

# Physical Protection in Aggregates and Organo-Mineral Associations Contribute to Carbon Stabilization at the Transition Zone of Seasonally Saturated Wetlands

Anna Kottkamp (✉ [akottkam@terpmail.umd.edu](mailto:akottkam@terpmail.umd.edu))

University of Maryland at College Park <https://orcid.org/0000-0001-6527-1942>

C Nathan Jones

The University of Alabama

Margaret A. Palmer

SESNYC: National Socio-Environmental Synthesis Center

Katherine L. Tully

University of Maryland at College Park <https://orcid.org/0000-0002-6190-2679>

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## Research Article

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1 Title: Physical protection in aggregates and organo-mineral associations contribute to carbon stabilization at the  
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4 Authors: Anna Kottkamp<sup>1</sup>, C. Nathan Jones, Margaret A. Palmer, Katherine L. Tully

5 Institutional Affiliations:

6 A.I. Kottkamp (ORCID 0000-0001-6527-1942)  
7 Department of Marine, Estuarine, and Environmental Sciences, University of Maryland, 1213 HJ Patterson  
8 Hall, College Park, MD 20742

9  
10 C. N. Jones  
11 Department of Biological Sciences, University of Alabama, 2098A Bevill Building, Tuscaloosa, AL 35487.

12  
13 M. A. Palmer  
14 National Socio-Environmental Synthesis Center, University of Maryland, 1 Park Place, Suite 300, Annapolis,  
15 MD 21401.

16  
17 K. L. Tully (Corresponding Author; 301-405-1469; kltully@umd.edu)  
18 Department of Plant Sciences and Landscape Architecture, University of Maryland, 2108 Plant Sciences  
19 Building, College Park, MD 20742.

20  
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28 **Abstract**

29 Wetlands store significant soil organic carbon (SOC) globally due to anoxic conditions that suppress SOC  
30 loss, yet this SOC is sensitive to climate and land use change. Seasonally saturated wetlands experience fluctuating  
31 hydrologic conditions that may also promote mechanisms known to control SOC stabilization in upland soils; these  
32 wetlands are therefore likely to be important for SOC storage at the landscape-scale. We investigated the role of  
33 physicochemical mechanisms of SOC stabilization in five seasonally saturated wetlands to test the hypothesis that  
34 these mechanisms are present, particularly in the transition between wetland and upland where soil saturation is  
35 most variable. At each wetland, we monitored water level and collected soil samples at five points along a transect  
36 from frequently saturated basin edge to rarely saturated upland. We quantified physical protection of SOC in  
37 aggregates and organo-mineral associations in mineral horizons to 0.5 m depth. As expected, SOC decreased from  
38 basin edge to upland. In the basin edge and transition zone, the majority of SOC was physically protected in  
39 macroaggregates. By contrast, overall organo-mineral associations were low, with the highest Fe concentrations (5  
40 mg Fe g<sup>-1</sup> soil) in the transition zone. While both stabilization mechanisms were present in the transition zone,  
41 physical protection is more likely to influence SOC stabilization during dry periods in seasonally saturated wetlands.  
42 As future climate scenarios predict changes in wetland wet and dry cycles, understanding the mechanisms by which  
43 SOC is stabilized in wetland soils is critical for predicting the vulnerability of SOC to future change.

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45

46 **1. Introduction**

47 Wetlands are a major component of the terrestrial carbon (C) cycle and sequester disproportionately large C  
48 stocks for their area (Mitsch et al. 2013). Given their C sequestration potential, wetland soils are the focus of current  
49 conservation and restoration efforts to help offset rising C emissions (Griscom et al. 2017). Research to date has  
50 primarily focused on soil organic C (SOC) of permanently saturated wetlands, where anoxia and subsequently slow  
51 decomposition are strong environmental drivers of SOC storage. However, a considerable yet frequently overlooked  
52 subset of wetlands are seasonally saturated and therefore experience dry, oxic conditions annually (e.g., vernal  
53 pools, Delmarva Bays, Prairie Potholes; Zedler 2003, Brooks 2005, Calhoun et al. 2017). Under the existing  
54 understanding of wetland SOC storage, periodic drying of seasonally saturated wetlands might be expected to  
55 stimulate C emissions and SOC loss (Miao et al. 2017). Contrary to this expectation, seasonally saturated wetland  
56 soils can sequester large stocks of SOC (e.g., Pearse et al. 2018; Tangen and Bansal 2020), though little is known  
57 about the mechanisms that support long-term SOC storage during dry periods in seasonally saturated wetlands. As  
58 wetting and drying cycles in wetlands are expected to become more extreme with changes in land use and climate  
59 (e.g., Fennessy et al. 2018; Lee et al. 2020), understanding the mechanisms of SOC storage in seasonally saturated  
60 wetlands is critical to predicting the vulnerability of SOC to future change.

61 Given the unique condition of annual drying in seasonally saturated wetlands, mechanisms that control  
62 SOC dynamics in upland (terrestrial) soils may also contribute to SOC stabilization during dry periods in these  
63 systems. Clay content has historically been used as proxy for SOC stabilization with mineral surfaces in upland soils  
64 (Keiluweit et al. 2018; Oades 1988) and has been assumed to also play a major role in wetland SOC sequestration  
65 (Daugherty et al. 2019). However, clay content alone does not clarify mechanisms of SOC stabilization, especially  
66 in seasonally saturated soils where C cycles are influenced by numerous hydrologic and biogeochemical factors  
67 (e.g., LaCroix et al. 2019). Beyond clay content, two physicochemical mechanisms of SOC stabilization are known  
68 to control long-term storage of SOC in upland soils: (1) physical protection of SOC within aggregates and (2)  
69 organo-mineral associations (Kaiser and Guggenberger 2000; Six et al. 2000; Moni et al. 2010; Schmidt et al. 2011).  
70 Because environmental conditions (i.e., anoxia) are dominant controls on SOC storage in saturated soils, the  
71 contribution of physicochemical mechanisms of SOC stabilization is understudied in wetlands. However, the  
72 fluctuating hydrologic and biogeochemical conditions of seasonally saturated wetlands warrant further study of  
73 physical protection of SOC in aggregates and organo-mineral associations in these dynamic ecosystems.

74 Physical protection of SOC in aggregates is a dominant control on SOC storage in upland soils. Aggregates  
75 are associations between soil particles such as clay, silt, and organic matter which may protect SOC for years to  
76 decades by limiting microbial access to C and reducing decomposition (Lehmann and Kleber 2015; Puget et al.  
77 2000). Aggregates may be physically separated into size classes associated with SOC of distinct ages and chemical  
78 composition: macroaggregates are a dynamic, younger fraction related to plant detritus, while microaggregates  
79 represent a more biologically processed, older fraction (Six et al. 2004; von Lützow et al. 2007, 2006). Research  
80 from terrestrial soils suggests that aggregate formation and stability are influenced by soil moisture and wet-dry  
81 cycles (Denef et al. 2001, 2002), organic inputs (Park et al. 2007), biological activity (Blankinship et al. 2016), and  
82 interactions with clay and mineral oxides (Wagner et al. 2007). Several studies indicate the potential for physical  
83 protection of SOC within aggregates in permanently saturated wetlands (e.g., Hossler and Bouchard 2010; Maietta  
84 et al. 2019), but the extent of aggregate-associated SOC in and around seasonally saturated wetlands is unknown.

85 Organo-mineral associations are a second type of physicochemical mechanism of SOC stabilization that  
86 has almost exclusively been studied in terrestrial soils (Kaiser and Guggenberger 2000). Strong interactions with  
87 iron (Fe) or aluminum (Al) can stabilize SOC for up to hundreds of years (Kleber et al. 2005; Kögel-Knabner et al.  
88 2008; Mikutta et al. 2006; Torn et al. 1997; Wagai and Mayer 2007). Associations between SOC and Fe depend on  
89 the type and amount of soil Fe, ranging from adsorption or coprecipitation of SOC with Fe (hydr)oxides to chelation  
90 (Kaiser and Guggenberger 2000; Mikutta et al. 2014; Sodano et al. 2017). Wetland soils tend to be depleted in  
91 mineral oxides due to reduction and subsequent leaching losses under saturated conditions (Mitsch and Gosselink  
92 2015). However, Fe (hydr)oxides may be retained in soils that are only intermittently saturated, such as the  
93 transition zone between seasonally saturated wetland and upland, thus promoting SOC stabilization via organo-  
94 mineral associations. Consequently, Fe and Al (hydr)oxides are better predictors of SOC levels than clay content in  
95 high moisture, acidic soils (Rasmussen et al. 2018). Thus, if Fe and Al (hydr)oxides are present in seasonally  
96 saturated wetlands, these associations could contribute to the stability of wetland SOC during dry periods.

97 Given the gaps in our understanding of how physicochemical mechanisms contribute to SOC storage  
98 during annual dry periods in seasonally saturated wetlands, our objectives were to (1) quantify total SOC and  
99 hydrologic characteristics, (2) examine the role of physical protection of SOC in aggregates, and (3) determine the  
100 extent of organo-mineral associations from the frequently saturated edge of the wetland basin to the rarely saturated  
101 upland. We investigated associations between SOC, aggregates, and Fe across a gradient of mean water level using

102 aggregate size fractionation and selective Fe extractions at five seasonally saturated wetlands. Across the wetland—  
103 upland gradient, we focus our study on the intermittently saturated transition zone, which dries annually with water  
104 table drawdown during the growing season. Transition zone soils contain intermediate SOC stocks compared to the  
105 basin and upland and are likely important sites for SOC storage on the landscape scale; however, the role of SOC  
106 stabilization mechanisms in transition zone soils has not been quantified (Webster et al. 2011; Fenstermacher 2012;  
107 LaCroix et al. 2019; Chanlabut et al. 2020). We hypothesized that both aggregate- and Fe-associated C would be  
108 greatest in the hydrologically dynamic transition zone. Given the annual drying of seasonally saturated wetland  
109 systems, we expect the physicochemical mechanisms of SOC stabilization studied here to be most relevant when  
110 under dry, oxic soil conditions, which may become increasingly prevalent under future climate change.

## 111 **2. Methods**

### 112 *2.1 Study sites*

113 This study was located on the Delmarva Peninsula in eastern U.S.A., which is bordered by the Chesapeake  
114 Bay on the west and the Atlantic Ocean on the east. Mean monthly temperature ranges from 1.3 °C (January) to 25.1  
115 °C (July). Mean annual precipitation is 1105 mm and is distributed evenly through the year (Supplementary Fig. 1).  
116 Study wetlands are on two properties of The Nature Conservancy which are < 5 km apart and contain numerous  
117 seasonally saturated freshwater wetlands. Water table fluctuations are primarily driven by seasonal  
118 evapotranspiration (Lee et al. 2020), with maximum surface water expression in late spring and wetland drawdown  
119 over the growing season from May to September (Brooks 2005).

120 We selected five seasonally saturated wetlands with similar mineral soils (Order: Ultisols, Suborders:  
121 Udults and Aquults), depressional shape (Fig. 1b), and vegetation. Given the acidity of the soils (Supplementary Fig.  
122 2), calcium carbonates were assumed to be negligible. Surrounding forest is comprised of *Acer rubrum*, *Quercus*  
123 *phellos*, *Liquidambar styraciflua*, and *Nyssa sylvatica* overstory with *Ilex opaca*, *Magnolia virginianica*, *Clethra*  
124 *alnifolia*, and *Vaccinium corymbosum* understory.

### 125 *2.2 Field methods*

#### 126 *2.2.1 Establishing sample points along transect at each wetland*

127           At each wetland, we established one 20–25 m transect with five evenly spaced sample points (4–5 m apart)  
128 extending from the frequently saturated “basin edge” (sample point A on the transect) to the rarely saturated  
129 “upland” (sample point E on the transect; Fig. 1a). The basin edge sample point was located at the minimum extent  
130 of ponded water observed during the 2017 water year, where A horizons were > 15 cm thick. The upland was  
131 designated as the elevated area with sandy soils around each wetland (Stolt and Rabenhorst 1987). Specifically, the  
132 upland sample point was located where (a) upland trees and understory vegetation were present and (b) no  
133 hydromorphic features were found in the upper 0.5 m of the soil (Supplementary Fig. 3).

### 134 *2.2.2 Characterizing hydrologic conditions at each sample point*

135           To characterize hydrologic conditions at sample points along the wetland transects, we used a combination  
136 of 1-year of continuous water level measurements, surveyed sample point elevations, and a simple interpolation  
137 procedure. Across all wetlands, we measured wetland and upland water levels at 15-minute intervals throughout  
138 water year 2018 (Oct 1, 2017–Sept 31, 2018) using HOBO U20 water level loggers (HOBO, Onset Computer  
139 Corporation, Bourne, MA). We measured wetland water levels in surface water stilling wells located near the center  
140 of each wetland, and we measured upland water level in wells located near the upland sample point (Fig. 1a). We  
141 also measured relative elevations of sample points along each transect using a TOPCON RL-H5A Laser Level  
142 (Topcon Positioning Systems, Livermore, CA). Finally, we estimated mean daily water level at each sample point  
143 by interpolating between the edge of surface water inundation and the upland water level in the well. Notably, this  
144 procedure accounts for the dynamic expansion and contraction of surface inundation along each transect. Further,  
145 while seasonally saturated wetlands may experience temporary groundwater mounding (Phillips and Shedlock 1993;  
146 Rosenberry and Winter 1997), this approach assumed linearity given the relatively small spatial scale and low-relief  
147 landscape. Interpolated water levels were comparable to observed water levels as measured biweekly from May to  
148 August 2018 ( $R^2 = 0.87$ , Supplementary Fig. 4). For each transect point at each wetland, we calculated the mean  
149 water level for water year 2018 from daily water levels ( $n = 365$  at each sample point). Higher, more positive values  
150 for mean water level indicate frequently saturated conditions at the basin edge, while lower, negative water levels  
151 indicate rarely saturated conditions at sample points towards the upland. A link to water level data, surveyed transect  
152 elevations, and interpolation procedure can be found in the Availability of Data and Material section.

153           We examined continuous hydrologic data (e.g., mean water level) instead of categorical transect points  
154 (e.g., A, B, etc.), as mean water level varied not only across the transect within wetlands but also at a given transect

155 point among wetlands. This allowed us to characterize the relationship between hydrologic conditions and  
156 physicochemical mechanisms of SOC stabilization (e.g., physical protection in aggregates and organo-mineral  
157 associations). In addition to mean water level, we quantified hydrologic variability as “saturation events” in the  
158 upper 0.5 m of soil by calculating the number of times that water level rose above a threshold (mean water level > -  
159 0.5 m) in water year 2018. To characterize reducing conditions, Fe oxide coated Indicator of Reduction In Soils  
160 (IRIS) films were deployed along each transect in April 2019 and analyzed after 30 days (Appendix A; Rabenhorst  
161 2018; Castenson and Rabenhorst 2006). In the upper 0.5 m of the soil profile, all metrics of hydrology and reducing  
162 conditions indicate frequently saturated conditions at sample points A and B, rarely saturated conditions at sample  
163 points D and E, and intermittently saturated conditions consistent with a transition zone near sample point C (Fig. 2;  
164 Supplementary Table 1).

### 165 2.2.3 Soil sampling

166 We used a gouge auger to collect soils to 1.0 m depth from sample points A–E along a transect at each of  
167 the five wetlands in June 2018. However, we focused our analysis on the first two mineral horizons from the upper  
168 0.5 m of soil as this zone experiences the largest fluctuations in water level (Fig. 2a) and contains higher SOC  
169 concentrations than deep soils. We elected to sample by pedogenic horizon instead of arbitrary depth limits given  
170 the differences in soil type between transect points. In general, the first mineral horizon was approximately 5–20 cm  
171 in depth and the second mineral horizon was approximately 20–45 cm in depth. O horizons were excluded from  
172 analysis as they were high in fresh organic matter inputs and low in mineral matrices necessary for stabilization.  
173 Three replicate cores from each of the A–E points were separated by horizon; replicates were combined and gently  
174 homogenized to obtain a representative sample for each horizon (Supplementary Fig. 3; Supplementary Table 2).  
175 Subsamples of 200 g were removed for aggregate analysis.

176 Additional samples for bulk density were collected using a McCauley peat sampler at sample points A and  
177 B and using a 3-inch diameter sharpened aluminum core at sample points C–E. Bulk density samples were separated  
178 into three depth increments (0–10 cm, 10–30 cm, and 30–50 cm) and corrected for compaction using a simple linear  
179 correction (Walter et al. 2016). Bulk density was calculated by dividing dry mass by core volume for each  
180 increment, subtracting roots and rocks (Poeplau et al. 2017).

### 181 2.3 Soil chemical analysis

182 Soil moisture levels can fluctuate daily in seasonally saturated wetlands, and, as it was not within the scope  
183 of the study to collect multiple soil samples through time, we decided to air dry all soils prior to analysis. Drying  
184 soils before aggregate analysis standardizes soils and allows soils to be stored while avoiding the effects of freezing  
185 on SOC. Standardization is necessary because aggregate stability in field-moist soils may be a function of  
186 antecedent water content (Gollany et al. 1991). However, we recognize that drying samples may promote the  
187 formation of stable macroaggregates (Beare and Bruce 1993) and novel Fe–C complexes following Fe<sup>2+</sup> oxidation  
188 (Kaiser et al. 2015). Therefore, by analyzing air-dried soils, we present the upper limit for both aggregate-associated  
189 C and Fe–C associations across the transect, to characterize the maximum potential contribution of physical  
190 protection and organo-mineral association to SOC stabilization in wetlands that dry seasonally.

191 Air dried soils were passed through a 2 mm sieve (excluding large roots and rocks) for analysis of pH,  
192 texture, C, nitrogen (N), and Fe. We measured pH using an Orion 9165BNWP Sure-Flow Combination pH probe on  
193 field moist soils with a 1:2 soil to solution (0.01 M CaCl<sub>2</sub>) ratio. Texture was determined by the hydrometer method  
194 (Gee and Bauder 1986). Bulk C (C<sub>bulk</sub>) and N were measured using dry combustion (LECO CHN-2000 analyzer;  
195 LECO Corp, St. Joseph, MI) with helium as the carrier gas and with a detection limit of 0.02% for total N and  
196 0.01% for total C.

### 197 *2.3.1 Physical protection of SOC determined by aggregate size class separation*

198 We examined water-stable aggregates on mineral horizons in the upper 0.5 m using the wet sieving  
199 protocol established by Six et al. (2000) and modified for wetland soils by Maietta et al. (2019). Across the transect,  
200 some soils were drier while others were saturated at the time of collection. Dry soils were passed through a 4.75 mm  
201 sieve within 24 h of sampling and dried for 7 days. Saturated soils were dried for 3 days at room temperature, passed  
202 through a 4.75 mm sieve, then dried at room temperature.

203 Soils were wet-sieved within two months of collection into four size classes: (1) large macroaggregates ( $\geq$   
204 2000  $\mu\text{m}$ ), (2) small macroaggregates (250–2000  $\mu\text{m}$ ), (3) microaggregates (53–250  $\mu\text{m}$ ), and (4) silt/clay particles (  
205  $< 53 \mu\text{m}$ ; includes only silt and clay sized particles not part of larger associations). Sieved samples were dried at  
206 room temperature with fans to evaporate excess water, then dried in a 65 °C oven for 7 days before weighing.  
207 Aggregate samples were then gently crushed, and roots were separated on a 1 mm sieve. Remaining mass for each  
208 size class was weighed and analyzed for C and N using dry-combustion (LECO CHN-2000 analyzer; LECO Corp,

209 St. Joseph, MI). We measured sand content for size classes  $> 53 \mu\text{m}$  on a 5 g subsample using the pipette method  
210 and subtracted sand mass to avoid overestimating aggregate mass (Appendix B; Elliott et al. 1991, Six et al. 2002).

### 211 *2.3.2. Organo-mineral associations between Fe–C determined by extractions*

212 We measured soil Fe concentrations using a dithionite-citrate-bicarbonate (DCB) extraction to estimate  
213 “total extractable” Fe oxides ( $\text{Fe}_{\text{DCB}}$ ) following the method by Darke and Walbridge (1994). A separate extraction  
214 with acid ammonium oxalate (AAO) was used to estimate “poorly crystalline” Fe species ( $\text{Fe}_{\text{AAO}}$ ; Coward et al.  
215 2018; Darke and Walbridge 1994; Hall et al. 2018). For both extractions, the supernatant was filtered through a  $0.45$   
216  $\mu\text{m}$  nylon filter (Tisch Scientific) and analyzed on an atomic absorption (AA) spectrometer on an air-acetylene flame  
217 (Perkin Elmer, Waltham, MA).

218 We also conducted inorganic Fe extractions on the mineral horizons to measure C associated with Fe  
219 (hydr)oxides (Darke and Walbridge 1994; Lopez-Sangil and Rovira 2013; Wagai et al. 2013) following the sodium  
220 dithionite-HCl (Dit-HCl) extraction procedure described for use in wetland soils by Maietta et al. (2019). The Dit-  
221 HCl extraction removes Fe oxides ( $\text{Fe}_{\text{Dit-HCl}}$ ) and is comparable to the DCB extraction of total extractable Fe oxides  
222 though it lacks an organic buffer, allowing us to measure concentrations of mineral-associated C in the extractant  
223 ( $\text{C}_{\text{Dit-HCl}}$ ; Appendix C). A preliminary DCB extraction of Al (Darke and Walbridge 1994) revealed low  
224 concentrations that were unchanged across the transect; thus Al was not explored further (mean  $\pm$  standard error:  $2.8$   
225  $\pm 0.3 \text{ mg Al g}^{-1}$  soil). As such, we use “organo-mineral associations” to refer to the relationship between C and Fe.

226 Extractions were conducted on a 0.5 g subsample of dried and ground soil. Extractants were filtered to  $0.45$   
227  $\mu\text{m}$  using a nylon filter (Tisch Scientific, North Bend, OH) and stored at  $4^\circ\text{C}$  until analysis of organic C  
228 concentration with a TOC/TN Analyzer (Shimadzu Corporation, Kyoto, Japan; Sugimura and Suzuki 1988). Fe  
229 concentrations were measured with AA spectrometry as above.

### 230 *2.4 Calculations and statistical approach*

231 We calculated the proportion of total C in bulk soil ( $\text{C}_{\text{bulk}}$ ) that was associated with each aggregate size  
232 class ( $\text{C}_{\text{aggregate}}/\text{C}_{\text{bulk}}$ ; Table 1, Eq. 1). We also calculated the proportion of  $\text{C}_{\text{bulk}}$  that was associated with extracted Fe  
233 and used this normalized value for analyses ( $\text{C}_{\text{Dit-HCl}}/\text{C}_{\text{bulk}}$ ; Table 1, Eq. 2). We tested the main effect of mean water  
234 level, where a higher mean water level indicates more frequently saturated soils towards the basin edge and a lower,

235 more negative water level indicates more rarely saturated soils towards the upland. We also tested the main effect of  
236 horizon by categorizing horizons in order of depth: first mineral horizon (1); second mineral horizon (2).

237 To test the effect of hydrology and horizon depth on SOC, physical protection in aggregates, and organo-  
238 mineral associations, we used linear mixed effects (LME) models with the *lmer* and *lmerTest* packages in R (Bates  
239 et al. 2015; Kuznetsova et al. 2017). In each test, mean water level (continuous) and horizon depth (categorical)  
240 were fixed effects, and we used backwards model selection to determine the best fixed effects structure (Zuur et al.  
241 2009). A random effect (wetland) was included as part of the study design (Barr et al. 2013). Assumptions of  
242 normality and homogeneity of variance of model residuals were checked graphically. All model results are in the  
243 Supplementary Information (Supplementary Tables 3–16).

244 Initial analysis revealed different patterns of Fe concentrations between frequently saturated basin edge  
245 soils (“frequently saturated;” mean water level > -0.5 m) and upland soils (“rarely saturated;” mean water level ≤ -  
246 0.5 m). Thus, for concentrations of Fe and associated C (e.g., Fe<sub>AAO</sub>, Fe<sub>DCB</sub>, Fe<sub>Dit-HCl</sub>, C<sub>Dit-HCl/C<sub>bulk</sub></sub>), we performed  
247 separate LME models on frequently saturated and rarely saturated soils to describe the effects of hydrology on  
248 organo-mineral associations.

249 We also used simple linear regression to test the relationship between SOC stocks and mean water level.  
250 Preliminary data exploration showed differences between 0–10 cm and 10–50 cm depth increments; therefore, we  
251 conducted separate analysis for SOC stocks from 0–10 cm and 10–50 cm. Initially, this LME model included  
252 wetland as a random effect to account for differing wetland intercepts. However, the random term did not  
253 significantly improve the model (log-likelihood ratio test; Zuur et al. 2009; P > 0.05). Therefore, we removed the  
254 random term and conducted a simple linear regression.

255 Statistical analysis was conducted using R statistical software v3.5.3 (R Development Core Team). Plots  
256 were created with the *tidyverse* and *cowplot* packages (Wickham 2017; Wilke 2019).

### 257 **3. Results**

#### 258 *3.1 High hydrologic variability at sample points with a mean water level near -0.5 m*

259 Temperature in water year 2018 was within the 30-year mean (mean monthly range: -1 °C in January to 26  
260 °C in July 2018; Supplementary Fig. 1). Precipitation over the year was 1366 mm, which is higher than average due  
261 to high precipitation in July and September 2018. Water levels varied ~1 m over the year at each sample point,

262 indicating dynamic hydrology typical for seasonally saturated wetlands of the region (Fig. 2a). Water levels rose  
263 from November to February and remained high until May. Water levels fell from June to September 2018 but rose  
264 in response to storm events in July and September.

265 Saturation within the upper 0.5 m of the soil profile was most variable at sample points with a mean water  
266 level of -0.5 m (Fig. 2b) compared to rarely saturated sample points with lower (more negative) mean water levels  
267 (e.g., yellow and green lines, Fig. 2a) or compared to frequently saturated sample points with higher (more positive)  
268 mean water levels (e.g., dark blue lines, Fig. 2a). Accordingly, low Fe oxide reduction on IRIS films at sample point  
269 C (Supplementary Table 1) indicates heterogeneity of reducing conditions at the transition between basin edge and  
270 upland.

### 271 *3.2 SOC concentrations greatest at basin edge and declined toward rarely saturated upland*

272 Soil  $C_{\text{bulk}}$  concentration decreased as water levels declined from basin edge to upland (i.e., rarely saturated  
273 soils;  $P = 0.001$ , Supplementary Table 3; Fig. 3a). With increasing horizon depth,  $C_{\text{bulk}}$  concentrations decreased, as  
274 is typical for soils not affected by fluvial processes ( $P < 0.0001$ , Supplementary Table 3). However, we observed  
275 contrasting relationships between SOC stocks and mean water level. In the top 0–10 cm, SOC stocks increased as  
276 water levels declined toward rarely saturated soils ( $P = 0.001$ , Supplementary Table 4; Fig. 3b). Conversely, in the  
277 10–50 cm depth increment, SOC stocks decreased with decreasing mean water level ( $P < 0.0001$ , Supplementary  
278 Table 4; Fig. 3c).

279 Bulk soil clay (%) decreased with decreasing mean water level ( $P < 0.0001$ , Supplementary Table 5;  
280 Supplementary Fig. 2) while percent sand tended to increase. For soil pH, there was a significant interaction  
281 between horizon and mean water level. As mean water level decreased, soil pH decreased in the first mineral  
282 horizon but increased in the second mineral horizon ( $P = 0.0017$ , Supplementary Table 6; Supplementary Fig. 2).

### 283 *3.3 Physical protection of SOC in macroaggregates is greatest at the frequently saturated basin edge*

284 Among size classes, small and large macroaggregates contained the highest  $C_{\text{aggregate}}$  concentrations, while  
285 microaggregates and silt/clay contained lower  $C_{\text{aggregate}}$  concentrations ( $P < 0.0001$ , Supplementary Table 7). In the  
286 large macroaggregate size class only,  $C_{\text{aggregate}}$  significantly decreased as water levels declined toward rarely  
287 saturated soils ( $P = 0.00098$ , Supplementary Table 8; Fig. 4a–d).

288 Normalized  $C_{\text{aggregate}}/C_{\text{bulk}}$  proportions ( $\text{mg C g}^{-1}$  bulk C; Table 1, Eq. 1) decreased with decreasing water  
289 level toward rarely saturated soils in the large macroaggregates but increased in all smaller size classes ( $P < 0.05$  for  
290 all tests, Supplementary Table 9; Fig. 4e–h). Among size classes, normalized  $C_{\text{aggregate}}/C_{\text{bulk}}$  proportions were highest  
291 in small macroaggregates and lowest in the smallest size classes ( $P < 0.0001$ ; Supplementary Table 7). Normalized  
292  $C_{\text{aggregate}}/C_{\text{bulk}}$  proportions declined in the deeper horizon in large macroaggregates but increased in microaggregates  
293 ( $P < 0.05$ , Supplementary Table 9; Fig. 4e).

#### 294 *3.4 Fe-associated C highest in rarely saturated soils, declined sharply toward the basin edge*

295 We conducted separate statistical analysis for frequently saturated soils (mean water level  $> -0.5$  m) and  
296 rarely saturated soils to understand the effect of mean water level on Fe species. In frequently saturated soils, Fe  
297 concentrations of all extracted species were low ( $< 0.5$  mg Fe  $\text{g}^{-1}$  soil; Fig. 5a; Supplementary Fig. 5). In rarely  
298 saturated soils,  $\text{Fe}_{\text{Dit-HCl}}$  concentrations were highest where mean water level was approximately 0.5 m below the soil  
299 surface (transition zone), but concentrations declined toward the upland ( $P < 0.0001$ , Supplementary Table 10; Fig.  
300 5a). We observed similar patterns in poorly crystalline  $\text{Fe}_{\text{AAO}}$  and total extractable  $\text{Fe}_{\text{DCB}}$  concentrations  
301 (Supplementary Table 11; Supplementary Table 12; Supplementary Fig. 5). Across the entire transect, the  
302 proportion of total extractable Fe oxides that are poorly crystalline ( $\text{Fe}_{\text{AAO}}/\text{Fe}_{\text{DCB}}$ ) was highest at the basin edge and  
303 decreased linearly towards upland soils ( $P < 0.0001$ , Supplementary Table 13; Supplementary Fig. 5).

304 Fe-associated C followed similar patterns to the extractable Fe species and was also analyzed separately for  
305 frequently and rarely saturated soils (Fig. 5b). In frequently saturated soils, Fe-associated  $C_{\text{Dit-HCl}}$  concentrations  
306 showed no significant changes with decreasing water level. In rarely saturated soils, Fe-associated  $C_{\text{Dit-HCl}}$   
307 concentrations decreased as mean water levels declined ( $P = 0.00035$ , Supplementary Table 14).

308 To understand the relative amount of total C associated with Fe across the basin edge–upland gradient, we  
309 calculated the normalized  $C_{\text{Dit-HCl}}/C_{\text{bulk}}$  proportion ( $\text{mg C g}^{-1}$  bulk C; Table 1, Eq. 2). When separated into frequently  
310 and rarely saturated soils, there was no significant association between mean water level and the proportion of  $C_{\text{Dit-}}$   
311  $\text{HCl}/C_{\text{bulk}}$  ( $P > 0.05$ , Supplementary Table 15; Fig. 5c). However, when we compared  $C_{\text{Dit-HCl}}/C_{\text{bulk}}$  proportions across  
312 frequently versus rarely saturated soils as separate groups, a greater proportion of bulk soil C was associated with Fe  
313 in rarely saturated soils than frequently saturated soils (Welch’s two sample t-test,  $P < 0.05$ ).

#### 314 4. Discussion

315 Seasonally saturated wetlands play an important role in SOC storage and stabilization (Pearse et al. 2018;  
316 Chanlabut et al. 2020), yet the mechanisms underpinning these processes during seasonally dry, oxic conditions are  
317 understudied. Here we show the potential for SOC in and around seasonally saturated wetlands to be stabilized  
318 during dry periods by physical protection within macroaggregates and, to a lesser extent, by organo-mineral  
319 associations with Fe. Specifically, the proportion of total C within large macroaggregates was highest at the  
320 frequently saturated wetland basin edge, suggesting that macroaggregates are likely to physically protect wetland  
321 SOC during seasonal drying. In contrast, Fe-associated C was low across the transect, peaking in the transition zone  
322 where saturation is most dynamic throughout the year. Overall, this research highlights the potential role of  
323 physicochemical mechanisms of SOC stabilization in wetlands that dry annually, indicating the need for research on  
324 how these mechanisms influence the resilience of SOC to future change.

##### 325 *4.1 SOC changes across the gradient from basin edge–upland*

326 Our results demonstrate that mean water level can be used as a continuous indicator of hydrologic  
327 conditions to examine linear and nonlinear trends in soil characteristics across wetlands, building on prior studies  
328 that study SOC across discrete topographic categories (e.g., Chanlabut et al. 2020; LaCroix et al. 2019; Pearse et al.  
329 2018; Webster et al. 2011; Webster et al. 2008). Low relief, wet forests are ubiquitous throughout the eastern U.S.,  
330 but given their complexity they have typically been neglected in forest C models (e.g., Hurtt et al. 2019) and may be  
331 poorly represented by upland SOC models (Trettin et al. 2001). As remote sensing and geospatial methods improve  
332 wetland inundation detection (e.g., Lang and McCarty 2009; Lang et al. 2013; Evenson et al. 2018), relationships  
333 between SOC and hydrology could lead to more accurate models of SOC stocks in seasonally saturated wetlands.

334 Surprisingly, we found contrasting trends by depth across the hydrologic gradient, wherein SOC stocks of  
335 surface soils (0–10 cm) were highest in the rarely saturated upland, while SOC stocks from 10–50 cm were greatest  
336 in frequently saturated soils of the basin edge (Fig. 3b–c). As expected, wetland SOC stocks from 10–50 cm  
337 decreased from basin edge to upland. In the lower depth increment, we anticipate that these strong trends in SOC  
338 storage are predominately due to trends in saturation and anoxia across the hydrologic gradient, as all metrics  
339 studied here indicate saturation and reducing conditions increase toward the basin edge (Supplementary Table 1).  
340 However, SOC stocks from 0–10 cm increased from basin edge to upland, which we anticipate is due to higher root  
341 growth in shallow organic horizons of upland soils than in the basin edge (LaCroix et al. 2019).

342 The trends in SOC stocks observed here suggest that the overall relationship between hydrology and SOC  
343 is a continuum and not a binary (e.g., wetland vs. upland). We observed greater SOC stocks from 10–50 cm in basin  
344 edge and transition zone soils than upland soils (Fig. 3). Our findings reinforce the SOC storage potential of  
345 intermittently flooded mineral soil wetlands found by Chanlabut et al. (2020) in which intermittently flooded soils  
346 contained higher SOC than permanently flooded soils. While the transition zone makes up a small portion of the  
347 area of an individual wetland, the transition zones of many seasonally saturated wetlands may contribute to  
348 important biogeochemical functions on the landscape scale (although see Tangen and Bansal 2020) and therefore  
349 must be included to improve wetland SOC models and expand targets of wetland conservation for SOC storage  
350 (e.g., Bridgham et al. 2006).

351 We observed strong relationships between mean water level and SOC characteristics, though soil texture  
352 also changed across the transect. Clay content has been used as a proxy for SOC stabilization (Oades 1988), as SOC  
353 may be protected from microbial decomposition by being bound to clay surfaces or sequestered in aggregates under  
354 anaerobic conditions (Weil and Brady 2017). However, bulk soil C was not correlated with percent clay along the  
355 transects studied here (Supplementary Fig. 6); instead, we found relationships between macroaggregate content and  
356 SOC (Fig. 4). Thus, our method of examining aggregate size classes can distinguish between the relative importance  
357 of SOC stabilization via free clay mineral surfaces (silt/clay fraction) and physical protection within aggregates  
358 (micro- and macroaggregate fractions). Our results suggest that percent clay alone as a proxy for C stabilization may  
359 not capture spatial variability of SOC in seasonally saturated wetlands. Rather, mean water level may be a better  
360 proxy for SOC in seasonally saturated wetlands due to its effects on anoxic conditions and SOC stabilization  
361 mechanisms in these systems.

#### 362 *4.2 Large macroaggregates contain the most SOC in basin edge soils*

363 Large macroaggregates (> 2 mm diameter) contained the most SOC across the transect and were related to  
364 hydrologic gradients, indicating that physical protection of SOC in large macroaggregates contributes to SOC  
365 stabilization in soils at the basin edge. Both C concentration ( $C_{\text{aggregate}}$ ) and the proportion of total C ( $C_{\text{aggregate}}/C_{\text{bulk}}$ )  
366 in large macroaggregates were highest in basin edge soils and were intermediate in the transition zones (Fig. 4).  
367 Large macroaggregates were greatest in surface soils, consistent with the expectation that large macroaggregates  
368 incorporate fresher organic inputs and are the most dynamic aggregate size class (Six et al. 2004). Because soils  
369 were dried prior to analysis for standardization, these results represent an upper limit for macroaggregate-associated

370 C in sampled wetlands. While few studies have quantified macroaggregates in mineral wetland soils, Cui et al.  
371 (2019), Maietta et al. (2019), and Hossler and Bouchard (2010) found that large macroaggregates store significantly  
372 more SOC than other aggregate size classes in organic wetland soils. Additionally, Maietta et al. (2019) found that  
373 macroaggregate C was highest in the lowest elevation marsh sample point, consistent with the increase in large  
374 macroaggregates toward the frequently saturated basin edge found in the present study. We note, however, that  
375 several of the wettest soils studied here did not contain large macroaggregates; therefore, we expect that hydrology,  
376 in addition to other factors such as root activity and microbial biomass (e.g., Blankinship et al. 2016), contribute to  
377 aggregate stability in wetland soils.

378         Physical protection of wetland SOC within macroaggregates may be promoted by the influx of organic-rich  
379 water and frequently saturated conditions at the basin edge. The influx of C-rich water provides organic constituents  
380 that may increase macroaggregate stability by increasing sorption between organic matter and minerals at aggregate  
381 surfaces and by strengthening intra-aggregate structures through the diffusion of DOC (Park et al. 2007). Therefore,  
382 we posit that relatively high dissolved organic C concentrations in the study wetlands (30 mg C L<sup>-1</sup>; Hosen et al.  
383 2018) promote aggregate stability. Soil saturation also influences aggregate formation and stability. Research from  
384 agricultural soils indicates that aggregates tend to increase under moderately high soil moisture compared to dry  
385 conditions (Blankinship et al. 2016). Further, multiple cycles of saturation and drying also tend to increase  
386 aggregates (Denef et al. 2001, 2002), which may be due to stronger interactions between organic matter and clay  
387 particles (Wagner et al. 2007), changes in the microbial community over time, or increases in macroaggregate  
388 cohesion (Cosentino et al. 2006). It is unclear how this extensive body of research on aggregates in agricultural soils  
389 applies to forest and wetland soils, as the wet-dry cycles in these studies are typically around 14 days (consistent  
390 with irrigation and subsequent drying), and agricultural soils are typically much drier on average than forested  
391 wetland soils. The influence of longer (annual) hydrologic cycles, such as those of seasonally saturated wetlands, is  
392 a critical area for new research on aggregates, as intact aggregates may mitigate the effects of wet-dry cycles and  
393 reduce C emissions during seasonally dry conditions (Navarro-García et al. 2012).

394         We found that the three smaller aggregate size classes were associated with the majority of SOC in rarely  
395 saturated, upland soils but were less responsive to the hydrologic gradient than large macroaggregates. The  
396 proportion of bulk soil C associated with smaller aggregate size classes ( $C_{\text{aggregate}}/C_{\text{bulk}}$ ) increased from basin edge to  
397 upland, but C concentration in smaller size classes did not change across the gradient (Fig. 4). As expected, SOC

398 associated with smaller aggregate size classes contained lower C concentrations and C:N than macroaggregates,  
399 indicating that microaggregates and silt/clay fractions consist of older, microbially derived C (Supplementary Table  
400 12; Jastrow 1996; Six et al. 2000). Notably, microaggregates likely contribute to long-term SOC stabilization  
401 through strong adsorption reactions on mineral surfaces, underscoring both the importance of smaller aggregate size  
402 classes to long-term physical protection of SOC as well as the connection between aggregation and organo-mineral  
403 associations (Arachchige et al. 2018; Totsche et al. 2018).

404 *4.3 Organo-mineral associations were relatively low across the transect, with minor accumulations in intermittently*  
405 *saturated transition zone soils*

406 Overall, total extractable Fe concentrations in our study wetlands are on the low end (0–5 mg Fe g<sup>-1</sup> soil) of  
407 the typical range for wetlands in the Northeast and mid-Atlantic (0–12 mg Fe g<sup>-1</sup> soil; National Cooperative Soil  
408 Characterization Database; LaCroix et al. 2019); thus, our results suggest a relatively minor potential for organo-  
409 mineral associations in transition zone and upland soils that is constrained by low Fe concentrations. As we dried  
410 soils prior to analysis, these values represent the maximum levels of Fe that we would expect to find in the sampled  
411 soils, regardless of saturation level. Unsurprisingly, we observed the lowest total extractable Fe concentrations in  
412 frequently saturated soils (< 1 mg Fe g<sup>-1</sup> soil), where saturated conditions promote reduction and translocation of Fe  
413 (Chen et al. 2017; Fiedler and Sommer 2004). In contrast, modest accumulations of total extractable Fe (up to 5 mg  
414 Fe g<sup>-1</sup> soil) were observed in the transition zone. Approximately 50% of the Fe oxides in the transition zone were  
415 poorly crystalline (Fe<sub>AAO</sub>/Fe<sub>DCB</sub>) and therefore more likely to have increased surface area available for transient  
416 association with C (Fig. 3; Supplementary Fig. 5; Hall et al. 2018; Coward et al. 2017). We expect that minor  
417 accumulations of Fe in transition zone soils are the result of periodic groundwater mounding that causes Fe to flow  
418 from uplands towards ponds (Phillips and Shedlock 1993) as well as seasonally dry conditions in upper transition  
419 zone soils that promote Fe (hydr)oxide retention. Additionally, research in permafrost soils suggests that vertical  
420 translocation may promote Fe accumulation at redox interfaces such as the upper soils of the transition zone  
421 (Herndon et al. 2017).

422 Constrained by low Fe (hydr)oxide levels, Fe-associated C concentrations were low overall (Fig S5) but  
423 were highest in the transition zone and decreased linearly toward the upland (C<sub>Di-HCl</sub>; Fig. 5b). Our results add to  
424 existing evidence that Fe may associate with SOC in places where Fe is present, such as redox interfaces like the  
425 transition zone (e.g., Riedel et al. 2013). While Fe-associated C concentrations were highest in the transition zone,

426 the proportion of total SOC that was Fe-associated ( $C_{\text{Dit-HCl}}/C_{\text{bulk}}$ ) was highest in the upland, indicating that a greater  
427 fraction of SOC is in stabilizing organo-mineral associations in upland soils (up to 30%; Fig. 5c). Of note, high  
428 proportions of poorly crystalline Fe and high ratios of C:Fe in all samples ( $> 1$ , Supplementary Table 16) indicate  
429 that most associations between C and Fe are transient and therefore susceptible to dissolution under seasonally  
430 saturated conditions when anoxic conditions are instead expected to drive SOC storage and stability (Kaiser and  
431 Guggenberger 2007; Wagai and Mayer 2007).

432         Organo-mineral associations in the frequently saturated basin soils were even lower ( $< 1 \text{ mg Fe g}^{-1} \text{ soil}$ )  
433 than in transition zone and upland soils, and only 10% of bulk soil C was associated with  $\text{Fe}_{\text{Dit-HCl}}$  at the basin edge  
434 (Fig. 5c). These results corroborate other work in mineral wetlands finding a lack of Fe (hydr)oxides in the wetland  
435 basin (LaCroix et al. 2019), suggesting that SOC in these frequently saturated soils are largely unprotected by  
436 organo-mineral associations. In fact, Fe may actually stimulate C loss under saturated conditions, as has been  
437 suggested in some drained and restored wetlands (Anthony and Silver 2020). As an alternative terminal electron  
438 acceptor, Fe facilitates anaerobic metabolism under reducing conditions. Further, the presence of reduced  $\text{Fe}^{2+}$  may  
439 both promote the oxidative activity of certain enzymes (Wang et al. 2017) and inhibit phenolic compounds which  
440 otherwise would constrain microbial decomposition (Wang et al. 2019). Thus, we expect the SOC-stabilizing  
441 influence of organo-mineral associations to promote SOC stabilization primarily during seasonally dry soil  
442 conditions, which may increase in frequency and magnitude in future.

443         While extractable Fe concentrations in the present study are relatively low, the role of organo-mineral  
444 associations in SOC stabilization may be underestimated in wet soils with greater concentrations of Fe (e.g., Kramer  
445 and Chadwick 2018). For example, one study found total extractable Fe concentrations of up to 25–30  $\text{mg Fe g}^{-1} \text{ soil}$   
446 in vernal pools (Hobson and Dahlgren 1998), though Fe is rarely reported in studies of wetland SOC and thus the  
447 extent of wetlands with Fe concentrations of this magnitude are unknown. Further, while our results represents the  
448 upper limit of Fe oxides in sampled wetlands, future research including in-situ measurements (e.g., Anthony and  
449 Silver 2020) could characterize temporal dynamics of Fe–C associations in seasonally saturated wetlands.

#### 450 *4.4 The stability of SOC in wetlands that dry seasonally at the landscape scale*

451         Of the two mechanisms studied here, a greater proportion of total SOC was physically protected within  
452 macroaggregates ( $C_{\text{aggregate}}/C_{\text{bulk}}$ ) than was associated with Fe ( $C_{\text{Dit-HCl}}/C_{\text{bulk}}$ ), indicating that macroaggregates  
453 contribute more to SOC storage in wetlands that dry seasonally. Given how little is known about the stability of

454 SOC in seasonally saturated wetlands, this study provides an initial assessment of physicochemical stabilization in a  
455 hydrologically dynamic wetland system and identifies an opportunity to further characterize physical protection of  
456 SOC in aggregates and organo-mineral associations under fluctuating saturation conditions.

457 Our research adds physicochemical mechanisms of SOC stabilization to the list of ecosystem processes  
458 (e.g. denitrification, gas fluxes) that may be enhanced in soils that experience dynamic saturation (Capps et al. 2014;  
459 Creed et al. 2013; Hefting et al. 2004; LaCroix et al. 2019; Ligi et al. 2014). More broadly, ecohydrological  
460 interfaces experience increased rates of biogeochemical processing and influence ecosystem response to  
461 environmental change (Krause et al. 2017). Coupled with the observation that small, depressional wetlands  
462 constitute a disproportionately high percentage of the total wetland perimeter length in many landscapes (Cohen et  
463 al. 2016), our results suggest that areas with fluctuating saturation, such as the transition zone, play an important role  
464 in landscape-scale SOC stabilization.

465 The future of stabilized SOC in wetlands is uncertain because the hydrologic regime of seasonally saturated  
466 wetlands is vulnerable to changes in land use and climate (Kolka et al. 2018). Under current hydrologic conditions,  
467 we expect SOC of seasonally saturated wetlands to be stabilized by physical protection in aggregates and, if there  
468 are abundant Fe or Al oxides, by organo-mineral associations particularly in the transition zone. However, as more  
469 extreme weather events are predicted with climate change (USGCRP 2017), increased incidence of severe  
470 precipitation, droughts, and warmer temperatures may lead to longer dry periods and more rapid transitions between  
471 wet and dry conditions. In transition zone and upland soils, we expect the effects of increased drying conditions to  
472 be moderated by associations between Fe and SOC in the short term. In the long term, increased storm events and  
473 subsequent leaching losses of Fe from upland soils may render SOC more susceptible to release as DOC (Kramer  
474 and Chadwick 2018) or loss via respiration. In basin soils, the low abundance of Fe-C associations may lead to  
475 enhanced SOC loss with increased drying (e.g., Fissore et al. 2009), though macroaggregates may mitigate some of  
476 these effects. Overall, our work highlights the complex implications of future hydrologic change on SOC emissions  
477 and stabilization in wetlands (Bridgham et al. 2006; Kolka et al. 2018).

## 478 **5. Conclusions**

479 Hydrologic conditions influence physicochemical mechanisms of SOC stabilization, which our research  
480 suggests may contribute to SOC stability during dry, oxic conditions in seasonally saturated wetlands. Overall, our  
481 results highlight distinct patterns in aggregate-associated C and, to a lesser extent, Fe-associated C from the basin

482 edge to the hydrologically dynamic transition zone, suggesting that these mechanisms merit further study in soils  
483 that experience both saturated and dry conditions throughout the year. These results are not only relevant for  
484 seasonally saturated wetlands in this study, but they likely extend to floodplain wetlands as well (Scott et al. 2019).

485         The role of global wetlands as a future C sink or source is uncertain, as climate and land use change alter  
486 wetland area and biogeochemical processing (e.g., Dahl 2011; Kolka et al. 2018; Moomaw et al. 2018). While the  
487 wetlands in this study are relatively small in area, up to one-fifth of wetland area in this region experiences  
488 seasonally dynamic hydrology (MD iMAP 2016), and seasonally saturated wetlands are globally ubiquitous  
489 (Calhoun et al. 2017). Therefore, the seasonally saturated areas of many small wetlands are likely to have a large  
490 cumulative impact on SOC stocks on the landscape scale. Additionally, small wetlands are also disproportionately  
491 subject to loss from human activity (e.g., Van Meter and Basu 2015), causing a decline in wetland perimeter-to-area  
492 ratios and altering SOC storage. While wetland conservation and restoration tend to focus on restoring hydrology of  
493 large wetlands, our results indicate that small, seasonally saturated wetlands have a capacity for SOC storage and  
494 stability that must not be overlooked in conservation efforts. Our research emphasizes hydrologically dynamic soils  
495 of seasonally saturated wetlands as a critical but understudied component of landscape-scale wetland SOC storage  
496 and highlights the role of physicochemical mechanisms of SOC stabilization in soils across the wetland–upland  
497 continuum.

498

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#### 504 **Conflicts of interest**

505         The authors have no conflicts of interest to declare that are relevant to the content of this article.

#### 506 **Ethics approval** NA

#### 507 **Consent to participate** NA

#### 508 **Consent to publication** NA

509 **Availability of Data and Material**

510 Water level data can be found at [www.doi.org/10.5281/zenodo.3879223](http://www.doi.org/10.5281/zenodo.3879223). All soil sample data is available in  
511 the Supplementary Information (SI).

512 **Code availability**

513 Code used for hydrologic analysis can be found at [www.doi.org/10.5281/zenodo.3879223](http://www.doi.org/10.5281/zenodo.3879223).

514 **Author's contributions**

515 **Kottkamp**: Conceptualization, Methodology, Formal analysis, Investigation, Resources, Writing- Original  
516 Draft, Visualization. **Palmer**: Conceptualization, Methodology, Formal analysis, Investigation, Resources, Writing-  
517 Review & Editing, Supervision, Project administration, Funding acquisition. **Jones**: Formal analysis, Data Curation,  
518 Writing- Review & Editing, Visualization. **Tully**: Conceptualization, Methodology, Formal analysis, Investigation,  
519 Resources, Writing- Review & Editing, Supervision, Project administration.

520 **Tables**521 **Table 1** Description of soil parameters, units, method of measurement or calculation, and their description

Property	Variable	Abbreviation	Unit	Source or Equation	Description
<b>Bulk Soil C</b>	Bulk soil C concentration	$C_{\text{bulk}}$	%; $\text{g C} \cdot \text{g}^{-1}$ soil	CHN Analyzer	Concentration of C in the bulk soil sample.
	Bulk soil C:N	$C:N_{\text{bulk}}$	ratio; $\text{g C} \cdot \text{g}^{-1}$ N	CHN Analyzer	Ratio of C:N in bulk soil.
	Soil C stock	C Stock	$\text{kg C} \cdot \text{m}^{-2}$	$C_{\text{bulk}} \cdot \text{bulk density} \cdot l$	Carbon stock in a given length increment, calculated by multiplying the C concentration in each depth increment by the bulk density and sample length.
<b>Aggregate Size Class</b>	Aggregate fraction mass		g aggregate	$\text{mass}_{\text{aggregate}}$	Mass of aggregate size class after air drying and removal of rocks (>1 mm).
	Aggregate C:N	$C:N_{\text{aggregate}}$	ratio; $\text{g C} \cdot \text{g}^{-1}$ N	CHN Analyzer	Ratio of C:N within an aggregate size class.
	Aggregate C concentration	$C_{\text{aggregate}}$	$\text{mg C} \cdot \text{g}^{-1}$ soil	CHN Analyzer; $(C \cdot \text{mass}_{\text{aggregate}}) / \text{mass}_{\text{total}}$	Concentration of C in each aggregate size class per gram of bulk soil, excluding rocks.
	Normalized aggregate-associated C	$C_{\text{aggregate}}/C_{\text{bulk}}$	$\text{mg C} \cdot \text{g}^{-1}$ C	$(C \cdot \text{mass}_{\text{aggregate}}) / \Sigma(C \cdot \text{mass}_{\text{aggregate}})$ (Eq. 1)	Proportion of C associated with an aggregate fraction, normalized to C of the entire sample. Indicates the relative proportion of C in each aggregate fraction, normalized for differences in bulk soil C across the transect.
<b>Bulk Soil Fe</b>	Poorly crystalline Fe	$\text{Fe}_{\text{AAO}}$	$\text{mg Fe} \cdot \text{g}^{-1}$ soil	Atomic absorption spectrometer	Concentration of poorly crystalline Fe species, extracted by AAO.
	Total extractable Fe	$\text{Fe}_{\text{DCB}}$	$\text{mg Fe} \cdot \text{g}^{-1}$ soil	Atomic absorption spectrometer	Concentration of total extractable Fe species (crystalline and poorly crystalline), extracted by DCB.
	Percent poorly crystalline Fe	$\text{Fe}_{\text{AAO}}/\text{Fe}_{\text{DCB}}$	%	$\text{Fe}_{\text{AAO}}/\text{Fe}_{\text{DCB}} \cdot 100$	Percent of total extractable Fe that is poorly crystalline.
<b>Fe-C Association</b>	Fe-associated C	$C_{\text{Dit-HCl}}$	$\text{mg C} \cdot \text{g}^{-1}$ soil	TOC Analyzer	Concentration of C extracted with Dit-HCl, an inorganic extraction for total extractable Fe.
	Extracted Fe	$\text{Fe}_{\text{Dit-HCl}}$	$\text{mg Fe} \cdot \text{g}^{-1}$ soil	Atomic absorption spectrometer	Concentration of Fe extracted with Dit-HCl.
	Normalized Fe-associated C	$C_{\text{Dit-HCl}}/C_{\text{bulk}}$	$\text{mg C} \cdot \text{g}^{-1}$ C	$C_{\text{Dit-HCl}}/C_{\text{bulk}}$ (Eq. 2)	Proportion of C associated with $\text{Fe}_{\text{Dit-HCl}}$ normalized to bulk soil C.
	Extractant C:Fe	$C_{\text{Dit-HCl}}:\text{Fe}_{\text{Dit-HCl}}$	ratio; $\text{g C} \cdot \text{g}^{-1}$ Fe	$C_{\text{Dit-HCl}}/\text{Fe}_{\text{Dit-HCl}}$	Ratio of extracted C to extracted Fe; indicates the kind of association between C and Fe.

522

523 **Figure Captions**

524 **Fig. 1** Cross-section schematic of theoretical sampling design (Panel A). Elevation gradient shown is exaggerated.  
525 Panel B shows actual location of wells and transects in wetlands that vary in topographic relief. Basin outline is the  
526 minimum extent of ponded water for each wetland

527  
528 **Fig. 2** Daily water levels at each wetland show differing hydrologic conditions across categorical transect points  
529 (Panel A). The black line indicates the soil surface, and the shaded grey band indicates upper 0.5 m of soil profile.  
530 As an indicator of hydrologic variability in at each sample point, we show the number of times (“event”) during the  
531 study year that the water table rose to within 0.5 m of the soil surface (i.e., mean water level above -0.5 m; Panel B)

532  
533 **Fig. 3** Soil C properties for all samples. Panel A shows bulk soil C (%) by horizon vs. mean water level, with  
534 significant models from the reduced LME presented as solid lines ( $P < 0.05$ ). Panels B and C depict SOC stocks (kg  
535 C m<sup>-2</sup>) for 0–10 cm and 10–50 cm depths, respectively, with significant models from the SLR presented as solid  
536 lines. Statistical results in Supplementary Tables 3–4

537  
538 **Fig. 4** Aggregate C ( $C_{\text{aggregate}}$ ) and normalized Aggregate-Associated C ( $C_{\text{aggregate}}/C_{\text{bulk}}$ ) vs. mean water level (from  
539 frequently saturated soils to rarely saturated soils), by aggregate size class. Solid lines represent significant  
540 differences across mean water level in the LME models. Statistical results in Supplementary Information

541  
542 **Fig. 5** Results from Dit-HCl extractions. Panel A shows  $\text{Fe}_{\text{Dit-HCl}}$  concentrations, Panel C shows Fe-associated  $C_{\text{Dit-}}$   
543  $\text{HCl}$  concentrations, and Panel C shows the normalized  $C_{\text{Dit-HCl}}/C_{\text{bulk}}$ . Data were separated at a mean water level of -  
544 0.5 m into frequently saturated soils and rarely saturated soils for analysis. Solid lines represent LME models where  
545 mean water level is significant. Only one line is shown for the LME model where differences between horizon are  
546 not significant. Statistical results in Supplementary Information

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## Supplementary Files

This is a list of supplementary files associated with this preprint. Click to download.

- [SupplementalAppendixWetlands2021.05.27.docx](#)