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Influence of aquifer heterogeneity on Cr(VI) diffusion and removal from groundwater

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1 **Abstract**

2 Previous studies show aquifer heterogeneity has an important influence on removal of
3 Cr(VI) in groundwater, but little research has revealed the role of aquifer heterogeneity
4 in Cr(VI) migration and how effective using emulsified vegetable oil is for Cr(VI)
5 removal in groundwater. We simulated a laboratory sand-packed box over a 50-day
6 period to research the effects of aquifer heterogeneity on Cr(VI) diffusion and also
7 injected emulsified vegetable oil (EVO) into the permeable reactive barrier (PRB) filled
8 with compost to investigate the influences of aquifer heterogeneity on Cr(VI) removal
9 from groundwater, with fixed conditions of simulated true water temperature of shallow
10 groundwater (19 ± 0.5 °C), hydraulic gradient (3‰), the Suzhou coal mining area (Anhui,
11 China). The results show that aquifer heterogeneity had the significant impact on Cr(VI)
12 diffusion with an overall diffusion direction of Cr(VI) that was from the upper left
13 corner to the lower right corner along the direction of the groundwater; permeable
14 reactive barrier would effectively remove Cr(VI) from groundwater in heterogeneous
15 aquifer due to the vertical movement of microorganisms between different aqueous
16 media; coarse sand and medium sand showed high performance in Cr(VI)
17 diffusion, with a slight superiority to fine sand; following a one-time EVO
18 injection, a considerably stable and uniform effective remove zone similar to the shape
19 of Σ was formed in the heterogeneous aquifer, and its Cr(VI) removal efficiency was
20 over 95%.

21 **Keywords:**

22 Groundwater; Heterogeneity; Emulsified vegetable oil; Cr(VI) removal;

23 Permeable reactive barrier; Biodegradation

24 **1. Introduction**

25 Chromium is one of the most frequently detected heavy metals in groundwater, surface
26 waters and industrial sites, which has been considered as one of the top 20 contaminants
27 on the superfund priority list of hazardous substances for the past 15 years (Grabarczyk
28 et al. 2006). Numerous industrial applications, such as production of alloys and mainly
29 steel production, metal finishing, electroplating, leather tanning, textile synthesis, wood
30 treatments and coal mining, usually cause chromium contaminated groundwater (Guo
31 et al. 2011, Sahinkaya & Kilic 2014, Yoon et al. 2011). The ground within the Suzhou
32 coal mine area (Huaibei coal mine area, northern Anhui Province, China) has
33 subsided to varying degrees due to long-term and large-scale coal mining, and
34 shallow pipeline has formed a relatively uniform drop funnel, which means coal
35 mining has severely changed the flow field of the shallow pipeline. Besides,
36 existing studies have shown that mine drainage, gangue piles and collapse
37 ponds formed by long-term coal mining poses a serious threat to surface water,
38 groundwater and soil. In particular, the shallow groundwater has been polluted
39 by conventional components such as ammonia nitrogen and heavy metals such
40 as Cr(VI) (Peng & Sun 2014). Cr(VI) is highly toxic to humans, animals, plants and
41 microorganisms and is associated with the development of various chronic health
42 disorders including organ damage, dermatitis and respiratory impairment (Xu et al.
43 2004), which has been classified as a Group A human carcinogen by the U.S.
44 Environmental Protection Agency (EPA) (Dhal et al. 2013). Therefore, Cr(VI)

45 contaminated treatment has attracted more and more attention. Recently the
46 permeable reactive barrier (PRB), with a low cost and an effective treatment in
47 contaminated groundwater remediation, has been widely studied and applied
48 (Thiruvengkatachari et al. 2008).

49 The PRB is an effective technology for in situ remediation of Cr(VI)
50 contaminated groundwater in the field (Blowes et al. 1997), the key part of
51 which is the reactive media filled in PRB and the reaction zone formed. The
52 most applied reactive media in PRB are zero-valent iron (ZVI) (Cang et al. 2009,
53 Fu et al. 2014, Qiu et al. 2020, Wei et al. 2016, Weng et al. 2007), activated
54 carbons (Huang et al. 2015, Song et al. 2019, Su et al. 2019), zeolites (Buenano
55 et al. 2017, Kesraoui-Ouki et al. 1993, Li et al. 2014, Peric et al. 2004),
56 bentonite (Wang et al. 2020, Zhang et al. 2012) and emulsified vegetable oil
57 (EVO) (Harkness & Fisher 2013, Watson et al. 2013, Wen et al. 2017). EVO is
58 an important biodegradable reactive media filled in PRB technology for Cr(VI)
59 reduction. It could stably form a large-scale microbial reaction zone for a long
60 period. Several scholars found that the effective migration distance of
61 emulsified oil in the aquifer could reach 8 m, which could create an effective
62 reaction zone within a certain range (Borden 2007, Boris et al. 2008, Clayton
63 & Borden 2009). Emulsified oil particles adsorbed on the surface of reactive
64 media could function stably in 3-5 years (Hughes 2007).

65 Previous studies have shown that emulsified vegetable oil (EVO) (Hiortdahl & Borden
66 2014) and aquifer heterogeneity have a significant effect on removing Cr(VI) from
67 groundwater. EVO, replacing organic matter (volatile solids, VS), is well distributed in
68 the sandbox after was injected into a corner of a sandbox that had heterogeneous
69 layered layers of fine-medium-coarse sand (from top to bottom) (Jung et al. 2006). A
70 one-time injection of EVO or amended EVO could effectively induce the occurrence
71 of dissimilatory iron reduction, and microbial activity is responsible for the process,
72 showing feasibility to remove Cr(VI) from groundwater (Wen et al. 2017). EVO has
73 proven efficient for anaerobic bioremediation of acid mine drainage for 300
74 days in column tests (Lindow & Borden 2005). Effluent concentrations from
75 PRBs in heterogeneous aquifers tend to be greater than effluent concentrations
76 predicted using the conventional one-dimensional deterministic plug flow
77 model typically used for design (Elder et al. 2002). The effects of geologic
78 heterogeneities in an aquifer containing an embedded clay lens and those of
79 stratified aquifer that comprises two geologic formations, on the remediation of
80 an organic contaminated aquifer through stochastic analysis (Lee et al. 2000).

81 Researchers used plexiglass sand columns or three-dimensional sandboxes to
82 carry out indoor experiments on the simulated restoration of Cr(VI) pollution
83 in groundwater and obtained a series of research results. To assess a reactive zone
84 with emulsified vegetable oil for the long-term remediation of Cr(VI)-contaminated
85 aquifer, experiments were carried out in a three-dimensional plexiglass sandbox with

86 fixed conditions of constant temperature (15.5 °C), hydraulic gradient (4‰), and
87 simulated groundwater flow rate (10 cm/d) (Dong et al. 2018). In order to investigate
88 the possibility of using the four considered materials (Fe^0 , Fe^0/Cu , AC and S/Z) inside
89 PRB for Cr(VI) from groundwater, four packed-column experiments were conducted.
90 A plexiglass column, with dimensions of 20 cm length and 2.5 cm inner diameter, was
91 used for packing the materials (Maamoun et al. 2020). The transport of Cr(VI) in the
92 presence of bentonite colloidal particles was studied in an experimental column filled
93 with sand, of which plexiglass column with a length of 50 cm, an inner diameter of 4.8
94 cm, and the outer diameter of 5 cm (Ghiasi et al. 2020).

95 In summary, previous studies have mostly focused on the effective diffusion
96 distance of emulsified vegetable oil, reactive media in PRB, service life and
97 removal effect, etc, but there have been no detailed reports about the influence
98 of heterogeneous aquifers of groundwater on Cr(VI) migration and Cr(VI)
99 removal from groundwater by emulsified vegetable oil. Also, previous
100 laboratory experiments were generally carried out at room temperature and
101 experimental conditions such as the actual water temperature and hydraulic
102 gradient of the groundwater in the field were not fully considered, which made
103 it difficult to apply the results of indoor experiments in situ remediations of
104 Cr(VI) contaminated groundwater in the field. In this study, we simulated a
105 laboratory-scale sandbox to examine the influence of aquifer heterogeneity on Cr(VI)
106 diffusion and also simulated PRB filled with compost and one-time EVO injection to

107 investigate the effectiveness of Cr(VI) removal, with fixed conditions of simulated true
108 water temperature of shallow groundwater (19 ± 0.5 °C), hydraulic gradient (3‰), the
109 Suzhou coal mining area (Anhui, China).

110 **2. Materials and methods**

111 **2.1. Materials**

112 EVO was prepared by blending 60% vegetable oil, 13% surfactant, and 27% water with
113 a booster mixer at 3000 rpm constant stirring for 10 minutes, put in a 55 °C
114 water bath for 10 minutes, taken out, cooled naturally to room temperature, and
115 stored in a cool place after sealing. The compost (purchased in a flower market)
116 was stored in a cool, ventilated and dry place. The sand was purchased from a
117 building materials market, dried after carefully sieved in a standard wire sieve
118 and made into coarse, medium, and fine sand according to the ratio shown in Table 1,
119 respectively. Table 1 describes the characteristics of packed sand in the
120 diffusion and removal study. The simulated Cr(VI) contaminated groundwater
121 sample with a 20 mg/L concentration (Li et al. 2019, Lv et al. 2019), which was
122 prepared with potassium dichromate (excellent grade pure) and deionized water.

123 **2.2. Experiment equipment**

124 A three-dimensional sandbox composed of plexiglass was used in this study. The
125 plexiglass sandbox (90 cm in length, 25 cm in width, and 48 cm in height) was
126 composed of an inlet tank, stratified aquifers (coarse, medium and fine sand), PRB,
127 injection well (a plexiglass column), outlet tank and other auxiliary equipment in order
128 from left to right as shown in Fig. 1. Fig. 1 shows the detailed experimental design.
129 There were 18 sampling ports from the front glass of the box, one water inlet and one
130 water outlet from each side of the box. The sampling ports were both laid every 10 cm
131 horizontally from left to right (6 columns), and every 10 cm vertically from bottom to
132 top (3 rows). Each sampling port was a plexiglass column (5 mm in inner diameter)
133 inserted into the sandbox 12 cm. One port of plexiglass columns was wrapped with
134 gauze inside the sandbox and the other port was connected with a rubber tube, of which
135 clamps stopped the rubber tube from water. There was 15 cm of fine sand, 15 cm
136 of medium sand and 13 cm of coarse sand were packed sequentially from
137 bottom to top. The sand finally reached a level of 43 cm and 2 cm of wet clay was
138 used to cover its surface tightly for reducing water evaporation. PRB (25 cm in
139 length, 20 cm in width, 43 cm in height) was laid 15.5 cm away from the right side of
140 the water inlet, and the filled reactive media was a mixture of quartz sand and compost
141 (4 g). After each 5-cm lift, filled reactive media was repeatedly compacted using a
142 wooden stick to ensure the homogeneity of each filling. In this way, both sides of the

143 PRB in the sandbox were packed with sand as the heterogeneous aquifer media. A glass
144 tube with holes all around was buried in the center of the PRB to simulate an injection
145 well in the field.

146 After the filling was completed, tap water was injected into the sandbox through
147 a peristaltic pump, and the aquifer was saturated with water by standing for two
148 days. Then, the rate of the peristaltic pump and outlet height were set to
149 maintain the inlet water level at 40 cm under the condition that the starting point
150 of the water level was the bottom surface of the water body at the inlet end, the
151 pore water flow rate was 10 cm/d, and the hydraulic gradient was 3‰.

152 **2.3. Experimental method**

153 Contaminated groundwater containing 20 mg/L of Cr(VI) replaced tap water as
154 a continuous water inflow after the water level of the sandbox stabilized.
155 Meanwhile, a glass rod was used to immediately and rapidly agitate the water
156 body in the inlet tank so that Cr(VI) contaminated groundwater uniformly filled
157 throughout the whole inlet tank, making it a typical source of lateral runoff
158 pollution. It was the first day that the experiment officially began.

159 The experiment lasted for 25 days from November 22 to December 18, 2019
160 (stage 1) with conditions of hydraulic gradient constant (3‰) and groundwater
161 temperature (19 ± 0.5 °C). During stage 1, water samples were collected from

162 sampling ports at 9 am every day. When the concentration of Cr(VI) of all water
163 samples reached a maximum value and remained stable for five days, we
164 believed there formed a stable Cr(VI) contaminated groundwater chemical field
165 in the sandbox aquifer. After that, a one-time injection of 8 L of emulsified
166 vegetable oil (5 g/L) was into the aquifer at a rate of 1500 mL/h by a peristaltic
167 pump through the injection well. With the other experimental conditions
168 unchanged, the experiment continued for 23 days from December 19, 2019 to
169 January 10, 2020 (stage 2). During stage 2, water samples were collected at 7
170 o'clock every evening, and its Cr(VI) concentration was determined by the 1,5
171 diphenylcarbohydrazide spectrophotometric method. When Cr(VI)
172 concentration of all water samples was 0 mg/L or reached a minimum value
173 and remained unchanged for 5 days, the experiment completed.

174 **3. Results and discussion**

175 **3.1. Effect of aquifer heterogeneity on Cr(VI) diffusion in groundwater**

176 A hill-shading profile of Cr(VI) concentration distribution in each water sample
177 of the heterogeneous aquifer corresponding to the I-I' section (stage 1) was
178 generated using the Kriging interpolation function in ArcGIS 10.1 software,
179 showing diffusion characteristics of Cr(VI) in heterogeneous aquifers on
180 different days during stage 1 (Fig. 2). In this figure, the aquifers are coarse,
181 medium and fine sand from top to bottom, respectively, the direction of water
182 flow is from left to right, the PRB is X-axis coordinate range from 7.5 cm to
183 27.5 cm, and different gradient colors from off-white to red indicate the
184 concentration of Cr(VI) from the low to the high.

185 Cr(VI) contaminants continuously diffused after entering heterogeneous
186 aquifers (Fig. 2), and the rate of horizontal diffusion was significantly faster
187 than vertical diffusion. Specifically, we studied the diffusion characteristics of
188 Cr(VI) contaminants in heterogeneous stratified aquifers by taking the orange
189 zone (OZ) (Fig. 2) as an example, where Cr(VI) concentration was equal to or
190 over 16 mg/L and less than 18 mg/L. Generally, the OZ mainly spread rapidly
191 to the right in the horizontal direction, but slowly downward in the vertical
192 direction. The results indicate the OZ usually diffused from the upper left corner
193 to the lower right corner along the direction of the groundwater flow. Also, Fig.

194 2 depicts that the diffusion characteristics of the OZ changed gradually over
195 time in a single homogeneous aqueous media. On day 1 to 2, the OZ migrated
196 in the direction of groundwater flow, entering the PRB of coarse sand and
197 medium sand. Owing to the uneven concentration of Cr(VI) contaminated
198 groundwater in the inlet tank, Cr(VI) diffused horizontally in the coarse sand
199 slightly slower than in the upper part of the medium sand, but its horizontal
200 diffusion rate in coarse sand and medium sand was still much faster than in fine
201 sand. Simultaneously, diffusion rates of Cr(VI) changed with the time the OZ
202 migrated to the water outlet (Fig. 2). The OZ reached the top border of coarse
203 sand, medium sand and fine sand for the first time in 3, 5 and 14 days,
204 respectively, and migrated to the bottom of each homogeneous aqueous media
205 for the first time in 5, 14 and 17 days, respectively. The results show that Cr(VI)
206 diffusion rates in aqueous media with different permeability were distinct, of
207 which the fastest was in coarse sand, the second was in medium sand, and the
208 slowest was in fine sand. The horizontal diffusion rates of Cr(VI) in coarse sand,
209 medium sand and fine sand were calculated to be 0.054 m/d, 0.036 m/d and 0.018 m/d,
210 respectively, and Cr(VI) concentration of the entire heterogeneous aquifer almost
211 reached 20 mg/L at the end of stage 1.

212 **3.2. Effect of aquifer heterogeneity on Cr(VI) removal effect**

213 The distribution of Cr(VI) removal efficiency in the sandbox on different days was
214 produced by the Kriging interpolation method during stage 2 shown in Fig. 3,
215 which can show the distribution of Cr(VI) removal efficiency in the
216 heterogeneous aquifer after EVO injection. The different gradient colors from
217 dark blue to red indicate Cr(VI) removal efficiency from high to low in the
218 figure.

219 Before injecting EVO (after stage 1 and before stage 2), Cr(VI) removal
220 efficiency of the entire heterogeneous aquifer was basically zero (Fig. 3). At 7
221 pm on day 27, a one-time injection of 8 L of EVO (5 g/L) accompanied with carbon
222 source facilitated microorganisms performance in the PRB on Cr(VI) removal.
223 In order to describe Cr(VI) removal effect of the entire heterogeneous aquifer
224 more conveniently, the zone of Cr(VI) removal efficiency above 95% was
225 defined as the ERZ (effective removal zone) in this paper. The ERZ is the light
226 blue and dark blue zones in Fig. 3. On day 28 to 45, whether it is seen from the
227 entire heterogeneous aquifer or the inside of a single sand layer, similar to
228 Cr(VI) diffusion in aquifers, diffusion of the ERZ was basically consistent with
229 the overall flow direction of groundwater. On day 46 to 50, there formed a
230 spatially stable ERZ. The results show that the groundwater flow field has an
231 important effect on the spatial distribution of the ERZ, and a stable ERZ could

232 be finally established in the entire heterogeneous aquifer downstream of the
233 injection well 18 days following EVO injection under the experimental
234 conditions. The stable ERZ was approximately 38 cm in length, 40 cm in height,
235 covered 1520 cm², its removal efficiency remained 97% approximately, and its
236 stable time was not less than five days.

237 The spatial distribution and diffusion characteristics of the ERZ were quite
238 different within the heterogeneous aquifer composed of coarse sand, medium
239 sand and fine sand. The longest and shortest distances between the ERZ and the
240 injection well in three sand layers are shown in Fig. 4. And Fig. 5 depicts the
241 area changes of the ERZ in sand layers.

242 As shown in Fig. 3, on day 28 to 29 (stage 2), the left border (that is, the shortest
243 distance in coarse sand in Fig. 4) of the ERZ reached the right border of the
244 injection well (about 20 cm from the right side of the water inlet), the right
245 border (the longest distance in coarse sand) of the ERZ reached the right edge
246 of the coarse sand (60 cm from the right side of the water inlet), and the ERZ
247 diffused throughout the entire coarse sand except for a small area in the lower
248 right corner of coarse sand; The left edge (the shortest distance in medium sand)
249 of the ERZ reached the right border of the injection well, the right boundary
250 (the longest distance in medium sand) of the ERZ only reached the middle of
251 the medium sand (about 42 cm from the right side of the water inlet) and spread

252 vertically as far as the bottom of the medium sand (about 18 cm from the bottom
253 of the sandbox), whereas, the ERZ did not diffuse into the fine sand. The results
254 indicate that the microorganisms in the compost filled in the PRB could
255 effectively remove Cr(VI) in a certain range downstream of the injection well
256 after obtaining the carbon source of EVO.

257 The ERZ generally diffused upward and to the upper right in coarse sand and
258 medium sand, but the top right corner of the ERZ in coarse sand tended to
259 spread to the upper left. This tendency was most obvious on day 30 when the
260 ERZ first diffused into the left side of the PRB upstream of the injection well
261 (about 10 cm from the right side of the water inlet) and almost filled the entire
262 coarse sand in the PRB. Besides, a small portion of the ERZ also appeared in
263 medium sand. The ERZ in the coarse sand had shrunk downstream of the injection
264 well on day 31 and diffused upstream of the injection well at the end of stage 2.
265 Whereas, the ERZ in the fine sand did not spread to the upstream of the injection well
266 during stage 2. On day 30, the ERZ in fine sand first appeared in the middle of
267 the PRB and spread around, and it finally merged with the ERZ in coarse sand and
268 medium sand on day 32, forming a unified ERZ in the heterogeneous aquifer (the ERZ
269 below, unless otherwise specified to the unified ERZ). Its shape resembled a pistol
270 with the muzzle pointing to the right.

271 On day 33 to 34, the morphology of the ERZ did not change much, but its area
272 changed slightly. The ERZ covered a local maximum area of 901.97 cm² on
273 day 34 (Fig. 5). Furthermore, the left and right boundaries of the ERZ generally
274 extended from the upper left corner to the lower right corner along the direction
275 of water flow on day 35. The area of the ERZ in the whole heterogeneous
276 aquifer decreased drastically due to the significant decrease of which in the
277 coarse sand and reached a minimum local area (559.87 cm²) on day 38. The left
278 edge of the EZR expand upstream and towards the PRB, while the right edge
279 of which kept spreading to the left bottom, and the area of the EZR started to
280 oscillate and increase gradually on day 39, reached a maximum area of 1357.87
281 cm² on day 47, and remained unchanged until the end of stage 2. Finally, the
282 ERZ was approximately in the shape of Σ , the left border of which was located
283 on the left side of the right border of the PRB. The ERZ filled the whole region
284 except for a small part in the fine sand at the lower left where was the entire
285 heterogeneous aquifer downstream of the right border of the PRB. The results
286 show that the EZR in the coarse sand rapidly covered the largest area 7 days
287 after EVO injection early in the experiment, but its area was rapidly reduced to
288 the smallest one in the following 2 days, and then it gradually increased due to
289 the interaction of microorganisms in medium sand and fine sand. Although the
290 area of the ERZ in medium sand and fine sand also fluctuated, the area of that
291 in the entire heterogeneous aquifer increased gradually and reached the

292 maximum value simultaneously with coarse sand at the end of stage 2 (Fig. 5).
293 This indicates that microorganisms in coarse sand, medium sand and fine sand
294 could have mutual flow vertically and form a stable and unified ERZ at the end
295 of the experiment. Meanwhile, aquifer heterogeneity had the least impact on
296 the ERZ.

297 Fig. 6, Fig. 7 and Fig. 8 show the variation of Cr(VI) removal efficiency over
298 time in the sampling ports of coarse sand, medium sand and fine sand,
299 respectively. Cr(VI) removal efficiency was calculated for the four materials using Eq.
300 (1)(Maamoun et al. 2020)

$$301 \quad \text{Removal efficiency} = \frac{C_0 - C_d}{C_0} \times 100 (\%)$$

302 where, C_0 and C_d are initial concentration and concentration at time d of Cr(VI),
303 respectively. The removal efficiency of Cr(VI) from the left to right 6 sampling
304 ports in the coarse sand were represented as C1-C6, and the representatives of
305 other sampling ports could be deduced by analogy.

306 On day 34, the removal efficiencies of Cr(VI) of coarse sand were basically
307 stable at 90% after EVO injection except sampling port C1 (Fig. 6). Sampling
308 port C1 was located at the center of the PRB and was close to the pollution
309 source, which would result in a low Cr(VI) removal efficiency. Due to the faster
310 water flow rate in the coarse sand and the more significant loss of

311 microorganisms and EVO, Cr(VI) removal efficiency of each sampling port
312 showed a decline tend downgradient on day 35, especially that of sampling port
313 C2 dropped to 40%, a considerable reduction in Cr(VI) biotransformation
314 efficiency. However, in the next 6 days, Cr(VI) removal efficiency of each
315 sampling port continuously increased, especially that of sampling port C2 rose
316 to 90% on day 41. Cr(VI) removal efficiency of the whole coarse sand
317 eventually reached equilibrium with around 90%, except that of sampling port
318 C1 was stable at 13% at end of stage 2.

319 As presented in Fig. 7, on day 32, the Cr(VI) removal efficiencies of sampling
320 ports M2 to M5 of medium sand exceeded 90%. Two sampling ports with
321 removal efficiency approaching 25% to 30% appeared. One of these was
322 sampling ports M6 mainly for a long distance from the PRB, fewer
323 microorganisms and nutrients, and the other was sampling ports M1 due to its
324 location in the upstream direction of the injection well. The Cr(VI) removal
325 efficiency of sampling ports M6 increased to 90% but that of sampling ports
326 M2 decreased from 90% to 20% on day 35 and kept fluctuating until it reached
327 equilibrium with 90% approximately on day 45. The variation of Cr(VI)
328 removal efficiency in the sampling ports of medium sand was similar to that of
329 coarse sand, except that the time of its overall Cr(VI) removal efficiency
330 recovered to 90% was 4 days later compared with coarse sand. The Cr(VI)

331 removal efficiencies of sampling port M2 to M6 were finally over 90% steadily,
332 only that of sampling port M1 was 12% steadily at the end of stage 2.

333 On day 35, the Cr(VI) removal efficiencies of sampling port F1, F5 and F6 were
334 below 20% while that of sampling port F3 and F4 of fine sand stabilized at 90%.

335 The Cr(VI) removal efficiencies of sampling port F5 and F6 slowly increased
336 and finally stabilized at 90% on day 45. The removal efficiency of sampling
337 port F2 rose slowly from 50% on day 29, and it fluctuated at 37% to 90% on
338 day 33 to 45 and recovered to 90% on day 47. The variation of Cr(VI) removal
339 efficiency in the sampling ports of fine sand was similar to that of medium sand,
340 except that the time of its overall Cr(VI) removal efficiency recovered to 90%
341 was 2 days later compared with medium sand. Finally, the removal efficiency
342 of Cr(VI) of sampling port F1 was stable at 13%, and that of other sampling
343 ports were stabilized at more than 90%, as illustrated in Fig. 8.

344 The above results indicate that under the experimental conditions, aquifer
345 heterogeneity significantly induced a change of removal efficiency of Cr(VI) in
346 each sand layer. The removal efficiencies of Cr(VI) of sampling port C2, M2
347 and F2 all decreased to varying degrees on day 35 and recovered to about 90%
348 on day 41, 45, and 47, respectively. This could be due to the vertical migration
349 of Cr(VI) contaminants in the three aqueous media with different permeability.
350 Following this, Cr(VI) removal efficiencies of the three aqueous media

351 decreased at different levels owing to microbial degradation and diffusion of
352 Cr(VI) contaminants. When Cr(VI) removal efficiency of fine sand was
353 exceedingly higher than that of medium sand and coarse sand, the migration of
354 Cr(VI) contaminants from fine sand to medium sand and coarse sand would be
355 aggravated. After that, on day 47, Cr(VI) removal efficiencies of the entire
356 heterogeneous aquifers were almost equal and stable at $90\% \pm 5\%$ (except that
357 of sampling port C1, M1 and F1), indicating that aquifer heterogeneity at this
358 time had a minimal impact on the removal of Cr(VI).

359 **4. Conclusion**

360 In this study, we investigated a simulated ERZ (effective reactive zone) for the in situ
361 bioremediation of Cr(VI) contaminated groundwater under simulated conditions of
362 hydraulic gradient (3‰), shallow groundwater temperature (19 ± 0.5 °C), Suzhou coal
363 mining area (Anhui, China). Results revealed that coarse sand and medium sand
364 showed high performance in Cr(VI) diffusion, with a slight superiority to fine
365 sand, with final horizontal diffusion rates values of 0.054 m/d, 0.036 m/d and
366 0.018 m/d, respectively. Meanwhile, Cr(VI) horizontal diffusion was faster than
367 vertical diffusion in the entire heterogeneous aquifer or the single relatively
368 homogeneous aqueous media. The diffusion of Cr(VI) in the vertical direction
369 could make the entire heterogeneous aquifer polluted. Furthermore, Cr(VI)

370 overall diffusion direction was from the upper left corner to the lower right
371 corner along the direction of groundwater flow.

372 Following EVO injection, a stable and uniform effective reactive zone(ERZ)
373 similar to the shape of Σ was formed in the heterogeneous aquifer downstream
374 of the injection well. The ERZ almost filled the entire heterogeneous aquifer at the
375 right boundary of the PRB, and its maximum area of section I-I' reached 1357.87 cm².
376 The results showed that aquifer heterogeneity had the apparently significant influence
377 on Cr(VI) removal efficiency of coarse sand, medium sand and fine sand, especially the
378 removal efficiency in the horizontal direction. Due to the vertical diffusion between
379 coarse sand, medium sand and fine sand, the microorganisms in different sand layers
380 could flow to each other, resulting in an effective Cr(VI) removal in the entire
381 heterogeneous aquifer.

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501 **Declarations**

502 ● Ethics approval and consent to participate (Not applicable)

503 ● Consent for publication (Not applicable)

504 ● Availability of data and materials

505 The datasets generated during and/or analysed during the current study are not
506 publicly available due to data confidentiality but are available from the
507 corresponding author on reasonable request.

508 ● Competing interests

509 The authors declare that they have no competing interests.

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514 ● Authors' contributions

515 Conceptualization: Weidong Zhao; Methodology: Weidong Zhao, Lei Ma; Formal
516 analysis and investigation: Weidong Zhao, Xinxiang Huang, Jianshi Gong; Writing
517 - original draft preparation: Xinxiang Huang; Writing - review and editing:
518 Weidong Zhao, Lei Ma; Funding acquisition: Jiazhong Qian, Lei Ma, Weidong
519 Zhao; Resources: Xinxiang Huang; Supervision: Weidong Zhao, Lei Ma.

Figures

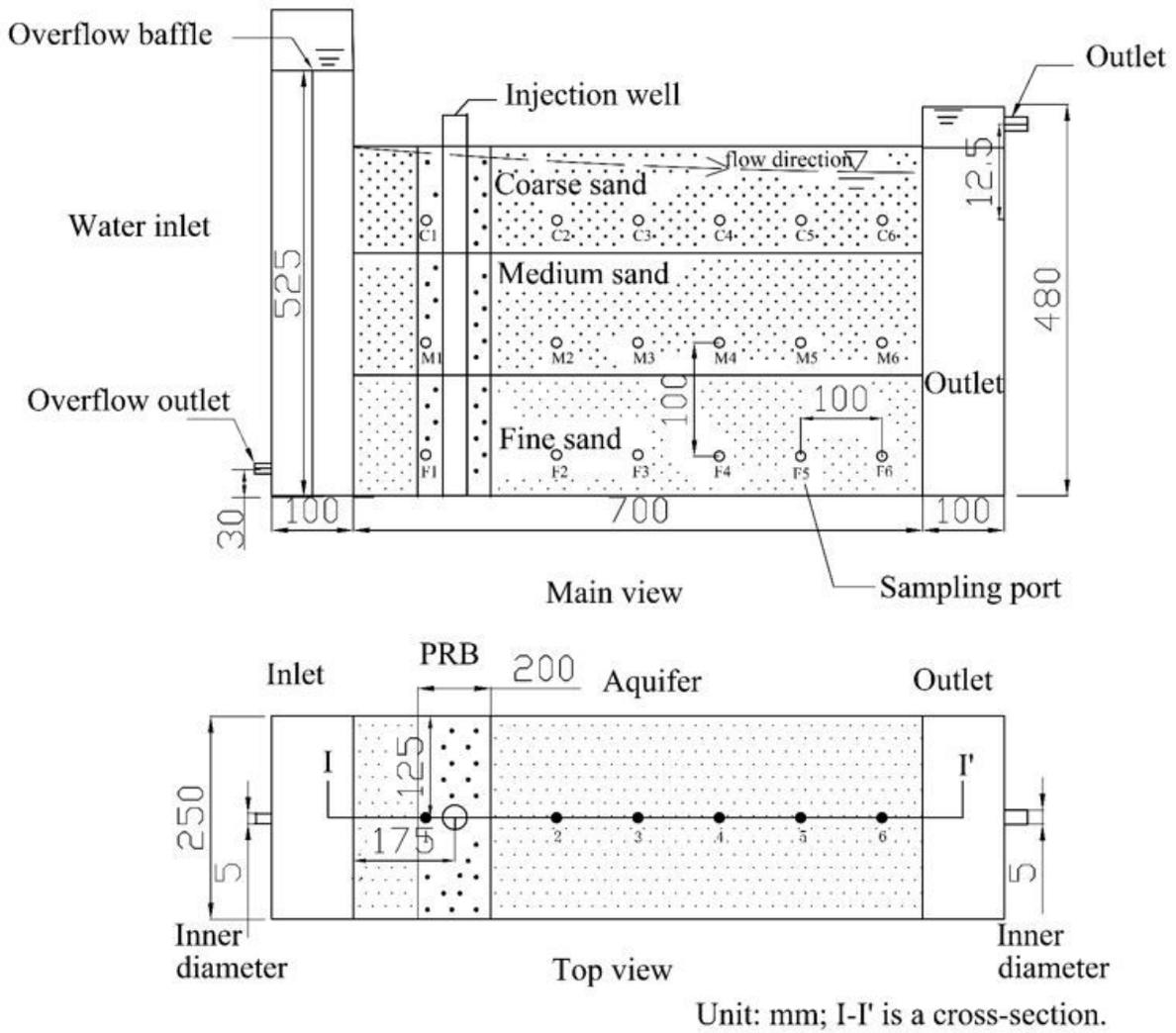


Figure 1

Main view and top view of Cr(VI) diffusion and removal experimental sandbox setup

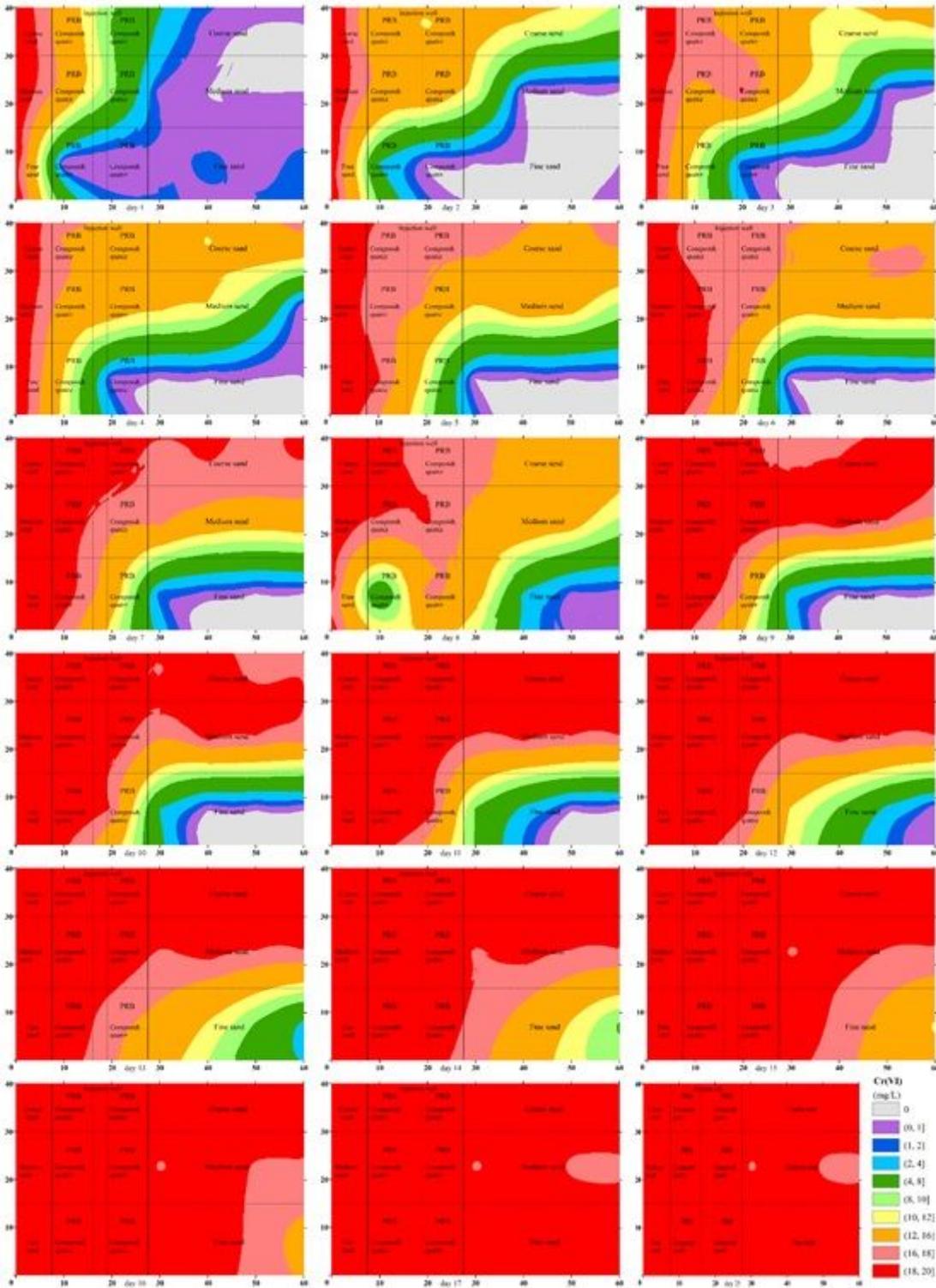


Figure 2

Diffusion of Cr(VI) in the sandbox on different days during stage 1

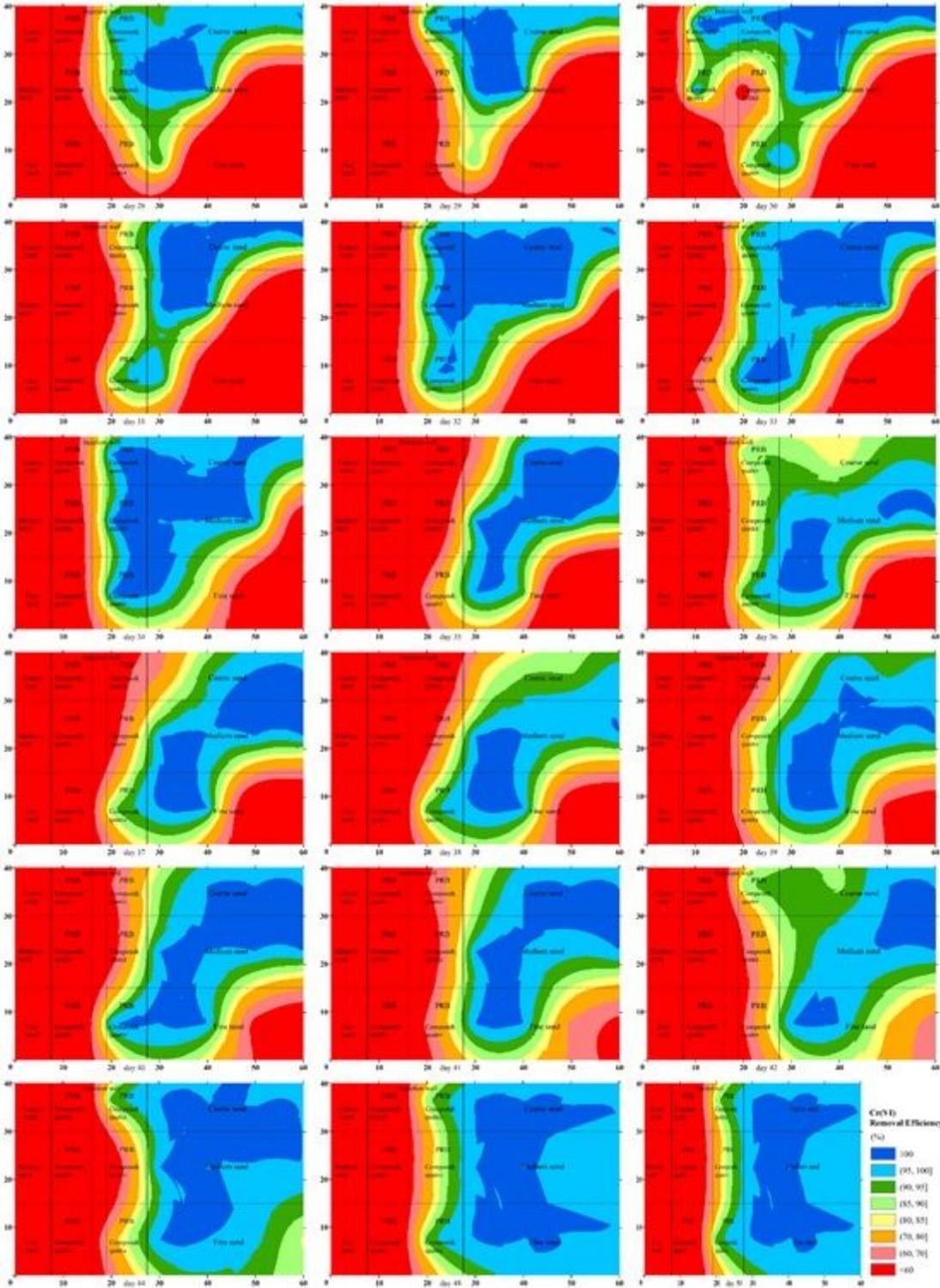


Figure 3

Distribution of Cr(VI) removal efficiency in the sandbox on different days during stage 2

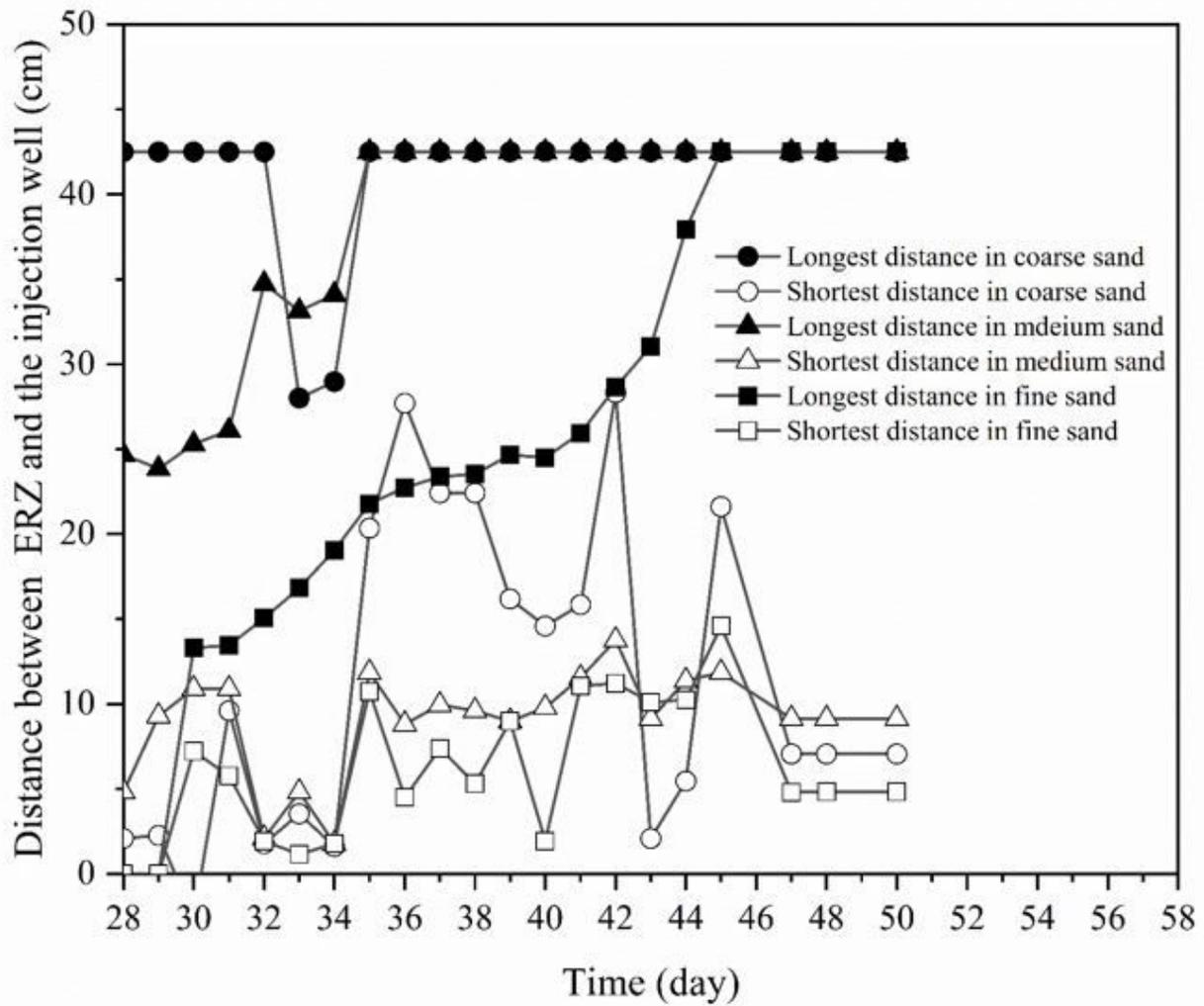


Figure 4

The longest and shortest distances between the ERZ and the injection well in three sand layers during stage 2

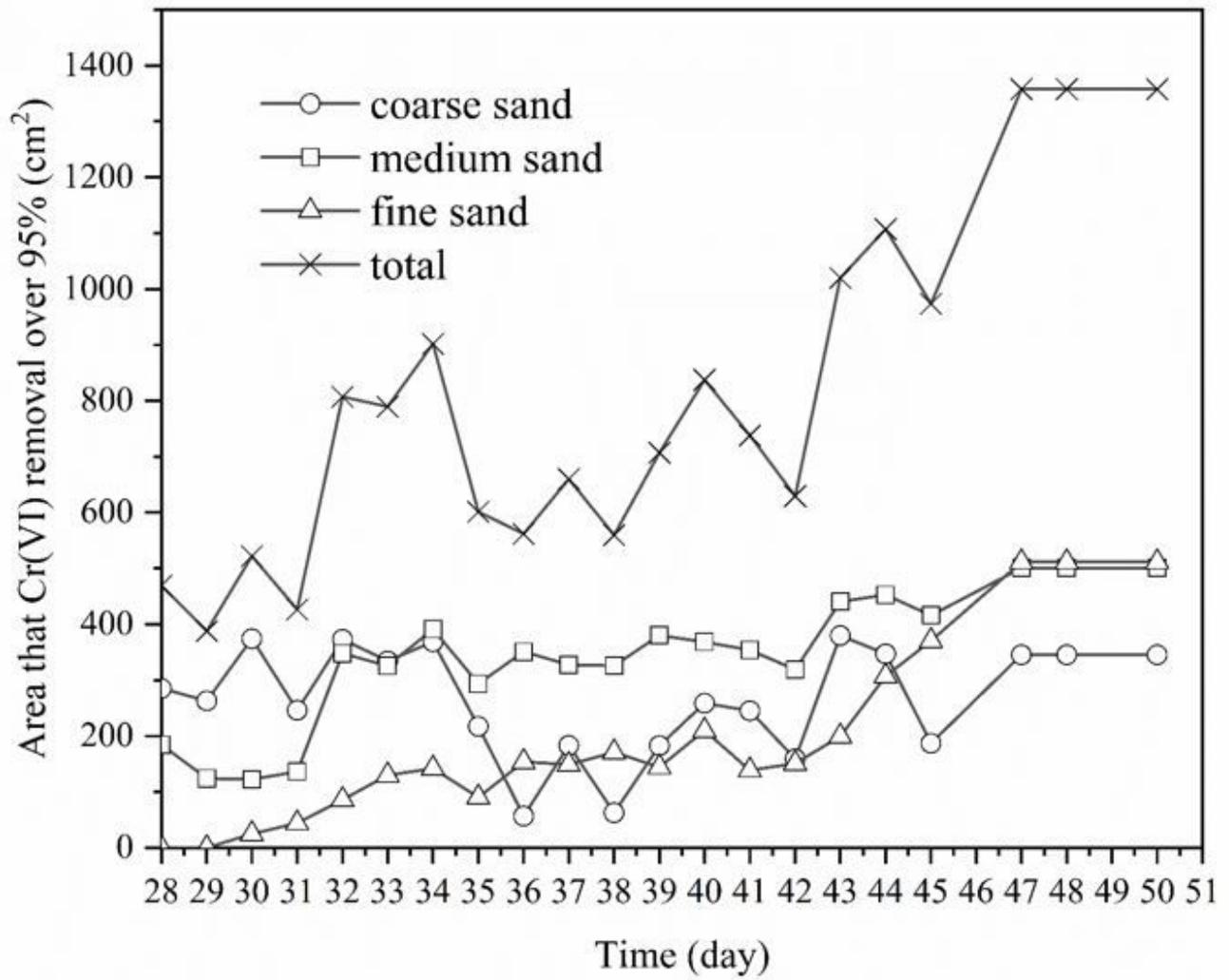


Figure 5

Area change of ERZ in sand layers

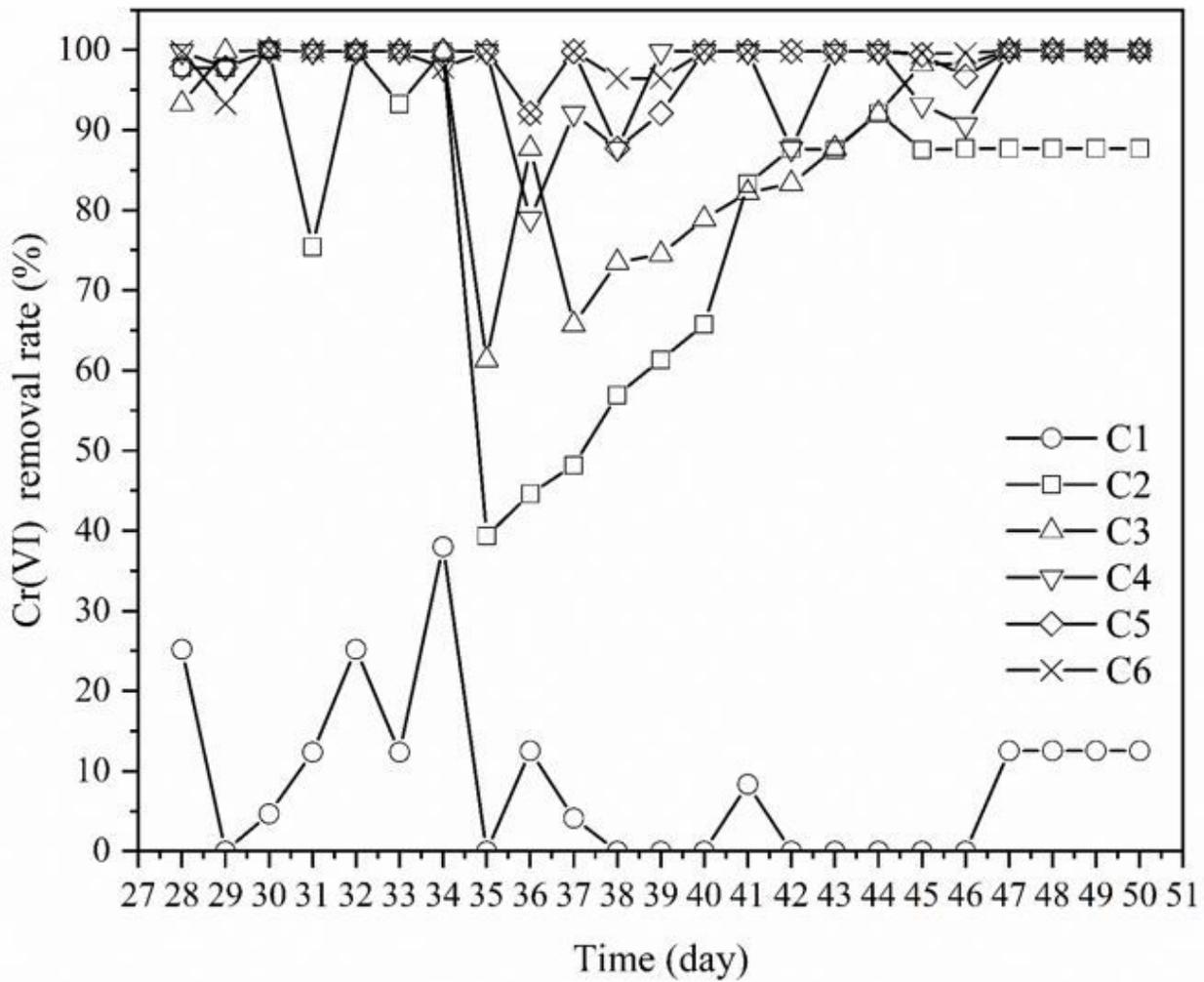


Figure 6

The variation of Cr(VI) removal efficiency in the sampling ports of coarse sand during stage 2 (Cr(VI) = 20 mg/L, $19 \pm 0.5 \mu$, hydraulic gradient = 3 ‰)

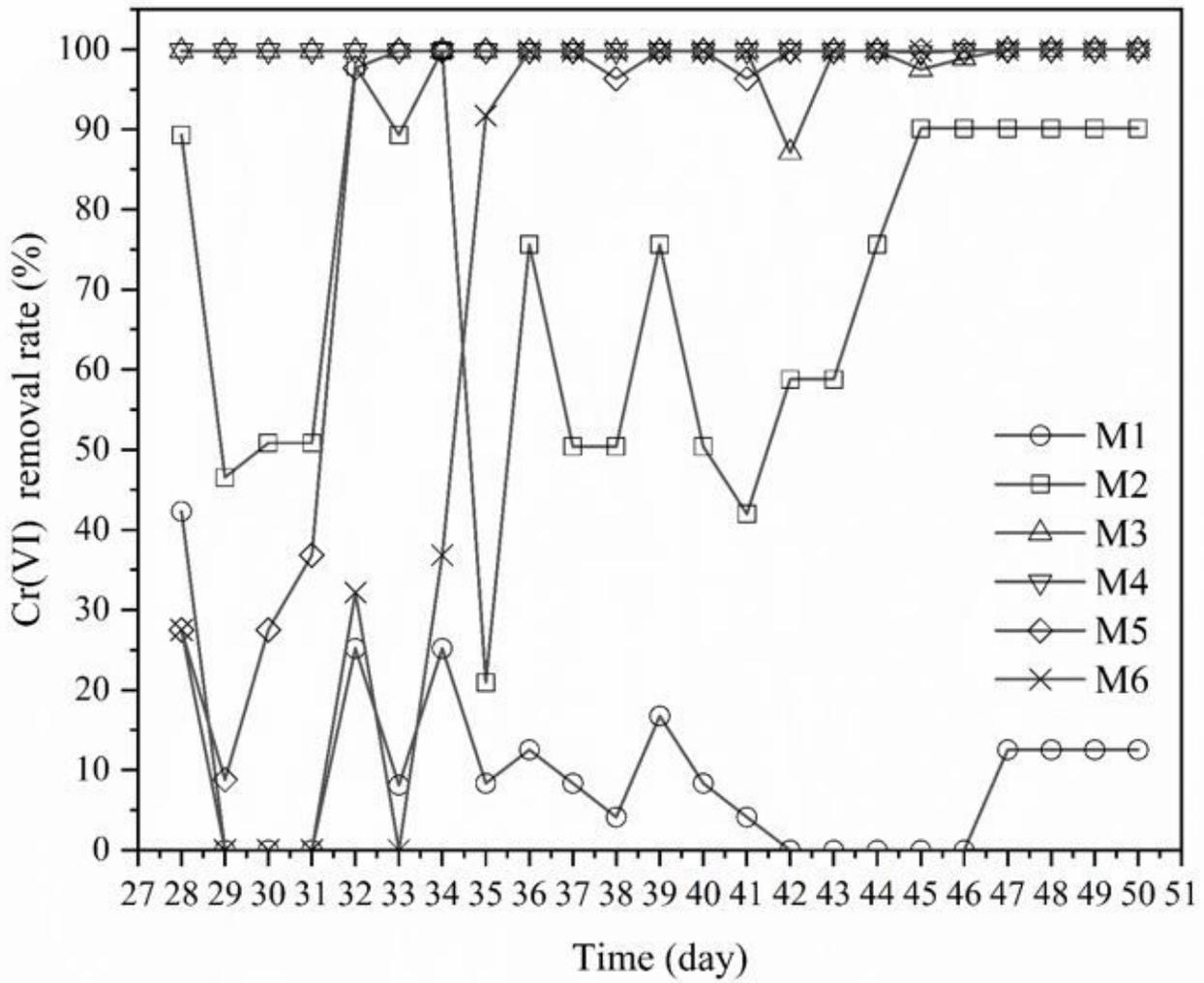


Figure 7

The variation of Cr(VI) removal efficiency in the sampling ports of medium sand during stage 2 (Cr(VI) = 20 mg/L, 19 ± 0.5 μ , hydraulic gradient = 3 ‰)

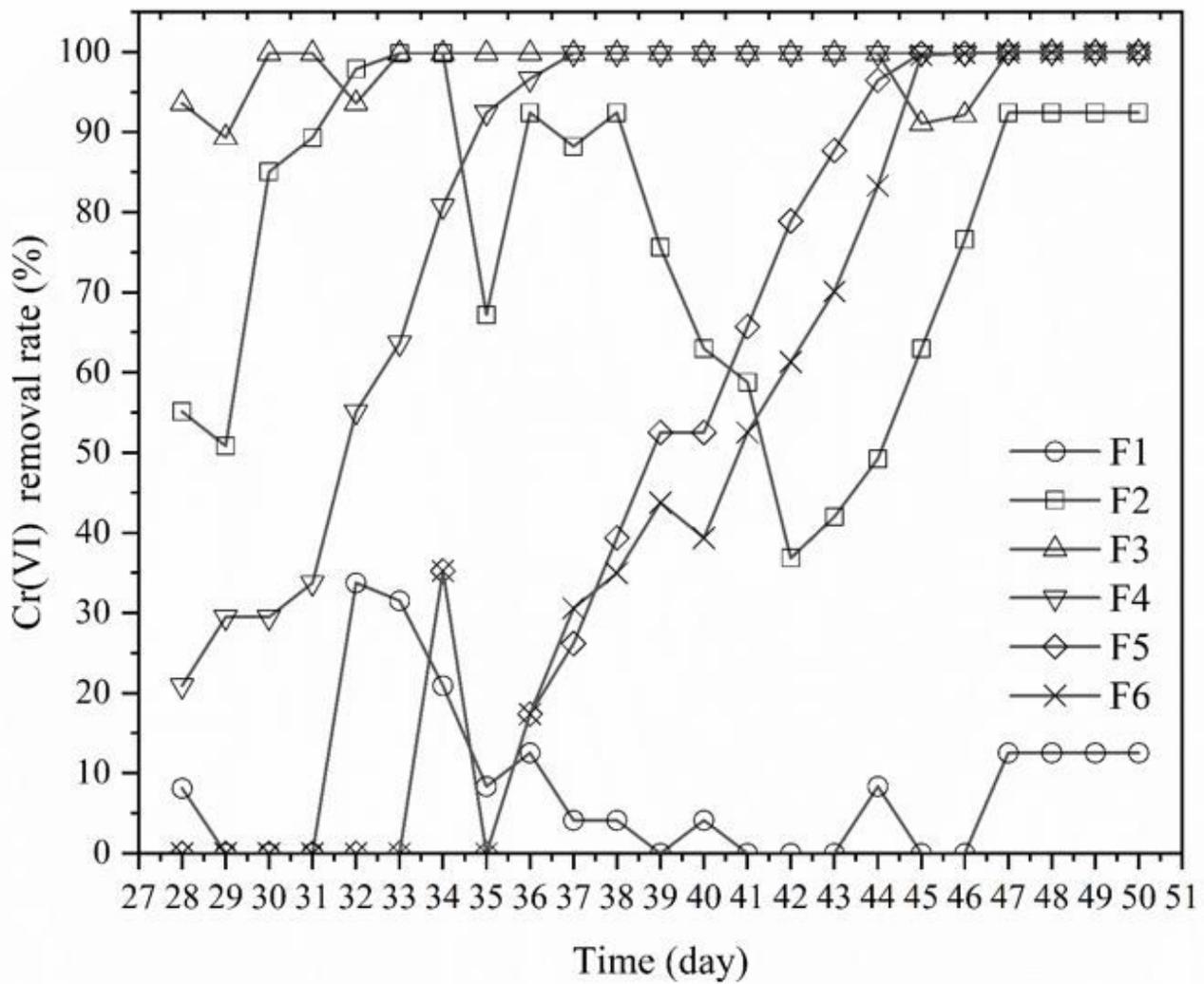


Figure 8

The variation of Cr(VI) removal efficiency in the sampling ports of fine sand during stage 2 (Cr(VI) = 20 mg/L, $19 \pm 0.5 \mu$, hydraulic gradient = 3 ‰)