

Sub-Ambient Radiative Cooling Realized Using CaCO₃ Microparticle-Based Single Layer Without Metal Reflector for Entire Day

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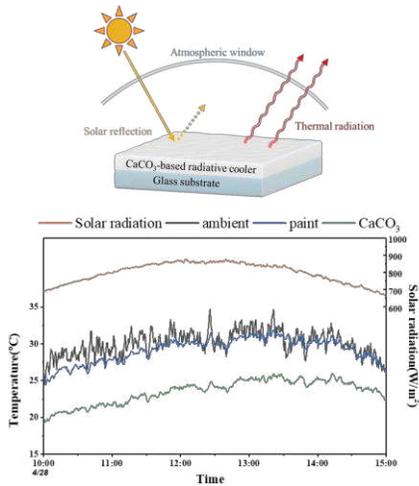
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16 **ABSTRACT**

17 Conventional cooling systems, that is, air conditioners, should be replaced because they consume
18 a substantial amount of energy and cause environmental pollution. In this context, radiative cooling
19 systems, which perform cooling without consuming any energy or causing environmental
20 pollution, are emerging as an alternative. However, most of the radiative coolers explored thus far
21 include metals, such as silver, that are used as solar reflectors, thereby entailing problems in terms
22 of practicality, mass production, cost, and light pollution. Herein, we propose calcium carbonate
23 (CaCO_3) micro-particle-based radiative cooling, which utilizes the high-energy band gap of
24 CaCO_3 for high-performance radiative cooling. As the cooler has only a single layer of a CaCO_3
25 composite without any metal reflector, it is mass-producible, cheap, and does not cause light
26 pollution. To demonstrate the cooling performance of CaCO_3 , optical properties and temperature
27 changes are measured and compared with those of commercial white paint. As a result, it is
28 demonstrated that the CaCO_3 -based radiative cooler has cooling power 93.1 W/m^2 in calculation
29 and can be cooled $6.52 \text{ }^\circ\text{C}$ and $3.38 \text{ }^\circ\text{C}$ under ambient temperature in daytime and nighttime
30 respectively. Thus, it can perform as radiative cooler in entire day.

31 **KEYWORDS:** radiative cooling, micro-particle, broadband mid-IR emitter, atmospheric transparency window,
32 passive cooling



33

34 1. Introduction

35 Cooling systems are a crucial necessity in modern industrial and residential complexes.
 36 Presently, most cooling systems incorporate air conditioners. However, air conditioners consume
 37 large amounts of energy and are a major cause of environmental pollution, such as the emission of
 38 chlorofluorocarbons, which lead to ozone depletion¹. Moreover, the most of energy used to drive
 39 the air conditioner obtained by consuming fossil fuels which emit greenhouse gases that cause
 40 global warming. Therefore, a new type of cooling system is required. In this scenario, radiative
 41 cooling is the key to an alternative cooling system. Notably, radiative cooling entails
 42 thermodynamic cooling without any energy consumption. Consequently, it does not burden the
 43 environment^{2,3}. The principle of passive radiative cooling is that the system radiates its energy in
 44 the form of infrared rays with a wavelength of 8–13 μm (referred to as the atmospheric window)
 45 and reflects sunlight which means the system is not heated by solar radiated energy^{4,5}. Unlike a
 46 conventional cooling system that transfers heat from the inside to the outside, a radiative cooler
 47 emits energy directly into space, at a temperature of 3 K, without entailing any chemical reactions
 48 or harmful effects to living beings. Because it radiates energy directly into space, it does not cause
 49 global warming, in contrast with air conditioners.

50 Substantial research on sub-ambient radiative cooling has been actively conducted over the past
51 few years, including investigating multilayer thin films^{2,5,6,7}, polymer-based radiative coolers^{8,9,10},
52 and photonic material structures^{6,11,12,13} and their various applications^{12,14,15,16,17}. However, most
53 radiative coolers include silver reflectors to reflect sunlight, which causes certain problems. Silver
54 is expensive and easily oxidized, and high vacuum equipment is required to deposit it. Oxidized
55 silver absorbs solar energy, which has a negative effect on radiative cooling. Using high vacuum
56 deposition, silver would be required to be deposited on a flat surface, which would impose a
57 constraint on the choice of the substrate; in addition, the process cost would also be high.
58 Furthermore, silver is a specular reflector that causes light pollution¹⁸. Smooth surface reflectors,
59 such as mirrors, reflect light parallelly, which can cause dizziness or safety hazards for pedestrians
60 and drivers. It can also damage the retina or iris, causing vision loss in certain cases. Owing to
61 these reasons, researchers have recently published radiative coolers without a metal reflector. In
62 order to reflect sunlight without a metal layer, white ceramic powders such as SiO₂, Al₂O₃ are
63 mixed with the polymer matrix^{19,20,21}, a porous polymer⁴ or a polymer composite²² is prepared to
64 induce scattering to increase the reflectivity. These can radiate and be cooled without metal
65 reflector, but because they are composed of a polymer matrix, they are susceptible to scratches due
66 to their low hardness. Therefore, in this study, the performance of a ceramic powder-based
67 radiative cooling device using CaCO₃ microparticles(MPs) will be discussed.

68 CaCO₃ is inexpensive and has excellent thermal and mechanical stability. Consequently, it is
69 used as a filler in several composite materials. Further, it is mainly used in plastics^{23,24,25,26},
70 dyes^{27,28}, and rubber.²⁹ It is also widely used in agricultural chemicals^{30,31}, medicine^{32,33}, food³⁴,
71 and construction materials^{35,36}. Moreover, due to its large band gap of 5 eV, it does not absorb

72 most of solar irradiance. We demonstrate the radiative cooling performance of CaCO_3 by
73 comparing it to that of a commercial white paint.

74 2. Principle of radiative cooling

75 The equation for radiative cooling is derived using the law of conservation of energy as
76 follows^{23,37}:

$$77 P_{\text{net}}(T) = P_{\text{rad}}(T) - P_{\text{atm}}(T_{\text{atm}}) - P_{\text{sun}} - P_{\text{non-rad}} \quad (1)$$

78 where

$$79 P_{\text{rad}}(T) = \int_0^{2\pi} \int_0^{\frac{\pi}{2}} \int_0^{\infty} I_{\text{BB}}(T, \lambda) \varepsilon(\lambda, \theta) \cos \theta \sin \theta d\lambda d\theta d\varphi \quad (2)$$

80 represents the hemispherical radiation power emitted from the radiative cooler.

$$81 P_{\text{sun}} = \int_0^{\infty} I_{\text{AM1.5}}(\lambda) \varepsilon(\lambda, \theta) d\lambda \quad (3)$$

82 is the solar irradiance power absorbed by the radiative cooler.

$$83 P_{\text{atm}}(T_{\text{atm}}) = \int_0^{2\pi} \int_0^{\frac{\pi}{2}} \int_0^{\infty} I_{\text{BB}}(T_{\text{atm}}, \lambda) \varepsilon(\lambda, \theta) \varepsilon_{\text{atm}}(\lambda, \theta) \cos \theta \sin \theta d\lambda d\theta d\varphi \quad (4)$$

84 is the power of the incident atmospheric radiation. According to Kirchhoff's law of thermal
85 radiation, for these equations, absorption and radiation are the same owing to thermodynamic
86 equilibrium.

$$87 P_{\text{non-rad}} = h_c(T_{\text{atm}} - T) \quad (5)$$

88 denotes the power lost by the radiative cooler because of convection and conduction.

89 Here, I_{BB} represents the blackbody radiation intensity and is expressed using the formula

$$90 I_{\text{BB}} = (2hc^2/\lambda^5) \times (e^{hc/\lambda k_B T} - 1)^{-1}, \text{ and } \varepsilon_{\text{atm}}(\lambda, \theta) = 1 - [t(\lambda)]^{1/\cos \theta} \text{ represents the}$$

91 angular atmospheric emissivity, where $t(\lambda)$ denotes the atmospheric transmittance in the
92 zenith direction. The constants h_c, h, c, k_B represent the heat transfer coefficient, the Planck
93 constant, speed of light, and Boltzmann constant, respectively. The cooling temperature is
94 defined as $\Delta T_{\text{cool}} = T - T_{\text{atm}}$, which is calculated by extracting the cooling temperature
95 under the condition $P_{\text{net}}(T) = 0$. As previously reported, T_{atm} is replaced by T_{amb} ^{38,39}. All
96 the calculation processes of the net cooling power and cooling temperature were performed
97 using MATLAB software based on absorptivity and emissivity values, with a wavelength
98 spacing of 0.01 μm .

99

100 **3. Experimental**

101 **3.1. Source materials**

102 dipentaerythritol penta-/hexa-acrylate, 1-hydroxycyclohexyl phenyl ketone 99% (Irgacure
103 184, photoinitiator), calcium carbonate powder (20–30 μm in diameter), and 1-methyl-2-
104 pyrrolidinone (NMP) were purchased from Sigma Aldrich. For comparison, commercial
105 white paint was acquired from Okong Company in spray form.

106

107 **3.2. Sample preparation**

108 CaCO_3 MP (5 g) was mixed in NMP (intrinsic solvent) (5 g) and stirred at 200 rpm with a
109 magnetic stirring bar for 30 min so that it would not agglomerate with each particle. After the
110 powders were dispersed, the DPHA monomer (0.25 g) and UV initiator (Irgacure 184) (0.025
111 g) were added to the solution and stirred at 200 rpm for 2 h. The glass substrate was treated
112 with UV-ozone, the prepared solution was dropped on it, and the glass with the solution was

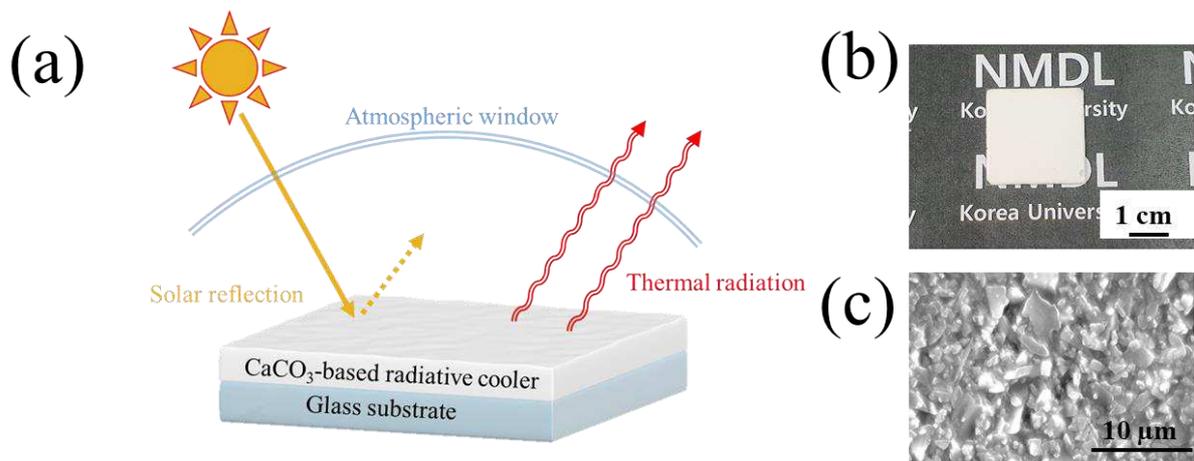
113 heated on a 100 °C hot plate for 2 h to evaporate the NMP solution. After the solvent had
114 dried completely, it was exposed to a UV lamp for 30 min to polymerize the DPHA.
115 To prepare paint-coated glass, the glass was spray painted to make it approximately 200 µm
116 thick.

117

118 **3.3. Sample Characterization**

119 An X-ray powder diffraction (XRD) scan (SmartLab, Rigaku, Japan) was performed using a
120 diffractometer and a Cu target (wavelength: 1.5412 Å) as the X-ray source at 9 kW. The surface
121 of the radiative cooler was observed at 15 keV using FESEM (Regulus8100, Hitachi, Japan).
122 Optical properties within a wavelength range of 3–2.5 µm were measured using a UV-Vis-NIR
123 spectrophotometer (Solidspec-3700, Shimadzu, Japan) in the integral sphere using Spectralon
124 diffuse reflectance standards (SRS-99-010, Labsphere, USA) as a baseline reflector. The
125 optical properties of the IR spectra (wavelength range of 3–15 µm) were measured through
126 Fourier transform infrared spectroscopy (FT-IR; Nicolet IS-50, Thermo Scientific, USA) and
127 an integral sphere (Mid-IR IntegratIR™, PIKE technologies, USA) using an Au reflector. The
128 absorbance A was determined using the equation $A=1-R-T$ after measuring the reflectance R
129 and transmittance T using the equipment described.

130 **4. Results and discussion**

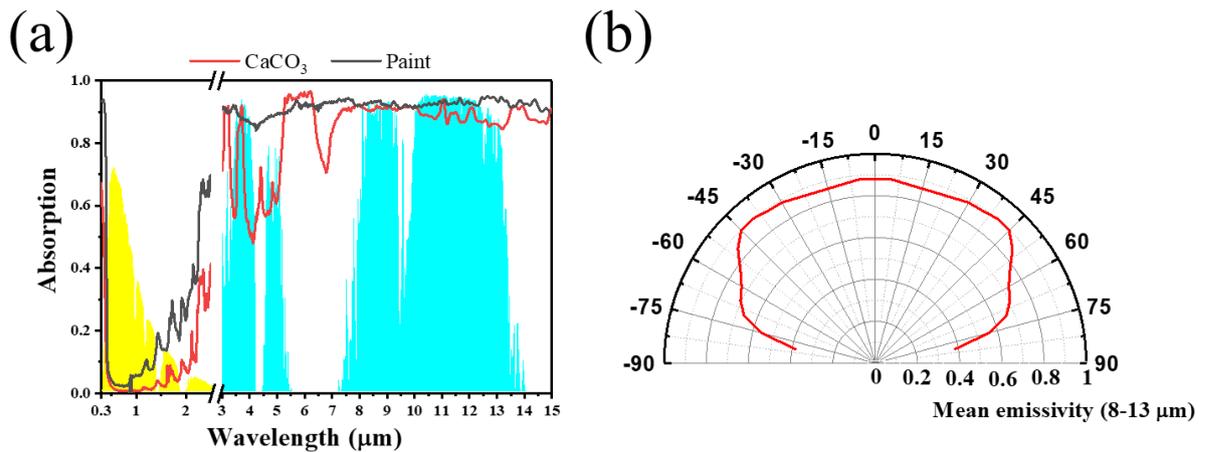


131 **Figure 1.** (a) Schematic of radiative cooling principle and radiative cooler structure. (b) Photo image and (c) SEM
 132 top view of radiative cooler, CaCO₃-coated glass.
 133

134

135 Figure 1(a) shows the operating principle of radiative cooling and structure of CaCO₃-based
 136 radiative cooler. By reflecting sunlight, it avoids being heating and sends heat energy to the
 137 outside by radiating infrared rays in the wavelength range corresponding to the atmospheric
 138 window, cooling to a lower temperature than the ambient temperature. As depicted in Figure
 139 1(b), the CaCO₃ layer is coated on glass placed over a black background, and the letters in the
 140 background are not visible, indicating that the transmittance is sufficiently low in the visual
 141 check. Figure 1(c) displays the surface of the CaCO₃ layer analyzed through field emission
 142 scanning electron microscopy (FESEM). Figure S2 depicts the absorption and transmittance of
 143 CaCO₃ with the dipentaerythritol penta-/hexa-acrylate (DPHA) binder with respect to the
 144 thickness. With an increase in the thickness, the transmittance tends to decrease. However, the
 145 absorption did not increase until the thickness reached 350 μm. Further, at 500 μm, the
 146 absorption started increasing, and approximately 8% of the solar incident power was absorbed.
 147 The emissivity within the atmospheric window region does not differ significantly based on
 148 the thickness. The ratio of absorbed solar incident power, average emissivity in the atmospheric

149 window region, and calculated net cooling power at 300 K in accordance with the thickness
 150 are listed in Table S1. The cooling power is the highest at a thickness of 130 μm ; however,
 151 because the transmittance in the solar radiation region is greater than 5%, the thickness of 350
 152 μm , at which the transmittance is low and the absorption does not increase significantly, is
 153 considered to be the most suitable and is adopted as the thickness used in this experiment.



154

155 **Figure 2.** (a) Spectral absorptivity and emissivity of CaCO₃- and commercial white-paint-coated glass. (b)
 156 Average emissivity of CaCO₃-coated glass in the atmospheric window (8–13 μm) with respect to angle θ .

157 Figure 2(a) presents the absorption properties of CaCO₃ and white-paint-coated glass. In the
 158 atmospheric window region, CaCO₃ and the paint have emissivities of 89.6% and 92.7%,
 159 respectively. The paint has a slightly better emissivity property. However, in the solar radiation
 160 region, CaCO₃ has better absorption properties than the paint. CaCO₃ absorbs 4.1% of the solar
 161 incident power, whereas the paint absorbs 11.1%. This difference in solar absorption affects
 162 the cooling performance. Using these optical properties, at 300 K, the net cooling power was
 163 calculated as 93.1 W/m^2 for CaCO₃ and 38.6 W/m^2 for paint. The calculation shows that CaCO₃
 164 demonstrates 2.4 times higher cooling performance than the paint. The ratio of absorbed solar
 165 incident power, average emissivity in the atmospheric window region, and net cooling power
 166 are presented in Table 1. Figure 2(b) depicts the average emissivity in the atmospheric window

167 region in accordance with the incident light angle. The decrease in emissivity starts at
168 approximately 50°, and the emissivity remains greater than 70% until the angle of 70° is
169 reached.

170 **Table 1.** Average absorption (0.3-2.5 μm), emissivity (8-13 μm) and net cooling power at 300 K

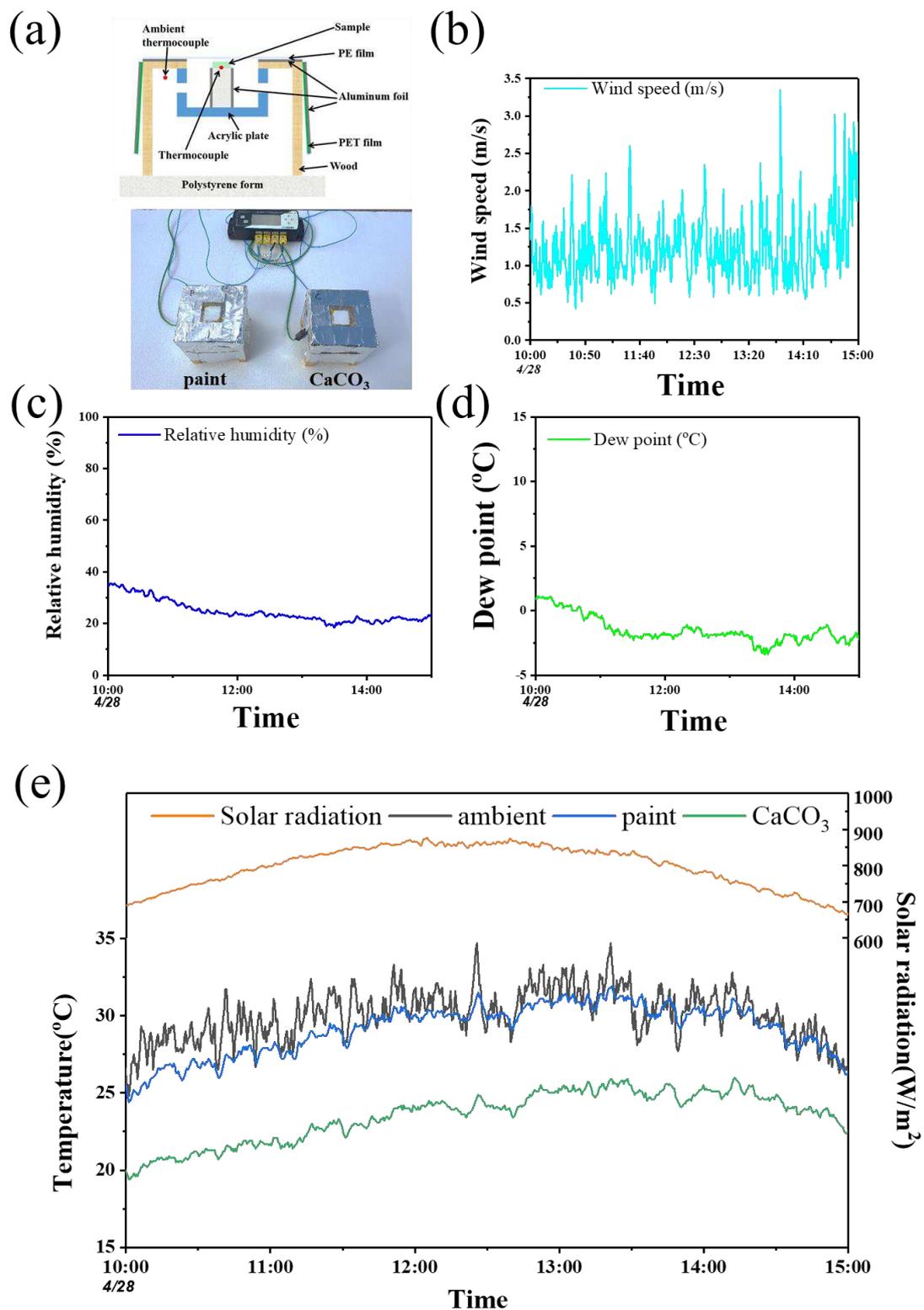
Sample	Solar power absorption ratio (0.3-2.5 μm)	Average emissivity (8-13 μm)	Net cooling power at 300 K (W/m ²)
CaCO ₃	0.041	0.896	93.1
paint	0.111	0.927	38.6

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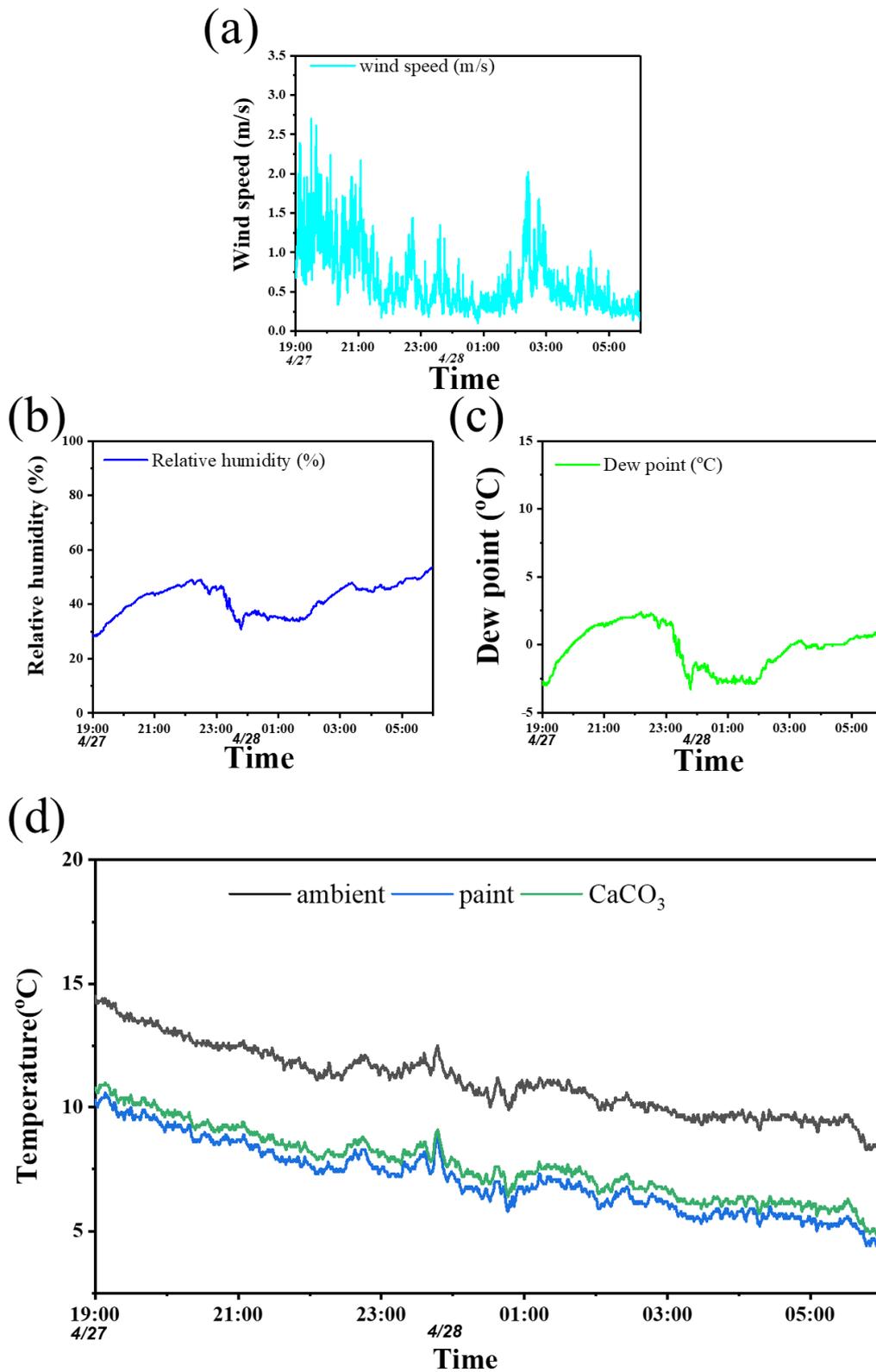
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Figure. 3. (a) Schematic diagram and photo image of outdoor measurement system for radiative cooling. Environmental conditions (b) wind speed, (c) Relative humidity, (d) dew point. (e) Radiative cooling performance of CaCO₃ and white paint in terms of temperature in daytime.

179

180 To measure the variation in the temperature of the surroundings over time, a measurement
181 chamber was prepared. A schematic of the chamber is presented in Figure 3(a). Acrylic plates
182 are attached to a frame made of wood, except for the top side of the frame. To prevent
183 conduction between the chamber frame and radiative cooler, a polystyrene pillar was attached
184 to the center of the acrylic plate, and a radiative cooler sample with K-type adhesive
185 thermocouples (ST-50, RKC INSTRUMENT INC., Japan) attached to the floor was placed on
186 the pillar. Several holes were drilled into the sides of the acrylic plates to allow the air to reach
187 an ambient temperature similar to the outside temperature; further, to reduce the influence of
188 wind, Polyethylene Terephthalate (PET) films were attached to the side of the frame. The top
189 side of the chamber was covered with low-density polyethylene (LDPE) to maximize radiative
190 cooling effect by preventing the influence of the wind, but light of almost all wavelengths could
191 be transmitted through the top side⁴⁰⁴¹. Finally, all the sides of the chamber were covered with
192 aluminum tape to reflect the light and prevent the chamber from being heated by the incident
193 solar radiation. A copper wire thermocouple was attached to the inside of the PET film, such
194 that the tip of the wire did not contact anything, to measure the ambient temperature. Although
195 the LDPE film has excellent transmittance, it does not have perfect transmittance, as depicted
196 in Figure S3(d), which causes a greenhouse effect that increases the inner temperature. Because
197 of the greenhouse effect, holes were drilled into the acryl plates to maintain a temperature
198 similar to that of the surroundings. The measurements were conducted on the roof of a new
199 engineering hall at Korea University, Seoul, South Korea. The temperature of the samples and
200 the ambient temperature were recorded every 30 s using a data logger (OM-CP-OCTTEMP-A,
201 OMEGA Engineering, USA). The solar irradiance, wind speed, relative humidity, and dew
202 point were also recorded every 30 s through a weather station (HD52.3D, DeltaOHM, Italy).

203 Figure 3(b), (c), and (d) depict the external daytime environmental conditions of wind speed,
204 relative humidity, and dew point data, respectively, over time. The temperature variations in
205 the CaCO₃- and paint-coated glasses, ambient temperature, and solar irradiation power from
206 10:00 to 15:00 are depicted in Figure 3(e). The figures indicate that the weather is clear, not
207 cloudy, and that the fluctuation in solar irradiation is not large. The temperature of the paint-
208 coated glass was lower by an average of 0.93 °C when compared with the ambient temperature
209 from 10:00 to 15:00. This cooling is not significantly meaningful, as the difference in
210 temperature was less than 1 °C. On the other hand, the temperature of the CaCO₃-coated glass
211 was lower than the ambient temperature by an average of 6.52 °C during the same time. Based
212 on optical properties, it is clear that paint can emit more energy than CaCO₃ through the
213 atmospheric window. However, it absorbs a substantial amount of solar radiated power, which
214 negatively affects the radiative cooling performance. On the other hand, although CaCO₃-
215 coated glass has lower emissivity than paints, it absorbs less solar radiated power. Therefore,
216 CaCO₃ can exhibit more cooling than that shown by paint under incident solar radiation.



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218
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Figure 4. Environmental conditions (a) wind speed, (b) Relative humidity, (c) dew point. (d) Radiative cooling performance of CaCO₃ and white paint in terms of temperature in nighttime.

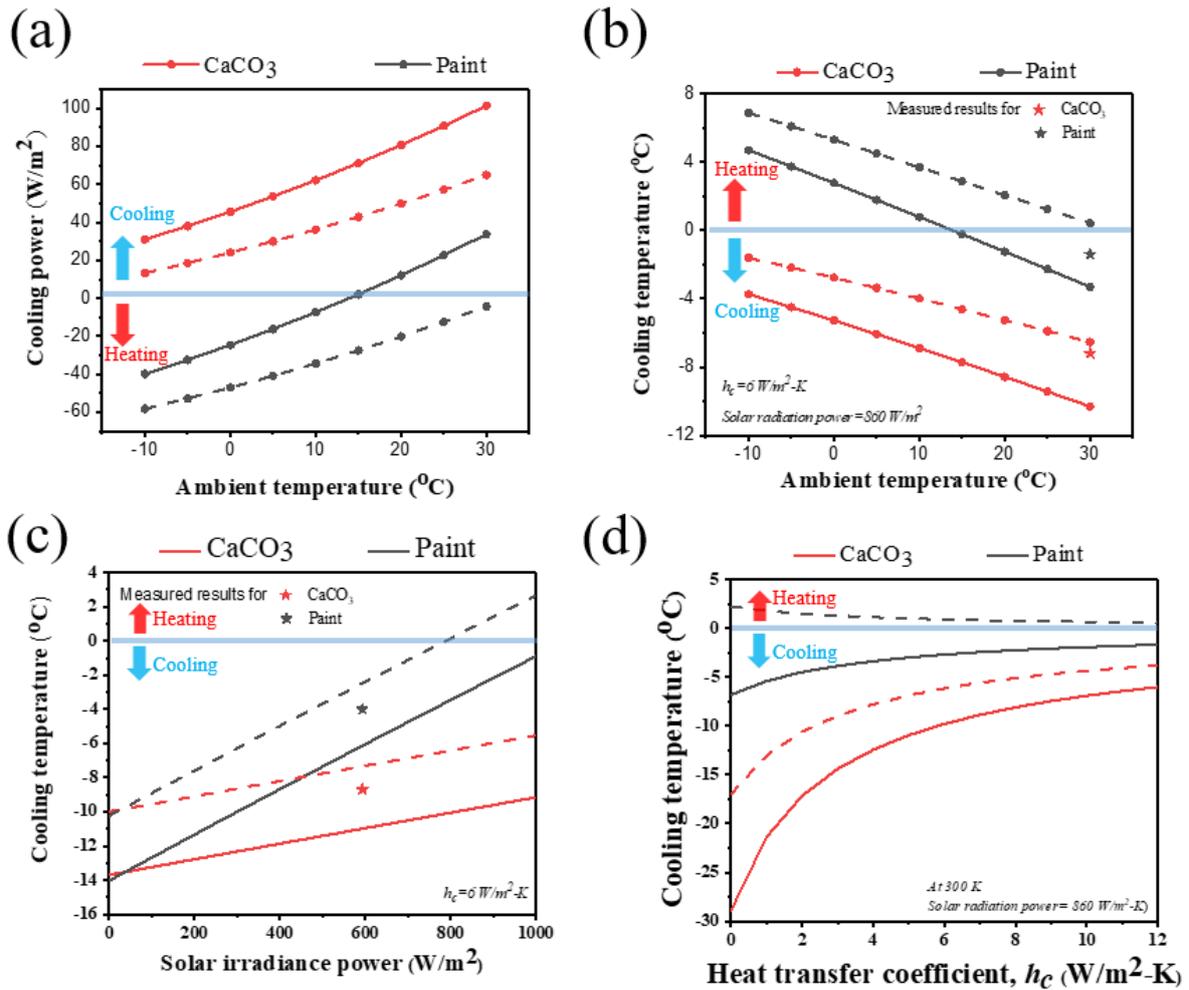
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221 Similarly, the wind speed, relative humidity, and dew point at nighttime (19:00 to 5:00) are
222 presented in Figure 4(a), (b), and (c), respectively. Figure 4(d) depicts the radiative coolers and
223 the ambient temperature. Because it is nighttime, the solar irradiance is 0, and the heating
224 caused by solar irradiance does not affect the performance of radiative cooling: only emission
225 does. Therefore, paint, which has a higher emissivity than that of CaCO₃, exhibits slightly
226 better cooling performance—CaCO₃ cooled by 3.38 °C and paint cooled by 3.93 °C. The
227 average cooling temperatures of the CaCO₃- and paint-coated glasses during daytime and
228 nighttime are listed in Table 2. The values indicate that both the samples can exhibit cooling to
229 below the ambient temperature during both daytime and nighttime.

230

Table 2. Average cooling temperature in daytime and nighttime.

Sample	Average cooling temperature (10:00-15:00)	Average cooling temperature (19:00-06:00)
paint	-0.93 °C	-3.93 °C
CaCO ₃	-6.52 °C	-3.38 °C



231

232 **Figure 5.** (a) Calculated cooling power of CaCO₃ and (b) cooling temperature in accordance with ambient
 233 temperature under the condition of $h_c = 6 \text{ W/m}^2\text{-K}$ using ideal (solid line) and mid-latitude winter (dotted line)
 234 atmospheric transmittance and measured cooling temperatures are represented with stars (red for CaCO₃/glass,
 235 gray for paint/glass). (c) Calculated temperatures of CaCO₃- and paint-coated glass in accordance with the solar
 236 irradiance power under the condition of $h_c = 6 \text{ W/m}^2\text{-K}$ and measured cooling temperatures are marked with stars
 237 (red for CaCO₃, gray for paint). (d) Calculated temperatures of CaCO₃- and paint-coated glass in accordance with
 238 convection coefficient h_c at 300 K using AM1.5 solar irradiance spectrum using ideal (solid line) and mid-latitude
 239 (dotted line) atmospheric transmittance.

240 Figure 5(a) and (b) depict the calculated cooling power and temperature in accordance with the
 241 ambient temperature at a steady state. The solar irradiance density and atmospheric
 242 transmittance used in the calculation are depicted in Figure S3. The heat transfer coefficient is
 243 assumed to be $6 \text{ W/m}^2\text{-K}$ in consideration of the wind speed (about 1-1.5 m/s) in Figure 3(b)
 244 under condition of the wind cover according to the literature equation⁴⁰. Under ideal
 245 transmittance, CaCO₃ and the paint can cool by approximately 10 °C and 4 °C, respectively,

246 and they have cooling powers of 93.1 and 38.6 W/m², respectively, at 300 K. However, the
247 measured results indicate a lower cooling performance when compared with the calculated
248 results. This is attributed to the effect of the weather at the time of the measurement and the
249 climatic characteristics of the mid-latitude region as well as the LDPE film, which make the
250 atmospheric transmittance lower than the ideal one. Based on the mid-latitude atmospheric
251 transmittance obtained by calculation using MODTRAN 6, the results indicate that CaCO₃ and
252 the paint can cool by approximately 6 and 0.6 °C, respectively, and that they have cooling
253 powers of 60 and 6 W/m², respectively. Compared with the actual measurement, these results
254 indicate a more similar cooling performance when using the mid-latitude atmospheric
255 transmittance, in contrast with the results obtained assuming ideal transmittance. Owing to the
256 difference between the assumed (888 W/m²) and actual solar irradiances (< 888 W/m²), the
257 measured cooling temperature is slightly higher than the calculated data. According to the
258 Stefan–Boltzmann law, the total energy radiated from a blackbody is proportional to the fourth
259 power of temperature. As depicted in Figure 3 and Figure 4, even though there is no sunlight
260 at night, the cooling temperature is less than that during the daytime because, based on this law,
261 the cooler radiates less energy. Figure 5(a) and (b) also indicate the same tendency, in
262 accordance with this law. Figure 5(c) depicts the cooling temperature with respect to the solar
263 irradiance power using ideal and mid-latitude atmospheric transmittance with a heat transfer
264 coefficient of 6 W/m²-K. When the solar irradiance power is zero, CaCO₃ can cool to
265 approximately 14 °C below ambient temperature under ideal atmospheric transmittance and to
266 10 °C below ambient temperature under mid-latitude transmittance, and the paint can cool to
267 almost the same extent as CaCO₃ at 300 K. Because there is no sunlight, only emissivity in the
268 atmospheric window affects the radiative cooling performance. Therefore, compared with
269 CaCO₃, the paint, which has higher average emissivity of 3%p, can cool slightly more.

270 However, as the solar irradiance power increases, the temperatures of CaCO₃ and the paint
271 differ gradually. When solar irradiance power reaches 1000 W/m², CaCO₃ can cool by
272 approximately 9 °C under ideal transmittance and by 5.3 °C under mid-latitude transmittance;
273 further, the paint can cool by 2.6 °C under ideal transmittance, and its temperature increases
274 to 1 °C under mid-latitude transmittance. Under strong solar irradiance, not only the emissivity
275 in the transmittance window but also the ability to reflect sunlight is significant. Therefore, It
276 is demonstrated that CaCO₃ has much better cooling performance than paint as the solar
277 irradiance power increases. There are slight differences between the calculated and measured
278 data because the actual and calculated values of the atmospheric transmittance and heat transfer
279 coefficient differ. Figure 5(d) depicts the cooling temperature of the CaCO₃- and paint-coated
280 glasses in accordance with the heat transfer coefficient, h_c , under the solar irradiance power of
281 888 W/m². When $h_c = 0$, which is similar to the case of vacuum, CaCO₃ can cool by 28 °C
282 under ideal transmittance and by 16 °C under mid-latitude transmittance, and the paint can cool
283 by 10 °C under ideal transmittance and by 0.7 °C under mid-latitude transmittance. As the
284 value of h_c increases, parasitic heat exchange occurs in the ambient environment and reduces
285 the cooling performance. Even when $h_c = 12$ W/m²-K, at which heat exchange occurs actively
286 via conduction and convection, CaCO₃ can cool by 6 °C under ideal transmittance and by 4 °C
287 under reduced transmittance. In contrast, the paint cooled by 2.4 °C under ideal transmittance,
288 and even at the reduced transmittance in h_c value is 0, it cooled by less than 1 °C because it
289 exhibits substantial absorption of solar irradiance. Thus, both actual measurements and
290 calculations show that CaCO₃ demonstrates better cooling performance than the paint.

291 **5. Conclusions**

292 In this work, we demonstrated the radiative cooling performance of CaCO₃ microparticles both
293 by calculation and experiment. As the proposed device has a powder-based layer, the scattering
294 effect maximizes the reflectance for solar irradiance wavelengths; further, owing to the intrinsic
295 optical properties of CaCO₃, this device can emit energy in the form of infrared (8–13 μm)
296 rays. Notably, this device absorbs only 4% of the radiated solar power and exhibits an average
297 emissivity of approximately 90% in the atmospheric window. Through actual measurements
298 of temperature variations, it is demonstrated that the CaCO₃ radiative cooler can cool by 6.5
299 °C during the daytime and by 3.4 °C during nighttime. Further, according to calculations, it can
300 cool by more than 10 °C during daytime. Furthermore, this structure, CaCO₃ composite single
301 layer, does not contain any metal layer as reflector, which enhances productivity, price
302 competitiveness and practicality. These results demonstrate that the proposed CaCO₃ can be
303 employed as a radiative cooler itself. Owing to the property of CaCO₃, it can be radiative cooler
304 as tile of paint forms and can be applied to building, car, cool bag or other any fields that need
305 being cooled.

306 **Associated content**

307 **Supporting Information**

308 XRD results for CaCO₃ powders; Optical properties of CaCO₃-based radiative cooler in
309 accordance with thickness; Solar power density and atmospheric transmittance window

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312 **CRedit authorship contribution statement**

313 **Hangyu Lim**: Conceptualization, Methodology, Writing-Original draft, Visualization, Formal
314 analysis, Investigation. **Dongwoo Chae**: Conceptualization, Software, Validation. **Soomin**
315 **Son**: Resources, Methodology. **Sucheol Ju**: Data Curation. **Jisung Ha**: Investigation. **Heon**
316 **Lee**: Supervision

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Figures

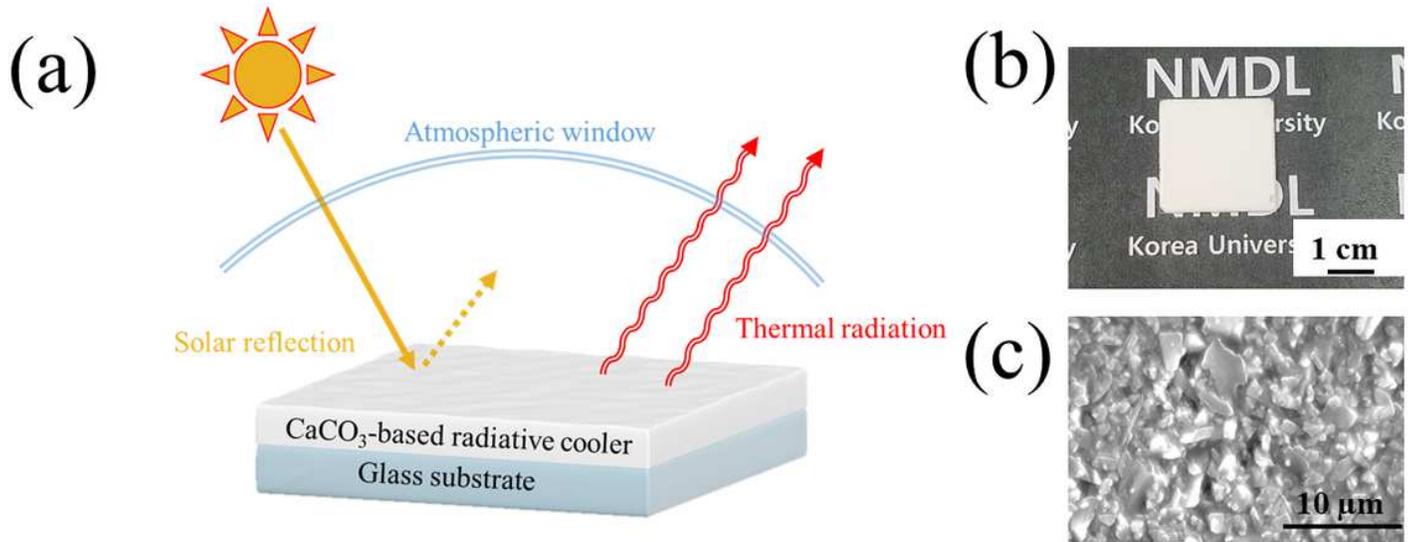


Figure 1

(a) Schematic of radiative cooling principle and radiative cooler structure. (b) Photo image and (c) SEM top view of radiative cooler, CaCO₃-coated glass.

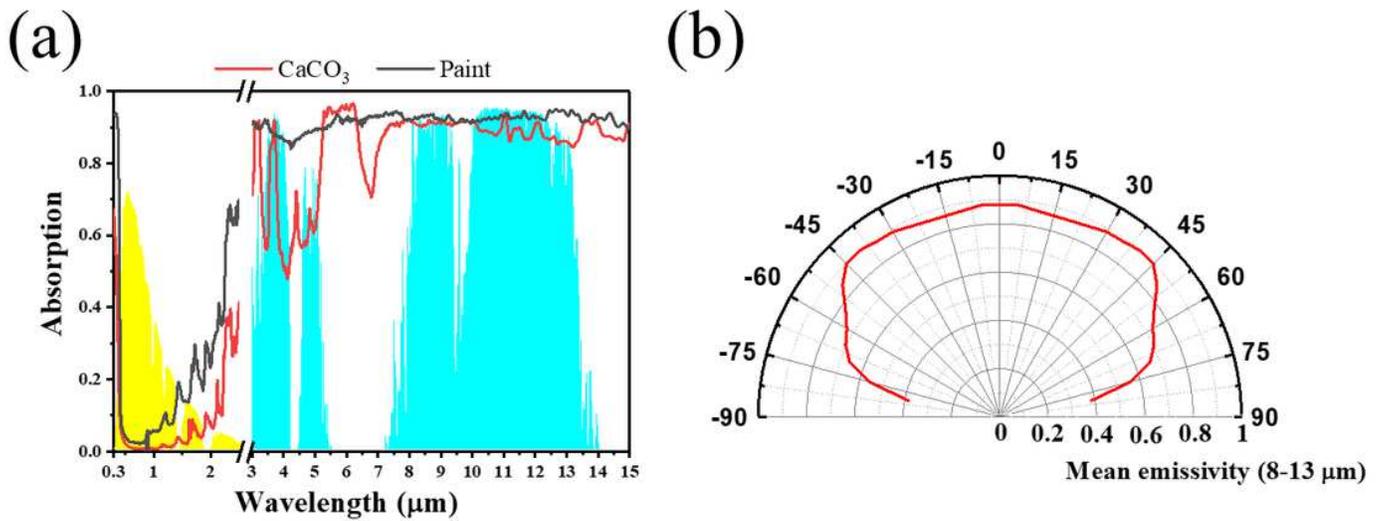


Figure 2

(a) Spectral absorptivity and emissivity of CaCO₃- and commercial white-paint-coated glass. (b) Average emissivity of CaCO₃-coated glass in the atmospheric window (8–13 μm) with respect to angle θ.

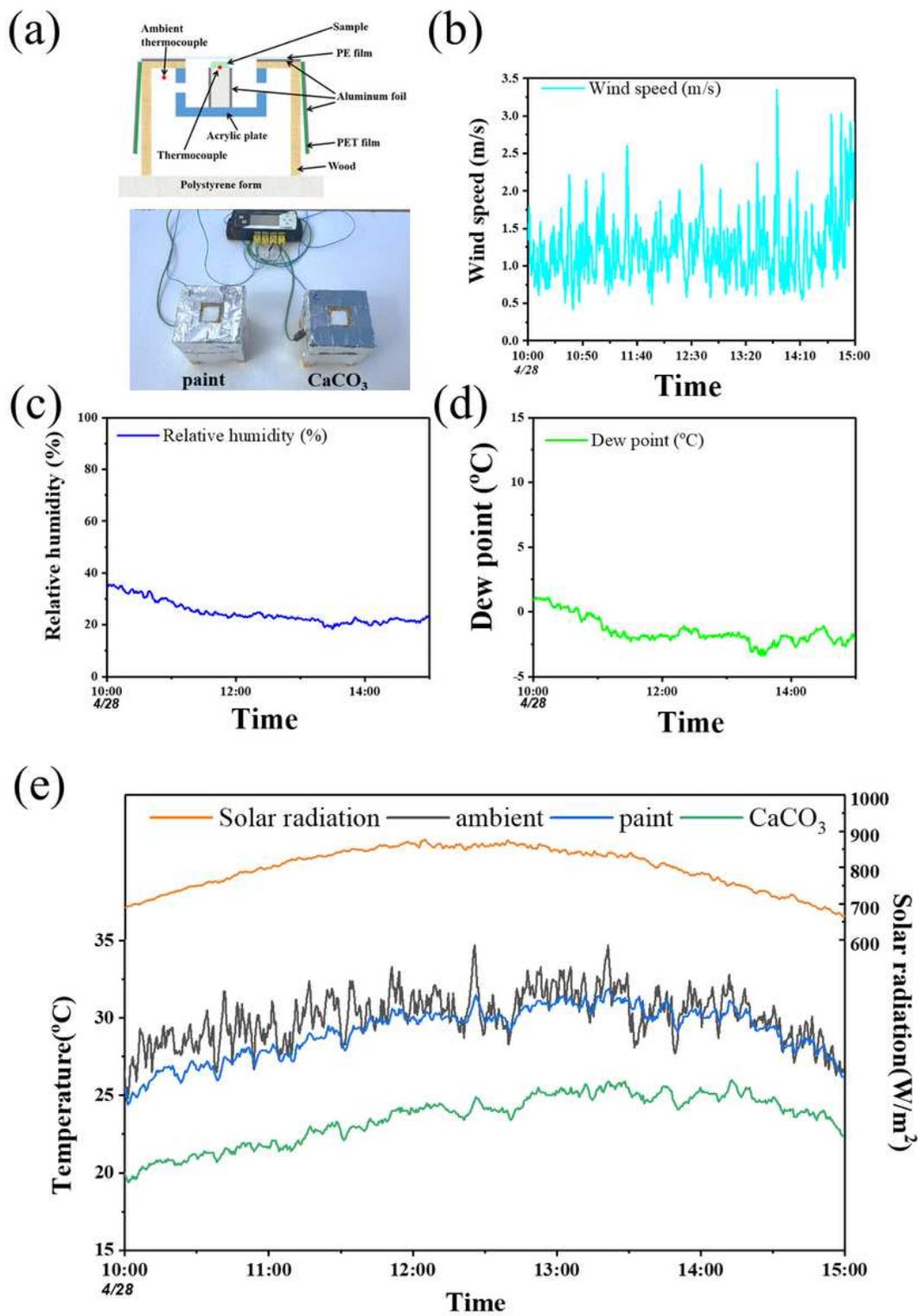


Figure 3

(a) Schematic diagram and photo image of outdoor measurement system for radiative cooling. Environmental conditions (b) wind speed, (c) Relative humidity, (d) dew point. (e) Radiative cooling performance of CaCO₃ and white paint in terms of temperature in daytime.

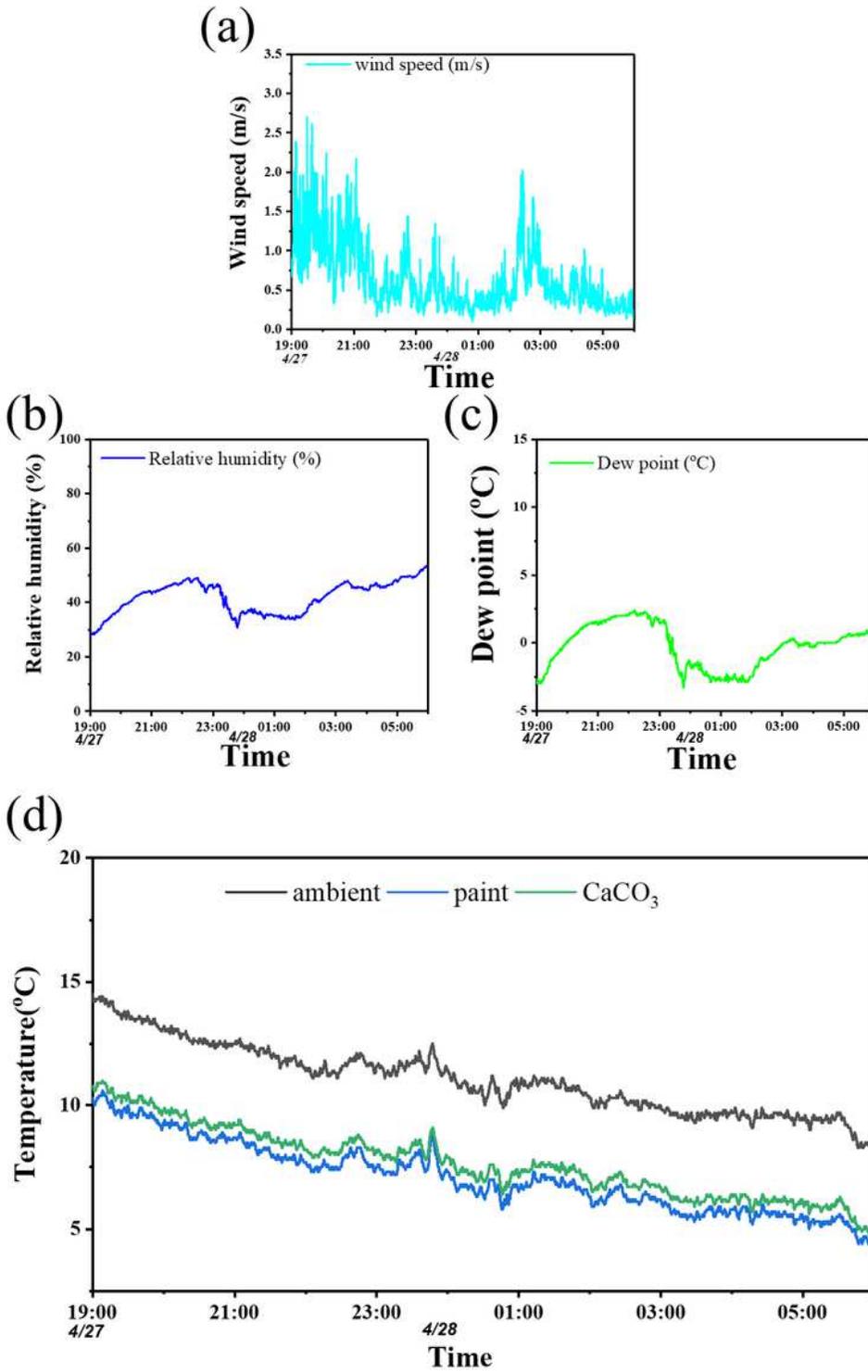


Figure 4

Environmental conditions (a) wind speed, (b) Relative humidity, (c) dew point. (d) Radiative cooling performance of CaCO₃ and white paint in terms of temperature in nighttime.

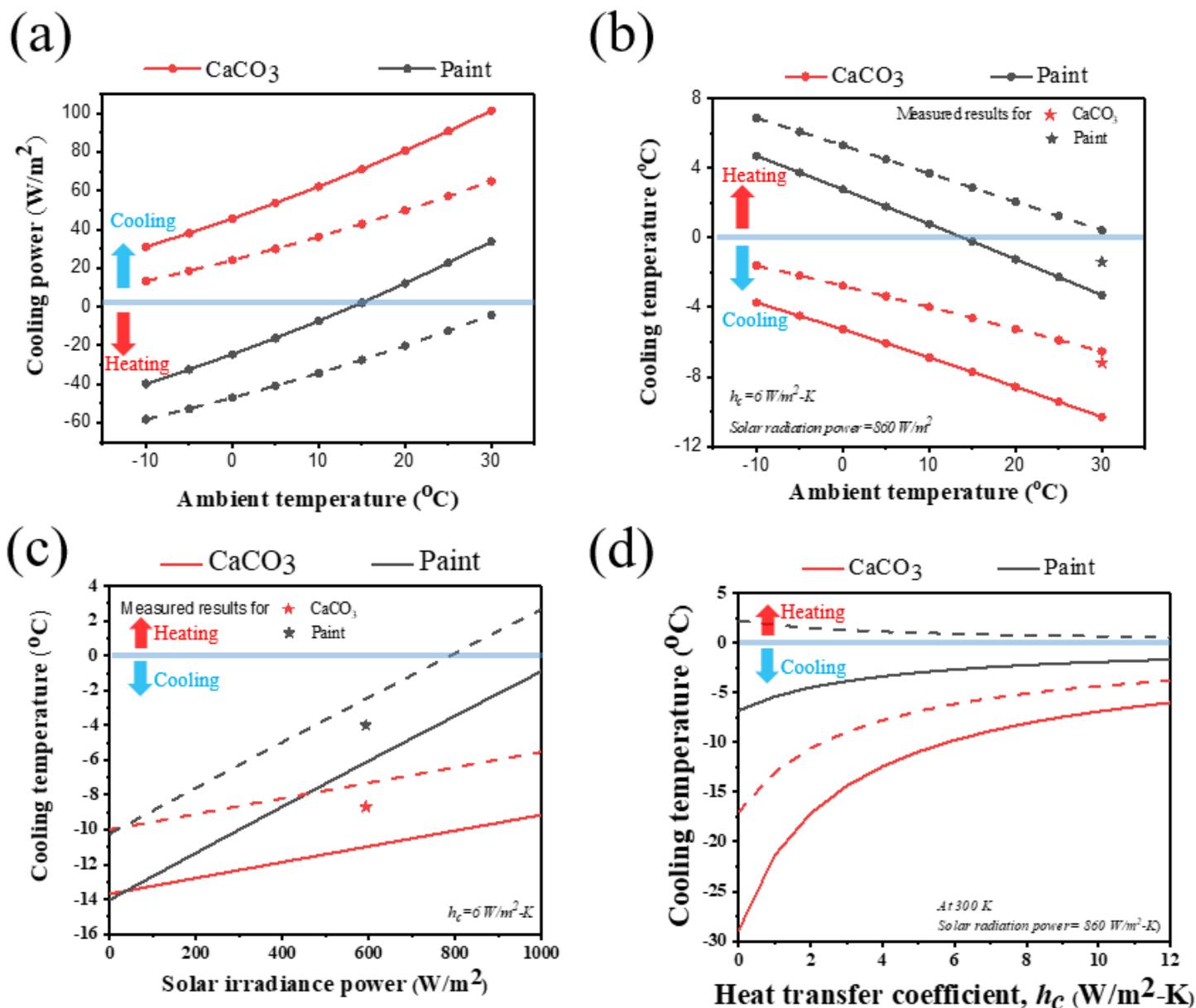


Figure 5

(a) Calculated cooling power of CaCO₃ and (b) cooling temperature in accordance with ambient temperature under the condition of $h_c = 6 \text{ W/m}^2\text{-K}$ using ideal (solid line) and mid-latitude winter (dotted line) atmospheric transmittance and measured cooling temperatures are represented with stars (red for CaCO₃/glass, gray for paint/glass). (c) Calculated temperatures of CaCO₃- and paint-coated glass in accordance with the solar irradiance power under the condition of $h_c = 6 \text{ W/m}^2\text{-K}$ and measured cooling temperatures are marked with stars (red for CaCO₃, gray for paint). (d) Calculated temperatures of CaCO₃- and paint-coated glass in accordance with convection coefficient h_c at 300 K using AM1.5 solar irradiance spectrum using ideal (solid line) and mid-latitude (dotted line) atmospheric transmittance.

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