

Performance Studies on Low GWP Refrigerants as Eco-friendly Alternatives for R134a in Household Refrigerator

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44 overall emissions of the equivalent CO₂, while indirect releases represent about ten times the total direct releases,
45 the total contribution to global warming by the refrigeration sector is approximately 20% (Mota-Babiloni et al.,
46 2020).

47 Due to its excellent thermophysical properties, R134a is the most commonly used refrigerant in
48 household refrigerators but its high GWP will have to be phased out earlier. The European Union recently
49 announced that alternative refrigerants would have a GWP of less than 150 in order to reduce greenhouse gas
50 emissions (Saji et al., 2019). In order to meet global environmental targets, conventional refrigerants must be
51 replaced by eco-friendly and energy-efficient refrigerants. The experimental tests carried out on household
52 refrigerator to replace R134a with R1234yf, the refrigeration capacity improved slightly, making R134a an
53 appropriate substitute for household refrigerator (Aprea et al., 2016). The R1234yf, R1234ze (E), and R600a are
54 low-GWP refrigerants that can be used to replace R134a. Among the refrigerants, R1234yf may be an alternate
55 solution for R134a due to similar efficiency and the same mass flow rate in household refrigerators (Righetti et
56 al., 2015). The experimental investigation carried out on R1234yf and R1234ze in an R134a system with various
57 evaporation and condensation temperatures. It indicates that R1234ze has a lower average refrigeration capacity
58 and COP than R134a by around 30% and between 2% and 8%, respectively. Similarly, R1234yf has a less cooling
59 capacity about of 9% and between 5% and 30% less COP than R134a (Navarro et al., 2013). Furthermore, the
60 flammability of these refrigerants is a big concern. Owing to its flammability, one of Europe's major car
61 manufacturers declined to use R1234yf.

62 There are various applications such as household refrigerator, beverage coolers and mobile air
63 conditioners, an azeotropic mixture R1234yf /R134a is offered to replace R134a. Adding 10-11% of R134a to
64 R1234yf, the mixture is non-flammable and it's COP, discharge temperature and refrigeration capacity are like
65 that of R134a (Yohan et al., 2013). The HFO/HFC mixtures are suitable alternatives of HFC refrigerants and it
66 has low GWP (GWP<150). It has been observed that adding 10% R134a to R1234yf will make the refrigerant
67 mixture to be non-flammable and it is suitable alternative for R134a due its energy efficiency (Aprea et al., 2016).
68 The HFO/HFC mixtures have improved performance, and their expected COP and exergy efficiency are 4% to
69 8.3% and 5.1% to 10.5% respectively higher than the HFO (Saji et al., 2019).

70 Therefore, the energy analysis on a household refrigerator operating with low GWP refrigerant mixtures
71 as environmental friendly alternatives for R134a has been studied using mathematical simulation. The household
72 refrigeration system has been modeled and the performance has been studied to find the best composition of the
73 mixture to operate the system. This simulation was carried out using MATLAB software, and the REFPROP
74 database was used to obtain thermophysical properties of the refrigerants. Due to the environmentally friendly
75 properties and flammability aspects, R1234ze/R134a (90/10) could be a good substitute for R134a in the
76 refrigerator to satisfy the Montreal and Kyoto Protocol expectations.

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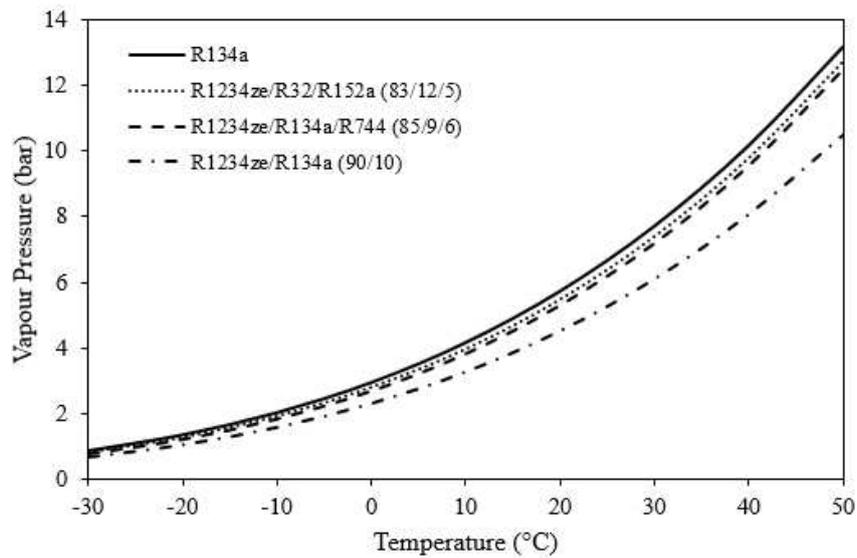
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82 **Refrigerant Selection**

83 One of the essential methods of identifying the suitable refrigerant is the selection of refrigerants based on
84 its thermo-physical properties. For operating temperatures -30°C to 50°C, the properties of refrigerants obtained
85 from REFPROP 9.0 have been plotted. The vapor pressures of HFO refrigerant mixtures are lower than that of
86 R134a as shown in Fig. 1. The vapor pressure of R1234ze/R134a is 21.5% lower than that of R134a among the
87 different mixtures. This can cause a reduced energy consumption. Fig. 2. shows that the latent heat of
88 R1234ze/R134a/R744 and R1234ze/R134a are 6.4% and 6.8% lower than R134a respectively. In the case of
89 R1234ze/R32/R152a mixture, the latent heat is 1.3% greater than the R134a. The thermo-physical properties used
90 in this analysis are shown in Table 1.

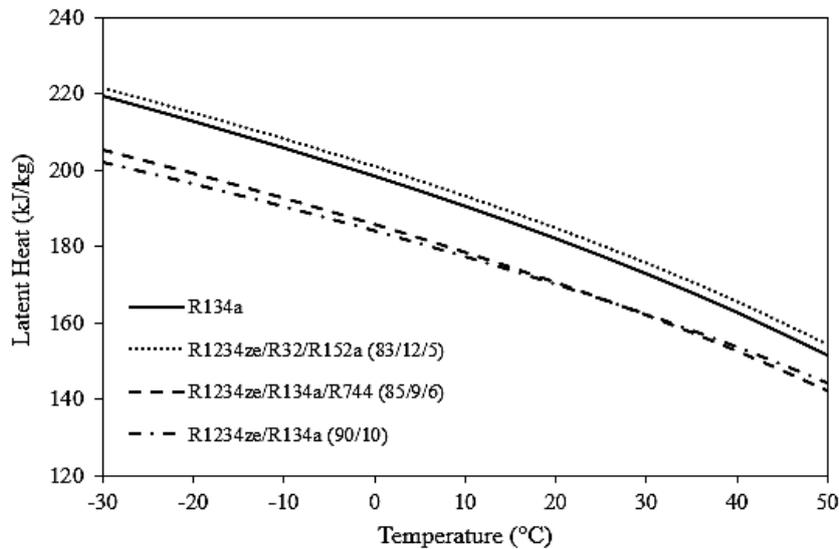
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Fig. 1 Variation of vapour pressure as a function of temperature



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Fig. 2 Variation of latent heat as a function of temperature

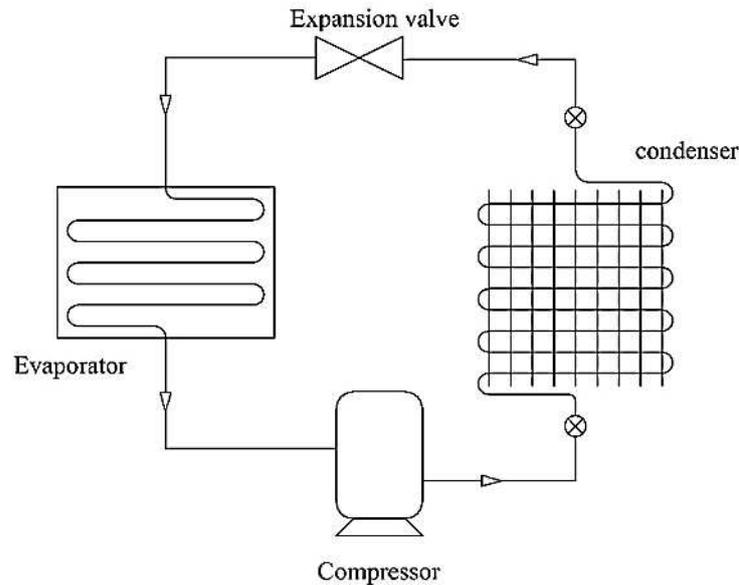
97 **Table 1.** Thermophysical properties of the refrigerants

Refrigerants	Composition (Mass fraction %)	GWP	Safety group	Critical Pressure (bar)	Critical Temperature (°C)	Boiling Point at 1 atm (°C)
R134a	0	1430	A1	40.6	101.06	-26.07
R1234ze/R32/R152a	83/12/5	92.88	A2L	41.38	101	-25.09
R1234ze/R134a/R744	85/9/6	133.86	A2L	47.13	103	-23.98
R1234ze/R134a	90/10	148.4	A2L	36	381.6	-20.6

98

99 **Analytical study**

100 Figure 3 shows a schematic diagram of the refrigeration system used throughout this analysis. It consists of
 101 a reciprocating compressor, air-cooled condenser, capillary tube, and evaporator. The simulation of the above
 102 components was performed using MATLAB software. To obtain the thermo-physical properties of refrigerants,
 103 REFPROP is directly interlinked with MATLAB. A 190 l household refrigerator has been considered in this
 104 modeling. The property variations are one dimensional, steady state condition, constant mass flow rate, polytropic
 105 process in compression, efficiency of electrical motor is 85% and pressure drop in the heat exchanger is negligible.
 106 The above assumptions are taken into account in this analysis to minimize the complexities of the simulation.



107

108 **Fig. 3** Schematic diagram of refrigeration system

109 **Compressor model**

110 A 100W reciprocating type compressor has been taken for the study and it is divided in to three control
 111 volumes such as compressor shell, swept volume and discharge tube (William and Doyle, 1988). The following
 112 equation is used to determine the swept volume:

113
$$\dot{m}_o h_o = Q_{comp} + \frac{dW}{dt} + \dot{m}_i h_i - (\Delta P)_V \quad (1)$$

114 The displacement volume is calculated at each time step using the following equation:

115
$$v(t) = V_{cylinder} + \frac{\pi D_{cyl}^2}{8} L_{swept} (1 - \cos(\omega t)) \quad (2)$$

116 The following equation is used to calculate compressor work:

$$117 \quad W_{comp} = \frac{n}{n-1} P_s V \left\{ \left(\frac{P_{discharge}}{P_{suction}} \right)^{\frac{n-1}{n}} - 1 \right\} \quad (3)$$

118 Condenser model

119 The condenser is divided into de-superheated region, two-phase region, and sub-cooled region. The
120 following correlations are used to measure the refrigerant side heat transfer coefficient in the single-phase region
121 (Minor et al., 2010):

122 In laminar region ($Re < 2100$)

$$123 \quad Nu = 1.86 Re^{0.33} Pr^{0.33} \left(\frac{D_i}{L} \right) \left(\frac{\mu_b}{\mu_w} \right)^{0.5} \quad (4)$$

124 In turbulent region ($Re > 10,000$)

$$125 \quad Nu = 0.023 Re^{0.8} Pr^{0.33} \left(\frac{D_i}{L} \right) \left(\frac{\mu_b}{\mu_w} \right)^{0.14} \quad (5)$$

126 The air-side heat transfer coefficient of the condenser is calculated using the correlations below (Marques
127 et al. 2014):

$$128 \quad Nu = 0.3 + \left(\frac{0.62 Ra^{0.167}}{\left[1 + \left(\frac{0.559}{Pr} \right)^{\frac{9}{16}} \right]^{\frac{1}{4}}} \right) \quad (6)$$

129

130 Evaporator model

131 The evaporator is divided into two-phase region and superheated region. The convective heat transfer
132 coefficient of single-phase region and two-phase region are computed by the equation no. (7), (8) & (9) (Wattelet
133 et al., 1994; Chang et al., 2000; Cooper, 1984; Navarro et al., 2013 & Nielsen et al., 2007):

134 In laminar region ($Re < 2100$)

$$135 \quad Nu = 1.86 Re^{\frac{1}{3}} Pr^{\frac{1}{3}} \left(\frac{D_i}{L} \right) \left(\frac{\mu_b}{\mu_w} \right)^{0.13} \quad (7)$$

136 In turbulent region ($Re > 10,000$)

$$137 \quad Nu = 0.023 Re^{0.8} Pr^{\frac{1}{3}} \left(\frac{D_i}{L} \right) \left(\frac{\mu_b}{\mu_w} \right)^{0.14} \quad (8)$$

$$138 \quad h_{evap} = 55 Mo^{-0.5} q^{0.67} Pr^{0.12} (\log Pr)^{0.55} + F h_l R \quad (9)$$

139 Capillary tube model

140 The flow of a capillary tube can be separated into single-phase and two-phase. Pressure drops in the capillary
141 tube are caused due to friction, momentum loss, and gravitational effect. Sudden contraction happens by the pressure
142 drop, is calculated by following equation (Perry and Chilton, 1984).

$$143 \quad P_i - P_{init} = \frac{G^2 V_f}{2} \left\{ \left(\frac{1}{C_{con}} - 1 \right) + \left(1 + \frac{A^2}{A_1^2} \right) \left(1 + \frac{V_f g}{V_f} \right) X \right\} \quad (10)$$

144 Coefficient of contraction is calculated by following equation:

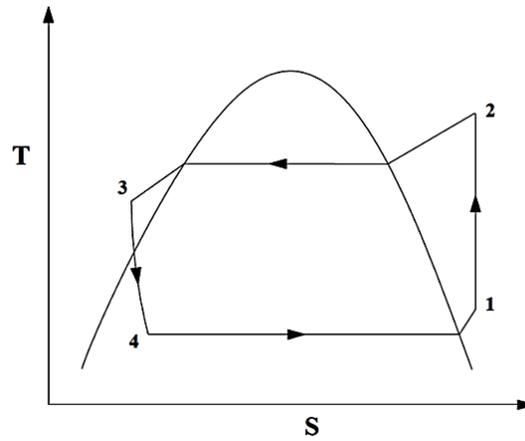
$$146 \quad C_c = 0.55 \left(\frac{A_{ma}}{A_1} \right)^3 - 0.242 \left(\frac{A_{ma}}{A_1} \right)^2 + 0.111 \left(\frac{A_{ma}}{A_1} \right) + 0.585 \quad (11)$$

$$147 \quad COP = \frac{Q_{evap}}{W_{comp}} \quad (12)$$

148 **Exergy Analysis**

149 The aim of the exergy analysis is to assess the irreversibility of each component of a household refrigerator.
 150 Fig. 4 depicts the T-S diagram and the exergy balance equations are taken from the previous studies (Bayrakci
 151 HC and Ozgur, 2009) and shown in Table 2.

152



153

154

Fig. 4 T-S diagram

155 **Table 2** Irreversibility equation of each Component

Components	Irreversibility
Compressor	$I_{comp} = \dot{m}\psi_1 + W - \dot{m}\psi_2$
Condenser	$I_{cond} = \dot{m}\psi_2 - \dot{m}\psi_3 - Q_{cond} \left(1 - \frac{T_o}{T_{cond}}\right)$
Capillary tube	$I_{capi} = \dot{m}\psi_3 - \dot{m}\psi_4$
Evaporator	$I_{evap} = \dot{m}\psi_4 + Q_{evap} \left(1 - \frac{T_o}{T_{evap}}\right) - \dot{m}\psi_1$

156

157 The total exergy destruction of the system is determined by:

158

$$I_{total} = I_{comp} + I_{cond} + I_{capi} + I_{evap} \quad (13)$$

159 **Table 3.** Operating conditions

Parameters	Household Refrigerator
Evaporator temperature (°C)	-18
Condenser Temperature (°C)	45
Ambient temperature range (°C)	22, 26, 30, 36 & 43

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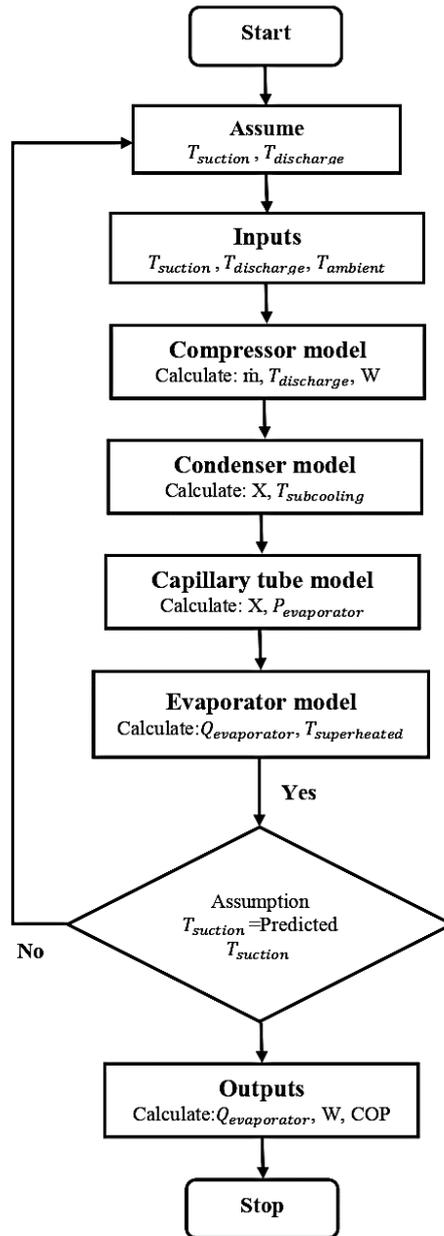


Fig. 5 Flow chart for simulation model

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164 **Validation**

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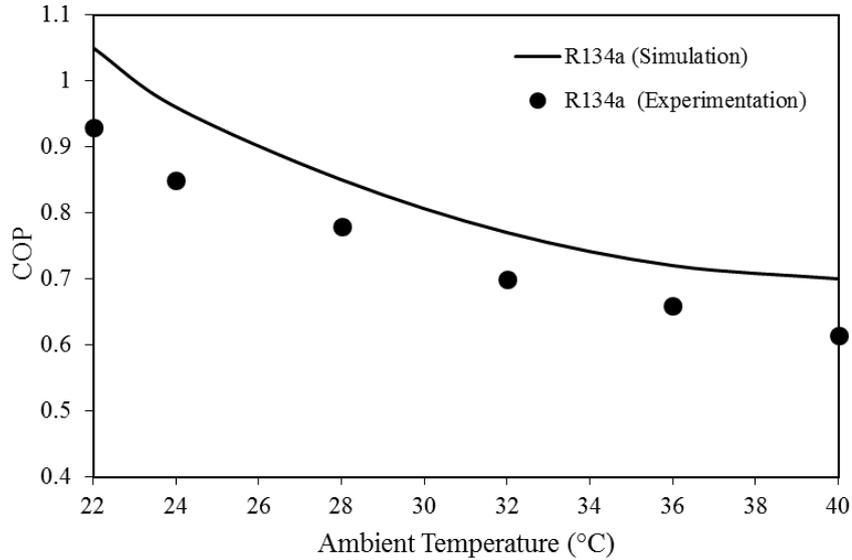
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The experimental values are compared to the simulation results in order to validate the simulation model, as shown in Fig. 6. The experimental COP of R134a has been taken from the previous studies (Saji et al., 2017). The figure indicates that the simulation COP deviates from the experimental COP by 9% to 12% for various ambient conditions at the evaporator temperature of -12°C. Since this difference is nominal, the validity of the mathematical approach has been verified.

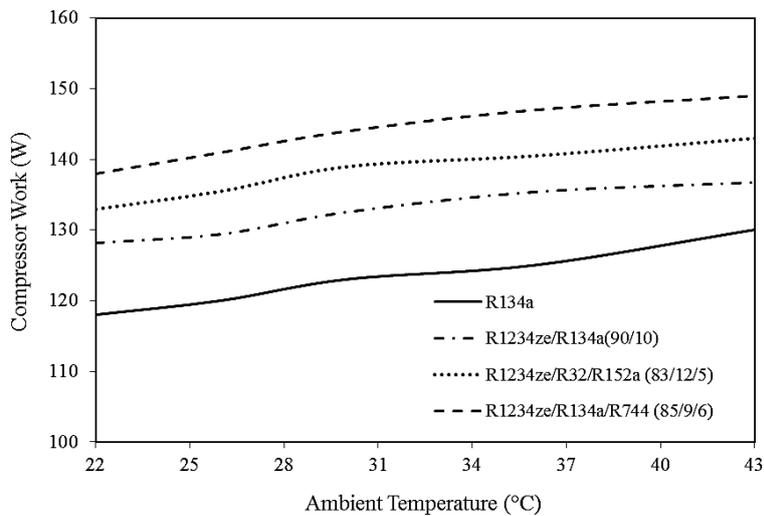


172
173 **Fig. 6** Variation of theoretical COP with ambient temperature

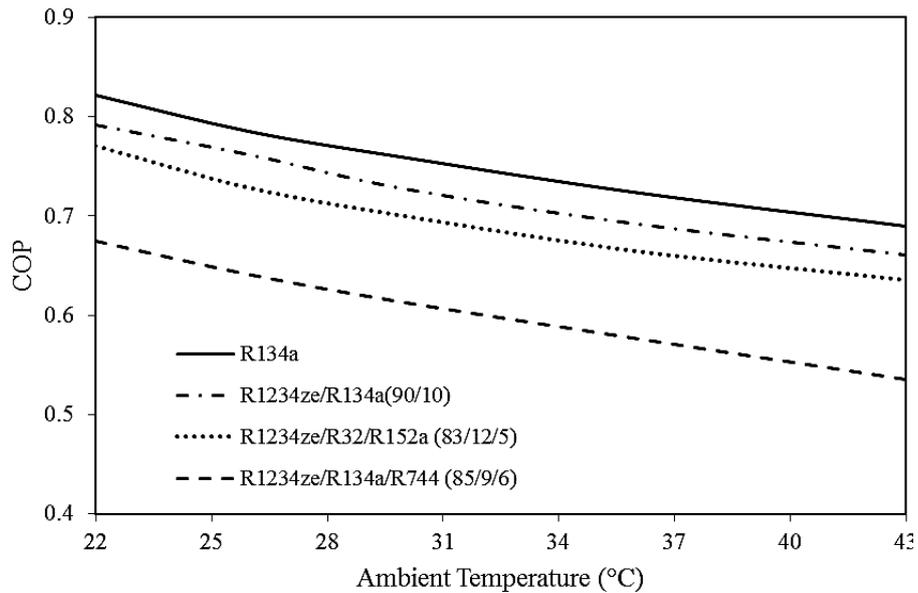
174 **Results and Discussion**

175 The energy analysis on a household refrigerator through environmentally friendly refrigerant mixtures
 176 was theoretically studied for different ambient temperatures at an evaporator temperature of -18 °C. The findings
 177 would examine the output parameters including the compressor work, COP, total exergy destruction, exergy
 178 efficiency and TEWI.

179 As seen in Fig. 7, the function of the compressor work increases as the atmospheric temperature
 180 increases. This is because the input pressure of compressor and mass flow rate increases. The R1234ze/R134a
 181 mixture has been observed to reduce compressor work from 4.3% to 8.5 % than other two mixtures. Because of
 182 low evaporator pressure and high molecular weight of the mixture. The difference of COP with ambient
 183 temperature was calculated and plotted using with HFO mixtures in Fig. 8. It has been noted that the COP of
 184 R1234ze/R134a mixture is higher than the other two mixtures by about of 16.4%. This may be because, the
 185 refrigerant mixture has a high latent heat.



186
187 **Fig. 7** Variation of compressor work with ambient temperature



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Fig. 8 Variation of COP with ambient temperature

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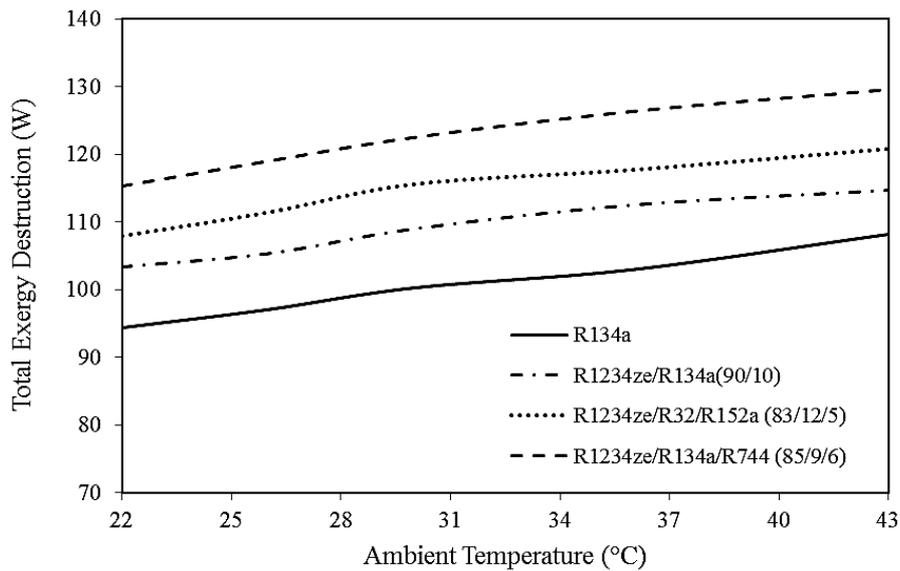
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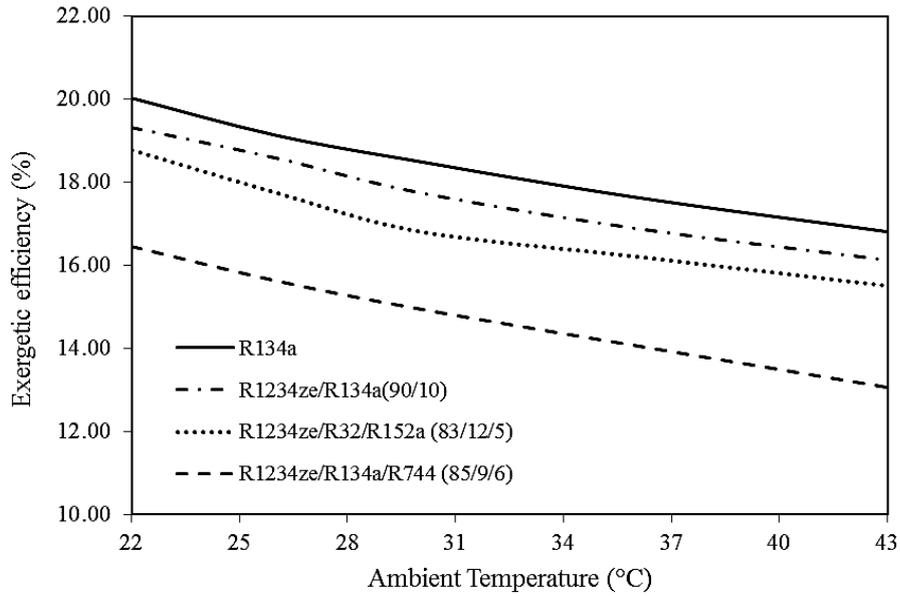
Fig. 9 indicates that as the function of the total exergy destruction increases as the atmospheric temperature increases. The previous studies have found a similar pattern (Siva et al., 2012). The overall irreversibility of the R1234ze/R134a mixture is decreases from 5.2% to 12.4% than the other two mixtures that can be attributed to the high compression ratio. Since the exergy efficiency is used to determine the quality of energy used by a system, it has been calculated for HFO refrigerant mixtures and shown in Fig. 10. The exergy efficiency of the R1234ze/R134a mixture improves from 4% to 16% as compared to the other two mixtures, and it also has a lower exergy efficiency at high ambient temperatures. Among all the HFO mixtures, R1234ze/R134a mixture shows better performance and it is very similar to the R134a.



198

199

Fig. 9 Variation of total exergy destruction with ambient temperature



200
201 **Fig. 10** Variation of exergetic efficiency with ambient temperature

202 **Total Equivalent Warming Impact**

203 The TEWI is an environmental index. It represents the quantity of direct and indirect effects that are
204 calculated using the following equation (Saji et al. 2020):

$$\begin{aligned}
 TEWI &= \text{Direct effect} + \text{Indirect effect} \\
 &= (m \times l \times S_l \times GWP_{100}) + (E \times S_l \times r)
 \end{aligned}
 \tag{14}$$

205
206
207 Table 4. shows that the R1234ze/R134a has a lower TEWI than R134a by about of 5.9%. This is because
208 of less energy efficiency, but direct emissions are significantly lower than R134a.

209 **Table 4.** TEWI for the both refrigerants

Refrigerant	R134a	R1234ze/R134a
Charge quantity (kg), m	0.135	0.129
Leakage rate per year (%), l	6.6	6.6
Service life (year), S_l	15	15
GWP	1430	148
Direct effect	191	19
Power consumption per year (kWh), E	701	756
CO ₂ emission (kgCO ₂ / kWh), r	0.89	0.89
Indirect effect	9358	10,093
TEWI	9549	10,112

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215 **Conclusion**

216 The low-GWP refrigerants such as R1234ze/R134a, R1234ze/R32/R152a, and
217 R1234ze/R134a/R744 have been studied theoretically in household refrigerator. Among the mixtures,
218 R1234ze/R134a (90/10) performs best, with estimated COP and exergetic efficiency 3.7% to 16.4% and 4% to
219 16% higher than the other mixtures. The R1234ze/R134a has a lower TEWI than R134a by about of 5.9%. This
220 is because of less energy efficiency, but direct emissions are significantly lower than R134a. Even though the
221 performance of the mixture R1234ze/R134a is slightly lower than R134a, but it is superior to other HFO mixtures.
222 It may also be an appropriate alternative for R134a in the household refrigerator to address environmental
223 problems according to the Kyoto Protocol.

224

Nomenclature	225
A : Area (m ²)	226
C : Coefficient	
COP : Coefficient of Performance	
D : Coil mean diameter (m)	
F : Empirical constant	
G : Mass flux (kg m ⁻² s ⁻¹)	
GWP : Global Warming Potential	
h : Specific enthalpy (J kg ⁻¹)	
L : Length (m)	
\dot{m} : Mass flow rate (kg s ⁻¹)	
Mo : Molecular weight (g kmol ⁻¹)	
n : Polytropic index	
Nu : Nusselt number	
P : Pressure (kPa)	
Pr : Prandtl number	
Q : Heat (J)	
Ra : Rayleigh number	
Re : Reynolds number	
t : time (s)	
T : Temperature	
TEWI : Total Equivalent Warming Impact	
V : Volume (m ³)	
W : Work (W)	
X : Refrigerant quality	
ω : Angular velocity (rad s ⁻¹)	
μ : Dynamic viscosity (Pa s)	
ψ : Exergy (J/kg)	

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231 **Authors' contributions**

232 Saji Raveendran Padmavathy handled conceptualization, methodology, data collection Murugan Paradesi
233 Chockalingam was responsible for writing, reviewing and final editing of the manuscript. Godwin Glivin was in
234 responsibility of data analysis and preliminary paper writing, while Nithyanandhan Kamaraj was in control of
235 result interpretation. Venkatesh Thangaraj aided in the analysis and editing of the manuscript. The methodology
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240 **Data availability**

241 The dataset generated during this work are not publicly available, however they are available upon reasonable
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243 **Compliance with ethical standards**

244 **Ethics approval and consent to participate:** Not Applicable.

245 **Consent for publication:** Not Applicable.

246 **Competing interests:** The authors declare that they have no competing interests.

247

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Figures

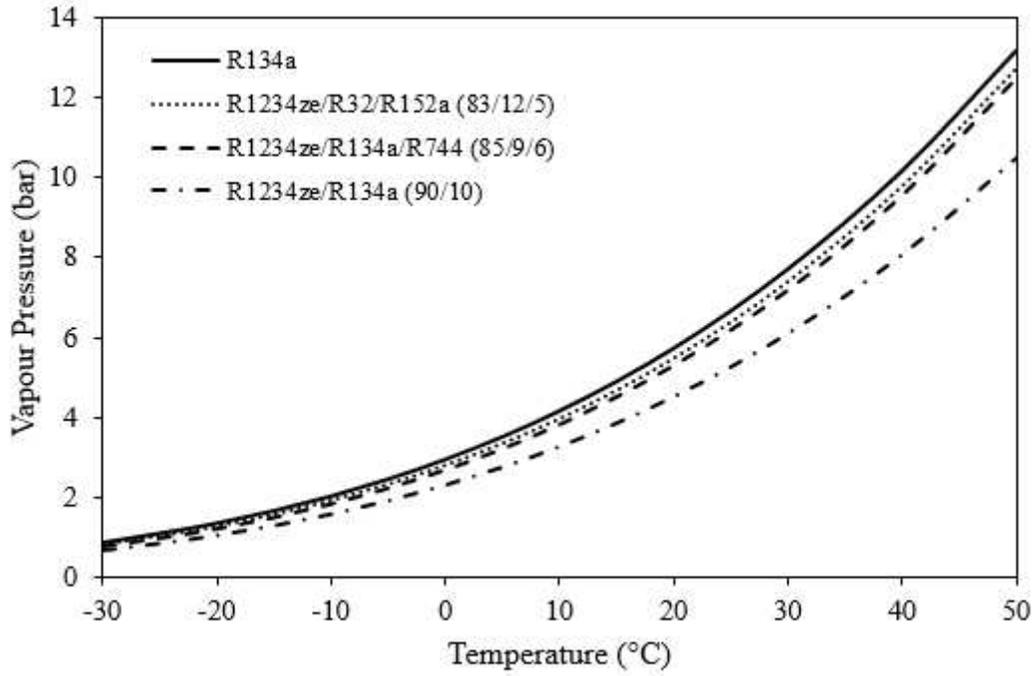


Figure 1

Variation of vapour pressure as a function of temperature

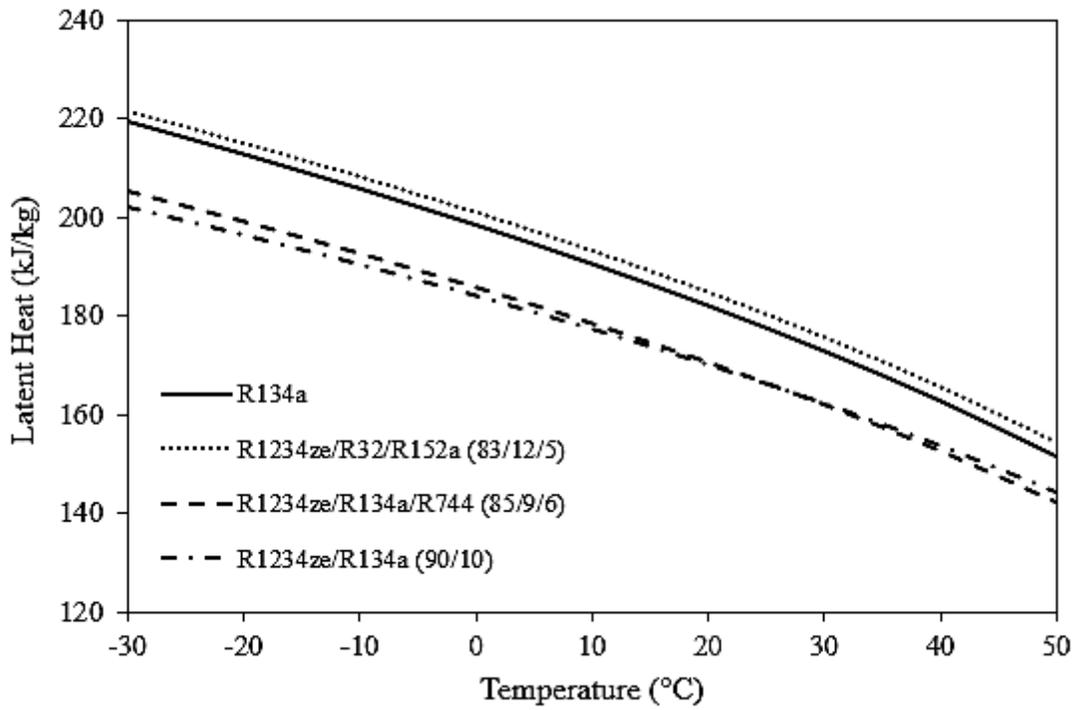


Figure 2

Variation of latent heat as a function of temperature

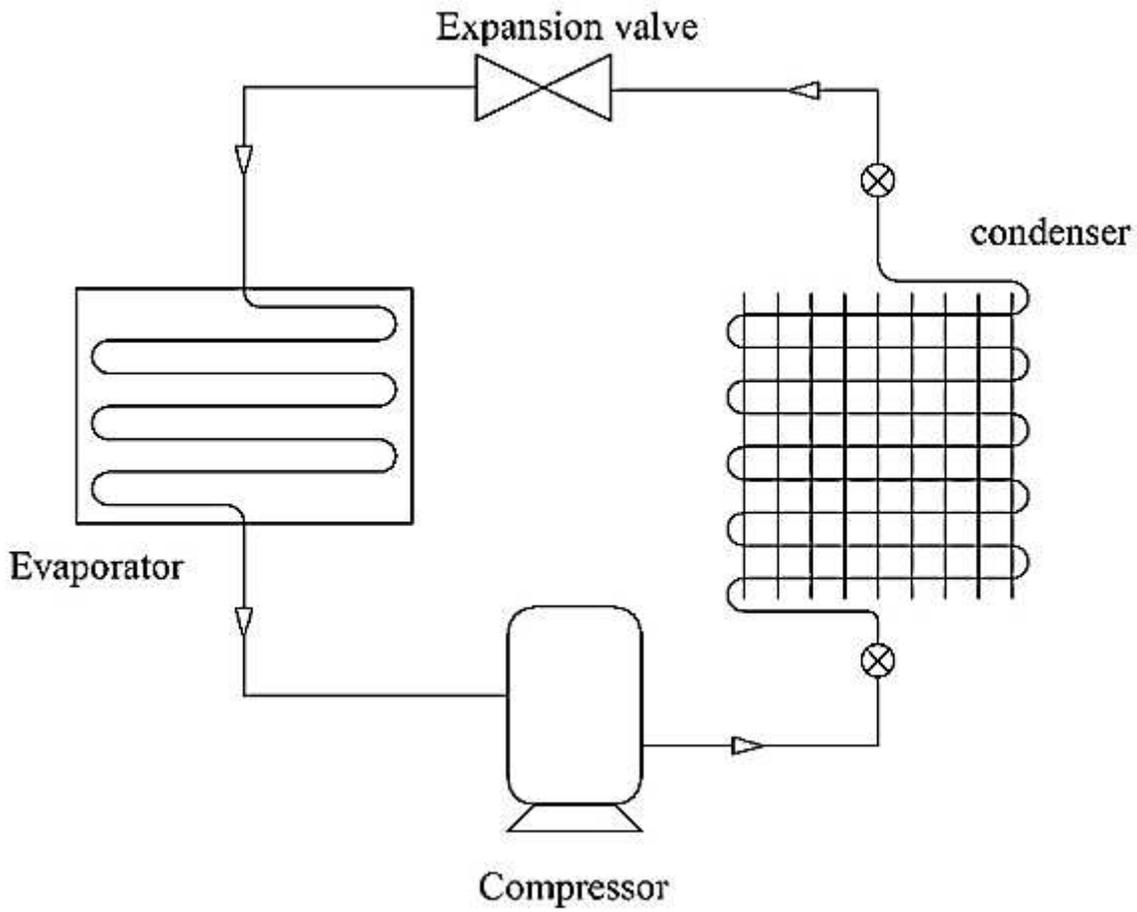


Figure 3

Schematic diagram of refrigeration system

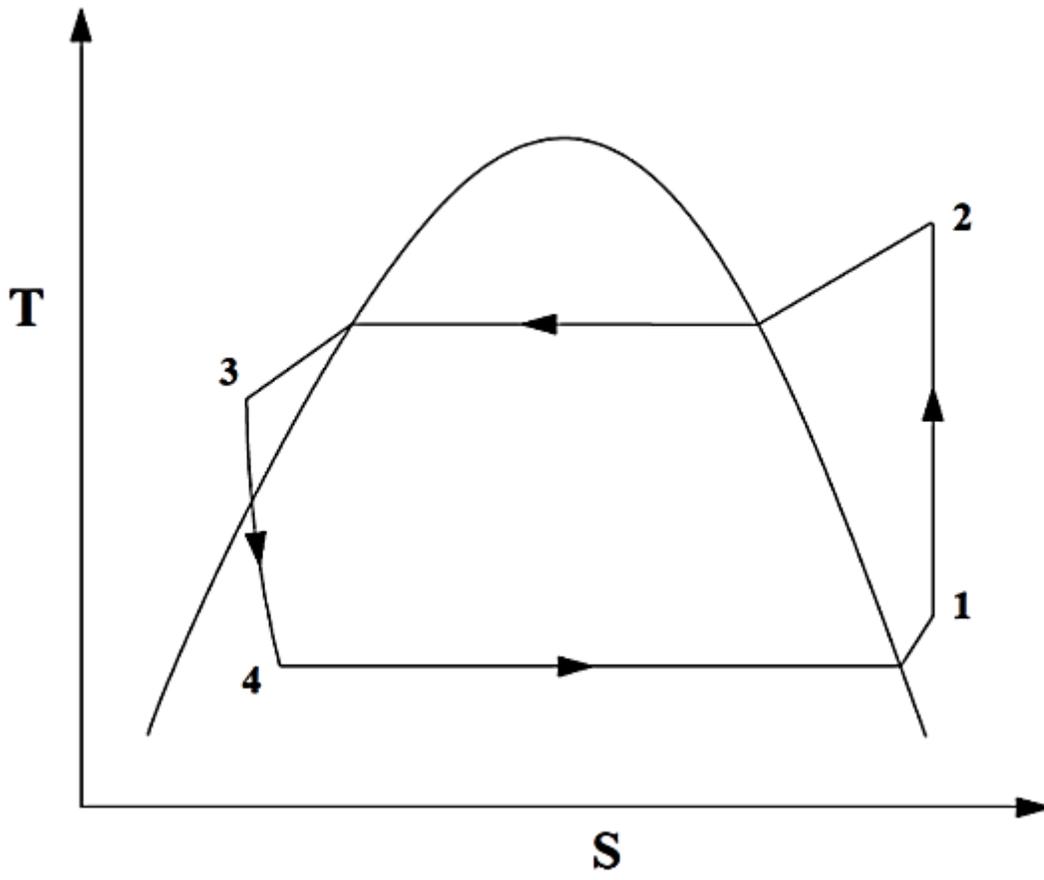


Figure 4

T-S diagram

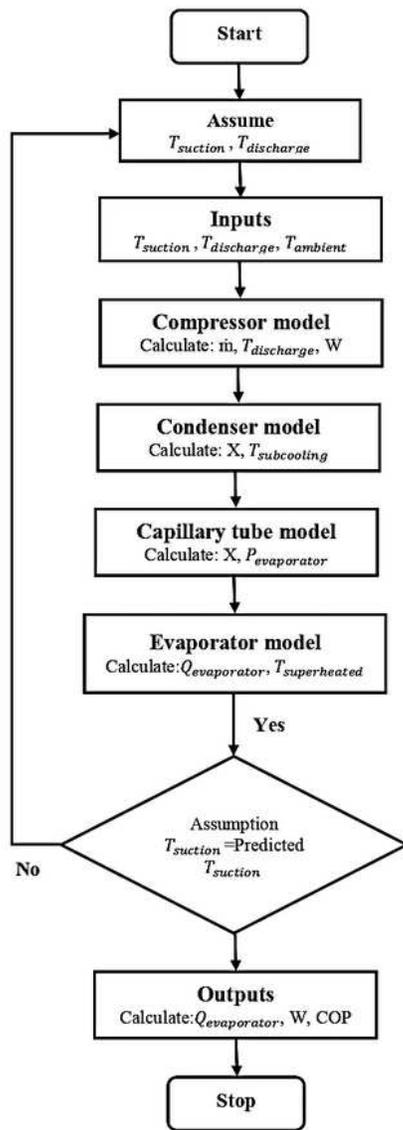


Figure 5

Flow chart for simulation model

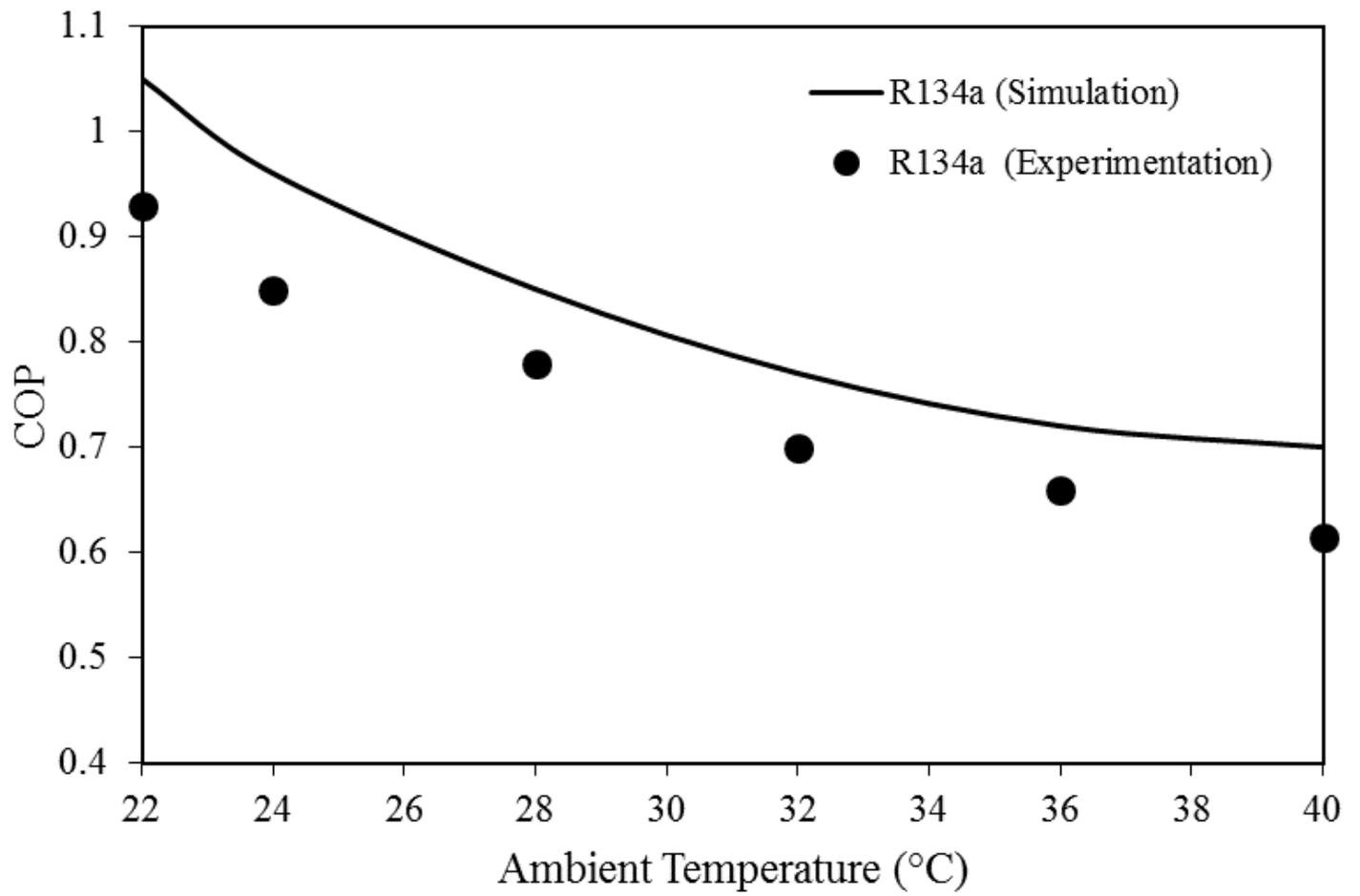


Figure 6

Variation of theoretical COP with ambient temperature

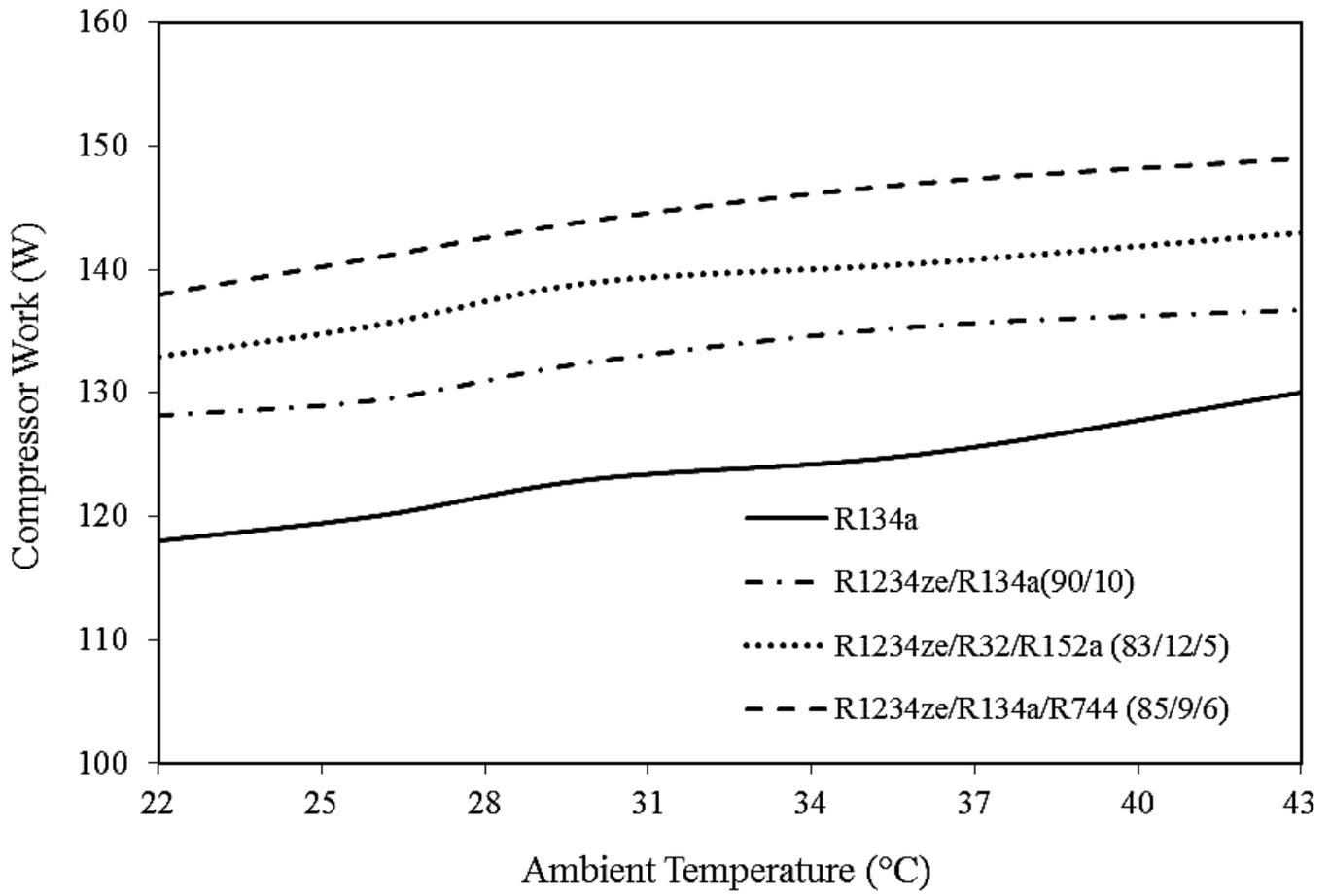


Figure 7

Variation of compressor work with ambient temperature

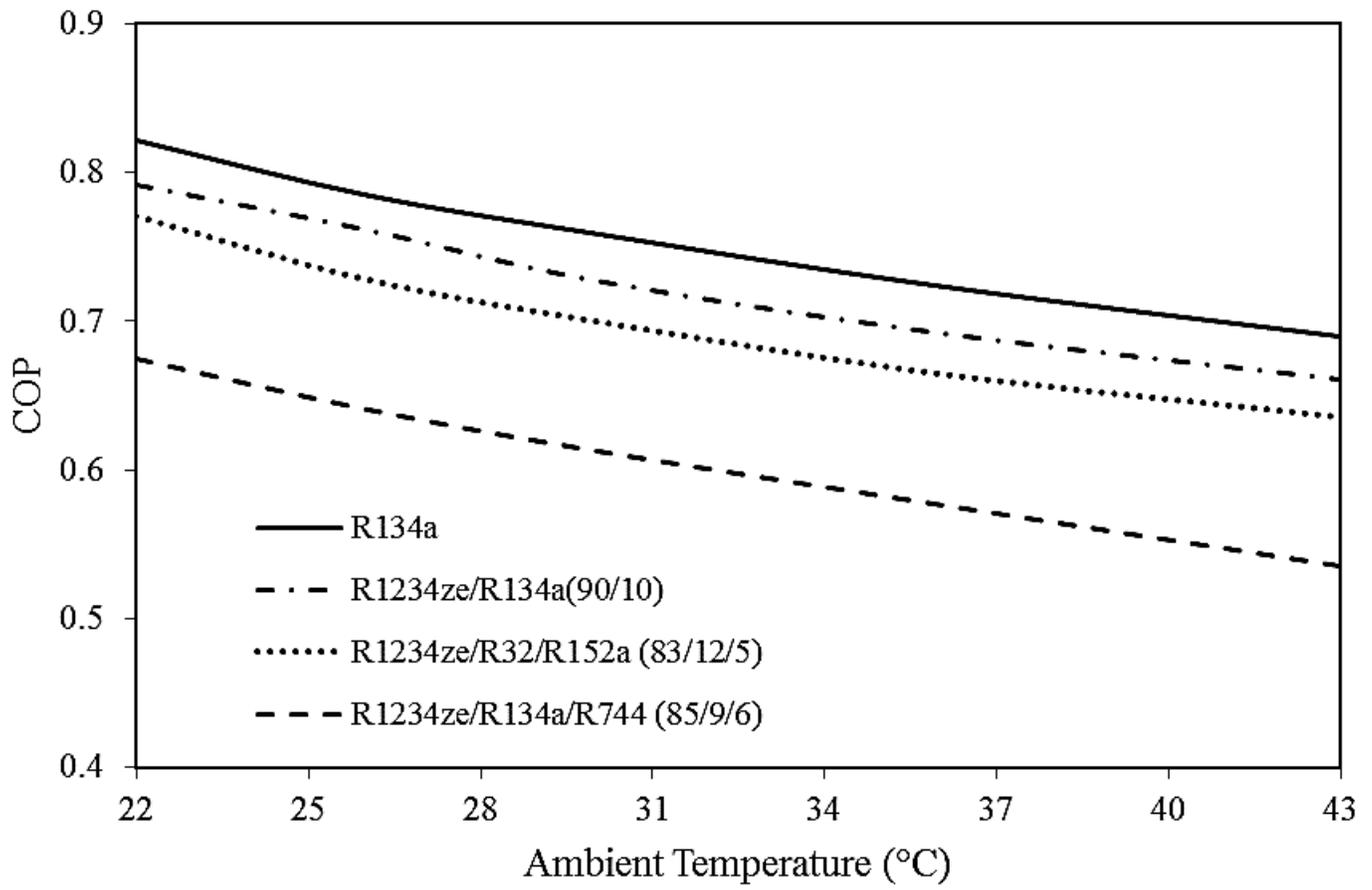


Figure 8

Variation of COP with ambient temperature

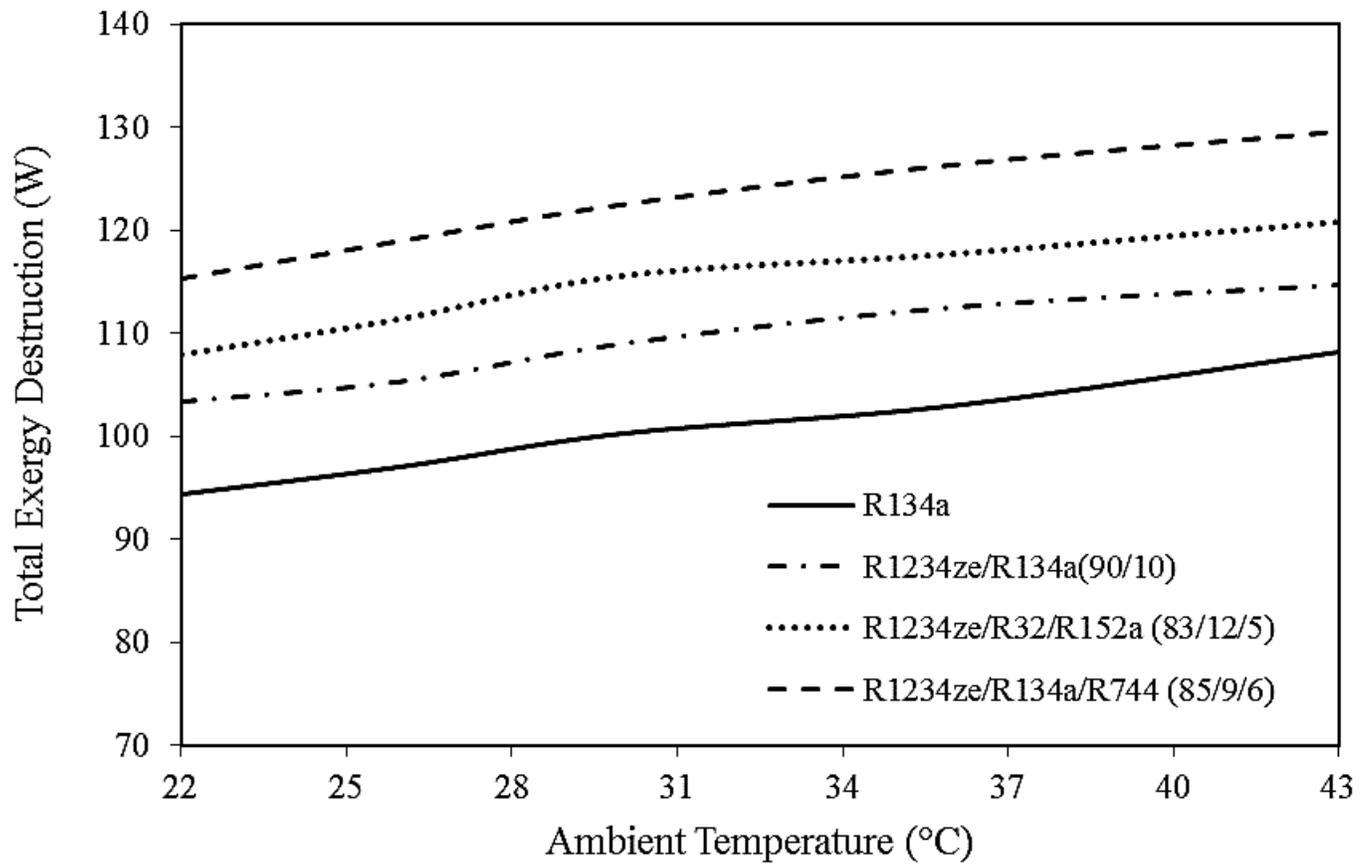


Figure 9

Variation of total exergy destruction with ambient temperature

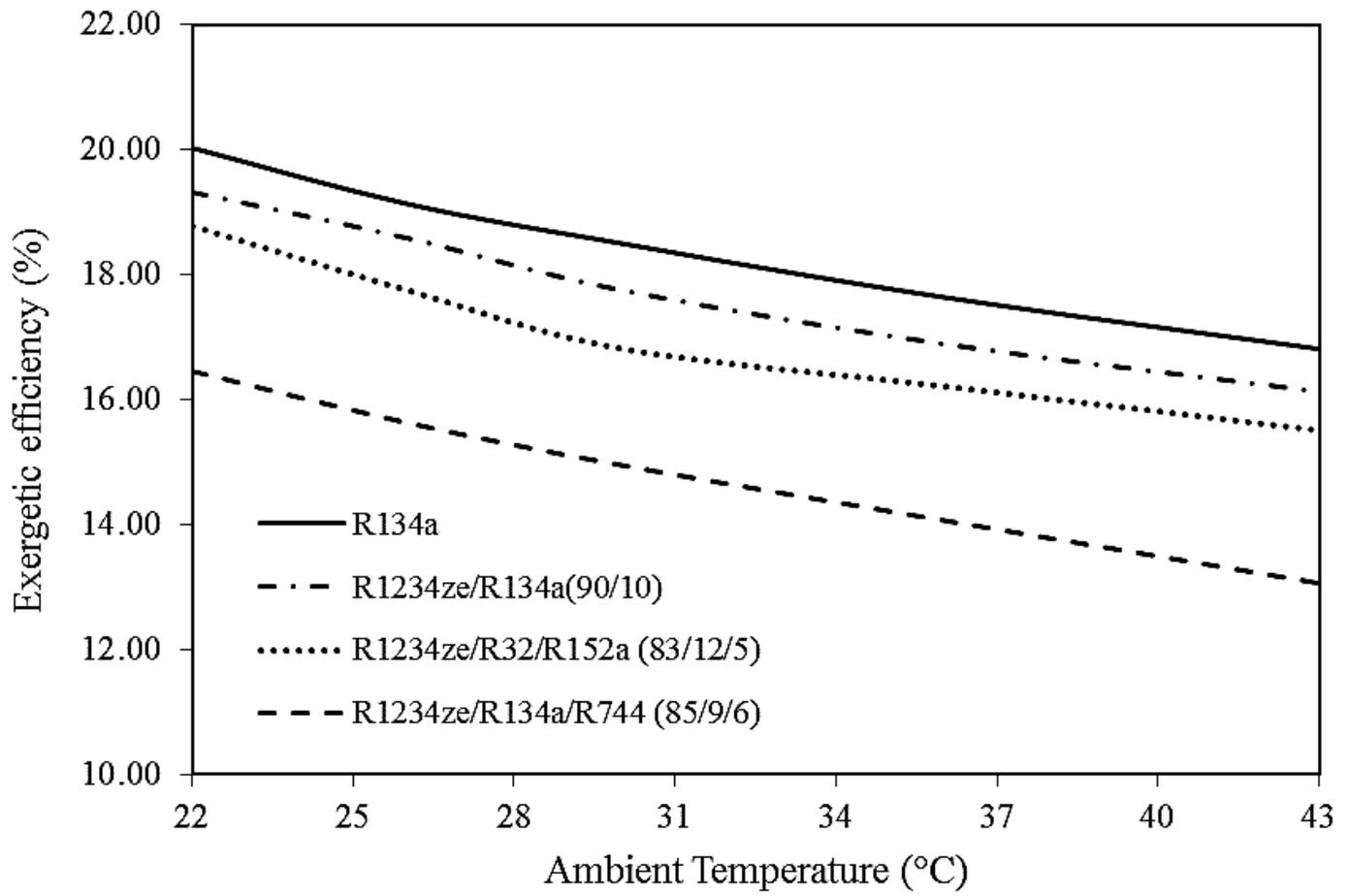


Figure 10

Variation of exergetic efficiency with ambient temperature