Field Verification of Low-Level Biochar Applications as Effective Ameliorants to Mitigate Cadmium Accumulation into *Brassica Campestris* L from Polluted Farmland Soil

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Abstract

Farmland soils in China have been reported to be diffusely contaminated, Cd has been recognized as a significant contributor to this issue and biochars have been reported to be effective in mitigating soil Cd pollution. However, previous studies have shown contradictory outcomes. Furthermore, in general, laboratory experiments and unrealistically large amounts of biochar (>10 t/ha) have been used. In this study, three biochars: rice straw biochar (RS), pig manure biochar (PM) and rice husk biochar (RH) were produced from readily available farm residues and characterized. These were used in a field experiment, at low applications rates of 1.8 and 3.6 t/ha, with rape (*Brassica campestris* L.). Batch adsorption experiments indicated Cd adsorption in the order RS biochar > PM biochar > RH biochar. Field experiment indicated biochar amendments to slightly changes in soil pH and CEC; yet led to considerable and significant decreases in extractable Cd concentrations (reductions of: 43%-51% (PM), 29%-35% (RS) and 17%-19% (RH)). Reduced extractable Cd correlated with lower Cd concentrations in rape plants. PM and RS biochars were the most effective in decreasing Cd phytoaccumulation into edible parts of rape (>68% reduction). It is highlighted that biochars were produced using a pyrolysis unit with an output of 20 ton/yr. Thus, assuming a working application rate of 2 ton/ha, the pyrolysis unit could service 10 ha/yr. While at a modest scale, this research demonstrates the genuine reality of biochar-based remediation solutions to contribute to the mitigation of diffuse Cd contamination in some of China's impaired farmland.

Highlights

- Biochar-based amelioration to the mitigation diffuse Cd contamination in farmland
- Pig manure (PM) and rice straw (RS) biochar amendment resulted in lower Cd concentrations in rape
- Rice husk (RH) biochar was less effective in lowering Cd concentrations in rape plants
- Low application rates, 1.8 t/ha of PM and RS, efficiently immobilized Cd in soil
- Doubling field application rates of biochar increased Cd immobilization by 10%

Introduction

Due to industrialization and urbanization, soil cadmium (Cd) pollution is a global issue (Liao et al. 2015). Often this pollution is localized in areas of industrial activity. However, long range transport from industrial sources and the use of tainted water resources for irrigation have resulted in farmland accumulating high levels of contaminant (Lu et al. 2015a). In China the latest official soil surveys conclude that 19.4% of national farmland soils (based on the sampling points) has been contaminated (MEP 2014). Cd, along with Hg, As, Cu, Pb, Cr, Zn and Ni, are recognized to be priority risk drivers. Cd can migrate from contaminated soils to food crops, which can significantly increase human health risk associated with Cd poisoning (Ochoa et al. 2020). There is therefore an urgent need to develop efficient and scalable techniques for the remediation of farmland diffusely polluted with potentially toxic elements (PTEs).

Biochar is the product of the pyrolysis of organic solid wastes under anoxic or limited oxygen conditions. Biochars can be produced from a wide range of organic feedstocks including crop straw, manure, wood chip, and sewage sludge (Wang and Wang 2019). Generally, biochar is alkaline, and contains abundant surface functional groups (Kloss et al. 2012). Several types of biochars have been reported to be efficient in immobilizing heavy metals through mechanisms such as adsorption, surface precipitation and electrostatic interaction (Chen et al. 2015; Alam et al. 2018). Biochar influence on Cd availability has been extensively studied over the last decade. For example, Lu et al. (2015b), Rehman et al. (2018), Li et al. (2019) and Bashir et al. (2020) all reported decreases in Cd bioavailability and reduced Cd accumulation into plants following biochar application to soil. In addition, biochars have been reported to improve soil properties, enhance soil fertility and increase crop yields (Laird et al. 2017). Thus, biochar amendment is considered as a cost-effective strategy for soil improvement and pollution abatement.

However, the properties of biochars can vary widely according to feedstock type used and the pyrolysis conditions applied (Liu et al. 2017; Qi et al. 2018; Zong et al. 2020). Zhao et al. (2013) produced twelve biochars pyrolyzed at 500°C and showed that both their surface areas and cation exchangeable capacities (CEC) varied considerably from 2.78 m²/g to 203 m²/g and 41.7 cmol/kg to 562 cmol/kg, respectively. As a consequence of these contrasting properties, the twelve biochars exhibited different capacities for metal ion adsorption (Xu et al. 2013; Bashir et al. 2018). In addition, it has been reported that different biochar types applied to soil give rise to contrasting changes to the soil properties such as pH and CEC (Alburguerque et al. 2014; Zhang et al. 2016; Yuan et al. 2020). These soil factors further influence metal ion behaviour (Lu et al. 2017; Azhar et al. 2019). For instance, Novak et al. (2014) showed pecan shell biochar (produced at 350 °C) to increase soil pH from 5.6 ± 0.1 to 6.3 ± 0.1, while addition of poultry litter biochar (also produced at 350 °C) increase pH in the same soil to 8.4 ± 0.1. In another study, Fellete et al. (2014) reported extractable (using diethylenetriamine pentaacetic acid (DPTA)) Cd concentrations from soils amended with orchard residue biochar to decrease by > 70%, while for fir tree biochar, Cd extractability decreased by only 12%.

As briefly outlined above, most of these studies were conducted in laboratory or greenhouse settings. However, there is to date little information and guidance on field-scale *in-situ* applications of biochar. Moreover, where field trial results have been reported they have shown inconsistent outcomes, (Cui et al. 2016; O'Connor et al. 2018); these different outcomes being attributed to variable field conditions and contrasting biochar properties. There are also issues pertaining to the realism of biochar application rates applied in many of the published reports. Many authors have used biochar at very large application amounts, for example, 10–40 t ha⁻¹ (Chen et al. 2016; Cui et al. 2016; He et al. 2019). In a few studies, application rates of biochar have been even higher (up to 100–180 t/ha; Chan et al. 2008; Jeffery et al. 2011; Rajkovich et al. 2011; Rondon et al. 2014). Given the availability of biochar resource and the associated cost of biochar application, such large rates of biochar application at a field scale would likely be unfeasible. Importantly, it has been reported that decreases in metal availability, following biochar application, are not proportional to the amount of biochar applied (Bian et al. 2014); and, in Jeffery's meta-analysis (1483 studies), a poor relationship between biochar application rate and crop yield was

reported ($r^2 = 0.1$; Jeffery et al. 2011). It might therefore be concluded that high application rates of biochar may not be necessary to achieve desired outcomes. Thus, there is a need to evaluate biochar efficacy to mitigate Cd phytoaccumulation using low application rates of biochar under field relevant conditions.

In this study, three biochars including rice straw biochar (RS), rice husk biochar (RH) and pig manure biochar (PM) were prepared and evaluated for their Cd adsorption capacities in the laboratory. The biochars were then used, in a field experiment at two low application rates (1.8 and 3.6 t/ha), under a crop of rape (*Brassica campestris* L.), to: i) evaluate their influence upon the accumulation of Cd from soils into the rape roots and edible parts; ii) to explore the mechanisms influencing Cd phytoaccumulation, and; iii) to provide evidence to assist the selection of biochar feedstock and biochar application rate for utilisation to abate risks associated with Cd pollution in soil.

Materials And Methods

2.1 Biochar preparation

Rice straw biochar (RS), rice husk biochar (RH) and pig manure biochar (PM) were produced using pilot-scale biochar pyrolysis equipment (L 4.5 m · W 0.8 m · H 2 m). The pyrolysis equipment comprised a pyrolysis chamber, a heating chamber, a feedstock supply system and a biochar output system. The equipment had an overall production capacity of 20 ton per year. Feedstocks were pyrolyzed in the pyrolysis chamber at 500°C for 1 hour under limited oxygen conditions. Thereafter, the biochars were ground through a 2 mm sieve. The physico-chemical properties of the three biochars were assessed using methods described in previous work (Zhang and Luo 2014; Zhang et al. 2016). Cd adsorption capacities of the biochars were characterized using batch equilibration adsorption experiments (full details are provided in the supporting information). Briefly, biochar (50 mg) was shaken with 25 mol/L – 500 mol/L $\text{Cd}(\text{NO}_3)_2$ solution (5 mL) at 25°C at 150 rpm for 48 hours. Thereafter the suspensions were collected, and the concentration of Cd measured by inductively coupled plasma optical emission spectrometer (ICP-OES) (Optima 7000DV; PerkinElmer Inc.).

2.2 Field study

The field experiment was conducted on an upland farm located in Zhouzai village (24°23'26.08"N, 117°43'26.25"E), Zhangzhou city, Fujian province in southern China. Local soils had been contaminated with potential toxic elements (PTEs); Cd being the primary pollutant. The concentration of Cd in the soil was 0.38 mg/kg; this concentration exceeded the regulatory limit for agricultural soils (0.30 mg/kg; MEP 2018).

In November 2016, soil (0–20 cm) was ploughed. Biochars were then spread on the soil surface, and thoroughly mixed with the soils using a tillage machine (1GQN-120; Weifang shengxuan machines corporation). The applications of biochar were 0, 1.8 and 3.6 t/ha, respectively. Each treatment was produced in triplicate. The area of each plot was 4 m². Two weeks after biochar application, the land was

rolled, and rape (*Brassica campestris* L.) seeds were sown. Irrigation and fertilizer management were performed according to the conventional practices of the local farmers and were identical on all plots.

2.3 Sampling and analysis

In January 2017, the rape plants were harvested. Three composite rape samples, each consisting of 10 plants randomly selected from each plot, were collected (resulting in nine composite rape samples for each treatment regime). Following their transfer to the laboratory, the rape samples were washed with deionized water. Thereafter, the rape samples were cut into two parts (the edible part and the root), and dried in the oven at 90°C. For the analysis of Cd concentrations in the plants, subsamples of the edible part, or root, were crushed, ground, and passed through a 0.2 mm sieve. Samples of the edible part or root were then digested with HNO₃ (65%, Merck, EMSURE™) using a protocol modified from Zhu et al. (2008). For quality control, plant reference material (GBW 10015; purchased from the National Research Center for Standards in China) was digested and measured alongside the experimental samples, as were reagent blanks. The concentration of Cd in the digestates were analysed using an inductively coupled plasma mass spectrometry (ICP-MS 7500cx; Agilent Inc.). Concentrations of Cd were reported on dry weight basis. The recovery of Cd from the reference material was 108–113%.

Nine soil samples (0–20 cm) were collected from each treatment. Soil samples were air-dried in the laboratory at room temperature, and then ground and passed through a 2 mm sieve. Subsamples of soils were ground and passed through a 0.15 mm sieve. The pH of soil samples was measured using a pH meter at the ratio of 1 g soil : 2.5 mL deionized water. The cation exchangeable capacity (CEC) of soil was measured using the NH₄OAc–NaOAc method (U.S. Environmental Protection Agency 1986). The concentrations of available Cd in soils were determined using a 0.01 mol/L CaCl₂ extraction method (Houba et al. 1996). Briefly, soil samples were extracted with 0.01 mol/L CaCl₂ solution at the ratio of 1 g soil : CaCl₂ solution (10 mL). After 2 hours shaking, the suspensions were centrifuged, and the concentrations of Cd in the supernatant were measured using inductively coupled plasma mass spectrometry (ICP-MS 7500cx; Agilent Inc.).

2.4 Data management and Statistic analysis

All data were analysed using Microsoft Excel. The fitting of batch adsorption data was performed with OriginLab 2018 software (OriginLab Corp., USA). Statistical analysis of the data was performed with IBM SPSS Statistics 22.0 software (IBM Corp., USA). Significant differences among treatments were analyzed using one-way analysis of variance (ANOVA) (at $p < 0.05$ level).

Results

3.1 Biochar characterisation

Properties varied across the three biochar types (Table 1). pH was always alkaline and ranged between 9.87 and 10.7; RS biochar had the highest pH (10.7). Carbon content in PM biochar was 20%, while the C

content in RS biochar or RH biochar was much higher (> 40%). N and O contents in PM biochar were the highest amongst the three biochars (1.5% and 20%, respectively), while the N and O contents in RH biochar were lowest (0.6% and 11%, respectively). The ash content of PM biochar was particularly high (68%; this accounting for the lower C content observed). The ash contents in the RS and RH biochars were much lower, 36% and 35%, respectively. The surface topography and pore structure of the biochars also varied (Table 1 and Figs. S1-S2). The surface area of biochars followed the order of RS biochar > PM biochar > RH biochar, and the average pore volume of biochars followed the order of PM biochar = RS biochar >>> RH biochar. The average pore size of PM biochar (12.7 nm) was the highest of the three biochars; average pore size in the RS and RH biochars were similar (~ 9.5 nm). Based on these chemical and physical properties, it was hypothesised that the RS biochar (with high pH, high carbon content and high surface area) would offer the best performance for Cd immobilisation (Chen et al. 2011; Shen et al. 2017; O'Connor et al. 2018).

Table 1
Biochar properties

	PM biochar	RS biochar	RH biochar
pH	9.87 ± 0.01 ^c	10.7 ± 0.2 ^a	10.0 ± 0.1 ^b
pH _{zpc}	9.48	9.53	8.91
C (%)	20.5 ± 0.6 ^a	41.7 ± 0.7 ^b	42.2 ± 0.1 ^b
N (%)	1.51 ± 0.03 ^a	1.06 ± 0.03 ^b	0.62 ± 0.01 ^c
O (%)	20.1 ± 0.7 ^a	12.0 ± 1.0 ^b	11.1 ± 0.4 ^b
H (%)	1.25 ± 0.03 ^a	2.01 ± 0.07 ^b	2.95 ± 0.05 ^c
Ash (%)	67.9 ± 0.4 ^a	35.9 ± 0.4 ^b	34.7 ± 0.2 ^b
Surface topography and pore structure			
BET surface area (m ² /g)	4.13	4.92	2.28
Average pore volume (m ³ /g)	0.014	0.012	0.006
Average pore size (nm)	12.7	9.45	9.72
Surface functional groups			
Carboxyl group (mmol/kg)	0.25 ± 0.0001 ^b	0.63 ± 0.13 ^a	0.25 ± 0.001 ^b
Phenolic group (mmol/kg)	0.50 ± 0.25 ^a	0.38 ± 0.18 ^a	1.17 ± 0.14 ^b
Lactonic group (mmol/kg)	0.75 ± 0.01 ^b	0.87 ± 0.13 ^b	1.33 ± 0.14 ^a
Total acid functional group (mmol/kg)	1.50 ± 0.25 ^a	1.88 ± 0.18 ^a	2.75 ± 0.25 ^b
Total base functional group (mmol/kg)	2.38 ± 0.13 ^b	7.75 ± 0.01 ^a	0.50 ± 0.001 ^c
dissimilar letters indicated significant differences (p < 0.05) for a given measurement			

Results of the batch equilibrium experiment indicated that Cd adsorption capacity of the three biochars followed the order: RS biochar > PM biochar > RH biochar (Fig. 1 and Table S1). Cd sorption onto both PM biochar and RH biochar were better fitted by the Freundlich isotherm model (rather than Langmuir isotherm model), while the RS biochar was fitted well with both the Langmuir and Freundlich isotherm models (Table S1). Given the model agreements, it was concluded that Cd adsorption onto the three biochars was likely attributable to chemical adsorption on heterogeneous surfaces (Zhang and Luo 2014). The value of 1/n in the Freundlich isotherm model followed the order of RS biochar < PM biochar < RH biochar. Since 1/n represented the adsorption affinity of metal ions onto adsorbent, this result

confirmed that the adsorption of Cd^{2+} ion onto RS biochar was the greatest, while RH biochar showed lowest affinity for Cd ions.

3.2 Effects of biochar amendment on the properties of field soils

The pH in biochar amended soils were increased (from 5.06 ± 0.03 in the control, to 5.26 ± 0.03 – 5.44 ± 0.06 (PM), 5.30 ± 0.07 – 5.30 ± 0.19 (RS) and 5.27 ± 0.06 – 5.37 ± 0.14 (RH)) (Fig. 2a). However, no significant difference ($p > 0.05$) in soil pH were observed among three biochars.

CEC in the PM and RS amended soils were, respectively, 23.9 cmol/kg – 24.8 cmol/kg and 23.0 cmol/kg – 24.3 cmol/kg ; these values were slightly higher, but not significantly different ($p > 0.05$), to that in the control (Fig. 2b). In contrast, soil amended with RH biochar had a lower, but not significantly different ($p > 0.05$), CEC (20.0 cmol/kg – 20.3 cmol/kg).

Available Cd concentration in the control soil was $21.5 \pm 1.9 \text{ } \mu\text{g/kg}$. Following biochar amendment, the available Cd concentrations were significantly lowered in all treatments ($p < 0.05$). However, greatest reductions in bioavailable Cd in soil were observed with PM amendment (43% and 51% reduction at 1.8 and 3.6 t/ha) and RS amendment (29% and 35% reduction at 1.8 and 3.6 t/ha) (Fig. 3). In the treatments with RH biochar amendment, bioavailable Cd concentration was reduced by only 17% and 19% in the 1.8 t/ha and 3.6 t/ha treatments, respectively.

3.3 Effects of biochar amendment on the growths of rape plants in filed

Biochar amendment influenced the growth of rape plants. Increases in dry weight of edible parts (Fig. S3a) and dry weight of roots (Fig. S3b) were observed in all instance following biochar amendment. However, these increases in biomass were marginal, and none were significantly different ($p > 0.05$) when compared to the control (except for the dry weight of root, in the treatment with 1.8 t/ha amendment of PM biochar).

3.4 Effects of biochar amendment on Cd accumulation into plants

In all instances, biochar amendment resulted in a decrease in the concentration of Cd in rape plant (Fig. 4). For the edible part of rape (Fig. 4a), the concentrations of Cd in the treatments with PM biochar were significantly ($p < 0.05$) decreased to $0.52 \pm 0.08 \text{ mg/kg}$ (1.8 t/ha) and $0.50 \pm 0.03 \text{ mg/kg}$ (3.6 t/ha); these levels were 70% and 68% of the control value. In the RS treatments, the Cd concentrations were significantly ($p < 0.05$) decreased to $0.51 \pm 0.09 \text{ mg/kg}$ (1.8 t/ha) and $0.51 \pm 0.02 \text{ mg/kg}$ (3.6 t/ha), respectively; these values were 70% of the control value. Following RH amendment, the concentrations of Cd in the edible part of rape were $0.66 \pm 0.04 \text{ mg/kg}$ (1.8 t/ha) and $0.67 \pm 0.03 \text{ mg/kg}$ (3.6 t/ha); these values were approximately 89% of the control value.

Regarding Cd concentrations in roots, treatments with PM biochar indicated decreased concentrations (0.60 ± 0.12 mg/kg (1.8 t/ha) and 0.55 ± 0.04 mg/kg (3.6 t/ha), compared to 0.74 ± 0.30 mg/kg in the control (Fig. 4b). Similarly, the Cd concentrations in the treatments with RS biochar and RH biochar were also observed to decrease (to 0.50 mg/kg – 0.52 mg/kg and 0.51 mg/kg – 0.65 mg/kg, in 1.8 t/ha and 3.6 t/ha, respectively). Where application rate was equivalent, no significant differences ($p > 0.05$) were observed across the three biochar types.

Discussion

The C content of the RS and RH biochar ($\sim 42\%$) was more than 2 times that of the PM biochar; while the content of N in PM biochar was 1.9 times the content in RS biochar, and 3.3 times the content in RH biochar (Table 1). In agreement with previous reports, the biochars prepared from manure contained lower C and higher N than the biochars prepared from plant residues (Xu et al. 2013; Zornoza et al. 2016). Such differences have previously been attributed to differences in the element composition of feedstocks (Zhao et al. 2013). In addition, the three biochars had different surface topography and pore structure (Fig. S1). The surface area, average pore volume and average pore size of PM biochar and RS biochar were all higher than those for RH biochar (Table 1), indicating PM biochar and RS biochar offer higher pore structures than RH biochar. This finding is in good agreement with previous literature highlighting that contrasting feedstock sources produced biochars with different physical and chemical properties (Sun et al. 2014; Hyväluoma et al. 2018).

The properties of biochar including surface area, distribution of surface functional groups, ash content and pH, play important roles during metal adsorption (Uchimiya et al. 2011; Jiang et al. 2016). As a consequence of differing physical and chemical properties, the sorption capacities of the three biochars varied (Fig. 1). Complexation of metals, through ion exchange interactions, with ionized surfaces and oxygen-containing functional groups (i.e. carboxyl ($-\text{COOH}$), hydroxyl ($-\text{OH}$), phenol ($\text{R}-\text{OH}$) groups) has been suggested as an important mechanism for metal sorption by biochar (Kołodziejńska et al. 2017). In this study, RS biochar had highest surface area and highest content of surface functional groups (Table 1), available to prompt interactions with metal ions. Similarly, since the BET surface area and the content of surface functional groups of RH biochar were lowest among three biochars (Table 1), the sorption of Cd^{2+} ions onto RH biochar was correspondingly lower. In addition, the pH of three biochars also influenced the equilibrium pH and thus the adsorption of Cd ions (Fig. S4). The sorption of more Cd^{2+} ion with increasing pH is consistent with Zhang and Luo (2014) who also reported this relationship. In the batch experiments, the addition of RS biochar with highest pH lead to highest equilibrium pH among three biochars (Table S3); this, likely, underpinned the greatest sorption of Cd^{2+} ions onto RS biochar. Overall the RH equilibrium pH was lower than RS or PM, and Cd^{2+} ion sorption was also lower.

When amended to soil, biochars increased soil pH in all instances, but the increases were not significantly different ($p > 0.05$) among the three biochars (Fig. 2a). In addition, biochar had limited influenced on soil CEC (Fig. 2b). These outcomes are likely due to the low amendment levels (1.8 and 3.6 t/ha). Nonetheless, all three biochars significantly ($p < 0.05$) reduced extractable Cd concentrations (decreases

followed the order: PM biochar > RS biochar > RH biochar) (Fig. 3). It is suggested, therefore, that changes in Cd availability were most likely linked to Cd ion interaction with biochar (rather than changes to the soil chemical environment). Reduced concentrations of available Cd were translated into observed reductions in Cd content in rape plants (Fig. 4 and Table S2). Importantly, the three types of biochar led to different outcomes for Cd-plant interactions. Treatment with RH biochar was relatively ineffective, while amendment with PM or RS biochar resulted in much more effective abatement of soil to plant transfer of Cd. Very little difference was observed where 1.8 and 3.6 t/ha application levels of the same biochar were compared. This observation suggesting, even at the lowest application rate (1.8 t/ha), that PM and RS biochars were effective ameliorants. The literature, in many cases, has reported metal-biochar-soil-plant interactions to result in large decreases in phytoaccumulation of metals into numerous crop types (Zhang et al. 2013; Puga et al. 2015; Xu et al. 2016; Younis et al. 2016; Zhang et al. 2016; Mohamed et al. 2017); while other cases this ameliorative influence has been reported to be minimal (Fellet et al. 2014; Hu et al. 2014; Lucchini et al. 2014; Kloss et al. 2015; Ree et al. 2015; Zhang et al. 2017). For instance, Zhang et al. (2017) reported that the amendment of biochar into heavily contaminated soils to have either no effect, or even to promote Cd accumulation into lettuce shoots. Wang et al. (2019) conducted a pot experiment with four types of biochar including wood biochar, rice straw biochar, Chinese walnut shell biochar and bamboo biochar. These authors reported none of the biochar amendments to influence Cu concentration of stems, leaves and roots of moso bamboo, while Cd concentrations decreased in all cases. The Zn accumulation in roots of moso bamboo was decreased in treatments with biochar, except bamboo biochar, while only wood biochar amendment reduced Zn concentration in plant stems and leaves. However, in the present study, RS and PM biochars (applied at low-levels: 1.8 and 3.6 t/ha) were established to be effective for the control of Cd phytoaccumulation, while RH biochar was observed to have only a limited effect.

While Cd sorption capacity of RS biochar was higher than that of PM biochar in the batch adsorption experiment, there were no significant difference between the decreased magnitude of Cd concentrations in rape plants where RS and PM treatments were compared (Fig. 4). These results highlight that the performance of these biochars in the batch adsorption experiments and in real soil systems were not consistent. This outcome is consistent with Uchimiya et al. (2010) who reported biochars produced from broiler litter manure at 350 °C (350BL) removed more Ni^{2+} and Cd^{2+} ions than biochars produced at 700 °C (700BL); but when applied (at 5% – 10 % (w/w)), to soil the 350BL treatment contained higher soluble metal concentrations than the 700BL treated soils. Although the 350BL had a higher adsorption capacity than the 700BL, its lower ability to increase soil pH underpinned the less effective immobilization of metals in soil by 350BL. Similarly, the pH increase, rather than primary Cd-biochar interaction, has also been proposed by other researchers (Houben et al. 2013; Rees et al. 2014). Thus, it is recommended that, before field scale deployment, biochar sorption capacity should be established in the presence of the soil it is intended to remediate.

In the present study, the pH of biochar amended soils were increased by 0.2–0.4 unit, while slight (although non-significant) changes in soil CEC were observed following biochar amendment. The

minimal affect is most likely due to the low level of biochar applied (1.8 t/ha – 3.6 t/ha). Given that soil chemical properties (pH and CEC) were largely unchanged it is suggested that the primary interactions between Cd and the biochars were likely responsible for the outcomes observed.

The application of RS biochar or PM biochar at low level (1.8 t/ha) was effective to mitigate the transfer of Cd from soil into crop. With a doubling in application rates, the decrease of soil available Cd concentrations were increased (on average by approximately extra 10%), but the concentrations of Cd in rape showed limited change. It is highlighted that in the present research, the soil was not heavily contaminated. The soil Cd concentration (0.38 mg/kg) only just exceeding the regulatory limit of 0.30 mg/kg (MEP 2018). It is therefore emphasized that the low application rates of biochar were directed at a small excess of Cd in the soil system (this likely underpinned the successful outcomes observed) and that the lower application rate was sufficient to accommodate the excess of Cd. It follows that should soil Cd concentrations are much higher, such an outcome might not transpire and larger applications of biochar could be needed to accommodate a greater excess of Cd in the soil system. This said, Nie et al. (2018) reported the low level (1.5–3.0 t/ha) application of sugarcane bagasse biochar decreased the concentrations of Cd, Pb and Cu in pak choi by 62–76%, 17.3–49.1% and 15–38%, respectively. In contrast to the present study, the concentrations of Cd, Cu and Pb in this experimental were 1.4 mg/kg, 278 mg/kg and 348 mg/kg; Cd being more than 4 times the regulatory limit. Overall, the results of the current research support low-level application (i.e. 1.8 t/ha) of biochar to mitigate the transfer of Cd from soil to rape plants.

It is highlighted that many of the results reported in the literature relate to biochars that are produced in small quantities in the laboratory (Albuquerque et al. 2014; Zornoza et al. 2016; Bashir et al. 2018; Azhar et al. 2019). In contrast, the present study considered biochars produced using a larger pyrolysis system. The pilot scale pyrolysis system used to produce the biochars for this present research had an output capacity of 20 ton per year. Thus, assuming an application rate of 2 t/ha, such a unit could service 10 ha p.a. It is emphasised that this scale of production, and low application rates, represents a realistic approach to support the production of biochar in quantities that would allow for meaningful field scale application. Given the extent of diffuse pollution associated with a considerable proportion of China's farmland (Lu et al. 2015a; Sun et al. 2019) and the predominance of small farms (< 0.6 ha) across much of China (particularly remote rural regions) (Zhang 2017), the collective research findings demonstrates the genuine reality of biochar-based remediation solutions to meaningfully contribute to mitigating some of the diffused contamination associated with tainted farmland.

Conclusions

Biochars derived from different feedstocks, under identical pyrolysis conditions, showed different sorption capacities for Cd²⁺ ions. Following amendment to soil, even at low application rates (1.8 t/ha), all biochars were observed to reduce the available concentrations of Cd in soil; slightly higher Cd immobilisation was observed at 3.6 t/ha (an average increase of 10%). These changes in Cd availability resulted in decreased Cd concentrations in rape plants. Biochars derived from pig manure (PM) or rice

straw (RS) led to much lower Cd concentrations in rape plants when compared to outcomes for rice husk (RH) biochar. Results underline that favorable Cd-biochar sorption capacities, established in the absence of soil under laboratory batch sorption conditions, did not necessarily translate into comparable success in metal pollution amelioration under field conditions. These results highlight the need to trial biochars, in the presence of the soil to be targeted for remediation, before full scale deployment is undertaken. Importantly, this research has validated an approach, that is relevant in terms of both biochar production rate and application rate, for meaningful engagement with the amelioration of Cd-tainted farmland in China at a realistic scale.

Declarations

Author contribution

C Cai and YC Zhang conceived and designed the study; JJ Fan and SN Lin performed the experiment; YW Hou analyzed the data; YC Zhang drafted the manuscript; BJ Reid and F Coulon revised the manuscript; C Cai approved the final version of manuscript.

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Ethics approval and consent to participate

Not applicable.

Consent for publication

Not applicable.

Availability of data and materials

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Competing interests

The authors declare no competing interests.

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Figures

Figure 1

Adsorption isotherm of Cd^{2+} ions onto PM biochar (circles), RS biochar (squares) and RH biochar (triangles).

Figure 2

Changes in the properties of soil and soil amended with biochars. “*” indicates a value to be significantly different to the control (CK) value ($p < 0.05$). Like uppercase letters indicate no significant difference ($p > 0.05$) between groups with biochar amendment at like application rates.

Figure 3

Available Cd concentrations in soil and soil amended with biochars. “*” indicates a value to be significantly different to the control (CK) value ($p < 0.05$). Like uppercase letters indicate no significant difference ($p > 0.05$) between groups with biochar amendment at like application rates.

Figure 4

Cd concentrations in rape plants grown in soil and soil amended with biochars. “*” indicates a value to be significantly different to the control (CK) value ($p < 0.05$). Like uppercase letters indicate no significant difference ($p > 0.05$) between groups with biochar amendment at like application rates.

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