

Analysis of *Achyranthes aspera* leaf extract and environmental safety evaluation to non target Nile tilapia fish fingerlings, *Oreochromis niloticus*.

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Research

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1 **Analysis of *Achyranthes aspera* leaf extract and environmental safety evaluation to**
2 **non target Nile tilapia fish fingerlings, *Oreochromis niloticus*.**

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25 **Abstract**

26 **Background:** Although plant molluscicides are biodegradable and less toxic to higher
27 animals, unregulated applications could affect other non target aquatic species. Eco
28 toxicological studies are required to evaluate whether they are safe to those economically
29 and ecologically important fish species. According to previous studies, *Achyranthes*
30 *aspera* is a molluscicidal plant with LC₉₀ of 96.5 mg/L and the current study is to
31 investigate its toxic effect to Nile tilapia fish fingerlings, *Oreochromis niloticus*.

32 **Methods:** Seven fish fingerlings were exposed to serial dilutions of *A. aspera* leaf
33 aqueous extract for 96 hours in duplicate setup. Phytoconstituents were identified by GC-
34 MS and lethal concentrations were determined by probit model. The NOAEC was
35 determined by hypothesis testing on the survival data.

36 **Results:** The respective piscicidal LC₁, LC₁₀ and LC₅₀ values were 897.4, 1063.9 and
37 1310.74 mg/L. The NOAEC was 1100 mg/L (p>0.05). GC-MS analysis revealed 12
38 phytoconstituents including a monoterpene.

39 **Conclusion:** The result shows that *A. aspera* is non toxic and hence is safe to Nile tilapia
40 fish especially at its molluscicidal dose limit (96.5 mg/L LC₉₀) corroborating to the plant's
41 target specific molluscicidal potential.

42 **Keywords:** *Achyranthes aspera*, molluscicidal, NOAEC, *Oreochromis niloticus*,
43 piscicidal, phytochemical

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48 **Background**

49 Awareness on the detrimental impact of synthetic pesticides on the environment coupled
50 with increasing prices is forcing researchers and communities to look for alternative
51 botanical products [1]. Ease of access at low cost and environmental friendliness are the
52 major attributes of many medicinal plants. Similar refreshment is being observed in the
53 search for molluscicidal plants against vector snails transmitting schistosomiasis and
54 other trematode parasites[2, 3].

55 Even though, plant molluscicides are generally considered as ecofriendly, rapidly
56 biodegradable and less toxic to higher animals, their indiscriminate use could cause risk
57 to many nontarget organisms. Toxic effects of these products at least to those important
58 species should be investigated beforehand. Lack of such ecotoxicological information is
59 a hindrance to their practical applicability[4, 5] Acute fish toxicity assessments were
60 performed on few molluscicidal plants. For instance, the toxicity effect of *Phytolacca*
61 *dodecandra* to aquatic invertebrates *Daphnia Magna* and fish (*Pimephales Promelas*)
62 was discussed in a study by Lambert et al., [6]. Similarly, toxicity of *Jatropha curcas* on
63 certain aquatic crustaceans and annelids was evaluated by Rug and Ruppel [7]. Acute
64 toxicity of *Moringa oleifera* to Nile tilapia *Oreochromis niloticus* fingerlings is also
65 studied [4, 8].

66 Beside toxicity studies, knowledge of the chemical constituents of such plants is desirable
67 to develop a more comprehensive information regarding its toxicological potential, safety
68 and efficacy [9]. In addition, identification of bioactive compounds is useful for further
69 understanding of its molluscicidal properties and for the synthesis of effective chemical
70 substances [10].

71 *Achyranthes aspera*, (Amaranthaceae), also known as “devil's horsehip” is stiff erect
72 perennial herb growing as a weed in many parts of Asia, Africa, America, Europe and
73 Australia [11–13]. It is a well-known medicinal plant in Ethiopia, India and other
74 countries as antifungal, antibacterial, antioxidant, antifertility and in the treatment of renal
75 dropsy, skin rash, chronic malaria, impotence, asthma, diabetes etc.[14–16]. Very
76 recently, The molluscicidal property of *A. aspera* is discovered by after investigation on
77 the aqueous and ethanolic extracts as well as in bait formulation [17, 18].

78 Phytochemical analyses carried out on *A. aspera* have revealed several compounds
79 including flavonoids, alkaloids, saponins and cardiac glycosides.[16, 19–21]. Saponins
80 and alkaloids are molluscicidal; also, piscicidal. Therefore, the plant’s molluscicidal
81 application requires further phytochemical analysis and ecotoxicological investigations.
82 This study is aimed to identify the major phytoconstituents present in *A. aspera* leaves
83 and investigate its acute toxicity to fingerlings of Nile tilapia, *O. niloticus*.

84 Nile tilapia is an African freshwater cichlid fish naturally occurring in rivers, dams and
85 lakes as an important ecological entity. It is an important food fish all over the world and
86 is most exploited species constituting 60-80% total fish capture [22–25]. Moreover, it is
87 one of the predominantly stocked fish in aquacultures, artificial lakes, reservoirs and
88 small water bodies.

89 **Materials and methods**

90 A standard non-renewed static test was adopted for this test to evaluate the fish acute
91 toxicity of *A. aspera* leaf aqueous extract in 96 hours exposure time.

92 ***Plant Material Collection and Processing***

93 Fresh leaves of *A. aspera* plant are collected in October 2017 from a natural habitat

94 located at 9°43'45.59" N, 39°37'2.71" E in Debre Berhan Zuria-Keyit District in Amhara
95 region, Ethiopia. The leaves were dried in the shade to a final weight and ground to a 200
96 µm mesh sized powder using an electric blender. For each serial dilution, crude aqueous
97 extraction was made at the time of experimentation, by soaking the required amount in
98 400 ml aged water in a conical flask and shaking overnight in an orbital shaker at 160
99 rpm. Finally, it was filtered using a clean cotton filtering cloth where the filtrate was then
100 transferred in to a volumetric flask, tightly capped and stored in the refrigerator at 4°C.

101 ***Test Animal Collection and Maintenance***

102 Fingerlings of *O. niloticus* were collected from an artificial fishpond in Dilla University
103 main compass, southern Ethiopia. The owner of the pond, Department of Biology, has
104 permitted us to take 110 fingerlings. A skilled person assigned to manage the pond, caught
105 the fish with care by a small handmade fishing net. Both male and female fingerlings of
106 length 5.9-6.6 cm (6.27 ± 0.2 cm) and mass 3.4-4.3 g (3.89 ± 0.25 g) were selected for this
107 test and the remaining were immediately returned in to the pond. Caught fish were
108 promptly put into clean plastic buckets containing the pond water and immediately
109 brought to the laboratory for acclimatization. They were maintained in two 80 cm x 60
110 cm x 60 cm aquaria for one week in aged tap water under continuous aeration and 12-
111 hour light, 12-hour dark photoperiod. They were fed with flakes of tasty soya (Pramukh
112 Agroindustry PLC, Ethiopia) as recommended [26]. Feeding was stopped 24 hours prior
113 to the start of experiment [27, 28].

114 ***Range Finding Test***

115 The range finding test was set according to the protocols defined by OECD [29] and EPA
116 [30]. Five serial dilutions of 100, 400, 800, 1200 and 1600 mg/L aqueous extract were
117 prepared in 10 L aged water. Three randomly selected healthy fish were exposed to each

118 test solution for 24 hours. The purpose of the range finding test is to identify the useful
119 concentration range that would probably produce mortality rates between 0 and 100%
120 and guess the range of concentrations for the definitive test.

121 ***Definitive Test***

122 Six serial dilutions: 600, 800, 1000, 1100, 1200 and 1400 mg/L were prepared in 30 L
123 aged tap water. According to the protocols [29, 31], seven healthy fishes were randomly
124 released in to each test solution for 96-hour exposure time. Only aged tap water was used
125 for negative control. The whole test was prepared in duplicates. Each solution and the
126 controls were intermittently aerated for about 40 minutes in every 2-3 hours.

127 ***Physicochemical Characteristics of Test Water***

128 Dissolved oxygen, temperature and pH were measured in every 24 hours using a
129 multiparameter probe. Before every measurement, the probe was calibrated according to
130 its operating manual. Total hardness was measured only once at the start by EDTA
131 titrimetric method. Eriochrome Black T (EBT) was used as indicator. First, 100 ml sample
132 water was buffered by 2 ml ammonium hydroxide to pH 10. When few drops of EBT was
133 added to this sample water, a wine-red color appeared. Upon titration with EDTA, the
134 color gradually changes to blue. At this instance of color change, the titration was
135 Immediately stopped and the final volume of the ample water was noted

$$136 \quad \text{Hardness (in mg/L as CaCO}_3\text{)} = (V \times N \times 50 \times 1000) / (SV)$$

137 Where: V = volume of titrant (ml); N = normality of EDTA; 50 = equivalent weight of
138 CaCO₃; SV = sample volume (ml)

139 The course of the experiment was monitored through constant and careful bench side
140 observation and cumulative mortalities were recorded at the end of 12, 24, 48, 72 and 96
141 hours according to OECD [29]. Dead fish were identified by absence of tail and gill
142 movements and loss of sensation to gentle prodding with a glass rod [32]. Dead fish were
143 immediately picked out from the solutions to minimize contamination.

144 ***Phytochemical Analysis of A. aspera by GC-MS***

145 The chemical profile of *A. aspera* leaf crude ethanolic extract was documented from GC-
146 MS analysis employing Agilent- Technologies (Little Falls, CA, USA) 6890N Network
147 Gas Chromatographic system, equipped with an inert XL Mass detector (Agilent-
148 Technologies 5975), auto injector (Agilent-Technologies 7683B series) and HP-5MS 5%
149 Phenyl Methyl Silox capillary column (27 m x 250 μm with film thickness x 0.24 μm).
150 The cured extract was clean up by silica gel and dissolved in HPLC grade n-hexane with
151 a ratio of 1:99. This mix was ready for GC-MS analysis. 1.0 μL sample was injected in
152 the split less mode. Helium was used as a carrier gas with a flow rate of 1 ml/min. The
153 temperature of column oven was programmed started from 40 $^{\circ}\text{C}$ for 0 min to 100 $^{\circ}\text{C}$ at
154 4 $^{\circ}\text{C}$ /min and from 100 to 310 $^{\circ}\text{C}$ at 10 $^{\circ}\text{C}$ /min while initial and final holdup time was
155 1 and 16 min, respectively. The temperature of the injector and MS transfer line were set
156 at 250 and 280 $^{\circ}\text{C}$, respectively. An electron ionization system (with ionization energy of
157 70eV) was used for GC/MS detection while the scanning mass was ranged from 33–500
158 m/z.

159 ***Data Analysis***

160 The phytoconstituents were identified by comparing the spectra of each unknown
161 component with the spectrum of known compounds archived in a database repository.

162 Effective doses (LC₁, LC₁₀ and LC₅₀) were determined from the mortality data by probit
163 regression method in IBM SPSS software, version 23. Mean and standard deviations of
164 fish weight and lengths as well as the water physicochemical parameters were computed
165 in Microsoft excel spreadsheet of Office 2016.

166 The NOAEC (No-Observed-Adverse-Effect-Concentration) is by definition, the highest
167 concentration at which survival of the test organism is not significantly different from the
168 control [30, 33]. In other words, it is the highest concentration from the serial dilutions in
169 which the number of fishes died is not significantly different from that in the control.
170 Therefore, the NOAEC was determined by hypothesis testing using Dunnett t test at 95%
171 confidence interval [30].

172 First: the mortality rates are recorded as response proportion data for each concentration
173 and control group.

174 $(RP) = \text{Number of survivors} / \text{Total exposed}$; where RP is Response Proportion

175 Second: The resulting RPs are transformed to arc sine values in radian as follows

176 (1) For RPs greater than zero or less than one

177
$$\text{Angle (in radians)} = \text{arc sine} \sqrt{(RP)}$$

178 (2) Modification of the arc sine when $RP = 0$.

179
$$\text{Angle (in radians)} = \text{arc sine} \sqrt{1/4n}$$

180 where n = number of animals per treatment replica.

181 (3) Modification of the arc sine when $RP = 1.0$.

182 $Angle = 1.5708 \text{ radians} - (\text{radians for } RP = 0)$

183 Third: Dunnett test is performed on the arc sine transformation of the corresponding
184 mortality data at 95% confidence interval to determine which maximum concentration
185 exhibited statistically insignificant difference from the control.

186 **Results**

187 ***Bench side observation***

188 The fingerlings remained very active in swimming and feeding in the course of
189 acclimatization. No fish has died in the aquaria and no one has exhibited any abnormal
190 operculum beating as signs of stress or suffocation. In few moments following loading of
191 fish to the aquaria, few fingerlings jumped out but were promptly picked up and returned.
192 Afterwards, they became calm and relaxed within 2-3 hours. Unlike the control groups,
193 fish exposed to the test solutions are constantly coming to the air water interface and
194 continuously gulp. They usually come to the surface and gulp in groups or rest at the
195 bottom unlike the control groups which were freely moving and chasing each other.

196 ***Physicochemical Properties of the Test Water***

197 Lower level of dissolved oxygen was detected in all test solutions other than the control
198 in every measurement. On the other hand, pH measurements of the first day were higher
199 than those in the other days. Standard deviation of the temperature recorded was less than
200 the 1.5 maximum limit set by OECD for a valid test [29]. In addition, all dissolved oxygen
201 measurements were above 60%.

202 ***Range Finding Test Results***

203 In the range finding test, the 100, 400 and 800 mg/L dilutions did not kill any fish. Rather,
204 fish death was observed at 1200 and 1600 mg/L solutions where, the former killed two

205 and the later killed all the three. Based on the result, five equidistant concentrations
 206 between 800 and 1600 mg/L were suggested for the definitive test. The researchers
 207 discussed on it and decided to include a 600 mg/L dilution and avoided the upper most
 208 1600 mg/L level. The underlying rationale was that, as the exposure time extends from
 209 24 to 96 hours, fish would become weaker due to prolonged starvation and hence, their
 210 resistance declines. On the other hand, prolonged exposure could aggravate toxic effects.
 211 Thus, serial dilutions of 600, 800, 1000, 1100, 1200 and 1400 mg/L were set for the
 212 definitive test.

213 ***Definitive Test Results***

214 The definitive test was started with the expectation that, the extended exposure time (96
 215 hours) will increase toxicity of the extract. As a result, fish mortality will be higher in the
 216 third and fourth day than in the first and second. However, contrary to this, the mortality
 217 data shows that most fish deaths were recorded in the first 48 hours (Table 1).

218 Table 1. Cumulative mortalities of *O. niloticus* fish fingerlings in different extract
 219 concentrations during 24,48, 72 and 96 hours of exposure

Test (mg/L)	conc. exposed	Number of fish	Fish cumulative mortalities at different time intervals			
			24 hours	48 hours	72 hours	96 hours
0 (Control)	14		0	0	0	0
600	14		0	0	0	0
800	14		0	0	0	0
1000	14		0	0	0	1
1100	14		0	1	1	1
1200	14		3	5	5	5
1400	14		3	6	7	9

220

221 When the test solutions were discarded at the end, a large amount of fish faeces is found
 222 decanted at the bottom of the containers but only in the experimental groups. No fish has

223 died in the control as well as in the lower 600 and 800 mg/L dilutions nor did the highest
 224 1400 mg/L concentration kill all fishes. Dying fish gradually loose activity and become
 225 sluggish, unable to escape when touched. They continuously swirl about unguided and
 226 finally sank down.

227 Table 2. Lethal concentrations and confidence intervals resulted from probit analysis

Lethal concentrations	Concentration in (mg/L)	Confidence interval (at 95%)	χ^2
LC1	897.43	642.48-1006.01	1.02
LC10	1063.87	903.01-1140.89	
LC50	1310.74	1229.53-1484.42	

228

229 Concentrations that can cause 1% and 10% mortalities as well as median lethal
 230 concentration with corresponding upper and lower limits, in 96-hour exposure were
 231 determined from the mortality data by probit analysis (Table 2). Within the 95%
 232 confidence interval, the calculated χ^2 is less than the corresponding tabular value asserting
 233 the model's goodness-of-fit.

234 The response proportions (RP), or survival proportions and the corresponding arc sine
 235 square root transformed values in each test solution of the two replicas are summarized
 236 in table 3. The results of Dunnett test on the arc sine square root transformed RP data
 237 showed that the highest concentration in which survival of the fingerlings is not
 238 significantly different from the control is the 1100 mg/L solution ($p>0.05$). This level is
 239 therefore, the NOAEC.

240 Table 3. Arc sine transformation of survival proportion data

Replicate	RP in each test concentration and the control
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		Control	600 mg/L	800 mg/L	1000 mg/L	1100 mg/L	1200 mg/L	1400 mg/L
Raw data	A	1.0	1.0	1.0	1.0	1.0	0.7	0.4
	B	1.0	1.0	1.0	0.9	0.9	0.6	0.3
Arc sine value	A	1.3807	1.3807	1.3807	1.3807	1.3807	1.0069	0.7137
	B	1.3807	1.3807	1.3807	1.1832	1.1832	0.8531	0.5640
	Mean	1.3807	1.3807	1.3807	1.2820	1.2820	0.9300	0.6389
	Var	0	0	0	0.0195	0.0195	0.0118	0.0112

241

242 *Phytochemical Analysis of A. aspera Leaf Ethanolic Extract*

243 The leaf powder was extracted by 85% ethanol using microwave assisted extraction
244 technique. The resulting extract was analysed by GC-MS and revealed 12 major
245 phytoconstituents (Fig. 1.).

246 **Discussion**

247 *Achyranthes aspera* is a well-known medicinal plant traditionally used for ailments of
248 various diseases in many parts of the world [14]. In addition, the molluscicidal potential
249 of this plant is recently discovered [17] But assessment on its negative toxic impact on
250 non target species is mandatory before applied in the aquatic environment for snail
251 control.

252 Here, the toxic effect of *A. aspera* leaf aqueous extract on an important fish species, *O.*
253 *niloticus* fingerlings, was studied to evaluate the possible adverse impact on survival of
254 such non-target and important species in case of its application against aquatic vector
255 snails. As a result, the NOAEC was determined to be 1100 mg/L ($p > 0.05$) and the LC₅₀
256 was 1310.74 mg/L. In addition, the respective LC₁ and LC₁₀ values were 897.43 mg/L
257 and 1063.87 mg/L.

258 Regarding the toxicity effect of different molluscicidal plants on fishes, *Sapindus*

259 *mukorossi* exhibited an LC₅₀ of 10 ppm while its molluscicidal LC₅₀ is 119.57 ppm [34].
260 *Jatropha gossypifolia* had piscicidal LC₅₀ of 10.490 mg/L [35] while its molluscicidal
261 LC₅₀ is over 100 ppm [36]. [36] The piscicidal LC₅₀ of Endod (*Phytolacca dodecandra*)
262 is 4.4 mg/L and its molluscicidal LC₅₀ is 10 ppm [37, 6]. Similarly, *Carica papaya*
263 exhibited piscicidal LC₅₀ of 700 ppm against *O. mossambicus* fish while its molluscicidal
264 LC₅₀ is from 619.1 to 2716.3 ppm [38]. These data indicate the molluscicidal median
265 lethal dose (LC₅₀) of each plant exceeds its respective piscicidal LC₅₀. And hence, they
266 are more toxic to those nontarget fishes than to the target snails. Such problem in
267 selectivity and target specificity limits application of these natural products for snail
268 control in habitats where fishes and snails co-exist.

269 On the contrary, the current study showed that the piscicidal LC₅₀ of *A. aspera* leaf
270 aqueous extract to Nile tilapia fingerlings is 1310.74 mg/L. Similarly, the NOAEC level
271 is 1100 mg/L. These concentrations are much higher than its molluscicidal LC₅₀ which is
272 72.4 ppm, according to Mandefro et al., [18]. The considerable gap between the
273 molluscicidal and piscicidal lethal doses of *A. aspera* indicates that the plant has very low
274 or negligible toxic effect on this economically and ecologically valuable fish, *O. niloticus*,
275 especially at its molluscicidal dose limits.

276 Several phytochemical analyses of the plant have been carried out by different
277 researchers, and they were able to identify different classes of saponins [11, 39, 40]. But
278 in the current GC-MS analysis, saponins are not detected (Fig.1). However, some
279 compounds identified are molluscicidal by nature. For example, *cis-p-mentha-1(7),8-*
280 *dien-2-ol* is an oxygenated monoterpene, a terpenoid saponin moiety [41]. The
281 molluscicidal effect of terpenoids is indicated in many literature [4, 42]. In addition, the
282 larvicidal property of eicosanoic acids, and phthalates is indicated in many literatures. In

283 general, bioactivity of crude extracts usually arises from synergistic effect of several
284 constituents.

285 In all measurement instances of the test water physico chemical parameters, lower
286 dissolved oxygen level was recorded in test solutions than in the control. This
287 phenomenon agrees with the studies of Ayuba et al.,[32] and Ojutiku et al., [43].
288 Biodegradation or decomposition of the phytochemicals leads to higher biological and
289 chemical oxygen demand (BOD, COD) and causes depletion of oxygen in the solutions
290 [30, 43].

291 Fingerlings in the test solution were seen motionless resting at the bottom or incessantly
292 gulping at the water air interface. These behavioural changes also happened in many
293 similar studies [27, 35, 44]. The reactions could be due to dissolved oxygen depletion by
294 the chemicals leading to oxidative stress [43]. It can also be a manifestation of neuro
295 toxicity and poisoning of the gills by the toxicants [35, 44].

296 **Conclusion**

297 As a general principle, molluscicides are directly applied in to the snail infested water
298 body where many other non-target and valuable species co-exist. As a result, they are
299 always subjected to deleterious toxicity by such chemicals applied in to their common
300 habitats. To mitigate such ecological damage, the molluscicide should be selective in
301 action or the dose limit applied for snail control is proved to be safe to such non-target
302 species.

303 Molluscicidal plants application in the field can be promoted when molluscicidal efficacy
304 studies are supported by eco toxicological safety data. This particular study showed that
305 *A. aspera* aqueous extract is almost nontoxic to *O. niloticus* fingerlings and is safe to this

306 fish when applied under its molluscicidal dose limit which is below 100ppm. Therefore,
307 it is possible to conclude that application of the plant product for snail control at
308 molluscicidal effective concentrations of below 100 ppm does not acutely toxify this fish
309 species. However, further studies involving additional non-target groups such as aquatic
310 invertebrates and mammals should be conducted to generate a more complete
311 ecotoxicological information.

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325 **Availability of data and materials**

326 All data generated or analyzed during this study are included in this
327 published article.

328 **Authors' contributions**

329 BM: Contributed in the conception of the study, carried out the experiment,

330 and wrote the manuscript. STM: Modified the experimental design,
331 supervised the experiment process, and edited the manuscript. AA: Carried
332 out the statistical analysis, organized literatures and edited the final version
333 of the manuscript. YT: Did the phytochemical analysis section and edited the
334 revised manuscript. WM: Carried out water physicochemical measurements and edited
335 the manuscript. DF: Did the phytochemical analysis using GC-MS. All authors read and
336 approved the final manuscript.

337 **Ethics approval**

338 The internal review board (IRB) of college of Public Health, Jimma University,
339 has issued ethical clearance for this study.

340 **Consent for publication**

341 Not applicable

342 **Competing interests**

343 All authors of this revised manuscript declare that they have no competing
344 interest regarding this research

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462 **Tables and Figures**

463 Figure 1. Phytoconstituents of *A. aspera* leaf ethanolic extract analysed by GC-MS.

- 464 (a) Cyclohexane, 1-methyi-4-(2-hydroxyethyl) C₉H₁₈O; (b) Benzene, (1-methylpropyl)
465 C₁₀H₁₄; (c) Cis-p-mentha-1(7),8-dien-2-ol C₁₀H₁₆O; (d) 1,2-Benzenedicarboxylic acid,
466 butyl 8-methylnonyl ester C₂₂H₃₄O₄; (e) Octadecanoic acid, 2-hydroxy-1,3-
467 propanediylester C₃₉H₇₆O₅; (f) Oleic acid,eicosyl eter C₃₈H₇₄O₂; (g) Benzene,1,3-diethyl

468 $C_{10}H_{14}$; (h) Naphthalene,2-methyl $C_{11}H_{10}$; (i) Naphthalene,1,7-dimethyl $C_{12}H_{12}$;
469 (j) Eicosanoic acid $C_{20}H_{40}O_2$; (k) Benzene, 1,2,3-trimethyl C_9H_{12} ; (l) Oleic acid
470 $C_{18}H_{34}O_2$.

471

Figures

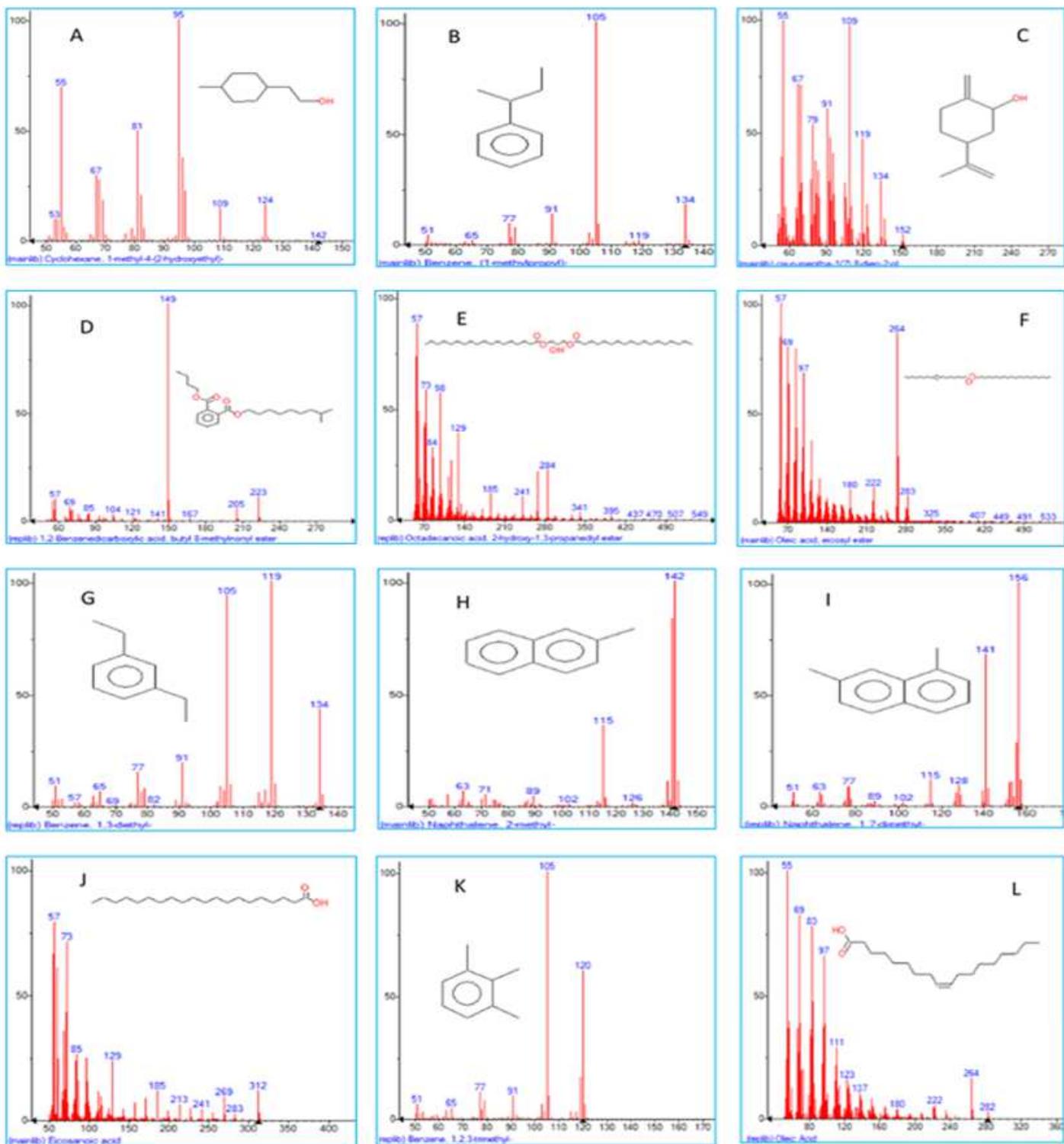


Figure 1

Phytoconstituents of *A. aspera* leaf ethanolic extract analysed by GC-MS. (a) Cyclohexane, 1-methyl-4-(2-hydroxyethyl) C₉H₁₈O; (b) Benzene, (1-methylpropyl) C₁₀H₁₄; (c) Cis-p-mentha-1(7),8-dien-2-ol C₁₀H₁₆O; (d) 1,2-Benzenedicarboxylic acid, butyl 8-methylnonyl ester C₂₂H₃₄O₄; (e) Octadecanoic acid, 2-hydroxy-

1,3-propanediylester C₃₉H₇₆O₅; (f) Oleic acid,eicosyl eter C₃₈H₇₄O₂; (g) Benzene,1,3-diethyl C₁₀H₁₄; (h) Naphthalene,2-methyl C₁₁H₁₀; (i) Naphthalene,1,7-dimethyl C₁₂H₁₂; (j) Eicosanoic acid C₂₀H₄₀O₂; (k) Benzene, 1,2,3-trimethyl C₉H₁₂; (l) Oleic acid C₁₈H₃₄O₂.