin University, Changchun 130021, China

6 <sup>2</sup>College of New Energy and Environment, Jilin University, Changchun 130021,  
7 China

8 <sup>3</sup>Department of Mine, Metallurgy and Geology Engineering, Engineering Division,  
9 University of Guanajuato, Guanajuato C.P. 36020, Mexico

10 **\*Corresponding author.** [Tel:+86-1800-431-9968](tel:+86-1800-431-9968); E-mail:bianjianmin@126.com  
11 (Jianmin Bian)

12 **Declarations**

13 •Funding

14 This research was funded by the National Key R&D Program of China, grant number  
15 2019YFC0409103, and the Key R&D Program of Science and Technology  
16 Department of Jilin Province, grant number 20190303076SF.

17 The funding body has funded sample collection tests and data analysis.

18 •Competing interests

19 Not applicable

20 •Availability of data and material

21 Not applicable

22 •Code availability

23 Not applicable

24 • Author contributions

25 Jialin Li: Data curation, Formal analysis, Methodology, Software, Roles/Writing -  
26 original draft.

27 Jianmin Bian: Conceptualization, Funding acquisition, Project administration,  
28 Resources, Supervision.

29 Yihan Li: Investigation, Validation.

30 Yuxi Ma: Investigation, Visualization.

31 Yanmei Li: Writing - review & editing.

32 All authors read and approved the final manuscript.

33 •Ethics approval

34 Not applicable

- 35 • Consent to participate
- 36 The author declares that they agree to participate
- 37 • Consent for publication
- 38 The author declares that they agree to publication

## 39 ABSTRACT:

40 Changbai Mountain area in China is an important mineral water storage and  
41 development area. The hydrochemical composition of mineral water is the decisive  
42 factor for mineral-water quality. Therefore, it is necessary to study the hydrochemical  
43 characteristics and genesis of mineral water in Changbai Mountain area. Considering  
44 the integrity of basin and groundwater system, the study area was classified into three  
45 areas where mineral springs are densely distributed. For these areas, according to the  
46 mineral water parameter contents, the springs are further classified into single-type  
47 (metasilicic acid) mineral spring and compound mineral spring. By examining 74  
48 mineral water samples collected from 2018 to 2020, the characteristics and formation  
49 mechanism of the hydrochemical components of the spring water were analyzed by  
50 mathematical statistics, model construction, cluster analysis, and principal component  
51 analysis. The results show that the formation of single-type mineral spring is  
52 controlled by rock weathering; compound mineral springs are the product of CO<sub>2</sub>-rich,  
53 weakly acidic, confined hot groundwater with high salinity and its mixing with  
54 shallow groundwater as it rises along the fracture. Volcanic geological process greatly  
55 influence the formation of the hydrochemical components of mineral springs on the  
56 North slope of Changbai Mountain. The mineral springs on the Longgang Mountain  
57 are greatly affected by human activities. The results of cluster analysis only  
58 considering hydrochemical components are consistent with the classification of the  
59 areas with concentrated distributions of mineral springs as determined by  
60 hydrogeological and geomorphological studies. The results of this study are useful for  
61 understanding the distribution, hydrochemical characteristics, and formation  
62 mechanism of mineral springs in Changbai Mountain area of China and provide the  
63 theoretical basis for the protection and development of mineral spring water.

## 64 Keywords:

65 Hydrochemical formation mechanism; Cluster analysis; PCA ; Metasilicic-acid  
66 mineral springs; Changbai Mountain.

## 67 1. Introduction

68 Mineral water is special groundwater formed under special hydrogeological  
69 conditions. It contains a certain amount of minerals and some trace elements and is

70 not polluted. It is a rare blue water resource. Mineral water generally has the  
71 characteristics of good oral feeling, fine water quality, and some minerals and trace  
72 elements needed by human body, and hence, it is favored by people for consumption.  
73 The Changbai Mountain in China is characterized by active geologic processes, and  
74 multiple stages of volcanic activities have resulted in the formation of basalt, trachyte,  
75 pumice, and other magmatic rocks here. These rocks constitute the lava platform and  
76 volcanic cone of Changbai Mountain and Longgang Mountain. The cracks and pores  
77 of volcanic rocks are very developed. The widely distributed volcanic eruptive clastic  
78 sediment favor the proliferation of forests and vegetation and promote the recharge of  
79 groundwater by meteoric water. These factors have created unique conditions for the  
80 formation and enrichment of mineral water in the Changbai Mountain area. During  
81 groundwater flow, hydrogeochemical interactions occur between groundwater and the  
82 surrounding rock and soil, dissolving some mineral components and trace elements in  
83 the volcanic rocks and forming abundant and high-quality mineral water resources.  
84 Mineral springs, mainly of the metasilicic-acid ( $H_2SiO_3$ ) type, are distributed radially  
85 around the main peak of Changbai Mountain. At present, this area is an important  
86 mineral water source in China and is among the three major high-quality mineral  
87 water sources in the world, along with the Alps and Caucasus mountains.

88 Since mineral spring water is directly discharged groundwater, the research  
89 methods used for groundwater can be applied to study mineral spring water. Research  
90 on the hydrochemical characteristics and formation mechanism of mineral springs  
91 mainly involves the use of graphic analysis, descriptive statistics, and multivariate  
92 statistical methods; in addition, hydrogen and oxygen isotopes and a variety of  
93 radioisotopes are used to analyze the formation time and cycles of cold spring and hot  
94 spring (Prtoljan, Kapelj et al. 2012, Haklidir 2013). The characteristics, origins, and  
95 influencing factors of the hydrochemical components of mineral springs are analyzed  
96 based on the statistical analysis of hydrochemical data and isotope method (Kopylova,  
97 Lepokurova et al. 2011, Chelnokov, Bragin et al. 2018). The dissolution rules of  
98 mineral components and the hydrogeochemical formation mechanism of mineral  
99 water were explored using TOUGHREACT and PHREEQC simulators and by water–  
100 rock interaction experiments (Choi, Yun et al. 2014, Yan, Xiao et al. 2016). The  
101 conceptual model of the groundwater system was constructed by using the method of  
102 system science, and the groundwater system was characterized (Kulkarni, Deolankar  
103 et al. 2000, Dafny, Burg et al. 2006, Daniele, Taucare et al. 2020). The origins,  
104 dynamics, and influencing factors of mineral spring flow was studied by wavelet  
105 analysis (Gao, Bian et al. 2016). Multifactor statistical analysis and comprehensive  
106 evaluation methods were used to evaluate the quality of mineral water as daily life

107 water (Leite, Stolberg et al. 2018). For mineral springs with industrial value and  
108 medical care value, the distribution of mineral springs, content and function of special  
109 components, and development and utilization were studied (Vinograd 2004).  
110 Considering the development and utilization value of mineral water resources, the  
111 hydrochemical characteristics, types, distribution, development, and utilization modes  
112 of mineral springs were studied (Barut, Erdogan et al. 2004, Corral, Galindo et al.  
113 2014).

114 Research on mineral spring water in Changbai Mountain area began in the 1980s.  
115 The initial research focused on the distribution characteristics of mineral water  
116 resources and the qualitative analysis of mineral spring formation. At present, the  
117 experiments and models, simulation and other technical means, isotope and system  
118 science, hydrological analysis and other methods are used to study the whole process  
119 of mineral water formation and its quantity and quality (Yan, Xiao et al. 2014, Zhang,  
120 Liang et al. 2017). Although the mineral springs in Changbai Mountain area are  
121 distributed throughout the region, most of the springs are concentrated in certain small  
122 areas. In recent years, the development of mineral water resources has increased, and  
123 mineral water development has gained importance as a major industry. The mineral  
124 water in this area is developed by delimiting the protected areas where mineral springs  
125 are concentrated and establishing a mineral water industry base to realize the  
126 centralized development of mineral water in the protected areas. Within a given area  
127 with a high density of mineral springs, the springs are interrelated in terms of water  
128 quantity, water quality, and dynamics, all the springs within the given area is a unified  
129 whole. Because of the continuous increase in the development and utilization,  
130 problems such as reduction in water quantity and change in water quality have been  
131 observed in some springs. This will also affect other mineral springs and underground  
132 aquifers within the same spring-concentrated area. It is necessary to analyze and  
133 explore the hydrochemical characteristics and formation mechanism of mineral  
134 springs to provide a theoretical basis for protecting mineral water resources and  
135 ensuring sustainable development.

## 136 2. Study area

137 Changbai Mountain is the highest mountain range in Northeast Asia and is  
138 located at the border between China and Korea (Fig.1) . It is the largest volcanic  
139 mountain range in the humid areas of China. The volcanic geological process is active,  
140 and the lava and pyroclastic rocks ejected form a giant volcanic cone and basalt  
141 platform with the Tianchi caldera as the center. The total area of the basalt platform is  
142 5500 km<sup>2</sup>, and the average thickness is about 180 m. The climate of the study area is

143 the East Asian monsoon climate with abundant rainfall and an annual average  
144 precipitation of more than 600 mm. Songhua River, Yalu River, and Tumen River  
145 originate in this region. The forest cover is more than 90%. The interception and  
146 absorption of rainwater by vegetation promotes the infiltration of rainwater.  
147 Groundwater is mainly supplied by atmospheric precipitation and converted to surface  
148 water and discharged into a low-lying valley. The main types of groundwater are  
149 basalt pore-fissure water, pore water in loose rock, karst-fissure water in carbonate  
150 rock, pore-fissure water in clastic rock, and fissure water in granite rock (Fig.1) .  
151 Basalt has the characteristics of pore development, good connectivity, and high silica  
152 content and is an ideal medium for groundwater storage and migration. Material  
153 exchange occurs between basalt and groundwater, forming abundant  
154 metasilicic-acid-type mineral water resources. Mineral springs are concentrated in the  
155 low basaltic platform of Longgang Mountain and basaltic plateau area of Changbai  
156 Mountain, forming a spring group. The low platform of the Longgang Mountain is  
157 located in the western part of the study area and has an altitude of 500–1200 m. In this  
158 platform, more than 170 craters were formed by the accumulation of basaltic trass.  
159 The Changbai Mountain basalt plateau is located under a giant volcanic cone, and  
160 around the volcanic cone, the altitude gradually decreases from a height of 1200 m to  
161 600 m (the valley).

### 162 3. Data and methods

#### 163 3.1 Sample collection and measurement

164 In all, 74 mineral water samples were collected in the months of July–October in  
165 2018–2020 (Fig.1) . The mineral water indexes were tested according to “Methods for  
166 examination of drinking natural mineral water” GB 8538-2016 (National Health  
167 Commission of the People's Republic of China. 2016). The sampling bottles were  
168 polyethylene bottles and glass bottles, which were washed with distilled water in  
169 advance. Before sampling, the sampling bottles were rinsed thoroughly with the  
170 spring water from the target sampling point thrice, and the pH value was measured on  
171 site. Subsequently, the water samples were sealed and labeled, stored at 4°C, and sent  
172 to the Testing Science Experimental Center of Jilin University and Jilin Provincial  
173 Institute of Geological Sciences for water chemistry testing.

174 The test items include  $\text{HCO}_3^-$ 、 $\text{Cl}^-$ 、 $\text{SO}_4^{2-}$ 、 $\text{Na}^+$ 、 $\text{K}^+$ 、 $\text{Ca}^{2+}$ 、 $\text{Mg}^{2+}$ 、 $\text{NO}_3^-$ 、 $\text{F}^-$ 、  
175 TDS、TH、TFe、 $\text{H}_2\text{SiO}_3$  , etc. Among them,  $\text{Na}^+$  and  $\text{K}^+$  were determined by FAAS;  
176  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  were determined by EDTA disodium salt titration;  $\text{HCO}_3^-$  was  
177 determined by double-indicator;  $\text{Cl}^-$  was titrated with silver nitrate;  $\text{SO}_4^{2-}$ 、 $\text{NO}_3^-$  and  
178  $\text{F}^-$  were determined by ion chromatography; TDS determination by electronic balance



208 analysis can be classified into Q-type clustering for the research object itself and  
 209 R-type clustering for variables and indicators of the research object. In this study,  
 210 SPSS software was used for Q-type clustering analysis of hydrochemical data. The  
 211 Euclidean distance was used to calculate the distance between different sample points.  
 212 Ward algorithm was used to calculate the distance between clusters. Because the order  
 213 of magnitude of different indicators is different, it is necessary to standardize the  
 214 whole range 0–1 to obtain the cluster tree of mineral water chemistry.

215 Here, Euclidean distance is the absolute distance between two points in  
 216 multidimensional space, and it is calculated as follows:

$$217 \quad d_{ij} = \|x_i - x_j\| = \sqrt{\sum_{k=1}^p (x_{ik} - x_{jk})^2}$$

218 (4)

219 Ward algorithm is mainly based on the idea of variance analysis, and it requires  
 220 that the sum of squares of deviation between the same clusters is as small as possible.  
 221 The method involves merging two clusters of objects close to each other until all  
 222 objects are merged. The calculation principle of this algorithm is as follows: First, all  
 223 sample points are divided into m classes ( $P_1, P_2 \dots P_m$ ), where  $P_n^{(m)}$  denotes the nth  
 224 index in  $P_m$ , and the index number of  $P_m$  is k. Let  $\overline{P^{(m)}}$  denote the center of gravity  
 225 of  $P_m$ . Then, the sum of squares of deviation of each sample point is as follows:  
 226

$$227 \quad S_m = \sum_{n=1}^k (P_n^{(m)} - \overline{P^{(m)}}) (P_n^{(m)} - \overline{P^{(m)}})$$

228 (5)

229 Then the intraclass sum of squares of deviation of m categories is as follows:

$$230 \quad S = \sum_{m=1}^m S_m = \sum_{m=1}^m \sum_{n=1}^k (P_n^{(m)} - \overline{P^{(m)}}) (P_n^{(m)} - \overline{P^{(m)}})$$

231 (6)

232 If the order of magnitude of the variable index is different, it is necessary to  
 233 transform the variable. The formula of standardized transformation for the full range  
 234 0–1 is as follows:

$$235 \quad Z_{ij} = \frac{x_{ij} - \bar{x}_j}{s_j} \tag{7}$$

236 Where,  $x_j = \frac{1}{n} \sum_{i=1}^n x_{ij}$   $s_j = \frac{1}{n-1} \sum_{i=1}^n (x_{ij} - \bar{x}_j)^2$

## 237 4. Results and discussion

### 238 4.1 Classification and hydrochemical characteristics of mineral 239 springs

#### 240 4.1.1 Classification of mineral springs

241 Water–rock interaction results in a type of dynamic, natural liquid mineral under  
242 the controls imposed by the rock medium and structure. This leads to a natural  
243 distribution of mineral water with fixed hydrochemical characteristics. The mineral  
244 springs in the study area have the characteristics of concentrated distribution in  
245 different hydrogeological and basin units. To analyze the similarity and differences in  
246 the chemical characteristics and hydrochemical origins of mineral water under the  
247 control of different hydrogeological and geomorphological conditions, digital  
248 elevation model (DEM) data were used. Using the hydrological analysis function of  
249 ArcGIS, the basin in the study area was extracted and divided. This process included  
250 the following steps: creating nondepression DEM; performing flow direction analysis  
251 and flow statistics; and defining the minimum surface runoff, river linking, grid river  
252 vectorization. By comprehensive consideration of basin division and mineral spring  
253 enrichment, three areas with a dense distribution of mineral springs were  
254 selected—the West Slope of Changbai Mountain (West slope), the North Slope of  
255 Changbai Mountain (North slope), and the Longgang Mountain (Longgang). Each of  
256 these areas is located in different hydrogeological and watershed units. The three  
257 areas form 26.55% of the research area, and contain 61 sampling points, accounting  
258 for 82.43% of the total number of sampling points (Fig.1). The mineral springs within  
259 any one area are regarded as a group for classification analysis.

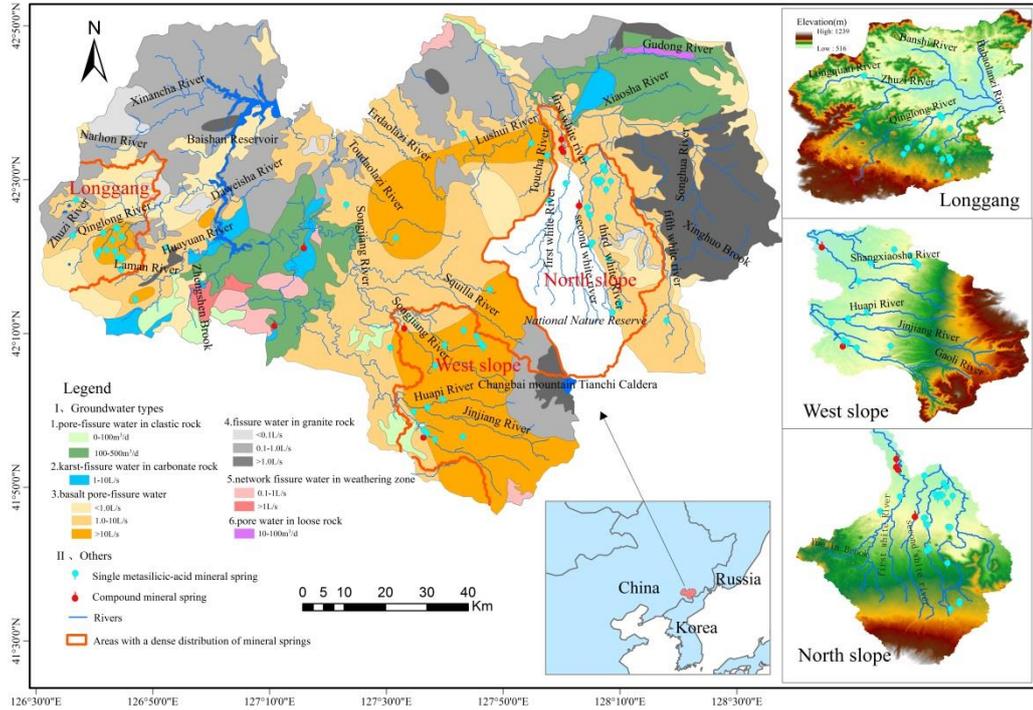


Fig.1.Hydrogeology conditions and sampling springs

Mineral spring water parameters include the contents of lithium, strontium, zinc, metasilicic, selenium, free carbon dioxide (CO<sub>2</sub>), total dissolved solids (TDS), and the thresholds of the parameters are listed in Table 1. If only one parameter meets the standard of “Drinking natural mineral water” GB8537—2018 (National Health Commission of the People's Republic of China. 2018), the spring is a single-type mineral spring. If more than one parameter meets the standard, it is a compound mineral spring. Compound mineral springs are relatively rare in the study area, and their special components are listed in Table 2.

Table1.Drinking natural mineral water parameter and thresholds

Types	Parameter	Threshold
metasilicic	H <sub>2</sub> SiO <sub>3</sub> / (mg/L)	≥25.0 (When the concentration is 25.0 mg/L-30.0 mg/L, the water temperature should be above 25 °C)
lithium	Li / (mg/L)	≥0.20
strontium	Sr / (mg/L)	≥0.20 (When the concentration is 0.20 mg/L-0.40 mg/L, the water temperature should be above)
zinc	Zn / (mg/L)	≥0.20
selenium	Se / (mg/L)	≥0.01
free carbon dioxide	CO <sub>2</sub> / (mg/L)	≥250
total dissolved solids	TDS/ (mg/L)	≥1000

Table2. Special components contents of compound mineral springs (all in mg/L)

Site	H <sub>2</sub> SiO <sub>3</sub>	CO <sub>2</sub>	TDS	Li	Sr
I	77.55	--	1365.66	--	0.33
II	40.30	396.60	2646.72	0.64	1.48
III	40.00	308.00	2343.00	0.20	4.33
IV	56.47	305.56	1402.40	--	--
V	50.05	734.00	1371.00	--	0.45
VI	96.20	658.11	2029.68	--	--
VII	98.80	1188.00	1672.00	0.19	0.44
VIII	107.83	1056	1672.00	0.21	0.52

273 The special component contents of each mineral spring were analyzed. The  
274 results show that the H<sub>2</sub>SiO<sub>3</sub>, CO<sub>2</sub>, and TDS of most of the compound mineral springs  
275 meet the quality standards of “Drinking natural mineral water” GB8537—2018, and  
276 the Li<sup>+</sup> and Sr<sup>2+</sup> contents of some mineral springs meet the standards. Single  
277 metasilicic-acid mineral springs are widely distributed in the study area. All the  
278 single-type mineral springs in this study are of the metasilicic-acid type, and the  
279 average content of metasilicic-acid is 43.79 mg/L. There are great differences between  
280 compound mineral springs and single-type mineral springs in terms of their in  
281 hydrochemical characteristics. According to the types of mineral springs and the area  
282 with dense distributions of mineral springs, the mineral springs in the study area were  
283 classified into five categories—compound mineral springs were regarded as a separate  
284 category. The single-type mineral springs were subdivided into four categories  
285 according to their locations: the spring groups of (1) the west slope of Changbai  
286 Mountain (West slope springs,) (2) the north slope of Changbai Mountain (North  
287 slope springs), (3) Longgang Mountain (Longgang springs), and (4) other regional  
288 mineral springs. Mineral springs of category 1–4 are all single metasilicic-acid type.  
289 Based on the hydrochemical data of various types of mineral springs, the  
290 hydrochemical characteristics and genesis of different types and regions of mineral  
291 springs were analyzed and studied.

#### 292 4.1.2 Hydrochemical types and component features

293 The minimum, maximum, average, and coefficient of variations of 14  
294 hydrochemical indexes, such as K<sup>+</sup>, Na<sup>+</sup>, and H<sub>2</sub>SiO<sub>3</sub> were calculated (Table.3) . The  
295 changes in the average contents of different hydrochemical indexes of all types of  
296 mineral springs show good consistency. Except for F and H<sub>2</sub>SiO<sub>3</sub>, the content of each  
297 component in the North slope springs is the lowest, and the TFe value of West slope  
298 springs is low. The hydrochemical characteristics of single-type mineral springs and  
299 compound mineral springs are quite different. The pH value of compound mineral  
300 springs is lower than that of the single-type mineral springs, and the F<sup>-</sup> content is

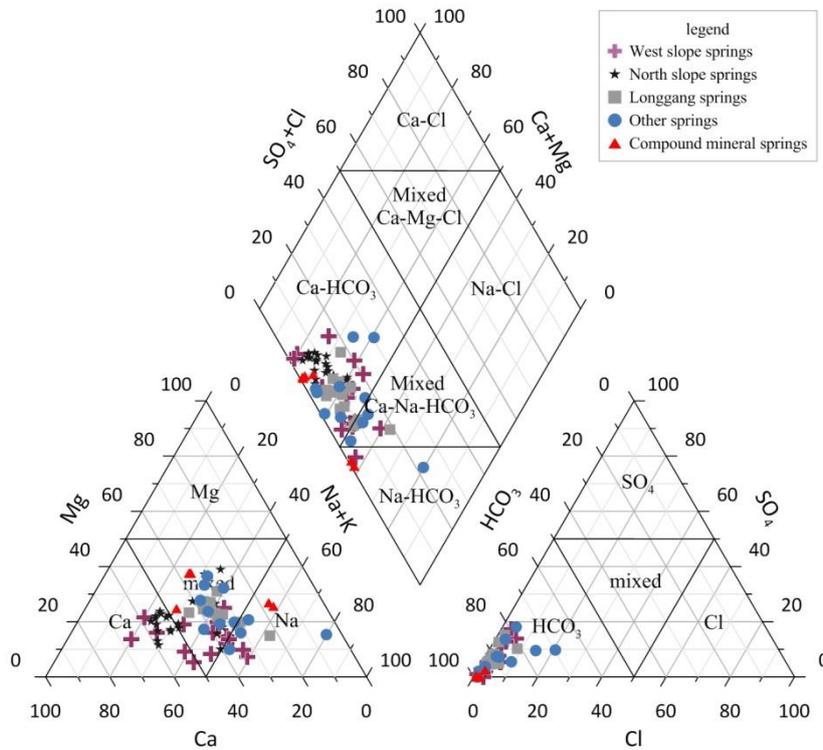
301 lower than that of the North slope springs; in comparison with the single-type mineral  
 302 spring, the content of  $\text{SO}_4^{2-}$ ,  $\text{K}^+$  is 2–5 times, that of  $\text{H}_2\text{SiO}_3$  is 1.5–1 times, the  
 303 content of  $\text{NO}_3^-$  is one order of magnitude lower, and the content of other  
 304 hydrochemical indexes is 1–2 orders of magnitude higher. The coefficient of variation  
 305 of Longgang springs is relatively small except for  $\text{NO}_3^-$ , and the  $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  
 306  $\text{Cl}^-$ ,  $\text{HCO}_3^-$ , TDS, TFe, and  $\text{H}_2\text{SiO}_3$  are the lowest among all types of mineral springs.  
 307 In addition to  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ , the coefficient of variation of the West slope springs is  
 308 relatively large, and the coefficients of variation of  $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{HCO}_3^-$ , TDS, and  
 309 pH are the largest among all types of mineral springs. The coefficients of variation of  
 310  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ , TFe,  $\text{H}_2\text{SiO}_3$ , and  $\text{F}^-$  in compound mineral springs are the largest.

311 **Table3.** Minimum (min), maximum (max), median (Md), and Standard Deviation (CV) values of  
 312 hydrochemical parameters in mineral springs of different types

Site	$\text{K}^+$	$\text{Na}^+$	$\text{Ca}^{2+}$	$\text{Mg}^{2+}$	$\text{Cl}^-$	$\text{SO}_4^{2-}$	$\text{HCO}_3^-$	$\text{NO}_3^-$	TDS	pH	TH	TFe	$\text{H}_2\text{SiO}_3$	$\text{F}^-$
West slope (n=16)														
Min	1.67	3.08	5.39	1.08	0.81	0.00	29.22	1.00	84.39	6.50	22.15	0.01	36.38	0.17
Max	15.19	66.25	24.39	16.76	7.25	8.92	300.78	8.08	484.93	7.85	199.00	0.38	84.64	0.79
Md	3.88	12.97	11.88	5.11	2.49	4.13	88.34	4.84	166.72	7.25	53.04	0.08	45.04	0.41
CV	1.01	1.19	0.44	0.93	0.66	0.50	0.85	0.39	0.66	0.06	0.81	1.21	0.30	0.47
North slope (n=23)														
Min	1.56	4.90	4.37	2.41	0.46	1.70	43.10	0.64	104.28	7.00	24.09	0.00	35.02	0.29
Max	3.60	15.60	18.53	22.14	4.91	7.66	184.91	10.00	294.44	7.89	137.45	0.26	59.13	1.34
Md	2.75	8.54	7.91	5.33	1.80	3.82	64.80	2.29	139.37	7.33	42.13	0.07	50.63	0.80
CV	0.18	0.28	0.47	0.73	0.63	0.43	0.42	0.81	0.27	0.03	0.53	1.10	0.12	0.32
Longgang (n=16)														
Min	2.42	5.76	10.98	5.96	1.76	4.26	70.12	1.87	142.50	6.67	62.06	0.01	30.46	0.07
Max	4.71	12.35	25.07	12.69	7.37	16.63	125.01	16.20	218.70	8.05	97.73	0.32	41.80	0.26
Md	3.79	9.43	15.81	9.20	3.29	7.83	104.42	4.84	180.24	7.58	78.74	0.11	36.49	0.16
CV	0.17	0.20	0.22	0.20	0.46	0.44	0.16	0.83	0.14	0.04	0.13	0.68	0.07	0.38
Others (n=11)														
Min	0.88	3.90	5.28	2.23	1.61	0.00	36.00	0.00	54.00	7.07	27.62	0.00	26.80	0.16
Max	6.50	15.85	22.76	22.10	13.49	8.88	216.99	5.00	322.24	8.15	145.97	0.29	63.01	0.36
Md	2.62	8.58	13.96	9.07	4.75	4.62	102.45	2.06	181.39	7.67	74.69	0.12	38.31	0.23
CV	0.67	0.45	0.42	0.80	0.97	0.53	0.61	0.83	0.47	0.05	0.56	0.79	0.25	0.22
Compound mineral springs (n=8)														
Min	6.53	43.75	98.08	48.77	5.89	0.00	678.13	0.00	1365.66	6.16	655.23	0.04	40.00	0.32
Max	25.68	214.05	181.61	152.36	42.60	46.06	1706.40	0.54	2646.72	6.98	903.65	3.30	107.83	1.50
Md	11.66	108.40	135.79	107.10	15.73	8.48	1239.34	0.12	1754.61	6.56	798.57	0.81	70.90	0.65

313 Note: n is the number of samples, and the unit of each index (except pH value) is mg/L.

314 The Piper trilinear diagram of various mineral springs were drawn using  
 315 Grapher, and their hydrochemical types were analyzed (Fig.2) . From the trilinear  
 316 diagram, it can be seen that  $Ca^{2+}$  accounts for the largest proportion of cations in the  
 317 North slope springs, and cations of other types of mineral springs are distributed in the  
 318 mixed area, and there is no dominant cation. The distribution of anions of all types of  
 319 mineral springs is concentrated, and  $HCO_3^-$  is the dominant anion. The hydrochemical  
 320 types of all types of mineral springs are basically consistent, including the  $HCO_3$ -Ca  
 321 type,  $HCO_3$ -Ca+Na type and  $HCO_3$ -Na type, among which  $HCO_3$ -Ca type accounts is  
 322 the dominant type.



323

324

Fig.2. Piper trigram diagram of mineral water

325

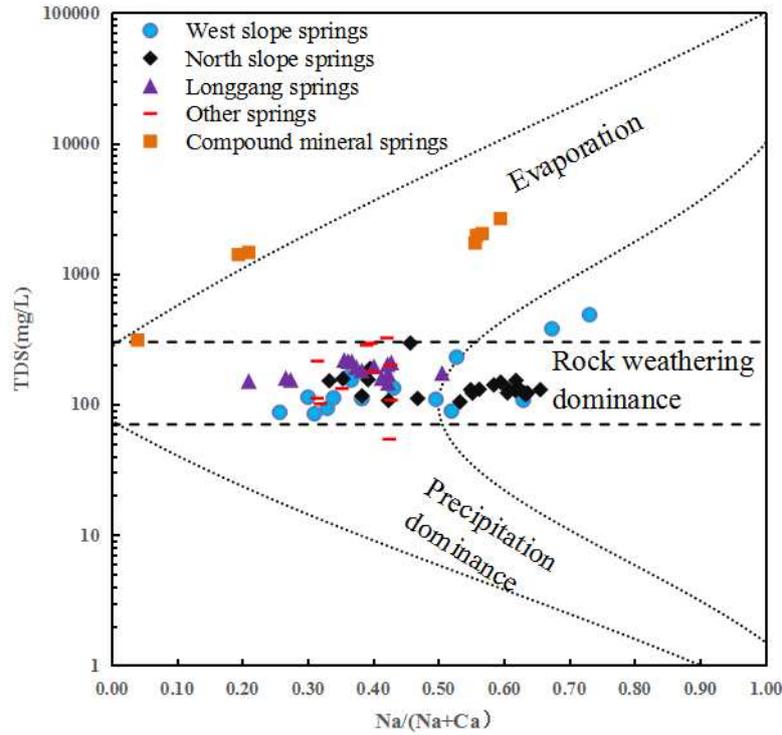
## 326 4.2 Hydrochemical formation causes

327 There are differences and similarities in the characteristics of the hydrochemical  
 328 components of different categories of mineral springs. This reflects the similarities  
 329 and dissimilarities in the hydrochemical formation mechanisms of mineral water. The  
 330 hydrochemical origins of various mineral springs were analyzed by using the Gibbs  
 331 diagram and lithologic end element ratios diagram. The hydrochemical characteristics  
 332 and genesis of single-type mineral springs and compound mineral springs are quite

333 different. Different hydrogeochemical processes have different effects on the  
334 formation of single-type mineral water chemical components in different areas with  
335 high densities of springs. There are obvious differences in their hydrochemical  
336 components. By using PCA, cluster analysis, and other hydrochemical analysis  
337 methods, combined with regional geological and hydrogeological conditions, the  
338 hydrochemical genesis of mineral springs in each category was explored.

#### 339 4.2.1 Controlling factors of hydrochemical formation

340 The Gibbs diagram is drawn based on the hydrochemical data of each mineral  
341 spring. According to the position of each spring in the Gibbs diagram, the main  
342 sources of the chemical components dissolved in water were analyzed, and the  
343 hydrochemical genesis of mineral springs was determined (Fig.3) . Most of the  
344 single-type mineral springs are located in the rock-weathering control area in the  
345 middle of Gibbs diagram, and rock weathering is the main factor determining the  
346 chemical composition of this type of mineral water. Meteoric water with very low  
347 mineral content leaches the surface soil and infiltrates the underground rock stratum.  
348 The groundwater flows through basalt and pyroclastic rock, and the mineral  
349 components in the weathered and broken rock are transferred to groundwater, forming  
350 the metasilicic-acid mineral water. Compound mineral springs are located in the  
351 evaporation zone and the periphery of the model. Because Changbai Mountain is a  
352 humid area, evaporation has little effect on the formation of mineral water. This type  
353 of mineral spring is not the product of evaporation and concentration, but it is affected  
354 by the special conditions during formation. The formation mechanism of  
355 hydrochemistry has changed, resulting in the atypical characteristics of hydrochemical  
356 components.



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**Fig.3.** Gibbs diagram of springs

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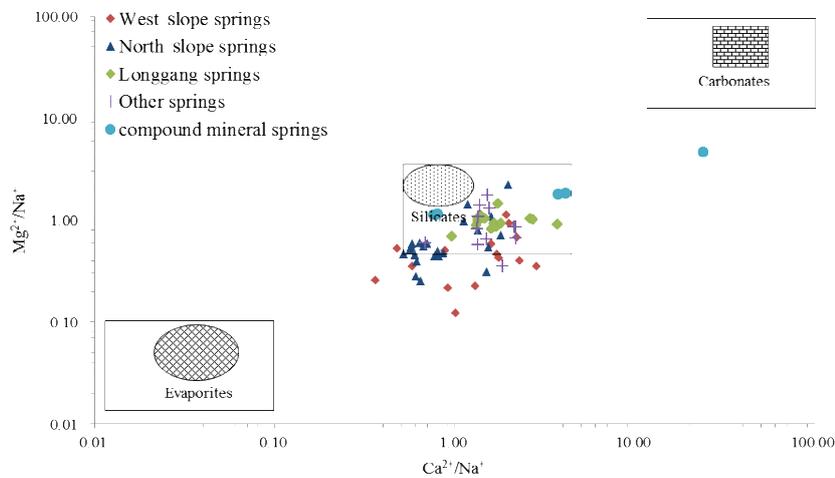
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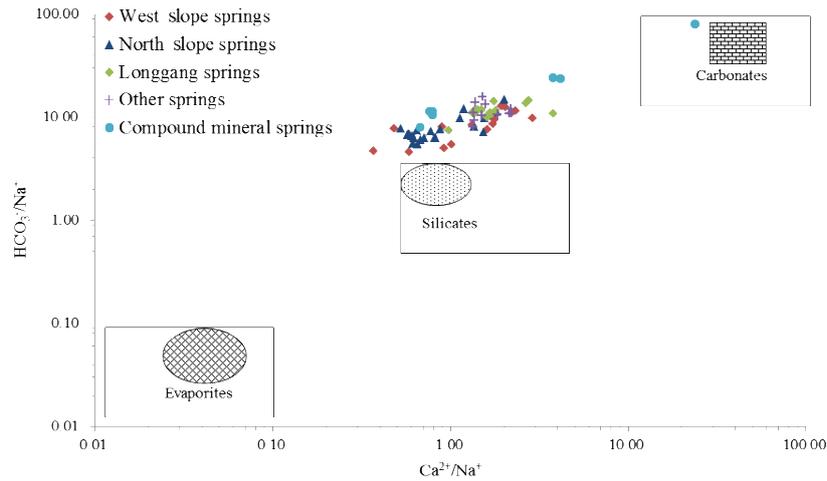
The lithologic end element ratios diagram (Fig. 4) for the mineral springs in the study area can be used to determine the rock types involved in water–rock interactions. Fig. 4 shows that the mineral springs mainly located in the silicate rock area and between the carbonate rock and silicate rock, and silicate rock is dominant. It shows that the hydrochemical components of mineral springs in Changbai Mountain are mainly controlled by silicate rocks and influenced by carbonate rocks. The influence of carbonate rocks on the West slope springs and the North slope springs is less than that of Longgang springs.



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(a)



(b)

**Fig.4.**Lithologic end element ratios diagram of different springs

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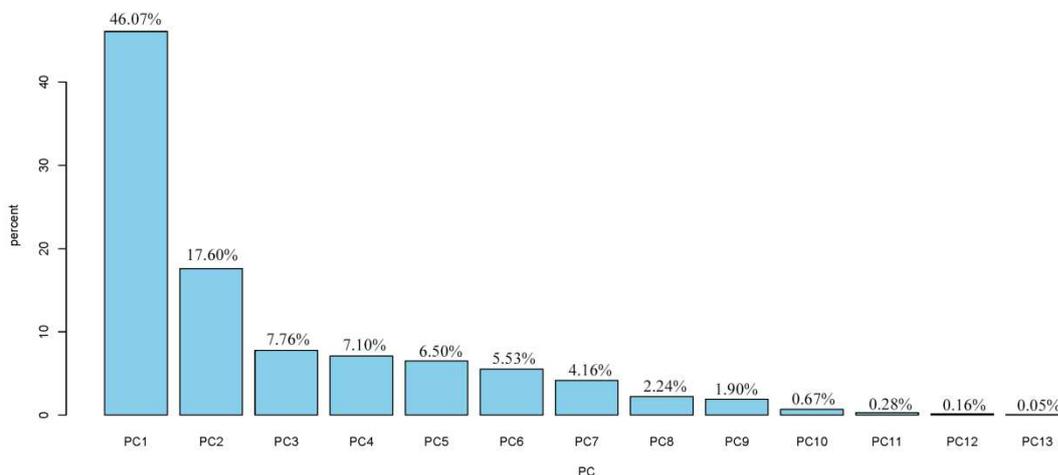
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#### 372 4.2.2 PCA of single-type mineral springs

373 The hydrochemical characteristics of mineral springs are controlled by the  
 374 dominant hydrogeochemistry reactions, but the formation process of mineral springs  
 375 is often affected by many factors. Different hydrogeochemistry reactions will produce  
 376 different hydrochemical characteristics and components (Yang, Li et al. 2016,  
 377 Ligavha-Mbelengwa and Gomo 2020). PCA can be used to judge the degree of  
 378 influence of different hydrochemical actions in the process of mineral spring  
 379 formation and the similarity and dissimilarity of each hydrochemical component  
 380 source. PCA is a statistical analysis method that simplifies multiple indicators into a  
 381 small number of comprehensive indicators. It uses a few variables to reflect the  
 382 information of the original variables as much as possible to ensure that the loss of  
 383 original information is small and the number of variables is as small as possible  
 384 (Lourenço, Ribeiro et al. 2010, Huang, Sun et al. 2013). Using R, which integrates  
 385 statistical analysis and graphic display, 13 hydrochemical indexes of single-type  
 386 mineral spring in the study area were simplified and analyzed, and the histogram of  
 387 factor contribution rate and double projection diagram of the load in PCA was drawn.

388 The factor contribution rate histogram can be used to help determine the optimal  
 389 principal component. The abscissa represents the main component, and the ordinate  
 390 represents the contribution rate. The first several principal components with larger  
 391 contribution rates are the main components that should be selected. According to  
 392 contribution rate of each factor the number of selected principal components is  
 393 determined. The histogram of contribution rate is shown in Fig.5. The contribution  
 394 rate of the first five principal components is relatively large, and the cumulative

395 contribution rate is 85%. Selecting the first five principal components can  
 396 comprehensively reflect most of the information of hydrochemical. The proportion of  
 397 variance, cumulative proportion and rotation factor load matrix of single-type mineral  
 398 springs are shown in Table4.



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 400

Fig.5. Histogram of contribution rate

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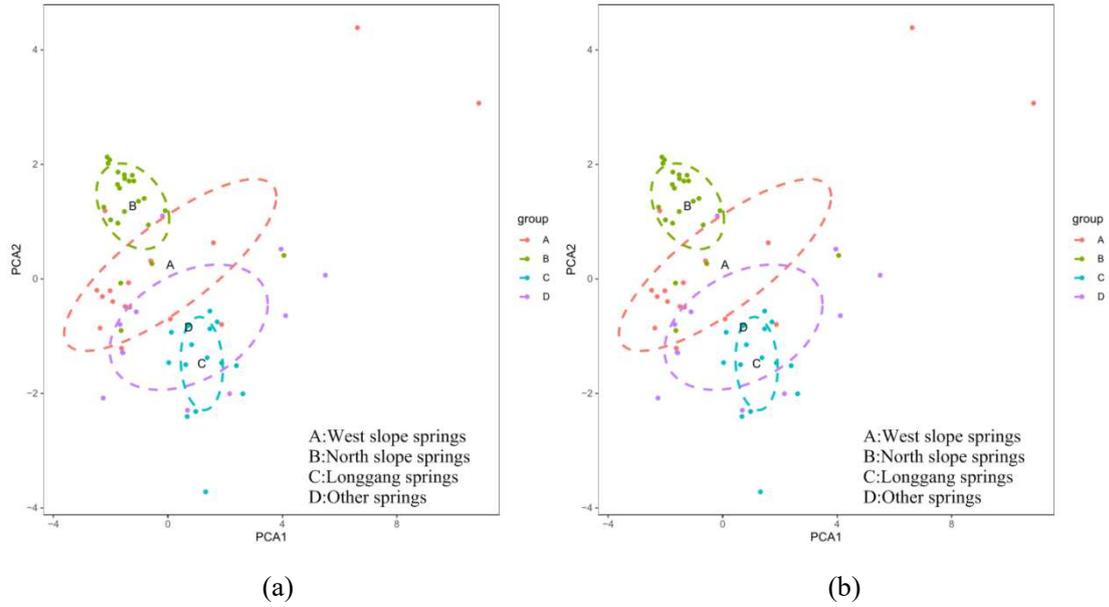
Table4. Rotation factor load matrix of PCA

Variables	PC1	PC2	PC3	PC4	PC5
K <sup>+</sup>	<b>0.331</b>	0.242	0.049	-0.097	0.193
Na <sup>+</sup>	<b>0.326</b>	0.246	0.011	-0.185	0.270
Ca <sup>2+</sup>	<b>0.331</b>	-0.247	0.144	0.026	-0.108
Mg <sup>2+</sup>	<b>0.353</b>	-0.020	-0.076	-0.038	-0.288
Cl <sup>-</sup>	0.126	-0.265	-0.483	-0.012	<b>0.579</b>
SO <sub>4</sub> <sup>2-</sup>	0.047	-0.390	0.063	0.178	<b>0.509</b>
HCO <sub>3</sub> <sup>-</sup>	<b>0.399</b>	0.076	-0.046	-0.054	-0.134
NO <sub>3</sub> <sup>-</sup>	0.082	-0.235	<b>0.818</b>	-0.195	0.204
TDS	<b>0.398</b>	0.113	-0.022	-0.010	0.002
TH	<b>0.391</b>	-0.071	-0.043	-0.076	-0.103
TFe	0.166	-0.054	0.134	<b>0.903</b>	-0.073
H <sub>2</sub> SiO <sub>3</sub>	0.056	<b>0.557</b>	0.050	0.247	0.258
F <sup>-</sup>	-0.155	<b>0.453</b>	0.200	0.014	0.242
Proportion of Variance	0.461	0.176	0.078	0.071	0.065
Cumulative Proportion	0.461	0.637	0.714	0.785	0.850

402 Table 4 shows that the contribution rate of the first factor is 46.1%, and it mainly  
 403 comprises K<sup>+</sup>, Na<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, HCO<sub>3</sub><sup>-</sup>, TDS, and TH; it represent the process that  
 404 mineral components in different types of rocks enter the groundwater, and various  
 405 hydrochemical components of groundwater increase with the increase in the water–

406 rock interaction time. Because of the different types of rock and soil, the flow length  
407 and the mineral contents in rock and soil, the factor loads of the main components in  
408 the first factor are relatively small. The contribution rate of the second factor is 17.6%,  
409 and it mainly comprises  $\text{H}_2\text{SiO}_3$  and  $\text{F}^-$  and represents the influence of volcanic  
410 activity and magmatic volatilization. In the volcanically active area, the release of F is  
411 closely related to deep magmatism and magma degassing. The solubility of silicate  
412 minerals increases with increase in temperature. In addition, the solubility of silicate  
413 minerals and the release capacity of F from rock and soil will also increase under  
414 weak alkaline conditions (Pachana, Zuddas et al. 2012, Sunkari, Abu et al. 2020).  
415 Hence,  $\text{H}_2\text{SiO}_3$  and  $\text{F}^-$  are the main components in the second factor. The contribution  
416 rate of the third factor is 7.8%, and it mainly comprises  $\text{NO}_3^-$ . Human activities are the  
417 main source of nitrate, and hence, the third factor represents the impact of human  
418 activities (Singh, Rishi et al. 2020). The contribution rate of the fourth factor is 7.1%,  
419 and it is mainly TFe, representing the high background value of iron in the rock  
420 matrix of Northeast China and the characteristics of pyroxene in trachyte and  
421 pantellerite that erupted in Holocene and are rich in Fe and poor in Mg (Li, Fan et al.  
422 2004). The contribution rate of the fifth factor is 6.5%, and it mainly comprises  $\text{Cl}^-$   
423 and  $\text{SO}_4^{2-}$ , representing evaporation concentration.

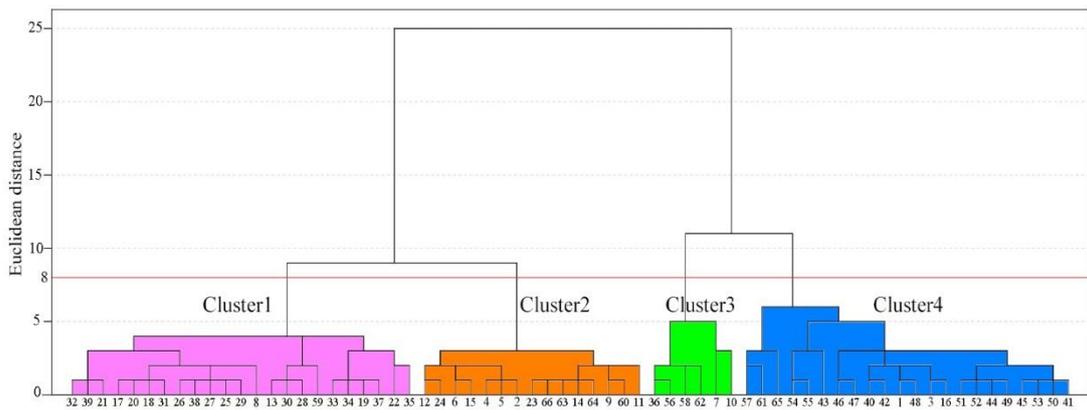
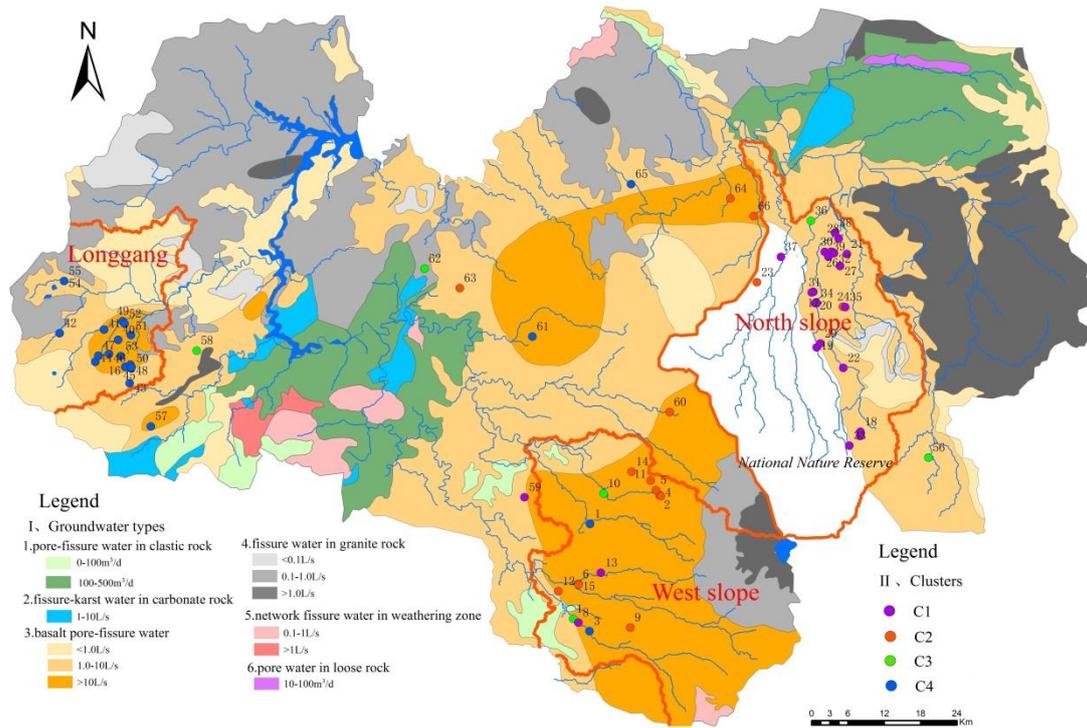
424 PC1, PC2, and PC3 are the three factors that have the greatest impact on the  
425 chemical components of regional single-type mineral water. To analyze the influence  
426 degree of the three principal factors on springs located in different  
427 spring-concentrated areas, the double projection diagram of load in PCA of each area  
428 was drawn by using R (Fig.6). Fig. 6 show that there is no significant difference in the  
429 influence degree of weathering-leaching processes (the first factor) and volcanic  
430 geology (the second factor) on the hydrochemical components of the West slope  
431 springs. Volcanic geology has a great influence on the formation of the hydrochemical  
432 components of the North slope springs, but less influence on the Longgang springs.  
433 This is attributed to the different active degrees of regional volcanic geology. The  
434 massive volcanic eruption time of Longgang Mountain Volcanic Group is about  
435 150,000 years ago. At present, the volcanic geological activity has weakened. The  
436 northern slope of Changbai Mountain has a developed fault structure, which  
437 comprises a ring structure and radial structure closely related to the giant volcano. The  
438 North slope springs are greatly affected by geological volcanic activity (Liu, Chu et al.  
439 2009). The nitrate produced by human activities (the third factor) has a relatively large  
440 impact on Longgang springs, but a relatively small impact on the springs in the other  
441 regions—this is consistent with the fact that there are relatively more human activities  
442 in Longgang mountain area.



**Fig.6.** Projection diagram of load

#### 4.2.3 Cluster analysis of single-type mineral springs

The Q-type cluster analysis of mineral water chemistry is a method to classify different types of samples according to the similarities and differences in hydrochemical components (Lambrakis, Antonakos et al. 2004, Ghesquière, Walter et al. 2015). Only according to the chemical composition of mineral water, all single-type mineral springs are regarded as a single type for Q-type cluster analysis. In the analysis, the selection of hydrochemical indexes is consistent with the PCA. A class comprises elements separated by a Euclidean distance less than 8; it is divided into four categories. Among these, cluster 1–2 and cluster 3–4 can be divided into two categories. On the whole, the content of hydrochemical components in cluster 3–4 is higher than that in cluster 1–2. The results of cluster analysis and the spatial distribution of mineral springs after clustering are shown in Fig.7, and the average contents of hydrochemical components of mineral springs in different clusters are listed in Table 5.



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**Fig.7.** Cluster analysis results and their spatial distribution

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**Table5.** Average contents of hydrochemical components in different clusters

cluster	n	K <sup>+</sup>	Na <sup>+</sup>	Ca <sup>2+</sup>	Mg <sup>2+</sup>	Cl <sup>-</sup>	SO <sub>4</sub> <sup>2-</sup>	HCO <sub>3</sub> <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>	TDS	TH	TFe	H <sub>2</sub> SiO <sub>3</sub>	F <sup>-</sup>
1	23	2.70	8.48	7.97	4.43	1.78	3.74	61.38	2.13	135.14	38.53	0.07	52.64	0.81
2	15	1.96	5.64	8.99	3.15	1.67	4.55	48.70	4.26	108.15	35.28	0.06	36.37	0.36
3	6	7.63	25.82	20.09	19.42	3.34	2.79	218.52	3.92	342.81	141.78	0.15	52.15	0.35
4	22	3.52	9.76	15.45	8.45	4.56	7.23	102.18	4.26	180.22	74.24	0.12	37.33	0.18

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Note: n is the number of samples, and the unit of each index is mg/L.

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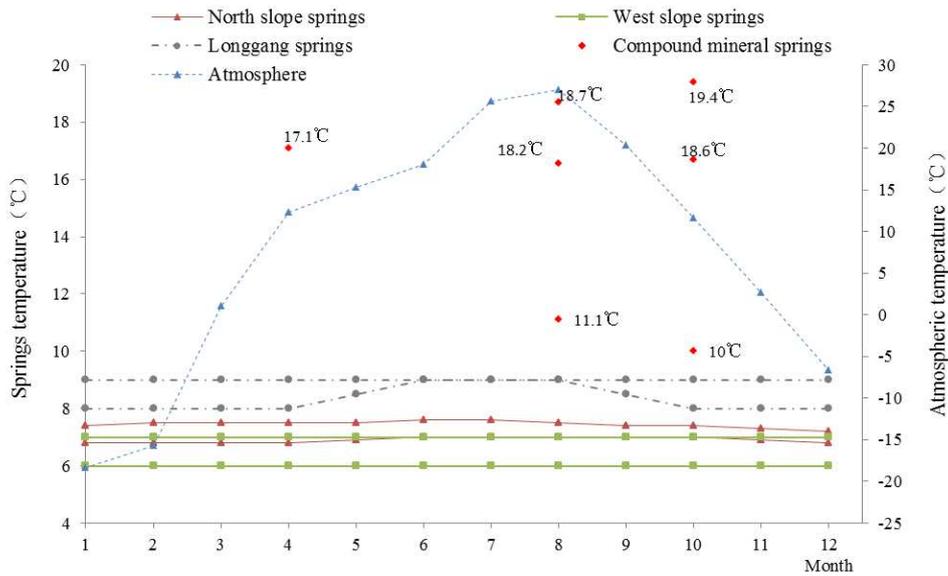
Figure 7 and Table 5 show that the springs in cluster 1 are mainly located in the North slope springs area, along the second White River and third White River from south to north, and most of the mineral springs in the North slope belong to cluster 1. The results show that the mineral springs in the upper reaches have a good hydraulic connection with those in the lower reaches, which are supplied with the groundwater

470 of the basalt platform in the south. The volcanic geological process (the second  
471 factor in PCA) has a greater impact on this cluster of mineral springs. Most of the  
472 springs in cluster 2 are located in the West slope springs area. Most of these springs  
473 are supplied by basalt pore-fissure water with a flow rate of more than 10 L/s and are  
474 located near the river. The contents of most hydrochemical components are low, while  
475 those of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  are relatively high, indicating that they are closely related to  
476 surface water. The springs in cluster 3 are scattered and few, and most of the  
477 components are present in relatively high contents. Some of the mineral springs are  
478 located in the bedrock area and the junction area between the basalt stratum and other  
479 types of strata, which are far from the supply area and concentrated areas of mineral  
480 springs. It shows that the flow path of this type of mineral spring is long, resulting in  
481 the inclusion of a large number of minerals from the rock and soil into groundwater.  
482 Most of the springs in cluster 4 are located in the Longgang springs area, and the  
483 distribution is relatively concentrated. Most of these springs are supplied by basalt  
484 pore-fissure water with a flow rate of more than 10 L/s. The contents of  $\text{H}_2\text{SiO}_3$  and  
485  $\text{F}^-$  are low, whereas those of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ , and  $\text{NO}_3^-$  are high. Human activities (the  
486 third factor in PCA) has a greater impact on this cluster. According to the topography,  
487 the area where Longgang springs are located is a valley surrounded by mountains on  
488 three sides. From the edge to the center of the valley, the amount of basalt pore-fissure  
489 water increased from less than 0.1 L/s to more than 10 L/s. It shows that the mineral  
490 spring classified as cluster 4 in Longgang mountain area is the atmospheric  
491 precipitation of the mountains around the valley that infiltrates into the ground, flows  
492 in the basalt under the action of topography, and is exposed at the bottom of the  
493 valley.

494 The results of cluster analysis show that the majority of North slope springs are  
495 in cluster 1; in the West slope springs, cluster 2 accounts for the majority, and all the  
496 Longgang springs are in cluster 4, which shows that the hydrochemical characteristics  
497 within a spring cluster are similar, and different spring clusters show relatively large  
498 differences. The result of cluster analysis is consistent with the division of the  
499 concentrated distribution area of mineral springs. Because the hydrochemical index is  
500 the only factor to be considered in cluster analysis, it shows that the hydrochemical  
501 composition of mineral water shows good consistency with the concentrated areas of  
502 mineral springs as determined by topography and landform. The division of the  
503 concentrated areas of mineral springs conforms to the basin conditions determined by  
504 topography and landform, as well as the hydrochemical characteristics of mineral  
505 springs.

#### 506 4.2.4 Formation of compound mineral springs

507 In the Gibbs map, some compound mineral springs are located at the periphery of  
508 the model, which shows that their formation is affected by special conditions.  
509 Tectonization and volcanic activity are strong in the study area. The ring structure and  
510 radial structure closely related to the volcanic activity are distributed around the main  
511 peak of the Changbai Mountain. There are many deep faults extending north–south,  
512 north–east, and north–west. Most of the groundwater in the deep fault is supplied by  
513 surface water and underground hot water, thereby affecting the mineral water quality  
514 (Chen, Ai et al. 2019). In the process of magmatism and metamorphism, CO<sub>2</sub> is  
515 released, lowering the pH value of groundwater. In addition, the temperature in the  
516 deep underground is higher. These factors increase the strength of the water–rock  
517 interaction. The groundwater circulation time in the deep fault is long, and a large  
518 number of minerals in the rock enter into the groundwater, increasing the content of  
519 each component and the TDS in the groundwater. According to the statistical analysis  
520 of the annual variation in some mineral spring temperatures (Fig.8) , the annual  
521 variation of regional atmospheric temperature is large, and the annual variation of  
522 mineral springs temperature is small. The temperature of single-type mineral springs  
523 is between 6 and 9°C, and the temperature of compound mineral springs is higher  
524 than that of single-type mineral springs. The temperature difference between different  
525 compound mineral springs is also large. Through field observation, it can be seen that  
526 free CO<sub>2</sub> overflow occurs in the compound mineral spring, and travertine is attached  
527 to the spring mouth, and the rock surface has red iron oxide, and the pores are  
528 relatively developed (Fig.9) . The above analysis shows that the compound mineral  
529 spring is supplied by the underground hot water in the deep fault formed by  
530 geological volcanic activity, and the difference in hydrothermal cycle characteristics is  
531 the main reason for the temperature difference between different compound mineral  
532 springs.

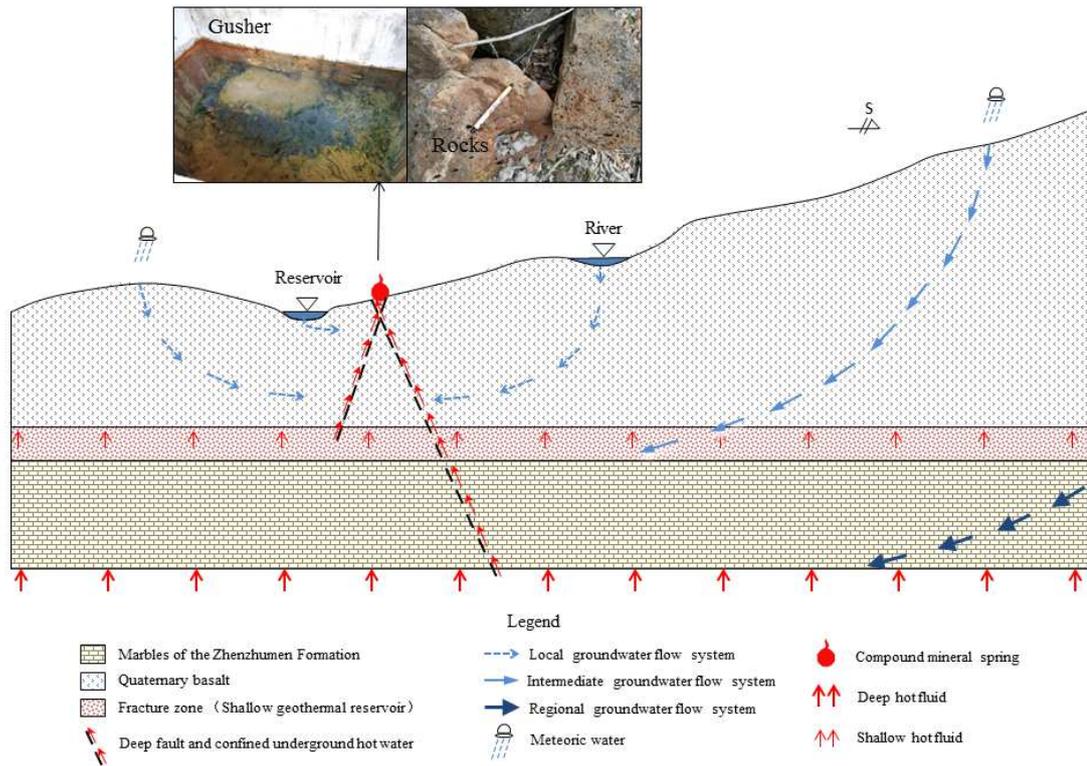


**Fig.8.** Annual variation of mineral spring temperature

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535 Based on the regional geological structure conditions, a high TDS compound  
 536 mineral spring near the north-south fault zone was selected. The mineral spring is  
 537 located at Site IV (Table 2). Its genesis was analyzed. The spring is located in the  
 538 second White River valley on the North slope of Changbai Mountain. Because of the  
 539 volcanic geological process of Changbai Mountain, north-south zonal geothermal  
 540 reservoirs are buried underground. The geothermal reservoirs include shallow  
 541 geothermal reservoirs in the fracture zone under the basalt caprock and a deep  
 542 geothermal reservoir formed by the development of karst and structural fractures in  
 543 the marble. Regional magma and magma chamber provide a stable heat source for  
 544 geothermal reservoirs. The genetic diagram was drawn according to the geothermal  
 545 geological conditions (Fig.9).



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**Fig.9.** Genetic diagram of compound mineral spring

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Figure 9 shows that there are local, intermediate, and regional groundwater flow systems in the region. Around the Changbai Mountain crater, radial and annular fractures are dense. Atmospheric precipitation and Tianchi lake water infiltrate into the rock fissures of high temperature and high pressure. Convection occurs under the action of temperature and pressure and rises along the marble fracture, forming a regional groundwater flow system from the main peak of Changbai mountain to the intersection of the fault structures in the river valley. The regional groundwater flow system is mainly affected by volcanic geology. The groundwater in the upper part of the basalt platform is recharged by precipitation and the surface runoff of the giant volcanic cone in the upstream. Under the control of topography, the groundwater flows downstream and accumulates in the basalt fracture zone, forming an intermediate groundwater flow system from the upper part of the basalt platform to the discharge area of the lower valley. The intermediate groundwater flow system is mainly controlled by topography and geological structural conditions. Mineral springs are located in the valley with a low terrain. Atmospheric precipitation infiltrates into the basalt and gathers in the valley under the action of gravity. There are rivers and reservoirs in the upstream and downstream of mineral spring. The water levels of surface water and reservoir are higher than the groundwater level; hence, they supply groundwater and form a local groundwater flow system. The local groundwater flow

567 system is mainly controlled by the terrain conditions. The regional and intermediate  
568 groundwater flow through the geothermal reservoirs and is heated; it then rises along  
569 the fracture connected with the mineral springs and mixes with the local groundwater  
570 flow with a lower temperature to form compound mineral springs. According to the  
571 above analysis, a part of the compound mineral springs in the study area is the product  
572 of dilution and subsequent mixing of the confined groundwater formed under the  
573 high-temperature and high-pressure conditions in the deep fault with the shallow  
574 groundwater as the confined groundwater rises along the fracture.

## 575 5. Conclusions

576 There are single metasilicic-acid mineral springs and compound mineral springs  
577 in Changbai Mountain area. Most of the single-type mineral springs are distributed in  
578 three areas. According to the types and distribution of mineral springs, the springs can  
579 be classified into five types: West slope springs, North slope springs, Longgang  
580 springs, other regional mineral springs, and compound mineral springs. The average  
581 values of hydrochemical indexes of all types of mineral springs show good  
582 consistency, and the hydrochemical type is mainly the  $\text{HCO}_3\text{-Ca}$  type. The content of  
583 metasilicic-acid in all mineral springs meets the standard of “Drinking natural mineral  
584 water” GB8537—2018, and the content of most hydrochemical indexes of compound  
585 mineral springs is 1–2 orders of magnitude higher than that of single-type mineral  
586 springs.

587 Using the general and special methods to explore the hydrochemical origin of  
588 mineral springs, it is concluded that the formation of single metasilicic-acid mineral  
589 springs is controlled by rock weathering, and that silicate rock is the main medium in  
590 the water–rock interaction. The hydrochemical formation mechanism of the  
591 compound mineral spring is different and shows the characteristics of high  
592 temperature and high  $\text{CO}_2$ . The formation of compound mineral springs is affected by  
593 the deep geothermal reservoirs of high temperature and high pressure and long-term  
594 groundwater circulation. It is the mixed product of the deep acidic hot water rising  
595 along the fault and mixing with the shallow groundwater.

596 PCA was used to judge the source of hydrochemical components and the  
597 influence of each factor on the hydrochemical components of mineral springs in  
598 different regions. The first five principal factors representing 85% contribution rate  
599 were extracted. The results show that the water–rock interaction of groundwater,  
600 regional volcanic geological process caused by volcanic activity and magmatic  
601 volatilization, human activities, high background value of iron in the rock, and  
602 evaporation concentration are the main reasons for the formation of the chemical

603 characteristics of mineral water, with the contribution rates of the factors are 46.1%,  
604 17.6%, 7.8%, 7.1%, and 6.5%, respectively. Volcanic geological process has a great  
605 influence on the formation of the chemical components of mineral water on the North  
606 slope springs, and the Longgang springs are greatly affected by human activities.

607 Using cluster analysis, the samples are classified into four categories: the mineral  
608 springs in cluster 1 are mainly located in the North slope springs area, and the mineral  
609 springs in cluster 1 on the North slope springs are supplied by the same groundwater  
610 aquifer. Most of the mineral springs on the West slope springs are in cluster 2, which  
611 is closely related to surface water. Some of the springs in cluster 3 are scattered at the  
612 junction of different strata. Longgang springs are all in cluster 4 and are distributed at  
613 the bottom of the valley, where enrichment and concentrated exposure of meteoric  
614 water is observed. The cluster analysis shows that the clustering results of only  
615 considering hydrochemical components are consistent with the division of the  
616 concentrated distribution area of mineral springs.

617 All the analysis results show that there are great differences between compound  
618 mineral springs and single-type mineral springs in the study area in terms of the  
619 hydrochemical characteristics and genesis. The compound mineral spring is a mixture  
620 of the confined groundwater formed in deep fractures and shallow cold groundwater,  
621 and the single-type mineral spring originates from meteoric water leaching the basalt.  
622 Mineral springs are distributed over the whole study area and concentrated in specific  
623 areas. Different hydrogeological conditions and human activities have different  
624 impacts on the mineral springs in different regions, and there are obvious differences  
625 in their hydrochemical characteristics.

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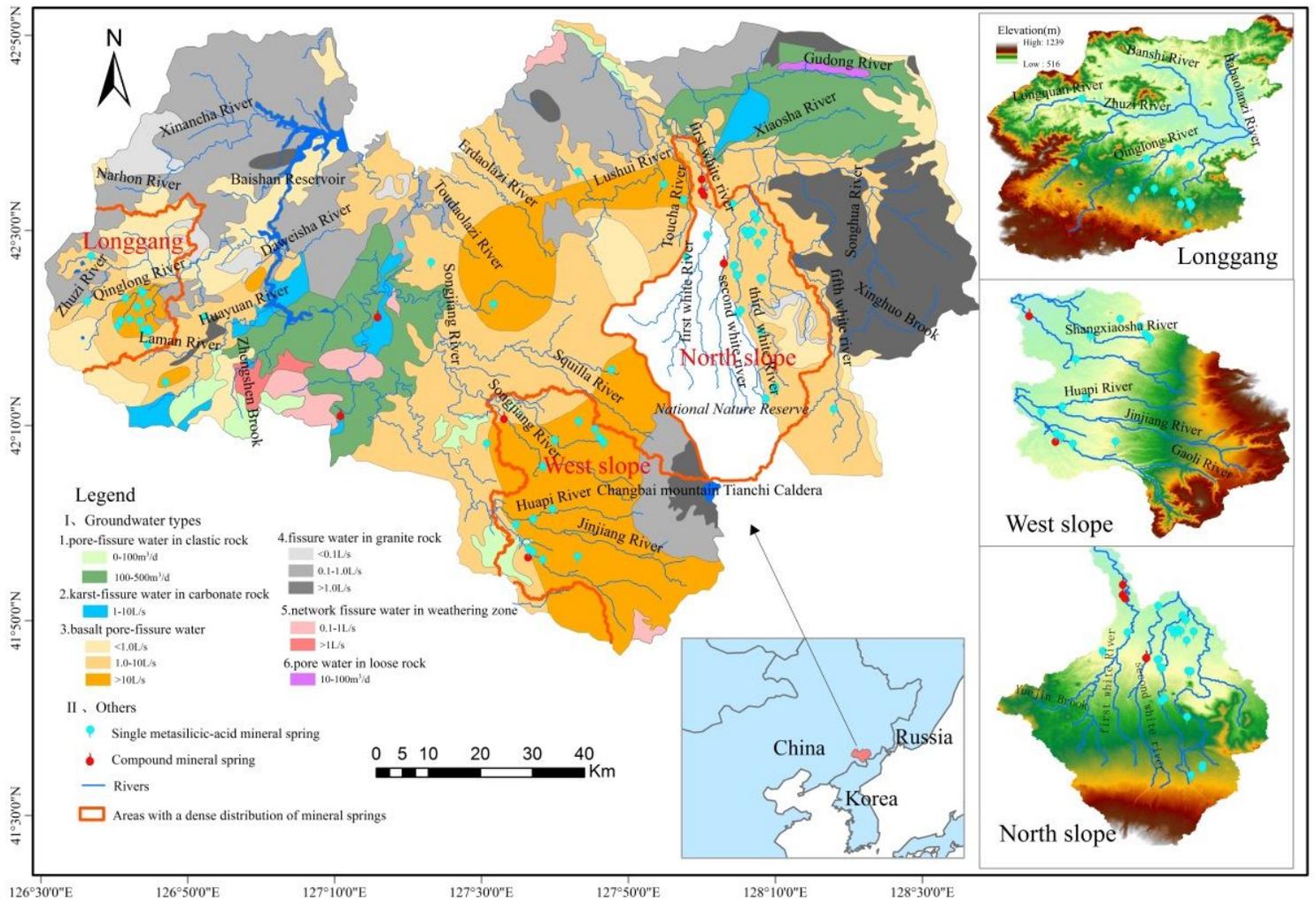
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# Figures



**Figure 1**

Hydrogeology conditions and sampling springs Note: The designations employed and the presentation of the material on this map do not imply the expression of any opinion whatsoever on the part of Research Square concerning the legal status of any country, territory, city or area or of its authorities, or concerning the delimitation of its frontiers or boundaries. This map has been provided by the authors.

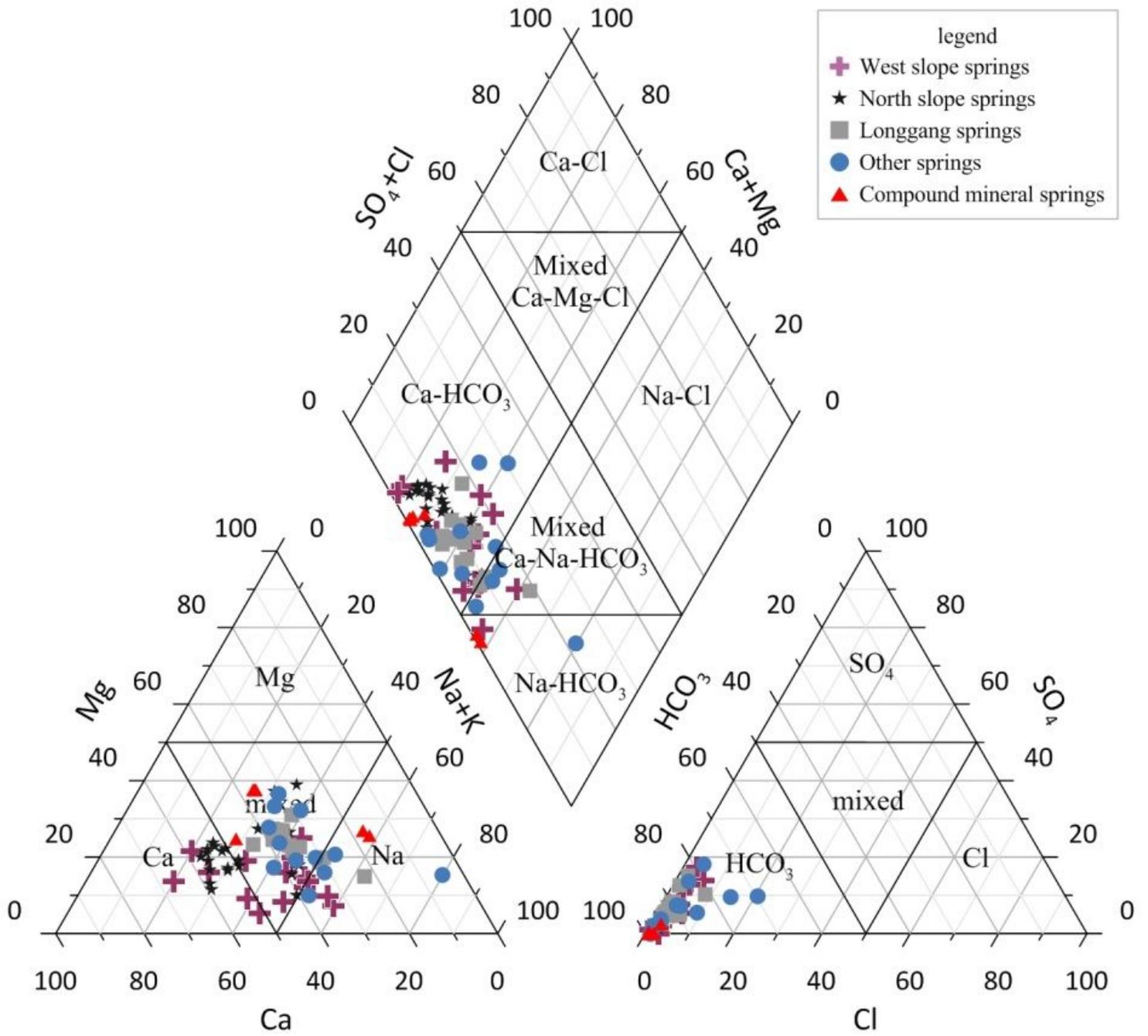


Figure 2

Piper trigram diagram of mineral water

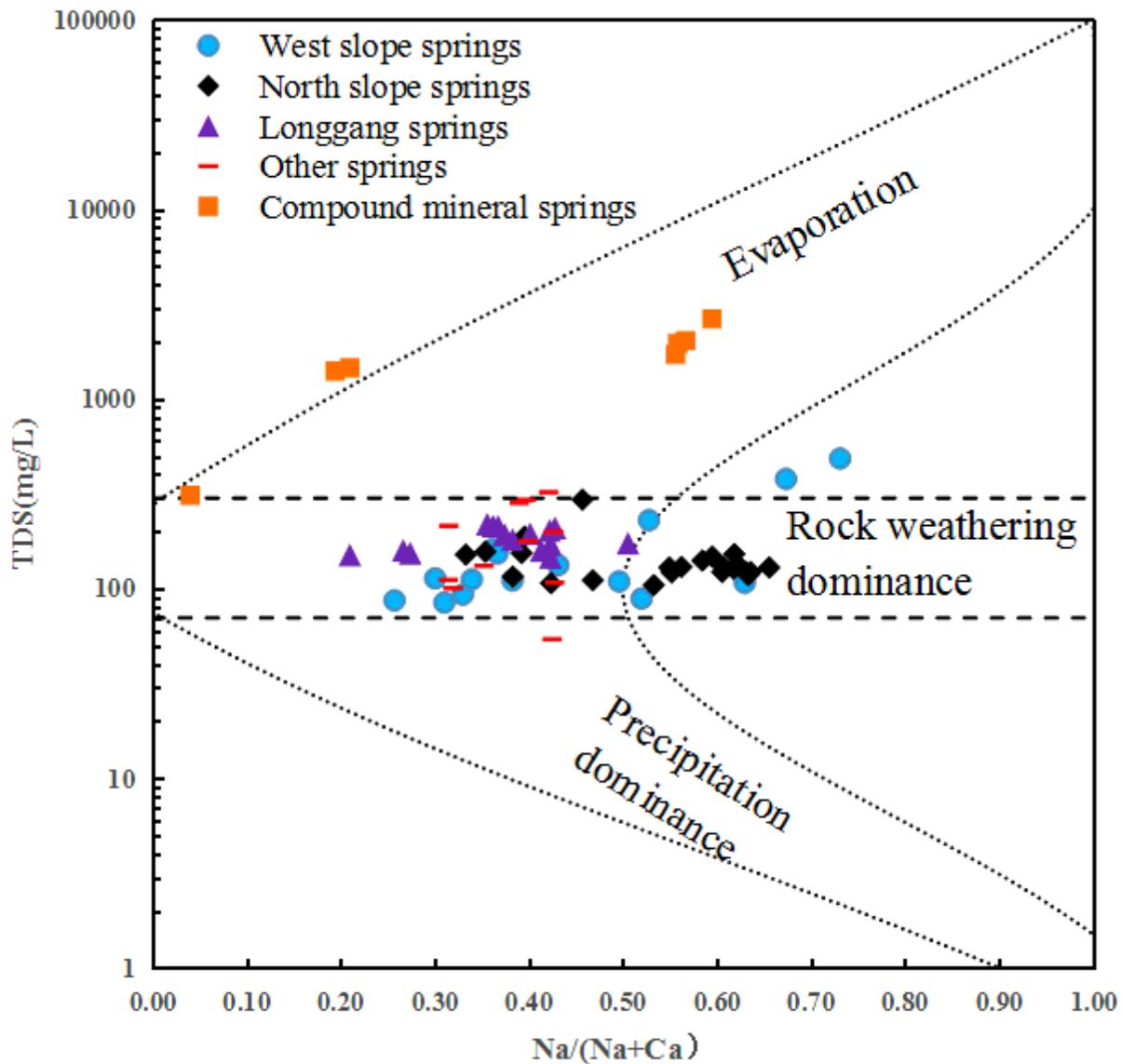
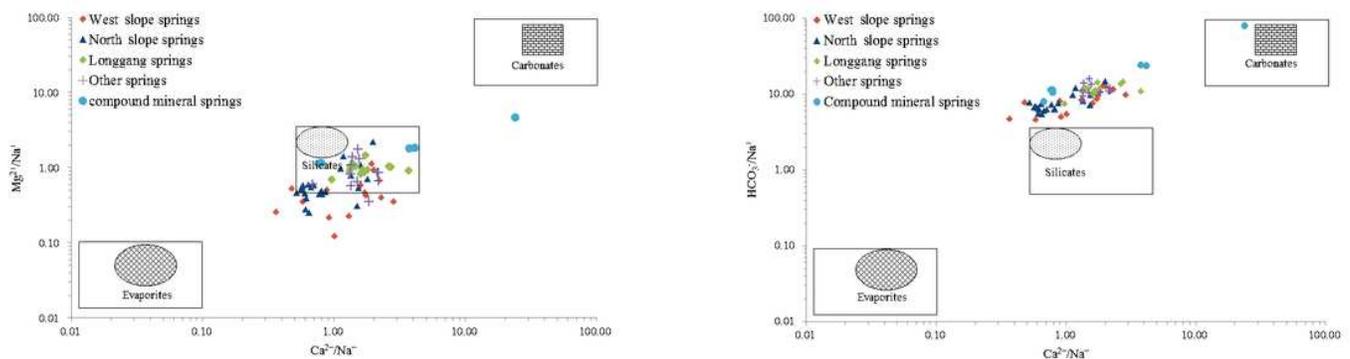


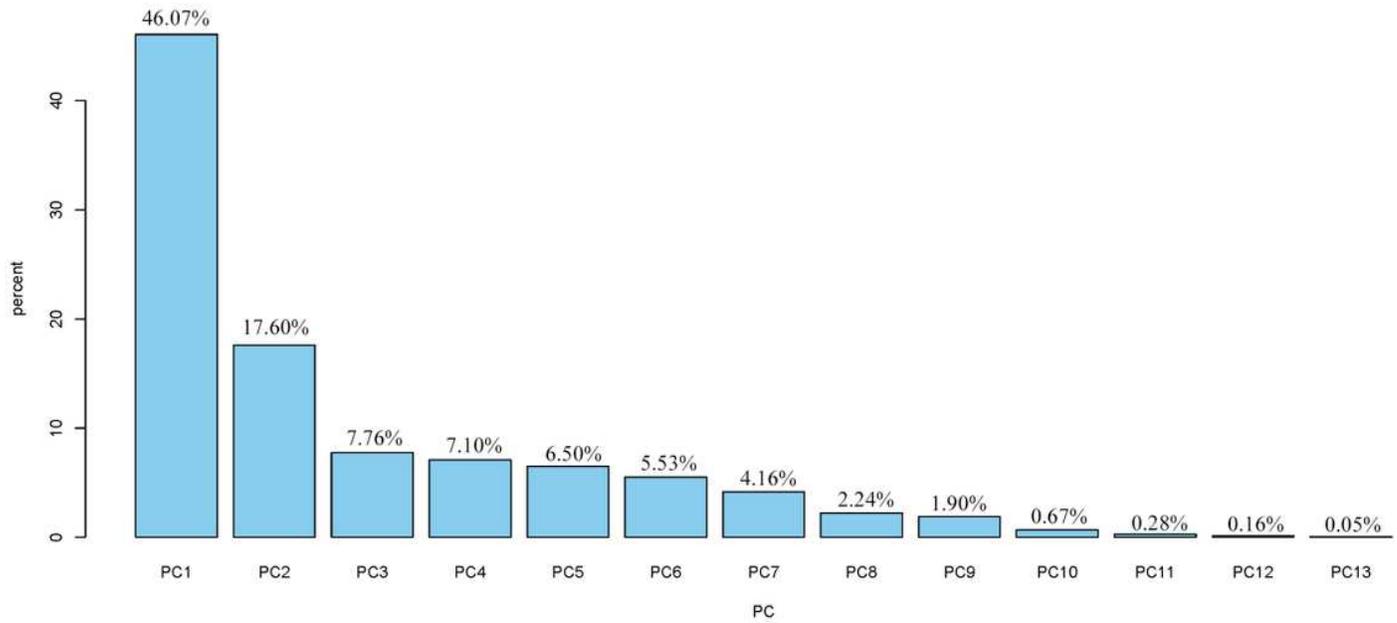
Figure 3

Gibbs diagram of springs



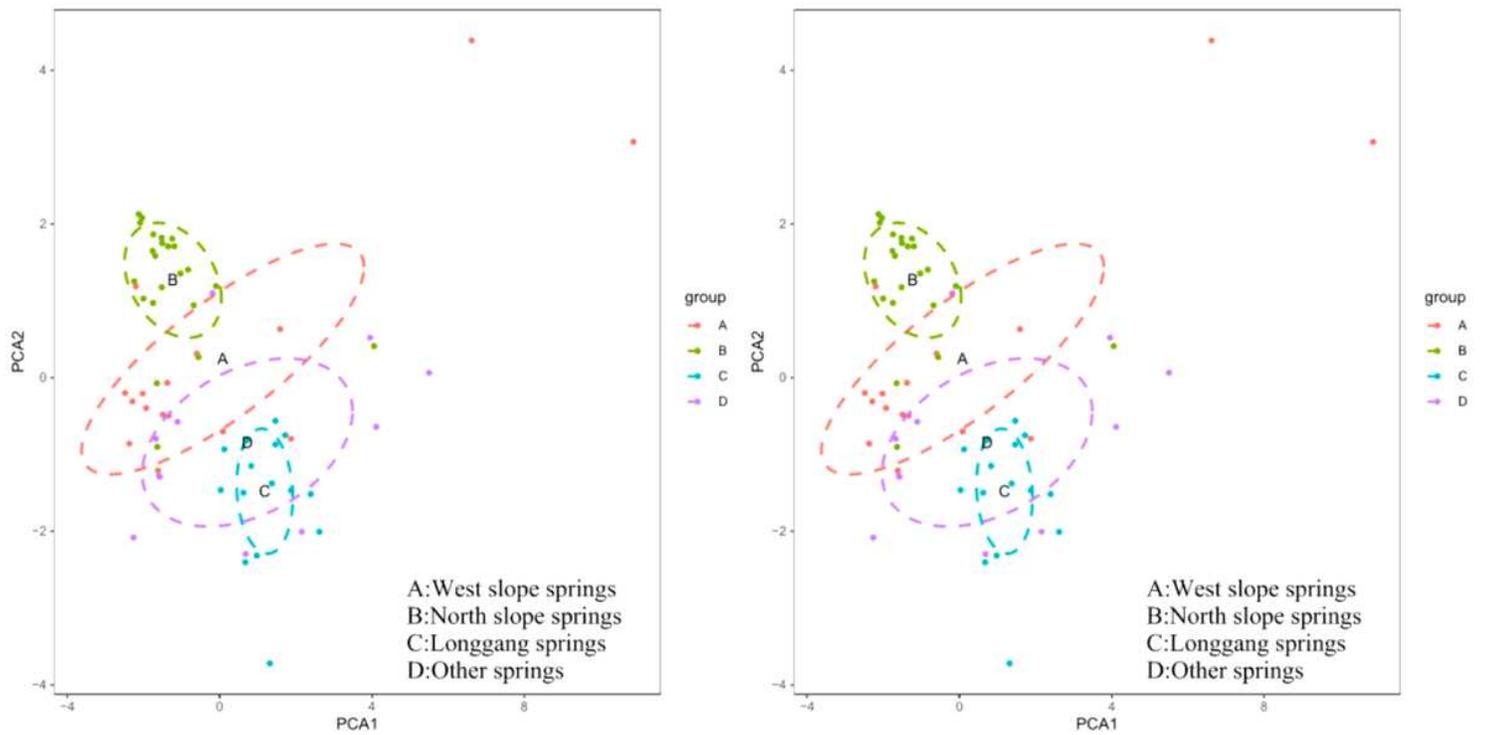
**Figure 4**

Lithologic end element ratios diagram of different springs



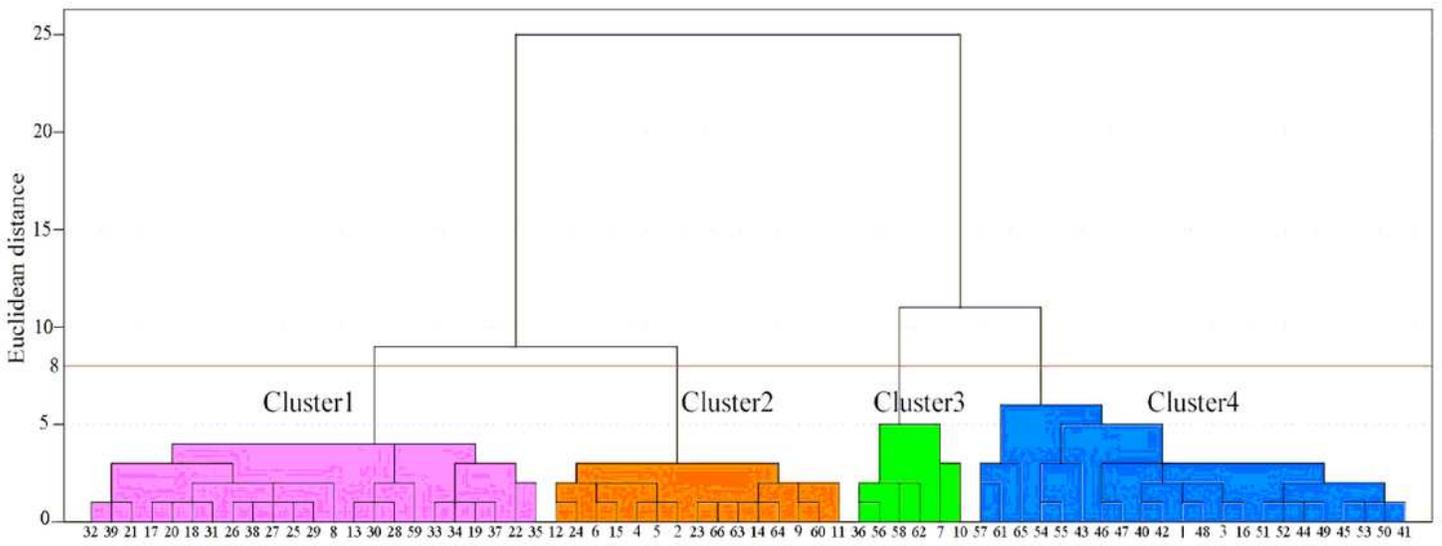
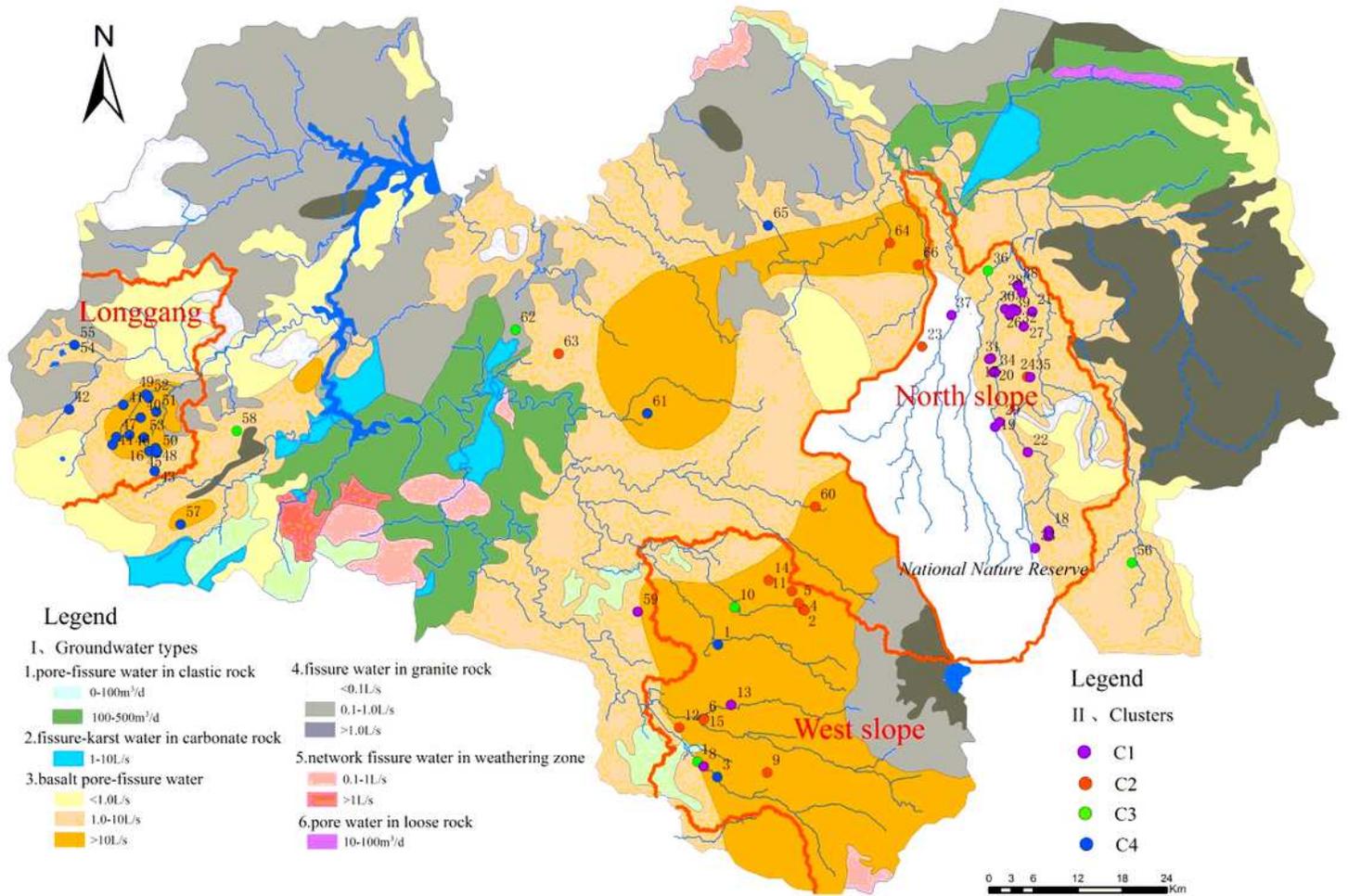
**Figure 5**

Histogram of contribution rate



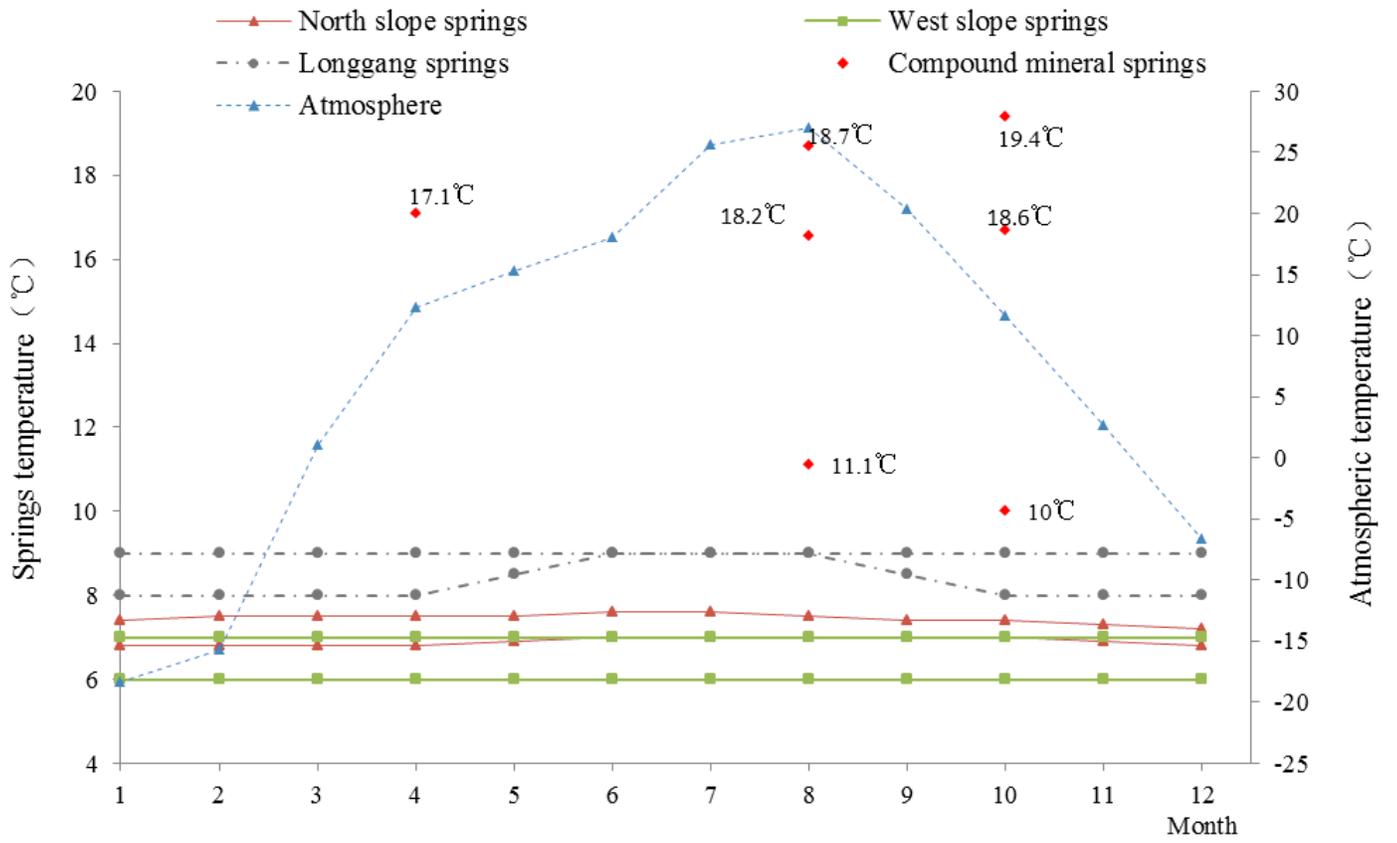
**Figure 6**

Projection diagram of load



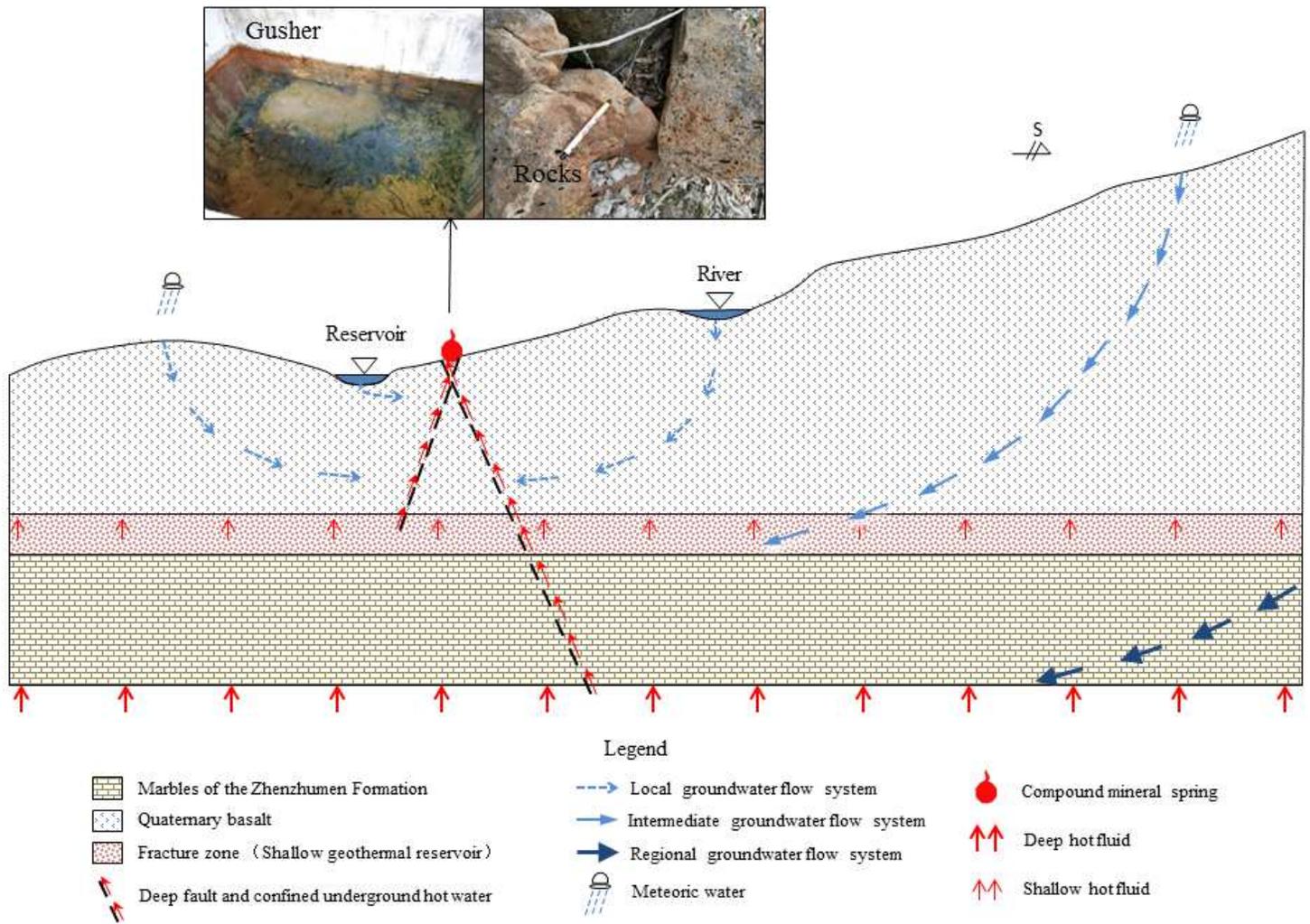
**Figure 7**

Cluster analysis results and their spatial distribution Note: The designations employed and the presentation of the material on this map do not imply the expression of any opinion whatsoever on the part of Research Square concerning the legal status of any country, territory, city or area or of its authorities, or concerning the delimitation of its frontiers or boundaries. This map has been provided by the authors.



**Figure 8**

Annual variation of mineral spring temperature



**Figure 9**

Genetic diagram of compound mineral spring