

Performance, Combustion and Emission evaluation of Direct Injection diesel engine fueled with ZnO dispersed waste cooking oil biodiesel

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Abstract

In present scenario, the fast depletion of fossil fuels and energy crises motivated the researchers towards finding an alternative sources for energy. In this work, the performance, combustion and emission of hydrothermally synthesized ZnO nanoparticles dispersed waste cooking oil biodiesel in single cylinder, four stroke, Direct Injection- Diesel Engine with eddy current dynamometer was investigated. The synthesized ZnO nano particles were characterized by XRD, FTIR, SEM and EDX to find crystallinity, functional groups and surface morphology. Waste cooking oil biodiesel was produced via conventional one step alkali catalyst Trans-esterification process. Test fuels were prepared with dispersion of ZnO nanoparticles in the concentration of 10ppm, 20ppm and 30ppm with B20 (20% waste cooking oil biodiesel and 80% diesel) with the aid of magnetic stirrer and ultra-sonication. The test results exposed that the addition of ZnO nanoparticles with B20 gave better Performance in terms of BTE and BSFC compared to B20. The addition of ZnO nano particles with 30ppm concentration was identified as better fuel among all tested fuels in this work. 30ppm addition of ZnO nano particles with B20 records 1.78 % increase in BTE and 10.34% decrease in BSFC. The emission like CO, HC and smoke were 20%, 15.4% and 17.39% lower than diesel. On the other hand NO_x was slightly higher than diesel but lower than B20.

1. Introduction

Biodiesel from various feedstock, such as vegetable oils and animal fats, is now the most promising and economically viable alternative fuel for diesel engines. Using biodiesel in a diesel engine causes problems such as greater fuel consumption, low power, increased NO_x and cold starting issues.(Hosseini and Wahid 2012) Modified fuel, improved engine design and post-treatment of engine exhaust are the main ways of addressing these problems.(Ahmed et al. 2020) Fuel alteration is an effective way to overcome the problems associated with the use of biodiesel in the engine through a cylinder strategy rather than a high cost of fitting and maintaining exhaust gas treatment. Additives are chemical substances that are introduced with Fuel as a means of adjusting the fuel properties.(Chacko and Jeyaseelan 2020; Tomar and Kumar 2020) Various fuel additives, including antioxidants, oxygenates, metal and metal oxides, cetane enhancers, lubricants and cold flow enhancers, have been widely used as fuel additives(Hosseinzadeh-Bandbafha et al. 2018). The addition of a fuel additive from 20 ppm to 500 ppm is at an affordable level. The greater size of the metal and metal oxide particles causes problems such as sedimentation, conglomeration and non-uniform distribution of the size. As a result, a nano-sized particle that is less than 100nm can be used as a diesel or biodiesel additive in an engine without any difficulty(Tomar and Kumar 2020). The addition of nano particles with base fuel tends to complete combustion by the catalytic action of particles which have larger active surface area for the chemical reactions(Saxena et al. 2017)

Many researchers have studied the impact of nano particles with diesel – biodiesel blends on the fuel properties, combustion, Performance and emission characteristics in diesel engine.(Kumar et al. 2017; Najafi 2018; Soudagar et al. 2018; Ağbulut et al. 2020). The impact of MgO nano particles on the cold flow properties, performance and emission characteristics with waste cooking oil biodiesel (Ranjan et al.

2018) was experimentally studied in a constant speed, single cylinder, water cooled, vertical compression ignition engine and found that, the addition of MgO in the concentration of 30 ppm gave better fuel properties such as cloud point, cold filter plugging point and pour point. The Specific fuel consumption was higher than petroleum based diesel (PBD) while emissions records lesser than B100 and PBD. The combustion characteristics are comparable with PBD.

Some of the researchers used Aluminium oxide nano particles as a fuel additive with diesel and biodiesel fuels. The performance, combustion and emission characteristics of diesel engine fueled with Aluminium oxide nano particles added jojoba methyl ester (El-Seesy et al. 2018) was investigated and concluded that addition of Al_2O_3 enhance the performance, combustion and emission characteristics. However, the preferable emission nature was obtained at 20mg/l dosing of Al_2O_3 while performance and combustion was better at 40 mg/l dosage. The comparison of emission and performance suggested that 30mg/l dosing gave better result in all aspect. The addition of Al_2O_3 with pungamia biodiesel (Sivakumar et al. 2018) was examined in constant speed, single cylinder, direct injection diesel engine and found, that Brake thermal efficiency has increased slightly and brake specific fuel consumption has reduced. The emissions like CO, UHC and smoke emission has decreased while NO_x has increased. The influence of Aluminium oxide nano particles with honge oil methyl ester on the performance and emission characteristics of diesel engine (Soudagar et al. 2020) was investigated and concluded that addition of 40 ppm Al_2O_3 resulted in increasing brake thermal efficiency by 10.57% and Brake specific fuel consumption, CO, HC and Smoke were reduced by 11.65%, 48.43%, 26.72% and 22.84% respectively while NO_x has increased about 11.27%.

Graphene oxide was used as a nano additive with Jatropha methyl ester and tested in single cylinder air cooled direct injection diesel engine (EL-Seesy et al. 2018) and found that BTE, peak cylinder pressure, the highest rate of pressure rise, and maximum heat release rate were increased by 17%, 8%, 6%, and 6% respectively compared to jatropha methyl ester. HC and CO were lower than pure jatropha methyl ester about 60% and 50% respectively and at full load condition, NO_x was reduced by 15%.

The effect of cerium oxide nano particles with mahua methyl ester (B20) in the concentration of 50 ppm, 100 ppm and 150 ppm in diesel engine was examined (Kumar et al. 2019) The brake thermal efficiency and combustion pressure data has increased slightly. At the same time, emissions like CO, HC and Smoke were reduced dramatically. The NO_x emission also decreased due to the oxidation of unburned Ce_2O_3 at the exhaust.

The experimental study was conducted by using ZnO and TiO_2 as a fuel additive with Calophyllum inophyllum methyl ester (CIME) as a emulsified fuel in CI engine and found that brake thermal efficiency was increased by 5 – 17% compared to pure CIME. CO, HC and smoke were drastically decreased when compared to pure CIME and diesel. On the other hand, NO_x was lower than pure CIME and higher than Diesel. (Nanthagopal et al. 2017) The effect of ZnO and ethanox as an fuel additive with Calophyllum inophyllum biodiesel in CI engine was examined and shown that addition of 100 ppm of ZnO enhance

the efficiency by 4.7% and reduce the NO_x emission by 12.6% at full load condition, due to the catalytic effect of nano particles and macro explosion of water molecules in the fuel emulsion.(Ashok et al. 2017).

From the above literature it is noted that the addition of metal oxide nanoparticles with diesel / biodiesel fuel, significantly affect the performance, combustion and emission characteristics of diesel engine. In this work, the impact of hydrothermally synthesized ZnO nano particles with waste cooking oil biodiesel on the performance, combustion and emission in single cylinder, water cooled, DI Diesel engine has investigated.

2. Experimental Setup

2.1. Materials and Methods

2.1.1. Preparation of Zinc Oxide nano particles

Zinc oxide nano particles were synthesized by hydrothermal method as shown in fig.1. (Prabhu et al. 2018) In this method zinc acetate is taken as precursor and triethylamine is used as a surfactant. All the reagents were purchased from sigma Aldrich in the analytical grade and used without further pretreatment. In this process required amount of Zinc acetate was dissolved in deionized water with the aid of magnetic stirrer then,5ml of triethylamine was added dropwise with continues stirring and finally got a white color precipitate. The obtained precipitate was undergoes hydrothermal treatment at 200⁰C for 12 hours in a Teflon lined autoclave at hot air oven. After hydrothermal treatment the sample was passively cooled to room temperature and collected by centrifugation. Finally the sample was collected after drying in a hot air oven about 80⁰C for 2 hours.

2.1.2. Preparation of waste cooking oil biodiesel

Biodiesel from waste cooking oil is an economical way of generating energy and handling waste oil generated in hotels, restaurants, university hostels, etc.(Zareh et al. 2017) Waste cooking oil biodiesel was produced by conventional Trans – esterification process as shown in fig.2. In typical process waste cooking oil and methanol were taken in the molar ratio of 6:1 with Sodium hydroxide as a base catalyst and stirred for 2 hours under 65⁰C with the help of magnetic stirrer, then it is allowed for gravitational separation for 12 hours. Layer of waste cooking oil methyl ester and glycerol were formed as upper and lower layers respectively during gravitational separation process. The waste cooking oil methyl ester was collected and cleaned with deionized water for further purification.(Abed et al. 2018; Ganapathi and Muralidharan 2020)

2.1.3. Test fuel preparation

The test fuels were prepared by adding the ZnO nanoparticles with B20 by means of agitation. In order to avoid the agglomeration of nano particles in the test fuel, ultra-sonication was adopted.(Gad and Jayaraj 2020; Manigandan et al. 2020) ZnO nano particles were taken in three proportions of 10 ppm, 20 ppm

and 30 ppm. Initially diesel and biodiesel were taken in the beaker in the ratio of 4:1 and mixed together with the aid of magnetic stirrer to get B20. The measured ZnO was poured into the B20 and stirred for 30 min, followed by ultra-sonication for 10 min as shown in fig.3 (Janakiraman et al. 2020) to produce three different fuels such as B20 + 10 ppm ZnO, B20 + 20 ppm ZnO and B20 + 30 ppm ZnO.

2.1.4. Test Engine Setup

The engine performance, combustion and emission of prepared test fuels were tested in a single cylinder, four strokes, water-cooled, and vertical diesel engine with a constant speed of 1500 rpm at 0%, 25%, 50%, 75% and 100% loading conditions as specified in table 1. The test was carried out at fuel injection timing of 23⁰CA before TDC with an injection pressure of 200 bar and 27⁰C of ambient temperature. The arrangement of the crank angle encoder, dynamometer, Computer, data acquisition system, five gas analyser and a smoke meter in the test engine RIG is as shown in fig.4.

Table 1

Engine Specifications

Parameters	Specifications
Manufacturer & Model	Kirloskar & TV1
No of cylinders	1
Cylinder arrangement	Vertical
No of Strokes	4
Power	5.2kW at 1500 rpm
Cooling type	water cooled
Stroke length	110 mm
Bore diameter	87.5 mm
Compression ratio	17.5:1
Loading type	Eddy current dynamometer with water cooling

3. Result And Discussion

3.1. Nano particles and fuel characterization

The prepared ZnO nano particles were characterized by XRD, FTIR, SEM, TEM and EDX to identify the crystal structure, surface composition, surface morphology, internal structure and elemental composition. The XRD studies confirmed the prepared material was hexagonal wurtzite crystal of ZnO

(JCPDS Card no. 16-1451). The diffraction peaks were present at the 2θ values of 31.770, 34.422, 36.253, 47.539, 56.603, 62.864, 66.380, 67.963 and 69.100 were indexed as the (100), (002), (101), (102), (110), (103), (200), (112) and (201) planes as shown in fig.5.(Cao et al. 2017; Prabhu et al. 2018; Hui et al. 2019) the surface composition was studied by FTIR. The spectra was recorded in the range of 400 – 4000 cm^{-1} as shown in fig.6. The strong and wide band present around 3400 cm^{-1} indicates the stretching vibration of O-H group due to the presence of water molecules. The intense transmission band located between 480 cm^{-1} to 580 cm^{-1} belongs to the stretching vibration of zinc and oxygen molecules.(Pudukudy and Yaakob 2014; Prabhu et al. 2018)

Fig.7. shows the SEM and TEM images which are used to determine the surface morphology as well as internal morphology of the prepared sample. The formation of spindle like structure was identified. The same morphology of ZnO material were reported in previous studies, (Zhou et al. 2013; Ameen et al. 2015; Hui et al. 2019) the TEM images confirms the Spindle like shape which is in good contract with SEM results. The elemental composition of prepared sample is shown in fig.8, it is evident that uniform distribution of Zinc and oxygen throughout the sample and the composition of Zinc is 59% and Oxygen is 41%.

The fuel properties such as Density, Kinematic viscosity, Calorific value, Flash point and Fire point for test fuels (diesel, B20, B20 + ZnO (10 ppm), B20 + ZnO (20 ppm), B20 + ZnO (30 ppm)) were identified as per the ASTM standards as shown in table.2.

Table 2

Test fuel characterization

Fuel Name / Properties	Density (Kg/m ³)	Kinematic Viscosity at 40 ^o C (cSt)	Calorific Value (kJ/kg)	Flash point (°C)	Fire Point (°C)
Diesel	830	1.8	42500	58	62
B20	865	2.7	40625	56	60
B20 + ZnO (10 ppm)	860	2.5	40723	45	50
B20 + ZnO (20 ppm)	862	2.6	41014	46	52
B20 + ZnO (30 ppm)	864	2.8	41306	48	54

3.2. Brake Thermal Efficiency

Brake thermal efficiency is used to find, how effectively an engine has convert chemical energy available in the fuel(Shrivastava et al. 2019). Fig.9. shows the variation of Brake Thermal Efficiency of test fuels with respect to the load. It is clear that the brake thermal efficiency has dramatically increases when increasing the load due to more combustion and reduction of BSFC at higher loading condition. Typically Lower BTE was recorded due to lacking in combustion of air / fuel mixture in the engine cylinder and the lower calorific value of the fuel. B20 accounts lower BTE than diesel fuel owing to irregular air fuel mixing due to their higher viscosity and density that leads to formation of large droplets during atomization of fuel.(Elkelawy et al. 2019) The addition of ZnO nano particles has led to a greater improvement on BTE due to better atomization of fuels, higher calorific value and shorter ignition delay compared to B20 at all load conditions, which is in good agreement with earlier studies(Sajin et al. 2019; Deepak Kumar et al. 2020; Hussain et al. 2020). At the same time an addition of ZnO nano particles with B20 records lower BTE than diesel. Adding 10 ppm, 20 ppm and 30 ppm of ZnO nano particles with B20 at full load condition, produces a greater BTE than B20 about 0.9%, 1.4% and 1.87% respectively.

3.3. Brake Specific Fuel Consumption

Fig.10. shows the variation of BSFC with respect to the load. The variation of BSFC depends on the density, calorific value and viscosity of fuel. BSFC of B20 is always higher than diesel at all loading condition. It may be due to higher viscosity and density of biodiesel that leads to longer ignition delay, compared to diesel(Rajak et al. 2019). The addition of ZnO nano additives in the concentration of 10 ppm tends to reduce the brake specific fuel consumption than B20 owing to higher calorific value and the better atomization of fuel. Further increasing the concentration of nano additives as 20 ppm and 30 ppm resulted in lower BSFC due to enhanced heat transfer and increasing the contact between fuel and oxidizer as higher surface to volume ratio of nano particles(Devarajan et al. 2019). The BSFC of B20 + ZnO (10 ppm), B20 + ZnO (20 ppm) and B20 + ZnO (30 ppm) were 0.28 kg/kW-hr, 0.27 kg/kW-hr and 0.26 kg/kW-hr respectively, which are 3.44%, 6.89% and 10.34% lower than B20 at full load condition.

3.4. Cylinder Pressure

Fig.11. describes the variation of cylinder pressure based on the crank angle at full load condition. The peak cylinder pressure is recorded as 74.69 bar by diesel fuel due to the lower viscosity, higher calorific value whereas B20 accounts 74.39 bar which is lower than diesel owing to higher viscosity and lower calorific value.(Devarajan et al. 2018) The addition of nano particles with B20 helps to increase the cylinder pressure. It may be due to shorter ignition delay, higher calorific vale and higher heat transfer rate of nano particles. The addition of ZnO with B20 in the ratio of 10 ppm, 20 ppm and 30

ppm produce peak pressure about 74.56 bar, 74.27 bar and 74.42 bar at full load condition respectively. This trend of increasing cylinder pressure with addition of nano particles were matched with previous studies.(Ashok et al. 2017; Najafi 2018)

3.5. Ignition Delay

The time period between start of fuel injection and start of combustion is known as ignition delay(Prabhakar et al. 2019) MFB is used to evaluate the ignition delay. The crank angle at which 10% of MFB is denoted as CA10 and the crank angle at which the 90% of MFB is denoted as CA90. The difference between CA10 and CA90 is called as combustion duration at the same time the difference between start of fuel injection and CA10 is known as ignition delay(Cooney et al. 2009; Kim et al. 2019). Since the ignition delay is inversely proportional to the cetane number, biodiesel and its blends often account for shorter ignition delays because it has higher cetane index(Aldhaidhawi et al. 2017; Shrivastava et al. 2019). The ignition delay has decreases when increasing the engine load, and the addition of nano additives tends to lower ignition delay as shown in fig.12. It may due to better catalytic activity of nano particles, as it has large surface area to volume ratio, thus enhance the mixing of fuel with oxidizer.(Aldhaidhawi et al. 2017; Nanthagopal et al. 2017)

3.6. Net Heat Release Rate

The heat release rate of the test fuels with respect to the crank angles at full load condition is as shown in fig.13. It shows that the heat release rate has increased when increasing the load due to the fact that engine drawn more fuel to generate more torque which leads to more heat generation(Reang et al. 2020).Heat release rate becomes positive when the combustion is initiated and increases gradually during the primary combustion of fuel. The heat release rate of B20 is lower than diesel at full load condition due to the higher viscosity and poor atomization of biodiesel. The addition of ZnO nano particles supports to higher cylinder pressure due to the higher surface to volume ratio of nano particles helps to complete combustion resulting in higher heat release rate.(Ranjan et al. 2018; Murugesan et al. 2020) The peak heat release rate for diesel , B20, B20 + ZnO (10 ppm), B20 + ZnO (20 ppm) and B20 + ZnO (30 ppm) were 40.56 J,39.09J, 42.03J, 40.83J and 40.91 J respectively at 7⁰ CA before TDC at full load condition.

3.7. CO emission

More carbon monoxide emissions have resulted from incomplete combustion caused by improper mixing of fuel and oxidant, lack of oxygen, and lower cylinder temperatures(Mehregan and Moghiman 2018). The variation of CO emission with respect to the load is as shown in fig.14. It shows that the CO emission

was decreases with increasing load, except full load condition for all fuel blends. It is due to lower air / fuel ratio and lower in-cylinder temperature at zero load condition has led to partial combustion, when increasing the load the surplus amount of oxygen and higher in cylinder temperature affirms complete combustion and resulted in lower emission of CO(Chandrasekaran et al. 2016; Fayad and Dhahad 2021). at the same time, engine drawn more fuel to generate extreme torque at full load condition that results in more emission of CO(Tan et al. 2012; Anchupogu et al. 2018). The CO emission of B20 records lower than diesel for all loading conditions due to more availability of oxygen in the biodiesel helps to enhance the combustion(Jiaqiang et al. 2018). Addition of ZnO nano particles helps to further reduction in CO which may due to the catalytic action of nano particles that helps to enhance the combustion process. The same way of reduction in CO when adding nano particles reported in earlier studies.(Murugesan et al. 2020; Ağbulut et al. 2021; PALANI et al. 2021). The CO emission of B20, B20 + ZnO (10 ppm), B20 + ZnO (20 ppm), and B20 + ZnO (30 ppm) records 10.6%,13.63%,17.4% and 20% lower than diesel fuel at full load condition.

3.8. CO₂ Emission

Higher oxygen content in the fuel and complete combustion results in greater emission of CO₂ which is inversely proportional to the emission of CO(Ashok et al. 2020; Soudagar et al. 2021). the variation of CO₂ with respect to the load for different test fuels are as shown in fig.15. It shows that CO₂ emission has slightly increases when increasing the load and B20 accounts maximum when compared to diesel at all loading condition owing to higher oxygen content in the biodiesel that tends to complete combustion(Ranjan et al. 2018). Further addition of ZnO nano particles with B20 helps to enhance the combustion that leads to more emission of CO₂ due to the higher surface to volume ratio of nano particles involved as a combustion catalyst. The results revealed that addition of ZnO nano particles with B20 generate more CO₂ for all loading conditions, which is evident that higher concentration of ZnO nano particles resulting in better combustion. The CO₂ emissions of B20, B20 + ZnO (10 ppm), B20 + ZnO (20 ppm), and B20 + ZnO (30 ppm) were 3.12%, 3.46%, 4.13% and 4.35% higher than diesel at full load condition. This results are in decent agreement with previous investigations.(Chandrasekaran et al. 2016; Fayad and Dhahad 2021)

3.9. Smoke Opacity

Fig.16.illustrated discrepancy of smoke opacity of diesel, B20 and various concentration of ZnO with B20. Smoke emission has been increasing due to partial combustion at higher loads as more fuel entering the cylinders to maintain constant speed.(Ranjan et al. 2018) Smoke opacity of B20 was lower at all loading conditions, because of higher oxygen levels which facilitates the better combustion and further addition of ZnO nano particles with B20 resulting in reduction of smoke emission than B20 owing

to shorter ignition delay and improved combustion due to larger surface to volume ratio of ZnO nano particles that act as combustion catalyst (Kumar et al. 2020; Venu and Appavu 2020a). The average reduction of Smoke emissions for B20, B20 + ZnO (10 ppm), B20 + ZnO (20ppm) and B20 + ZnO (30 ppm) were noted that, 2.92%, 3.79%, 9% and 15.4% when compared to diesel. This similar results were noted in previous investigations.(Prabu and Anand 2016; El-Seesy et al. 2018; Sadhik Basha 2018; Gad et al. 2021)

3.10. HC Emission

Lower cylinder temperature, deposits on the combustion chamber walls, non-stoichiometric Air / fuel ratio and incomplete combustion are the factors that greatly contributes to the more release of UHC (Prabu and Anand 2016; Hosseini et al. 2017). In all cases, the emission of UHC were noted in the descending order of Diesel, B20, B20 + ZnO (10 ppm), B20 + ZnO (20ppm) and B20 + ZnO (30 ppm) as shown in fig.17. owing to higher oxygen content in the biodiesel helps to complete combustion. However, ZnO nano particles with the concentration of 30 ppm added with B20 generates lower UHC than all other fuels. it is due to that large surface to volume ratio, higher heat transfer rate, oxygen availability and catalytic action of nanoparticles tends to complete combustion(Venu and Appavu 2020a, b).

3.11. NOx emission

Fig.18. depicts the NOx emission pattern of tested fuels. NOx is produced owing to higher combustion temperatures, availability of oxygen and lower ignition delay(Gad and Jayaraj 2020) the NOx emission of B20 was higher than all other fuels regardless of loading condition. It is due to the more oxygen content in the biodiesel that increases the local combustion temperature.(Hoseini et al. 2020) The addition of ZnO nano particles with B20 reduces the NOx emission due to the shorter ignition delay(Rangabashiam et al. 2020). At the same time, increasing the concentration of ZnO nano particles leads to increase in NOx emission due to the availability of more oxygen in ZnO. The NOx emissions of B20, B20 + ZnO(10 ppm), B20 + ZnO (20 ppm) and B20 + ZnO (30 ppm)were 12.66%, 4.01%, 7.41% and 7.47% higher than diesel at full load condition.

4. Conclusion

ZnO nanoparticles were prepared by hydrothermal method and waste cooking oil biodiesel was prepared via Trans-esterification process. ZnO nano particles were mixed with B20 in the concentration of 10 ppm, 20 ppm and 30 ppm with the aid of magnetic stirrer and ultrasonicator. The prepared fuels were tested in single cylinder, four stroke, water cooled and vertical diesel engine at different loading conditions. The following conclusion were made.

1. The addition of Zinc Oxide nano particles helps to increase the calorific value and reduce the viscosity compared to B20. The shorter ignition delay was noted when the concentration of nano particles increases.
2. The addition of nano particles helps to improve the BTE and reduce the fuel consumption compared to B20.
3. The addition of nano particles reduces the emissions like CO, HC and smoke opacity when compared to diesel and B20.
4. The emission of NO_x has increased when increasing the concentration of nano particles at the same time NO_x was lower than B20.

Nomenclature

B20-20% biodiesel + 80% Diesel

ZnO-Zinc Oxide

ppm-parts per million

B20 + ZnO (10 ppm)-20% biodiesel + 80% Diesel + 10 ppm of ZnO

B20 + ZnO (20ppm)-20% biodiesel + 80% Diesel + 20 ppm of ZnO

B20 + ZnO (30 ppm)-20% biodiesel + 80% Diesel + 30 ppm of ZnO

BTE-Brake Thermal Efficiency

BSFC-Brake specific Fuel Consumption

CO-Carbon Monoxide

CO₂-Carbon Dioxide

UHC -Unburned Hydrocarbon

NO_x -Oxides of Nitrogen

CA -Crank Angle

MFB -Mass Fraction Burned

TDC- Top Dead Centre

XRD - X-Ray Diffraction

FTIR- Spectroscopy-Fourier Transform Infrared Spectroscopy

SEM- Scanning Electron Microscope

TEM -Tandem Electron Microscope

EDX- Spectroscopy-Energy Dispersive X - ray Spectroscopy

Declarations

Conflict of Interest

The authors declare no conflict of interest.

Availability of data and material

No data associated with this article.

Code availability

Not applicable.

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Figures

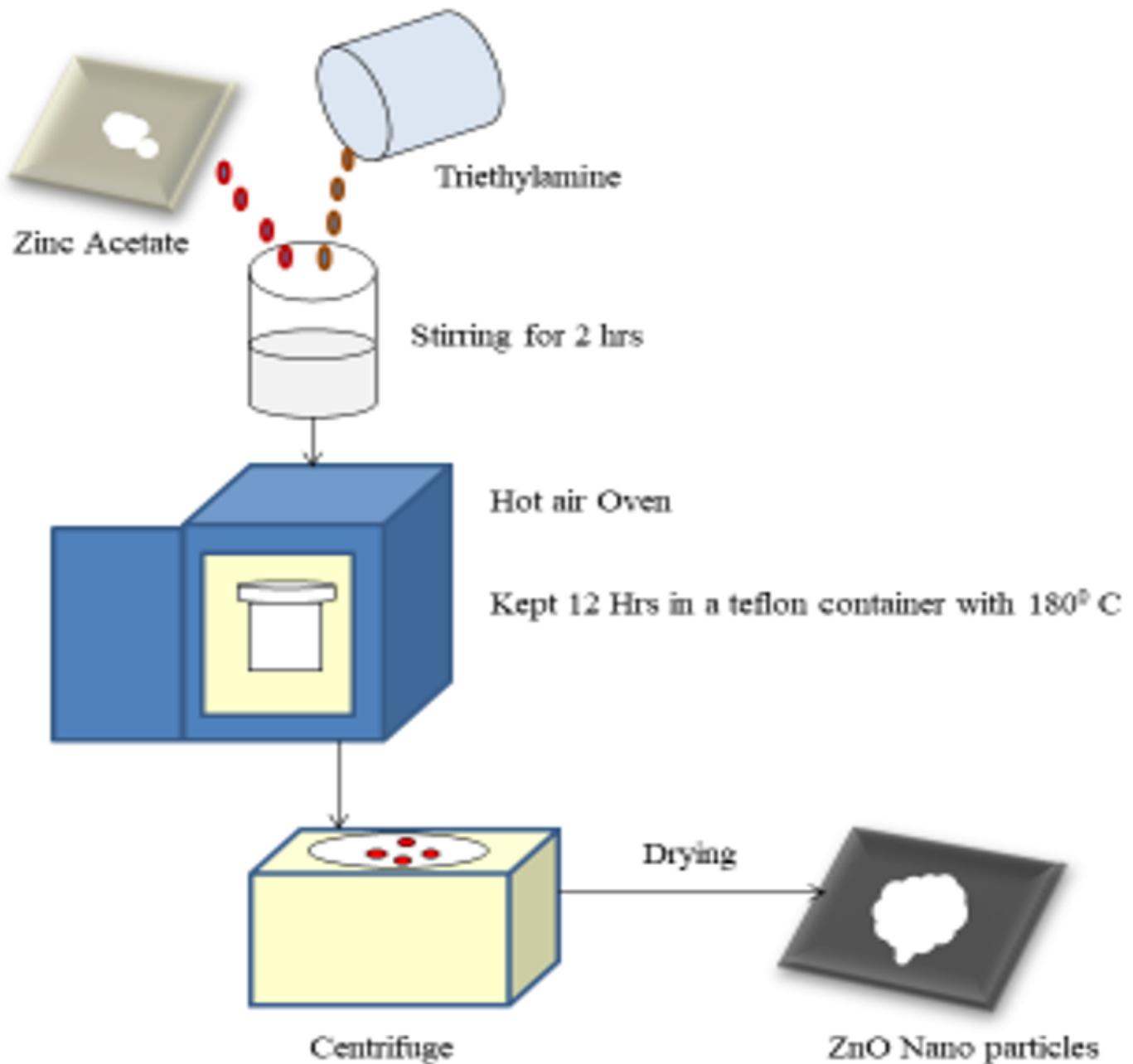


Figure 1

Synthesis of Zinc Oxide nano particles

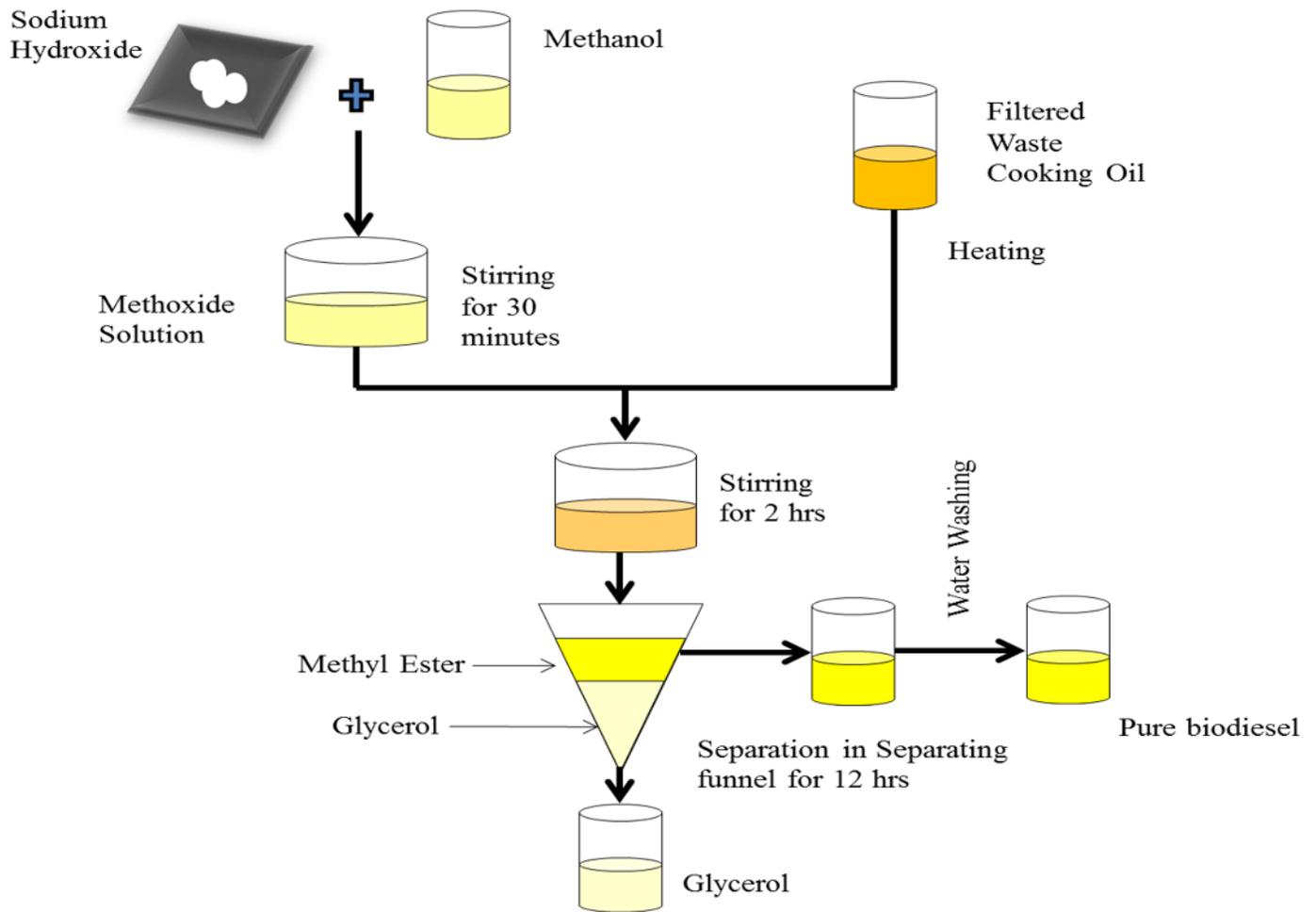


Figure 2

Biodiesel Preparation

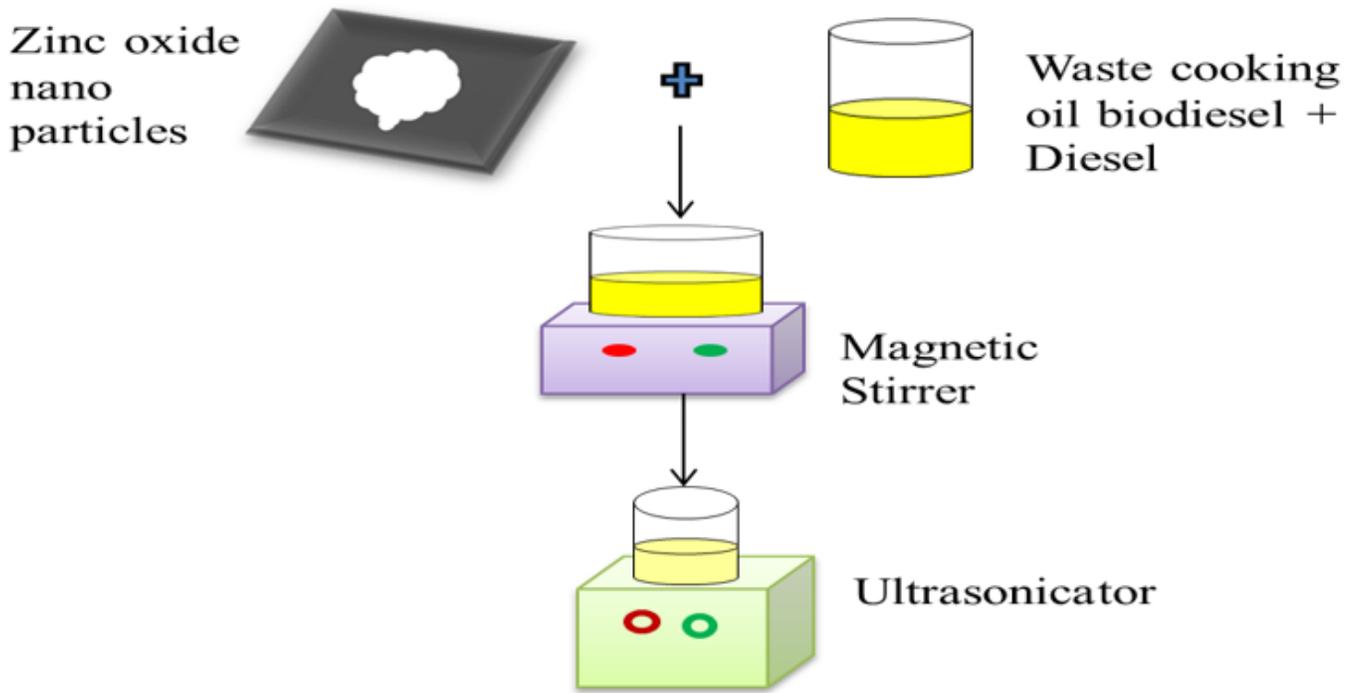


Figure 3

ZnO Addition with Biodiesel

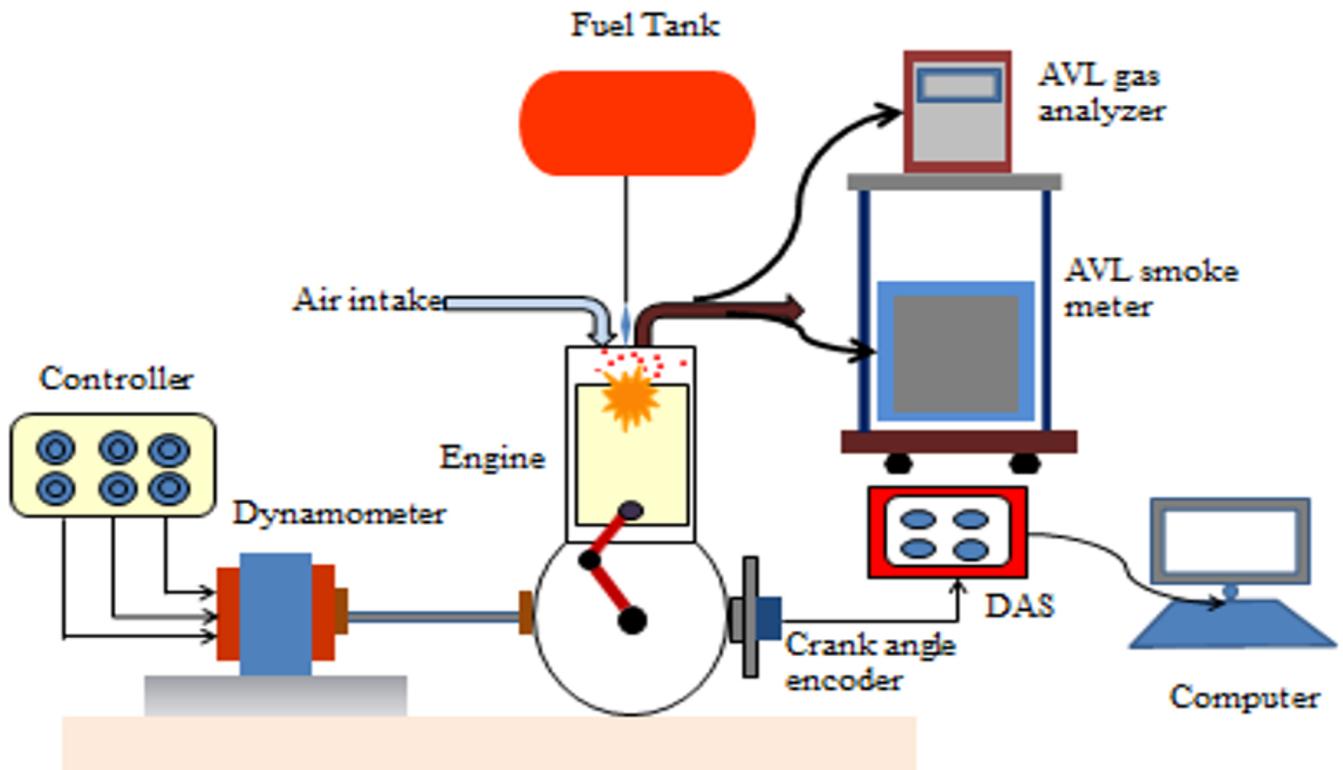


Figure 4

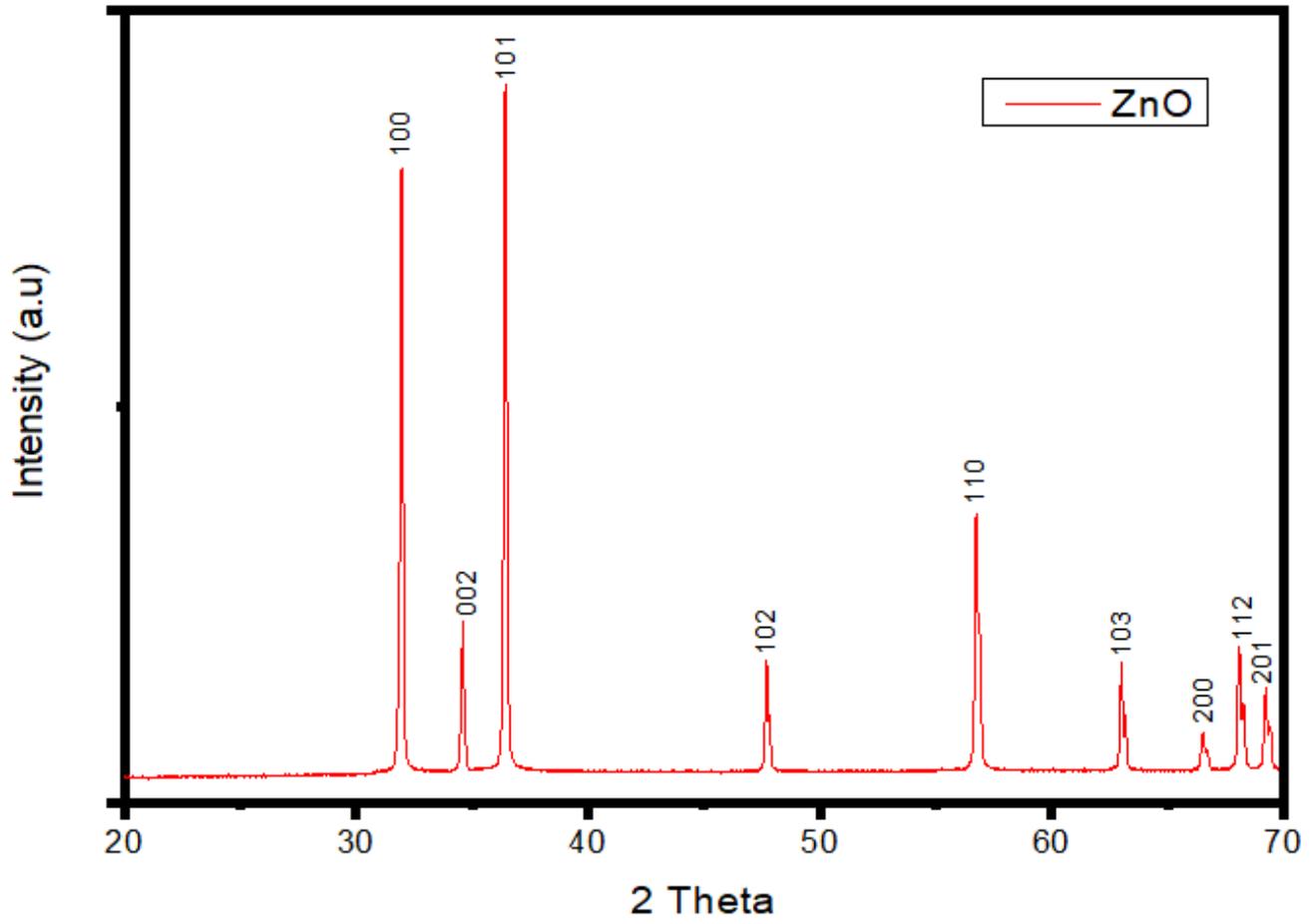


Figure 5

XRD Pattern of ZnO

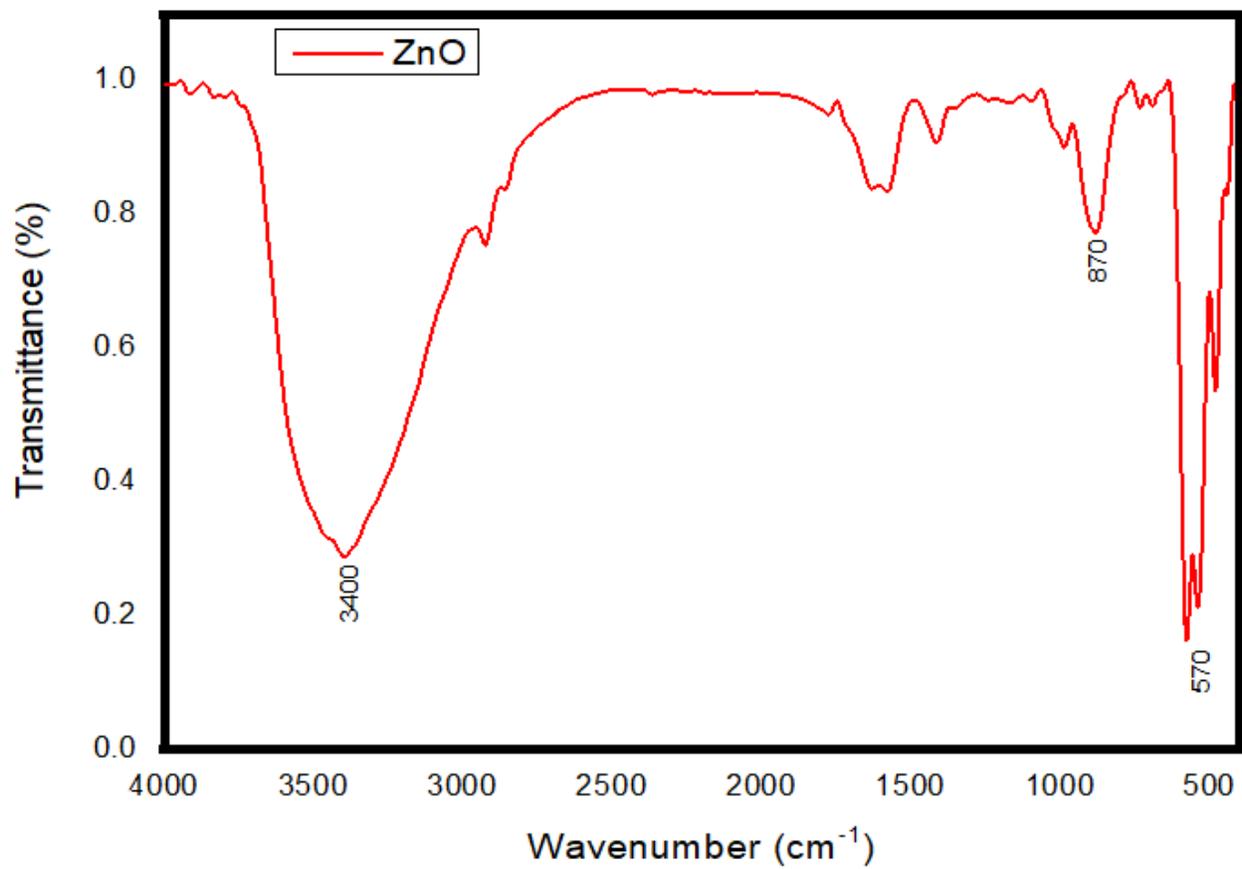


Figure 6

FTIR Spectrum of ZnO

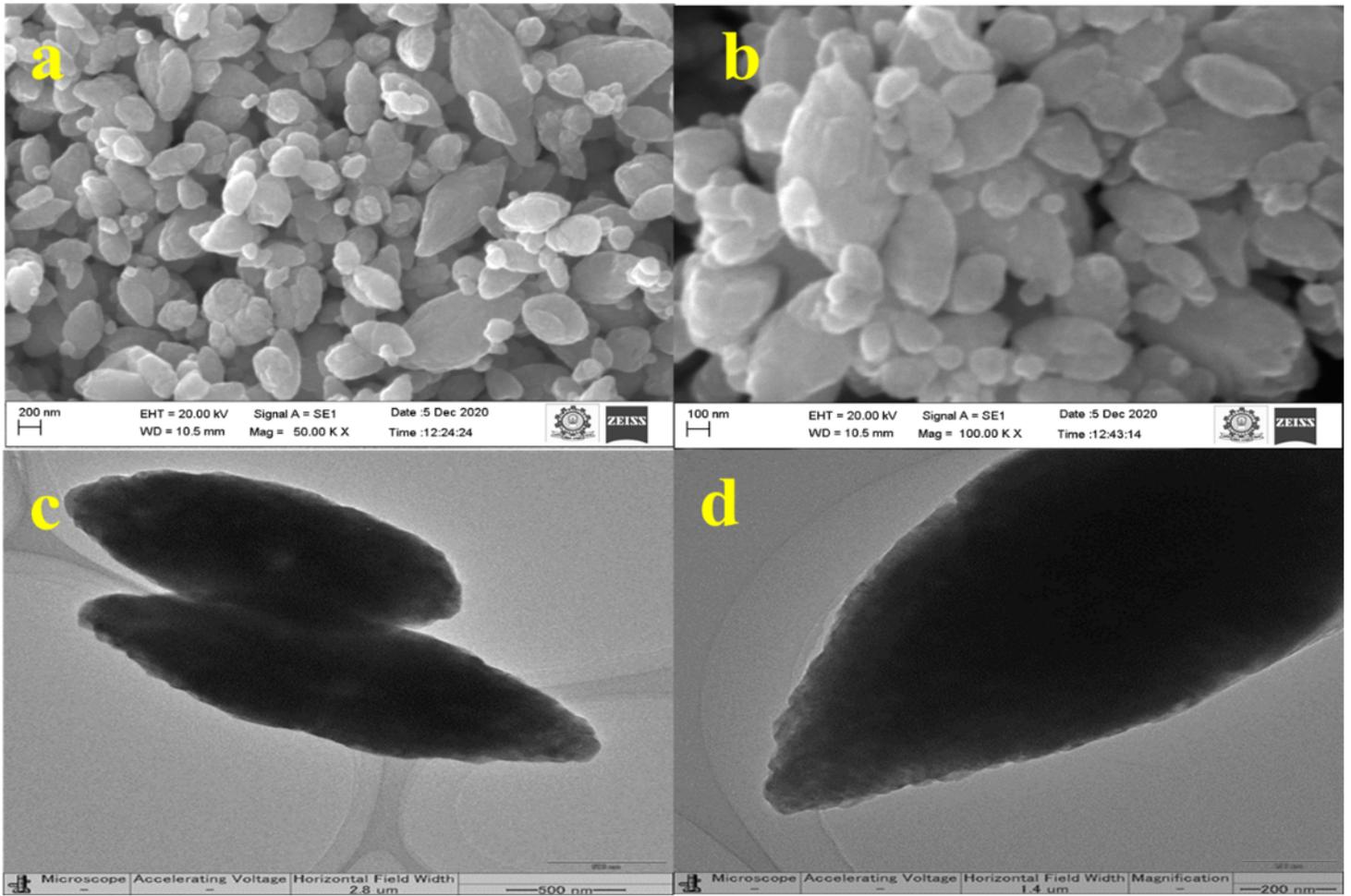


Figure 7

a, b SEM images and c, d TEM images of ZnO

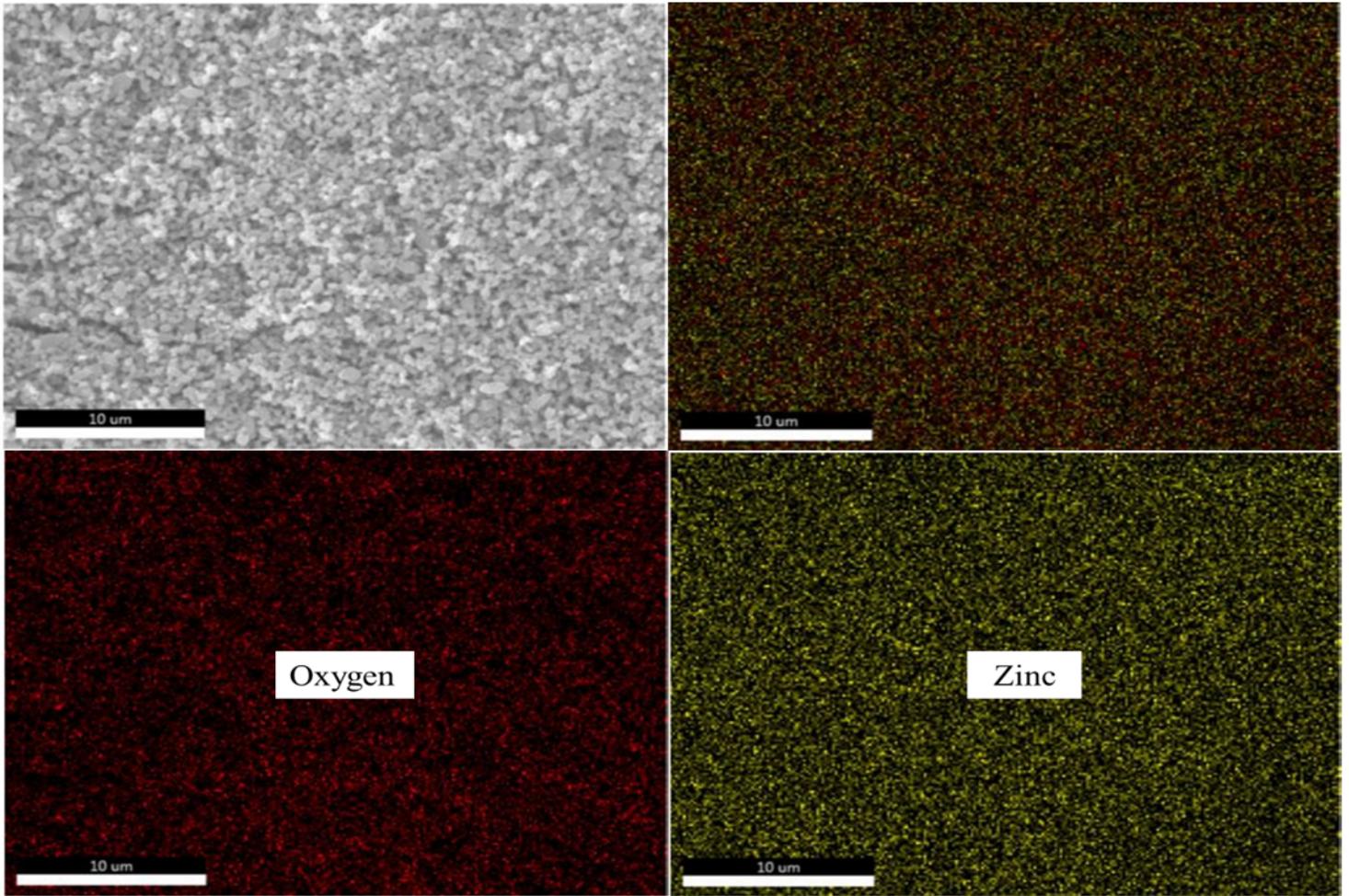


Figure 8

EDX mapping of ZnO

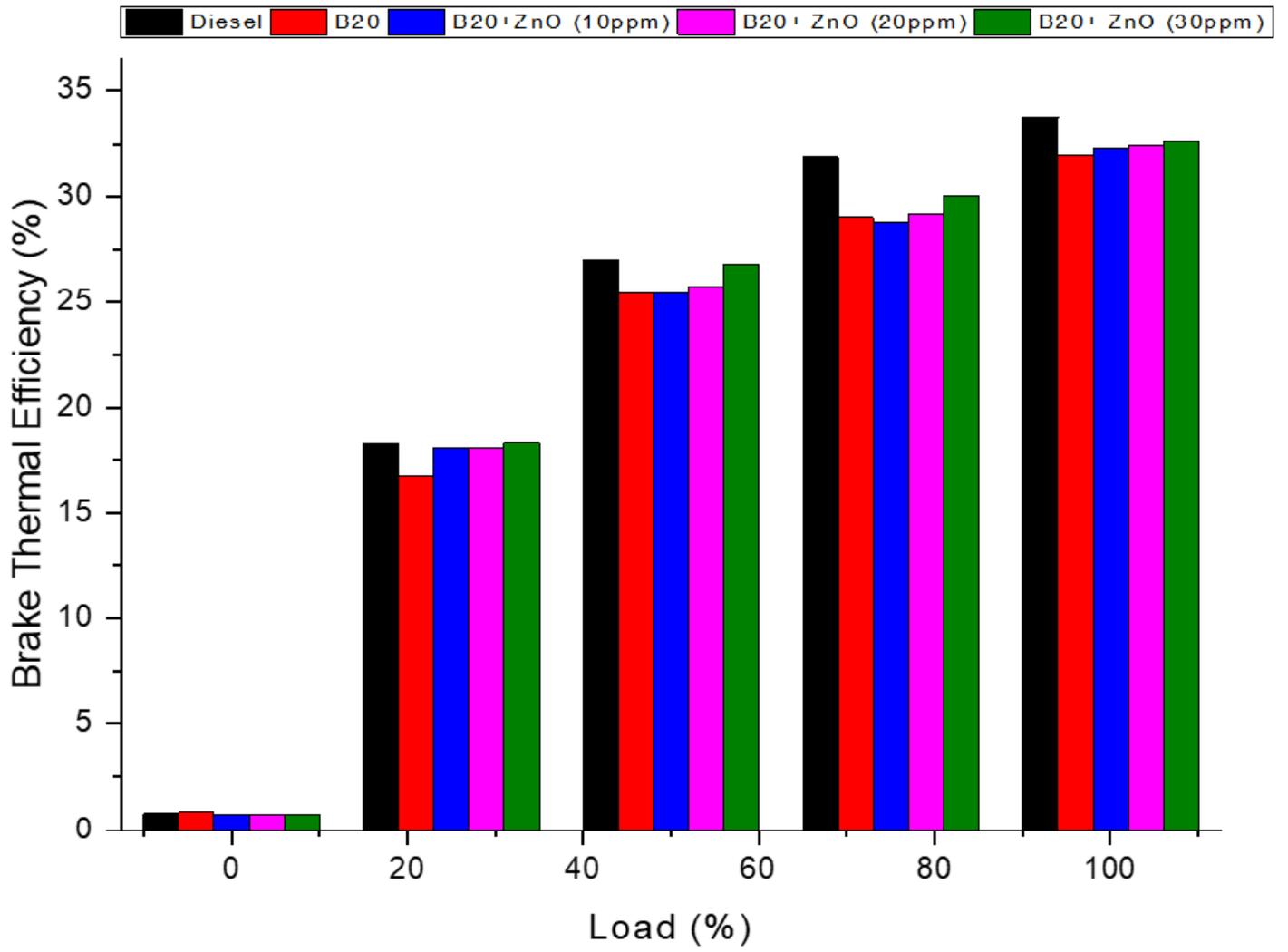


Figure 9

Brake Thermal Efficiency

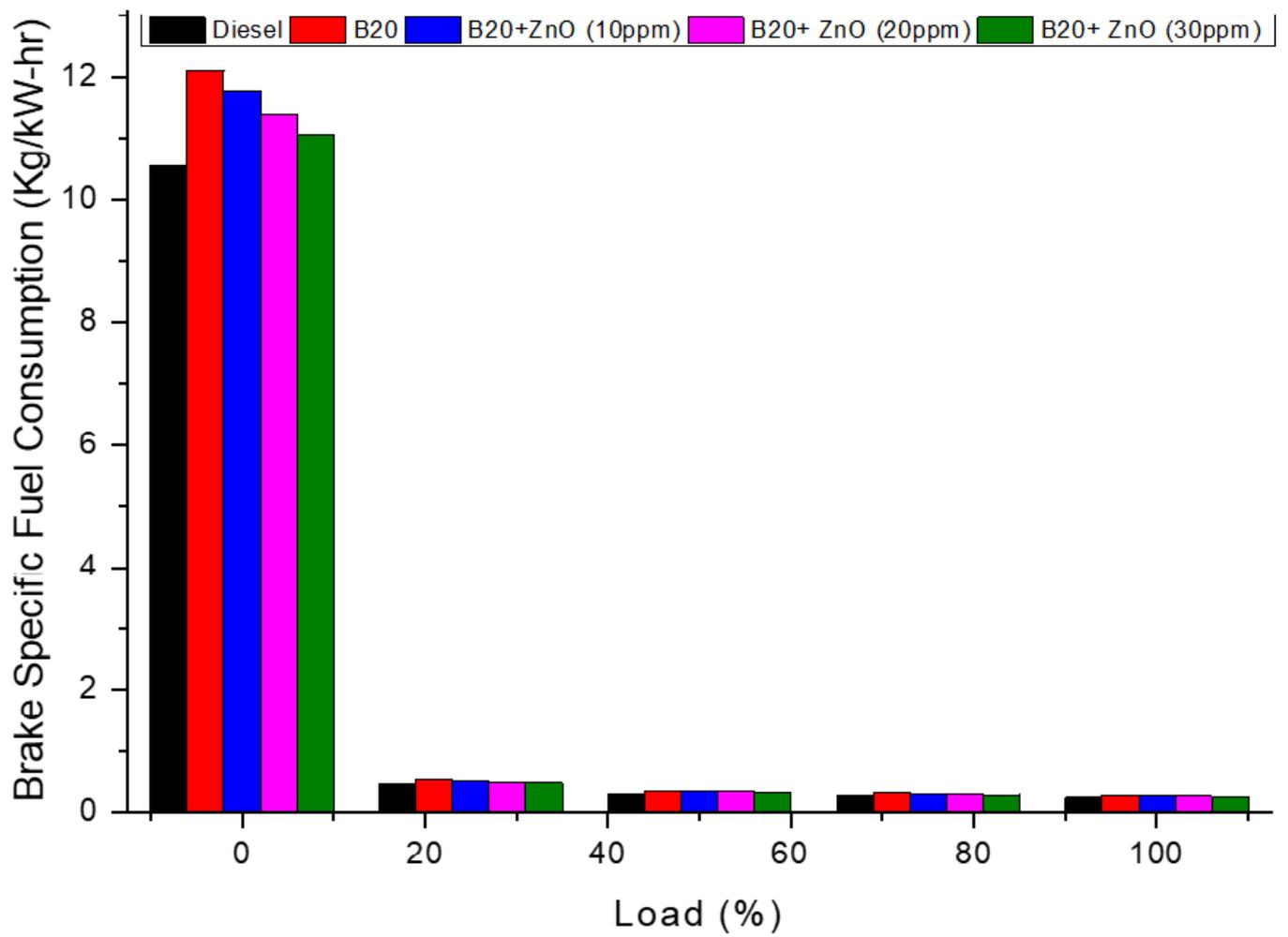


Figure 10

Brake Specific Fuel Consumption

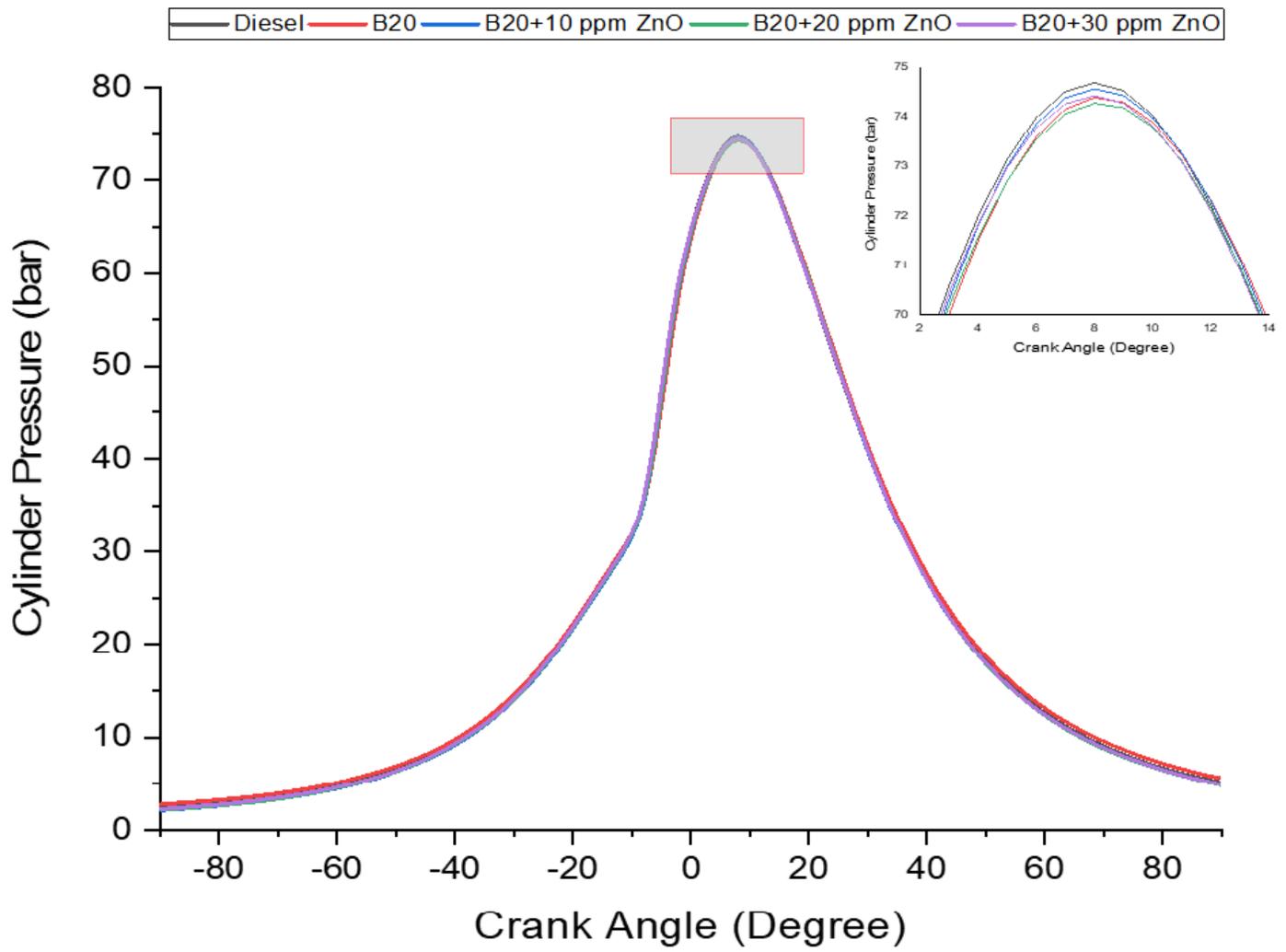


Figure 11

Brake Specific Fuel Consumption

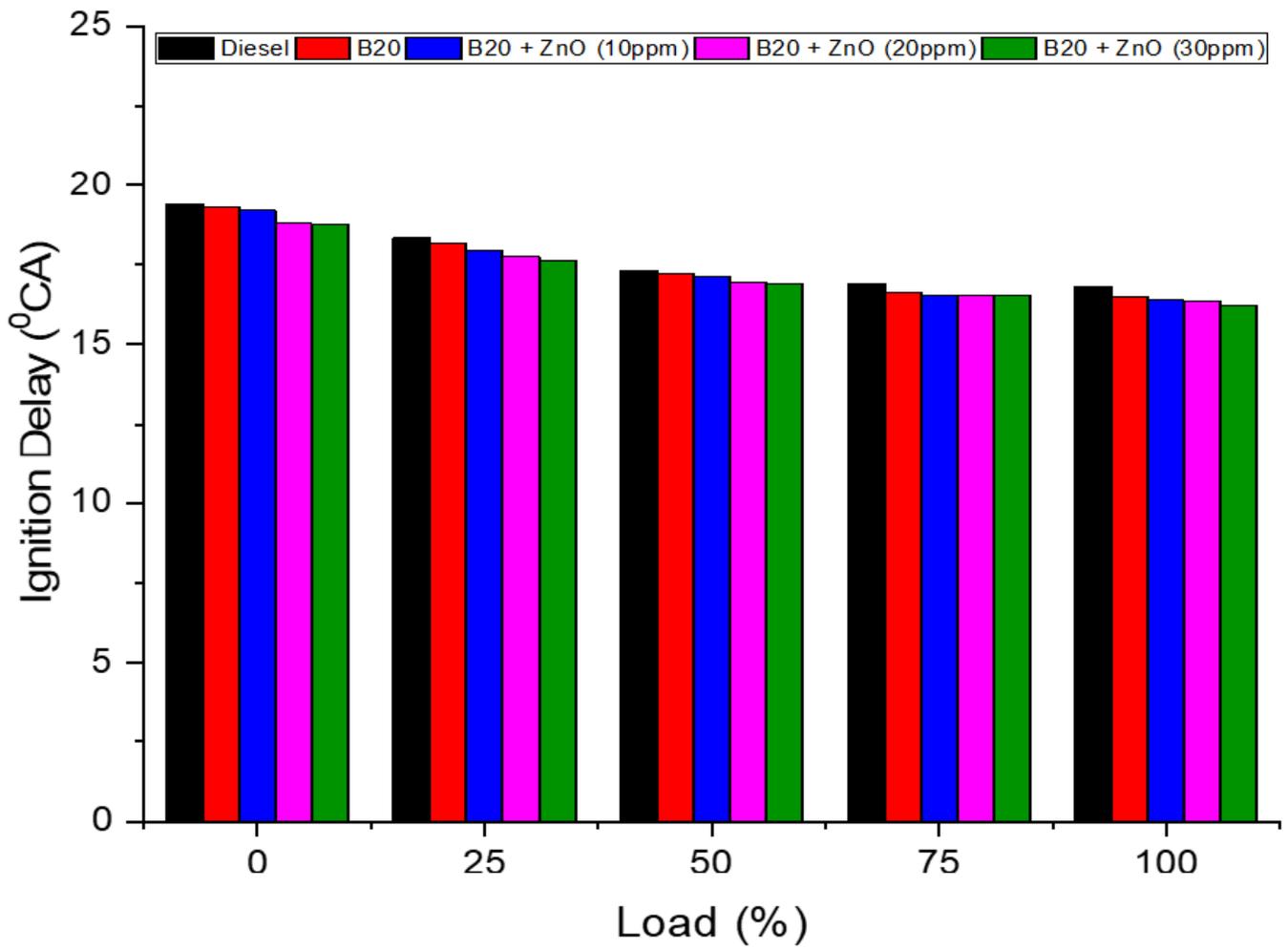


Figure 12

Ignition Delay

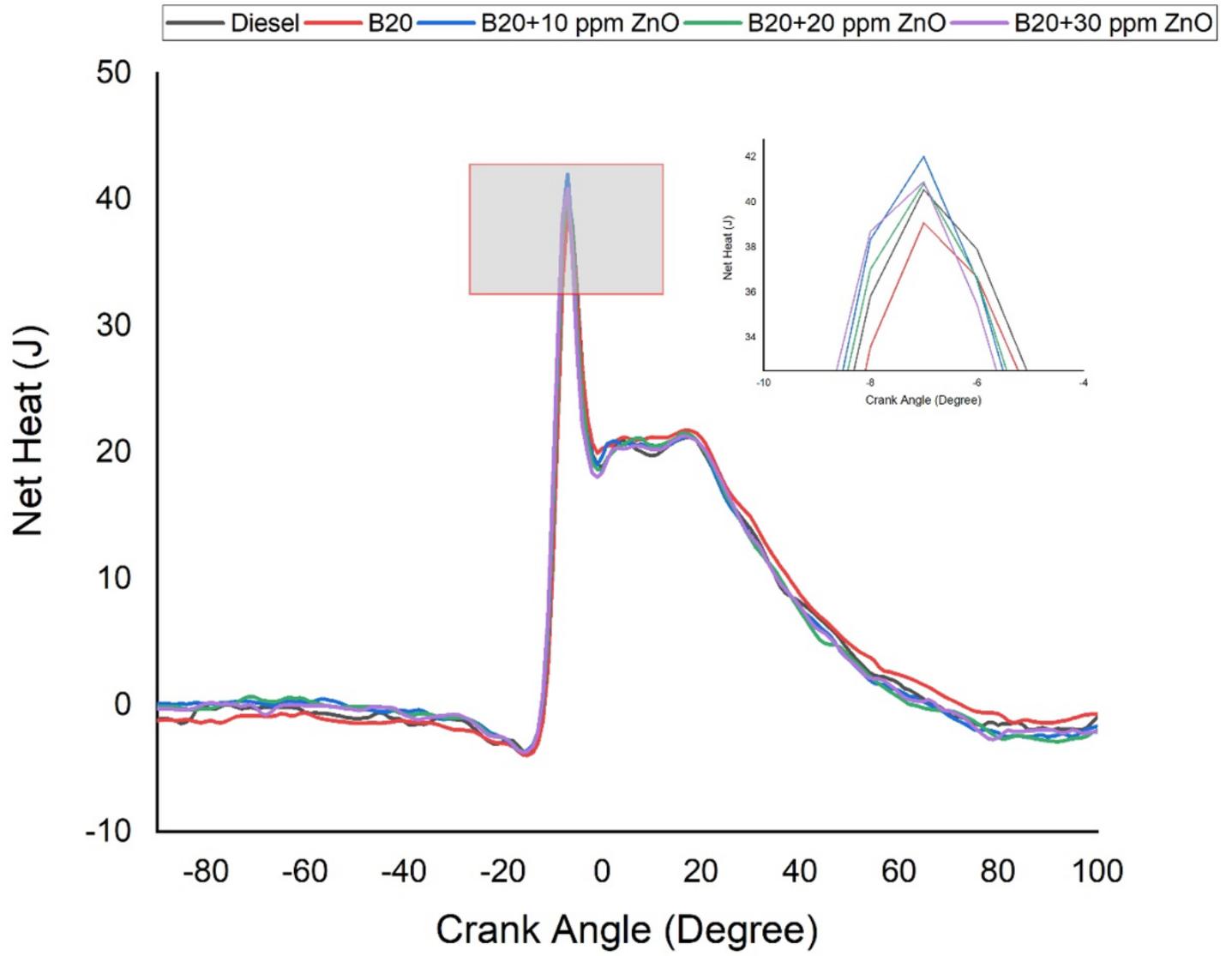


Figure 13

Net Heat Release Rate

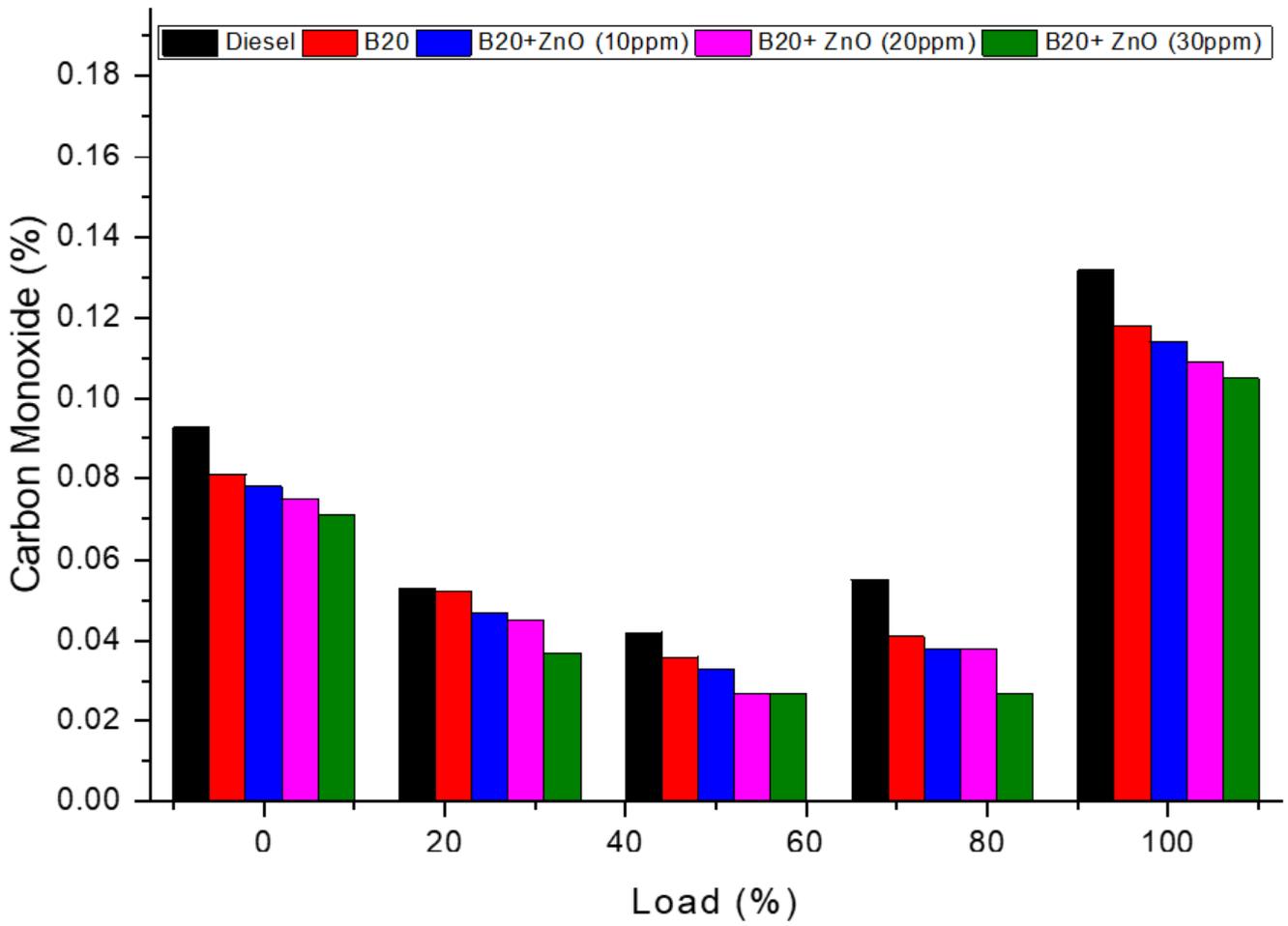


Figure 14

CO Emission

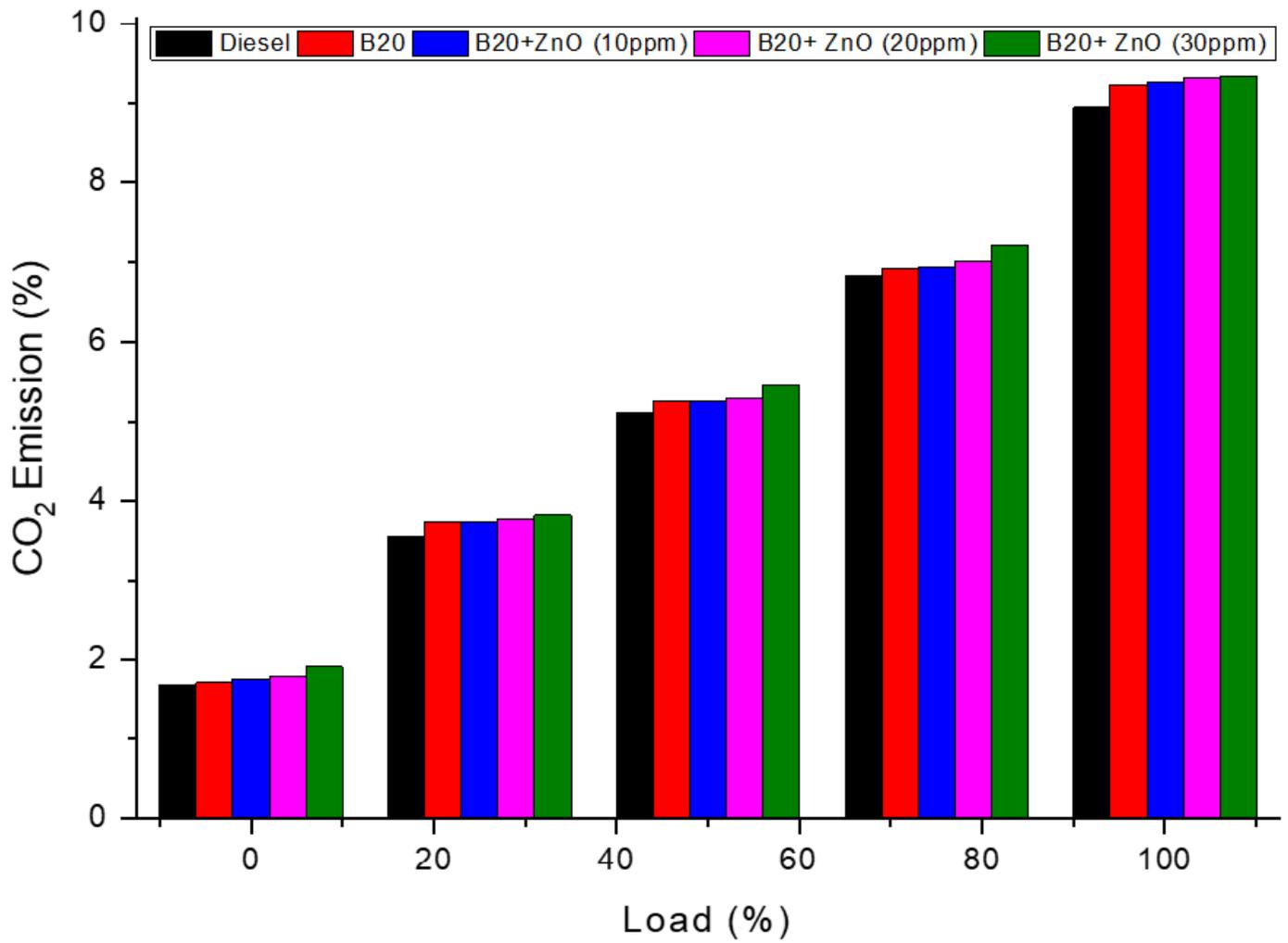


Figure 15

CO₂ Emission

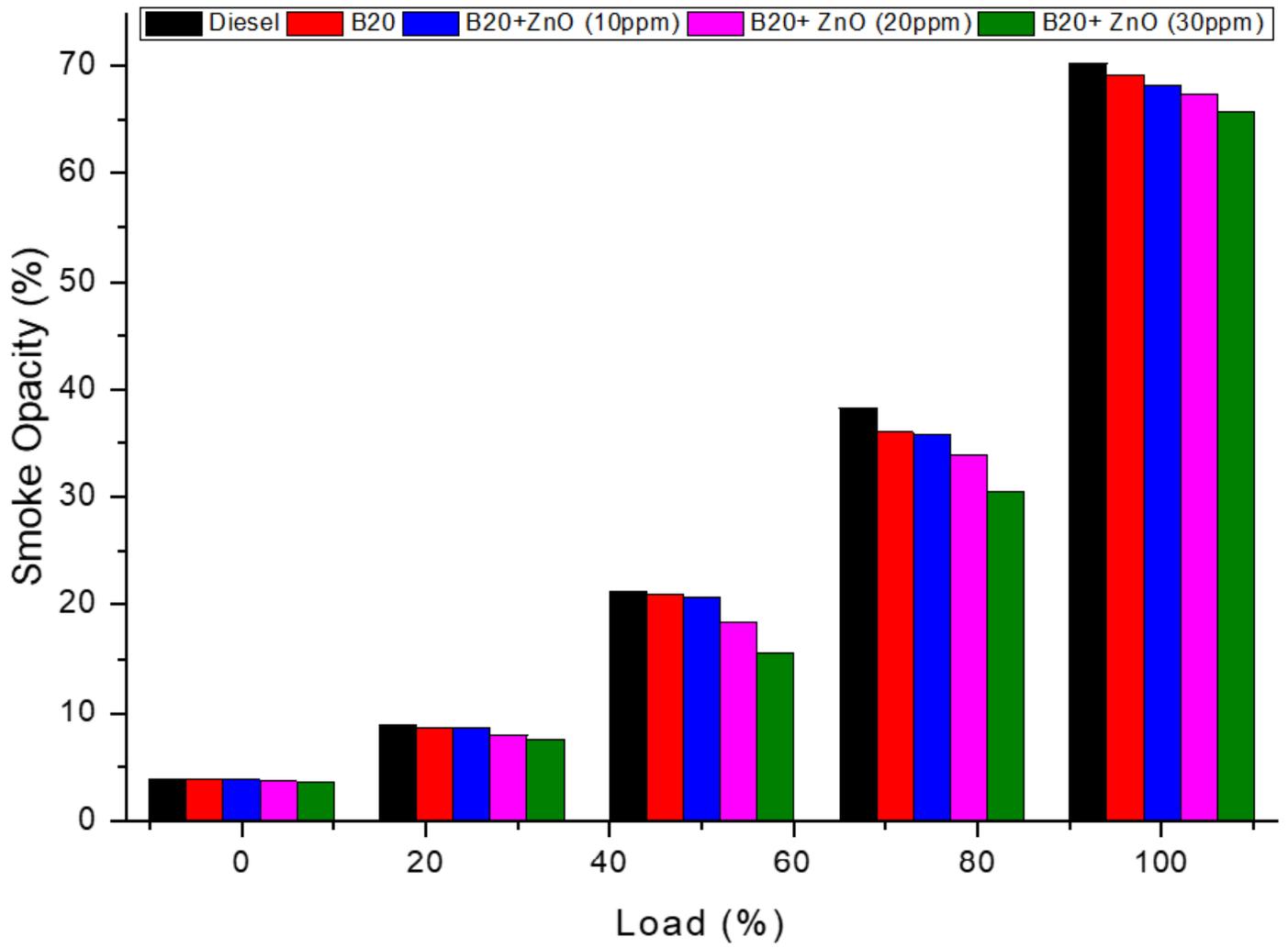


Figure 16

Smoke Opacity

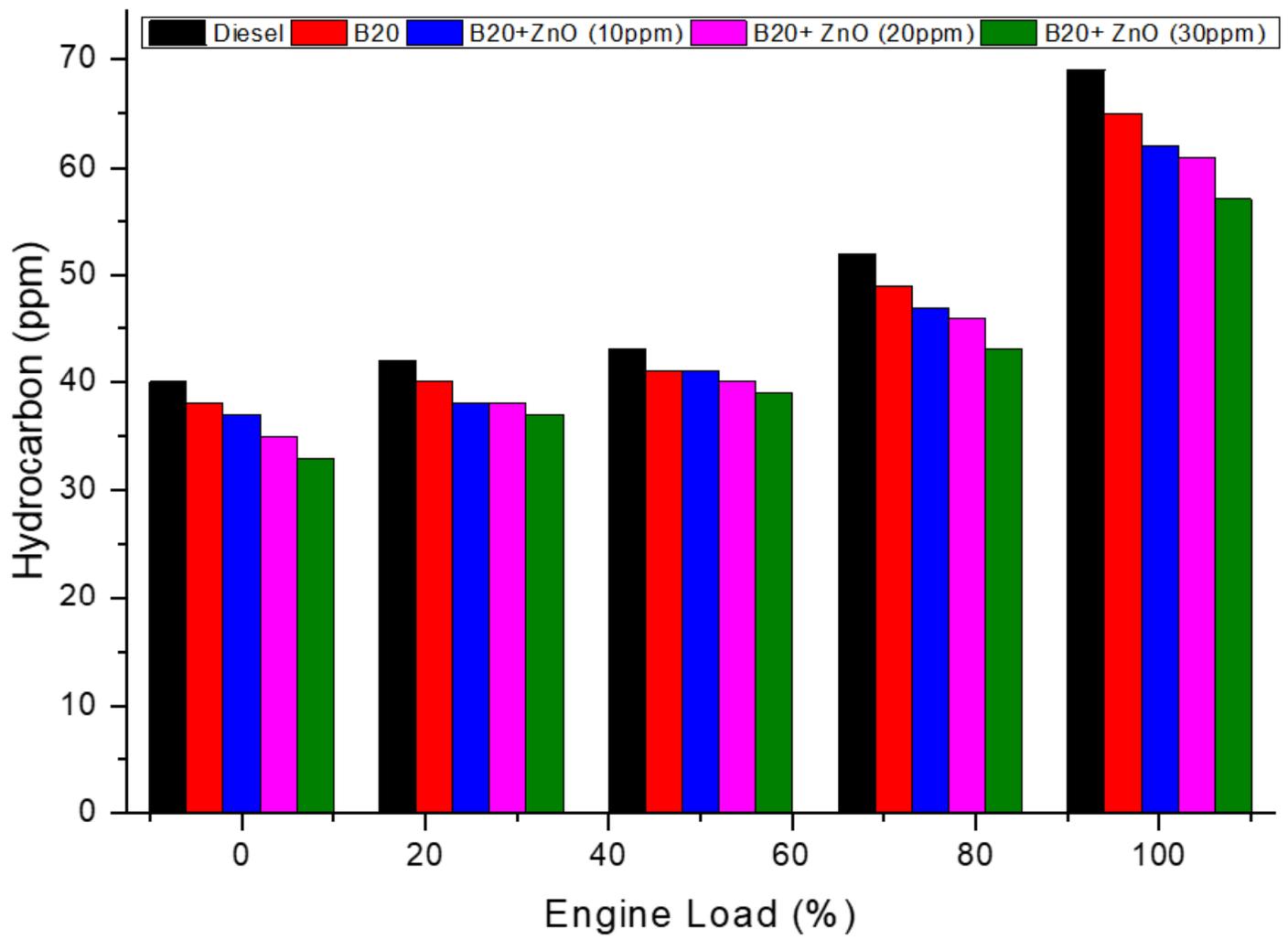


Figure 17

Hydrocarbon

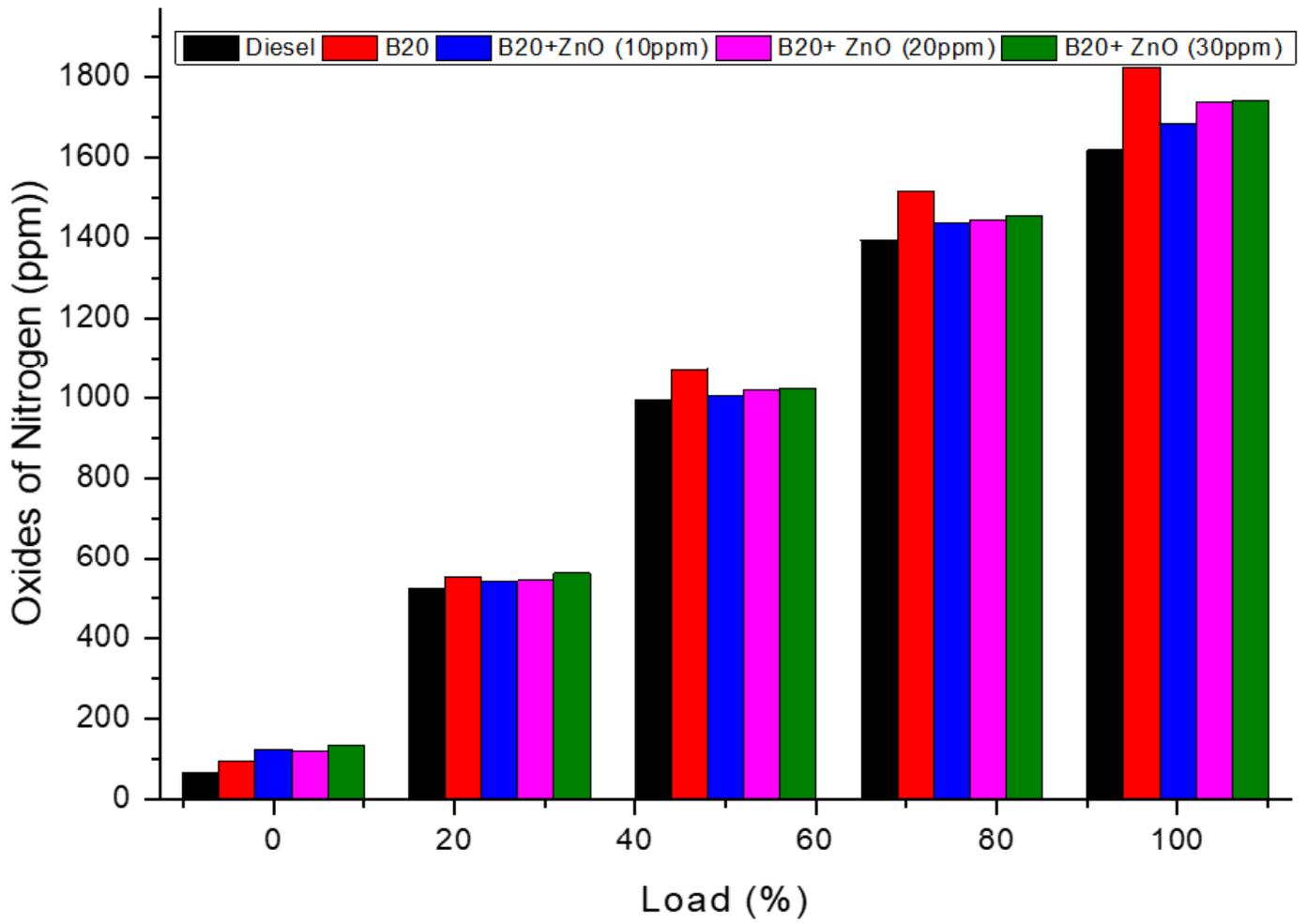


Figure 18

NOx Emission