

Molecular Bonding Regulated Defect Passivation for Modulating Efficiency and Stability of Methylammonium Free Inverted Perovskite Solar Cells

yasuhiro Shirai NIMS Masatoshi Yanagida National Institute for Materials Science Andrey Lyalin Hokkaido University https://orcid.org/0000-0001-6589-0006 Tetsuya Taketsugu Hokkaido University https://orcid.org/0000-0002-1337-6694 Kenjiro Miyano NIMS DHRUBA KHADKA (Khadka.b.dhruba@nims.go.jp) National Institute for Materials Science

Article

Keywords: MA-free, bifunctional molecule, piperazine dihydriodide, surface passivation, defect, device stability

Posted Date: July 18th, 2023

DOI: https://doi.org/10.21203/rs.3.rs-3025256/v1

License: (c) This work is licensed under a Creative Commons Attribution 4.0 International License. Read Full License

Additional Declarations: There is NO Competing Interest.

Version of Record: A version of this preprint was published at Nature Communications on January 29th, 2024. See the published version at https://doi.org/10.1038/s41467-024-45228-9.

Abstract

Molecular passivation is a prominent approach for improving the power conversion efficiency and operation stability of perovskite solar cells. Herein, diammonium iodide functional molecules with an aryl or alkyl core are introduced on 3D-perovskite, and its interfacial passivation effect is explored. It showed that piperazine dihydriodide (PZDI) with alkyl core-electron rich -NH terminal is propitious to mitigate surface and bulk defects and modify surface chemistry or interfacial energy band leading to increase carrier extraction. Benefiting from superior PZDI passivation, the device efficiency has been scaled to 23.17% (area ~1 cm²) with superior operational device stability. We also achieved a certified efficiency of ~21.50% (area ~1.024 cm²). The theoretical calculation suggests that PZDI entangles onto the film's surface with -NH₂I anchor, and reinforces the adhesion. Device analysis corroborates that a stronger bonding interaction attenuates the defect densities in the perovskite film and suppresses ion migration, which is supported by the first-principle calculations. This work demonstrated that the bifunctional molecules with stronger surface adsorption play a crucial role in triggering defect mitigation, which paves the way for the design of bonding-regulated molecular passivation for enhancing device performance and stability.

Introduction

Exceptional optoelectronic properties of halide perovskite (HP) have hiked the power conversion efficiency (PCE) of halide perovskite solar cells (HPSCs) by over 25%, approaching to Shockley–Queisser limit.¹ Deluge of experimental efforts on stoichiometric engineering, crystallinity improvement, interface passivation, and carrier transport engineering have been used in the course of rapid progress.^{2,3} However, HP films are prone to degradation under external factors (such as thermal/humidity stress, oxygen, and light) and intrinsic phenomena.^{4,5} These deleterious characteristics have exerted big challenges to practicality. Indeed, the HPSC degradation stems from deleterious surface chemistries on the surface of HP film or at the device interface.⁶ To address these detrimental defect chemistries, molecular passivators have been of great interest in improving the PCE and device stability.^{7,8}

Several reports have documented the multifunctional molecular passivation approach to mitigate the charge defects on perovskite surface or bulk comprising under-coordinated Pb²⁺ ions, A-site vacancies, and halide vacancies.^{9–11} The molecular passivator with multifunctional derivatives consisting of amine,^{12,13} Lewis acids/bases,^{14,15} supramolecules,^{16–18} ionic polymer,^{8,19} etc have been used for interfacial passivation of HPSCs. The ammonium comprising functional additives with alkyl or aryl halide or pseudohalide counterparts have demonstrated a significant enhancement in PCE and operational stability.^{20–27} The additive having stronger adsorption on the HP surface modulates the grain nucleation and growth or alleviates defect chemistry at interface and bulk which is crucial for a highly efficient and stable device.^{11,28} The function additives also remove molecular iodine existing in perovskites and quench the iodine and the Pb-related deep trap sites.^{29,30} Despite lots of additives being wielded for optimization of perovskite growth and defect passivation, the influence of bonding interaction

between additives to perovskite is rarely explored. Thus, it is paramount to get a characteristic insight into the effect of molecular interaction of additives for quenching detrimental defect dynamics induced by the soft ionic nature of HP.

In this work, we report on passivation strategy regulating through bonding defect passivators by introducing bifunctional molecules with an aryl core (1,4-phenylenediamine dihydrodide (PEDAI)) or alkyl core (piperazine dihydriodide (PZDI)) onto the 3D-HP film of the precursor formulation of $FA_{0.84}Cs_{0.12}Rb_{0.04}Pbl_3$ (methylammonium/bromine (MA/Br)-free HP film). Density functional theory (DFT) calculations show that the alkyl core amine has stronger bonding interaction with uncoordinated Pb^{2+} and iodine ion traps. It is found that the bifunctional molecular passivation significantly affects the film morphology and surface chemistries by virtue of their basicity and adsorption energy. Thus, the PZDI additive effectively quenches the surface or bulk defects in the HP film resulting in a longer carrier lifetime and better interface quality. Consequently, the HPSCs with PZDI treatment resulted in an enhanced large area (> 1cm²) device performance from 19.68 to 23.17% with an increase in the device parameters and reduced V_{OC} deficit of 0.328 V. Interestingly, although HPSC with PEDAI treatment lowers down the PCE to 19.21%, the operational device stability is superior to the control device under thermal and moisture stress. The work gets insight into the device characteristics with synergetic effect in the surface chemistry, elemental distribution, photophysics, and defect profile.

Results and discussion

The chemical structures of diammonium iodide molecule (DIM) with aryl and alkyl chains used as a bonding-regulated molecular passivator are depicted in Fig. 1a. These molecules in isopropanol were introduced on the 3D-HP film as an interfacial passivation layer (IPL) as shown in schematics (Fig. 1b, c). The surface morphology of MA/Br-free HP film (treated without and with PEDAI or PZDI) was studied with scanning electron microscope (SEM) measurement (Fig. 1d-f). One can notice a slight change in the grain size with an overlayer surface grown on the film. The HP film with PEDAI passivation grows with unevenly distributed small crystallite. It suggests that the DIM with an alkyl core has beneficial surface coverage compared to the aryl core. The film with PZDI treatment forms well-covered surface and grain boundaries which is propitious for the elimination of localized defects in perovskite film.^{7,11}

We collected X-ray diffraction (XRD) results of control and surface-treated perovskite films to study the crystal growth. (Fig. 1g, S1a,b) shows XRD of the PEDAI or PZDI-treated HP films with reference to the control film. The PZDI-treated HP films show a dominant (110) characteristic diffraction peak of the α -phase of FA- HP. It suppresses the δ -Cs/RbPbl₃ phase and residual Pbl₂. Moreover, additional peaks at 20 < 10° are dominantly grown on the HP films with a higher content of DIM indicating the evolution of a 2D phase of PEDAI and PZDI interacting with Pbl₂.³¹ The characteristic XRD peak at lower 20 on the HP film with PEDAI treatment grows with higher intensity indication higher tendency for the formation of 2D phase compared to alkyl counterpart.

Figure 1h depicts the absorption spectra of the control and DIM-treated HP films (varying concentration; Fig. S1c, d). No significant difference was seen at the band edge of the HP film (insets in Fig. 1h, Fig. 1a, c). The PL spectra (Fig. 1i) of HP with PZDI demonstrated an intensified peak while PEDAI suppressed it. The varying PL intensity of HP films treated with different concentrations of DIM (Fig. S2) correlates with improvement in opto-physical quality. One can see a negligible shifting of PL characteristics peak (~ 817–818 nm) suggesting no incorporation of these additives in the 3D-HP lattice. It is also consistent with XRD patterns.

To get more insight into the 2D-phase formation, we prepared HP film by mixing DIM in perovskite precursor solution as depicted in supporting information (Figure S3a₁, b₁). XRD patterns (Figure S3a₂, b₂) demonstrated a more intensified XRD peak at $2\theta < 10^{\circ}$ for the HP film prepared with mixed DIM compared to that with surface treatment. Importantly, the HP film with mixed PEDAI was found to be grown with much higher XRD peak intensity at ~ 4.8° indicating more preference for 2D phase formation. SEM images (Figure S3a₃, b₃) display nano-sheet-like features in the HP mixed with PEDAI implicates PEDAIbased 2D phase. This is consistent with the small crystallite observed in PEDAI passivated film. Interestingly, unlike PEDAI mixed case, the HP film with mixed PZDI grows with overlayer on crystal grain or at grain boundaries rather than flake structure which is in line with the passivated film. The PL spectra of respective HP films (Figure S3c) also showed characteristic PL peaks at higher energy regimes suggesting the existence of DIM-based 2D phases. The PL characteristic features of respective films are parallel to XRD patterns. These results suggest the HP film with PEDAI can preferably form 2D phases rather than forming a well-covered 3D-HP surface. It is believed that these characteristics differences could dominantly affect the device performance.

To investigate the effect of surface treatment using DIM with anyl or alkyl core on the photovoltaic properties, we fabricated HPSCs with inverted device configuration as depicted in Fig. 2a. A typical crosssectional image of a complete device is displayed in Fig. 2b. The current density-voltage (J-V) curves for the best HPSCs without and with PEDAI or PZDI treatment are shown in Fig. 2c with a large device area of \approx 1 cm². The J-V characteristics with varying concentrations of DIM are given in the supporting information (Figure S4). The device parameters have been summarized in Table 1(Table S1, S2, Supporting Information). The HPSCs with PZDI-treated HP (1 mg/ml) demonstrated champion PCE of ~ 23.17% with negligible J-V hysteresis. However, unlike our expectation, the PCE of the PEDAI passivated device rolls off with reduced J_{SC} and FF with reference to the control device as given in Table 1. A decrease of J_{sc} and an increase of V_{oc} (both small) seem to be typical behavior of slight passivation of interface defects. It suggests that PEDAI has an interface passivation effect to some extent. These results corroborate that the amine in DIM with any and alkyl core greatly affects the device results. It is a consequence of the chemical interaction of the HP with DIM that drives the film guality impacting morphology, surface chemistry, and defect profile. It is known that PEDAI has delocalized lone pair of electrons of nitrogen atoms while that in PZDI is localized. The nitrogen site in DIM enhances the surface adhesion on the HP film which is favorable for defect mitigation. This could result in a stark

characteristic difference in HPSCs. We will discuss more insight in the succeeding paragraph accounting for these aspects.

We validated HPSCs with MA/Br-free HP film with PZDI treatment of PCE \approx 21.47% (area \approx 1.024 cm²) under standard conditions (accredited independent photovoltaic test laboratory, AIST PV Lab, Japan). The official certified data is given in Figure S8 (Supporting Information). Our certified PCE of the champion device has a record-level device efficiency for inverted p-i-n configuration of HPSCs with MA/Br-free HP for a large area of > 1 cm². For comparative evaluation, the few certified device reports with an area > 1 cm² are tabulated (Table S3, Supporting Information), where the champion HPSC compares favorably amongst the reports.

Table 1 Solar cell data and V_{OC} deficit of the HPSCs with MA/Br-free perovskite (without and							
with surface treatment (PEDAI or PZDI)). E_g^a calculated from EQE data.							
Condition	a		J _{SC}	V _{OC} FF	PCE	V _{OC} deficit	
	E_{g}		(mAcm ⁻²)	<i>(V)</i>		(%)	a
	(eV)						(E _g / q -V _{OC}) (V)
Control	1.513	F	23.62	1.104	0.717	18.69	0.402
		R	23.56	1.112	0.750	19.65	
PEDAI	1.517	F	22.98	1.142	0.703	18.46	0.381
		R	22.79	1.145	0.736	19.21	
PZDI	1.515	F	24.76	1.185	0.772	22.65	0.327
		R	24.78	1.188	0.787	23.17	

Figure 2d presents the PCE statistics of the control device and with PEDAI or PZDI treatments (Figure S5,6,7 and Table 1,2; Supporting Information). These results indicate better device reproducibility with respective DIM passivation with varying concentrations. For PZDI, the device performance is found to be improved with an increase in all device parameters. However, the device with a higher concentration of PZDI rolls off with a lower V_{OC} and *FF*. Although the HPSCs with PEDAI treatment are inferior to the control device, the device with a higher concentration of PEDAI further deteriorates by dropping V_{OC} and FF as like the PZDI case. This identical trend could be due to the accumulation of 2D HP phase on the surface unevenly. This observation is in line with other reports.^{18,31}

The external quantum efficiency (EQE) data for HPSCs without and with DIM passivation are shown in Fig. 2e. The EQE response for the device with PZDAI presents a better spectral response in the absorption energy band range of the HP layer ($\lambda > 450$ nm) and interfacial regime ($450 > \lambda > 330$ nm).³² It is attributed to the betterment in the bulk and interface quality of PZDI-treated HPSC. Note that the integrated current

values from EQE spectra are 23.04, 22.68, and 24.08 mA/cm² for the control and PEDAI or PZDAI-treated HPSCs, which are in the range of the J_{SC} of respective devices. We also calculated the bandgap (E_g) of the HP absorber layer from EQE analysis ($E_g^a \sim 1.513$, 1.516, and 1.515 eV for the control, PEDAI, and PZDI; Figure S9a-c, Supporting Information) are in close agreement with the E_g obtained from the absorption spectra (Figure S9d-f, Supporting Information) and PL spectra (Fig. 1i).

To explore the characteristic insight, we analyzed the photo response of the HPSCs. Figure 2f presents the V_{OC} variation with logarithmic of light intensity (ln(l)). The slopes are estimated to be 1.40, 1.34, and 1.16 k_BTq⁻¹ for the control, PEDA, and PZDI-passivated devices, respectively. A device with a higher slope signifies more charge recombination at open circuit conditions. It suggests that the HPSCs with PZDI experience reduced trap-assisted recombination which ameliorates the device performance. We recorded the TPV response as displayed in Fig. 2g by triggering V_{OC} with transient photo illumination. The TPV decay signal analysis reveals a longer carrier lifetime for the PZDI-treated device (12.34 µs). While the device with PEDAI (7.62 µs) shows a slight increase in carrier lifetime compared to the control device (6.22 µs). It suggests that the PZDI treatment passivates the defect in the HP film with propitious surface chemistry and mitigating defect.

To understand the carrier lifetime, we measured the time-resolved photoluminescence (TRPL) spectra and fitted them with bi-exponential decay equation; ${}^{31}I(t) = A_0 + A_1 e^{-\frac{(t-t_0)}{\tau_1}} + A_2 e^{-\frac{(t-t_0)}{\tau_2}}$, where A_0 is a constant for the baseline offset, A_1 and A_2 are the relative amplitude. The decay time, τ_1 and τ_2 accounts for the nonradiative recombination at the interface and radiative recombination at the bulk layer.³³ The HP film with PZDI treatment shows a significantly longer lifetime ($\tau_1 \approx 331$ ns and $\tau_2 \approx 2285$ ns) compared to that of the HP with PEDAI ($\tau_1 \approx 85$ ns and $\tau_2 \approx 783$ ns) or control ($\tau_1 \approx 98$ ns and $\tau_2 \approx 678$ ns). Interestingly, the PEDAI-treated film shows only a small difference in carrier lifetime compared to the control film. It corroborates that PEDAI is not as effective as PZDI for the attenuation of a deleterious defect in HP film. These results indicate the surface treatment with PZDI is propitious for defect passivation due to stronger localized nitrogen bonding in HP film and hence leads to the superiority of device performance.

To understand surface energy, we measured ultraviolet photoelectron spectroscopy (UPS). Figure 3a, b demonstrates the cutoff energy corresponding to work function (φ) and the onset energy (E_i) calculated from the UPS results. The band structure has been constructed by combining with optical bandgap and UPS result (Figure S10). The values of φ and E_i are found to be slightly increased with DIM treatment. The results demonstrate a downshift of E_V (by 0.287 or 0.278 eV) and E_C (0.283 or 276 eV) levels for PEDAI or PZDI-treated film. It modulates the band alignment with NiOx/MeO-2PACz and C₆₀.^{13,22} Indeed, the interfacial band alignment is beneficial for effective carrier transport resulting in better device performance. Although the PZDI or PEDAI shows a similar effect in surface energy, there is a significant improvement in device parameters for the device with PZDI treatment. It suggests that the surface energy modification by PEDAI has only a minimal effect on device performance.

To explore surface chemistry, we carried out X-ray photoelectron spectroscopy (XPS) measurements. In the C 1s XPS core (Fig. 3c), the binding energies centered at ~ 284.4, 286.2, and 287.4 eV are assigned to C-C/C = C, C-N-C, and N = C-N, respectively. The C-N-C characteristic peak for PZDI treated film indicates its dominant interaction with the HP surface. The N 1s XPS cores for the corresponding film (Fig. 3d) also indicate the respective chemical binding characteristics. The surface passivated film shows a small shift of XPS characteristic core of Pb 4f and I 3d (~ 0.11 and 0.18 eV for PEDAI and 0.16 and 0.29 eV for PZDI; respectively, Figure S11) towards higher binding energy. It indicates a stronger ionic bonding induced on the film surface with PZDI treatment. The Cs 3d and Rb 3d core levels demonstrate almost similar spectral features indicating only a weak interaction with an alien molecule. This surface analysis implicates that the bifunctional surface passivator establishes stronger interaction with nitrogen bonding to uncoordinated Pb²⁺ or lodine antistites.^{31,34} The functionality of the PZDAI molecule is superior due to higher electron density in the vicinity of the N-atom compared to PEDAI with an aryl core.

To explore the spatial distributions of molecular passivator, time-of-flight secondary ion mass spectrometry (ToF-SIMS) was used to track the ionic distribution of PEDA⁺ and PZD⁺ distribution in the HP. The characteristic ionic species are shown in Fig. 3e-g. It shows an identical ionic distribution in the perovskite bulk. The characteristic signals from PEDAI or PZDAI are found to be significantly higher on the surface with a deep gradient to the bulk (Figure S12). The 3D maps (Fig. 3h-j) demonstrate that the PEDA⁺ and PZD⁺ cations introduced by surface treatment are mainly distributed on the HP top surface. Importantly, the PZD⁺ additive shows uniform surface coverage with negligible bulk diffusion. In contrast, the PEDA⁺ ions are found to have uneven distribution on the top surface with enriched bulk diffusion compared to the PZDI case. These results are analogous to the surface feature of SEM images of respective films. This observation corroborates that the PZDI or PEDAI mainly passivates the surface defect density by virtue of its functional characteristics to modify the surface chemistry of the pristine perovskite layer.

To get insight into the defect densities, we investigate admittance spectroscopy of HPSCs with surface treatment. Mott-Schottky (*M-S*) plot and carrier profile (N_{CV}) were extracted from capacitance-voltage (C-V) data.³⁵⁻³⁸ Fig. 4a exhibits the *M-S* plots fully depleted curves for *V*>diffusion potential (V_D) suggesting intrinsic characteristic junction. One can see a slight hysteresis in the *M-S* curve near to V_D for the control device which almost disappeared for surface-passivated device. It is attributed to the reduction in ionic polarization at the interface. The V_D value for the PSC with PZDI (1.148 V) is greater than PEDAI (1.065 V) and control device (0.991 V) which is parallel to the V_{OC} of the respective device. The carrier profile (N_{CV}) (Fig. 4b) extracted from *C-V* data analysis comprises the free carrier and defect density. It showed a carrier distribution in a bulk (N_{CV}^B) in the range of ~3.46–6.94 ×10¹⁵ cm⁻³. The N_{CV}^B with PZDI is slightly lower by some fraction. The carrier profile at the edge accounts for the interface defect density profile (N_{CV}^{IF} ~15.26×10¹⁷, ~11.56×10¹⁷, and ~2.43×10¹⁷ cm⁻³ for control, PEDAI, and PZDI treated devices, respectively). The interface defect density is suppressed by 6 times for the PZDI-

passivated device. It corroborates that the PZDI effectively attenuates the recombination centers leading to the improvement in the device parameter.

For the quantitative analysis of the defect profile, we investigated thermal admittance spectroscopy, an effective technique for estimating optoelectronic properties; the defect level and defect density to thin film solar cells (HPSCs^{39,40}, chalcogenide solar cells⁴¹, and organic solar cells⁴²). Figure 4c shows the capacitance-frequency (*C-f*) spectra measured at room temperature (under dark). All device reveals a plateau regime (1 to 100 kHz) with a slightly lower value for PZDI treated device that could stem from the HP accounting for defect density. Besides that, the lower frequency capacitance response is much steeper for the control device which is attributed to the interfacial charge accumulation or ionic polarization. It implicates a suppression of interfacial charge accumulation for the HPSCs with PZDI treatment.

Furthermore, we measured the temperature-dependent capacitance-frequency (C-f-T) spectra (Figure S12) to analyze the defect density profile. The trap state (E_t) is calculated from the Arrhenius plot⁴³ by analyzing the resonance frequency (ω_o) obtained from the C-f-T analysis as given in Figure S12. The Arrhenius plots (Fig. 4d-f) revealed shallower defect states in the PZDI-treated device ($E_{t3}, E'_{t3} \sim 0.154, 0.374 \text{eV}$) compared to the PEDAI-treated ($E_{t2}, E'_{t2} \sim 0.212, 0.408 \text{eV}$) or a control device ($E'_{t1}, E'_{t2} \sim 0.241, 0.423 \text{eV}$). We calculated the defect density profiles (Figure. 4g-i) using the equation, ${}^{43,44}N_t$ (E_ω)= $-\frac{V_D}{qW}\left(\frac{\omega}{k_BT}\frac{dC}{d\omega}\right)$, where, V_D , W, q, and ω denote the diffusion potential, the space charge region width, elementary charge, and applied frequency, respectively.

We found that the integrated trap densities for the control device $(N_{t1}, N'_{t1} \sim 1.08 \times 10^{17}, 9.89 \times 10^{16} \text{ cm}^{-3})$ are attenuated for the PEDAI $(N_{t2}, N'_{t2} \sim 7.38 \times 10^{16}, 8.49 \times 10^{16} \text{ cm}^{-3})$ or PZDI $(N_{t3}, N'_{t3} \sim 3.22 \times 10^{16}, 4.03 \times 10^{16} \text{ cm}^{-3})$ treated devices. These results are in the range of reported trap densities for the perovskite film $(10^{16} \cdot 10^{19} \text{ cm}^{-3})^{13.45}$. On other the hand, the defect densities in our devices are more than 10^{6} than a single crystal $(10^{10} \text{ cm}^{-3})$ which call for more effort to lower the trap densities to achieve superior film quality. From the comparative analysis, the trap densities (N'_{t1}) primarily assigned for defects in the bulk are decreased by ~ 2.5 times in the device with PZDI treatment or 1.65 times in the device with PEDAI treatment indicating the improved bulk quality of HP film. The shallower trap state profile (N_{t1}) is assumed to have defects at the surface or GBs in the HP film. These shallower defect densities are found to be significantly lowered in the PZDI or PEDAI-treated devices. These results consolidate that the PEDAI or PZDI stays on the film surface or diffuses into the bulk through the GBs to passivate the defect states due to molecular interaction with the characteristics of nitrogen terminals. We found that the PZDI passivation is rather efficient for mitigating the defect chemistries at the surface, GBs, and bulk compared to PEDAI. Thus, the capacitance spectra analysis well agrees with the advantageous properties of HP film induced with surface treatment as discussed in previous paragraphs.

To clarify the effects of the DIM passivator on HP film, we have performed the first-principles calculations based on density functional theory (DFT). The full details on computational and theoretical methods are

given in Supporting Information. The pseudo-cubic structure of FAPbl₃ was used as a model of the bulk structure (Figure S17). The perovskite's surface was modeled by a 2x2 slab of Pbl₂-terminated surface (001) with five PbI₂ layers and a vacuum region of ~ 25 Å (Figure S18a). The corresponding total density of electronic states (DOS) calculated for the defect-free Pbl₂ terminated surface is shown in Figure S18b. In the case of full surface coverage, two PEDAI (PZDI) molecules (Figures S19 and S20) can be accommodated with I atoms of PEDAI (PZDI) adsorbing on top of Pb atoms of the topmost PbI₂ surface layer as shown in Figures S21a, b. Both molecules are tilted with N atoms forming a plane parallel to the perovskite surface. Our calculations demonstrate that PEDAI molecules prefer to form a chain along the (100) direction, while PZDI molecules form a chain along the (010) direction. This could be correlated to the different distribution tendencies of DIM in the HP film as seen in ToF-SIMS results (Fig. 3h-j). Interaction of PEDAI and PZDI with the surface results in a slight distortion of the surface PbI₂ layers and rotation of the FA molecules in the first and second Pbl₂-FAI bilayers, which is especially noticeable for the PEDAI@FAPbI₃ case (Figure S21a). As one can see adsorption of PEDAI or PZDI molecules does not introduce any defect states in the forbidden zone of perovskite (DOS for PEDAI and PZDI; Figures S21c, d), slightly increasing the band gap from 1.53 eV calculated for the pure FAPbl₃, up to 1.60 eV (1.59 eV) for the perovskite covered by PEDAI (PZDI) which is analogous to slightly higher surface band energy obtained from UPS analysis (Figure S10).

Figure 5 shows the role of PEDAI/PZDI passivation of the defected (IPb antisite) PbI2-terminated surface of FAPbl₃ defined in the previous studies.^{18,46} Indeed, the I_{Pb} antisite defect results in the formation of an unoccupied defect state 0.1 eV above the Fermi level as well as some defect states in the middle of the forbidden zone (0.4–1.0 eV above the Fermi level), as it is seen from the analysis of the total DOS of the perovskite surface with I_{Pb} defect, presented in Figs. 5d,e by black lines. The detrimental defect states can be effectively passivated with PZDI treatment. Thus, in the case of PZDI passivation, the density of the defect states is considerably reduced, slightly shifting down the bottom of the conductivity zone by 0.06 eV toward the Fermi level (Fig. 5e). On the other hand, the passivation of the IPb defect with PEDAI, does not eliminate the low-lying defect state at 0.1 eV above the Fermi, as well as introduces the narrow defect state at 1.66 eV, with the edge of the conductivity band shifted to 1.78 eV. Therefore, from the analysis of the electronic structure of the passivated surface with IPb antisite defect, one can suggest that PZDI passivation should lead to considerably better solar-cell performance. This observation is in line with the calculated defect profile (Fig. 4). Since PZDI or PEDAI, both contain cations and anions, these could also interact with other charge defects. From the analysis of the Mulliken charges (Figures S19, S20), it is found that PZDAI, the charge distribution in the tail is strongly polarized, with both of I in the -NH₂I anchor possessing an excess of the negative charge of -0.68|e|, while the NH₂ counterpart is positively charged, with the net charge of + 0.34|e|. Here, |e| is an elementary charge. On the other hand, in the case of the more stable trans-isomer the PEDAI molecule one of the -NH₃I anchor is polarized with the Mulliken charge on I equal to -0.64|e| and charge on the NH₃ counterpart of +0.49|e|, while another -NH₃I anchor is overall almost neutral with a little polarization. In the case of the cis-isomer of PEDAI both of the -NH₃I anchors are overall almost neutral with a little polarization of charges between I and N. This feature

explains the difference in the interaction of PEDAI and PZDI molecules with the perovskite surface and their ability to quench the defects. Theoretical analysis of the change in Gibbs free energy upon adsorption demonstrates that PZDI molecules bind considerably stronger to the surface with I_{Pb} antisite defect in comparison with PEDAI, with the binding energy 1.54 eV per molecule (1.32 eV for PEDAI). Thus, our theoretical analysis corroborates that PZDI passivates the defective surface with a stronger quenching tendency forming the stable film covering the surface.

Despite continuous breakthroughs in device efficiency, the stability of HPSCs is still a stumbling block to their competitive reliability in practical applications. To evaluate the device stability, we tracked the device parameters of HPSCs (encapsulated) at the maximum power point tracking (MPPT) conditions under 1 sun irradiation under heat, light, or humidity stress. As shown in (Fig. 6a, b), the device with PZDI treatment demonstrated superior operational device stability under respective monitoring conditions. Interestingly, despite the improvidence in improving device performance, the PEDAI-treated device showed better device stability compared to the control device suggesting the beneficial effect of surface passivation. At elevated temperature (~ 60 ± 5 °C; ~ 35-40% RH), the performance of the control device dropped to ~ 57.82% of initial PCE in 1000 hours which significantly lowered to ~ 37.80% in 200 hours under heat and moisture stress (T=35 \pm 5 °C; RH~ 60–65%). Similarly, HPSCs with PZDI treatment retained ~ 89.48% and 86.74% of original PCE under respective aging conditions. While the PEDAI treated device demonstrated comparatively better device stability than the control device retaining 74.32% and 72.67% of the original PCE under respective aging conditions. These data corroborate that the surface treatment with DIM multifunctional molecules significantly improves the device stability under thermal and humidity stress as a consequence of propitious surface chemistry and interfacial surface modulation with strong adsorption energy.¹⁷ This observation indicates that the superior device stability stems from better interface quality and moisture stability in the surface-treated HPSCs.

To consolidate the superior moisture stability data, the water contact angles were taken to study the hydrophobicity of respective HP films (Figure S6c-e). We noticed a significant drop in water contact angle from 66.42° ($t_o \sim 0$ s) to 42.36° ($t \sim 1$ min) for the control film. For the PEDAI-treated HP film, it shows a higher contact angle of 84.50° ($t_o \sim 0$ s) which retains at 78.40° after one minute. Similarly, the PZDI-treated HP film demonstrates a contact angle of 90.60° ($t_o \sim 0$ s) to 86.20° ($t \sim 1$ min). It corroborates that the HP films with surface treatment result in excellent moisture tolerance. The moisture resistivity of the passivated film is attributed to its dense distribution on the film surface (Fig. 3h-j), which agrees with the trend of device stability under higher humidity stress.

Moreover, to contemplate the interfacial deterioration under aging conditions, the capacitance-voltage curves of aged devices were measured. A more pronounced C-V hysteresis was seen for the control HPSC (Fig. 6f-h) suggesting a deteriorated interface compared to the HPSCs with surface passivation. This observation substantiates that the control HPSC degrades due to the corrosion of the interfacial junction and increasing dominance of accumulated ions at the interface.⁴⁷ A sharp transition of the *M*–*S* curve in the device with PZDI treatment indicates a smaller depletion layer capacitance (C_{dl}) that is attributed to

low interfacial defect density. It retains a more stable interfacial junction that stems from intact bulk capacitance (C_g). ^{48,49} One can see a plateau capacitance (C_s) region for $V > V_D$ which is correlated to interfacial charge accumulation and electrode polarization. A suppressed *C-V* hysteresis in PZDI-treated HPSC signifies the suppression of ionic motion or interfacial charge accumulation analysis.⁶ Interestingly, the M-S characteristic features for the aged PEDAI device are not as intact as PZDI treated device. These characteristic disparities indicate that alky amine is rather meticulous for surface passivation as supported by theoretical calculations. Thus, this work corroborates that localized electron density in alkyl amine enhances the interfacial adhesion stabilizing the interface and bulk that is benign for device efficiency and operational stability.

Conclusions

In summary, we introduced a very effective approach to mitigate the defects in MA-free perovskites using bifunctional molecules with amine terminal. It is found that the diammonium iodide with aryl or alkyl core amine has a remarkable effect on both the efficiency and stability of inverted HPSCs. The PEDAI with delocalized lone pair of electrons with an aryl core exerts weak interaction on the surface defects. The PZDI molecules with alkyl core amine are rather propitious for the improvement in film quality inducing a surface chemical environment for quenching the surface and bulk defect due to stronger interaction with localized electron density. Consequently, the PCE of HPSCs with PZDI treatment has been boosted from 19.68–23.17% with a large area of $\geq 1~{\rm cm}^2$ and a reduced $V_{\rm OC}$ deficit of 0.328 V. The molecular distribution of surface-treated film reveals a uniform surface coverage of PZDI with scant diffusion through the bulk layer from ToF-SIMS mapping. Moreover, the PZDI treated-perovskite film exhibited superior operational device stability under heat and humidity stress compared to PEDAI or control devices. This report corroborates that the amine with an alkyl core is crucial for enabling molecular interaction for defect passivation. And hence it paves a new way for designing multifunctional molecular passivator for HPSCs and other optoelectronic applications.

Experimental Section

Materials and precursor solution:

All the reagents and solvents were used as purchased with detailed given in supporting information. The sputtering target for NiO_x deposition (purity > 99.9%) was obtained from Kojundo Chemical Laboratory Co. Ltd, Japan.

Fabrication of MA/Br-free HP; FA $_{0.84}$ Cs $_{0.12}$ Rb $_{0.04}$ PbI $_3$:

MA/Br-free halide perovskite: the precursor solution (1.05 M) was prepared by dissolving FAI, CsI RbI, PbI₂, and 5-AVAI in (0.84, 0.12, 0.04, 1, 0.001) molar ratios in dimethylformamide (DMF) and dimethyl

sulfoxide (DMSO) (4:1) for 2 h at 60 °C. The HP film deposition and surface passivation strategy details are given in supporting information.

Perovskite solar cell Fabrication:

We fabricated the p-i-n device structured HPSC using ITO substrate. The device fabrication was completed in device structure; ITO/ sputtered-NiO_x/MeO-2PACz/MA-Br free-HP film/surface passivation with molecular passivator/C60-evaporation/BCP/Ag-evaporation. Four sets of HPSCs in ITO substrate (4.5 cm \times 3.5 cm) with an active device area of ~ 1.26 x 1.26 cm² were encapsulated using UV-curable resins before the subsequent measurements. A detailed of each layer deposition is mentioned in the supporting information.

Materials and device Characterizations:

The perovskite films were characterized using XRD, SEM, XPS, UPS, PL spectroscopy, fluorescence lifetime spectrometer, UV-vis spectroscopy, TEM, water contact angle, and time-of-flight secondary ion mass spectrometry.

HSPSCs were characterized by measuring current density-voltage (J-V) curves, EQE spectra, TPV, and capacitance spectra.

The details of characterization conditions and instrumentation used in this report are given in the Supporting Information.

Solar Cell certification:

HSPC certification was carried out in The National Institute of Advanced Industrial Science and Technology (AIST) (ISO / IEC 17025 accreditation laboratory), Japan.

Theoretical and computational methods.

We performed density functional theory (DFT) calculations have been performed using the fully constrained meta-generalized-gradient approximation (meta-GGA) SCAN⁵² for the exchange – correlation functional and the projector-augmented wave (PAW) formalism as implemented in the Vienna ab initio simulation package (VASP).^{53,54} A detailed computational procedure is described in the Supporting Information.

Declarations

Conflicts of interest

There are no conflicts to declare.

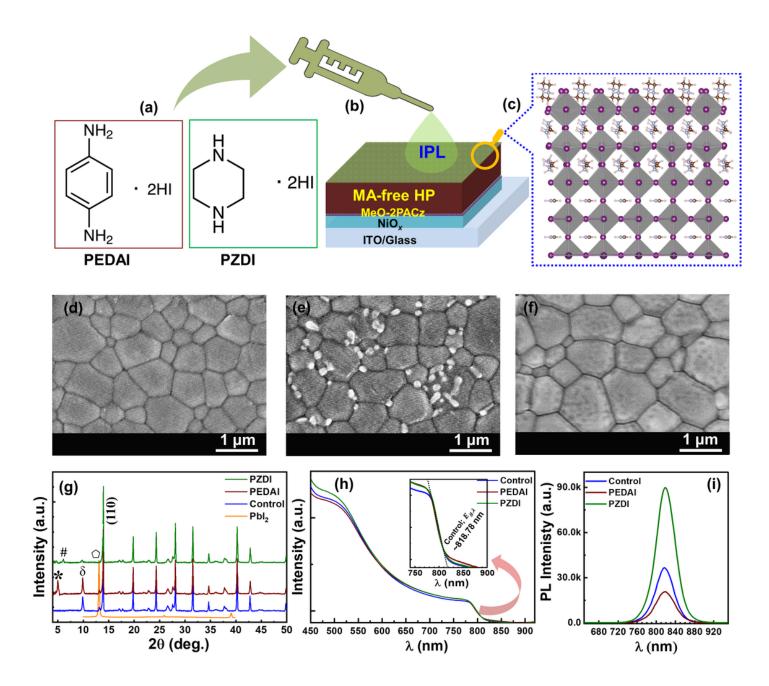
References

- 1. Almora, O. *et al.* Device Performance of Emerging Photovoltaic Materials (Version 3). *Adv. Energy Mater.***13**, 2203313 (2023).
- Seo, J., Song, T., Rasool, S., Park, S. & Kim, J. Y. An Overview of Lead, Tin, and Mixed Tin–Lead-Based ABI 3 Perovskite Solar Cells. *Adv. Energy Sustain. Res.* 2200160 (2023) doi:10.1002/aesr.202200160.
- 3. Kumar Jena, A., Kulkarni, A. & Miyasaka, T. Halide Perovskite Photovoltaics: Background, Status, and Future Prospects. *Chem. Rev.***119**, 3036–3103 (2019).
- 4. Yang, T. *et al.* Achieving Long-Term Operational Stability of Perovskite Solar Cells with a Stabilized Efficiency Exceeding 20% after 1000 h. *Adv. Sci.***6**, 1900528 (2019).
- Khadka, D. B., Shirai, Y., Yanagida, M., Uto, K. & Miyano, K. Analysis of degradation kinetics of halide perovskite solar cells induced by light and heat stress. *Sol. Energy Mater. Sol. Cells*246, 111899 (2022).
- Khadka, D. B., Shirai, Y., Yanagida, M. & Miyano, K. Insights into Accelerated Degradation of Perovskite Solar Cells under Continuous Illumination Driven by Thermal Stress and Interfacial Junction. ACS Appl. Energy Mater. 4, 11121–11132 (2021).
- 7. Liu, Y. *et al.* Ultrahydrophobic 3D/2D fluoroarene bilayer-based water-resistant perovskite solar cells with efficiencies exceeding 22%. *Sci. Adv.***5**, eaaw2543 (2019).
- 8. Bai, S. *et al.* Planar perovskite solar cells with long-term stability using ionic liquid additives. *Nature***571**, 245–250 (2019).
- 9. Motti, S. G. et al. Defect Activity in Metal-Halide Perovskites. Adv. Mater.0, 1901183 (2019).
- 10. Moia, D. & Maier, J. Ion Transport, Defect Chemistry, and the Device Physics of Hybrid Perovskite Solar Cells. *ACS Energy Lett.***6**, 1566–1576 (2021).
- 11. Zhu, J. *et al.* Formamidine disulfide oxidant as a localised electron scavenger for >20% perovskite solar cell modules. *Energy Environ. Sci.***14**, 4903–4914 (2021).
- 12. Wang, F. *et al.* Phenylalkylamine Passivation of Organolead Halide Perovskites Enabling High-Efficiency and Air-Stable Photovoltaic Cells. *Adv. Mater.***28**, 9986–9992 (2016).
- 13. Zheng, X. *et al.* Defect passivation in hybrid perovskite solar cells using quaternary ammonium halide anions and cations. *Nat. Energy***2**, 17102 (2017).
- 14. Bi, D. *et al.* Multifunctional molecular modulators for perovskite solar cells with over 20% efficiency and high operational stability. *Nat. Commun.***9**, 4482 (2018).
- 15. Li, T. *et al.* Multiple functional groups synergistically improve the performance of inverted planar perovskite solar cells. *Nano Energy***82**, 105742 (2021).
- 16. Abate, A. *et al.* Supramolecular Halogen Bond Passivation of Organic–Inorganic Halide Perovskite Solar Cells. *Nano Lett.***14**, 3247–3254 (2014).

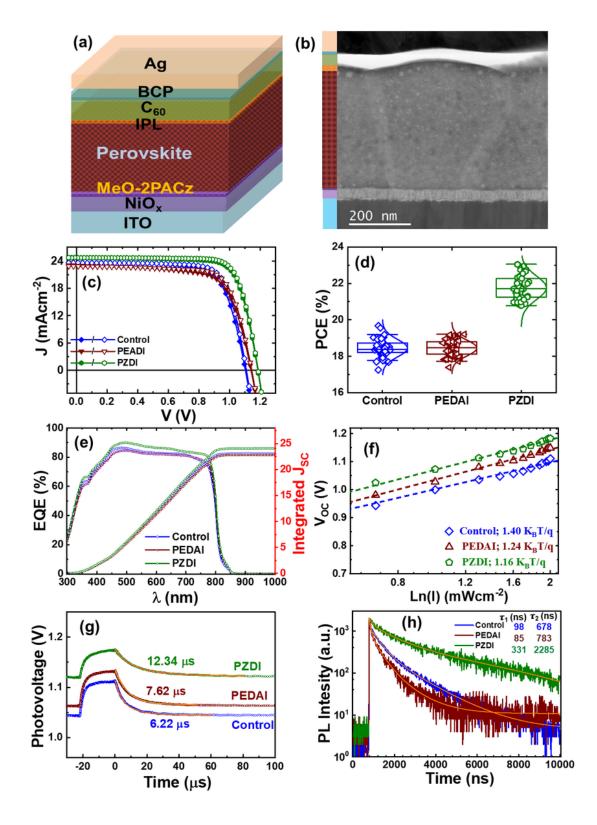
- Ruiz-Preciado, M. A. *et al.* Supramolecular Modulation of Hybrid Perovskite Solar Cells via Bifunctional Halogen Bonding Revealed by Two-Dimensional 19 F Solid-State NMR Spectroscopy. *J. Am. Chem. Soc.***142**, 1645–1654 (2020).
- Khadka, D. B., Shirai, Y., Yanagida, M., Tadano, T. & Miyano, K. Interfacial Embedding for High-Efficiency and Stable Methylammonium-Free Perovskite Solar Cells with Fluoroarene Hydrazine. *Adv. Energy Mater.* 12, 2202029 (2022).
- 19. Zhu, X. *et al.* Ionic-Liquid-Perovskite Capping Layer for Stable 24.33%-Efficient Solar Cell. *Adv. Energy Mater.***12**, 2103491 (2022).
- Fu, C. *et al.* From Structural Design to Functional Construction: Amine Molecules in High-Performance Formamidinium-Based Perovskite Solar Cells. *Angew. Chemie Int. Ed.*61, e2021170 (2022).
- Pham, N. D. *et al.* Tailoring Crystal Structure of FA 0.83 Cs 0.17 Pbl 3 Perovskite Through Guanidinium Doping for Enhanced Performance and Tunable Hysteresis of Planar Perovskite Solar Cells. *Adv. Funct. Mater.*29, 1806479 (2019).
- 22. Gharibzadeh, S. *et al.* Two birds with one stone: dual grain-boundary and interface passivation enables >22% efficient inverted methylammonium-free perovskite solar cells. *Energy Environ. Sci.***14**, 5875–5893 (2021).
- Alanazi, A. Q. *et al.* Atomic-Level Microstructure of Efficient Formamidinium-Based Perovskite Solar Cells Stabilized by 5-Ammonium Valeric Acid Iodide Revealed by Multinuclear and Two-Dimensional Solid-State NMR. *J. Am. Chem. Soc.***141**, 17659–17669 (2019).
- 24. Xu, W. *et al.* Rational molecular passivation for high-performance perovskite light-emitting diodes. *Nat. Photonics***13**, 418–424 (2019).
- 25. Yang, G. *et al.* Stable and low-photovoltage-loss perovskite solar cells by multifunctional passivation. *Nat. Photonics***15**, 681–689 (2021).
- 26. Lin, Y.-H. *et al.* A piperidinium salt stabilizes efficient metal-halide perovskite solar cells. *Science***369**, 96–102 (2020).
- Zhang, J., Wu, S., Liu, T., Zhu, Z. & Jen, A. K. Y. K. -Y. Boosting Photovoltaic Performance for Lead Halide Perovskites Solar Cells with BF 4 – Anion Substitutions. *Adv. Funct. Mater.*29, 1808833 (2019).
- 28. Pan, T. *et al.* Surface Energy Regulated Growth of α-phase Cs0.03 FA0.97PbI3 for Highly Efficient and Stable Inverted Perovskite Solar Cells. *Adv. Mater.* 2208522 (2023) doi:10.1002/adma.202208522.
- 29. Chen, S., Xiao, X., Gu, H. & Huang, J. lodine reduction for reproducible and high-performance perovskite solar cells and modules. *Sci. Adv.***7**, (2021).
- 30. Zhang, W. *et al.* Enhanced optoelectronic quality of perovskite thin films with hypophosphorous acid for planar heterojunction solar cells. *Nat. Commun.***6**, 10030 (2015).
- 31. Hou, M. *et al.* Aryl Diammonium Iodide Passivation for Efficient and Stable Hybrid Organ-Inorganic Perovskite Solar Cells. *Adv. Funct. Mater.***30**, 2002366 (2020).

- Nakane, A. *et al.* Quantitative determination of optical and recombination losses in thin-film photovoltaic devices based on external quantum efficiency analysis. *J. Appl. Phys.***120**, 064505 (2016).
- 33. Oliver, R. D. J. *et al.* Understanding and suppressing non-radiative losses in methylammonium-free wide-bandgap perovskite solar cells. *Energy Environ. Sci.***15**, 714–726 (2022).
- Chen, J., Kim, S.-G., Ren, X., Jung, H. S. & Park, N.-G. Effect of bidentate and tridentate additives on the photovoltaic performance and stability of perovskite solar cells. *J. Mater. Chem. A*7, 4977–4987 (2019).
- Khadka, D. B., Shirai, Y., Yanagida, M., Masuda, T. & Miyano, K. Enhancement in efficiency and optoelectronic quality of perovskite thin films annealed in MACI vapor. *Sustain. Energy Fuels* 1, 755– 766 (2017).
- Matsushita, A., Yanagida, M., Shirai, Y. & Miyano, K. Degradation of perovskite solar cells by the doping level decrease of HTL revealed by capacitance spectroscopy. *Sol. Energy Mater. Sol. Cells*220, 110854 (2021).
- Khadka, D. B., Shirai, Y., Yanagida, M. & Miyano, K. Pseudohalide Functional Additives in Tin Halide Perovskite for Efficient and Stable Pb-Free Perovskite Solar Cells. *ACS Appl. Energy Mater.***4**, 12819– 12826 (2021).
- 38. Miyano, K., Yanagida, M. & Shirai, Y. Impedance Spectroscopy Revisited. *Adv. Energy Mater.***10**, 1903097 (2020).
- 39. Ye, S. *et al.* A Breakthrough Efficiency of 19.9% Obtained in Inverted Perovskite Solar Cells by Using an Efficient Trap State Passivator Cu(thiourea)I. *J. Am. Chem. Soc.***139**, 7504–7512 (2017).
- 40. Khadka, D. B., Shirai, Y., Yanagida, M. & Miyano, K. Attenuating the defect activities with a rubidium additive for efficient and stable Sn-based halide perovskite solar cells. *J. Mater. Chem. C***8**, 2307–2313 (2020).
- Khadka, D. B., Kim, S. Y. & Kim, J. H. A Nonvacuum Approach for Fabrication of Cu2ZnSnSe4/In2S3 Thin Film Solar Cell and Optoelectronic Characterization. *J. Phys. Chem.* C119, 12226–12235 (2015).
- Zhang, K. *et al.* Reducing Limitations of Aggregation-Induced Photocarrier Trapping for Photovoltaic Stability via Tailoring Intermolecular Electron–Phonon Coupling in Highly Efficient Quaternary Polymer Solar Cells. *Adv. Energy Mater.* 2103371 (2021) doi:10.1002/aenm.202103371.
- Walter, T., Herberholz, R., Müller, C. & Schock, H. W. Determination of defect distributions from admittance measurements and application to Cu(In,Ga)Se2 based heterojunctions. *J. Appl. Phys.*80, 4411–4420 (1996).
- Khadka, D. B., Shirai, Y., Yanagida, M., Noda, T. & Miyano, K. Tailoring the Open-Circuit Voltage Deficit of Wide-Band-Gap Perovskite Solar Cells Using Alkyl Chain-Substituted Fullerene Derivatives. ACS Appl. Mater. Interfaces10, 22074–22082 (2018).
- 45. Saidaminov, M. I. *et al.* High-quality bulk hybrid perovskite single crystals within minutes by inverse temperature crystallization. *Nat. Commun.***6**, 7586 (2015).

- 46. Liu, N. & Yam, C. First-principles study of intrinsic defects in formamidinium lead triiodide perovskite solar cell absorbers. *Phys. Chem. Chem. Phys.* **20**, 6800–6804 (2018).
- 47. Lopez-Varo, P. *et al.* Device Physics of Hybrid Perovskite Solar cells: Theory and Experiment. *Adv. Energy Mater.* 1702772 (2018) doi:10.1002/aenm.201702772.
- 48. Tsai, H. *et al.* High-efficiency two-dimensional Ruddlesden–Popper perovskite solar cells. *Nature***536**, 312–316 (2016).
- 49. Zhang, X. *et al.* Inorganic CsPbI 3 Perovskite Coating on PbS Quantum Dot for Highly Efficient and Stable Infrared Light Converting Solar Cells. *Adv. Energy Mater.***8**, 1702049 (2018).
- 50. Chen, B., Yang, M., Priya, S. & Zhu, K. Origin of J-V Hysteresis in Perovskite Solar Cells. *J. Phys. Chem. Lett.***7**, 905–917 (2016).
- 51. Yanagida, M., Shirai, Y., Khadka, D. B. & Miyano, K. Photoinduced ion-redistribution in CH 3 NH 3 Pbl 3 perovskite solar cells. *Phys. Chem. Chem. Phys.* **22**, 25118–25125 (2020).
- 52. Sun, J., Ruzsinszky, A. & Perdew, J. P. Strongly Constrained and Appropriately Normed Semilocal Density Functional. *Phys. Rev. Lett.***115**, 036402 (2015).
- 53. Kresse, G. & Furthmüller, J. Efficient iterative schemes for ab initio total-energy calculations using a plane-wave basis set. *Phys. Rev. B***54**, 11169–11186 (1996).
- 54. Kresse, G. & Joubert, D. From ultrasoft pseudopotentials to the projector augmented-wave method. *Phys. Rev. B***59**, 1758–1775 (1999).



Schematic illustration of surface treatment; a) chemical structure of (1,4-phenylenediamine dihydrodide (PEDAI) and piperazine dihydriodide (PZDI), b) surface passivation of MA-free ($FA_{0.84}Rb_{0.04}Cs_{0.12}PbI_3$) 3D-HP film, c) illustration of interfacial interaction of diammonium iodide molecular passivator and HP surface. SEM images of HP films d) without passivation, e) with PEDAI, and f) with PZDI passivation, respectively. g) XRD patterns, h) absorption spectra, and i) PL spectra without and with IPL treatment on the HP films.



HPSCs with surface treatment. a) Schematics of the device structure. b) STEM-cross-sectional images of devices. c) *J-V* characteristics of the control and IPL treatment (best IPL content; 1mg/ml); (open/filled symbols (forward/reverse) scan direction). d) Statistical box for *PCE* of the HPSCs without and with surface passivation (PEDAI or PZDI). These data consist of 50 devices from 6 batches. e) EQE spectra of

devices. f) Light intensity vs V_{OC} plot. g) Transient photovoltage (TPV) decay curves of the respective devices. h) TRPL decay spectra for corresponding films.

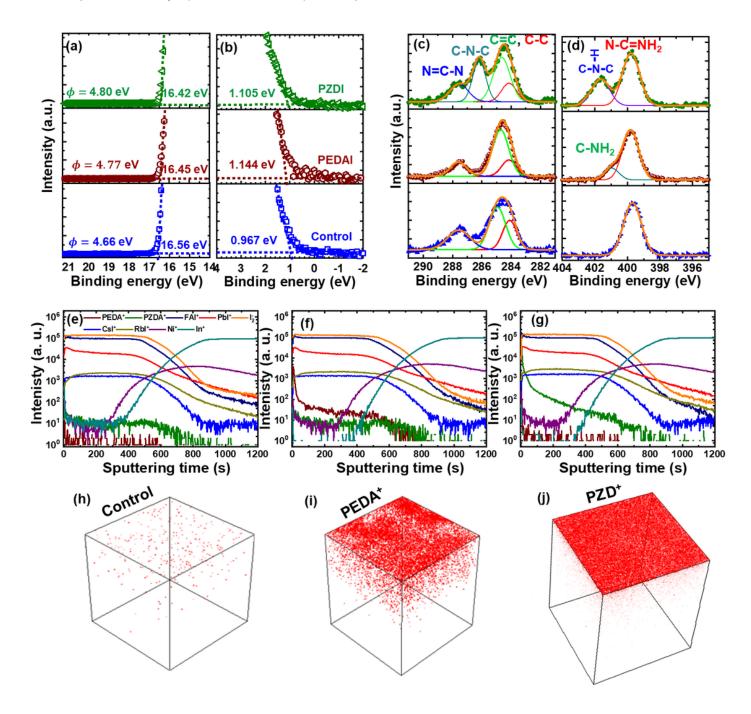
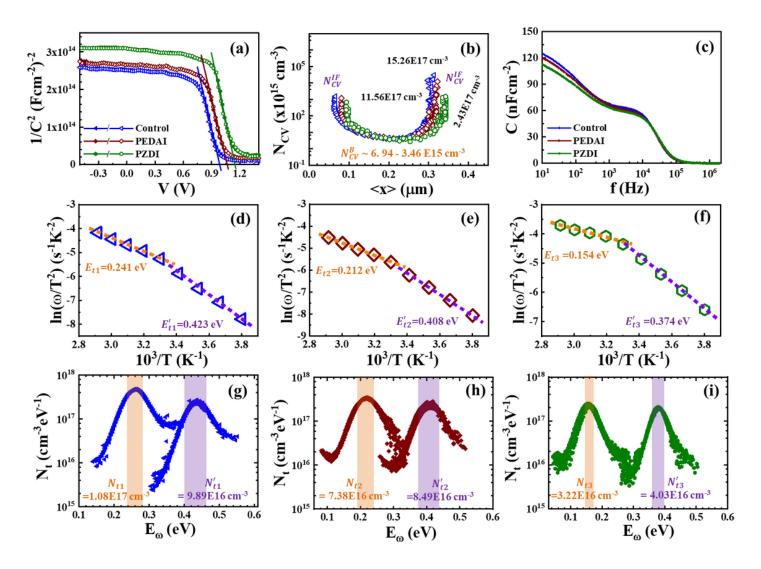
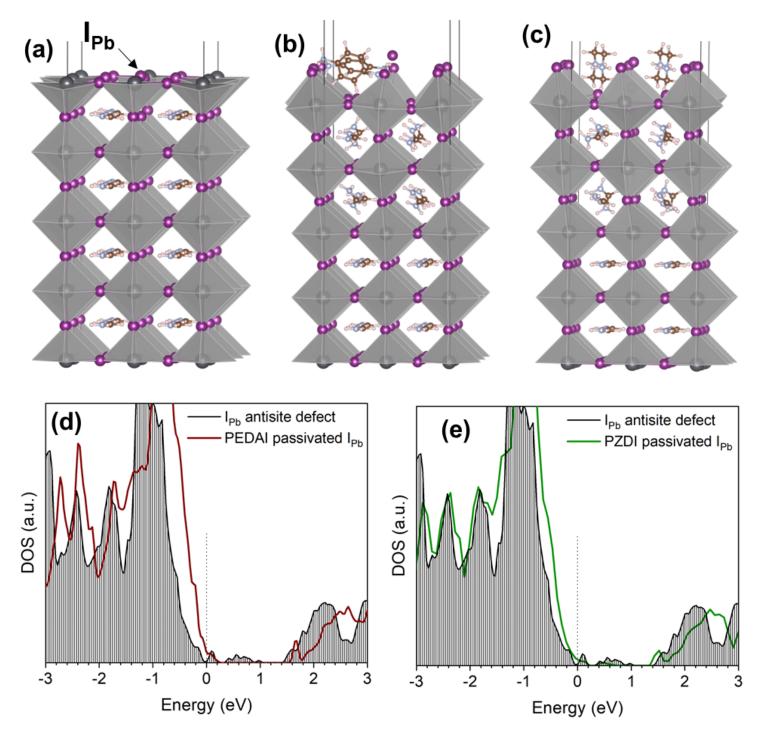


Figure 3

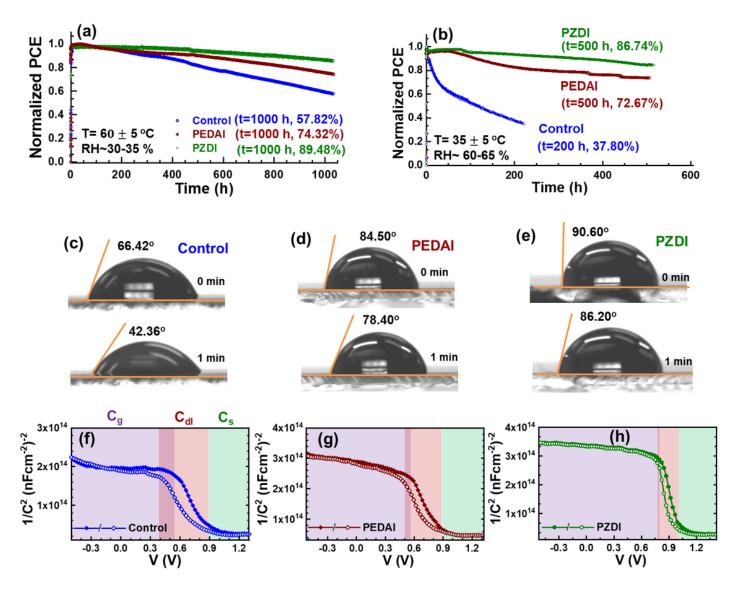
UPS spectra of the HP films; control, PEDAI, and PZDI treatment (a, b). XPS-spectra analysis; c) C-1s core and d) N-1s core. ToF-SIMS depth profiles of; e) control, f) PEDAI, and g) PZDI passivated HP films. Reconstructed 3D maps; h-j) the distributions of passivated molecules in HP film. There are selected ionic species; ITO (In⁺), NiO_x (Ni⁺), HP control (FAI⁺, PbI⁺, CsI⁺, RbI+) and with PEDAI (PEDA²⁺) or PZDI (PZD²⁺).



Capacitance characteristics of HPSCs. a) Mott-Schottky plots (open/filled symbols forward/reverse scan direction), b) carrier distribution calculated from C-V data, c) capacitance-frequency (C-f) response, d-f) Arrhenius plots, and g-i) defect profile (N_{t}).



Optimized structures of (a) I_{Pb} antisite defect, (b) PEDAI, and (c) PZDI adsorbed on the PbI₂-terminated surface of FAPbI₃ with I_{Pb} antisite defect. The total DOS calculated for the PbI₂-terminated surface of FAPbI₃ with I_{Pb} antisite defect passivated with (d) PEDAI and (e) PZDI. Black lines correspond to the total DOS calculated for the unpassivated perovskite surface with the I_{Pb} defect.



Stability of the control, PEDAI, and PZDI passivated HPSCs; stability monitoring a) operational tracking under MPPT conditions: T=60±5°C; 30–35% RH and b) T=35±5 °C; RH~ 60–65%. Images of the water contact angle on the surface of f) control, g) PEDAI, and h) PZDI perovskite films at different water loading times (initial (0 min) and after 1 min). c-e) M-S plots (open/filled symbols: reverse scan direction) of aged HSPCs (T=60±5 °C; RH~ 30–35%). The color-shaded region represents the characteristic capacitance regime. Overlapped shaded regions represent the lagging of the M-S curve induced by interfacial deterioration.

Supplementary Files

This is a list of supplementary files associated with this preprint. Click to download.

- nrphotovoltaicreporting.pdf
- 3PSCsSIDB.docx

• TOC.png