

# Catalyst H-ZSM 5 in Amines addition on Carbonyl Compounds $\alpha$ , $\beta$ -unsaturated catalyzed by H-ZSM 5

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## Research Article

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# Abstract

## Background

Zeolites are crystallized aluminum silicates. Their structure consists of an arrangement of TO4 tetrahedral (T = Si, Al ...), creating a regular nonporous system made up of channels with a diameter of less than 1 nm and cages (the channel intersections). These materials were discovered by the mineralogist in 1756. These porous materials are used in the paper industry, in jewelry, and in other fields. Certain applications require pure materials with specific properties, hence the need to develop their synthesis. Zeolites are obtained hydrothermally at temperatures below 200°C. The dimensions of the pores are of the order of a nanometer, close to the dimensions of certain organic molecules which make these zeolitic materials find applications in various fields, separation by molecular sieving, purification by adsorption, cation exchange and catalysis.

## Result

We are successful in the reaction of addition the series of amines to unsaturated alkenes, catalyzed by zeolite H-ZSM-5, the results show good products obtained with excellent yields, in short time, pure products can be obtained after filtering the catalyst and then removing the excess amines, the catalyst is recoverable, reactivated and reused several times for other addition reactions, the products were defined by  $^1\text{H}$  NMR spectroscopy.

## Conclusion

According to the results obtained. It can be said that the addition of various aliphatic or aromatic amines reacted successfully with alkenes, in the presence of the catalyst H-ZSM-5, to give the corresponding adducts with excellent yields in time and in perfect conditions, solvent-free and room temperature, so our H-ZSM-5 catalyst is effective, can meet our goal of adding amines to unsaturated carbonyl components.

## 1. Introduction

Zeolites are natural (volcanic rock) or synthetic materials [1]. There are about fifty natural zeolites and more than 150 synthetic zeolites [2]. The activity and the selectivity of the catalyst are influenced by the characteristics of the catalyst when these characteristics depend on the method of preparation [3]. A method often used in the preparation of the catalyst is the impregnation method [4]. Zeolites are stable at high temperatures and have a broad porous structure [5], which opens new perspectives for the treatment of large molecules especially in the field [6] of petroleum refining and petrochemical catalysis processes [7].

The addition of a series of amines to  $\alpha$ ,  $\beta$ -unsaturated alkenes, [8] an interesting methodology for [9] for the efficient coupling of amines with a wide range of  $\alpha$ ,  $\beta$ -unsaturated carbonyl components [10], to obtain the excellent products show the role of a material like HMZ5 in reactions in organic chemistry [11].

With this addition we can synthesize similar products, where other reactions do not work

The use of this addition of amines to  $\alpha$ ,  $\beta$ -unsaturated alkenes in vast areas such as biology [12], medicine, agriculture.

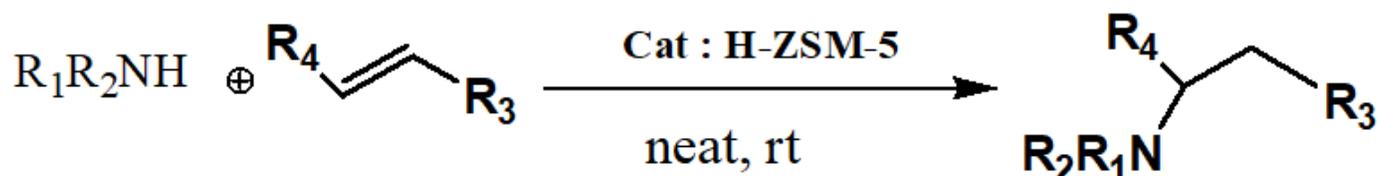
Following the addition Michael [13], there is a practical method has been developed for the addition of aliphatic or aromatic amines to alkenes [14], catalyzed by a heterogeneous catalyst [15], in this sense we can be carried out our addition amines on  $\alpha$ ,  $\beta$ -unsaturated alkenes [16].

A comparative study shows that the aza-Michael addition reaction [17] (Scheme1) leads to the formation of the C-C or C-heteroatom bond. From these results [18], we carried out a series of amine addition on  $\alpha$ , $\beta$ -unsaturated alkenes [19], active or not active in a dry medium catalyzed by zeolite H-ZSM-5 [20]. Table 1 shows the 1, 4 - addition of a series of amines on  $\alpha$ , $\beta$ -unsaturated alkenes, ate catalyzed by zeolite H-ZSM-5 [21], in dry medium and a temperature T= 25°C.

## 2. Materials And Methods

In an Erlenmeyer flask, we add (1eq) of alkenes, to the (3eq) of amines with stirring, we add a quantity of (0.1) mg of H-ZSM-5 catalyst at an ambient temperature T = 25 ° C, and without solvent.

The reaction is followed by C.C.M, the products obtained are recovered by filtration and washed with dichloromethane, and the excess of the starting reagents are removed using a rotary evaporator. The products obtained are analyzed by proton nuclear magnetic resonance,  $^1\text{H}$  N.M.R and represented in Figure 01 (N1-(3-bromopropyl) ethane-1, 2-diamine), Figure 02(3-(2-aminoethylamino)-3-phenylpropan-1-ol), Figure 03(-(3-chlorophenylamino)-3-phenylpropan-1-ol).



### Scheme1: Aza-Michael addition

## 3. Spectral Data

## N.M.R. Spectra of Product Entry 1 of table

RF= 0.80 (Hexane-EtoAc): 2/1

RMNH<sup>1</sup> (CDCl<sub>3</sub>, 300 MHz) d: 1.91(m, J=5.2 Hz, CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-Br) 2.0 (s, 3H, NH<sub>2</sub>, NH) ; 2.55 (t, 2H, NH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-Br) ; 2.77-2.81 (m, J=7.6 Hz, 4H, NH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-NH) , 3.30 (t, 2H, CH<sub>2</sub>-CH<sub>2</sub>-Br)

## N.M.R. Spectra of Product Entry 02 of table

RF= 0.82 (Hexane-EtoAc): 2/1

RMNH<sup>1</sup> (CDCl<sub>3</sub>, 300 MHz) d: 2.0 (s, 4H, NH<sub>2</sub>, NH, OH) ; 1.8-2.81 (m, J=5.4 Hz, 4H, NH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-NH) ; 2.77 (t, J=7.6 Hz, 2H, OH-CH<sub>2</sub>-CH<sub>2</sub>) 2.90(t, 1H, NH<sub>2</sub>-(CH-Ph)-CH<sub>2</sub>) ; 7.08-7.21(m, 5H, Ph)

## N.M.R. Spectra of Product Entry 03 of table

RF= 0.70 (Hexane-EtoAc): 2/1

RMNH<sup>1</sup> (CDCl<sub>3</sub>, 300 MHz) d: 1.5 (m, 7 Hz, 6H, m, p, CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-C<sub>5</sub>H<sub>10</sub>N) ; 2.4 (t, J=5.6 Hz, 4H, ortho, N-CH<sub>2</sub>-CH<sub>2</sub>-N C<sub>5</sub>H<sub>10</sub>N) ; 2.5 (t, J=7.6 Hz, 2H, CH<sub>2</sub>-CH<sub>2</sub>-CO<sub>2</sub>CH<sub>3</sub>) 2.6-2.7 (t, J=7.2 Hz, 2H, N-CH<sub>2</sub>-CH<sub>2</sub>) ; 3.65 (s, 3H, CH<sub>3</sub>).

## 4. Results And Discussion

We are successful in the reaction of addition the series of amines to unsaturated alkenes, catalyzed by zeolite H-ZSM-5, the results show good products obtained with excellent yields, in short time, pure products can be obtained after filtering the catalyst and then removing the excess amines, the catalyst is recoverable, reactivated and reused several times for other addition reactions, the products were defined by <sup>1</sup>H NMR spectroscopy.

## 5. Conclusion

According to the results obtained. It can be said that the addition of various aliphatic or aromatic amines reacted successfully with alkenes, in the presence of the catalyst H-ZSM-5, to give the corresponding adducts with excellent yields in time and in perfect conditions, solvent-free and room temperature, so our H-ZSM-5 catalyst is effective, can meet our goal of adding amines to unsaturated carbonyl components.

## Declarations

## Ethics approval and consent to participate

We, the authors of the manuscript, acknowledge ethical approval and consent to share

## **Consent for publication**

Yes, we agree to publish

## **Availability of data and materials**

There is no data with the manuscript

## **Competing interests**

The work is very important in the biochemistry and synthetic he carried out to synthesize a product which destroys the fungus of date palm disease

All authors declare that there is no conflict of interest, or funding from any party for this work.

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## **Authors' contributions**

### **1 Author**

Implementation of the idea, and carrying out this work

### **2 Author**

Work in the laboratory on the reactions

### **3 Author**

Correct mistakes

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## Table

Due to technical limitations, table 1 is only available as a download in the Supplemental Files section.

## Figures



Figure 1

N1-(3-bromopropyl) ethane-1, 2-diamine

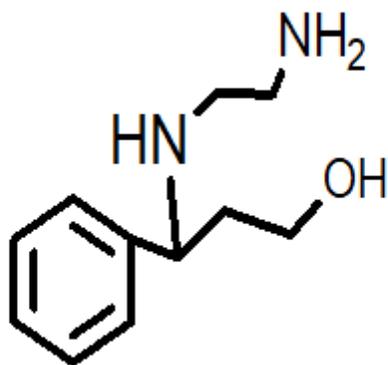


Figure 2

3-(2-aminoethylamino)-3-phenylpropan-1-ol

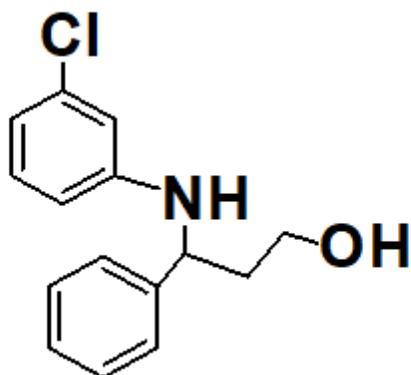
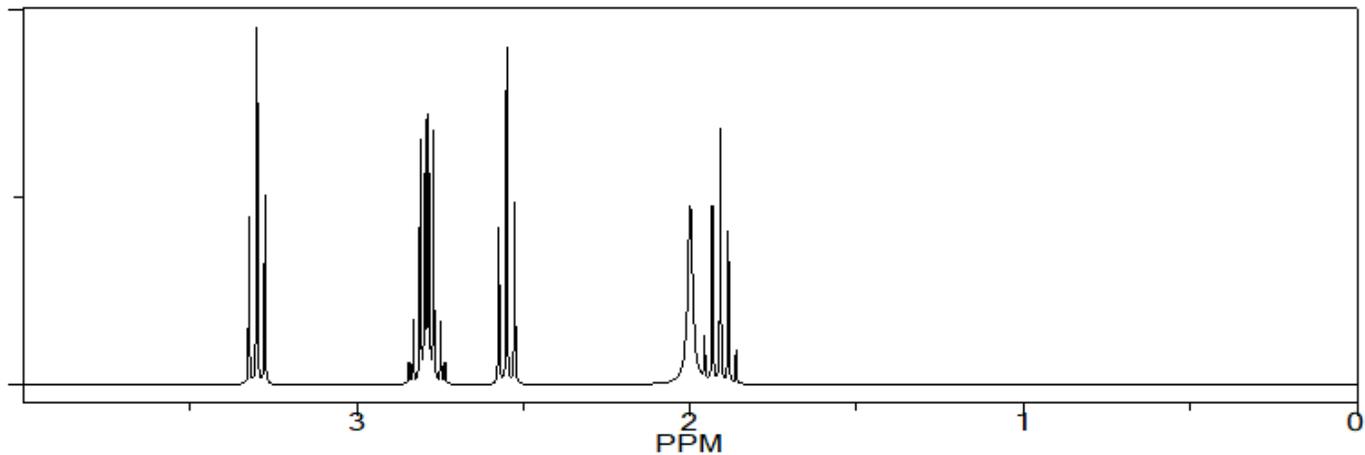


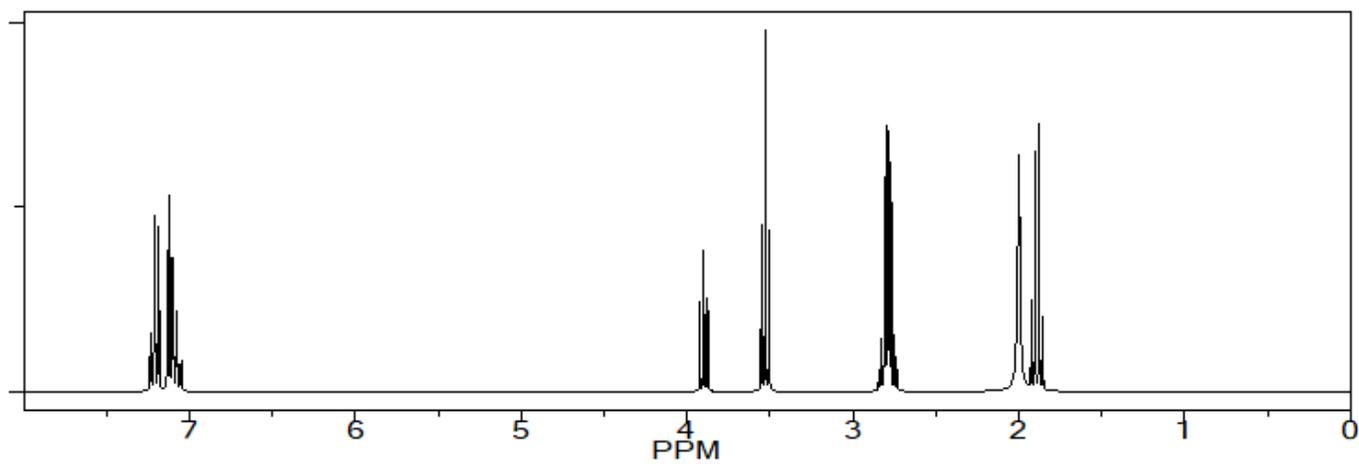
Figure 3

3-(3-chlorophenylamino)-3-phenylpropan-1-ol



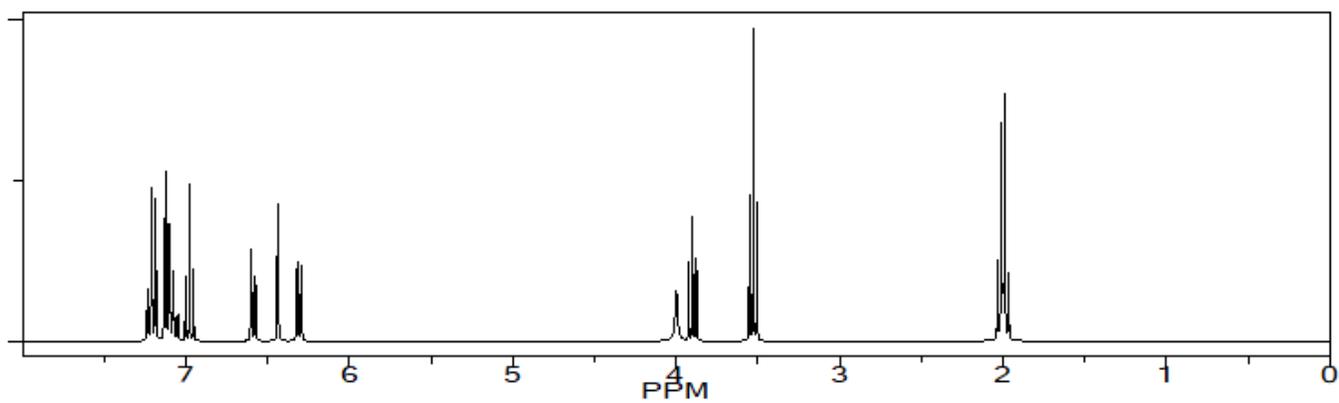
**Figure 4**

N.M.R. Spectra of Product Entry 1



**Figure 5**

N.M.R. Spectra of Product 02



**Figure 6**

N.M.R. Spectra of Product 03

## Supplementary Files

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