

1 The impact of scrubber washwater on inland waters

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18 ABSTRACT

19 **Background:** The International Maritime Organization (IMO) has set limits on sulphur content in
20 fuels for marine transport. However, vessels continue to use these residual high sulphur fuels in
21 combination with exhaust gas cleaning systems (EGCS or scrubbers). Next to high sulphur,
22 combustion of these fuels also results in higher emissions of contaminants including metals and
23 PAHs. In scrubbers, exhaust gases are sprayed with water in order to remove SO_x, resulting in
24 acidic washwater with elevated contaminant concentrations discharged in the aquatic ecosystem.
25 The number of vessels with scrubbers is increasing rapidly, but knowledge on washwater quality
26 and impact are limited.

27 **Results:** The scrubber washwater is found to be acidic with elevated concentrations of e.g. zinc,
28 vanadium, copper, nickel, phenanthrene, naphthalene, fluorene and fluoranthene. Model
29 calculations on the effects of scrubber (20% of vessels) discharge on aquatic systems showed a
30 decrease in pH of 0.015 units and an increase in surface water concentrations for e.g. naphthalene
31 (110% increase) and vanadium (17% increase).

32 **Conclusions:** The IMO established sulphur regulations to mitigate the impact of high sulphur
33 emissions of the maritime sector. However, the use of open loop scrubbers as an abatement
34 technology will not reduce their contribution to the acidification. In addition, different types of
35 scrubbers discharge washwater that is acute toxic for aquatic organisms. However, washwater is
36 diluted and the compounds for which a large increase in surface water concentrations was
37 calculated (Naphthalene > Phenanthrene > Fluorene > Acenaphthene > Vanadium) were not the
38 compounds that already exceed their respective Water Quality Standards (WQS). Nevertheless,
39 existing WQS exceedances of 'priority hazardous substances' (Water Framework Directive) that
40 are also identified in the washwater indicate that coastal waters and estuaries, often with large

41 ecological value, are already under pressure. In these areas the discharge of scrubber washwater
42 should be discouraged.

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44 **Keywords:** marine traffic, pollution, EGCS, SECAs, water quality, acidification

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59 1. Background

60 International shipping significantly contributes to air pollution and the emissions of SO_x, NO_x and
61 PM (particulate matter) from marine transport account for an increasing contribution to global
62 anthropogenic emissions¹. This has a number of environmental consequences², such as ocean
63 acidification³ and disturbance of climate regulation⁴, and also impacts human health in coastal
64 regions through deteriorated air quality^{5, 6}. To mitigate these effects, the IMO (International
65 Maritime Organization) has included annex VI (Prevention of Air Pollution) to the International
66 Convention for the Prevention of Pollution from Ships (MARPOL). This has resulted in a global
67 cap on sulphur (S) in fuel oil of 3.5% (mass percentage) from 2012 and a maximum S content of
68 0.1% in dedicated SO_x Emission Control Areas (SECA's) from 2015 onwards⁷. A new stringent
69 global limit on fuel S content of 0.5% came into force on January 2020.

70 To comply, ships can use compliant low S fuel oil or alternative fuels which are low in S, such as
71 liquefied natural gas (LNG) or methanol. The IMO S limits only apply to atmospheric emissions.
72 Consequently, it is allowed to continue to use high S fuels in combination with an exhaust gas
73 cleaning system (EGCS or scrubber). In scrubbers, the exhaust gases of vessels are sprayed with
74 liquid in order to remove the SO_x before it will be emitted to the air. Scrubbers are capable of
75 removing up to 95% of the SO_x in the exhaust gases and meet the IMO S exhaust limits⁸. Two
76 types of scrubbers dominate the market: 'open loop' and 'closed loop' scrubbers. In an open loop
77 system, also referred to as seawater scrubbing technology, the exhaust gases are sprayed with
78 seawater at a high flow rate, and the SO_x in the exhaust gas is trapped and converted to sulphurous
79 acid (SO₃²⁻) and sulphuric acid (SO₄²⁻). The washwater generated in the scrubber is discharged in
80 the surrounding surface water. Alternatively, closed loop systems use freshwater as the scrubbing
81 medium, which is pre-treated with sodium hydroxide (NaOH). This washwater recirculates in the

82 scrubbing system. The scrubbing capacity is maintained by dosing extra NaOH and periodically
83 discharging smaller volumes of washwater. Often, ‘hybrid systems’ are installed, whereby vessels
84 can shift the scrubber operation between open or closed loop mode.

85 Given the fairly recent changes in the IMO S regulations, the amount of vessels equipped with
86 scrubbers is still limited, but changing rapidly. According to Clarksons Worldfleet Register,
87 consulted in November 2019, nearly 3000 scrubbers have already been installed, which
88 corresponds to 3% of the total number of vessels and 16% of the gross tonnage. This implies that
89 mainly large vessels invest in a scrubber. Additionally, 15% in numbers or 35% in gross tonnage
90 of all vessels ordered at this moment (November 2019) will have a scrubber installed. From an
91 economical perspective, scrubbers are an attractive option, particularly for larger vessels^{9, 10}. In
92 order to comply, the choice between using the more expensive low S fuels or the installation of a
93 scrubber depends largely on the price difference between both, low S fuels and common heavy
94 fuels¹¹. Under most conditions, the scrubber installation costs are recouped within the span of
95 maximum several years^{12, 13}. The number of scrubbers is predicted to continue to increase after the
96 implementation of the more restrictive global sulphur cap in 2020.

97 The use of scrubbers result in a shift of the environmental impact of S from emissions to the
98 atmosphere towards a direct discharge into aquatic systems. Further, the high S fuels used by
99 vessels with scrubbers are generally heavy fuel oils (HFO), which are residual fuels incurred during
100 the distillation of crude oil. Together with high S emissions, these fuels are known to result in
101 higher emissions of other hazardous species including metals and polycyclic aromatic
102 hydrocarbons (PAHs) compared to low S distillates such as marine gas oil (MGO). These
103 contaminants originate from higher concentrations of e.g. metals and PAHs in the fuel and larger
104 emissions during combustion of this residual fuel¹⁴. In general, scrubbers are found to reduce the

105 atmospheric emissions of SO_x or PM to a level that is comparable to emissions when operating on
106 MGO¹⁵⁻¹⁹. But scrubbers are an end-of-pipe solution and a substantial part of the emitted
107 compounds will be trapped in the scrubber washwater, discharged in the surrounding surface water
108 with potential consequences for aquatic ecosystems^{17, 20-22}. Existing studies are limited, mainly
109 focus on open marine systems and conclude that the overall impact of scrubber use on pH changes
110 and contaminant concentrations is expected to be small under most conditions^{21, 23-25}. Yet, the long
111 term accumulation of contaminants caused by scrubber discharges can be of concern in aquatic
112 systems where ships are numerous and discharge into a relatively restricted water bodies including
113 coasts, estuaries or harbours^{26, 27}.

114 Data on washwater contaminant concentration are scarce, often proprietary and rarely published.
115 In present study, an extensive dataset on washwater contaminant concentrations and acidity is
116 compiled, based on own measurements and received and published datasets. This data allowed us
117 to calculate the impact of scrubber use on water quality for two scenarios (10% and 20% scrubber
118 use) for the Antwerp harbour docks and the Scheldt estuary. While the IMO regulatory framework
119 primarily focuses on atmospheric emissions, also the discharge of washwater is regulated to a
120 certain extent. Washwater discharge criteria were set for pH (min. of 6.5, measured at 4 m from
121 the overboard discharge point), for PAHs (max. 50 µg L⁻¹ PAH Phe equivalent at a flow rate of 45
122 m³ MWh⁻¹) and turbidity (max. 25 NTU (Nephelometric turbidity units) above the inlet water
123 turbidity) (IMO, Resolution MEPC.184(59) and MEPC.259(68)). No criteria for metals are
124 included. However, in current Belgian legislation the discharge of contaminated water from ships
125 into their surrounding surface water is only accepted in several exceptional cases (e.g. wastewater
126 from kitchens)²⁸. Consequently, the use of open loop scrubbers or closed loop scrubbers with bleed-
127 off discharge are not allowed in Belgian inland waters. To comply, vessels need to operate on
128 compliant low sulphur fuel or use scrubbers in closed loop mode with the boundary condition that

129 no washwater is discharged (zero discharge mode). While there is no current impact of scrubbers
130 on Belgian water bodies, it is important to get insight in possible consequences for European ports,
131 rivers, estuaries and coastal regions in order to streamline legislation. Ahead of the implementation
132 of the SECA's, shipowners have already criticized the uncertainty and the inconsistency between
133 the Member States on the use of scrubbers in order to comply with the requirements of the Sulphur
134 Directive. In an open letter, the European Community Shipowners' Association (ECSA) urges that
135 establishing legal certainty about proper compliance and enforcement together with a fair level
136 playing field between shipping operators and between transport modes are a must²⁹.

137 2. Methods

138 2.1. Sampling

139 Washwater samples were collected from two separate marine vessels operating a scrubber. The
140 first vessel was equipped with a hybrid scrubber and was sampled on two occasions in October
141 2014: when at berth in the port of Antwerp, operating in closed loop mode and when sailing on the
142 Scheldt estuary in open loop mode. The second vessel had an open loop scrubber and was sampled
143 twice in October 2015: when sailing at the North Sea and when manoeuvring in the port of
144 Antwerp. Discharge and sampling of washwater in these Belgian waters was permitted by the
145 Flemish Environmental Agency for present research. Detailed information on scrubber type, fuel
146 and operating conditions can be found in the datasheet (Table S1). Samples were taken from a tap
147 close to the scrubber outlet. Right after sampling the temperature (°C) and pH were measured with
148 a temperature-pH electrode connected to a portable multi meter (HQ30D, Hach, US). Water
149 samples were collected in 1 L glass bottles for PAH analysis and in 0.25 L high density
150 polyethylene (HDPE) bottles for metal analysis. All samples were stored cool during transport.
151 Metal analyses were performed after acid digestion with HCl and HNO₃ by inductively coupled

152 plasma optical emission spectrometry (ICP-OES) following standard method ISO 11885.
153 Measured metals are Arsenic (As), Cadmium (Cd), Chromium (Cr), Copper (Cu), Mercury (Hg),
154 Lead (Pb), Nickel (Ni), Zinc (Zn) and Vanadium (V). PAHs were determined by gas
155 chromatography/mass spectrometry (GC/MS) following standard method EPA 8270 D. Measured
156 PAHs are Acenaphthene (Ace), Acenaphthylene (Acy), Anthracene (Ant), Benzo(a)anthracene
157 (B(a)A), Benzo(a)pyrene (B(a)P), Benzo(b)fluoranthene (B(b)F), Benzo(g,h,i)perylene (B(ghi)P),
158 Benzo(k)Fluoranthene (B(k)F), Chrysene (Chr), Dibenzo(a,h)anthracene (D(ah)A), Fluoranthene
159 (Fluoran), Fluorene (Flu), Indeno(1,2,3c-d)pyrene (I(123cd)P), Naphtalene (Naph), Phenantrene
160 (Phe) and Pyrene (Pyr). These results were combined with additional datasets on pH, metal and
161 PAH concentrations: (1) received from the same shipping companies as we sampled, (2) resulting
162 from sampling campaigns organised by the exhaust gas cleaning system association (EGCSA) and
163 (3) from literature. All data was combined in a database (Table S1). Parameters with many values
164 below limit of quantification (BLOQ) were excluded from further analysis (Table S2).

165 2.2. Scenario's and model calculation

166 To calculate the impact of discharged washwater on the water quality two scenarios were defined.
167 Scenario LOW assumed that 10% of the total ship emissions were treated by scrubbers and vessels
168 discharged at average washwater concentrations. Scenario HIGH assumed a 20% treatment by
169 scrubbers and discharge at 90th percentile of washwater concentrations (Table 1 for scenario HIGH,
170 Table S3 for scenario LOW).

171 The calculated scenarios included all fuel used by the vessels, main and auxiliary engines. For
172 manoeuvring and berthing in harbours, auxiliary engines are typically used. As these auxiliary
173 engines are not always connected to the scrubber, the calculated changes in metal and PAH
174 concentrations in the harbour docks surface water could be an overestimation.

175 The contaminant input J_c (kg y^{-1}) to the water bodies (harbour docks and estuary) was calculated
176 as:

$$177 \quad J_c = Q_w * P_t * F_s * C_c \quad (\text{Eq. 1})$$

178 With Q_w (L MWh^{-1}) the discharge of washwater (closed or open loop) per unit of generated power
179 of the vessel, P_t (GWh y^{-1}) is the total power generated by the vessel in a certain water body
180 (harbour docks or Scheldt estuary), F_s the share of total emissions treated by scrubbers (0.1 for
181 scenario LOW or 0.2 for scenario HIGH), and C_c (kg L^{-1}) is the concentration of the contaminant
182 in the washwater. The power P_t generated by the vessels in the Scheldt estuary (182 GWh y^{-1}) and
183 Antwerp harbour docks (472 GWh y^{-1}) is well constrained (data provided by the Antwerp port
184 Authority). An average washwater discharge was calculated for open loop ($51 \pm 30 \text{ m}^3 \text{ MWh}^{-1}$,
185 $N=44$) and closed loop ($0.28 \text{ m}^3 \pm 0.17 \text{ m}^3 \text{ MWh}^{-1}$, $N=7$), based on the available data (Table S1).
186 Note however that some vessels with closed loop systems will not discharge any washwater, as
187 they retain the washwater on board and deliver it to port reception facilities on shore (personal
188 communication shipping companies). The concentrations of metals and PAHs measured in the inlet
189 water were not used for the calculation because values were often below limit of quantification
190 (BLOQ). Additionally, concentrations of e.g. zinc in inlet water samples were sometimes
191 unrealistically high (Table S1), which can be caused by sampling inlet water from a valve on board
192 which could have resulted in elevated metal concentrations originating from the metal tubing of
193 the scrubber system. Instead, available data on total metal and PAH concentrations in the harbour
194 docks were used and average values were subtracted from the outlet concentrations prior to the
195 calculation of fluxes and changes in surface water concentrations (see Table 1 for concentrations).
196 All outlet washwater concentrations reported as BLOQ were replaced by the respective LOQ/2 and
197 included in the calculations.

198 To estimate the impact of exhaust scrubber discharge on contaminant concentrations, total
199 alkalinity (TA) and pH in the docks, a dynamic model was set-up. In this model, the docks are
200 considered well mixed. The water body is affected by influx of freshwater (16 m³ s⁻¹, fixed water
201 composition based on observations), by exchange with the adjacent estuary over the locks (fixed
202 water composition of the estuary based on observations, dispersive flux proportional to
203 concentration difference, fixed bulk dispersion coefficient of 100 m³ s⁻¹, see appendix), by the
204 scrubber efflux, and by gaseous exchange with the atmosphere.

205 Changes in contaminant concentrations are calculated as:

$$206 \quad \Delta C_c = \frac{J_c}{Q_r} \quad (\text{Eq. 2})$$

207 With ΔC_c (μg L⁻¹) the mean concentration change in the receiving water body caused by scrubber
208 discharge, J_c (kg y⁻¹) the contaminant input from scrubbers calculated with Eq. 1 and Q_r (L y⁻¹) is
209 the flowrate of water through the receiving water body. The mean flow rate of the Scheldt estuary
210 was 100 m³ s⁻¹, for the docks of the port Antwerp Q_r was the sum of the flow rate (16 m³ s⁻¹) and
211 the dispersive exchange D through the locks (100 m³ s⁻¹). Calculation assumptions were that all
212 discharged contaminants were evenly distributed in the water column and stayed in suspension, i.e.
213 during their stay in the docks (average residence time = 19 days) and in the estuary (average
214 residence time = 2.5 months). Calculated increase in concentrations were compared with surface
215 water concentrations measured during regular water quality monitoring programs from 2015-2016
216 in the receiving water body (data from the Flemish Environmental Agency and Antwerp Port
217 Authority; in the docks n=15 (metals) and n=15 (PAHs), and in the Scheldt estuary n=115 (metals)
218 and n=20 (PAHs).

219 For changes in TA, SumCO₂ and H₂SO₄, the CO₂(aq) in scrubber effluent was assumed to be in
220 equilibrium with CO₂ in flue gas, assumed to have a fixed partial pressure of 0.1 atm³⁰. The
221 carbonate balance in the scrubber effluent was computed at observed effluent water temperature
222 (T= 25 °C). Scrubber flux of H₂SO₄ and TA were determined by assuming that all S in the exhaust
223 is captured by the washing process. Further, 20% of vessels equipped with an open loop scrubber,
224 2.1% S content in fuel, and a total of 90 x 10⁶ ton fuel use per year for all vessels in the right bank
225 of the Antwerp harbour were assumed. The model was run to steady state to assess the difference
226 in water composition in the docks with and without scrubbers. Carbonate balances were computed
227 with the R package AquaEnv³¹. The model was integrated with the R package deSolve³². The model
228 code and scripts to run the scenario analyses are available on GitHub [link to the model will be
229 made available upon acceptance].

230 3. Results and Discussion

231 3.1. Scrubber use

232 For some vessels the installation of a scrubber is considered an attractive option to comply with
233 the IMO regulations on reduction of sulphur emissions¹². The number of vessels with a scrubber is
234 currently still rather limited, but increasing rapidly. Also in Belgium an increasing number of
235 vessels with scrubbers call at the ports of Zeebrugge, Ghent and Antwerp. From January till
236 November 2019 about 350 unique vessels equipped with a scrubber visited the port of Antwerp,
237 with a total of approximately 1250 calls (information Antwerp Port Authority and Clarksons
238 Worldfleet Register). This corresponds to 8.7% of the total number of vessels and 9.5% of the total
239 number of calls. Scrubber types of vessels in the Antwerp port are evenly distributed between open
240 loop and hybrid scrubbers.

241 In present study, the impact of washwater in the harbour docks is calculated with 10% and 20%
242 scrubbers as two different future scenarios. It is however difficult to estimate the future use of
243 scrubbers. Since the installation of a scrubber is economically profitable under most scenarios¹⁰, a
244 further increase can be expected. However, the fuel market is changing rapidly and low sulphur
245 heavy fuels which are cheaper than distillates are becoming available and might influence scrubber
246 interest¹.

247 3.2.Scrubber washwater quality

248 To get insight in the concentrations of contaminants present in scrubber washwater a large database
249 was compiled with results from different sources, including own samples, received datasets and
250 literature (Table S1). Metal and PAH concentrations are found to be elevated compared to surface
251 water concentrations or Water Quality Standards (WQS) (Fig. S1, S2). The variation in
252 concentrations within parameters is large with a concentration range of four orders of magnitude
253 for some metals and a range of two orders of magnitude for some PAHs. The high variation in
254 washwater concentrations can be attributed to many different factors including scrubber type,
255 additives, fuel origin³³, fuel sulphur content³⁴, engine load¹⁴ or the presence of treatment facilities
256 before the washwater is discharged. In general, the contaminants originate from the fuel, lubricant
257 oil or combustion process, are transported to the smokestack, washed out by the scrubber water
258 and end up in the washwater. Metal concentrations in fuels are found to vary substantially and are
259 related to the crude oil origin and refinery process³³. Since a substantial part of the metals in the
260 fuel is expected to end up in the scrubber washwater^{15, 17}, the fuel origin will directly affect the
261 washwater metal concentrations. Vessels with scrubbers usually operate on high sulphur fuel oil
262 (HSFO). These are residual fuels that are known to contain higher concentrations of metals
263 compared to distillate fuel (DF), e.g. MGO³⁵. The metals V and Ni and to a lesser extent Cu are

264 typically tracers for residual fuel. For Zn the fuel and the lubricant oil were found to contribute
265 equally to the emissions^{35, 36}. Also in present study Zn, V, Ni and Cu are the metals that are
266 measured in the highest concentrations in the scrubber washwater with average values and standard
267 deviation (STDEV) of 1745 $\mu\text{g L}^{-1}$ (STDEV 5460 $\mu\text{g L}^{-1}$) for Zn, 752 $\mu\text{g L}^{-1}$ (STDEV 2437 $\mu\text{g L}^{-1}$)
268 ¹) for V, 670 $\mu\text{g L}^{-1}$ (STDEV 1533 $\mu\text{g L}^{-1}$) for Ni and 160 $\mu\text{g L}^{-1}$ (STDEV 417 $\mu\text{g L}^{-1}$) for Cu.
269 Besides, the concentration of Cr was found to be high in the washwater of several scrubbers
270 operating in closed loop, while for most other washwater samples no elevated concentrations were
271 measured. It is not clear where the high concentrations originate from, as Cr is generally not present
272 in fuel³⁵. It is possible that corrosion or abrasion of the scrubber installation, stimulated by the
273 acidic washwater, is a source of Cr, as was previously suggested for Cu^{21, 23}.

274 The higher emissions originating from combusting HSFO compared to DF, as reported for metals,
275 is even more pronounced for PAHs, with atmospheric emissions found to be 200 times higher when
276 operating on HSFO¹⁴. In the case of HSFO, the PAHs in the exhaust generally originate directly
277 from the fuel¹⁴. The PAH concentrations in the emissions of marine engines are found to be
278 dominated by Phe, Naph, Fluoran and Flu³⁷, which corresponds to the PAHs that were measured
279 in high concentration in the washwater of present study with average values and STDEV of 2744
280 ng L^{-1} (STDEV 2895 ng L^{-1}) for Naph, 2001 ng L^{-1} (STDEV 1678 ng L^{-1}) for Phe, 708 ng L^{-1}
281 (STDEV 587 ng L^{-1}) for Flu and 186 ng L^{-1} (STDEV 217 ng L^{-1}) for Fluoran.

282 Large variation in concentrations between washwater samples could be observed and can to a large
283 extend be explained by the differences that exist between scrubbers operating in open or closed
284 loop (Fig. 2). As the washwater in closed loop scrubbers circulates within the system, contaminants
285 accumulate over time, resulting in higher concentrations of metals and PAHs in the washwater
286 compared to open loop mode (2-way ANOVA; metals: $F_{1,323}= 26,7$; $p<0,001$, PAH: $F_{1,475}= 7.27$;
287 $p=0,007$) (Fig. 1). Metal concentrations in closed loop washwater are on average 43 times higher

288 for metals and 1.3 times higher for PAHs compared to open loop washwater. However, in closed
289 loop scrubbers the scrubbing capacity is kept high by dosing sodium hydroxide resulting in a low
290 volume of water needed to trap SO_x efficiently. Closed loop systems discharge discontinuous and
291 lower volumes of washwater (bleed off) with a calculated average volume of 0.28 m³ MWh⁻¹
292 (STDEV=0.17, N=7). In contrast, open loop systems need a large volume of surface water to ensure
293 removal of SO_x from the exhaust with discharge volumes that are roughly 200 times higher
294 (calculated average 51 m³ MWh⁻¹, STDEV=30, N=44). The circulation of water and smaller
295 washwater volumes when operating in closed loop allows efficient treatment of the washwater
296 using a hydrocyclone with removal of particles before discharge. Hereby, contaminants are
297 scavenged in a sludge fraction that is stored and delivered to port reception facilities resulting in a
298 lower total discharge of contaminants to the surrounding surface water (6 times for metals and 183
299 times for PAHs) for scrubbers operating in closed loop mode (differences are significant for metals:
300 2-way ANOVA; F_{1,323}= 6.56; p=0,011 and PAHs: F_{1,475}= 30.4; p<0,001) (Fig. 2). The differences
301 between metals and PAHs indicate that PAHs are trapped much more efficient in the sludge fraction
302 than metals by hydrocyclone treatment in closed loop mode. Also for scrubbers operating in open
303 loop, the use of washwater treatment systems is reported (Table S1). However, treatment of the
304 large washwater flow rates is less straightforward. A vessel sailing with 15 MW engine power will
305 discharge roughly 200 L s⁻¹. The large variation in concentrations and the limited number of
306 scrubbers that reported an open loop system with treatment in the dataset does not allow to draw
307 conclusions on differences in concentrations between open loop with and without treatment. The
308 acidity of the washwater in closed loop mode can be controlled by dosing the scrubbing media
309 NaOH resulting in higher average pH values in the discharged water (6.8, STDEV=1.7, n=6)
310 compared to the average pH values in open loop mode (4.8, STDEV=1.4, n=21).

311

312 3.3.Impact on water quality

313 Generally, total acidifying potential and emissions of hazardous substances of vessels with
314 scrubbers operating on HSFO are higher than from vessels operating on low sulphur compliant
315 fuels. A substantial part of these emissions are directly discharged with the washwater into
316 receiving aquatic ecosystems. The discharge of washwater has impacts on different spatial and
317 temporal scales. Right after discharge there will be an impact in the immediate vicinity of the vessel
318 (acute toxicity, small spatial and temporal scale), while on a far longer time scale, pollutants will
319 be dispersed throughout the larger water body, which then leads to an increase in mean pollutant
320 levels throughout the port or estuary (chronic toxicity, larger spatial and temporal scale).

321 The concentrations of most PAHs and all metals in the undiluted closed loop washwater largely
322 exceed their WQS (Tabel 1,2) and are expected to be acutely toxic for most aquatic organisms.
323 Acute toxic effects of scrubber washwater on phyto- and zooplankton are reported, even at
324 concentrations of metals and PAHs much lower than the concentrations reported in present work²⁰.
325 ²². The synergistic effects caused by the mixture of metals and PAHs combined with low pH in
326 scrubber washwater result in higher toxicity than estimated from the effect thresholds of the
327 individual compounds^{20, 38}. However, the effects of washwater are strongly influenced by dilution
328 with surrounding surface water. Buhaug et al. (2006) modelled that washwater at 50 m behind the
329 vessels will be diluted 2000 times for vessels sailing in open sea and 1750 times during port
330 operation at lower speed. The extent of dilution will depend on vessel activity (at berth,
331 manoeuvring, sailing) and physical characteristics of the receiving water body such as dimensions
332 and flow rate, which complicates the prediction of scrubber washwater toxicity. When applying
333 the dilution factor of 2000 on washwater metal and PAH concentrations almost no compounds will
334 exceed their WQS whereby no acute toxicity is expected.

335 The increase in metal and PAH concentrations in aquatic ecosystems that originate from scrubber
336 washwater is expected to be higher in inland waterbodies such as estuaries, rivers or harbours
337 compared to large open marine systems. The accumulation of metals and PAHs in the surface water
338 of the Antwerp harbour docks and the Scheldt estuary was calculated for a ‘scenario LOW’ (10%
339 open loop scrubbers and average washwater concentrations) and a ‘scenario HIGH’ (20% open
340 loop scrubbers and 90th percentile washwater concentrations) (Fig. 3). In particular the
341 concentration of several PAHs (Flu, Naph and Phen) in the surface water of the Antwerp harbour
342 docks was simulated to increase due to scrubber discharge. An increase in concentration of 23%
343 under the ‘scenario LOW’ and 110% under the ‘scenario HIGH’ was calculated for naphthalene.
344 The mean concentration of vanadium in the docks would increase with 3.4% under scenario LOW
345 and 17% under scenario HIGH. The time vessels spend in the Scheldt estuary is shorter than in the
346 harbour docks, which results in lower total amount of fuel use, a lower volume of scrubber
347 washwater discharged and a smaller effect on metal and PAH concentrations in the surface water
348 compared to the docks. For the Scheldt estuary mean naphthalene concentrations are calculated to
349 increase with 3.0% (scenario LOW) to 14% (scenario HIGH). For vessels with scrubbers in closed
350 loop mode a large part of the metals and PAHs is removed from the washwater, trapped in the
351 sludge fraction and delivered on shore, with a smaller increase in pollutant concentrations as a
352 consequence (Table 1).

353 Some of the pollutants that are present in scrubber washwater are already exceeding (Flu, Pyr) or
354 close to exceedance (Ni, Zn) of their respective WQS in the surface water of the harbour docks or
355 the Scheldt estuary (Fig 3). However, the compounds for which a large increase in concentrations
356 was calculated (Naph>Phe>Flu>Ace>V) are not the compounds that are expected to pose the
357 highest risk, based on the exceedances of the WQS. Nevertheless, several pollutants that are
358 measured in elevated concentrations and discharged with the scrubber washwater are identified as

359 'priority substances' (Fluoran, Naph, Ni) or 'priority hazardous substances' (Ant, B(a)P, Cd) by
360 the European Water Framework Directive (WFD) and as such are of major concern for European
361 Waters. WQS exceedances of these compounds indicate that these aquatic systems are under
362 pressure of high contaminant concentrations and progressive reduction of pollution from priority
363 substances and the cessation or phasing-out of discharges, emissions and losses of priority
364 hazardous substances is required³⁹. Many European coasts and estuaries are part of Natura 2000,
365 the largest coordinated network of protected areas in the world to safeguard valuable and threatened
366 species and habitats. Mainly in these areas with large ecological values the discharge of scrubber
367 washwater should be restricted. In addition, the use of scrubbers deflect attention from
368 development of cleaner fuels⁴⁰. However, also the emissions from vessels operating on low sulphur
369 fuel are variable and subject to changes. With the sulphur regulations, new types of low sulphur
370 heavy fuel oils (hybrid fuels, intermediate fuels or ECA fuels) have entered the market. How these
371 fuels influence emission of metals and PAHs is not clear yet. It will, likely be necessary to limit
372 the use of all low quality fuels, with high and low sulphur content and instead encourage the use
373 of distillate fuels, mainly in coasts, estuaries and inland water bodies with large ecological value¹.

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384 **Table 1.** Summarising numbers on scrubber washwater concentrations, fluxes and impact on water
 385 quality for scenario high (20% scrubbers).

	Units	CLOSED loop - Scenario HIGH			OPEN loop - Scenario HIGH				
		Conc. Docks (1)	WQS (2)	Discharge conc. (90th perc.)	Flux kg y ⁻¹ (20% scrubbers)	Conc. Increase docks	Discharge conc. (90th perc.)	Flux kg y ⁻¹ (20% scrubbers)	Conc. Increase docks
Cr		3.34	5	10120	266	0.073	45.0	108	0.055
Cu		8.38	7	1780	47	0.013	130	313	0.160
Ni	µg L ⁻¹	5.86	4*	6060	159	0.044	127	305	0.159
Zn		32.9	20	25000	657	0.180	460	1106	0.561
V		3.94	4	9100	240	0.065	324	779	0.421
Ace		4.92	60	745	1.95E-02	5.33E-06	648	1.56	0.845
Acy		7.20	4000	185	4.68E-03	1.28E-06	536	1.29	0.695
Ant		2.67	100*	446	1.17E-02	3.19E-06	308	0.741	0.401
Fluoran		9.51	6.3*	661	1.72E-02	4.69E-06	478	1.15	0.616
Flu	ng L ⁻¹	3.72	2000	2370	6.23E-02	1.70E-05	1200	2.89	1.57
Naph		8.24	2000*	6370	1.68E-01	4.58E-05	6960	16.7	9.14
Phe		7.95	100	6970	1.83E-01	5.01E-05	3700	8.90	4.85
Pyr		13.0	40	554	1.42E-02	3.89E-06	1220	2.93	1.59
Total PAH		58.1		22200	5.83E-01	1.59E-04	13620	32.7	17.8

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387 (1) Average values for total concentrations in the harbour docks

388 (2) Water Quality Standards (WQS) from the EU WFD (*) or Flanders (dissolved concentrations for metals, total
 389 concentrations for PAHs)

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391 3.4. Acidification

392 Marine transport related emissions of NO_x and SO_x cause acidification of terrestrial and marine
 393 ecosystems⁴¹. The contribution of anthropogenic N and S depositions to ocean acidification
 394 account only for a few percent of the acidifying impact of the global anthropogenic emissions
 395 (mainly caused by CO₂)⁴². However, in certain restricted areas such as coastal waters with
 396 important shipping lanes or large harbours, the acidifying effect caused by NO_x and SO_x can exceed
 397 the effect of overall anthropogenic CO₂ emissions⁴¹. In open loop scrubber systems the natural
 398 buffering capacity (alkalinity) of the sea or river water is used to neutralize the acid ions. Mean
 399 alkalinity in coastal waters varies between 2100 and 2400 µmol/l⁴³ and is high enough to guarantee
 400 high SO_x removal efficiencies. Due to a calcium rich bedrock, the mean alkalinity in the surface
 401 water of the Scheldt estuary (4400 µmol L⁻¹) and the docks (3400 µmol L⁻¹) is high. In closed loop
 402 systems the acidity of the washwater is buffered by dosing NaOH to the circulating washwater in
 403 order to have a bleed of that is neutral (pH around 6-8). Since vessels with scrubbers operate on

404 high sulphur fuel (up to 3.5%) and the acidifying sulphur compounds are discharged directly into
405 the surface water, their acidifying capacity is much larger than vessels operating on low sulphur
406 fuel (0.1% in SECAs). In addition, the elevated concentrations of CO₂ in the exhaust gas result in
407 elevated dissolved inorganic carbon concentrations in the washwater with additional acidification
408 of the receiving water bodies. Model simulations with scenario HIGH (20% open loop scrubbers)
409 show a decrease in pH of 0.015 units caused by washwater discharge (Fig. 4). The alkalinity will
410 comparably decrease slightly with 6 μmol L⁻¹ or 0.16% and total sulphate concentrations will
411 increase with 3 μmol L⁻¹ or 0.08%. For the Baltic Sea, the water pH was calculated to decrease by
412 open loop scrubber use with roughly 0.0015 units (50% scrubbers scenario) to 0.003 units (100%
413 scrubbers scenario)⁴¹.

414 Since preindustrial times global ocean pH decreased with approximately 0.1 units with related
415 negative consequences for marine ecosystems⁴⁴. Among many other sources, SO_x emissions of
416 marine transport contributes to this acidification. The IMO established sulphur regulations to
417 mitigate the impact of high sulphur emissions of the maritime sector. However, the use of open
418 loop scrubbers as an abatement technology will not reduce their contribution to the acidification.

419

420 4. Conclusions

421 The number of vessels with a scrubber is increasing rapidly. These vessels usually operate on high
422 sulphur residual fuels that are known to result in higher emissions of hazardous substances such as
423 metals and PAHs compared to compliant low sulphur distillates. In scrubbers, sulphur, metals and
424 PAHs are washed out of the atmospheric exhaust resulting in discharge of acidic washwater with
425 elevated contaminant concentrations. This washwater is found to be acute toxic for aquatic
426 organisms and a substantial long term increase in the concentrations of Naph, Phe, Flu, Ace and V
427 following scrubber washwater discharge was calculated for inland waterbodies such as estuaries,

428 rivers or harbours. The compounds for which a large increase in concentrations was calculated
429 (Naph>Phe>Flu>Ace>V) are not the compounds that are expected to pose the highest risk, based
430 on the exceedances of the WQS. Nevertheless, several pollutants that are discharged with the
431 scrubber washwater are identified as ‘priority substances’ or ‘priority hazardous substances’ by the
432 European Water Framework Directive and as such are of major concern for European Waters.
433 WQS exceedances of these compounds indicate that many European aquatic systems are already
434 under pressure. As such, mainly in coast and estuaries with large ecological values the discharge
435 of scrubber washwater should be restricted.

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447 **Figure Legends**

448 **Figure 1.** Metal (A) and PAH (B) concentrations ($\mu\text{g L}^{-1}$) in washwater from scrubbers operating
449 in closed loop (shaded boxes) and open loop (grey boxes). Boxplots with 5th and 95th percentile
450 (whiskers), 25th, median and 75th percentile and outliers (dots). Y-axis in logarithmic scale.

451 **Figure 2.** Total discharge (g MWh^{-1}) of metals (A) and PAHs (B) in washwater from scrubbers
452 operating in closed loop (shaded boxes) and open loop (grey boxes). Boxplots with 5th and 95th
453 percentile (whiskers), 25th, median and 75th percentile and outliers (dots). Y-axis in logarithmic
454 scale.

455 **Figure 3.** Increase in metal and PAH surface water concentrations (%) in the harbour docks (A)
456 and the Scheldt estuary (B) caused by open loop scrubber discharge compared to current
457 concentrations (grey, 100%) calculated with the scenario LOW (white) and scenario HIGH (black).
458 The ratio between water quality standards (WQS) and current (grey) metal and PAH surface water
459 concentrations in the docks (C) and the Scheldt estuary. The calculated concentration increase
460 caused by open loop scrubber discharge calculated by the scenario LOW (white) and scenario
461 HIGH (black).

462 **Figure 4.** Changes in total alkalinity (TA, $\mu\text{mol L}^{-1}$), H_2SO_4 ($\mu\text{mol L}^{-1}$) and pH in the surface
463 water from the Antwerp harbour docks caused by open loop scrubber discharge calculated with
464 scenario HIGH (20% open loop scrubbers).

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477 **List of abbreviations**

478 IMO: International Maritime Organization

479 EGCS: exhaust gas cleaning systems

480 PM: particulate matter

481 MARPOL: International Convention for the Prevention of Pollution from Ships

482 S: sulphur

483 SO_x: sulphur (x)oxide

484 SECA Emission Control Area

485 LNG: liquefied natural gas

486 NaOH: sodium hydroxide

487 HFO: heavy fuel oil

488 PAH: polycyclic aromatic hydrocarbons

489 MGO: marine gas oil

490 ECSA: European Community Shipowners' Association

491 HDPE: high density polyethylene

492 ICP-OES: inductively coupled plasma optical emission spectrometry

493 GC/MS: gas chromatography/mass spectrometry

- 494 As: Arsenic
- 495 Cd: Cadmium
- 496 Cr: Chromium
- 497 Cu: Copper
- 498 Hg: Mercury
- 499 Pb: Lead
- 500 Ni: Nickel
- 501 Zn: Zinc
- 502 V: Vanadium
- 503 Ace: Acenaphthene
- 504 Acy: Acenaphthylene
- 505 Ant: Anthracene
- 506 B(a)A: Benzo(a)anthracene
- 507 B(a)P: Benzo(a)pyrene
- 508 B(b)F: Benzo(b)fluoranthene
- 509 B(ghi)P: Benzo(g,h,i)perylene
- 510 B(k)F: Benzo(k)Fluorathene

- 511 Chr: Chrysene
- 512 D(ah)A: Dibenzo(a,h)anthracene
- 513 Fluoran: Fluoranthene
- 514 Flu: Fluorene
- 515 I'123cd)P: Indeno(1,2,3c-d)pyrene
- 516 Naph: Naphtalene
- 517 Phe: Phenantrene
- 518 Pyr: Pyrene
- 519 EGCSA: exhaust gas cleaning system association
- 520 BLOQ: below limit of quantification
- 521 TA: total alkalinity
- 522 WQS: Water Quality Standards
- 523 HSFO: high sulphur fuel oil
- 524 DF: distillate fuel
- 525 STDEV: standard deviation
- 526 WFD: Water Framework Directive
- 527

528 DECLARATIONS

529 *Ethics approval and consent to participate:* Not applicable.

530 *Consent for publication:* Not applicable.

531 *Availability of data and material:* All data generated or analysed during this study are included in
532 this published article and its supplementary information files.

533 *Competing interests:* The authors declare that they have no competing interests.

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535 *Authors' contributions:* JT collected samples, performed chemical analysis, interpreted the data and
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544 and data on the water quality and to the Royal Belgian Institute of Natural Sciences for their advice.

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547 **Supporting Information.**

548 File 1: TableS1: an Excel file that contains the compiled database on scrubber use and washwater
549 quality. A total of 127 samples, based on own sampling, received datasets and literature.

550 File 2: A Word document with information on calculations, 2 Tables and 2 Figures.

551 Table S2: all measured parameters with number of values below the limit of quantification

552 Table S3: Summarising numbers on scrubber washwater concentrations, fluxes and impact on
553 water quality for scenario low (10% scrubbers). The numbers of scenario high (20% scrubbers) are
554 included in the manuscript (Table 1).

555 Figure S1: All scrubber washwater metal concentrations

556 Figure S2: All scrubber washwater PAH concentrations

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- 568 1. Andersson, K.; Brynolf, S.; Lindgren, J. F.; Wilewska-Bien, M., *Shipping and the Environment: Improving Environmental Performance in Marine Transport*. Springer: Berlin, 2016; p 425.
- 569 2. Claremar, B.; Haglund, K.; Rutgersson, A., Ship emissions and the use of current air cleaning
570 technology: contributions to air pollution and acidification in the Baltic Sea. *Earth System Dynamics*
571 **2017**, 8, (4), 901-919.
- 572 3. Hassellöv, I.-M.; Turner, D. R.; Lauer, A.; Corbett, J. J., Shipping contributes to ocean acidification.
573 *Geophysical Research Letters* **2013**, 40, (11), 2731-2736.
- 574 4. Capaldo, K.; Corbett, J. J.; Kasibhatla, P.; Fischbeck, P.; Pandis, S. N., Effects of ship emissions on
575 sulphur cycling and radiative climate forcing over the ocean. *Nature* **1999**, 400, 743.
- 576 5. Viana, M.; Hammingh, P.; Colette, A.; Querol, X.; Degraeuwe, B.; Vlieger, I. d.; van Aardenne, J.,
577 Impact of maritime transport emissions on coastal air quality in Europe. *Atmospheric Environment* **2014**,
578 *90*, 96-105.
- 579 6. Corbett, J. J.; Winebrake, J. J.; Green, E. H.; Kasibhatla, P.; Eyring, V.; Lauer, A., Mortality from
580 ship emissions: A global assessment. *Environmental Science & Technology* **2007**, 41, (24), 8512-8518.
- 581 7. EC, The sulphur content of marine fuels 2012/33/EU. **2012**.
- 582 8. Andreasen, A.; Mayer, S., Use of Seawater Scrubbing for SO₂ Removal from Marine Engine
583 Exhaust Gas. *Energy & Fuels* **2007**, 21, (6), 3274-3279.
- 584 9. Lindstad, H. E.; Rehn, C. F.; Eskeland, G. S., Sulphur abatement globally in maritime shipping.
585 *Transportation Research Part D: Transport and Environment* **2017**, 57, 303-313.
- 586 10. Nikopoulou, Z., Incremental costs for reduction of air pollution from ships: a case study on North
587 European emission control area. *Maritime Policy & Management* **2017**, 44, (8), 1056-1077.
- 588 11. Jiang, L.; Kronbak, J.; Christensen, L. P., The costs and benefits of sulphur reduction measures:
589 Sulphur scrubbers versus marine gas oil. *Transportation Research Part D: Transport and Environment*
590 **2014**, 28, 19-27.
- 591 12. Panasiuk, I.; Turkina, L., The evaluation of investments efficiency of SO_x scrubber installation.
592 *Transportation Research Part D: Transport and Environment* **2015**, 40, 87-96.
- 593 13. Carr, E. W.; Corbett, J. J., Ship Compliance in Emission Control Areas: Technology Costs and Policy
594 Instruments. *Environmental Science & Technology* **2015**, 49, (16), 9584-9591.
- 595 14. Sippula, O.; Stengel, B.; Sklorz, M.; Streibel, T.; Rabe, R.; Orasche, J.; Lintelmann, J.; Michalke, B.;
596 Abbaszade, G.; Radischat, C.; Gröger, T.; Schnelle-Kreis, J.; Harndorf, H.; Zimmermann, R., Particle
597 Emissions from a Marine Engine: Chemical Composition and Aromatic Emission Profiles under Various
598 Operating Conditions. *Environmental Science & Technology* **2014**, 48, (19), 11721-11729.
- 599 15. Fridell, E.; Salo, K., Measurements of abatement of particles and exhaust gases in a marine gas
600 scrubber. *Proceedings of the Institution of Mechanical Engineers, Part M: Journal of Engineering for the*
601 *Maritime Environment* **2014**, 230, (1), 154-162.
- 602 16. Di Natale, F.; Carotenuto, C., Particulate matter in marine diesel engines exhausts: Emissions and
603 control strategies. *Transportation Research Part D: Transport and Environment* **2015**, 40, 166-191.
- 604 17. Endres, S.; Maes, F.; Hopkins, F.; Houghton, K.; Mårtensson, E. M.; Oeffner, J.; Quack, B.; Singh,
605 P.; Turner, D., A New Perspective at the Ship-Air-Sea-Interface: The Environmental Impacts of Exhaust
606 Gas Scrubber Discharge. *Frontiers in Marine Science* **2018**, 5, (139).
- 607 18. Bengtsson, S.; Andersson, K.; Fridell, E. *Life cycle assessment of marine fuels. A comparative*
608 *study of four fossil fuels for marine propulsion*; Technical report no 11:125; Chalmers University of
609 Technology: Gothenburg, Sweden, 2011.
- 610 19. Andersson, K.; Brynolf, S., Fuels in the Baltic Sea after SECA, Report Trafikanalys. **2016**.
- 611

- 612 20. Koski, M.; Stedmon, C.; Trapp, S., Ecological effects of scrubber water discharge on coastal
613 plankton: Potential synergistic effects of contaminants reduce survival and feeding of the copepod
614 *Acartia tonsa*. *Marine Environmental Research* **2017**, *129*, 374-385.
- 615 21. Kjølholt, J.; Aakre, S.; Jürgensen, C.; Lauridsen, J., Assessment of possible impacts of scrubber
616 water discharges on the marine environment. Report of The Danish Environmental Protection Agency.
617 **2012**.
- 618 22. Ytreberg, E.; Hassellöv, I.-M.; Nylund, A. T.; Hedblom, M.; Al-Handal, A. Y.; Wulff, A., Effects of
619 scrubber washwater discharge on microplankton in the Baltic Sea. *Marine Pollution Bulletin* **2019**, *145*,
620 316-324.
- 621 23. Hufnagl, M.; Liebezeit, G.; Behrends, B., Effects of Sea Water Scrubbing. Final report. **2005**.
- 622 24. Buhaug, O.; Fløgstad, H.; Bakke, T., MARULS WP3: Washwater Criteria for seawater exhaust gas-
623 SOx scrubbers. MARINTEC REPORT. **2006**.
- 624 25. USEPA, Exhaust Gas Scrubber Washwater Effluent. Regulatory Document. US Environmental
625 Protection Agency. **2011**.
- 626 26. Lange, B.; Markus, T.; Helfst, L., Impacts of scrubbers on the environmental situation in ports and
627 coastal waters. *Dessau-Roßlau* **2015**.
- 628 27. den Boer, E.; 't Hoen, M., Scrubbers – An economic and ecological assessment. *Delft, CE Delft*
629 **2015**.
- 630 28. Belgian Government, Wet oppervlaktewateren. 26 maart 1971 - Wet op de bescherming van de
631 oppervlaktewateren tegen verontreiniging. **1971**.
- 632 29. ECSA, European Community Shipowners' Association Open letter to EU Member States and the
633 European Commission. 18/06/2014. Implementation of the EU Sulphur Directive must be harmonised
634 and realistic. **2014**.
- 635 30. Gotze, H.; S., N.; E., U., Onboard Measurements Of Diesel Engine Exhaust Gas Components.
636 *Transactions on the Built Environment* **1997**, *24*.
- 637 31. Hofmann, A. F.; Soetaert, K.; Middelburg, J. J.; Meysman, F. J. R., AquaEnv: An Aquatic Acid-Base
638 Modelling Environment in R. *Aquatic Geochemistry* **2010**, *16*, (4), 507-546.
- 639 32. Soetaert, K.; Petzoldt, T.; Setzer, R. W., Solving Differential Equations in R: Package deSolve.
640 *Journal of Statistical Software* **2010**, *33*, (9), 1-25.
- 641 33. Agrawal, H.; Eden, R.; Zhang, X.; Fine, P. M.; Katzenstein, A.; Miller, J. W.; Ospital, J.; Teffera, S.;
642 Cocker, D. R., Primary Particulate Matter from Ocean-Going Engines in the Southern California Air Basin.
643 *Environmental Science & Technology* **2009**, *43*, (14), 5398-5402.
- 644 34. Winnes, H.; Moldanova, J.; Anderson, M.; Fridell, E., On-board measurements of particle
645 emissions from marine engines using fuels with different sulphur content. *Proceedings of the Institution*
646 *of Mechanical Engineers Part M-Journal of Engineering for the Maritime Environment* **2016**, *230*, (1), 45-
647 54.
- 648 35. Celo, V.; Dabek-Zlotorzynska, E.; McCurdy, M., Chemical Characterization of Exhaust Emissions
649 from Selected Canadian Marine Vessels: The Case of Trace Metals and Lanthanoids. *Environmental*
650 *Science & Technology* **2015**, *49*, (8), 5220-5226.
- 651 36. Moldanová, J.; Fridell, E.; Winnes, H.; Holmin-Fridell, S.; Boman, J.; Jedynska, A.; Tishkova, V.;
652 Demirdjian, B.; Joulie, S.; Bladt, H.; Ivleva, N. P.; Niessner, R., Physical and chemical characterisation of
653 PM emissions from two ships operating in European Emission Control Areas. *Atmos. Meas. Tech.* **2013**, *6*,
654 (12), 3577-3596.
- 655 37. Contini, D.; Gambaro, A.; Belosi, F.; De Pieri, S.; Cairns, W. R. L.; Donato, A.; Zanutto, E.; Citron,
656 M., The direct influence of ship traffic on atmospheric PM2.5, PM10 and PAH in Venice. *Journal of*
657 *Environmental Management* **2011**, *92*, (9), 2119-2129.

- 658 38. Gauthier, P. T.; Norwood, W. P.; Prepas, E. E.; Pyle, G. G., Metal–PAH mixtures in the aquatic
659 environment: A review of co-toxic mechanisms leading to more-than-additive outcomes. *Aquatic*
660 *Toxicology* **2014**, *154*, 253-269.
- 661 39. EC, Water Framework Directive on priority substances 2013/39/EU. **2013**.
- 662 40. Lindstad, H. E.; Eskeland, G. S., Environmental regulations in shipping: Policies leaning towards
663 globalization of scrubbers deserve scrutiny. *Transportation Research Part D: Transport and Environment*
664 **2016**, *47*, 67-76.
- 665 41. Turner, D. R.; Edman, M.; Gallego-Urrea, J. A.; Claremar, B.; Hassellöv, I.-M.; Omstedt, A.;
666 Rutgersson, A. J. A., The potential future contribution of shipping to acidification of the Baltic Sea. **2018**,
667 *47*, (3), 368-378.
- 668 42. Doney, S. C.; Mahowald, N.; Lima, I.; Feely, R. A.; Mackenzie, F. T.; Lamarque, J.-F.; Rasch, P. J.,
669 Impact of anthropogenic atmospheric nitrogen and sulfur deposition on ocean acidification and the
670 inorganic carbon system. **2007**, *104*, (37), 14580-14585.
- 671 43. Lee, K.; Tong, L. T.; Millero, F. J.; Sabine, C. L.; Dickson, A. G.; Goyet, C.; Park, G. H.; Wanninkhof,
672 R.; Feely, R. A.; Key, R. M., Global relationships of total alkalinity with salinity and temperature in surface
673 waters of the world's oceans. *Geophysical Research Letters* **2006**, *33*, (19).
- 674 44. Doney, S. C.; Fabry, V. J.; Feely, R. A.; Kleypas, J. A., Ocean Acidification: The Other CO₂ Problem.
675 *Annual Review of Marine Science* **2009**, *1*, 169-192.

676