

Photosynthetic linear electron flow drives CO₂ assimilation in maize leaves

Ginga Shimakawa

Kobe University

Chikahiro Miyake (✉ cmiyake@hawk.kobe-u.ac.jp)

Kobe University <https://orcid.org/0000-0002-2426-2377>

Article

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1 ***Corresponding Author:** Chikahiro Miyake

2 Department of Biological and Environmental Science, Faculty of Agriculture, Graduate School of
3 Agricultural Science, Kobe University, 1-1 Rokkodai, Nada, Kobe 657-8501, Japan
4 Fax: +81-78-803-5851; e-mail: cmiyake@hawk.kobe-u.ac.jp

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6 **Title:** Photosynthetic linear electron flow drives CO₂ assimilation in maize leaves

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8 **Running head:** Regulation of photosynthesis in C₄ plants

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10 **Author's Addresses:** Ginga Shimakawa^{1,#} and Chikahiro Miyake^{1,2,*}

11 ¹Department of Biological and Environmental Science, Faculty of Agriculture, Graduate School of
12 Agricultural Science, Kobe University, 1-1 Rokkodai, Nada, Kobe 657-8501, Japan

13 ²Core Research for Environmental Science and Technology, Japan Science and Technology Agency, 7
14 Goban, Chiyoda, Tokyo 102-0076, Japan

15

16 **Footnotes:** #Present address: Research Center for Solar Energy Chemistry, Osaka University, 1-3
17 Machikaneyama, Toyonaka, Osaka 560-8631, Japan

18 **Abstract**

19

20 Photosynthetic organisms commonly develop the strategy to keep the reaction centre chlorophyll of
21 photosystem I, P700, oxidised for preventing the generation of reactive oxygen species in excess light
22 conditions. In photosynthesis of C₄ plants, CO₂ concentration is kept at higher levels around ribulose
23 1,5-bisphosphate carboxylase/oxygenase (Rubisco) by the cooperation of the mesophyll and bundle
24 sheath cells, which enables them to assimilate CO₂ at higher rates and to survive under drought stress.
25 However, the regulatory mechanism of photosynthetic electron transport for P700 oxidation is still
26 poorly understood in C₄ plants. Here we assessed gas exchange, chlorophyll fluorescence,
27 electrochromic shift, and near infrared absorbance in the intact leaves of NADP-malic enzyme subtype
28 of C₄ plants maize in a comparison with the C₃ plant field mustard. Instead of the alternative electron
29 sink due to photorespiration, photosynthetic linear electron flow was strongly limited between
30 photosystems I and II dependent on the proton gradient across the thylakoid membrane (ΔpH) in
31 response to the suppression of CO₂ assimilation in maize. The increase of ΔpH for P700 oxidation was
32 caused by the regulation of proton conductance of chloroplast ATP synthase but not by promoting
33 cyclic electron flow, which was supported by linear relationships among CO₂ assimilation rate, linear
34 electron flow, P700 oxidation, ΔpH , and the oxidation rate of ferredoxin. At the scale of intact leaves,
35 the ratio of PSI to PSII was estimated almost 1:1 in both C₃ and C₄ plants. Overall, the photosynthetic
36 electron transport was regulated for P700 oxidation in maize through the same strategies as in C₃
37 plants only except for the capacity of photorespiration despite the structural and metabolic
38 differences in photosynthesis between C₃ and C₄ plants.

39

40 **Keyword:** Photosynthesis; Linear electron flow; C₄ plants; P700 oxidation; Ferredoxin

41 Introduction

42

43 In chloroplasts of plant leaves, photosynthetic CO₂ assimilation is driven in the Calvin-Benson cycle
44 utilizing NADPH and ATP produced by light energy¹. In the photosynthetic electron transport system,
45 light energy is absorbed by chlorophyll in photosystems (PS) I and II, producing the photo-oxidised
46 reaction centre chlorophylls, P700⁺ and P680⁺, to initiate photosynthetic linear electron flow (LEF)
47 from PSII to PSI *via* the plastoquinone (PQ) pool, the cytochrome (Cyt) *b₆/f* complex, and plastocyanin
48 (PC). On the electron acceptor side of PSI, NADP⁺ is reduced to NADPH using electrons from PSI *via*
49 ferredoxin (Fd) and Fd-NADP⁺ reductase. In PSII, H⁺ is released by H₂O oxidation in the luminal side of
50 the thylakoid membrane; in the Cyt *b₆/f* complex, the Q-cycle pumps stromal H⁺ to the luminal side.
51 Both generate a proton gradient (Δ pH) across the thylakoid membrane to produce ATP *via* the
52 chloroplast ATP synthase. In the Calvin-Benson cycle, ribulose 1,5-bisphosphate (RuBP)
53 carboxylase/oxygenase (Rubisco) catalyses the carboxylation of RuBP to produce two molecules of 3-
54 phosphoglycerate (3-PGA), which are then metabolised in the Calvin-Benson cycle with NADPH and
55 ATP. In this process, RuBP is regenerated. On the other hand, Rubisco competitively catalyses the
56 oxygenation of RuBP, when available CO₂ is limited in the presence of sufficient O₂, to produce 3-PGA
57 and 2-phosphoglycolate (2-PG). 2-PG is finally metabolised to 3-PGA in cooperation with the
58 chloroplasts, peroxisomes, and mitochondria using Fd⁻ and ATP. This is the so-called photorespiration,
59 in which CO₂ is released from glycine in the mitochondria. The production and consumption of both
60 NADPH and ATP are normally balanced to poise the redox state of the photosynthetic electron
61 transport system in C₃ plants^{2,3}. In C₄ plants, the capacity of photorespiration had been degenerated
62 during the evolutionary history^{4,5}, instead of the CO₂ concentrating mechanism, where CO₂ is
63 incorporated into phosphoenolpyruvate (PEP) in mesophyll cells and then transported in the form of
64 malate into bundle sheath cells that specifically express Rubisco⁶. Malate is converted to CO₂ and
65 pyruvate around Rubisco by NADP-malic enzyme (ME) with NADP⁺ as the electron acceptor, and PEP
66 is regenerated in mesophyll cells theoretically with two additional ATP^{7,8}. Because of the structural
67 and metabolic complexities, the regulation of photosynthetic electron transport has been poorly
68 understood in C₄ plants.

69 Oxidation of P700 is a universal physiological response in photosynthetic organisms to
70 prevent photo-oxidative damage derived from reactive oxygen species (ROS) to PSI by dissipating
71 excess light energy as heat⁹ because the inhibition of PSI can be a lethal event for photosynthetic
72 organisms¹⁰. Actually, P700 is kept oxidised by a variety of molecular mechanisms in response to
73 excess light conditions such as high light and CO₂ limitation¹¹. In C₃ plants, P700 oxidation is supported
74 on the donor side by the suppression of electron transport in the Cyt *b₆/f* complex, dependent on Δ pH

75 ^{12, 13}. On the acceptor side, photorespiration replaces CO₂ assimilation, to function as an electron sink
76 for P700 oxidation under limited CO₂ conditions ¹⁴. In C₄ plants, there is little electron sink by
77 photorespiration even at the CO₂ compensation point, different from C₃ plants ^{5, 15}. It is still not clear
78 how P700 remains oxidised in C₄ plants when CO₂ assimilation is suppressed. One important
79 hypothesis that should be tested is that cyclic electron flow around PSI (CEF) functions to make ΔpH
80 to keep P700 oxidised in C₄ plants. Since CEF is mediated by the electron transport from Fd⁻ to PQ,
81 theoretically it is capable of pumping H⁺ from the stroma to the lumen of the thylakoid membrane in
82 the Q-cycle; this also results in an additional ATP production that is not linked to NADP⁺ reduction ¹⁶.
83 Especially in C₄ plants, the ratio of PSI to PSII is much higher in isolated bundle sheath cells than in the
84 mesophyll cells, which gives the hypothesis that CEF in bundle sheath cells contributes not only to
85 keeping P700 oxidised but also to meeting the additional ATP demand for C₄ photosynthesis.
86 Unfortunately, there is not currently a method available to directly measure the electron transport
87 rate *via* CEF. The CEF activity has been indirectly estimated mainly from the quantum yield of PSI,
88 which is inevitably under/overestimated dependent on the redox state of PC ². Therefore, the extent
89 of CEF activity remains controversial.

90 In this study, we evaluated gas exchange, chlorophyll fluorescence, electrochromic shift
91 (ECS), and near infrared (NIR) absorbance in the NADP-ME subtype of C₄ plants maize to investigate
92 how photosynthetic electron transport is regulated for P700 oxidation. Although maize did not show
93 the significant capacity of photorespiration, P700 oxidation was tightly coupled with LEF like the C₃
94 plant field mustard (komatsuna, hereafter mustard), which was supported by the linear relationships
95 among CO₂ assimilation rate, LEF, P700 oxidation, ΔpH, and the Fd⁻ oxidation rate. The ratio of PSI to
96 PSII was estimated *in vivo* approximately 1:1 in the both C₃ and C₄ plants. These results suggest that
97 maize regulates photosynthetic electron transport for P700 oxidation tightly associated with LEF like
98 C₃ plants but strongly relies it on the ΔpH-dependent suppression of electron transport on the donor
99 side of PSI instead of photorespiratory electron sink on the acceptor side.

100 **Results**

101

102 Photosynthetic CO₂ assimilation and dark respiration were analysed in both the C₃ plant mustard and
 103 the C₄ plant maize simultaneously with chlorophyll fluorescence, ECS, and NIR absorbance. In this
 104 study, we analysed *in vivo* photosynthetic parameters at a constant light intensity and different CO₂
 105 partial pressures to simply investigate the effects of limitation of electron sink on the photosynthetic
 106 electron transport. Additionally, the C₃ and C₄ intact leaves were measured at atmospheric (21 kPa)
 107 and low (1 kPa) O₂, where photorespiration is inhibited. We note that in maize CO₂ is first incorporated
 108 into PEP in mesophyll cells, different from C₃ plants, but the decrease of the CO₂ partial pressure in
 109 the intercellular space (Ci) finally results in the limitation of the carboxylation reaction of Rubisco in
 110 bundle sheath cells.

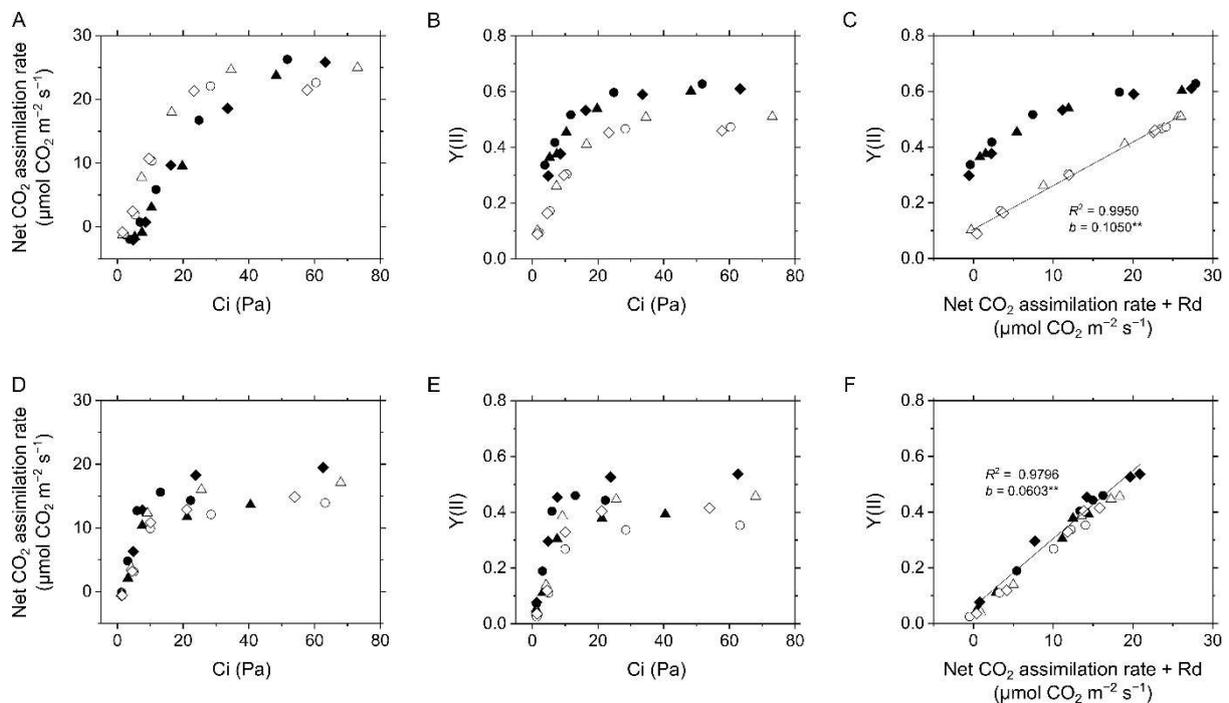


Fig. 1. Photosynthetic CO₂ assimilation and linear electron flow in the C₃ plant mustard (A–C) and the C₄ plant maize (D–F). (A, D) Net CO₂ assimilation rate at various intercellular CO₂ partial pressures (Ci). (B, E) Effective quantum yield of PSII, Y(II), at various Ci. (C, F) Relationship of Y(II) with CO₂ assimilation rate. Dark respiration rate is presented as Rd. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent estimated linear regression of the data at 1 kPa (C) and 21 kPa O₂ (F) (R^2 , coefficient of determination). The y -intercepts (b) were tested based on the null hypothesis: ** $p < 0.005$.

111 In mustard, the net CO₂ assimilation rate was higher at 1 kPa than at 21 kPa O₂ when CO₂
 112 assimilation was limited by Ci (Fig. 1A), whereas the effective quantum yield of PSII, Y(II), was kept
 113 high uncoupled from CO₂ assimilation (Fig. 1B and C). However, maize did not show any difference in
 114 net CO₂ assimilation and Y(II) between different O₂ partial pressures (Fig. 1D and E), which is in
 115 accordance with a small contribution, if any, of photorespiration to the capacity of electron sink. As a
 116 result, Y(II) has a linear relationship with the sum of net CO₂ assimilation rate and dark respiration rate
 117 (Rd) in maize (Fig. 1F). These typical photosynthetic characteristics of C₃ and C₄ plants have already

118 been established in previous studies, and extra Y(II) in mustard is known to be due almost exclusively
 119 to photorespiration^{5, 15, 17, 18, 19, 20}. In the other word, Y(II) reflects LEF mainly derived from CO₂
 120 assimilation and photorespiration in angiosperms^{13, 21, 22}. The inferred reduction level of the PQ pool
 121 (1 – qL) and non-photochemical quenching (NPQ) were also calculated from chlorophyll fluorescence
 122 in mustard and maize, both of which increased at low Ci (Supplemental Fig. S1). In mustard, limiting
 123 photorespiration at 1 kPa O₂ caused the further increase of 1 – qL at low Ci, whereas NPQ slightly
 124 decreased in this condition (Supplemental Fig. S1A and B).

125 Next, we evaluated the oxidation of P700 in the relationship with LEF reflected in Y(II) (Fig. 2).
 126 In mustard at 21 kPa O₂, the oxidation level of P700 increased monotonously with the decrease in Y(II)
 127 (Fig. 2A) although the PQ pool was suggested to be reduced at low Ci (Supplemental Fig. S1B). That is,
 128 photosynthetic electron transport is limited between PSII and PSI, presumably at the Cyt *b₆/f* complex.
 129 However, at 1 kPa O₂, P700 started to be kept reduced when Y(II) < 0.3 (Fig. 2A), which suggests that
 130 the electron sink by photorespiration is required for P700 oxidation¹⁴. Unlike mustard, P700 continued
 131 to remain oxidised in maize even at 1 kPa O₂ except under an extreme CO₂ limitation (< 1.5 Pa)(Fig.
 132 2B). The inverse proportional relationship in the redox states of the PQ pool and P700 suggested that
 133 LEF was limited by the suppression of electron transport at the Cyt *b₆/f* complex in maize as in C₃ plants
 134 (Fig. 2 and Supplemental Fig. S1).

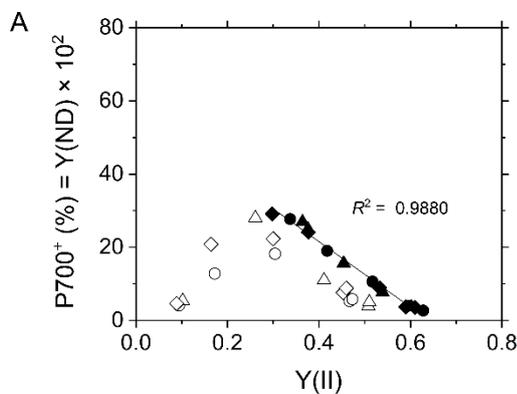
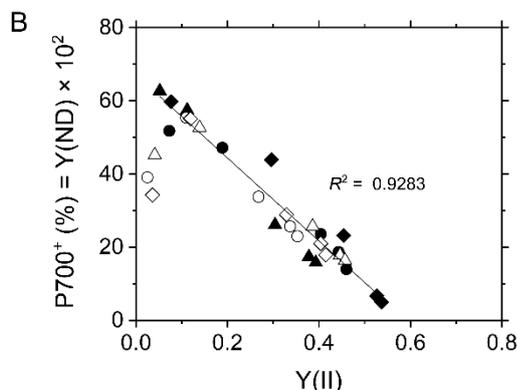


Fig. 2. Relationship of P700 oxidation with effective quantum yield of PSII, Y(II), at various intercellular CO₂ partial pressures in the C₃ plant mustard (A) and the C₄ plant maize (B). Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent estimated linear regression of the data at 21 kPa O₂ (R^2 , coefficient of determination).



135 To investigate molecular mechanisms for P700 oxidation in response to the limitation of
 136 electron sink, we evaluated the thylakoid membrane potential by ECS analysis in the transition from

137 light to dark at the steady-state photosynthesis (Fig. 3). We note that the ECS parameters are
 138 dependent on the properties of the leaves, not only the density of chloroplasts, but also the content
 139 of light-harvesting complexes that house the shifted pigments. Therefore, it is difficult to make any
 140 quantitative conclusions for the differences in the amplitudes of ECS parameters between mustard
 141 and maize. In the so-called dark-interval relaxation kinetics, *proton motive force (pmf)* in the light is
 142 defined as the total rapid (< 1 s) change in the ECS signal upon rapidly switching off the light, which
 143 increased with the limitation of CO₂ in both mustard and maize at 21 kPa O₂ (Fig. 4A and D). However,
 144 in mustard *pmf* was not enhanced in response to the suppression of CO₂ assimilation at 1 kPa O₂,

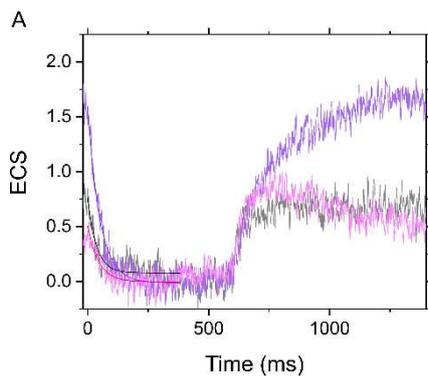
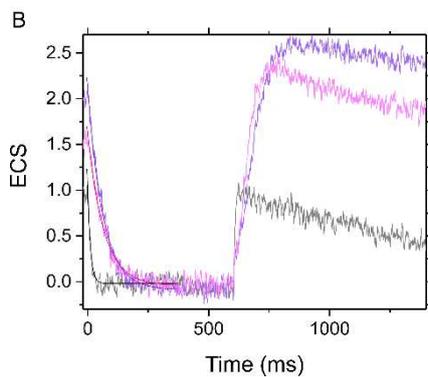


Fig. 3. Dark-interval relaxation kinetics of electrochromic shift (ECS) in the C₃ plant mustard (A) and the C₄ plant maize (B) under ambient air (40 Pa CO₂, 21 kPa O₂; black), low CO₂ (1 Pa CO₂, 21 kPa O₂; purple), and low CO₂/O₂ (1 Pa CO₂, 1 kPa O₂; pink). Red actinic light (550 μmol photons m⁻² s⁻¹) was turned off at 0 ms for 600 ms during the steady-state photosynthesis. The kinetics were fit to mono exponential decay (*R*², coefficient of determination: 0.7017, 0.9454, and 0.6868 in A; 0.8814, 9868, and 0.9801 in B respectively).



145 where photorespiration is inhibited (Fig. 4A). That is, photorespiration functions as the electron sink
 146 to sustain LEF, supporting the increase of *pmf* in C₃ plants¹⁴. Nevertheless, we found the increase of
 147 *pmf* independent of photorespiration in maize (Fig. 4D). Proton conductance of the ATP synthase (*g_H⁺*)
 148 is calculated as the rate constant of the mono-exponential ECS decay²³, which increased with Ci and
 149 were then saturated as in the trend of CO₂ assimilation rate regardless of O₂ partial pressures in both
 150 mustard and maize (Fig. 4B and E). Further, the initial decay rate of the ECS changes is termed as
 151 relative light-driven proton flux through the chloroplast ATP synthase, the so-called *v_H⁺*²⁴, which is
 152 correlated with the sum of net CO₂ assimilation rate and Rd at 1 kPa O₂ in mustard (Fig. 4C). In mustard
 153 at 21 kPa O₂, a part of *v_H⁺* was uncoupled from the CO₂ assimilation rate (Fig. 4C), like the case of Y(II)
 154 (Fig. 1C), which suggested that *v_H⁺* has almost the linear relationship with LEF derived from CO₂
 155 assimilation and photorespiration in C₃ intact leaves. In the case of maize leaves, the relationship of
 156 *v_H⁺* with photosynthetic CO₂ assimilation does not seem to be linear, different from that of Y(II), at

157 both 21 and 2 kPa O₂ (Fig. 4F). We note that v_{H^+} decreased with photosynthetic CO₂ assimilation larger
 158 in lower Ci conditions (Fig. 4F), implying that ATP is utilized in a different manner at different Ci levels.

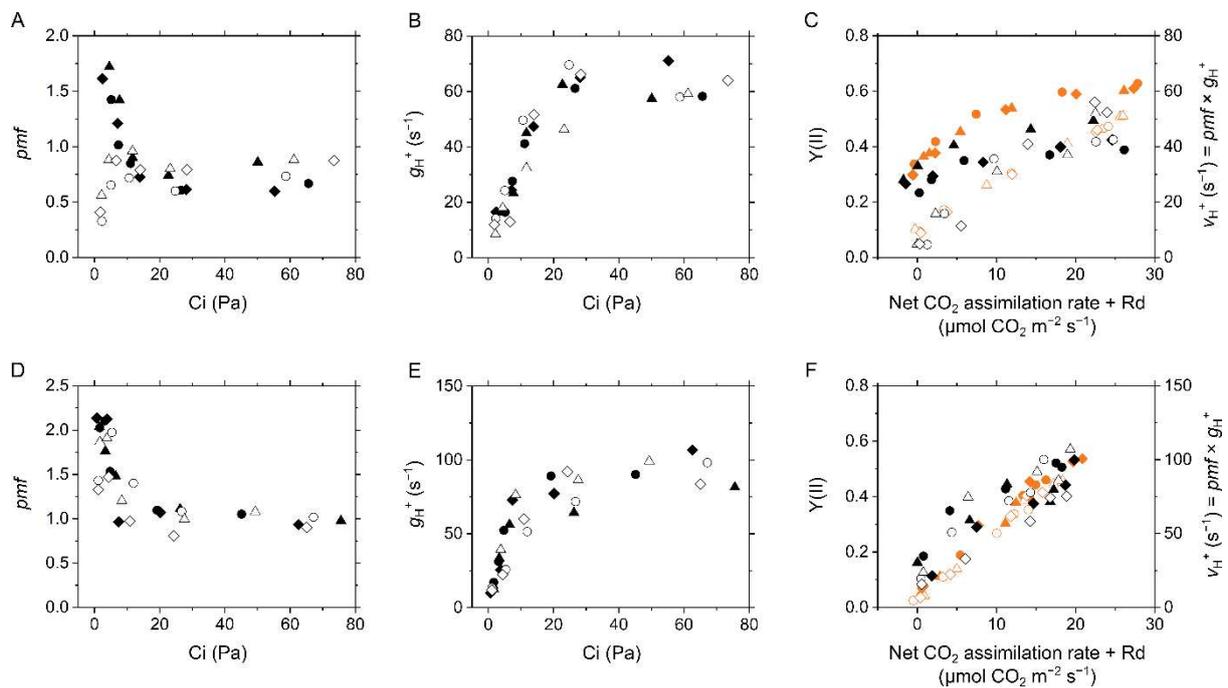


Fig. 4. Electrochromic shift (ECS) parameters in the C₃ plant mustard (A–C) and the C₄ plant maize (D–F). (A, D) Proton motive force (*pmf*) at various intercellular CO₂ partial pressures (Ci). (B, E) Proton conductance of the chloroplast ATP synthase (g_{H^+}) at various Ci. (C, F) Relationship of proton efflux rate *via* the ATP synthase (v_{H^+}) with CO₂ assimilation rate. Effective quantum yield of PSII, Y(II), are also shown in orange symbols. Dark respiration rate is presented as Rd. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols).

159 Theoretically, *pmf* includes two components: ΔpH and transmembrane difference in the
 160 electric field ($\Delta\Psi$), which can be distinguished by the post-illumination transient change in ECS, and
 161 the former component triggers the suppression of electron transport in the Cyt *b*₆/*f* complex²⁵. Like
 162 P700 oxidation, ΔpH showed the linear relationship with LEF at 21 kPa O₂ in both mustard and maize.
 163 The formation of ΔpH was disturbed at 1 kPa O₂ in mustard and also in maize at very low Ci (Fig. 5A
 164 and C). The increase of ΔpH was associated with NPQ (Supplemental Fig. S1B), which was in agreement
 165 with that NPQ at PSII is stimulated by ΔpH ²⁶. There were two possibilities assumed for the lumen
 166 acidification: (1) H⁺-pumping from stroma into the thylakoid lumen was promoted; or (2) H⁺-leakage
 167 from the lumen to stroma was blocked. Both mustard and maize showed the linear relationship of g_{H^+}
 168 with LEF reflected in Y(II) (Fig. 5B and D), which suggested that limiting g_{H^+} leads to the increase in ΔpH ,
 169 resulting in P700 oxidation.

170 In this study, we investigated the redox state of Fd at the steady state of photosynthesis in
 171 both mustard and maize using a Klas-NIR spectrophotometer. The maximum amplitude of photo-
 172 reducible Fd was determined by the standard method in advance as shown in Supplemental Fig. S2.
 173 The decay of Fd⁻ in the transition from light to dark was mono-exponentially fit, giving the amplitude
 174 and the oxidation rate of Fd⁻ (Fig. 6). Exceptionally, a biphasic decay manner was recognised in the
 175 kinetics at 1 Pa CO₂ and 1 kPa O₂ in maize (Supplemental Fig. S3). The slow component remains to be

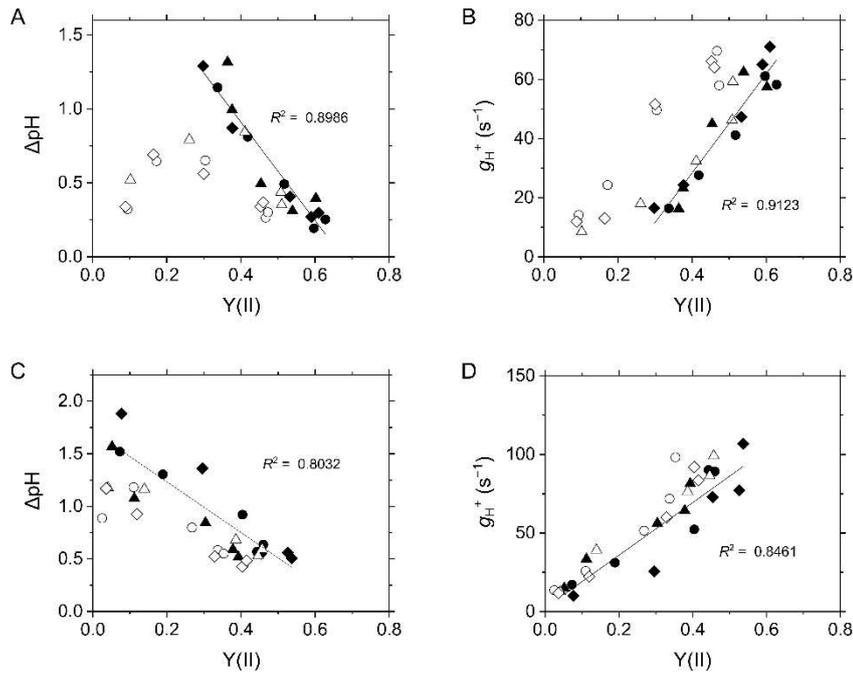


Fig. 5. Relationships of the proton gradient across the thylakoid membrane (ΔpH ; A, C) and the proton conductance of the chloroplast ATP synthase (g_{H^+} ; B, D) with effective quantum yield of PSII, $Y(\text{II})$, at various intercellular CO_2 partial pressures in the C_3 plant mustard (A, B) and the C_4 plant maize (C, D). We note that ΔpH and g_{H^+} were separately measured from $Y(\text{II})$ at the same ambient CO_2 partial pressures. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O_2 (open symbols). Solid lines represent estimated linear regression of the data at 21 kPa O_2 (R^2 , coefficient of determination).

176 identified, but it was clearly negligible compared to LEF because the estimated half time was more
 177 than minutes. In mustard, Fd was strongly reduced under CO_2 limitation at 1 kPa O_2 , whereas it was
 178 totally kept oxidised at 21 kPa O_2 (Fig. 6A and 7A), indicating that photorespiration functions as the
 179 electron sink to relieve the acceptor-side limitation of PSI. In maize, Fd^- was gradually accumulated
 180 with the decrease in C_i regardless of O_2 (Fig. 6B and 7B). The intact leaves of mustard showed a linear
 181 relationship of Fd^- oxidation rate with $Y(\text{II})$ and its y -intercept was close to zero (Fig. 7C), which is in
 182 agreement with the recent study on C_3 plants²⁷. Further, the same trend was observed also in maize
 183 leaves (Fig. 7D). That is, LEF estimated from $Y(\text{II})$ clearly corresponds to the electron transport *via* Fd
 184 in the situations where ΔpH increased and P700 was oxidised in response to the limitation of electron
 185 sink in both mustard and maize.

186 We also plotted effective quantum yield of PSI, $Y(\text{I})$, against $Y(\text{II})$, which is a conventional
 187 method to evaluate CEF activity used in numerous previous and recent reports. In mustard at 21 kPa
 188 O_2 , we observed a linear relationship between $Y(\text{I})$ and $Y(\text{II})$ with extra $Y(\text{I})$ to $Y(\text{II})$ at lower C_i (Fig. 8A).
 189 Additionally, $Y(\text{I})$ showed no linear relationship with $Y(\text{II})$ in mustard at 1 kPa O_2 and maize at both 21
 190 and 1 kPa O_2 , resulting in the extra $Y(\text{I})$ (Fig. 8A and C), different from the relationship of Fd^- oxidation
 191 rate with $Y(\text{II})$. Here, we also evaluated the oxidation of PC and plotted it against $Y(\text{II})$. In mustard, PC
 192 was kept more oxidized at lower C_i , but it was reduced where photorespiration was inhibited (Fig. 8B),
 193 like P700 (Fig. 2A). In maize PC was reduced with the decrease in $Y(\text{II})$ less than 0.2, although it was

194 more oxidized at lower C_i if $Y(II)$ was >0.2 , finally giving a curved plot (Fig. 8D). Interestingly, the extra
 195 $Y(I)$ to $Y(II)$ was coincided with the oxidation of PC (Fig. 8).

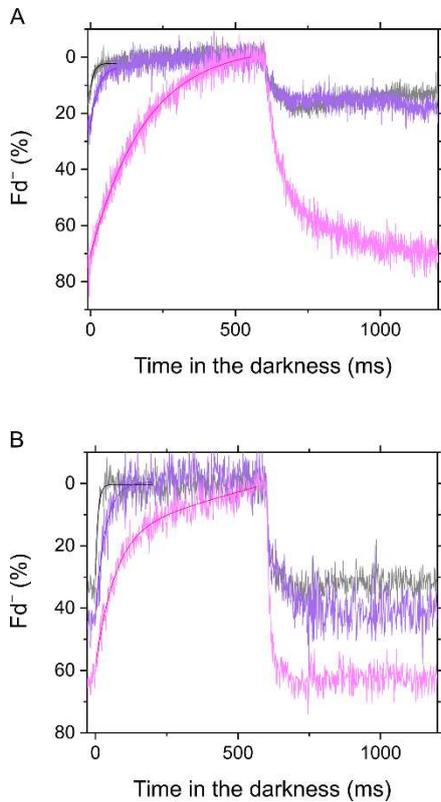


Fig. 6. Dark-interval relaxation kinetics of ferredoxin (Fd⁻) in the C₃ plant mustard (A) and the C₄ plant maize (B) under ambient air (40 Pa CO₂, 21 kPa O₂; black), low CO₂ (1 Pa CO₂, 21 kPa O₂; purple), and low CO₂/O₂ (1 Pa CO₂, 1 kPa O₂; pink). Red actinic light (550 μmol photons m⁻² s⁻¹) was turned off at 0 ms for 600 ms during the steady-state photosynthesis. The kinetics were fit to mono exponential decay (R^2 , coefficient of determination: 0.6628, 0.8816, and 0.9788 in A; 0.6612 and 0.8307 in B). Only Fd⁻ kinetics in maize under low CO₂/O₂ was fit to biphasic exponential decay (R^2 : 0.9405).

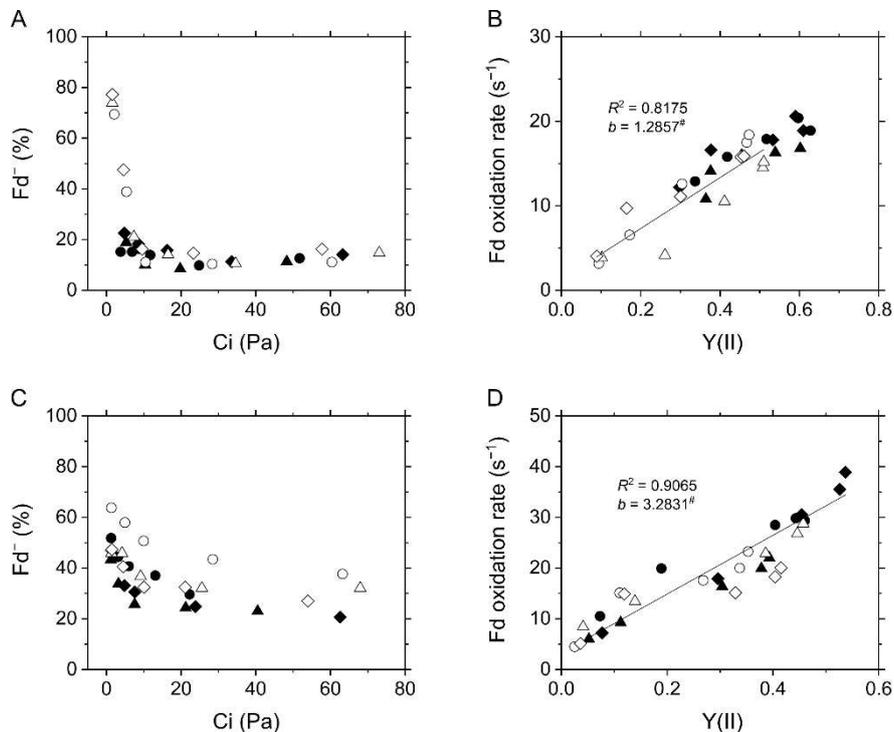


Fig. 7. *In vivo* measurement for the redox state of ferredoxin (Fd) in the C₃ plant mustard (A, B) and the C₄ plant maize (C, D). (A, C) The Fd reduction during the steady-state photosynthesis at various intercellular CO₂ partial pressures (C_i). (B, D) Relationship of Fd⁻ oxidation rate with effective quantum yield of PSII, $Y(II)$, at various C_i . Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent the estimated linear regressions of the data at 1 kPa (B) and 21 kPa O₂ (D), respectively (R^2 , coefficient of determination). The y -intercepts (b) were tested based on the null hypothesis: $^{\#}p > 0.05$.

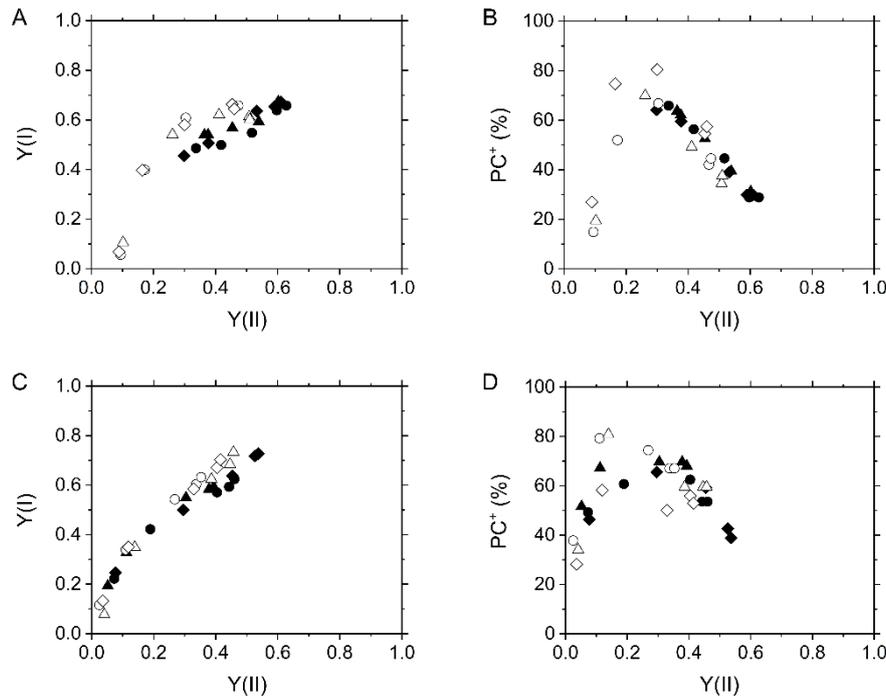


Fig. 8. Relationships of effective quantum yield of PSI, Y(I) (A, C), and plastocyanin (PC) oxidation (B, D) with effective quantum yield of PSII, Y(II), at various intercellular CO₂ partial pressures in the C₃ plant mustard (A, B) and the C₄ plant maize (C, D). Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols).

196 The linear relationships among CO₂ assimilation rate, Y(II), P700 oxidation, ΔpH, and Fd⁻
 197 oxidation rate (Fig. 1, 2, 5, and 7) suggested that the regulation of photosynthetic electron transport
 198 is tightly associated with LEF in C₄ plants, similarly to C₃ plants, which is somehow unexpected if we
 199 assume that maize has a large activity of CEF. Originally, CEF in C₄ plants is hypothesized based on the
 200 biochemical fact that there is less grana structure containing PSII in the isolated bundle sheath cells.
 201 It should be noted that in this study we analysed a variety of photosynthetic parameters at the scale
 202 of intact leaves that is the mixture of mesophyll and bundle sheath cells. Here, we spectroscopically
 203 estimated the ratio of PSI to PSII *in vivo* in the intact leaves of mustard and maize. A short-saturation
 204 flash induces ECS dependent on the photochemical reaction at PSII and PSI. During far-red light
 205 illumination, where PSI is selectively excited, the short-saturation flash induces ECS presumably
 206 originated only from PSII²⁸. Actually, the ECS amplitude induced by the short-saturation flash
 207 decreased with the intensity of far-red light, finally reaching approximately 50% of the initial
 208 amplitude in both mustard and maize (Fig. 9), indicating that the ratio of PSII to PSI is about 1:1 at the
 209 scale of leaves. The same results were obtained from the field-grown sunflower (C₃ plant) and maize
 210 (Supplemental Fig. S4).

211 To further test if there is the significant amount of PSI uncoupled from PSII in bundle sheath
 212 cells of maize, the connectivity of PSI with PSII was roughly estimated *in vivo* using a Klas-NIR
 213 spectrophotometer. A short-saturation flash was applied to excite PSII and PSI after accumulating
 214 P700⁺ by the far-red light illumination, resulting in the decay kinetics of P700⁺ (Fig. 10). The P700⁺

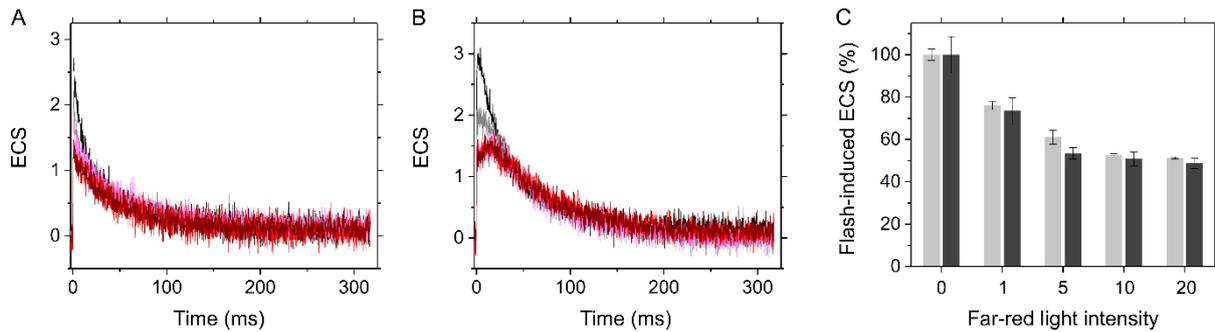


Fig. 9. Electrochromic shift (ECS) induced by a 5 μ s-short saturation flash during far-red light illumination in the C_3 plant mustard (A) and the C_4 plant maize (B). Far-red light was provided at various intensities (0, black; 1, grey; 5, pink; 10, red; and the maximum 20, wine red; the values defined by the Walz software). (C) The flash-induced ECS changes normalized by the values without far-red light illumination as 100%. The data of mustard (light grey) and maize (dark grey) are shown as the mean with the standard deviation ($n = 3$, biological replicates).

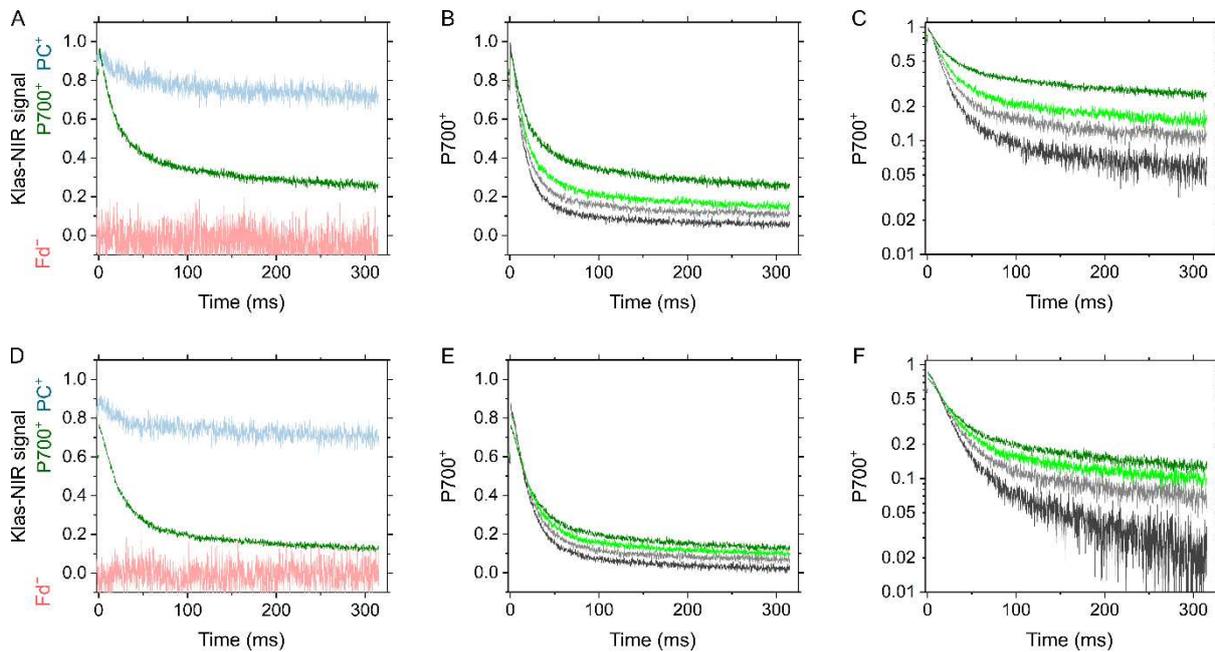


Fig. 10. Effects of a short-saturation flash on the redox state around PSI in the C_3 plant mustard (A–C) and the C_4 plant maize (D–F). (A, D) Reduction kinetics of $P700^+$ (green) in response to a 5 μ s-short saturation flash after far-red light illumination for 10 s. Far-red light was provided at the maximum intensity (20, the value defined by the Walz software). Deconvoluted signals to plastocyanin (PC^+ , blue) and ferredoxin (Fd^- , red) are also shown. All the Klas-NIR signals were normalized within the range from the minimum 0 to the maximum 1. The maximum reduction/oxidation levels of each component were determined as shown in Supplemental Fig. S2. The signal to Fd^- has negative values. (B, E) Reduction kinetics of $P700^+$ by a short saturation flash at different lengths (5, green; 10, light green; 20, light grey; and 50 μ s, grey respectively) after far-red light illumination for 10 s. The kinetics are also shown with the logarithmic scale at y-axis (C, F). The representative traces of independent experiments ($n = 3$, biological replicates) are shown.

215 reduction kinetics was likely to involve in more than two components, but the rapid decay within 100
 216 ms should reflect the electron transport from PSII²⁹. Interestingly, a part of $P700^+$ (10–20%) took more
 217 than second to be reduced (data not shown), which may be due to a redox equilibration between PC
 218 and $P700$ as implied from the reduction of a part of PC (ca. 20%) within 100 ms (Fig. 10A and D). The
 219 redox state of Fd did not change in response to the short-saturation flash (Fig. 10A and D). The ratio
 220 of $P700^+$ rapidly reduced was rather larger in maize (ca. 80%) than in mustard (ca. 65%), both of which
 221 increased with the length of the short-saturation flash (5–50 μ s) to approximately 90% (Fig. 10B, C, E,
 222 and F). These results implied that there is no significant amount of PSI having “less connectivity” with
 223 PSII uniquely to maize but not to mustard at the scale of the intact leaves. It should be noted that $P700$

224 was kept more reduced in maize than in mustard during far-red light illumination (Fig. 10A and D, and
225 Supplemental Fig. S2), and a part of P700⁺ (ca. 20%) was immediately reduced within 50 μ s just after
226 the illumination with 5 μ s short-saturation flash (Fig. 10D), which was likely to be too fast for CEF,
227 considering the turnover of the Cyt *b*₆/*f* complex³⁰.

228 Discussion

229

230 We experimentally characterized the *in vivo* regulatory mechanisms of photosynthetic electron
231 transport for P700 oxidation in maize, which showed the different O₂ dependency but could be
232 understood based on the same model for C₃ plants although photosynthesis in C₄ plants is driven in
233 co-operation of mesophyll and bundle sheath cells, and metabolically different from that in C₃ plants.
234 Oxidation of P700 is the universal strategy for photosynthetic organisms to suppress the generation
235 of ROS at the acceptor side of PSI, which is mainly regulated in C₃ plants by the donor-side mechanism,
236 i.e. the ΔpH-dependent suppression of electron transport at the Cyt *b₆/f* complex, and by the acceptor-
237 side one, photorespiration. Since maize did not show the enough capacity of photorespiration as an
238 alternative electron sink, LEF was linearly suppressed with the decrease in photosynthetic CO₂
239 assimilation (Fig. 1). Nevertheless, P700 is kept oxidised with the suppression of CO₂ assimilation,
240 which is linearly associated with the increase in ΔpH by limiting g_{H^+} (Fig. 5). In these processes, the Fd⁻
241 oxidation rate showed the linear relationship with Y(II) (Fig. 7). The metabolic compartmentation in C₄
242 photosynthesis and the higher ratio of PSI to PSII in isolated bundle sheath cells have made it
243 complicated to consider the regulation of photosynthetic electron transport in C₄ plants. However,
244 the ratio of PSI to PSII was estimated almost 1:1 at the scale of the intact maize leaves, similarly to C₃
245 plants (Fig. 9). Further, the ratio of P700⁺ rapidly reduced by a short-saturation flash was similar
246 between mustard and maize (Fig. 10). All these results supported the robustness of P700 oxidation
247 tightly associated with LEF in the C₄ plant maize, as previously proposed in C₃ plants, with the
248 exception of little photorespiratory electron sink on the acceptor side of PSI¹³.

249 Instead of the photorespiratory electron sink, maize strongly relies P700 oxidation on the ΔpH-
250 dependent suppression of electron transport at the Cyt *b₆/f* complex (Fig. 2B and 5C). The decrease in
251 g_{H^+} was linearly correlated with LEF reflected in Y(II) (Fig. 5D) and Fd⁻ oxidation rate (Fig. 7D). These
252 results suggest that lumen acidification for P700 oxidation was attributed to the decrease of g_{H^+} but
253 not to an additional H⁺-pumping from stroma to the thylakoid lumen, for example, by CEF. The
254 different dependencies of P700 oxidation on the regulation of g_{H^+} led to the different threshold of LEF
255 for keeping P700 oxidised between mustard and maize. Contrary to mustard, which needed about
256 50% of the maximum Y(II) for P700 oxidation, maize did not exhibit breakdown of P700 oxidation even
257 when Y(II) was close to zero (Fig. 2). The linear proportional relationship between Y(II) and P700
258 oxidation indicated that the C₄ plant maize does not require an electron sink for P700 oxidation (Fig.
259 2B). The insensitivity to O₂ of P700 oxidation has been observed also at various irradiances at CO₂-
260 saturated conditions³¹. Nevertheless, an extreme condition of 1 kPa O₂ and very low CO₂ (< 1.5 Pa)
261 partially disturbed P700 oxidation in maize (Fig. 2B), which may be due to the slight but certain

262 electron flux *via* photorespiration or the Mehler reaction in C₄ plants^{18,32}. In the other word, C₄ plants
 263 ultimately utilize O₂ for P700 oxidation maybe in some extremely stressed conditions. In C₃ plants,
 264 photorespiration relieves the electron transport limitation on the acceptor side of PSI and also sustains
 265 to produce ΔpH (Fig. 2A, 5A and C, and Supplemental Fig. S1A and B). It should be trade-off to rely
 266 P700 oxidation mainly on the regulations on the donor or acceptor sides of PSI. The contribution of
 267 the regulation of g_{H^+} to P700 oxidation would be different among different types of C₄ plants and C₃-
 268 C₄ intermediates associated with the capacity of photorespiration⁵. Whereas P700 oxidation with the
 269 suppression of photosynthesis is the phenomenon commonly observed in a variety of photosynthetic
 270 organisms, the dominant molecular mechanisms are rich in diversity, which has been already
 271 diversified among different species of cyanobacteria, the progenitor of oxygenic photosynthesis¹⁰,
 272 and has changed during the evolutionary history of photosynthetic green and red plastid lineages.
 273 Interestingly, the strategy for P700 oxidation in maize can be categorized on the view of O₂-usage into
 274 the same type of that in some secondary algae derived from red algae, which do not need O₂ for P700
 275 oxidation³³.

276 What is the determinant for the strong contribution of the ΔpH-dependent donor side
 277 mechanism to P700 oxidation in maize? Lumen acidification should be controlled by both H⁺-pumping
 278 and H⁺-consumption rates in plant leaves. Based on the fact that the CEF activity *via* Fd was negligible
 279 (Fig. 7B and D), the H⁺-pumping rate into the thylakoid lumen is estimated from LEF with the rate
 280 constant (k_{H^+}), which is correlated with Y(II). Meanwhile, the H⁺-consumption rate is equal to the ECS
 281 parameter v_{H^+} , reflecting ATP consumption by CO₂ assimilation and photorespiration in C₃ plants.
 282 Finally, the change in *pmf* is presented as the following equation in C₃ plants¹³:

283

$$284 \quad d(\textit{pmf})/dt = (k_{H^+} \times \textit{LEF}) - (\textit{pmf} \times g_{H^+})$$

$$285 \quad \text{At the steady state,} \quad \textit{pmf} = (k_{H^+} \times \textit{LEF})/g_{H^+}$$

286

287 That is, lumen acidification occurs where LEF is sustained more than g_{H^+} . In mustard, the decrease in
 288 Y(II) (ca. 50%) was much smaller than that in g_{H^+} (ca. 85%) at 21 but not at 1 kPa O₂ (Fig. 5B), indicating
 289 that photorespiration contributes to sustaining LEF, resulting in the H⁺ accumulation into the thylakoid
 290 lumen (Fig. 5A and 11)^{14, 34}. In maize LEF decreased almost concomitantly with g_{H^+} , different from
 291 mustard (Fig. 5D). Nevertheless, H⁺ was accumulated in the thylakoid lumen to cause P700 oxidation
 292 at both 21 and 1 kPa O₂ (Fig. 2B and 5C). Overall, different from C₃ plants, *pmf* in maize cannot be
 293 formulated as the above equation (Fig. 11). In the other words, there may be an additional mechanism
 294 to pump H⁺ into the thylakoid lumen independent of the electron transport via Fd, which remains to
 295 be further investigated in future works. One possibility that cannot be excluded is the effect of the

296 Mehler reaction¹⁸ on ΔpH at the steady state of photosynthesis even though the activity is very low.
 297 It should be also noted that the relationship of v_{H^+} with photosynthetic CO_2 assimilation was different
 298 from that of $Y(II)$ (Fig. 4F), which is presumably due to the different ATP consumption for regenerating
 299 PEP at various C_i . The strong contribution of the ΔpH -dependent donor side mechanism to P700
 300 oxidation in maize should be also related to the different ATP utilization between C_3 and C_4 plants.

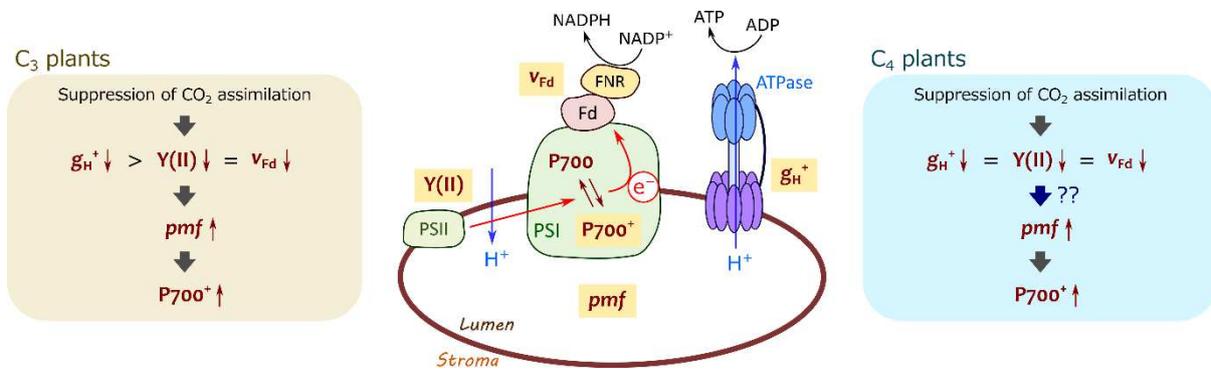


Fig. 11. A brief illustration of mechanism for P700 oxidation in C_3 and C_4 plants. In C_3 plants, proton conductance of chloroplast ATP synthase (g_{H^+}) decreases with the suppression of photosynthetic CO_2 assimilation greater than photosynthetic linear electron flow reflected in effective quantum yield of PSII, $Y(II)$, and ferredoxin (Fd^-) oxidation rate (v_{Fd}), resulting in the increase in *proton motive force* (*pmf*) to induce P700 oxidation. In C_4 plants, *pmf* increases with the suppression of photosynthetic CO_2 assimilation, although g_{H^+} decreases concomitantly with $Y(II)$ and v_{Fd} .

301 The present results are unlikely to follow the hypothesis that CEF is driven in the bundle sheath
 302 cells of NADP-ME subtype of C_4 plants at a comparable flux to LEF in the mesophyll cells. Although
 303 there is the report that the PSII activities in the bundle sheath cells of NADP-ME subtype of C_4 plants,
 304 including maize, sorghum, and Flaveria, are almost equal to those in the mesophyll cells³⁵, numerous
 305 studies have followed and documented that PSII activity is nearly negligible in isolated bundle sheath
 306 cells in NADP-ME subtype of C_4 plants, especially in maize^{36, 37, 38, 39, 40, 41, 42, 43}, having constructed the
 307 dogma that CEF is responsible for the additional ATP production in the bundle sheath cells of C_4 plants.
 308 Actually, recent modelling studies select the CEF model for simulating C_4 photosynthesis⁴⁴. However,
 309 it should be concerned that these studies follow the ratio of PSI to PSII in the experimentally
 310 differentiated mesophyll and bundle sheath cells, except for the semi-quantitative evaluation by
 311 immunocytology⁴⁵. Overall, the PSI:PSII ratio has yet not been quantitatively understood at the scale
 312 of intact leaves, and there is still no evidence for the energetic contribution of CEF to photosynthesis
 313 in C_4 plants. Another problem is that *in vivo* CEF activity has been ordinarily evaluated using the
 314 effective quantum yield of PSI that is easily under/overestimated (Fig. 8)². Considering the
 315 experimental results by the previous and present studies, we propose that the amount of PSI in bundle
 316 sheath cells is presumably much smaller than that in mesophyll cells at the scale of C_4 intact leaves.

317 It is still unclear how the additional ATP demand is met in C_4 photosynthesis. Chapman et al.
 318 (1980), one of important landmarks in C_4 photosynthesis research, has shown that in the presence of
 319 exogenously added malate photosynthetic CO_2 assimilation proceeds in the isolated bundle sheath
 320 cells of maize with PSII inhibited by 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU). The PSII-

321 independent ATP production can reach about 80% of the ATP production of the illuminated cells in
322 the *in vitro* system³⁸, which clearly suggests that there is a malate-dependent ATP source in the cells.
323 It should be also considered that the NAD(P)H dehydrogenase complex plays an important role for C₄
324 photosynthesis⁴⁶. It has been recently accepted that the intermediates of the Calvin-Benson cycle
325 shuttle between the mesophyll and bundle sheath cells with certainly large fluxes^{43, 47}, which can
326 make it more complicated how much ATP and NADPH are needed respectively in these two types of
327 cells in the light. In both mustard and maize, *pmf* was formed to some extent even in the conditions
328 where photosynthesis and photorespiration are almost completely suppressed under 1 kPa O₂ and
329 very low Ci (Fig. 4A and D), which suggested that ATP is, at least, not limited. We also note that the
330 stoichiometry of ATP and NADPH may not need to be necessarily satisfied because excess NADPH
331 should not be accumulated as long as P700 oxidation system works².

332 For the last decades, CEF has been frequently evaluated by the comparison between Y(II) and
333 Y(I). The extra Y(I) to Y(II) has been believed to be a conventional indicator to CEF activity. Nevertheless,
334 recently it has been clearly shown that Y(I) can be easily over-estimated by the oxidation of PC². In
335 this study, the extra Y(I) was totally coincided with the PC oxidation in both mustard and maize at
336 various CO₂ and O₂ partial pressures (Fig. 8). These facts indicate that Y(I) must not be utilized to
337 evaluate CEF activity. A variety of alternative methods, including the comparison of Fd⁻ oxidation rate
338 with Y(II), should be considered in future works.

339 **Materials and Methods**

340

341 **Plant materials**

342 Field mustard (komatsuna, *Brassica rapa*) and maize (*Zea mays*) were grown under long-day
343 conditions (14 h-light, 24 °C, 300 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$, white fluorescent lamp/10 h-dark, 22 °C). Seeds
344 were planted in pots that contained a 5:3:2 mix of Metro-Mix 350 (Sun Gro Horticulture, Agawam, MA,
345 USA), Akadama, and vermiculit with 1000-fold diluted Hyponex solution (Hyponex, Osaka, Japan) used
346 as a watering solution. For the experiments in Supplemental Fig. S4, the field-grown sunflower
347 (*Helianthus annuus*) and maize were used.

348

349 **Gas exchange, chlorophyll fluorescence, and spectroscopic analyses**

350 Exchanges of CO_2 and H_2O were measured using a GFS-3000 equipped with a 3010-DUAL gas exchange
351 chamber (Walz, Effeltrich, Germany) in which ambient air was saturated with water vapor at $18.0 \pm$
352 0.1 °C, and the leaf temperature was maintained at 25 ± 2 °C. C_i was calculated based on the previous
353 report⁴⁸.

354 Chlorophyll fluorescence and near infrared absorbance were simultaneously measured
355 coupled with gas exchange analysis using a Klas-NIR spectrophotometer (Walz)⁴⁹. Chlorophyll
356 fluorescence parameters were calculated as follows⁵⁰: F_o , minimum fluorescence from a dark-adapted
357 leaf; F_m' , maximum fluorescence from a light-adapted leaf; F' , fluorescence emission from a light-
358 adapted leaf; $Y(\text{II}) = (F_m' - F')/F_m'$, effective quantum yield of PSII; $q_L = (F_m' - F')/(F_m' - F_o') \times (F_o'/F')$,
359 fraction of "open" PSII centres (with Q_A oxidised) on the basis of a lake model for the PSII
360 photosynthetic apparatus; $\text{NPQ} = (F_m - F_m')/F_m'$, non-photochemical quenching. Pulse-amplitude
361 modulated green measuring light (540 nm , $<0.1 \mu\text{mol photons m}^{-2} \text{s}^{-1}$) was used. To obtain F_m' , a
362 saturation flash (630 nm , $8,000 \mu\text{mol photons m}^{-2} \text{s}^{-1}$, 300 ms) was applied. Red actinic light (630 nm ,
363 $550 \mu\text{mol photons m}^{-2} \text{s}^{-1}$) was supplied using a chip-on-board LED array. The signals for P700^+ , PC^+ ,
364 and Fd^- were calculated based on the deconvolution of four pulse-modulated dual-wavelength
365 difference signals in the near infrared region ($780\text{--}820$, $820\text{--}870$, $840\text{--}965$, and $870\text{--}965 \text{ nm}$). The
366 redox state of P700 was evaluated as the ratio of P700^+ to the total P700, termed $Y(\text{ND})$ ⁵¹. Both P700
367 and PC are kept completely reduced and Fd is fully oxidised in dark conditions. For the determination
368 of total photo-oxidizable P700 and PC, the saturation flash was applied after 10 s illumination with a
369 far-red light (740 nm ; Supplemental Fig. S2)⁵². Total photo-reducible Fd was determined by the
370 illumination with a red actinic light ($450 \mu\text{mol photons m}^{-2} \text{s}^{-1}$) after plant leaves were adapted to the
371 dark for 5 min (Supplemental Fig. S2)⁵². For the analysis of a dark-interval relaxation kinetics, the red
372 actinic light ($550 \mu\text{mol photons m}^{-2} \text{s}^{-1}$) was temporarily turned off for 600 ms at the steady-state

373 photosynthesis²⁷. The oxidation rate of Fd^- was estimated as relative values by a Klas-NIR
374 spectrophotometer by a linear fitting for the initial decay of Fd^- . Recently, it has been shown that the
375 reduced iron-sulphur clusters in PSI, F_A/F_B , can contribute to the Klas-NIR signal attributed to Fd^- in
376 almost the same manner *in vitro*⁵³. In this study, we concluded that the F_A/F_B contribution to the Fd^-
377 signal was not critical problem in the measurement for the Fd^- oxidation rate because Fd^- was kept
378 oxidised more than 50% at the steady state of photosynthesis except for the condition where the
379 electron acceptor side of PSI was extremely limited at low CO_2 and O_2 in mustard.

380 ECS was measured simultaneously with gas exchange using a Klas-100 spectrophotometer
381 (Walz)⁵⁴. The ECS signal was calculated from two pulse-modulated dual-wavelength difference signals
382 using the following equation: $(\Delta I/I_{521.4-507.6} + \Delta I/I_{520.2-534.8})/2$. Red actinic light was temporarily turned
383 off for 600 ms in a dark-interval relaxation kinetics analysis during the steady-state photosynthesis to
384 determine ECS parameters⁵⁵. The total rapid (<1 s) change in ECS signal upon rapidly switching off
385 actinic light was defined as *pmf*. The parameter v_H^+ was estimated by a linear fitting for the initial
386 phase of the mono-exponential decay of *pmf* in the transition from light to dark, giving the rate
387 constant of the decay g_H^+ by the calculation. The ΔpH component of *pmf* was estimated also by turning
388 red actinic light off for 30 s at the steady state of photosynthesis²⁵, which are used in Fig. 5A and C.
389 The amplitude of ECS was normalised by the ECS change induced by a 5 μs -short saturation flash²⁸.

390 All of *in vivo* spectroscopic measurements were based on the assumption that the absorption
391 coefficient and the amplification factor⁵⁶ of each targeted molecule are not different between
392 mesophyll and bundle sheath cells. All statistical analyses were performed using Origin 2017
393 (Lightstone, Tokyo, Japan).

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397

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399

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402 English corrections.

403 **Figure legends**

404

405 **Fig. 1.** Photosynthetic CO₂ assimilation and linear electron flow in the C₃ plant mustard (A–C) and the
406 C₄ plant maize (D–F). (A, D) Net CO₂ assimilation rate at various intercellular CO₂ partial pressures (C_i).
407 (B, E) Effective quantum yield of PSII, Y(II), at various C_i. (C, F) Relationship of Y(II) with CO₂ assimilation
408 rate. Dark respiration rate is presented as R_d. Experiments were conducted independently three times
409 as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open
410 symbols). Solid lines represent estimated linear regression of the data at 1 kPa (C) and 21 kPa O₂ (F)
411 (R^2 , coefficient of determination). The y -intercepts (b) were tested based on the null hypothesis:
412 $**p < 0.005$.

413

414 **Fig. 2.** Relationship of P700 oxidation with effective quantum yield of PSII, Y(II), at various intercellular
415 CO₂ partial pressures in the C₃ plant mustard (A) and the C₄ plant maize (B). Experiments were
416 conducted independently three times as shown in different symbols (biological replicates) at 21 kPa
417 (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent estimated linear regression of the
418 data at 21 kPa O₂ (R^2 , coefficient of determination).

419

420 **Fig. 3.** Dark-interval relaxation kinetics of electrochromic shift (ECS) in the C₃ plant mustard (A) and
421 the C₄ plant maize (B) under ambient air (40 Pa CO₂, 21 kPa O₂; black), low CO₂ (1 Pa CO₂, 21 kPa O₂;
422 purple), and low CO₂/O₂ (1 Pa CO₂, 1 kPa O₂; pink). Red actinic light (550 μmol photons m⁻² s⁻¹) was
423 turned off at 0 ms for 600 ms during the steady-state photosynthesis. The kinetics were fit to mono
424 exponential decay (R^2 , coefficient of determination: 0.7017, 0.9454, and 0.6868 in A; 0.8814, 9868,
425 and 0.9801 in B respectively).

426

427 **Fig. 4.** Electrochromic shift (ECS) parameters in the C₃ plant mustard (A–C) and the C₄ plant maize
428 (D–F). (A, D) Proton motive force (pmf) at various intercellular CO₂ partial pressures (C_i). (B, E) Proton
429 conductance of the chloroplast ATP synthase (g_{H^+}) at various C_i. (C, F) Relationship of proton efflux
430 rate *via* the ATP synthase (v_{H^+}) with CO₂ assimilation rate. Effective quantum yield of PSII, Y(II), are also
431 shown in orange symbols. Dark respiration rate is presented as R_d. Experiments were conducted
432 independently three times as shown in different symbols (biological replicates) at 21 kPa (closed
433 symbols) and 1 kPa O₂ (open symbols).

434

435 **Fig. 5.** Relationships of the proton gradient across the thylakoid membrane (ΔpH ; A, C) and the proton
436 conductance of the chloroplast ATP synthase (g_{H^+} ; B, D) with effective quantum yield of PSII, Y(II), at

437 various intercellular CO₂ partial pressures in the C₃ plant mustard (A, B) and the C₄ plant maize (C, D).
438 We note that ΔpH and g_{H⁺} were separately measured from Y(II) at the same ambient CO₂ partial
439 pressures. Experiments were conducted independently three times as shown in different symbols
440 (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent
441 estimated linear regression of the data at 21 kPa O₂ (R^2 , coefficient of determination).

442

443 **Fig. 6.** Dark-interval relaxation kinetics of ferredoxin (Fd⁻) in the C₃ plant mustard (A) and the C₄ plant
444 maize (B) under ambient air (40 Pa CO₂, 21 kPa O₂; black), low CO₂ (1 Pa CO₂, 21 kPa O₂; purple), and
445 low CO₂/O₂ (1 Pa CO₂, 1 kPa O₂; pink). Red actinic light (550 μmol photons m⁻² s⁻¹) was turned off at 0
446 ms for 600 ms during the steady-state photosynthesis. The kinetics were fit to mono exponential decay
447 (R^2 , coefficient of determination: 0.6628, 0.8816, and 0.9788 in A; 0.6612 and 0.8307 in B). Only Fd⁻
448 kinetics in maize under low CO₂/O₂ was fit to biphasic exponential decay (R^2 : 0.9405).

449

450 **Fig. 7.** *In vivo* measurement for the redox state of ferredoxin (Fd) in the C₃ plant mustard (A, B) and
451 the C₄ plant maize (C, D). (A, C) The Fd reduction during the steady-state photosynthesis at various
452 intercellular CO₂ partial pressures (C_i). (B, D) Relationship of Fd⁻ oxidation rate with effective quantum
453 yield of PSII, Y(II), at various C_i. Experiments were conducted independently three times as shown in
454 different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid
455 lines represent the estimated linear regressions of the data at 1 kPa (B) and 21 kPa O₂ (D), respectively
456 (R^2 , coefficient of determination). The y-intercepts (*b*) were tested based on the null hypothesis:
457 #*p*>0.05.

458

459 **Fig. 8.** Relationships of effective quantum yield of PSI, Y(I) (A, C), and plastocyanin (PC) oxidation (B,
460 D) with effective quantum yield of PSII, Y(II), at various intercellular CO₂ partial pressures in the C₃
461 plant mustard (A, B) and the C₄ plant maize (C, D). Experiments were conducted independently three
462 times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂
463 (open symbols).

464

465 **Fig. 9.** Electrochromic shift (ECS) induced by a 5 μs-short saturation flash during far-red light
466 illumination in the C₃ plant mustard (A) and the C₄ plant maize (B). Far-red light was provided at various
467 intensities (0, black; 1, grey; 5, pink; 10, red; and the maximum 20, wine red; the values defined by the
468 Walz software). (C) The flash-induced ECS changes normalized by the values without far-red light
469 illumination as 100%. The data of mustard (light grey) and maize (dark grey) are shown as the mean
470 with the standard deviation (*n* = 3, biological replicates).

471

472 **Fig. 10.** Effects of a short-saturation flash on the redox state around PSI in the C₃ plant mustard (A–C)
473 and the C₄ plant maize (D–F). (A, D) Reduction kinetics of P700⁺ (green) in response to a 5 μs-short
474 saturation flash after far-red light illumination for 10 s. Far-red light was provided at the maximum
475 intensity (20, the value defined by the Walz software). Deconvoluted signals to plastocyanin (PC⁺, blue)
476 and ferredoxin (Fd⁻, red) are also shown. All the Klas-NIR signals were normalized within the range
477 from the minimum 0 to the maximum 1. The maximum reduction/oxidation levels of each component
478 were determined as shown in Supplemental Fig. S2. The signal to Fd⁻ has negative values. (B, E)
479 Reduction kinetics of P700⁺ by a short saturation flash at different lengths (5, green; 10, light green;
480 20, light grey; and 50 μs, grey respectively) after far-red light illumination for 10 s. The kinetics are also
481 shown with the logarithmic scale at y-axis (C, F). The representative traces of independent
482 experiments (*n* = 3, biological replicates) are shown.

483

484 **Fig. 11.** A brief illustration of mechanism for P700 oxidation in C₃ and C₄ plants. In C₃ plants, proton
485 conductance of chloroplast ATP synthase (g_{H^+}) decreases with the suppression of photosynthetic CO₂
486 assimilation greater than photosynthetic linear electron flow reflected in effective quantum yield of
487 PSII, Y(II), and ferredoxin (Fd⁻) oxidation rate (v_{Fd}), resulting in the increase in *proton motive force* (*pmf*)
488 to induce P700 oxidation. In C₄ plants, *pmf* increases with the suppression of photosynthetic CO₂
489 assimilation, although g_{H^+} decreases concomitantly with Y(II) and v_{Fd} .

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Figures

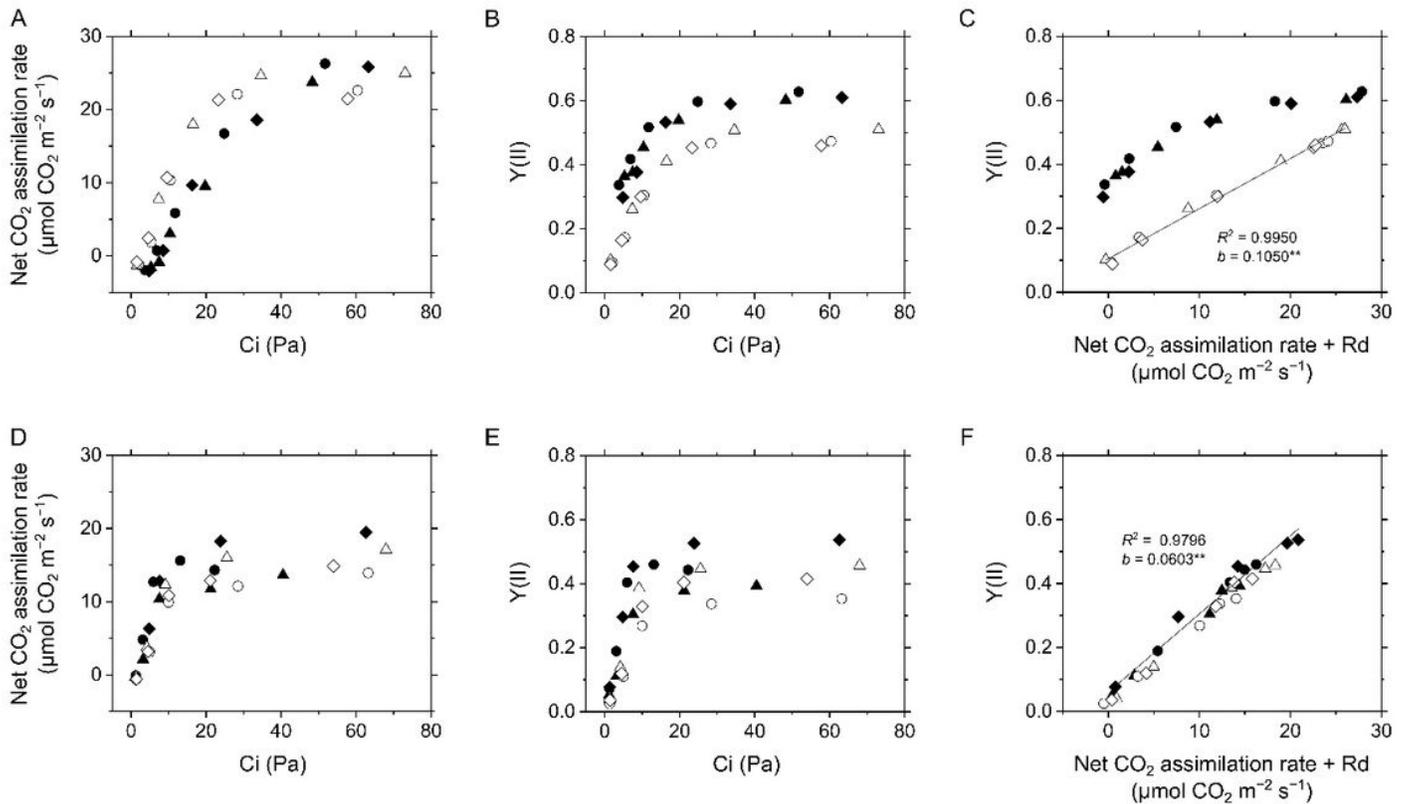


Figure 1

Photosynthetic CO₂ assimilation and linear electron flow in the C3 plant mustard (A–C) and the C4 plant maize (D–F). (A, D) Net CO₂ assimilation rate at various intercellular CO₂ partial pressures (Ci). (B, E) Effective quantum yield of PSII, Y(II), at various Ci. (C, F) Relationship of Y(II) with CO₂ assimilation rate. Dark respiration rate is presented as Rd. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent estimated linear regression of the data at 1 kPa (C) and 21 kPa O₂ (F) (R², coefficient of determination). The y-intercepts (b) were tested based on the null hypothesis: **p<0.005.

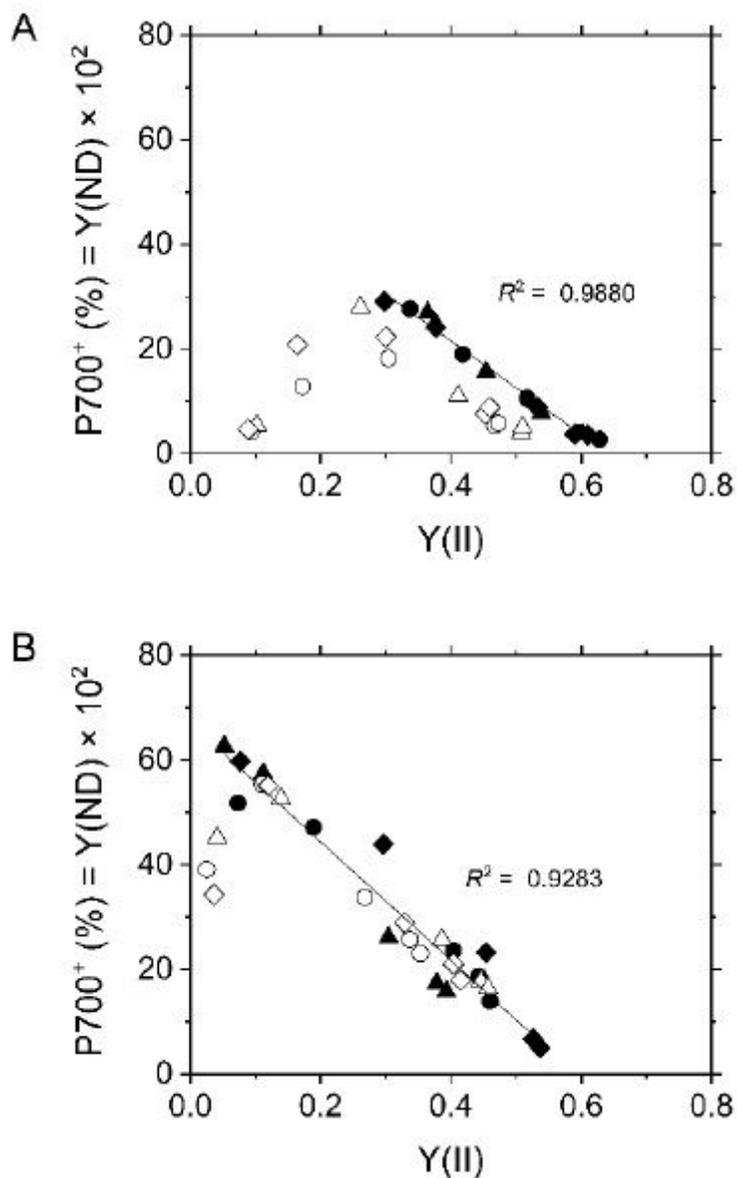


Figure 2

Relationship of P700 oxidation with effective quantum yield of PSII, Y(II), at various intercellular CO₂ partial pressures in the C₃ plant mustard (A) and the C₄ plant maize (B). Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent estimated linear regression of the data at 21 kPa O₂ (R^2 , coefficient of determination).

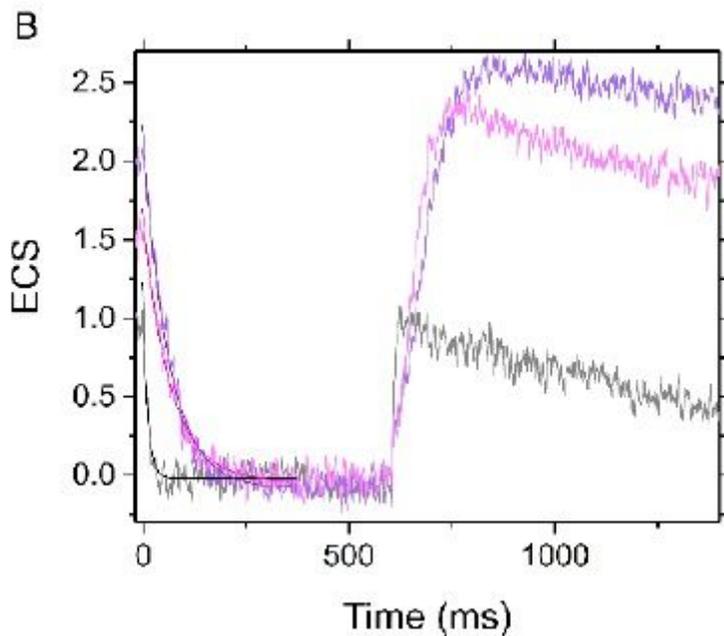
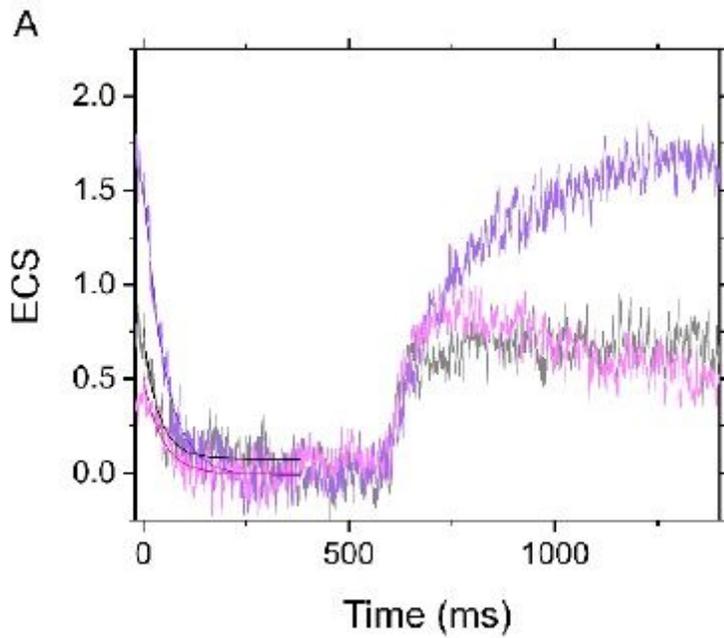


Figure 3

Dark-interval relaxation kinetics of electrochromic shift (ECS) in the C3 plant mustard (A) and the C4 plant maize (B) under ambient air (40 Pa CO₂, 21 kPa O₂; black), low CO₂ (1 Pa CO₂, 21 kPa O₂; purple), and low CO₂/O₂ (1 Pa CO₂, 1 kPa O₂; pink). Red actinic light (550 μ mol photons m⁻² s⁻¹) was turned off at 0 ms for 600 ms during the steady-state photosynthesis. The kinetics were fit to mono exponential decay (R², coefficient of determination: 0.7017, 0.9454, and 0.6868 in A; 0.8814, 9868, and 0.9801 in B respectively).

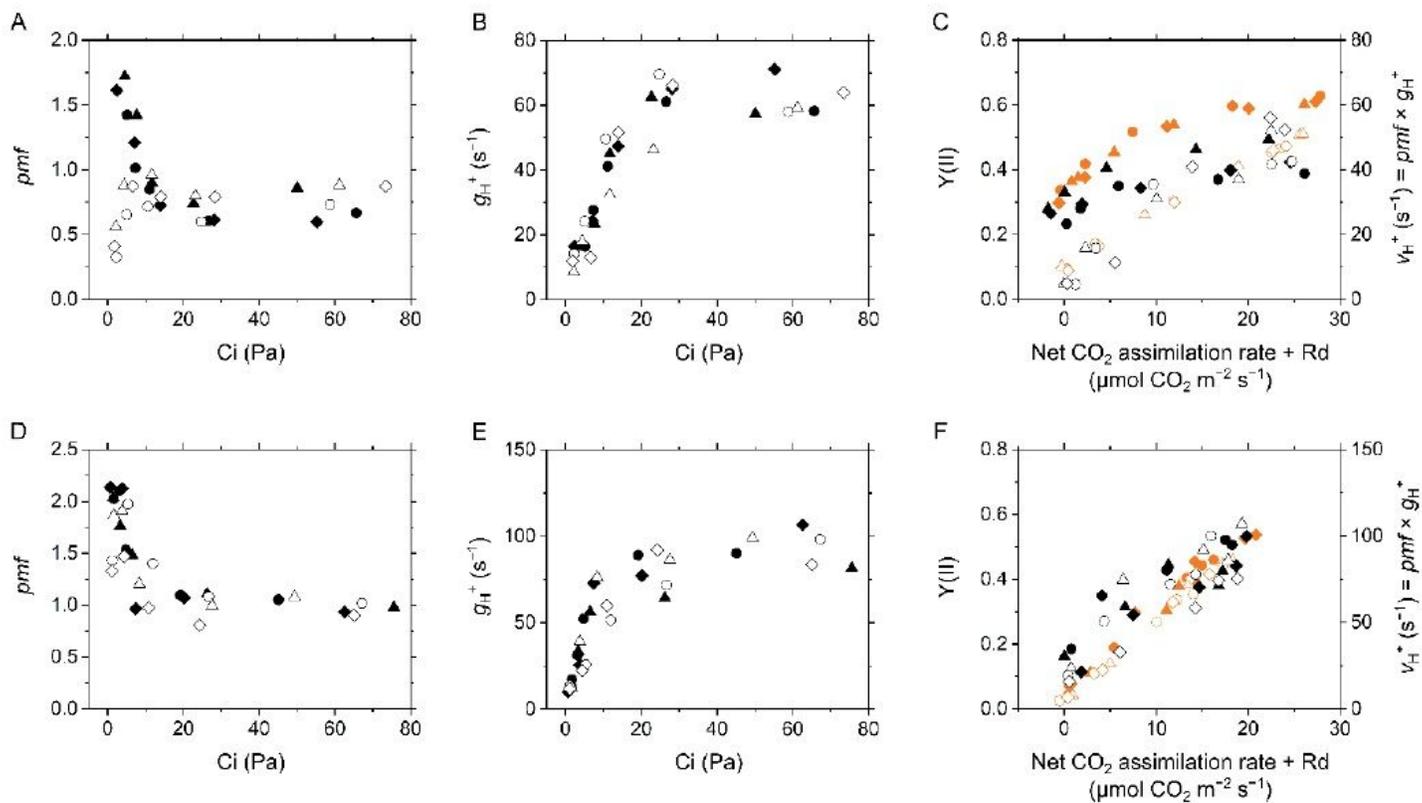


Figure 4

Electrochromic shift (ECS) parameters in the C3 plant mustard (A–C) and the C4 plant maize (D–F). (A, D) Proton motive force (pmf) at various intercellular CO₂ partial pressures (Ci). (B, E) Proton conductance of the chloroplast ATP synthase (g_{H^+}) at various Ci. (C, F) Relationship of proton efflux rate via the ATP synthase (v_{H^+}) with CO₂ assimilation rate. Effective quantum yield of PSII, Y(II), are also shown in orange symbols. Dark respiration rate is presented as Rd. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols).

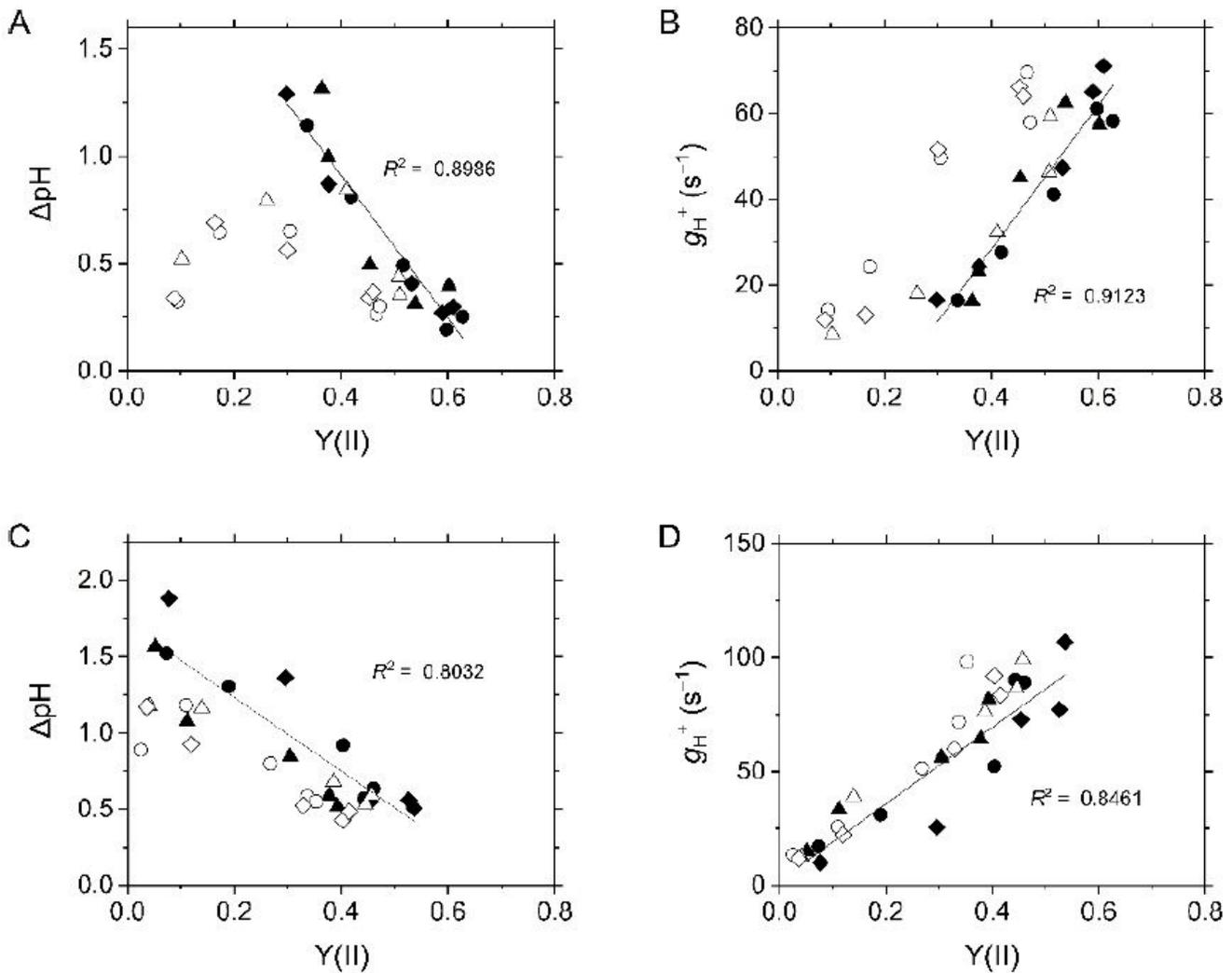


Figure 5

Relationships of the proton gradient across the thylakoid membrane (ΔpH ; A, C) and the proton conductance of the chloroplast ATP synthase (g_{H^+} ; B, D) with effective quantum yield of PSII, $Y(\text{II})$, at various intercellular CO $_2$ partial pressures in the C $_3$ plant mustard (A, B) and the C $_4$ plant maize (C, D). We note that ΔpH and g_{H^+} were separately measured from $Y(\text{II})$ at the same ambient CO $_2$ partial pressures. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O $_2$ (open symbols). Solid lines represent estimated linear regression of the data at 21 kPa O $_2$ (R^2 , coefficient of determination).

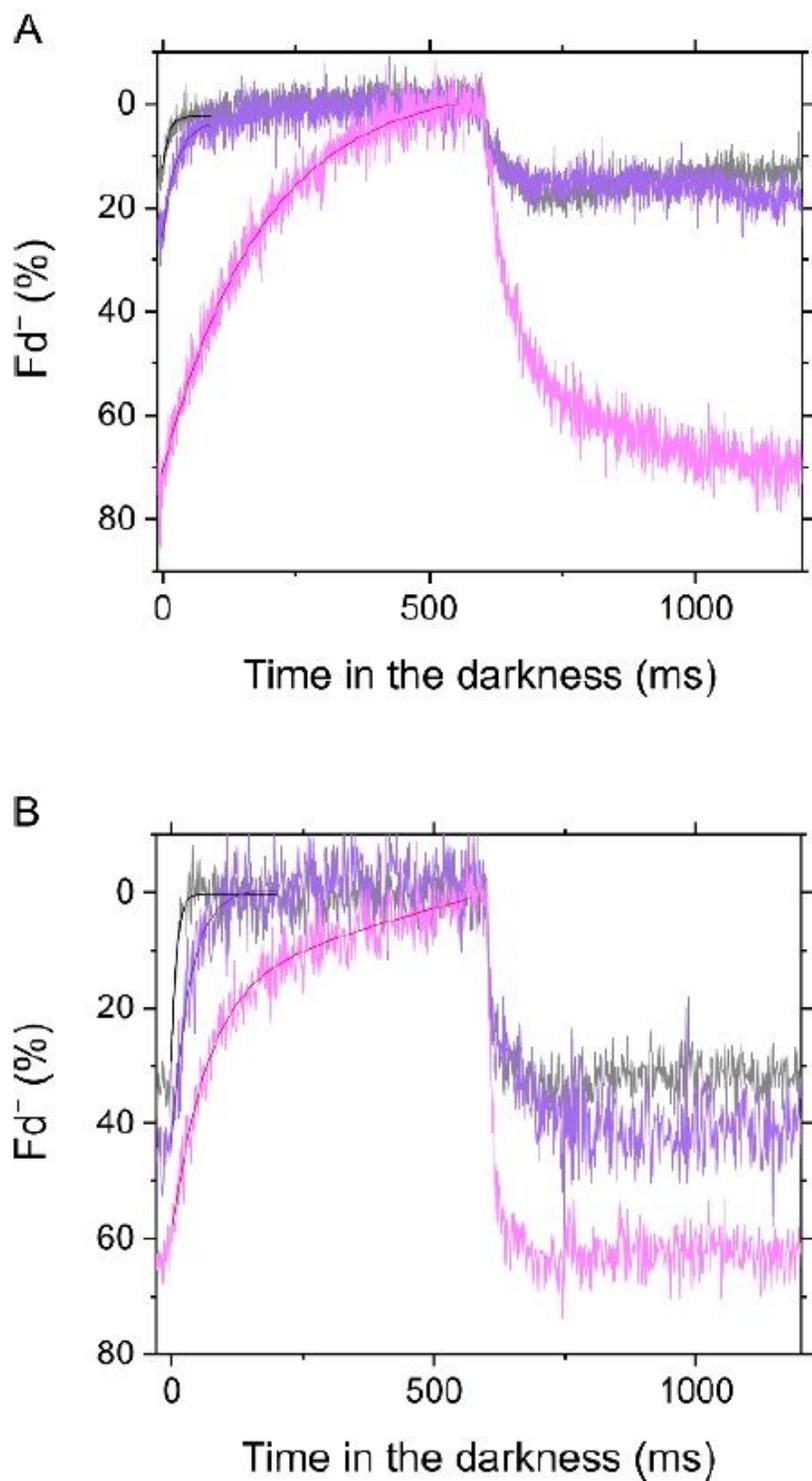


Figure 6

Dark-interval relaxation kinetics of ferredoxin (Fd^-) in the C3 plant mustard (A) and the C4 plant maize (B) under ambient air (40 Pa CO_2 , 21 kPa O_2 ; black), low CO_2 (1 Pa CO_2 , 21 kPa O_2 ; purple), and low CO_2/O_2 (1 Pa CO_2 , 1 kPa O_2 ; pink). Red actinic light ($550 \mu\text{mol photons m}^{-2} \text{s}^{-1}$) was turned off at 0 ms for 600 ms during the steady-state photosynthesis. The kinetics were fit to mono exponential decay (R2,

coefficient of determination: 0.6628, 0.8816, and 0.9788 in A; 0.6612 and 0.8307 in B). Only Fd- kinetics in maize under low CO₂/O₂ was fit to biphasic exponential decay (R²: 0.9405).

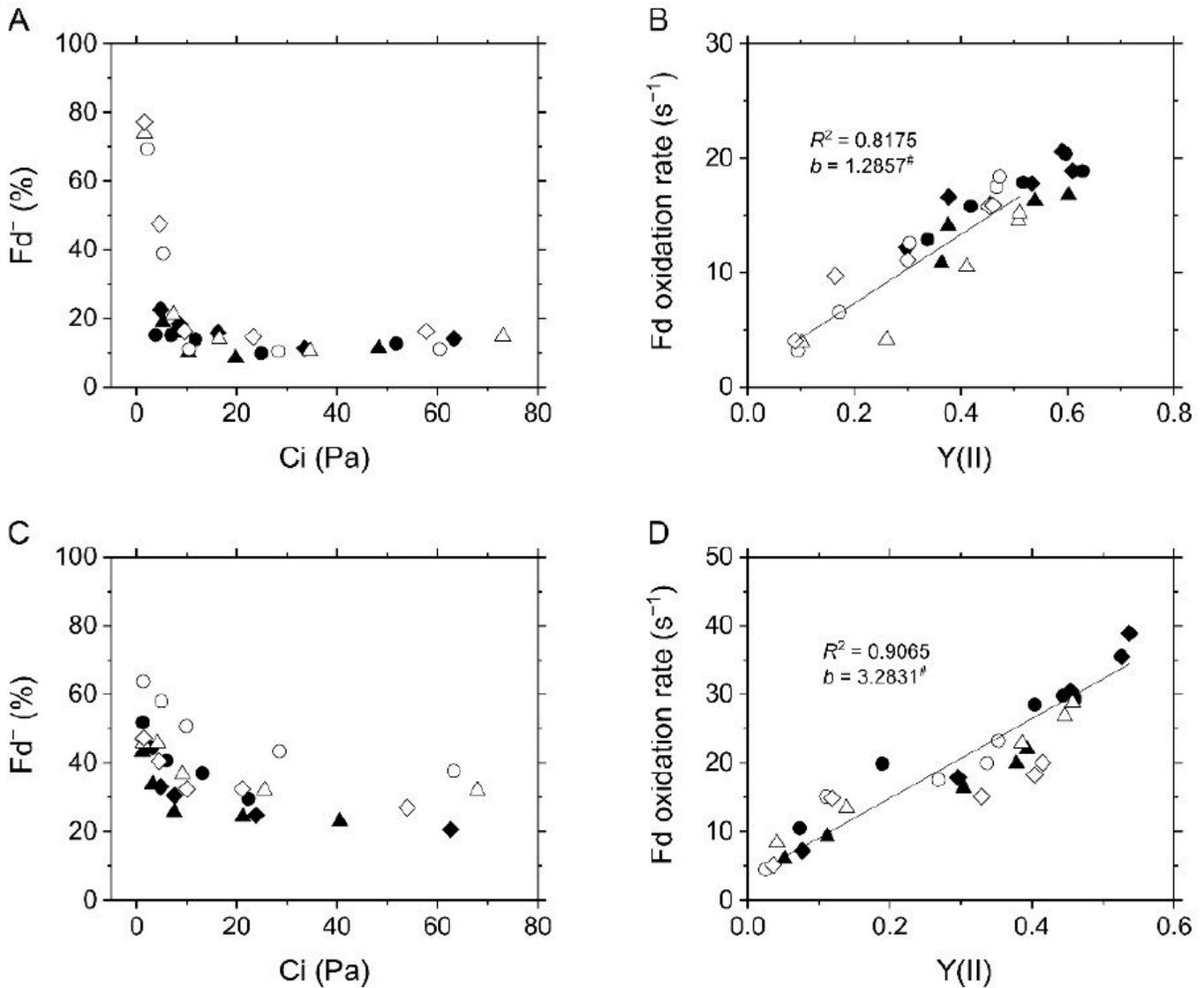


Figure 7

In vivo measurement for the redox state of ferredoxin (Fd) in the C₃ plant mustard (A, B) and the C₄ plant maize (C, D). (A, C) The Fd reduction during the steady-state photosynthesis at various intercellular CO₂ partial pressures (Ci). (B, D) Relationship of Fd- oxidation rate with effective quantum yield of PSII, Y(II), at various Ci. Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O₂ (open symbols). Solid lines represent the estimated linear regressions of the data at 1 kPa (B) and 21 kPa O₂ (D), respectively (R², coefficient of determination). The y-intercepts (b) were tested based on the null hypothesis: #p>0.05.

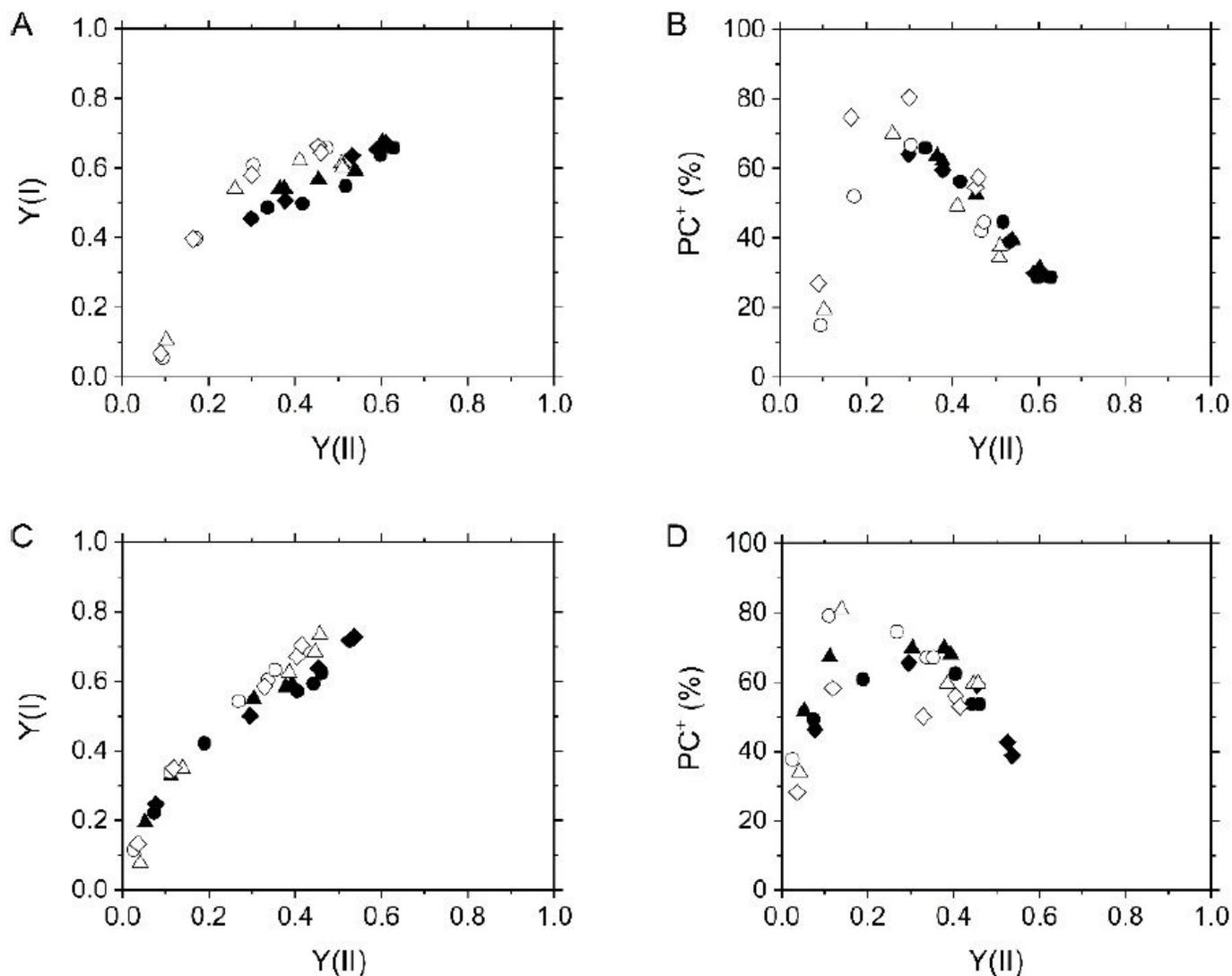


Figure 8

Relationships of effective quantum yield of PSI, $Y(I)$ (A, C), and plastocyanin (PC) oxidation (B, D) with effective quantum yield of PSII, $Y(II)$, at various intercellular CO_2 partial pressures in the C3 plant mustard (A, B) and the C4 plant maize (C, D). Experiments were conducted independently three times as shown in different symbols (biological replicates) at 21 kPa (closed symbols) and 1 kPa O_2 (open symbols).

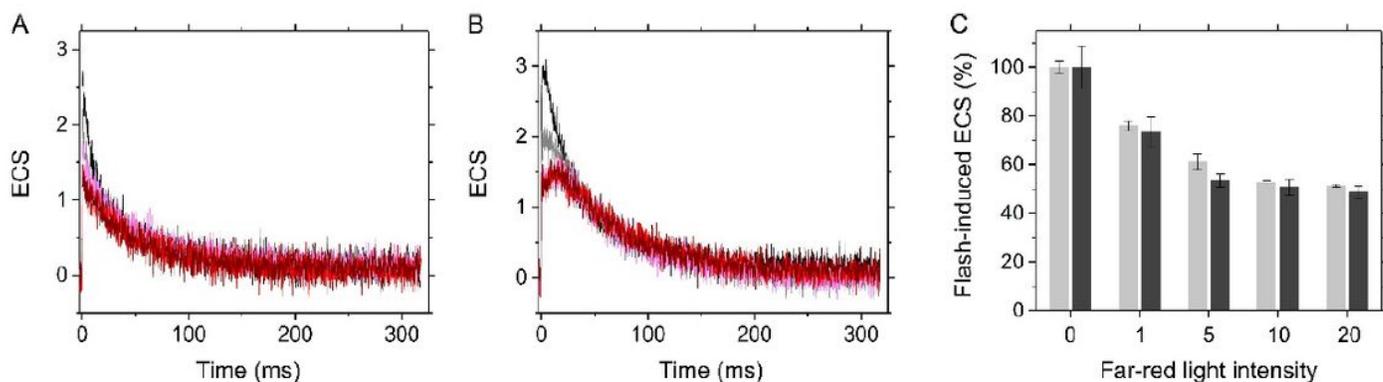


Figure 9

Electrochromic shift (ECS) induced by a 5 μ s-short saturation flash during far-red light illumination in the C3 plant mustard (A) and the C4 plant maize (B). Far-red light was provided at various intensities (0, black; 1, grey; 5, pink; 10, red; and the maximum 20, wine red; the values defined by the Walz software). (C) The flash-induced ECS changes normalized by the values without far-red light illumination as 100%. The data of mustard (light grey) and maize (dark grey) are shown as the mean with the standard deviation ($n = 3$, biological replicates).

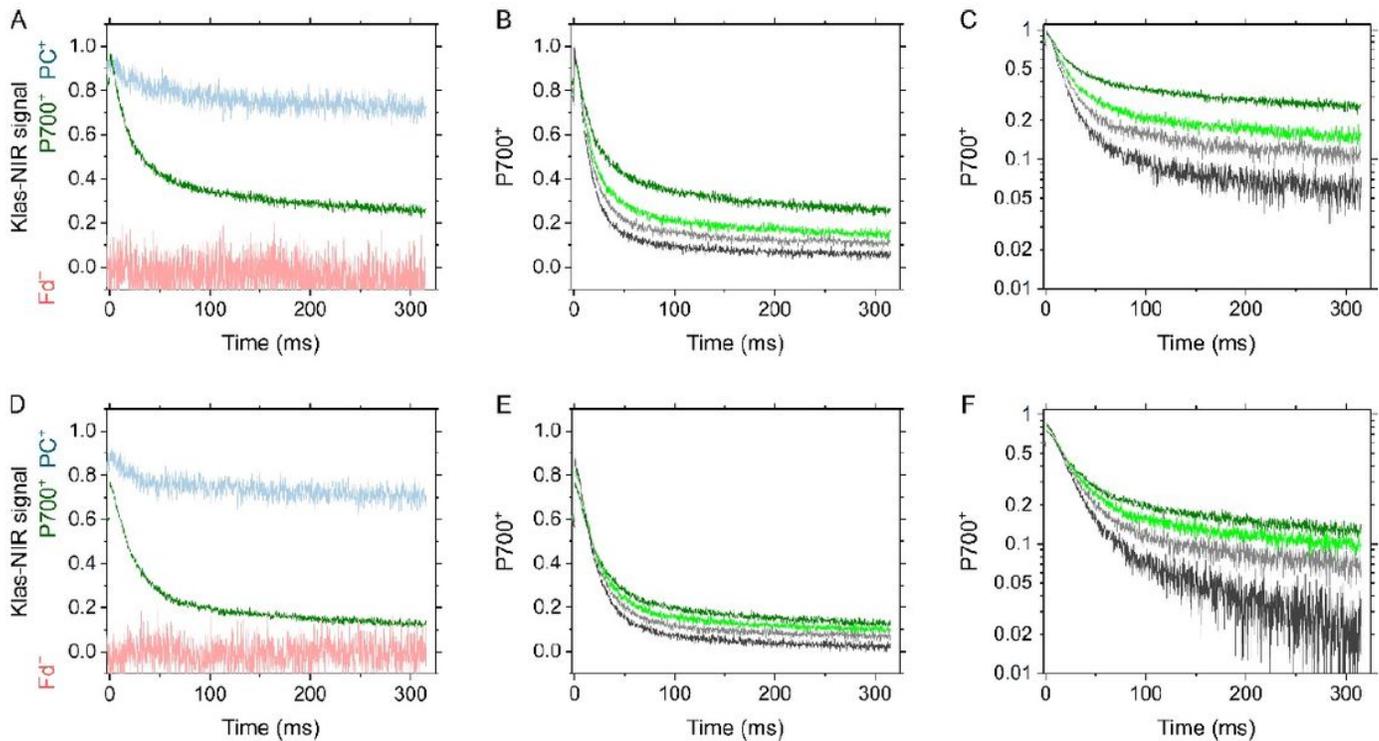


Figure 10

Effects of a short-saturation flash on the redox state around PSI in the C3 plant mustard (A-C) and the C4 plant maize (D-F). (A, D) Reduction kinetics of P700+ (green) in response to a 5 μ s-short saturation flash after far-red light illumination for 10 s. Far-red light was provided at the maximum intensity (20, the value defined by the Walz software). Deconvoluted signals to plastocyanin (PC+, blue) and ferredoxin (Fd-, red) are also shown. All the Klas-NIR signals were normalized within the range from the minimum 0 to the maximum 1. The maximum reduction/oxidation levels of each component were determined as shown in Supplemental Fig. S2. The signal to Fd- has negative values. (B, E) Reduction kinetics of P700+ by a short saturation flash at different lengths (5, green; 10, light green; 20, light grey; and 50 μ s, grey respectively) after far-red light illumination for 10 s. The kinetics are also shown with the logarithmic scale at y-axis (C, F). The representative traces of independent experiments ($n = 3$, biological replicates) are shown.

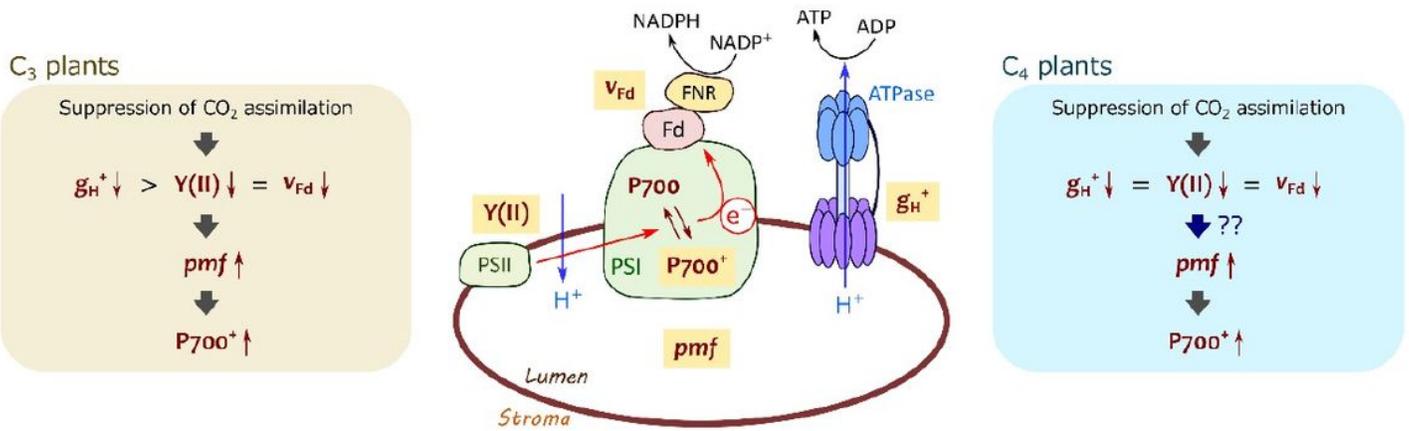


Figure 11

A brief illustration of mechanism for P700 oxidation in C₃ and C₄ plants. In C₃ plants, proton conductance of chloroplast ATP synthase (g_{H^+}) decreases with the suppression of photosynthetic CO₂ assimilation greater than photosynthetic linear electron flow reflected in effective quantum yield of PSII, Y(II), and ferredoxin (Fd-) oxidation rate (v_{Fd}), resulting in the increase in proton motive force (pmf) to induce P700 oxidation. In C₄ plants, pmf increases with the suppression of photosynthetic CO₂ assimilation, although g_{H^+} decreases concomitantly with Y(II) and v_{Fd} .

Supplementary Files

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