

Production of bioethanol from wheat straw via optimization of co-culture conditions of *Bacillus licheniformis* and *Saccharomyces cerevisiae*

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Abstract

Bioethanol production has been a challenge for the researchers to enhance the bioethanol yield. In this study, we are reporting an efficient novel method to produce bioethanol. The process comprises co-culture technique to produce bioethanol from wheat straw by co-culturing *Bacillus licheniformis* and *Saccharomyces cerevisiae*. Simultaneous saccharification and fermentation allows wheat straw hydrolysis by cellulase enzyme produced by *Bacillus licheniformis* and conversion of produced reducing sugar into ethanol by *Saccharomyces cerevisiae*. Pre-treatment of wheat straw and optimization of co-culturing parameters like, time, pH, substrate concentrations and nitrogen source concentrations gave a net yield of 5.67% v/v bioethanol. Scale up of optimised media to fermenter has resulted in a significant enhancement of bioethanol production to 18.64% (v/v).

1. Introduction

In the present state, world's requirement for fuel is increasing at high rate in contrast to the fossil fuels [1]. Production of fuel from waste products have the potential to provide a solution for cost effectiveness and sustainability. Ethanol is an eco-friendly fuel in contrast to other substrates which causes green-house gas emissions when employed as fuel [2]. It is now being used as an additive to biodiesel and have increased the biodiesel's energy yield for automobile industry [3]. And is being employed as fuel for many industrial applications in many countries [4]. In pharmaceutical industry, ethanol is a major solvent for the synthesis of drugs and their derivatives [5]. Due to increase in demand of ethanol and high cost of production using traditional substrates like molasses and malt, ethanol produced is insufficient and expensive. There is need of a cost effective and an abundant source of ethanol [6]. A process is required which can simply convert substrate into ethanol with less complexity and cost effectiveness.

Production of ethanol from waste lignocellulosic agricultural products would be able to provide an opportunity to replace the conventional expensive methods of production [7]. As lignocellulosic substrate is one of the most abundant, renewable, and low-cost substrates for ethanol production. Wheat straw is excessively available in nature as an agricultural waste and contains 35–45% cellulose [8]. However, it is required to give a pre-treatment to the raw wheat straw to facilitate hydrolysis as well as ethanol production. It has been observed that pre-treatment of lignocellulosic substrate has increased the availability of the substrate for glucose production and increased the ethanol yield [9]. Various pre-treatment methods have been employed by the researchers involving physical, chemical, and enzymatic pre-treatments [10]. Bioethanol production via enzymatic pre-treatment with cellulolytic enzymes followed by microbial fermentation is an efficient method. However, enzymatic methods are expensive and are substrate specific [11]. Physical and chemical pre-treatments of raw cellulose sources have been optimized and is cost-effective [10].

After pre-treatment of the wheat-straw, cellulose fibres have been observed to become more prone to microbial hydrolytic degradation in comparison to raw substrate [12]. Microbial hydrolytic degradation involves the release of cellulolytic enzymes by the microbes to produce monosaccharides. In the next

step for bioethanol production, monosaccharide rich medium is inoculated with the ethanol producing microbes. Several bacterial and yeast strains have been employed and optimized to carry out the stepwise production of bio-ethanol production. To decrease the number of steps involved in bioethanol production, researchers have developed genetically engineered strains which can both degrade as well as convert degraded cellulose to ethanol [13]. In addition, immobilisation is another technique employed to produce ethanol where one strain is immobilised for the degradation of lignocellulose substrate and microbes are added in mobile phase for ethanol production [14]. However, cost for using two different systems for separate enzyme production and the instability of genetically engineered strains is major problem being faced by the industry. Co-culture is a technique which has been used by researchers where combination of two microbes have been employed to carry out the stepwise conversion in a controlled way [15]. Simultaneous saccharification and fermentation has been explored by the researchers by employing various combinations of microorganisms and enzymes. SSF of alkali treated paddy field was carried out xxxxxx and coworkers by using 2g/l commercial enzymes (Cellulase T "Amano 4" : Cellulase "Onozuka 3S": Pectinase G "Amano") and 1M of circinelloide at 10% solid biomass loading at 28°C and pH of 5.5 and

Have produced 30.5 g/l of ethanol in 36 hours [16]. However, employment of commercial enzymes would affect the cost effectiveness of the ethanol production. Also ,microbes have been employed by the researchers to produce cellulases for the hydrolysis of complex cellulose substrates. In case of microbes, the major challenge is the selection of microbial strain due to difference in the working temperature and pH of the hydrolytic enzymes and the fermenting strains. Most common fermenting strains employed is *Saccharomyces cerevisiae*, which ferments at 30-37°C and 6–7 pH. Microbe like fungal strains have been utilized for the hydrolysis of complex cellulose substrates. However, the optimum temperature and Ph for the activity of cellulases are 40-60°C and pH 4.8–5.5 [17]. *B. licheniformis* has been studied extensively for its potential for producing cellulases to degrade complex cellulose substrates [18, 19].

In this study, co-culture technique is used to produce ethanol in one system to produce ethanol from wheat straw instantaneously. Process of ethanol production was optimized by varying the fermentation time, pH, substrate concentration and nitrogen source concentration to enhance the yield of bioethanol. *Bacillus licheniformis* was employed for the hydrolytic degradation of lignocelluloses and *Saccharomyces cerevisiae* for bioethanol production for simultaneous saccharification and fermentation (SSF). Simultaneous saccharification and fermentation took place in co-culture, where *Bacillus licheniformis* and *Saccharomyces cerevisiae* grow together but *Saccharomyces cerevisiae's* growth totally is dependent on glucose produced by the *Bacillus licheniformis*. The process was further scaled up to bioreactor, where SSF has resulted in the production of 18.64% v/v bioethanol under optimized conditions.

2. Material And Methods

2.1 Materials

Soluble starch, sodium thiosulphate, phenol and sulphuric acid were purchased from Lobachemie Pvt. Ltd. Wheat straw was procured from local farmer of PBW621 variety. Potassium dichromate and peptone was purchased from central drug house, India. Yeast extract, agar and ammonium nitrate were purchased from Titan Pvt. Ltd, India. Dextrose, 3,5-dinitrosalicylic acid and sodium meta bisulphite were purchased from Molychem Pvt. Ltd. and potassium iodide and potassium hydroxide was purchased from Thomas Baker (chemicals) Pvt. Ltd., India. Rochelle salt was obtained from Qualikems Pvt. Ltd., India.

Saccharomyces cerevisiae (MTCC No. 464) and *Bacillus licheniformis* (MTCC No. 429) was procured from IMTECH, Chandigarh.

2.2 Preparation of wheat straw media

The PBW621 variety of wheat straw was dried in hot air oven and grounded to proper size using a blender. 5 mg/ml of wheat straw and 2.5 mg/ml of ammonium nitrate was added in 100ml of double distilled water and was autoclaved at 15 Psi at 121°C.

2.3 Alkali pretreatment of wheat straw media

Pretreatment of wheat straw media was carried out by treating wheat straw media with 1% w/v potassium hydroxide and followed by 24 hours incubation at room temperature. Further the alkali treated media was autoclaved at 15 psi and 121°C [20, 21].

2.4 Glucose/ Reducing sugars estimation by DNS (Dinitrosalicylic acid) method

Reducing sugars were quantified as per earlier reported method by Miller [22]. 1 ml of supernatant was taken in a test tube and 3 ml DNS reagent was added to it. The tubes were heated in a water bath at 80°C for 20 minutes. Further, the tubes were cooled to room temperature and optical density (O.D) was measured at 540nm by using UV spectrophotometer (Sistrionics). A standard curve was prepared by using 1mg/ml glucose solution for the estimation of reduced sugars.

2.4 Simultaneous saccharification and fermentation (SSF)

In a typical reaction, the nascent and pretreated wheat straw media was inoculated with *Bacillus licheniformis* (10^8 CFU/ml) and incubated in an incubator shaker at 30°C for 48 hours. Further, the media was co-cultured with *Saccharomyces cerevisiae* (10^8 CFU/ml) and incubated in an incubator shaker at 30°C for 120 hours. The effect of different parameters on production of ethanol was optimized by changing the co-inoculation time (24, 48, 72 hours), substrate concentration (2.5, 5.0, 7.5 and 10.0 mg/ml), pH of media (pH 5, 6, 7, 8) and concentration of nitrogen source (0.03, 0.06, 0.09 and 0.12 M). Samples were taken for ethanol quantification at every 24 hours interval from each flask using potassium dichromate redox titration method and gas chromatography [23]. The best optimized reaction condition was scaled up from 100 ml to 1.5 liters pre-sterilized fermenter. The fermenter conditions were set to 30°C and agitation speed at 121 rpm [24]. Aeration was given for cell growth followed by anaerobic condition as per earlier reported method [25].

2.5 Scale up of SSF

The BIOAGE fermenter of 3 litres capacity was filled with 2 litres distilled water and autoclaved. Further, autoclaving along with the fermentation medium was done after setting the pH to 7 and media was allowed to cool down to room temperature. *Bacillus licheniformis* was inoculated in 1.5 litres of fermentation medium with 10^8 CFU/ml. The fermenter conditions were set to 30°C and agitation speed was set at 121 rpm. After 72 hrs *Saccharomyces cerevisiae* was inoculated. Aeration was given for cell growth followed by anaerobic condition at 10^8 CFU/ml [14]. For determining the concentration of ethanol sample was centrifuged and supernatant was collected separately. And sample was analysed by GC [15].

2.7 Ethanol estimation by potassium dichromate method

1 ml of the fermented broth was centrifuged and dispersed into round bottom flasks containing 30 ml of distilled water fixed to a distillation column cooled down by cold tap water. A conical flask containing 25 ml of potassium dichromate solution (33.768 g of $K_2Cr_2O_7$ dissolved in 400 ml of distilled water with 325 ml of sulphuric acid and volume raised to 1 litre) was fixed to the other end of the distillation column to collect the distillate. A heating mantle with the temperature adjusted to 78°C was used to heat the round bottomed flask. 20 ml of distillate was collected in each sample and the flasks were kept in a water bath maintained at 62.5°C for 20 minutes. The flasks were cooled to room temperature and 30 ml distilled water was added to it to make the final volume 50 ml. this sample is now used for measuring the optical density at 600 nm using UV spectrophotometer. A standard curve was prepared under similar set of conditions by using standard solution of ethanol containing 2 to 12% (v/v) ethanol in distilled water. The percentage ethanol concentration of ethanol produced was obtained by comparing its optical density with the standard ethanol density curve.

2.7 Bioethanol estimation by Gas chromatography (GC) method

Ethanol present in the fermentation broth was quantified by gas chromatography method. Agilent 6890 gas chromatograph equipped with Flame Ionization Detector (FID) was employed for the separation and quantification of ethanol. A Zebron column was fitted into the instrument to provide on column injection. The detector and injector temperature were maintained at 260°C. The gas chromatograph was connected to an integrator and computer system to determine area of ethanol and internal standard peak.

Formula to find ethanol% v/v from GC data analysis:-

Response factor = Peak area / sample amount

Amount of analyte = peak area / response factor

w/v value is then converted to % v/v

3. Results And Discussion

For ethanol production, wheat straw media was prepared and pre-treated with alkali to facilitate the hydrolysis and fermentation process as per earlier reported method as shown in Fig. 1. The steam and alkali pre-treatment of wheat straw demonstrated an increase in glucose production from 0.184 mg/ml to 0.223 mg/ml compared to untreated wheat straw after 24 hours of co-culturing (Fig S1). Alkali pre-treatment has been observed to increase the degradation of hemicellulose and lignin. Increase in degradation is due to the removal of acetate group from the hemicellulose, increasing the access of hydrolytic enzymes to cellulose [26]. In addition to this, alkali addition results in lignin solubilization which also enhance the enzyme susceptibility [27]. Matrix of cellulose and lignin are held together by hemicellulose chains in a lignocellulosic complex. During pre-treatment, the lignin-cellulose matrix breaks down and reduce the crystallinity by enhancing the proportion of the amorphous phase [28].

After pre-treatment of the wheat straw media, the media was inoculated with *B. licheniformis* to produce reducing sugars by enzymatic hydrolysis. Later, media was inoculated with *S. cerevisiae* for co-culturing to produce bioethanol via SSF. Time point for the inoculation of *S. cerevisiae* for co-culturing was optimized by inoculating the *S. cerevisiae* at 0th hour, 24th hour and 48th hour after the inoculation of *B. licheniformis*. It was observed that inoculation at 48th hour has produced highest bioethanol in comparison to other time points (shown in figure S2). Owing to sufficient time provided for *B. licheniformis* to produce reducing sugars helps in enhancing the ethanol production. Enzymatic hydrolysis plays an important role in enhancing the production of reducing sugars and would further increase the ethanol production. For the better enzymatic activity, environmental conditions play an important role. pH and temperature are the most important factors and have been optimized to improve enzymatic activity as well as growth of micro-organisms. In this study, pH was optimized for co-culturing by performing SSF at different pH (5,6,7 and 8) at 37°C. in Fig. 2, higher production of reducing sugars as well as ethanol demonstrates the suitable pH required for the SSF. The ethanol production was increase from $1.04 \pm 0.15\%$ to $2.46 \pm 0.18\%$ v/v, when the pH was changed from 5 to 7 after 96 hours of SSF. However, further increase in pH to 8 has decreased the ethanol production to $2.18 \pm 0.034\%$ v/v (Fig. 2A). *In situ* analysis of ethanol and glucose was carried out up to 96 hours of SSF as shown in Fig. 2B. Decrease in glucose concentration with increase in ethanol production demonstrated the conversion carried out by *S. cerevisiae* under anaerobic conditions. Ethanol concentration in wheat straw medium has increased from 0.55% v/ v after 24 hours to 2.46% v/v at pH 7. Gupta et al have also demonstrated that *B. licheniformis* demonstrated > 90% cellulase activity at pH 7 in case of water soluble carboxymethylcellulose (CMC) [29].

Substrate concentration in wheat straw medium plays an important role in improving the growth as well as production of enzymes in microbes. To optimize the substrate concentration, 2.5, 5.0, 7.5 and 10.0 mg/ml of substrates concentrations were prepared to perform SSF. Ethanol production increases from 2.91% to $4.29 \pm 0.27\%$ v/v (96 hours SSF) after increasing substrate concentration from 2.5 to 5.0 mg/ml as shown in Fig. 3A. However, further increase in substrate concentration to 7.5 and 10 mg/ml has decreased the ethanol production to $3.79 \pm 0.014\%$ and $3.27 \pm 0.007\%$ v/v, respectively. Glucose estimation at different substrate concentration demonstrates a decrease in glucose production when the

substrate concentration was increased from 5 mg/ml to 7.5 and 10.0 mg/ml as shown in Fig. 3B. The decrease in ethanol production could be due to catabolite repression that tends to lower the yield of glucose, thereby, lowering the ethanol production [30].

Apart from wheat straw, a nitrogen source was added in the form of ammonium nitrate to prepare wheat straw medium. Nitrogen source is essential to boost up the production of proteins/enzymes required for the intra-cellular processes during the growth and division of the micro-organisms. Ammonium nitrate was used as a nitrogen source for SSF as it has been observed to support higher cellulase synthesis by micro-organisms in comparison to other nitrogen sources [31]. For bioethanol production, nitrogen source was optimized by performing SSF at 0.03, 0.06, 0.09 and 0.12 M concentrations of ammonium nitrate in wheat straw medium. Bioethanol production increased from 4 ± 0.014 % to 5.67 ± 0.28 %, on increasing nitrogen source concentration from 0.03 M to 0.06 M as shown in Fig. 4A. Whereas further increase in nitrogen source concentration has declined the bioethanol production to 4.76 ± 0.34 % and 3.14 ± 0.23 % v/v for 0.09 M and 0.12 M, respectively. Decline in bioethanol production could be due to catabolite repression, which has decreased the cellulase production and have resulted in lower glucose production. Similar behaviour has been reported by Hernandez and coworkers while studying the effect of nitrogen source effect on ethanol production in *S. cerevisiae* [32].

Optimization of co-culture time, substrate concentration, pH of wheat straw medium and nitrogen source concentration demonstrated an increase in bioethanol production. For better control over the SSF process and the process parameters, the process was scaled up to 1.5 litre fermenter level to further improve the bioethanol production. SSF process was carried out by employing the optimized parameters in BIO-AGE 3 litres fermenter. Bioethanol produced was characterized through gas chromatography, which demonstrated a sharp increase in ethanol production from 5.67 % to 18.64 % v/v (Shown in figure S6). Studies by Mosier et al have demonstrated similar rise in ethanol production after scaling up o fermenter level [21].

4. Conclusion

Optimized co-culture conditions for *B. licheniformis* and *S. cerevisiae* were successfully established to produce bioethanol. The alkaline pre-treatment of wheat straw medium was observed to be a contributing factor along with other factors including pH, substrate concentration and nitrogen source concentration, towards enhancement in the production of bioethanol. The optimization of the given process parameters showed that the highest ethanol production by co-culturing of *B. licheniformis* and *S. cerevisiae* was achieved after 48 hours of co-culturing at a pH of 7, with nitrogen source and substrate concentration 0.06M and 5 mg/ml, respectively. Scale up of the SSF process to fermenter level has resulted in further enhancement of bioethanol production from 5.8% v/v to 18.64% v/v.

Declarations

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Data availability

The datasets generated during and/or analysed during the current study are available from the corresponding author on reasonable request.

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Figures

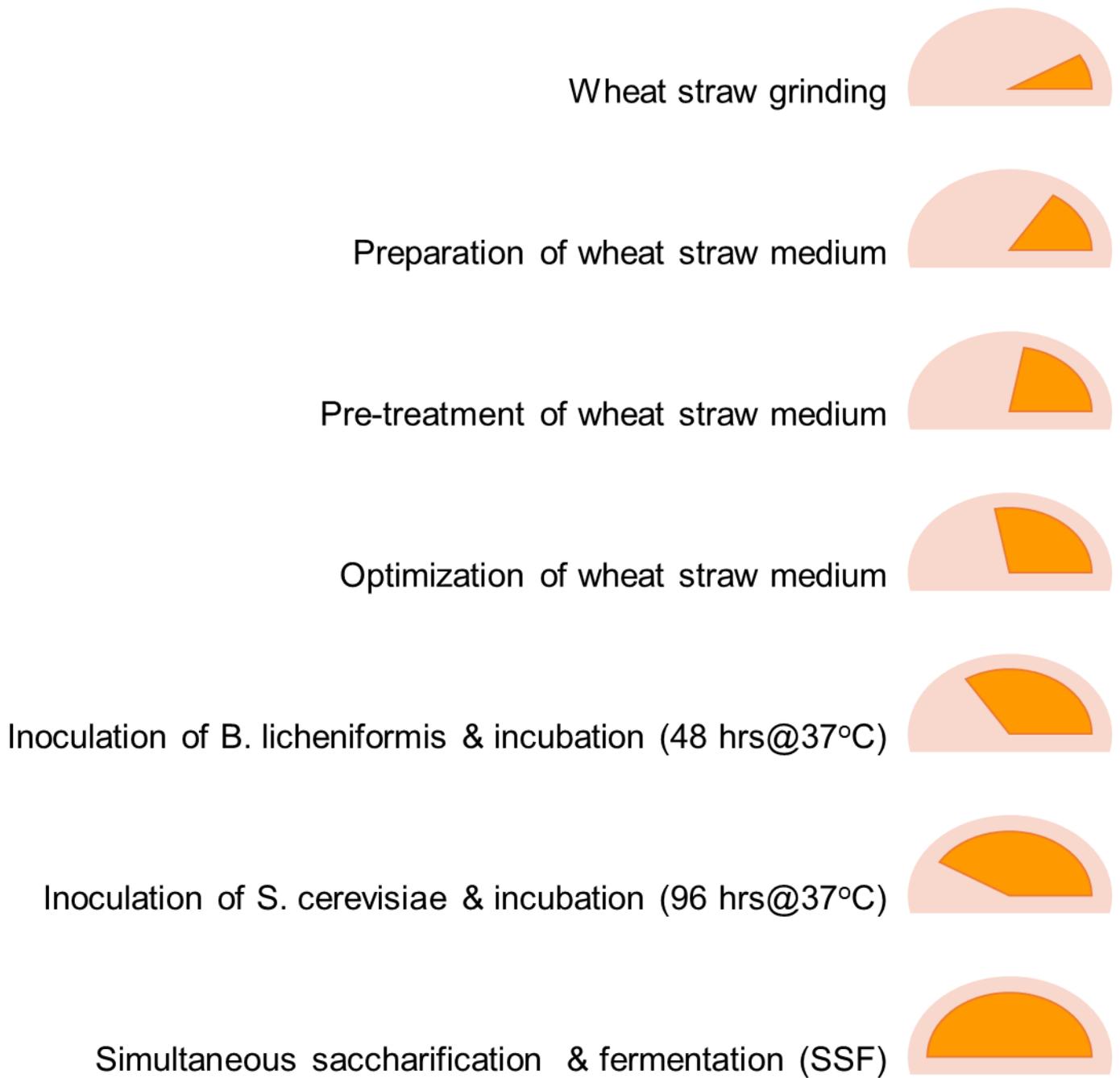


Figure 1

Schematics to demonstrate the Simultaneous saccharification and fermentation (SSF) for bioethanol production from wheat straw

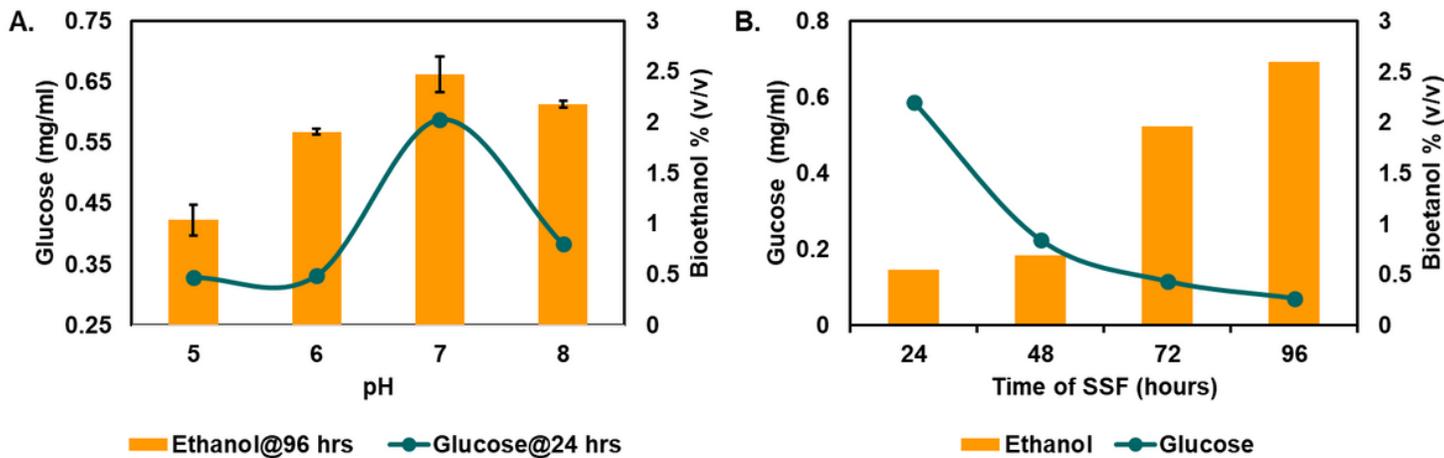


Figure 2

Optimization of bioethanol production by varying pH of the wheat straw medium (A) and in situ analysis of glucose and bioethanol produced during SSF at pH 7

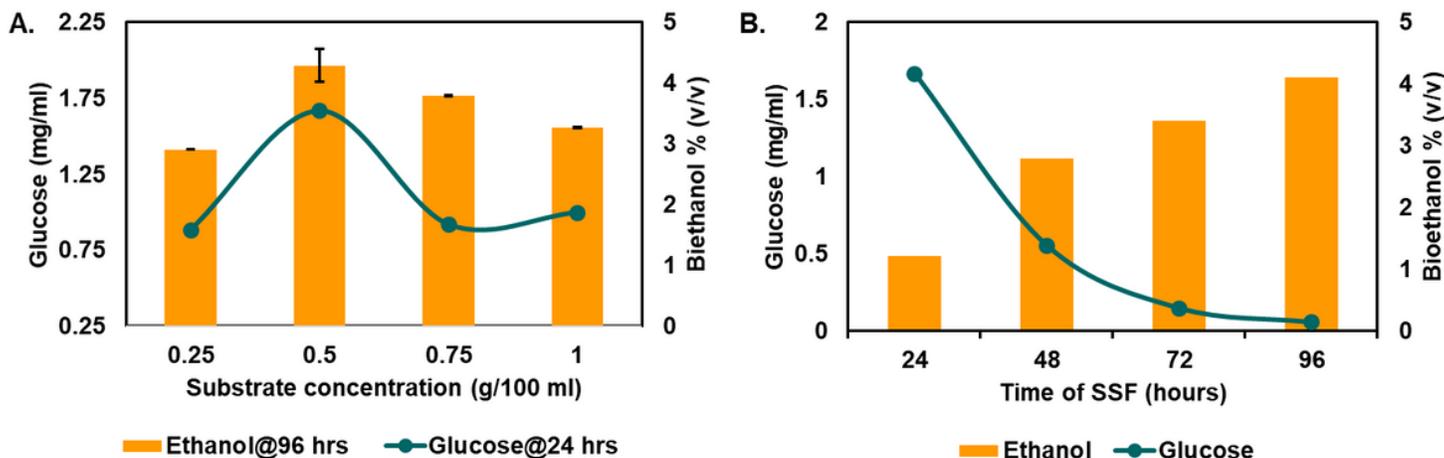


Figure 3

(A) optimization of substrate concentration in wheat straw medium for bioethanol production; (B) In situ analysis of glucose and bioethanol during SSF at 0.5 g/100 ml substrate concentration and pH 7.

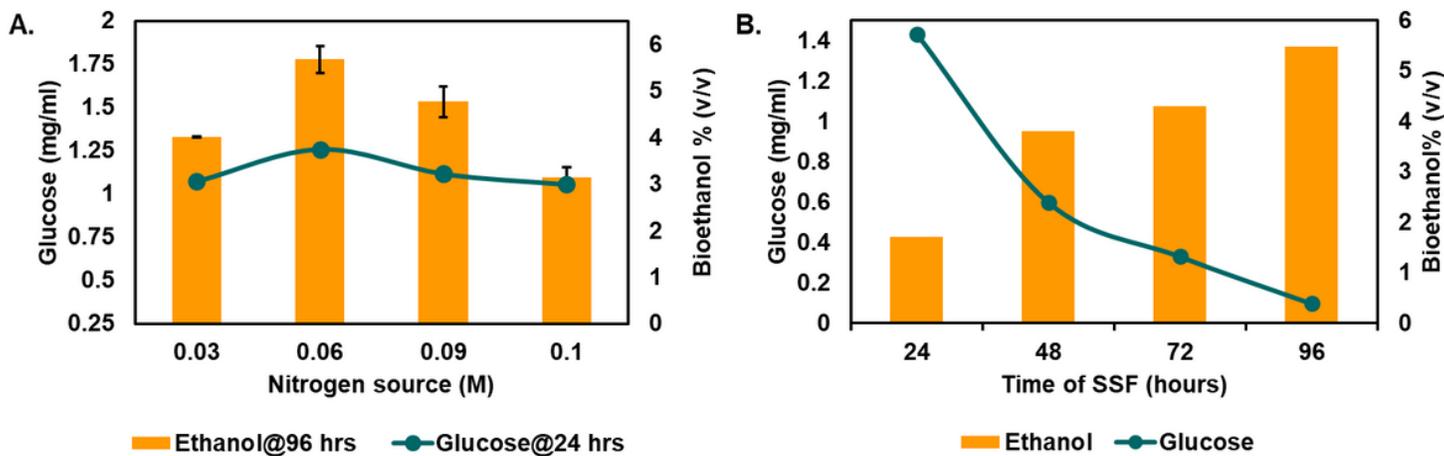


Figure 4

(A) Optimization of nitrogen source concentration in wheat straw medium for bioethanol production; (B) In situ analysis of glucose and bioethanol during SSF at pH 7, 0.06 M nitrogen source.

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