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# Modification of N-Terminal a-Amine of Proteins via Biomimetic ortho-quinone-mediated Oxidation

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#### 1 Modification of N-Terminal α-Amine of Proteins via Biomimetic ortho-quinone-

#### 2 mediated Oxidation

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#### 15 ABSTRACT

16 Naturally abundant quinones are important molecules, which play 17 essential roles in various biological processes due to their reduction potential. In 18 contrast to their universality, the reactions between quinones and proteins remain 19 sparse. Herein, we report the development of unprecedented strategy to protein 20 modification via a biomimetic quinone-mediated oxidation at the N-terminus. By 21 exploiting unique reactivity of an ortho-quinone reagent, the  $\alpha$ -amine of protein 22 N-terminus was oxidized to generate aldo or keto handle for orthogonal 23 conjugation. Its applications have been demonstrated using a range of proteins, including myoglobin and ubiquitin. The effect of this method was further 24 25 highlighted via the preparation of a series of 17 MIP-1<sup>β</sup> analogs, followed by 26 preliminary anti-HIV activity and cell viability assays, respectively. This method 27 offers a fast, efficient and complementary approach to existing strategies for 28 protein N-terminus modification.



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Figure 1: Development of protein N-terminus modification via quinone-mediated
 transamination: A) Natural catalytic cycle of CuAOs, B) Oxidation of primary amines
 with ortho-quinone to form imines, C) Biomimetic quinone-mediated transamination
 inspired by CuAOs.

34 Oxidation reactions play an important role in organic chemistry and are widely involved in crucial biological transformations.<sup>1-2</sup> In line with the broad interest in the 35 36 development of more selective and mild oxidation reactions, ortho-quinone cofactors 37 of copper amine oxidases (CuAOs) have been extensively studied due to their high 38 catalytic efficiency. In the catalytic cycle, the Tyr side chain is converted into lysyl 39 tyrosylquinone, which can readily oxidize primary amines into aldehydes via a 40 quinone-mediated transamination pathway using  $O_2$  as a co-oxidant to complete the catalytic cycle (Fig. 1A).<sup>3-4</sup> Recently, significant progress has been achieved toward the 41 design of quinone-based catalysts with major contributions from Corey,<sup>5</sup> Fleury,<sup>6</sup> 42 Kobayashi,<sup>7</sup> Stahl<sup>8-10</sup> and Luo (Fig. 1B).<sup>11-13</sup> These biomimetic quinone-based catalysts 43 44 show specific chemoselectivity toward the dehydrogenation of primary, secondary, tertiary amines or other reactions.<sup>14-16</sup> The applications of quinone oxidation have been 45 46 largely limited to small molecule transformations despite the significant progress in this area of research.<sup>17-18</sup> On the other side, in nature, the interactions between quinones and 47 48 proteins are important but complex in fundamental process of life. Early work was 49 provided by Mason in 1955, who reported that o-quinones can react with proteins to form a colored complex via the N-terminal residue.<sup>19</sup> In addition, many research studies 50 51 have been carried out on the role of quinone in protein cross-linking.<sup>20-22</sup> However, the 52 relationships between quinones and proteins have not been fully elucidated on a 53 molecular level. Nevertheless, the intrinsic design of quinones may provide an effective 54 opportunity to modify proteins under physiological conditions.

55 Existing methods used to modify proteins are mainly confined to nucleophilic 56 amino acids, such as Cys and Lys, and hence result in heterogeneous conjugates because of the high frequency of these amino acids.<sup>23-25</sup> In this context, the emergency of 57 selective modification of the N-terminus of proteins has led to the single site 58 functionalization of proteins.<sup>26-30</sup> Although significant progress has been made, many 59 60 current methods relied on assistance of a side-chain functional group, such as  $\beta$ nucleophilically-functionalized Cys<sup>31-35</sup> or Ser/Thr.<sup>36-38</sup> Transamination reactions 61 62 performed under physiological conditions have been elegantly developed by Francis and co-workers utilizing pyridoxal-5-phosphate (PLP)<sup>39-46</sup> or N-methylpyridinium-4-63 carboxaldehyde benzenesulfonate salt (Rapoport's salt)<sup>47</sup> as oxidants, which can 64 65 convert the N-terminus of proteins into aldo or keto functionalities for oxime ligation or other bio-conjugation reactions.<sup>48-50</sup> However, the efficiency of these methods relies 66 on the specific N-terminal sequence<sup>51</sup> and prolonged reaction times in the presence of 67 PLP can potentially result in denaturation.<sup>52</sup> Thus, given the generality of quinone and 68 69 its derivates in the effective oxidation of amines, we hope to investigate whether or not 70 the quinones may be suitable for the selective oxidation of the N-terminal  $\alpha$ -amine of 71 protein to aldehyde or ketone with fast kinetics as well as a wide scope of amino acids. 72 Herein, we report a selective, efficient and rapid method for modifying the N-terminus 73 of proteins via a quinone-mediated oxidation of N-terminal  $\alpha$ -amine of complex 74 peptides and proteins under physiological conditions (Fig. 1C). Several examples have 75 been demonstrated using a range of peptides and proteins, including ubiquitin and 76 myoglobin. Moreover, we prepared a library of macrophage inflammatory protein- $1\beta$  $(MIP-1\beta)^{53-54}$  analogues using a combination of native chemical ligation and quinone-77 78 mediated transamination, where the late-stage modification of the N-terminus of MIP-79 1β can lead to a 20-fold increase in its anti-HIV-1 activity.

80 To begin our journey, a range of quinone derivates 1-6 were prepared with 81 different functionalities with an aim toward fine-tuning their reactivity to be selective 82 for the  $\alpha$ -amine selective rather than the  $\epsilon$ -amine of Lys, as well as other functionalities.<sup>7</sup>, <sup>11-12</sup> We tested the transamination reaction of model peptide 7 (1 mM) with the sequence 83 of GFHAKGY in an aqueous solution buffered at pH 6.5 using 10 mM (saturated) of 84 85 each of the quinone (1-6) as the oxidant (Fig. 2A). The oxidized product was 86 subsequently reacted with EtONH<sub>2</sub> to form its corresponding oxime 8. 4-tert-Butyl-5-87 methoxy-4-*tert*-butyl-o-benzylquinone 1 achieved near quantitative conversion (>90%) 88 to generate product 8 at pH 6.5 in 3 h with <2% Lys oxidation observed *via* LCMS due 89 to the extraordinary low pKa of the N-terminal amine compared to that of the Lys side chain amine .<sup>26</sup> 4-*tert*-Butyl-o-benzylquinone 2 could also achieve 55% conversion at 90 91 pH 6.5; the major by-product 9 was attributed to the undesired Lys side chain oxidation 92 (>40%) (Fig. 2B). Further optimization of the pH of the transamination reaction with 93 either quinone 1 or 2 showed that methoxyquinone 1 has an optimal pH range of 6.0-6.5, while the side reaction during using quinone 2 can be minimized by lowering 94 reaction pH to 4.0 (90% conversion to give target product 8) (Fig. 2C). Thus, 95 96 methoxyquinone 1 serves as the most effective oxidant (see Section 8 in the SI for the 97 discussion of the mechanism). A time-course monitoring study was performed to verify 98 that the reaction reached to completion in 3 h at room temperature (see section 3.1.3. in the SI). Although an early work demonstrated that using Cu<sup>2+</sup> salts is necessary for N-99 termini oxidation with glyoxylate as reagent,<sup>55</sup> addition of Cu<sup>2+</sup> in this reaction did not 100 101 increase the yield of the oxidative product formed. Interestingly, when compared with 102 such reactions with small molecules, quinone-mediated transamination can readily push 103 towards completion (90% conversion) without formation of the imine derivative of the 104 starting material, which may potentially quench the reaction.



Figure 2. Optimization of the reaction conditions used in the biomimetic transamination reaction. A) Screening of oxidants 1-6 used for the transamination reaction with model peptide 7; B) LC trace obtained for the transamination reaction using quinone 1 (in yellow) or 2 (in blue); C) Optimization of the pH of the transamination reaction using quinone 1 (in yellow) or 2 (the rate of conversion was calculated by integrating the related peaks of the LC trace measured at 280 nm, although TIC showed a >99% conversion rate of product 8).

113 Next, to investigate the scope of the N-terminal residue, we expanded the N-114 terminal residue of peptide 7 to the remaining 19 amino acid residues 10-28. As it shows 115 in Fig. 3, Arg peptide 10, Leu peptide 11, Met peptide 12, Glu peptide 13 and Lys 116 peptide 14 reacted cleanly and gave satisfactory conversion (>60%) in 3 h to afford their desired products without any significant observation of the expected side-products. 117 Asp peptide 15 underwent decarboxylation during the oxidation and thereafter formed 118 119 the methylated product (R = Me, see SI). Ser peptide 16, Thr peptide 17 and Trp peptide 120 18 were oxidized, but all formed the side-chain cleaved oxime product 8. Peptides 19-23 (containing Ala, Phe, Ile, Tyr and Val as N-termini, respectively) can also be 121 122 converted into their desired oxime products, but with lower conversions. Asn peptide 123 24 and Gln peptide 25 were oxidized in the first step, but the resulting side-chain 124 cyclized ketal product could not be effectively reopened using ethoxyamine, which 125 caused a decrease in conversion to the desired oxime product. His peptide 26 failed to 126 be oxidized due to the quinone oxidant 1 remaining bonded to the peptide, which was

in line with that previously reported and attributed to a Pictet-Spengler-type side
reaction.<sup>40</sup> Unfortunately, Pro peptide 27 was not oxidized as a secondary amine. For
comparison, Cys peptide 28 decomposed (see section 3.2. in the SI for predicted
structures of all by-products).





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136 After screening the scope of N-terminal residue in the reaction, we next applied 137 this method to more complex therapeutic peptides and proteins. The first target chosen 138 was COVID-19 spike protein 319-347 fragment, which was located in its receptor 139 binding domain. The peptide has an Arg residue at its N-terminus and a free cysteine 140 residue, which originally forms a disulfide bond with another Cys in the spike protein. 141 However, the free Cys residue was not tolerated under our oxidative conditions. An 142 Acm group was introduced to the peptide during this stage. Due to the low solubility of 143 this peptide, the transamination step was carried out in 2 M Gn·HCl buffer with all the 144 other reaction conditions unchanged with the exception of the reaction time, which was 145 also extended to 6 h. The reaction proceeded smoothly to afford the oxime product 29 with 85% conversion (Fig. 4A). Tetracosactide, a therapeutic peptide bearing an N-146 147 terminal Ser, one Met and four Lys residues in its sequence, was successfully oxidized under the standard reaction conditions in 3 h. The desired oxime product 30 was formed
with 80% conversion monitored by LCMS (Fig. 4B).

150 The scope of this transformation was next examined using myoglobin, which 151 contains 18 Lys residues in the sequence (Fig. 4C). This reaction was conducted in 152 aqueous buffer at pH 6.0 to minimize over-oxidation of the Lys side chain and thus the 153 reaction time was extended to 5 h. Due to the limitation of the chromatographic 154 techniques used, total ion counts (TICs) were used to determine the conversion rate 155 after a simple workup procedure (see section 3.5 in the SI). The desired ethyl oxime 156 product **31a** was observed with 70% conversion, 15% unreacted starting material and 157 15% of the over-oxidized by-product. Furthermore, a biotinylated tagcwas synthesized 158 and incorporated to myoglobin. However, due to the low reactivity of this tag, an excess 159 of oxidant 1 could not be quenched immediately, which then led to further oxidation 160 during the oxime formation step. To address this issue, once the first transamination 161 reaction was complete, an ethyl acetate extraction step was carried out to remove the 162 excess oxidant. The resulting solution underwent the oxime formation step overnight 163 to afford a 40% conversion of biotinylated protein **31b**.

In nature, many proteins are expressed with Met as their N-terminus. Therefore,
we examined this transformation with ubiquitin, which contained an N-terminal Met.
In 2 M Gn·HCl buffer, ubiquitin was easily converted into its oxime product 32 with
77% conversion (Fig. 4D).

With all the examples above, we confirmed that ortho-quinone 1 was able to oxidize peptides or proteins with various N-termini. Therefore, it can serve as an alternative strategy to existing methods and used to expand the application of transamination in biological research.

172 Considering the fast kinetics and wide N-termini tolerance of our transamination
173 reaction, we applied this methodology to a medicinal target, macrophage inflammatory



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Figure 4. Site selective peptide modification via 175 quinone-mediatated transamination reaction. A) Reaction conditions for COVID-19 spike protein 176 fragment 29: 1) Transamination: 2 M Gn·HCl, oxidant 1 (10 mM, sat.), 200 mM 177 Na<sub>2</sub>HPO<sub>4</sub>, pH 6.5, 6 h, 25 °C; 2) Oxime formation: EtONH<sub>2</sub>·HCl (100 mM), pH 4.0, 3 178 h, 25 °C. B) Reaction conditions for tetracosactide 30: 1) Transamination: oxidant 1 179 (10 mM, sat.), 200 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 6.5, 3 h, 25 °C; 2) Oxime formation: 180 EtONH<sub>2</sub>·HCl (100 mM), pH 4.0, 3 h, 25 °C; C) Reaction conditions for myoglobin 181 182 31a-b: 1) Transamination: oxidant 1 (10 mM, sat.), 200 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 6.0, 5 h, 25 °C; 2) Oxime formation for **31a**: EtONH<sub>2</sub>·HCl (100 mM), pH 4.0, 3 h, 25 °C; oxime 183 formation for **31b**: extraction with EtOAc (2 x 1.0 mL), then biotinylated tag **S4** (100 184 185 mM), pH 4.0, 16 h, 25 °C. D) Reaction conditions for ubiquitin **32**: 1) Transamination: 2 M Gn·HCl, oxidant 1 (10 mM, sat.), 200 mM Na<sub>2</sub>HPO<sub>4</sub>, pH 6.5, 3 h, 25 °C; 2) Oxime 186 formation: EtONH<sub>2</sub>·HCl (100 mM), pH 4.0, 3 h, 25 °C. 187

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protein-1β (MIP-1β). MIP-1β (CCL4)<sup>56</sup> exhibits potent anti-HIV-1 activity<sup>57</sup> by binding
to the hydrophobic transmembrane helix bundle of CCR5 via its N-terminal domain.<sup>58-61</sup> Due to the increasing viral resistance and inefficient therapies, the discovery of new
HIV-1 inhibitors is still highly coveted.<sup>62-63</sup> Therefore, diversification of the N-terminal
region of CCLs can be greatly beneficial toward the development of new peptide HIV-
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1 inhibitors.<sup>64-67</sup> However, current methods rarely facilitate such direct modifications
of its N-termini. Thus, the late-stage modification of CCL N-termini will afford an ideal
strategy for this area of research.

196 Thus, a merging of late-stage oxidation and state-of-the-art peptide ligation 197 method will allow the fast generation of a protein library for the rapid screening of 198 potential drugs. We synthesized a truncated MIP-1 $\beta$  4-96 protein **33**, which contained 199 Gly4 and the full-length form of MIP-1 $\beta$  **34** with the mutation of Ala1 to Gly for 200 oxidation (Fig. 5B). Both proteins were chemically synthesized *via* native chemical 201 ligation<sup>68-69</sup> and re-folded using Cys/Cys<sub>2</sub> redox<sup>70</sup> buffer (see SI) on a >30 mg scale (see 202 section 4 for the synthesis of MIP-1 $\beta$  variants **33** and **34**).

203 Having established a facile protocol to prepare MIP-1ß proteins on a large-scale, 204 we next performed the protein N-terminal transamination-oxime ligation reaction to 205 modify the protein. Various alkyl alkoxyamines were chosen to optimize the anti-HIV-206 1 activity of MIP-1β due to their hydrophobicity (Fig. 5A). Hexyl, (*E*)-hexyl, (*Z*)-hexyl, 207 geranyl and decyl hydroxylamine were used to investigate the influence of the 208 hydrocarbon chains with different conformation or chain lengths toward fine-tuning the 209 bioactivity (Fig. 5C). In addition, we also synthesized PEGylated hydroxylamine as a 210 hydrophilic tag for comparison. The oxidation reactions were performed in 2 M 211 guanidine buffer, because this concentration of guanidine can solubilize target proteins 212 without denaturing them. Oxidant 1 was added into the target protein 33 or 34 in buffer 213 solution at pH 6.5 and the solution was incubated at 25°C for 3 h. Subsequently, after 214 the addition of the alkoxyamine, the mixture was readjusted to pH 6.5 and reacted for a further 16 h to afford modified proteins 35-49 (Fig. 5C). Because MIP-1ß proteins 215 and MeONH2·HCl or EtONH2·HCl were dissolved well in this buffer solution, the 216 217 ligations proceeded smoothly and gave high isolated yields (43% for 35, 44% for 42 218 and 47% for 43 respectively). Reactions with fatty alkoxyamines became biphasic due 219 to their poor solubility. Therefore, vigorous stirring was necessary during these 220 reactions. After 24 h of reaction, a small amount of starting material was still present

(~20% by HPLC analysis), which suggested that these lipo-hydroxylamines did not mix
well even in an aqueous buffer and hence resulted in a decreased yield (28–38%).
PEGylated variants (41 and 49) were extremely low yielding (23 and 25%, respectively)
due to unknown side reactions. Transamination with quinone 1 is fully compatible with
the disulfide bonds in these proteins without disrupting their biological activity, as
shown by HPLC-MS and the anti-HIV-1 assay.



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Figure 5. Late-stage modification of MIP-1 $\beta$  proteins *via* quinone-mediated selective oxidation of N-terminal amine. A) Reaction scheme for the transamination of MIP-1 $\beta$  protein 33 and 34; B) The sequences of protein 33 and 34; C) A summary of the transamination/oxime ligation of protein 33 and 34 and the isolated yields for all reactions; D) Anti-HIV-1 activity of selected MIP-1 $\beta$  analogs with EC50 <5000 nM.

With all the modified MIP-1 $\beta$  variants in hand, we performed an anti-HIV-1 activity assay to evaluate their inhibitory activity. Recombinant MIP-1 $\beta$  (**Rh-CCL4**) was used as the primary standard with a measured half maximal effective concentration (EC<sub>50</sub>) value of 3.3  $\mu$ M (Fig. 5D). Protein **33** (4-69) resulted in a complete loss of anti-HIV-1 activity within the concentration range studied (up to 5  $\mu$ M). Interestingly, ethylated variant **35** regained its activity with an EC<sub>50</sub> value of 200 nM. Full-length Gly1-69 protein **34** also displayed increased inhibitory activity with an EC<sub>50</sub> value of 240 157 nM upon substitution of Ala1 with Gly, which suggested that a branched methyl 241 group decreases the binding affinity between MIP-1 $\beta$  and CCR5. Further extension of 242 full-length protein 34 with methoxyamine and ethoxyamine showed comparable 243 activities against HIV-1 by an order of magnitude (42:  $EC_{50} = 2.7 \mu M$  and 43:  $EC_{50} =$ 2.3 µM), whereas all the lipo-variants and PEGylated variants exhibited no activity 244 245 within the concentration range studied (all  $EC_{50} > 5 \mu M$ , see SI). Besides, a cell viability 246 assay with MTT was also performed to explore the cytotoxicity of the synthetic proteins 247 toward Tzm-bl cells. In our investigations, all the synthetic proteins exhibited 248 cytotoxicity with 50% cytotoxic concentration (CC<sub>50</sub>) values of >50  $\mu$ M (see SI), which 249 indicated that the proteins showed no cytotoxic behavior. In our future studies, we will 250 further optimize the N-terminal domain with hydroxyamines bearing other 251 functionalities and thereafter explain the exact mechanism of how these modified proteins exhibit their inhibitory activity against HIV-1 both in vitro and in vivo. 252

#### 253 CONCLUSION

254 In summary, we have demonstrated a highly selective and mild method for 255 modifying the N-terminus of protein via a biomimetic quinone-based transamination 256 reaction. The key features of this reaction include rapid reaction time, high selectivity and a broad scope of amino acids. Covid-19 spike protein fragment (Arg), 257 258 tetracosactide (Ser), myoglobin (Gly) and ubiquitin (Met) along with a range of model 259 peptides were given as examples. Furthermore, the preparation of several MIP-1 $\beta$ 260 analogs via native chemical ligation and late-stage N-terminal modification has 261 demonstrated its potential utility in medicinal chemistry. We believe this method will 262 be a useful tool for protein modification and medicinal chemistry studies.

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## **Figures**



## Figure 1

Development of protein N-terminus modification via quinone-mediated transamination: A) Natural catalytic cycle of CuAOs, B) Oxidation of primary amines with ortho-quinone to form imines, C) Biomimetic quinone-mediated transamination inspired by CuAOs.



## Figure 2

Optimization of the reaction conditions used in the biomimetic transamination reaction. A) Screening of oxidants 1-6 used for the transamination reaction with model peptide 7; B) LC trace obtained for the

transamination reaction using quinone 1 (in yellow) or 2 (in blue); C) Optimization of the pH of the transamination reaction using quinone 1 (in yellow) or 2 (the rate of conversion was calculated by integrating the related peaks of the LC trace measured at 280 nm, although TIC showed a >99% conversion rate of product 8).



## Figure 3

Landscape for the transamination of 20 canonical amino acids attached on the N-terminal residue of the model peptides. Reaction conditions: 1) Transamination: oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.5, 3 h, 25 °C; 2) Oxime formation: EtONH2·HCl (100 mM), pH 4.0, 3 h, 25 °C.



### Figure 4

Site selective peptide modification via quinone-mediatated transamination reaction. A) Reaction conditions for COVID-19 spike protein fragment 29: 1) Transamination: 2 M Gn·HCl, oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.5, 6 h, 25 °C; 2) Oxime formation: EtONH2·HCl (100 mM), pH 4.0, 3 h, 25 °C. B) Reaction conditions for tetracosactide 30: 1) Transamination: oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.5, 3 h, 25 °C; 2) Oxime formation: EtONH2·HCl (100 mM), pH 4.0, 3 h, 25 °C; C) Reaction conditions for myoglobin 31a-b: 1) Transamination: oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.5, 2) Oxime formation: oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.5, 3 h, 25 °C; 2) Oxime formation: oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.0, 5 h, 25 °C; 2) Oxime formation for 31a: EtONH2·HCl (100 mM), pH 4.0, 3 h, 25 °C; oxime formation for 31b: extraction with EtOAc (2 x 1.0 mL), then biotinylated tag S4 (100 mM), pH 4.0, 16 h, 25 °C. D) Reaction conditions for ubiquitin 32: 1) Transamination: 2 M Gn·HCl, oxidant 1 (10 mM, sat.), 200 mM Na2HPO4, pH 6.5, 3 h, 25 °C; 2) Oxime formation: EtONH2·HCl (100 mM), pH 4.0, 3 h, 25 °C. D) Reaction



## Figure 5

Late-stage modification of MIP-1 $\beta$  proteins via quinone-mediated selective oxidation of N-terminal amine. A) Reaction scheme for the transamination of MIP-1 $\beta$  protein 33 and 34; B) The sequences of protein 33 and 34; C) A summary of the transamination/oxime ligation of protein 33 and 34 and the isolated yields for all reactions; D) Anti-HIV-1 activity of selected MIP-1 $\beta$  analogs with EC50 <5000 nM.

## **Supplementary Files**

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• Slquinoneoxidation0824.pdf