

A Comparative Study of Water Quality and Trophic State Contribution on Natural and Artificial Lakes Global Warming Potential

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1 **A comparative study of water quality and trophic state contribution on**
2 **natural and artificial lakes global warming potential**

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9 **Abstract**

10 Natural lakes and reservoirs are emitters of GHGs in the atmosphere
11 contributing to 31% of the annual CO₂ emissions of those from fossil fuel
12 combustion. Measurements of GHGs emissions in reservoirs demonstrate
13 that hydropower may actually not be as “green” as once thought. It is
14 estimated that emissions from reservoirs may be equivalent to 7% of the
15 global warming potential (GWP) of other documented anthropogenic
16 emissions. Aim of this work is to assess the impact of water quality
17 deterioration and the subsequent increment of biological productivity of a
18 waterbody on GHGs emissions. Therefore, the trophic state, the carbonic
19 GHGs emissions and the GWP of one natural lake domestic wastewater
20 receiver and two different age hydroelectric reservoirs, located in North
21 West Greece, were studied. Gross emissions of CO₂ and CH₄ were in-situ
22 measured using a static floating chamber and specific emissions as well as
23 GWP were calculated. Furthermore, water quality and trophic state were
24 evaluated based on the application of ΥΔΩΡ (hydōr) Water Quality Index
25 and Florida Trophic State Index using physicochemical characteristics

26 measurements. Data statistical interpretation revealed that CH₄ has strong
27 positive correlation with GWP, temperature, water quality and trophic state.
28 There is a seasonal variation of GWP that follows the seasonal variation of
29 CH₄ emissions induced by water temperature. Specific CH₄ emission rate
30 presents the most reliable indicator for assessing the impact of a waterbody
31 in terms of GWP, especially of a hypertrophic one. Water quality and trophic
32 state indices can be used for a rough comparison of GWP between
33 waterbodies with the same climatic conditions.

34 **Keywords:** Global warming potential, GHGs emissions; waterbodies, water
35 quality, water quality index, trophic state index.

36 **1. Introduction**

37 Anthropogenic activities, such as fossil fuel combustion, livestock,
38 deforestation and creation of artificial wetlands significantly increase the
39 concentration of greenhouse gases (GHGs) in the atmosphere and enhance
40 greenhouse effect ^{1,2}. Natural lakes, freshwater reservoirs and other surface
41 waters are of particular interest, since lentic ecosystems emit 110-810 Tg
42 CO₂-C and 69–112 Tg CH₄-C annually, constituting a great subject of
43 discuss in the heart of a worldwide debate ^{2,3,4}. Their contribution on annual
44 CO₂ emissions is estimated at approximately 31% of those from fossil fuel
45 combustion ⁵. Measurements of GHGs emissions in reservoirs demonstrate
46 that hydropower may actually not be as “green” as once thought. It is
47 estimated that emissions from reservoirs may be equivalent to 7% of the
48 global warming potential (GWP) of other documented anthropogenic
49 emissions ⁶.

50 GHGs emissions from aquatic systems are highly influenced by
51 waterbody's trophic state, which in turn is a function of nutrients availability,
52 usually phosphorous and less often nitrogen, as well as other parameters
53 such as seasonal variations, grazing of phytoplankton by zooplankton,
54 mixing depth of water, etc. ⁷⁻¹¹. According to European Environment Agency
55 ¹² the trophic status of a waterbody is classified into 5 categories,
56 Oligotrophic, Oligo/Mesotrophic, Mesotrophic, Meso/Eutrophic and
57 Eutrophic. The term Hypereutrophic has been also used for characterizing
58 highly eutrophic waterbodies including artificial lakes mainly the first years of
59 their formation ^{13,14}.

60 In freshwater reservoirs, carbon dioxide and methane are formed due
61 to organic matter imported from the catchment area or produced due to
62 plants and soils decomposition ¹⁴. More specifically, CO₂ is produced mainly
63 in oxic (aerobic) conditions and sometimes in anoxic conditions in the water
64 column, impounded soils and sediments of the reservoir, whereas is
65 consumed (carbon fixation) by aquatic primary producers in the euphotic
66 zone of a reservoir ^{15, 16}. GHGs emissions are highly influenced by organic
67 carbon in soils and topography. Reservoirs that flood peatlands emit more
68 greenhouse gases in the long term than reservoirs built in canyons where
69 little area is flooded, with limited peat deposits ⁶.

70 In general, carbon dioxide emissions increase from oligotrophic to
71 eutrophic lakes. Methane is produced under anaerobic conditions, primarily
72 in the sediments, a portion of which is oxidized to CO₂ by methanotrophic
73 bacteria in both water column and sediments ¹⁷. The fraction of CH₄, which

74 is oxidized before being emitted to the atmosphere, varies across freshwater
75 aquatic systems depending on oxygen levels and temperature ¹⁸.

76 Since nutrient availability is directly related to organic matter
77 decomposition, monitoring and assessing the trophic state (status) of an
78 aquatic system is essential to evaluate the environmental impact of
79 reservoirs. The trophic status of an aquatic system indicates its
80 environmental health and is expressed by a basic classification scale
81 showing rather its biological productivity than its water quality. Trophic
82 status can be calculated using Trophic State Indices (TSI), combining
83 quality parameters, usually water clarity, algal activity, phosphorus and
84 nitrogen availability ¹⁹.

85 On the other hand, water quality assessment of an aquatic system is
86 a more complicated procedure due to the numerous physicochemical and
87 biological parameters that may affect it ²⁰. The plethora of factors needed to
88 yield a single result of the overall water quality, such as (i) the large number
89 of data necessary for the qualitative evaluation, (ii) the special knowledge
90 and expertise required, as well as the (iii) difficulties arising in combining
91 qualitative parameters characterized by different significance and expressed
92 in different units of measurement and concentration ranges, make the
93 assessment process particularly difficult ²¹. A widely applied methodology
94 for conveying the different water quality parameters in one single expression
95 is the calculation of a Water Quality Index (WQI). A WQI is a number, a
96 scale, a word, a symbol or a color that expresses the water quality of an
97 aquatic system at a specific area in a specific period ^{22, 23}.

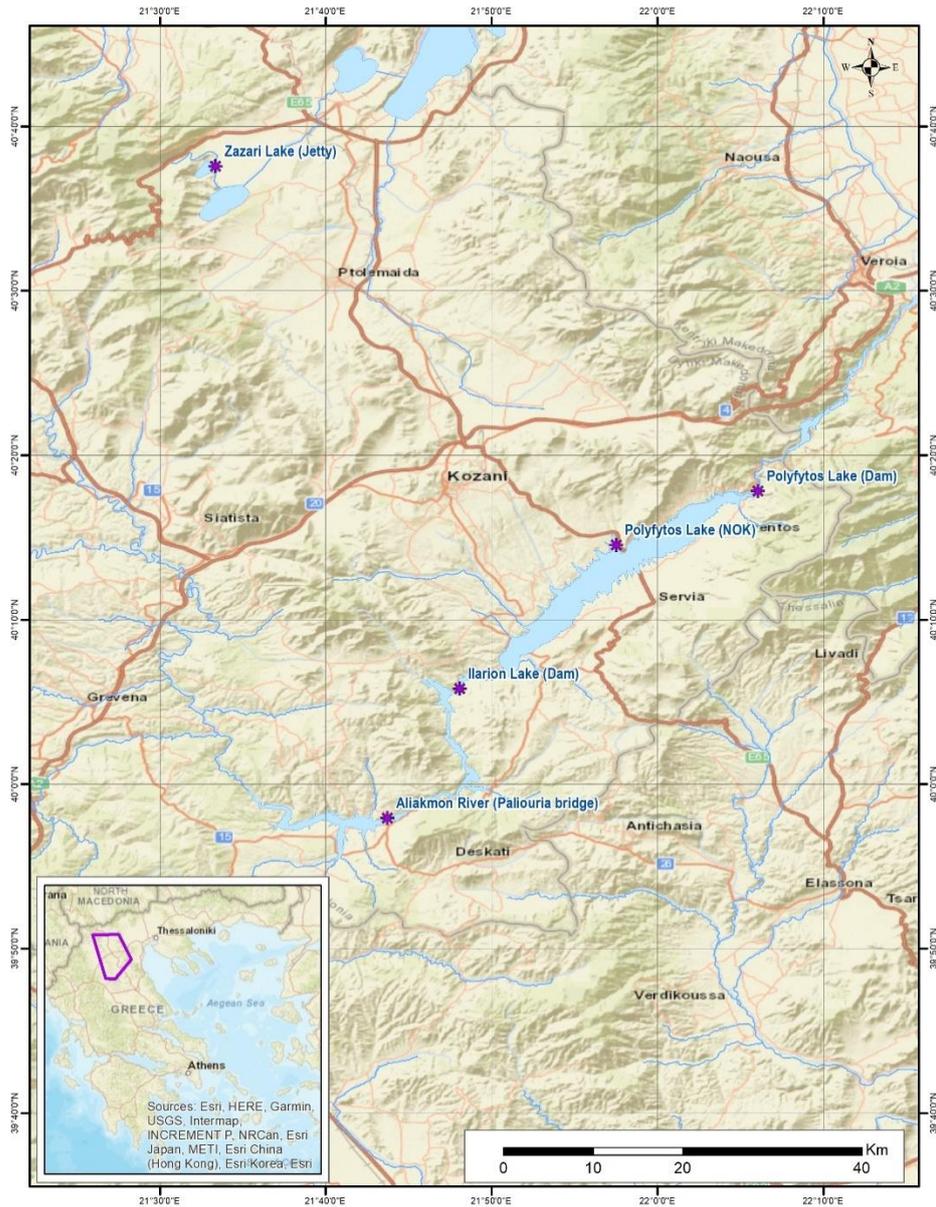
98 According to literature review, there are limited data regarding the
99 correlation of a waterbody's water quality and trophic state with GHGs
100 emissions and subsequently GWP. Aim of this work is to correlate water
101 quality and trophic state indices as well as GHGs emissions with GWP in
102 order to obtain indicators for easily assessing GWP of waterbodies. For this
103 purpose, one natural lake and two hydroelectric reservoirs were studied.
104 Physicochemical characteristics and gross emissions of CO₂ and CH₄ were
105 measured. Water quality and trophic state indices, as well as specific
106 emissions and GWP were calculated and statistically interpreted.

107 **2. Materials and methods**

108 The research methodology of this work includes the selection of sampling
109 stations and the measurement of water physicochemical parameters, CO₂
110 and CH₄ emissions in the collected samples. Thereafter, calculation of
111 specific emission rates and GWP was carried out, whereas water quality
112 and trophic state indices were also calculated. The obtained data were
113 statistically processed using SPSS software and interpreted accordingly.

114 **2.1. Study area**

115 A natural lake (Zazari Lake) and two hydroelectric reservoirs (Ilarion and
116 Polyfytos Reservoirs) were selected for the comparative assessment of
117 GHGs emissions, water quality and trophic state. All three waterbodies are
118 located in the area of North-West Greece presenting the same climatic
119 conditions. (Figure 1).



120
121

Figure 1. Study area map and sampling stations.

122
123
124
125
126

Zazari Lake is located in a relatively small flat area approximately 60 km North-West of Ilarion and Polyfytos Reservoirs, covering a surface of approximately 1.7 km², with a maximum depth of approximately 6 m and is used mainly for irrigation and fishing. Five years ago, Zazari Lake was the final receiver of domestic wastewater from neighboring villages.

127
128

Ilarion and Polyfytos Reservoirs were created in 2012 and 1975 respectively, for the production of hydroelectric energy. Both reservoirs are

129 fed with water from Aliakmon River, which is the longest river in Greece
130 (approximately 300 km).

131 Ilarion Reservoir is a relatively young and small reservoir downstream
132 Aliakmon River, created in stiff rocky hills, forming a deeper canyon-like lake
133 (terrain slopes between 15° to 45°). It covers a total surface of 22 km² with a
134 maximum depth of 65 m. Prior to its impoundment, 23.1% was river, 53.03%
135 was non-forest soil (grassland, rocks, and agricultural land), and 23.96%
136 was forest (plane and oak trees) ²⁴. Limited agricultural and livestock
137 activities and absence of other anthropogenic pressures characterize the
138 area.

139 Polyfytos Reservoir is significantly older and larger than Ilarion,
140 having a surface of 73 km² with a maximum measured depth of 22 m. It was
141 created in a rather flat agricultural area where Neraida village was located.
142 Agricultural and livestock activities characterize the area around the
143 reservoir prior and post their impoundment ^{24, 25}.

144 **2.2. Sampling and measurements**

145 Five sampling stations were selected (Figure 1), one in Zazari Lake, two in
146 Polyfytos Reservoir, one in Ilarion Reservoir and one in Aliakmon River
147 (sampling station Paliouria Bridge, western end of Ilarion Reservoir). The
148 maximum measured depth of the sampling stations varied from 4.5 m to 6 m
149 for Zazari Lake, 12 m to 14 m for Polyfytos NOK, 10 m to 12 m for Polyfytos
150 Dam and 5 m to 7 m for Ilarion Dam. The measured depth of Aliakmonas
151 River at the sampling station Paliouria Bridge ranged from 0.3 m to 1 m. All
152 samplings were carried out from January 2018 to February 2019; samples
153 were collected from surface (0.3 m) and maximum depth in every sampling

154 station. Table 1 summarizes all the necessary details of the sampling plan
 155 performed in North-West Greece.

156 **Table 1.** Sampling stations, sampling points, number of samples and type of
 157 performed measurements.

Sampling station	Sampling point	Number of samples	Type of Measurements
Ilarion Reservoir	Ilarion Dam	10	Physicochemical measurements in water and in situ gross CO ₂ and CH ₄ emission measurements
Polyfytos Reservoir	Polyfytos Dam	10	Physicochemical measurements in water and in situ gross CO ₂ and CH ₄ emission measurements
	Nautical Club of Kozani (Polyfytos NOK)	10	
Zazari Lake	Jetty (South-West coast)	8	Physicochemical measurements in water and in situ gross CO ₂ and CH ₄ emission measurements
Paliouria Bridge	Aliakmon River	6	Physicochemical measurements in water

158 A sub-surface water sampler was used for the collection of water
 159 samples. Water temperature and dissolved oxygen were measured in situ
 160 with a properly calibrated optical dissolved oxygen meter (Hach HQ30d).

161 All water samples were analyzed within a few hours from sampling by
 162 the accredited, according to ISO 17025, Environmental Chemistry & Water
 163 and Wastewater Treatment Laboratory, of the University of Western
 164 Macedonia, Greece. The parameters of pH, Conductivity, Color, Turbidity,
 165 Ammonium ions (NH₄⁺), Nitrite ions (NO₂⁻), Total suspended solids (TSS),
 166 Hexavalent Chromium (Cr⁺⁶), Total organic Carbon (TOC), Total Kjeldahl
 167 Nitrogen (TKN), Total Phosphorus (TP), Hardness, Alkalinity and
 168 Chlorophyll- α were analyzed according to standard methods ²⁶. The
 169 parameters of Cl⁻, NO₃⁻, SO₄⁻², K⁺, Na⁺, Ca⁺² and Mg⁺² were analyzed by Ion

170 Chromatography (IC), according to ISO 10304-1 (2007) and ISO14911
171 (1998) ^{27, 28}, using a Metrohm Chromatographer (model 881 Compact IC
172 pro), equipped with an electrical conductivity detector and two Metrosep
173 columns (models A SUPP 5 250 and C 2 250/4).

174 For the collection of GHGs emissions, specifically CO₂ and CH₄, a
175 static floating chamber was used. The static floating chamber was a close-
176 ended, stainless steel, rectangular box, 0.50 m in height, 0.50 m in width,
177 and 1 m in length, equipped with a floating buoy, constructed using a 0.20 m
178 Ø rubber tube, filled with polyurethane foam. Gas samples were collected
179 after 0 min, 50 min, 90 min, 130 min, and 180 min in low permeability, multi-
180 layered foil bags (Supel™ - Inert Multi-Layer Foil Bag), suitable for gas
181 sample collection. Gas chromatography for the determination of CO₂ and
182 CH₄ concentrations was performed using a SHIMADZU 14B, equipped with
183 thermal conductivity (TCD) and flame ionization (FID) detectors as well as
184 with a Molecular Sieve 13× (10 ft. × 1/8 in.) and a Porapack QS (10 ft. × 1/8
185 in.) column for gas separation. Both columns were heated at 80°C.

186 The specific GHGs emission rate was calculated according to
187 Equation (1), using CO₂ and CH₄ concentration measurements with the
188 static floating chamber previously described.

$$189 \quad C = \rho \times \frac{dC}{dt} \times \frac{273}{273+T} \times H \quad (1)$$

190 where: C is the specific gas emission rate (mg m⁻² h⁻¹); ρ is the gas density
191 in standard conditions (kg m⁻³); dC/dt is the linear regression of gas
192 concentration in static floating chamber; H is the height (m) from the
193 chamber top to the water surface; T is the air temperature (°C) inside the
194 chamber ²⁹.

195 **2.3. Characterization of water quality**

196 Water quality classification was carried out using a new WQI developed at
197 the department of Chemical Engineering - University of Western Macedonia,
198 named *ΥΔΩΡWQI* (hydōr = water in ancient Greek). *ΥΔΩΡWQI* is
199 considered to be a sensitive and flexible tool for assessing water quality, as
200 it (a) prioritizes monitoring parameters, (b) redefines weight coefficients
201 based on experts' opinion, (c) inserts in its calculation the ideal values, (d)
202 takes into consideration both legislation and experts' opinion and (e) is
203 sensitive to marginal conditions, in contrast to other water quality indices ³⁰.
204 ³¹. *ΥΔΩΡWQI* is calculated by Equation 2.

205
$$ΥΔΩΡWQI = \frac{\sum_{i=1}^n [Average(q_i \times RW_i)] + \sum_{i=1}^n [R_{i,e} \times Average(q_{i,e} \times RW_i)]}{\sum_{i=1}^n RW_i} \quad (2)$$

206 where: q_i is the Sub-index of sample for i parameter; $q_{i,e}$ is the Sub-index of
207 sample for i parameter exceeding permitted value; RW_i is the relative weight
208 of i parameter (obtained from Mavromatidou et al., 2020)³⁰; $R_{i,e}$ is the ratio
209 of samples exceeding permitted value of parameter i to total number of
210 samples of parameter i ; n is the number of control parameters

211 The sub-index is calculated according to Equation (3), which is taking
212 under account the measured physicochemical and microbiological water
213 quality parameter values, the standard values deriving from legislation limits
214 and the desired or ideal values.

215
$$q_i = \frac{100 \times |C_i - V_{io}|}{|S_i - V_{io}|} \quad (3)$$

216 where: C_i is the measured value of the i water quality parameter; S_i is the
217 permitted i water quality parameter value obtained from legislation; V_{io} is
218 the ideal value of i water quality parameter.

219 The parameters included in the calculation of $Y\Delta\Omega PWQI$ were:
 220 Hexavalent Chromium, *E-coli*, Lead, Arsenic, Cadmium, Fecal Coliforms,
 221 Total Pesticides, Mercury, Ammonium, COD, Nitrite Ions, Nitrate Ions, Total
 222 Phosphorus (TP) and Chlorides (Total Pesticides, Lead, Arsenic, Cadmium
 223 and Mercury measurement data were provided from the Greek Ministry of
 224 Rural Development and Food, (2020 a,b)^{32, 33}). The permitted values of
 225 control parameters were obtained from the limit values of European
 226 directives and national legislation regarding quality of internal surface,
 227 ground and drinking waters. For ideal values, the Lower Detection Limit
 228 (LDL) of each parameter was used. The classification of water quality
 229 according to $Y\Delta\Omega PWQI$ is presented in Table 2.

230 **Table 2.** Characterization of lakes according to $Y\Delta\Omega PWQI$ results
 231 (Mavromatidou et al., 2020).

$Y\Delta\Omega PWQI$ scale	Water Quality Characterization (Based on physicochemical and microbiological characteristics)
$0 \leq Y\Delta\Omega PWQI \leq 10$	Excellent
$10 < Y\Delta\Omega PWQI \leq 25$	Good
$25 < Y\Delta\Omega PWQI \leq 50$	Fair
$50 < Y\Delta\Omega PWQI \leq 100$	Marginal
$Y\Delta\Omega PWQI > 100$	Poor

232 **2.4. Characterization of trophic state**

233 Trophic state characterization of every waterbody was based on the
 234 calculation of a TSI, more specifically, the Florida TSI (*FTSI*)^{13, 34}. *FTSI* is
 235 based on the same rationale as that of Carlson (1977)³⁵, who introduced the
 236 concept of classifying the trophic state of a waterbody using measurement
 237 data regarding Turbidity (Secchi depth), Chlorophyll concentration and TP
 238 concentration. *FTSI* excluded turbidity, while included the parameter of Total
 239 Nitrogen (TN) as a third indicator. Turbidity was excluded because it was

240 causing problems in dark-water lakes and estuaries, where dark-waters
241 rather than algae diminish transparency ³⁴.

242 In order to apply *FTSI*, each waterbody's limiting nutrient, must be
243 determined, which is usually phosphorous and less often nitrogen ¹³. The
244 limiting nutrient is determined based on the ratio of TN concentration to TP
245 concentration. For $TN/TP < 10$ the limiting nutrient is Nitrogen, for $TN/TP >$
246 30 is Phosphorus, while for $10 \leq TN/TP \leq 30$ there is no limiting nutrient
247 (nutrient balanced waterbody) ³⁶. Regarding *FTSI* calculation in Polyfytos
248 reservoir, mean values of TP, TN and Chlorophyll α from the two sampling
249 stations were used.

250 For the determination of the overall trophic state index using *FTSI*, a
251 separate component (sub-index) for each parameter was calculated,
252 depending on the limiting nutrient, using the appropriate formula (Equation 4
253 to Equation 8), followed by the appropriate combination of the sub-indexes,
254 as indicated in Equation A to Equation C. The equation for the calculation of
255 Chlorophyll *a* sub-index ($CHLA_{TSI}$) is the same (Equation 4) regardless the
256 limiting nutrient. On the contrary, the nutrient status of a lake dictates the
257 empirical formula that is used for Nitrogen and for Phosphorus sub-indices
258 calculation. For nutrient-balanced lakes ($10 \leq TN/TP \leq 30$) Equation 5 and
259 Equation 6 are used; for phosphorus-limited lakes ($TN/TP > 30$), Equation 7
260 is used, whereas for nitrogen-limited lakes, Equation 8 is used

261 *FTSI sub-index formulas:*

262 $CHLA_{TSI} = 16.8 + [14.4 \times \ln(CHLA)]$ (4)

263 $TP_{TSI} = [18.6 \times \ln(TP \times 1000)] - 18.4$ (5)

264 $TN_{TSI} = 56 + [19.8 \times \ln(TN)]$ (6)

265 $TP2_{TS} = 10 \times [2.36 \times \ln(TP \times 1000) - 2.38]$ (7)

266 $TN2_{TSI} = 10 \times [5.96 + 2.15 \times \ln(TN + 0.0001)]$ (8)

267 *FTSI overall index formulas:*

268 $TSI = \frac{CHLA_{TSI} + \frac{TN_{TSI} + TP_{TSI}}{2}}{2}$ (A)

269 $TSI = \frac{CHLA_{TSI} + TP2_{TSI}}{2}$ (B)

270 $TSI = \frac{CHLA_{TSI} + TN2_{TSI}}{2}$ (C)

271 Based on the results of overall *FTSI*, the waterbodies are classified
 272 according to their trophic state as “Oligotrophic”, “Mid-Eutrophic”,
 273 “Eutrophic” and “Hypereutrophic” and corresponds to “Good”, “Fair” and
 274 “Poor” water quality respectively (Table 3). A "Good" lake water quality is
 275 one that meets all lake-use criteria (swimmable, fishable and supports
 276 healthy habitat); a “Fair” lake water quality can be considered highly
 277 productive and a reasonable lake for fishing and most water sports, while a
 278 “Poor” lake water quality means that probably the lake use criteria are not
 279 meet ¹³.

280 **Table 3.** Classification and characterization of lakes according to *FTSI* results
 281 ¹³.

<i>FTSI</i> scale	Trophic State Classification	Water Quality Characterization (Based on trophic status)
0 to 59	Oligotrophic through Mid-Eutrophic	Good
60 to 69	Mid-Eutrophic through Eutrophic	Fair
70 to 100	Hypereutrophic	Poor

282 **2.5. Global Warming Potential Calculation**

283 The Global Warming Potential was developed to allow comparisons of the
284 global warming impacts of different gases. Each greenhouse gas has different
285 GWP. CO₂, by definition, has a GWP of 1 regardless of the time period used,
286 because it is the gas being used as the reference. CH₄ is considered to have
287 20 times higher GWP than that of CO₂ on over a 100 to 150-year period ³⁷.

288 The calculation of GWP in relation to season and for a monthly step,
289 was carried out using the average values of CO₂ and CH₄ specific emissions
290 for the year 2018, as well as from 2014 to 2016 as described in our previous
291 work ²⁴.

292 **2.6. Data evaluation and statistical analysis**

293 The samples' analysis results were statistically processed and interpreted
294 using the SPSS statistical software. For testing the significant differences
295 among the values of the measured parameters corresponding to depth and
296 surface samples, the independent variables *t*-test was used. A significance
297 level of 0.05 or 95% confidence interval was set. For the evaluation of
298 correlation between measured and calculated parameters, the Spearman's
299 rank correlation hypothesis test was applied setting a significance level of
300 0.05.

301 **3. Results and Discussion**

302 **3.1. Assessment of significant differences between sampling stations**

303 Independent variable *t*-test between the measured physicochemical
304 parameters of surface and maximum depth samples at all sampling stations
305 (comparative data from the first three samplings) showed that:

306 - Polyfytos and Ilarion Reservoir presented no significant statistical
307 difference (Sig. > 0.05) for none of the twenty-two examined parameters.

308 This may be attributed to the low hydraulic residence time due to water
309 discharges as well as the continuous mixing of the waterbody's layers
310 from dam ^{25, 38}.

311 - In Paliouria Bridge, due to its low depth, no stratification occurred and
312 hence no significant differences in water quality were observed; none of
313 the twenty-two examined parameters presented significant statistical
314 differences between surface and depth samples. Paliouria Bridge
315 sampling station in Aliakmon River is a river to lake transition area that
316 inundated last and has significant changes in depth due to dam operation
317 and Aliakmon River flow ²⁵.

318 - In Zazari Lake, only chlorophyll *a*, exhibited significant statistical difference
319 between surface and depth samples (*t*-test, Sig.= 0.001). This may be
320 attributed to water stratification of the lake, due to phototrophic organisms'
321 growth. The relatively high turbidity and color in Zazari Lake as well as its
322 relatively low mixing of water, indicative of natural lakes, may cause higher
323 growth rate of phototrophic organisms at the surface of the lake in relation
324 to those present in the deeper levels, where the transfer of
325 photosynthetically active radiation (PAR) is significantly obstructed ^{5, 39}.
326 Leach et al. (2018) found that the variation in chlorophyll *α* concentrations
327 with depth in lake water is influenced more by light attenuation than thermal
328 stratification ⁴⁰. Reservoirs differ from natural lakes in terms of hydraulic
329 residence time (HRT), loads of total suspended solids, and productivity ³⁸,
330 ⁴¹. Moreover, hydrological variability induced by anthropogenic
331 manipulation of inflow/outflow can affect the variation with depth in
332 constituents in lakes and reservoirs.

333 Due to the non-significant statistical differences between surface and
 334 depth regarding the examined parameters in all sampling stations, as well
 335 as due to the fact that the differences in chlorophyll *a* measurements
 336 between surface and depth in Zazari Lake did not affect either the
 337 characterization in terms of water quality and trophic state or the results of
 338 statistical correlations, all subsequent water samples were collected only
 339 from the maximum depth.

340 Independent variable *t*-test between the measurements of
 341 physicochemical parameters at the two sampling stations of Polyfytos
 342 Reservoir (Figure 1, Polyfytos Dam and Polyfytos NOK) showed that there
 343 are non-significant differences (Sig > 0.05) regarding 20 of the 22 examined
 344 parameters. According to the results of the statistical analysis only nitrates
 345 and turbidity presented significant statistical differences (Sig. = 0.002 and
 346 0.029 respectively). Nevertheless, these differences in nitrates and turbidity
 347 measurements between the two sampling stations of Polyfytos Reservoir did
 348 not affect either the characterization in terms of water quality and trophic
 349 state or the results of statistical correlations. Thus, both Polyfytos sampling
 350 stations were accounted as one for the calculation of *ΥΔΩPWQI* and *FTSI*,
 351 as well as for the statistical analysis between all sampling stations.

352 In Table 4, the average values of each physicochemical parameter in
 353 every sampling period along with their standard deviation are presented.

354 **Table 4.** Average values and standard deviation of physicochemical
 355 parameters at maximum depth of each sampling station

Parameter	Zazari Lake	Std. Dev.	Ilarion Reservoir	Std. Dev.	Polyfytos Reservoir	Std. Dev.	Paliouria Bridge	Std. Dev.
pH	8.17	0.61	8.13	0.31	8.17	0.34	8.22	0.18
D.O., % saturation	118	6.20	82	10.41	102	20.67	100	15.34

Conductivity, $\mu\text{S/cm}$	175.66	11.37	388.25	20.89	405.17	29.83	402.05	56.57
Color, Hazen	8.62	5.08	10.34	4.46	6.99	3.72	24.05	9.12
Turbidity, NTU	5.51	8.70	1.77	1.04	1.18	1.42	75.30	57.70
TSS, mg/l	11.60	1.27	2.63	1.80	3.1	1.87	22.20	17.07
NO_3^- , mg/l	0.53	0.39	1.18	0.99	1.11	0.86	1.90	0.78
NO_2^- , $\mu\text{g/l}$	7.32	5.73	23.78	14.36	12.48	11.31	52.75	21.05
NH_4^+ , mg/l	0.11	0.21	0.11	0.22	0.02	0.03	0.03	0.03
Alkalinity, mg/l	92.96	80.44	179.16	15.66	205.60	20.53	204.20	23.05
Hardness, mg/l	63.21	15.06	213.75	21.05	216.95	65.50	148.80	46.67
Cl^- , mg/l	4.09	1.31	5.47	0.88	6.56	2.70	6.40	1.58
TOC, mg/l	4.49	6.34	5.19	3.90	3.73	3.28	5.75	1.34
TKN, mgN/l	4.65	2.62	1.40	0.61	1.78	0.90	1.89	0.30
Ca^{+2} , mg/l	15.60	0.71	34.80	3.41	38.61	5.36	22.75	12.87
Chlorophyll <i>a</i> , $\mu\text{g/l}$	18.89	1.70	1.40	0.76	2.55	1.98	0.52	0.63
Cr^{+6} , $\mu\text{g/l}$	6.15	1.49	2.43	1.01	0.76	1.40	4.70	7.12
TP, mgP/l	1.29	1.41	0.007	0.004	0.45	0.49	0.083	0.055
Na^+ , mg/l	7.98	0.60	5.46	2.08	5.88	1.12	5.81	6.14
K^+ , mg/l	3.46	0.25	1.57	0.22	1.65	0.06	1.40	1.28
Mg^{+2} , mg/l	4.45	0.30	25.80	0.35	22.74	1.07	22.20	1.69
SO_4^{-2} , mg/l	8.50	1.20	22.34	5.42	19.16	4.15	18.3	3.34

356 The high standard deviation values in various parameters of Table 4
357 are indicative of changes of water quality between samplings. Nevertheless,
358 the high differences in measured values did not hinder the results regarding
359 the characterization of the waterbodies in terms of water quality and trophic
360 state, since the application of *YΔΩPWQI* and *FTSI* using average parameter
361 values plus their standard deviation presented similar results and identified
362 the differences between the three waterbodies.

363 As it can be seen in Table 4, all sampling stations of Aliakmon River
364 basin except Zazari Lake present similar water quality characteristics.
365 Statistical analysis of all physicochemical parameters between all sampling
366 stations showed significant differences (*t*-test, $p < 0.05$) only in Zazari Lake.

367 The similar water quality of sampling stations at Aliakmon River basin is
 368 attributed to their common feed. More specifically, water of Paliouria Bridge
 369 supplies Ilarion Reservoir, which thereafter replenishes Polyfytos Dam.
 370 Higher values of Color, Turbidity and TSS were only reported in Paliouria
 371 Bridge sampling station, due to the shallow depth (0.3- 1 m) and the flow of
 372 Aliakmon River.

373 **3.2. Assessment of water quality, trophic state and carbonic emissions**

374 The waterbodies quality and trophic state classification according to
 375 *YΔΩPWQI* and *FTSI* respectively, as well as CO₂ and CH₄ specific
 376 emissions are presented in Table 5. *YΔΩPWQI* and *FTSI* were calculated as
 377 described in Materials and Methods, sections 2.3. and 2.4., based on the
 378 average measured values (Table 4). The range of CO₂ and CH₄ specific
 379 emissions as well as the average specific emissions were calculated as
 380 described in Materials and Methods, section 2.2., based on measured
 381 emission rates using a static floating chamber.

382 **Table 5.** Measurement data; classification / characterization of waterbodies
 383 based on *FTSI* and *YΔΩPWQI* calculation.

Parameter	Ilarionas Reservoir	Polyfytos Reservoir	Zazari Lake
TN/TP ratio	137	4.07	3.71
Nutrient status	Phosphorus-Limited Lake	Nitrogen-Limited Lake	Nitrogen-Limited Lake
<i>FTSI</i>	20.1	51.4	76.3
<i>FTSI</i> classification	Oligotrophic through Mid-Eutrophic	Oligotrophic through Mid-Eutrophic	Hypereutrophic
<i>FTSI</i> characterization	GOOD	GOOD	POOR

<i>ΥΔΩΡWQI</i>	21.4	45.2	112.5
<i>ΥΔΩΡWQI</i> characterization	GOOD	FAIR	POOR
Specific CO ₂ emission rate (mg m ⁻² d ⁻¹)	65.8 to 1144.8	98.0 to 761.4	-731.9 to 2000.7
Average specific CO ₂ emission rate (mg m ⁻² d ⁻¹)	389.8	391.9	309.7
Specific CH ₄ emission rate (mg m ⁻² d ⁻¹)	0 to 19	0 to 43.2	0 to 2236.9
Average specific CH ₄ emission rate (mg m ⁻² d ⁻¹)	4.6	13.3	673.4

384 The water quality of the older and considered stabilized Polyfytos
385 Reservoir and the younger Ilarion Reservoir is characterized as “Fair” and
386 “Good” respectively, according to *ΥΔΩΡWQI* (Table 5). On the contrary, in
387 our previous work (2012-2015 monitoring period) and during Ilarion
388 Reservoir maturation period, Polyfytos Reservoir was characterized by
389 better water quality than Ilarion ²⁵. Taking into consideration that Ilarion
390 Reservoir has formed in a rocky area (canyon), this improvement in water
391 quality suggests that it has entered in stabilization period approximately six
392 years after its creation, earlier than the ten years period suggested by other
393 studies ^{6, 42}. This highlights the importance of site selection in the
394 construction of water reservoirs for hydroelectric energy production, since
395 both specific morphology and absence of anthropogenic activities can lead
396 to a good water quality only a few years after inundation of an area.

397 According to the results of *FTSI*, Polyfytos and Ilarion Reservoirs are
398 classified as oligotrophic through mid-eutrophic waterbodies, with Ilarion
399 Reservoir presenting better trophic state than Polyfytos (Table 5). This is in

400 accordance with the previous results of *ΥΔΩΡWQI*, where young Ilarion
401 Reservoir exhibited better water quality indicative of its stabilization.

402 The respective average specific CO₂ and CH₄ emissions in Polyfytos
403 Reservoir are 391.9 mg m⁻² d⁻¹ and 13.3 mg m⁻² d⁻¹ and in Ilarion Reservoir
404 389.8 mg m⁻² d⁻¹ and 4.6 mg m⁻² d⁻¹. The emissions of CO₂ and CH₄ were
405 relatively low in both Polyfytos and Ilarion Reservoirs, regardless their age
406 difference, presenting non-significant statistical (*t*-test) difference in terms of
407 specific CO₂ emission rate (Sig. = 0.499) and specific CH₄ emission rate
408 (Sig. = 0.985). Nevertheless, Polyfytos Reservoir, exhibited three times
409 higher productivity in terms of specific CH₄ emission rate when compared to
410 Ilarion emissions. This is attributed to the relatively flat area with fertile soils
411 that was impounded and to agricultural and livestock activities that increase
412 inflow of organic matter in the reservoir contributing to higher GHGs
413 emissions ⁴³. GHGs specific emissions from these two in-line reservoirs
414 were also calculated in our previous work ²⁴ for collected data of the years
415 2014-2016. During this period, it was found that the average specific CO₂
416 emission rate was 563.3 and 496.3 mg m⁻² d⁻¹, whereas the average specific
417 CH₄ emission rate was 38.5 and 28.9 mg m⁻² d⁻¹ in the case of Polyfytos and
418 Ilarion respectively. Two years after the 2014-2016 monitoring period, the
419 lower GHGs emissions measured in Ilarion Reservoir (see Table 5), as well
420 as the limited biomass impoundment (canyon-like rocky area) indicate that
421 this reservoir has reached its stabilization period, which is in accordance
422 with the results obtained from the comparison of water quality indices
423 between the two reservoirs. The fact that Ilarion Reservoir has entered its
424 stabilization period only six years after its formation renders the selection of

425 the reservoir location a crucial parameter on the assessment of how green
426 is an investment on hydroelectric energy.

427 In Zazari Lake the respective average specific CO₂ and CH₄
428 emissions are 309.7 mg m⁻² d⁻¹ and 673.4 mg m⁻² d⁻¹. Zazari natural lake (a)
429 exhibits “Poor” water quality according to *ΥΔΩΡWQI*, reflecting the
430 degradation of organics due to domestic wastewater disposal (five years
431 ago) from nearby villages (b) is characterized as hypereutrophic lake and (c)
432 exhibits significantly higher GHGs emissions, especially regarding CH₄
433 emissions comparing with Ilarionas and Polyphytos Reservoirs (Sig. = 0.043
434 and Sig. = 0.033 respectively). It is worth mentioning that, despite the
435 relatively low depth at Zazari Lake sampling station (4.5 m to 6 m)
436 methanogenesis occurred at significantly high rates (mean value 673.4 mg
437 m⁻² d⁻¹). This is attributed to the hypertrophic state of Zazari Lake, which
438 results in the creation of anoxic conditions, enhancing anaerobic biological
439 processes ⁴⁴. Davidson et al. (2015) also found that nutrients concentration
440 in a shallow eutrophic lake, such as Zazari Lake, have a profound effect in
441 GHGs emission ⁴⁵. Domestic pollution of Zazari Lake ceased five years ago.
442 Considering the significantly higher CH₄ emissions of Zazari Lake, as well
443 as the deteriorated water quality and trophic state compared to that of
444 Polyfytos and Ilarion Reservoirs, it is concluded that Zazari Lake can be
445 characterized as a young artificial reservoir that has not yet been stabilized.

446 According to the results of this study, Ilarion Reservoir has entered its
447 stabilization period (lower emissions) six years after its formation, while
448 Zazari Lake five years after domestic pollution ceased still emits significant

449 quantities of GHGs. Consequently, both morphology and anthropogenic
450 activities affect the stabilization time of a waterbody.

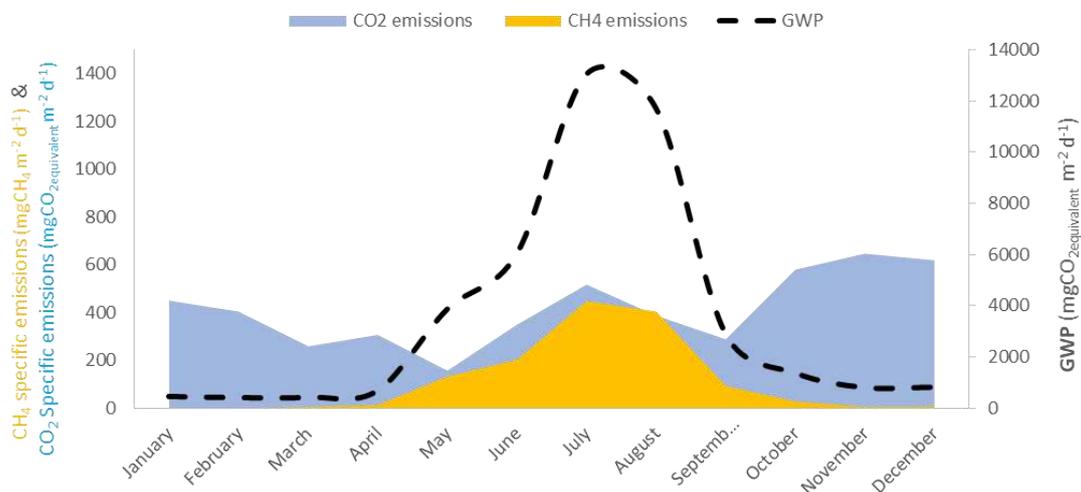
451 According to Spearman statistical test, temperature presents strong
452 positive correlation with CH₄ specific emissions (Corr. Coef. = 0.727, Sig. <
453 0.001) and GWP (Corr. Coef. = 0.823, Sig. < 0.001), indicative of the positive
454 effect of temperature on methanogenesis.

455 On the other hand, CO₂ specific emissions present significant negative
456 correlation with turbidity (Corr. Coef. = 0.728, Sig. < 0.001), a parameter that
457 is indicative of the presence of microorganisms in a waterbody. This negative
458 correlation as well as the absence of a similar strong negative correlation
459 between CO₂ emissions and chlorophyll α shows that the presence of
460 autotrophic microorganisms play a major role in carbon fixation by
461 assimilating CO₂ ⁴⁶. The growth of chemolithotrophic bacteria that can fixate
462 CO₂ or/and oxidise CH₄ could be triggered by the increased availability of
463 specific nitrogenous nutrients, as indicated by the strong positive correlation
464 (Sig. < 0.03) between turbidity and ammonium, nitrites and nitrates.
465 Furthermore, the strong negative correlation of sulphates with the emission
466 rates of CO₂ (Corr. Coef. = -0.964, Sig. = 0.001) and CH₄ (Corr. Coef. = -
467 0.794, Sig. = 0.008), as well as the positive correlation of sulphates with
468 turbidity (Corr. Coef. = 0.420, Sig. = 0.037) indicate the presence of sulphate
469 reducing chemolithotrophs and their effect on CO₂ fixation and CH₄ oxidation
470 ⁴⁶.

471 **3.3. Variability of GHGs emissions and statistical interpretation**

472 The strong relation between carbonic GHGs emissions and seasonal changes
473 of temperature is also evident in Figure 2 regarding the study area at North-

474 West Greece. The average water temperature of the studied waterbodies was
 475 7.2 °C to 14.8 °C during the colder period (November to March), while during
 476 the warmer period (April to October) ranged from 18.8 °C to 25.1 °C. These
 477 differences in temperature result in a seasonal pattern regarding GWP of the
 478 waterbodies. As it can be seen from Figure 2, GWP increases in warmer
 479 period, when both CO₂ and CH₄ emissions are recorded, whereas it
 480 decreases in colder period due to methanogenesis inhibition and the sole
 481 production of CO₂. This seasonal pattern is also observed in CH₄ emissions,
 482 but is not observed in CO₂ emissions which are greatly affected by the
 483 presence of algae (carbon fixation) and a plethora of physicochemical
 484 parameters, such as salinity, pH, alkalinity and water temperature ⁴⁷.
 485 Interestingly, GWP is steeply increased, following CH₄ specific emissions in
 486 warmer period, further confirming the results of statistical process.



487
 488 **Figure 2.** Monthly average specific emission rates* and GWP* of Ilarion,
 489 Polyfytos and Zazari sampling stations. *calculated from current (2018)
 490 measurement data (present study) and older (2014 to 2016) measurement
 491 data ²⁴

492 The emissions of CO₂ and CH₄ from Zazari Lake follow the same
493 seasonal pattern with that of GWP, but they were quantitatively higher than
494 those of Polyfytos and Ilarion Reservoirs (Table 5). Negative CO₂ emissions
495 were observed in Zazari Lake, which are attributed to the increased
496 microbial activity in the hypereutrophic Zazari Lake and the subsequent
497 carbon fixation. Compared to that of oligotrophic-mesotrophic lakes of Ilarion
498 and Polyfytos, Zazari Lake exhibits up to 14 times more Chlorophyll *a*
499 concentration and significantly worse water quality attributed to domestic
500 pollution from neighboring villages. Thus, seasonal GHGs emissions of a
501 waterbody and especially of a hypertrophic one, should be assessed in terms
502 of CH₄ emissions or/and GWP in order to count for the impact of microbial
503 activity on GHGs emissions.

504 The statistical correlation between *YΔΩPWQI*, *FTSI* and the
505 parameters expressing GHGs emissions i.e., GWP, specific CO₂ emissions
506 and specific CH₄ emissions, revealed the link between water quality, trophic
507 state and GHGs emissions.

508 In more detail, water quality in terms of *YΔΩPWQI* presented strong
509 positive correlation (Spearman test, Sig.<0.05) with GWP (Cor. Coef. =
510 0.645, Sig. = 0.022) and specific CH₄ emissions (Corr. Coef. = 0.621, Sig. =
511 0.031). *FTSI* presents strong positive correlation with GWP (Corr. Coef. =
512 0.813, Sig. = 0.012) and specific CH₄ emissions (Corr. coef. = 0.823, Sig. =
513 0.008). Neither *YΔΩPWQI* nor *FTSI* presented statistical correlation with
514 specific CO₂ emissions, which as mentioned in previous section is only
515 correlated negatively with turbidity, an indicator of biological growth. The

516 aforementioned results further support the use of CH₄ specific emissions as
517 an indicator for the assessment of GWP that a waterbody exhibits.

518 **4. Conclusions**

519 According to trophic state index, both the six years old Ilarion Reservoir, as
520 well as the forty-three years old Polyfytos Reservoir are classified as
521 oligotrophic through mid-eutrophic reservoirs with “Good” water quality
522 characterization. On the other hand, Zazari Lake, which is considered as a
523 five years old reservoir, is classified as a hypereutrophic Lake with “Poor”
524 water quality characterization.

525 The application of the water quality index revealed that water quality of
526 Ilarion Reservoir is better than of Polyfytos Reservoir, characterized as
527 “Good”, whereas water quality of Polyfytos Reservoir is characterized as
528 “Fair”. Zazari Lake exhibited the worst water quality, characterized as “Poor”.
529 These results highlight the impact of reservoir’s topography and of
530 anthropogenic pressures on trophic state, water quality and the subsequent
531 stabilization time of a reservoir.

532 The emissions of CO₂ and CH₄ were relatively low in Ilarion and
533 Polyfytos Reservoirs. Both reservoirs presented non-significant statistical
534 difference in terms of specific emissions and water quality, despite the fact
535 that Polyfytos Reservoir exhibited three times higher specific CH₄ emissions
536 and worst water quality according to the applied indices. On the contrary,
537 Zazari Lake presented significant statistical differences compared to Ilarion
538 and Polyfytos Reservoirs, both in terms of specific emissions and water
539 quality exhibiting one to two orders of magnitude higher CH₄ emissions and
540 two to three classes better water quality respectively.

541 Water quality and trophic state indices exhibited strong positive
542 correlation with GWP and more specifically with specific CH₄ emissions. On
543 the contrary, no statistical correlation between either indices and CO₂ specific
544 emissions was found, as a result of carbon fixation. The strong positive
545 correlation between GWP, specific CH₄ emissions and temperature justifies
546 the steep increment of GWP in warmer period, with CH₄ emissions
547 contributing up to 97% of the waterbodies' GWP, depending on seasonal
548 water temperature variations.

549 Consequently, CH₄ specific emissions, as well as *YΔΩPWQI* and *FTSI*
550 can be used as indicators for the assessment of GWP that a waterbody
551 exhibit. Quantifying this relation may result in the creation of a rapid GWP
552 assessment tool for researchers and engineers involved in fresh water and
553 reservoir management.

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563 **6. References**

564 1. Dimitriou, K. *et al.* Greenhouse gases (CO₂ and CH₄) at an
565 urban background site in Athens, Greece: Levels, sources and impact of

566 atmospheric circulation. *Atmospheric Environment*. **253**, 118372 (2021)
567 <https://doi.org/10.1016/j.atmosenv.2021.118372>.

568 2. Tremblay, A., Varfalvy, L., Garneau, M. & Roehm, C. (Eds).
569 Greenhouse gas Emissions-Fluxes and Processes: hydroelectric reservoirs
570 and natural environments. (Springer Science & Business Media, 2005).

571 3. Del Sontro, T., Beaulieu, J. J. & Downing, J. A. Greenhouse gas
572 emissions from lakes and impoundments: Upscaling in the face of global
573 change. *Limnology and Oceanography Letters*. **3(3)**, 64-75 (2018)
574 <https://doi.org/10.1002/lol2.10073>.

575 4. Beaulieu, J. J., DelSontro, T. & Downing, J. A. Eutrophication
576 will increase methane emissions from lakes and impoundments during the
577 21st century. *Nature communications*. **10(1)**, 1-5 (2019)
578 <https://doi.org/10.1038/s41467-019-09100-5>.

579 5. Li, Y., Shang, C., Zhang, W., Niu, L., Wang, L. & Zhang, H. The
580 role of fresh water eutrophication in greenhouse gas emissions: A review.
581 *Science of the Total Environment*. **768** 144582 (2021)
582 <https://doi.org/10.1016/j.scitotenv.2020.144582>

583 6. Louis, V.L.St., Kelly, C.A., Duchemin, É., Rudd, J.W. &
584 Rosenberg, D.M. Reservoir Surfaces as Sources of Greenhouse Gases to the
585 Atmosphere: A Global Estimate: Reservoirs are sources of greenhouse gases
586 to the atmosphere, and their surface areas have increased to the point where
587 they should be included in global inventories of anthropogenic emissions of
588 greenhouse gases. *BioScience*. **50(9)**, 766-775 (2000)
589 [https://doi.org/10.1641/0006-3568\(2000\)050\[0766:RSASOG\]2.0.CO;2](https://doi.org/10.1641/0006-3568(2000)050[0766:RSASOG]2.0.CO;2).

- 590 7. Prairie, Y.T. *et al.* Greenhouse gas emissions from freshwater
591 reservoirs: what does the atmosphere see? *Ecosystems*. **21(5)**, 1058-1071
592 (2017) <https://doi.org/10.1007/s10021-017-0198-9>.
- 593 8. Prasad, D. & Siddaraju, G. Carlson's Trophic State Index for the
594 assessment of trophic status of two Lakes in Mandya district. *Adv. Appl. Sci.*
595 *Res.* **3(5)**, 2992-2996 (2012) [https://www.imedpub.com/articles/carlsons-](https://www.imedpub.com/articles/carlsons-trophic-state-index-for-the-assessment-of-trophic-status-of-twolakes-in-mandya-district.pdf)
596 [trophic-state-index-for-the-assessment-of-trophic-status-of-twolakes-in-](https://www.imedpub.com/articles/carlsons-trophic-state-index-for-the-assessment-of-trophic-status-of-twolakes-in-mandya-district.pdf)
597 [mandya-district.pdf](https://www.imedpub.com/articles/carlsons-trophic-state-index-for-the-assessment-of-trophic-status-of-twolakes-in-mandya-district.pdf).
- 598 9. Juutinen, S. *et al.* Methane dynamics in different boreal lake
599 types. <http://urn.fi/URN:NBN:fi-fe2016092624351> (2009).
- 600 10. Liikanen, A. *et al.* Spatial and seasonal variation in greenhouse
601 gas and nutrient dynamics and their interactions in the sediments of a boreal
602 eutrophic lake. *Biogeochemistry*. **65(1)** 83-103 (2003)
603 <https://doi.org/10.1023/A:1026070209387>.
- 604 11. Huttunen, J. T. *et al.* Fluxes of methane, carbon dioxide and
605 nitrous oxide in boreal lakes and potential anthropogenic effects on the
606 aquatic greenhouse gas emissions. *Chemosphere*. **52(3)**, 609-621 (2003)
607 [https://doi.org/10.1016/S0045-6535\(03\)00243-1](https://doi.org/10.1016/S0045-6535(03)00243-1).
- 608 12. European Environment Agency Publications: 3.3. Reservoir and
609 lake eutrophication. [https://www.eea.europa.eu/publications/92-9167-056-](https://www.eea.europa.eu/publications/92-9167-056-1/page006.html)
610 [1/page006.html](https://www.eea.europa.eu/publications/92-9167-056-1/page006.html) (2016).
- 611 13. Paulic, M., Hand, J. & Lord, L. WATER-QUALITY
612 ASSESSMENT FOR THE STATE OF FLORIDA SECTION 305(B) MAIN
613 REPORT; FLORIDA DEPARTMENT OF ENVIRONMENTAL PROTECTION
614 DECEMBER 1996.

615 <http://www.hillsborough.wateratlas.usf.edu/upload/documents/1996%20Water>
616 [Quality%20Assessment%20for%20the%20State%20of%20Florida%20Sectio](http://www.hillsborough.wateratlas.usf.edu/upload/documents/1996%20Water)
617 [n%20305\(b\)%20Main%20Report.pdf](http://www.hillsborough.wateratlas.usf.edu/upload/documents/1996%20Water)

618 14. Duchemin, E., Lucotte, M., Canuel, R. & Chamberland, A.
619 Production of the greenhouse gases CH₄ and CO₂ by hydroelectric reservoirs
620 of the boreal region. *Global Biogeochemical Cycles*. **9(4)**, 529-540 (1995)
621 <https://doi.org/10.1029/95GB02202>.

622 15. Sanchez, L. F., Guenet, B., Marinho, C. C., Barros, N. & de
623 Assis Esteves, F. Global regulation of methane emission from natural lakes.
624 *Scientific reports*. **9(1)**, 1-10 (2019) [https://doi.org/10.1038/s41598-018-](https://doi.org/10.1038/s41598-018-36519-5)
625 [36519-5](https://doi.org/10.1038/s41598-018-36519-5).

626 16. Goldenfum, J. A. GHG Measurement Guidelines for Freshwater
627 Reservoirs in: *The UNESCO/IHA Greenhouse Gas Emissions from*
628 *Freshwater Reservoirs Research Project*. (International Hydropower
629 Association, IHA, 2010).

630 17. Zigah, K.P., Kirsten, O., Brand, A., Dinkel, C., Bernhard, W. &
631 Schubert, C.J. Methane oxidation pathways and associated methanotrophic
632 communities in the water column of a tropical lake. *Limnology and*
633 *Oceanography*. **60(2)**, 553-572 (2015) <https://doi.org/10.1002/lno.10035>.

634 18. Thottathil, S.D., Reis, P.C.J. & Prairie, Y.T. Methane oxidation
635 kinetics in northern freshwater lakes. *Biogeochemistry* **143**, 105–116 (2019)
636 <https://doi.org/10.1007/s10533-019-00552-x>.

637 19. El-Serehy, H. A., Abdallah, H. S., Al-Misned, F. A., Irshad, R.,
638 Al-Farraj, S. A. & Almalki, E. S. Aquatic ecosystem health and trophic status
639 classification of the Bitter Lakes along the main connecting link between the

- 640 Red Sea and the Mediterranean. *Saudi journal of biological sciences*. **25(2)**,
641 204-212 (2018) <https://doi.org/10.1016/j.sjbs.2017.12.004>.
- 642 20. Abbasi, T. & Abbasi, S.A. Approaches to WQI formulation in
643 *Water Quality Indices* 9-24 (Elsevier, 2012).
- 644 21. Zotou, I., Tsihrintzis, V. A. & Gikas, G. D. Performance of Seven
645 Water Quality Indices (WQIs) in a Mediterranean River. *Monitoring and*
646 *Assessment*. **191(8)**, 505 (2019) <https://doi.org/10.1007/s10661-019-7652-4>.
- 647 22. Trikoilidou, E., Samiotis, G., Tsikritzis, L., Kevrekidis, T. &
648 Amanatidou, E. 2017. Evaluation of water quality indices adequacy in
649 characterizing the physico-chemical water quality of lakes. *Environmental*
650 *Processes*. **4(1)**, 35-46 (2017). <https://doi.org/10.1007/s40710-017-0218-y>
- 651 23. Noori, R., Berndtsson, R., Hosseinzadeh, M., Adamowski, J. F.
652 & Abyaneh, M. R. A critical review on the application of the National
653 Sanitation Foundation Water Quality Index. *Environmental Pollution*. **244**,
654 575-587 (2019). <http://dx.doi.org/10.1016/j.envpol.2018.10.076>.
- 655 24. Samiotis, G., Pekridis, G., Kaklidis, N., Trikoilidou, E.,
656 Taousanidis, N. & Amanatidou, E. Greenhouse gas emissions from two
657 hydroelectric reservoirs in Mediterranean region. *Environmental Monitoring*
658 *and Assessment*. **190(6)**, 363. (2018b) [https://doi.org/10.1007/s10661-018-](https://doi.org/10.1007/s10661-018-6721-4)
659 [6721-4](https://doi.org/10.1007/s10661-018-6721-4).
- 660 25. Samiotis, G., Trikoilidou, E., Tsikritzis, L. & Amanatidou, E.
661 Comparative water quality assessment between a young and a stabilized
662 hydroelectric reservoir in Aliakmon River, Greece. *Monitoring and*
663 *Assessment*. **190(4)**, 234 (2018a) <https://doi.org/10.1007/s10661-018-6602-x>.

- 664 26. APHA-American Public Health Association, American Water
665 Works Association, Water Pollution Control Federation, & Water Environment
666 Federation, Standard methods for the examination of water and wastewater
667 (American Public Health Association, 2017)
- 668 27. ISO-International Organization for Standardization. Water
669 quality-determination of dissolved Li⁺, Na⁺, NH₄⁺, K⁺, Mn²⁺, Ca²⁺, Mg²⁺,
670 Sr²⁺ and Ba²⁺ using ion chromatography-Method for water and waste water
671 ISO 14911:1998.
- 672 28. ISO-International Organization for Standardization.
673 Determination of dissolved anions by liquid chromatography of ions—Part 1:
674 Determination of bromide, chloride, fluoride, nitrate, nitrite, phosphate and
675 sulphate. ISO 10304-1: 2007.
- 676 29. Brooker, M.R., Bohrer, G. & Mouser, P.J. Variations in potential
677 CH₄ flux and CO₂ respiration from freshwater wetland sediments that differ
678 by microsite location, depth and temperature. *Ecological Engineering*. **72**, 84–
679 94 (2014)
- 680 30. Mavromatidou C. *et al.* 2020. A water quality assessment tool
681 for decision making, based on widely used water quality indices. 4th EWaS
682 Conference, Corfu, in press (2020).
683 <http://dx.doi.org/10.1016/j.ecoleng.2014.05.028>
- 684 31. Gikas, G.D., Sylaios, G.K., Tsihrintzis, V.A., Konstantinou, I.K.,
685 Albanis, T. & Boskidis, I. Comparative evaluation of river chemical status
686 based on WFD methodology and CCME water quality index. *Science of the*
687 *Total Environment*. **745**, 140849 (2020)
688 <https://doi.org/10.1016/j.scitotenv.2020.140849>.

- 689 32. Ministry of Rural Development and Food of Greece (a).
690 Monitoring of Chemical Quality of Water (Surface and Groundwater) for
691 Irrigation on a River Basin Scale for the Rivers of Macedonia, Thrace and
692 Thessaly, Greece: Results for Aliakmon River Basin (in Greek),
693 [http://www.minagric.gr/ardefatika/files/results/geol/15.RESULTS_ALIAKMONA.](http://www.minagric.gr/ardefatika/files/results/geol/15.RESULTS_ALIAKMONA.pdf)
694 [pdf](http://www.minagric.gr/ardefatika/files/results/geol/15.RESULTS_ALIAKMONA.pdf) (2020)
- 695 33. Ministry of Rural Development and Food of Greece (b).
696 Monitoring of Chemical Quality of Water (Surface and Groundwater) for
697 Irrigation on a River Basin Scale for the Rivers of Macedonia, Thrace and
698 Thessaly, Greece: Results for Vegoritida's catchment area (in Greek).
699 http://www.minagric.gr/ardefatika/files/results/hydro/10.%20HYDRO_VEGORIT
700 [IDAS.pdf](http://www.minagric.gr/ardefatika/files/results/hydro/10.%20HYDRO_VEGORIT) (2020)
- 701 34. Huber, W. C. *et al.* A classification of Florida lakes. Water
702 Resources Research Center Publication, 72 (1982)
703 <https://ufdc.ufl.edu/UF00000142/00001>.
- 704 35. Carlson, R. E. A trophic state index for lakes 1. *Limnology and*
705 *oceanography*. **22(2)**, 361-369 (1977)
706 <https://doi.org/10.4319/lo.1977.22.2.0361>.
- 707 36. Richardson, J. Water Quality Report for Selected Lakes and
708 Streams. Leon County Public Works. Division of Engineering Services.
709 Florida's Capital County, United States. p. 23.
710 http://cms.leoncountyfl.gov/Portals/0/publicworks/engservices/docs/WQR_Ma
711 [y2013_comp.pdf](http://cms.leoncountyfl.gov/Portals/0/publicworks/engservices/docs/WQR_Ma) (2013).
- 712 37. EPA, 2021, Understanding Global Warming Potential.
713 www.epa.gov. (2021)

- 714 38. Walker, W.W. Empirical Methods for Predicting Eutrophication in
715 Impoundments, Rep. 3, Phase II. Model Refinements. Tech. Rep. E-81-9.
716 U.S. Army Engineer Waterways Experiment Station, Vicksburg, MS.
717 [https://www.semanticscholar.org/paper/Empirical-Methods-for-Predicting-](https://www.semanticscholar.org/paper/Empirical-Methods-for-Predicting-Eutrophication-in-Walker/c04aaf1218df008dac5e7067bd9e680ef7163977)
718 [Eutrophication-in-Walker/c04aaf1218df008dac5e7067bd9e680ef7163977](https://www.semanticscholar.org/paper/Empirical-Methods-for-Predicting-Eutrophication-in-Walker/c04aaf1218df008dac5e7067bd9e680ef7163977)
719 (1985).
- 720 39. Yu, H., Tsuno, H., Hidaka, T. & Jiao, C. Chemical and thermal
721 stratification in lakes. *Limnology*. **11**, 251–257 (2010)
722 <https://doi.org/10.1007/s10201-010-0310-8>.
- 723 40. Leach, T.H. *et al.* Patterns and drivers of deep chlorophyll
724 maxima structure in 100 lakes: the relative importance of light and thermal
725 stratification. *Limnol. Oceanogr.* **63(2)**, 628–646 (2018)
726 <https://doi.org/10.1002/lno.10656>.
- 727 41. Søballe, D.M., Kimmel, B.L., Kennedy, R.H. & Gaugush, R.F.
728 Biodiversity of the Southeastern United States: Aquatic Communities.
729 (JohnWiley & Sons Inc., 1992).
- 730 42. Cotterill, N. G. & Thornton, J. A. Hydroclimate development
731 following impoundment in a tropical African manmade lake (Lake Robertson,
732 Zimbabwe). *Journal of the Limnological Society of Southern Africa*, **11**, 54–61
733 (1985) <https://doi.org/10.1080/03779688.1985.9632829>.
- 734 43. Samiotis, G., Trikoilidou, E., Michailidis, A., Zagana, E. &
735 Amanatidou, E. Comparative water quality assessment of a new and an old
736 reservoir in Aliakmon River using both statistical tools and a modified NSF
737 water quality index. 13th International Conference on Protection and
738 Restoration of the Environment. (2016).

739 [742 44. Giles, J. Methane quashes green credentials of hydropower.
743 *Nature*. **444\(30\)**, 524–525 \(2006\).](https://www.semanticscholar.org/paper/Comparative-water-quality-
740 <u>assessment-of-a-new-and-a-Samiotis-
741 <u>Trikoilidou/9ee15da0e417ace6085b6a87d4fe78382ffd47a6.</u></u></p></div><div data-bbox=)

744 45. Davidson, T.A. *et al.* Eutrophication effects on greenhouse gas
745 fluxes from shallow lake mesocosms override those of climate warming.
746 *Global Change Biology*. **21(12)**, 4449-4463 (2015) doi: 10.1111/gcb.13062.

747 46. Alfreider, A., Baumer, A., Bogensperger, T., Posch, T., Salcher,
748 M.M. & Summerer, M. CO₂ assimilation strategies in stratified lakes: Diversity
749 and distribution patterns of chemolithoautotrophs. *Environ. Microbiol.* **19(7)**,
750 2754–2768 (2017) https: doi:10.1111/1462-2920.13786.

751 47. Cole, J. J. & Prairie, Y. T. Dissolved CO₂ in *Encyclopedia of*
752 *Inland Waters*. **2**, 30-34 (Elsevier, 2009). [https://doi.org/10.1016/B978-
012370626-3.00091-0](https://doi.org/10.1016/B978-
753 012370626-3.00091-0).

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