

Phytochemical Evaluation of Hibiscus Sabdariffa Powder, Jam and Yoghurt

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Research

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1 **Phytochemical Evaluation of *Hibiscus Sabdariffa* Powder, Jam and Yoghurt**

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20 **ABSTRACT**

21 **Background**

22 *Hibiscus sabdariffa* is popularly known as food and herbal drink with numerous health
23 benefits. The phytochemical compounds present in *Hibiscus sabdariffa* calyces are
24 important in developing nutraceutical foods. In this study hibiscus jam and yoghurt
25 were produced from dried hibiscus calyces' powder.

26 **Methods**

27 The phytochemical content and antioxidant activity of these products were then
28 analysed in terms of Total phenolic content (TPC), Total flavonoid content (TFC),
29 Condensed tannins (CT) and DPPH Scavenging activity.

30 **Results**

31 The results showed the presence of all phytochemical compounds (TPC, TFC, CT)
32 and antioxidant activity in all *Hibiscus sabdariffa* products. The hibiscus calyces
33 powder showed the highest phytochemical contents of 35.24 mg GAE. g⁻¹, 0.91 mg
34 QE. g⁻¹, 2.85 mg CAE. g⁻¹ and 48.2 % inhibition for TPC, TFC, CT and DPPH
35 Scavenging activity, respectively. Hibiscus jam and Hibiscus yoghurt had
36 phytochemical contents of 6.44 and 4.81 mg GAE. g⁻¹, 0.19 & 0.24 mg QE. g⁻¹, 1.40 &
37 0.66 mg CAE. g⁻¹ and 26.2 & 39.3 % inhibition for TPC, TFC, TC and DPPH
38 Scavenging activity, respectively.

39 **Conclusions**

40 The results of the current study showed that there is potential in using *Hibiscus*
41 *sabdariffa* to develop functional foods.

42 **Key Words:** *Hibiscus sabdariffa*, phytochemicals, antioxidants, calyces, therapeutic

43 **BACKGROUND**

44 Since ancient times, plant-based traditional medicine has played a major role in the
45 therapy of many diseases (1). The use of herbal extracts as medicine for the treatment
46 of many disease is well documented (2,3). Currently, the food market encourages
47 industries to develop products that have functional, nutritional, and therapeutic
48 properties (4). In efforts to meet food market demands, *Hibiscus sabdariffa* or “Mutete”,
49 has gained popularity as a miracle plant with potential medicinal benefits (5). This is
50 mainly due to the therapeutic effects it has against many diseases including cancer
51 (6).

52 *Hibiscus sabdariffa*, is an annual herbaceous subshrub that belongs to the
53 *Malvaceae* family (Formaggio *et al.*, 2015). The plant is native to India but also grown
54 in many parts of the world including Africa (3). It is widely cultivated for its strong fibre
55 (3), and popular for the edibility and therapeutic effects of its leaves and calyces (8,9).
56 The calyces and leaves have been used to make food (juice, jam, jellied
57 confectionaries, ice cream, chocolates and flavouring agents) and herbal medicine
58 (herbal tea to sooth colds, clear a blocked nose, clear mucous, as an astringent,
59 promoting kidney function, aiding digestion, as a general tonic, diuretic, and antipyretic
60)(8,10–12). The calyces and leaves contain bioactive molecules including flavonoids,
61 anthocyanins, alkaloids, saponins, steroid, sterols and tannins (3,10,11). The seeds
62 are a great source of lipid-soluble antioxidants, particularly γ -tocopherol (1). These
63 bioactive molecules have choleric, febrifugal, hypertensive and diuretic effects,
64 decreasing blood viscosity, stimulating intestinal peristalsis and reducing blood
65 pressure (5). The *Hibiscus sabdariffa* extract has been effective in treating abscesses,
66 bilious conditions, cancer, coughs, kidney stones and *Mycobacterium Tuberculosis*

67 (1,13). The tea has been used to lower blood pressure (Chopra et al., 1986),
68 cholesterol level, and to prevent cardiovascular disease (1).

69 The *Hibiscus sabdariffa* plant deserves attention especially in Namibia where it is
70 underutilised. The plant's tremendous health benefits also present an opportunity for
71 the development of potential nutraceutical and functional food products. The aim of
72 this study was, therefore, to evaluate the potential of using *Hibiscus sabdarrifa* to
73 develop foods that have nutraceutical and functional properties

74 **MATERIALS AND METHODS**

75 **Collection of samples**

76 *Hibiscus sabdariffa* calyces were used as the plant material in the study. Air-dried
77 *Hibiscus sabdariffa* calyces were purchased from a commercial farm 20 km Northeast
78 of Otavi, Namibia. The calyces were transported to the University of Namibia's Food
79 Science and Technology Department and kept at 7°C before use.

80 **Preparation of H. sabdariffa calyces' powder**

81 Dried *Hibiscus sabdariffa* calyces were ground using pestle and mortar. Following this,
82 the ground calyces were sieved using a 0.5 mm aperture sieve to obtain a fine powder.

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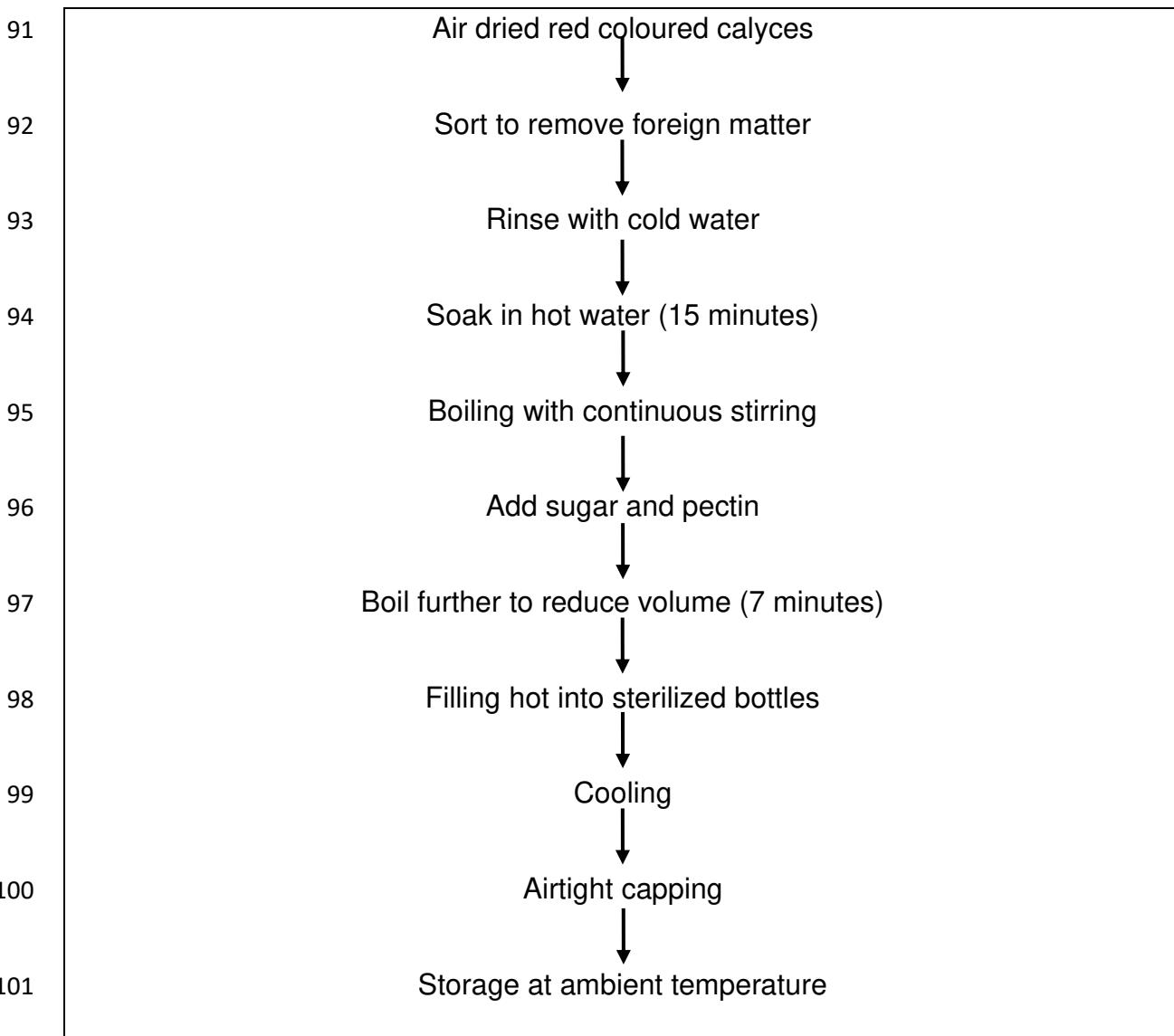
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88 **Preparation of Mutete Jam**

89 The jam was prepared by boiling the clean hibiscus pulp with enough sugar and pectin
90 to a thick consistency (Figure 1).



102 Figure 1. Preparation of Hibiscus jam

103 **Preparation of Mutete Yoghurt**

104 Yoghurt was prepared using commercial thermophilic starter culture containing
105 *Lactobacillus bulgaricus* and *Streptococcus thermophiles*. In brief, fresh good quality
106 cow milk was pasteurised at 85-90°C for 30 minutes to kill unwanted microorganisms

107 as well as denature whey proteins (Albumins and Globulins). Stabilizers were
108 subsequently added at the rate of 5-10 millilitre per Litre of milk after which, 6% white
109 sugar was added and mixed while hot. It was then cooled to 43-45°C. The starter
110 culture was then added, mixed, and incubated to 43°C for 6-8 hrs until coagulation.
111 The yoghurt was cooled to 5-7°C overnight to set. The yoghurt was then gently
112 agitated to obtain a smooth and thick texture before 15% of the hibiscus powder was
113 added

114 **Phytochemical analysis**

115 Phytochemical analysis in terms of total phenolic content, total flavonoid content,
116 condensed tannins, and antioxidant activity were done using spectrophotometric
117 techniques.

118 **Sample extraction**

119 Sample extraction was done following a method outlined by Rooney and Waniska
120 (2004) using two extraction solvents, 1% Hydrochloric acid (v/v) in methanol. About
121 0.5 g of the sample was weighed into 50 ml centrifuge tubes. To the tubes, 5 ml of 1%
122 HCl in methanol was added. The mixture was sonicated at 25°C for 10 minutes. After
123 sonication, the tubes were centrifuged at 4000 rpm for 5 minutes. The supernatant
124 was decanted in a separate centrifuge tube and extraction was repeated using another
125 5 ml of 1% HCl in methanol. The extracts were stored at -4°C until further analysis.
126 Extraction was done in duplicates and determinations were done in triplicates.

127 **Determination of Total Phenolic Content (TPC)**

128 Total phenolic contents were determined using the Folin-Ciocalteu method described
129 by Mohd-esa et al., (2010). Extract sample of 0.5 ml was mixed with 0.1 ml of 0.5N

130 Folin-Ciocalteu reagent (2.5 ml of the 2N Folin-Ciocalteu reagent original bottle mixed
131 with 7.5 ml of distilled water) and gallic acid stock solution (50 mg of gallic acid into
132 100 ml of methanol in a volumetric flask). The mixture was then incubated for 15
133 minutes at room temperature in the dark after which 2.5 ml of sodium carbonate (20
134 g/100 ml) was added to the test tubes and incubated once more for 30 minutes at
135 room temperature in the dark. The absorbance of the sample extract was measured
136 at 760 nm using UV/VIS spectrophotometer. The total phenolic content was expressed
137 as gallic acid equivalents (GAE) in milligrams per gram of the sample. It was calculated
138 using the equation $y = 1.3016x - 0.0275$ with $R^2 = 0.9925$, generated from the gallic
139 acid standard curve. Where Y is the absorbance of the sample extract and X is the
140 unknown concentration of the sample extract.

141 **Determination of Total Flavonoids Content (TFC)**

142 Total flavonoids content was determined using Aluminium Chloride method described
143 by Chang, Yang, Wen, and Chern (2002). For the standard curve, quercetin stock
144 solution was prepared by dissolving 100 mg of quercetin into 100 ml of methanol in a
145 volumetric flask. Volumes of quercetin stock solution (0, 0.0315, 0.0625, 0.125, 0.25,
146 0.5, 1, 2, and 4 ml) were pipetted into separate volumetric flasks and made up to 50
147 ml with methanol to have concentrations of 0, 0.000625, 0.00125, 0.0025, 0.005, 0.01,
148 0.02, 0.04 and 0.08 mg/ml. The appropriate number of test tubes were prepared for
149 each sample to replicate, blank and standard solutions. In those test tubes, 0.5 ml of
150 1.2% (w/v) aluminium chloride and 0.5 ml of 120 mM potassium acetate (1.1778 g into
151 100 ml distilled water) were added. One ml of sample extracts or blank (extracting
152 solvent) or standard solutions was added to the test tubes containing aluminium
153 chloride and potassium acetate solution. The mixture was incubated at room

154 temperature for 30 minutes. The absorbance of the samples, blank and standard
155 solutions were read at 415 nm using a UV/VIS spectrophotometer. The total flavonoids
156 content of the sample extracts was calculated using the equation $y = 31.046x + 0.0311$
157 with $R^2 = 0.9975$, generated from the quercetin standard curve. Where Y is the
158 absorbance of the sample extracts and X is the unknown concentration of the sample
159 extract.

160 **Determination of Condensed Tannins (CT)**

161 Condensed tannins were determined using the Vanillin-HCl method described by
162 Price, Van Scyoc and Butler (1978). For the standard curve, the catechin stock
163 solution was prepared by dissolving 10 mg of catechin into 100 ml of methanol in a
164 volumetric flask. Volumes of catechin stock solution (0, 1, 2, 3, 5 and 10 ml) were
165 pipetted into separate volumetric flasks and made up to 10 ml with methanol to have
166 concentrations of 0, 0.01, 0.02, 0.03, 0.05 and 0.1 mg/ml. The standard solutions,
167 sample extract and vanillin reagent (4% HCl (v/v) in methanol and 0.5% (w/v) vanillin
168 in methanol) were kept in the water bath at 30°C for 20 minutes before mixing the
169 reactants. One ml of sample extract or blank (extracting solvent) or standard solutions
170 was mixed with 5 ml of vanillin reagent in test tubes and maintained at 30°C in the
171 water bath for 20 minutes. The absorbance of the sample extract, blank and standard
172 solutions was read at 500 nm using a UV/VIS spectrophotometer. The condensed
173 tannins of the sample extracts were calculated using the equation $y = 0.222x - 0.0011$
174 with $R^2 = 0.9987$, generated from the catechin standard curve. Where Y is the
175 absorbance of the sample extracts and X is the unknown concentration of the sample
176 extract.

178 **Determination of Antioxidant Capacity**

179 The antioxidant activity of hibiscus products (hibiscus calyces' powder, hibiscus jam
180 and hibiscus yoghurt) was evaluated by the free radical scavenging activity of the
181 products using the 2,2-diphenyl-2-picrylhydrazyl (DPPH) free radical according to the
182 method described by McCune and Johns (2002). One millilitre of 0.3 mM DPPH
183 solution (0.012 g into 100 ml of methanol) was added to test tubes containing 5 ml
184 hibiscus product. For the blank sample, 1.0 ml of extracting solvent was added to a
185 test tube containing 1.0 ml of 0.3 mM DPPH solution and 1.0 ml of methanol. Quercetin
186 was used as positive control and standard concentration solutions (0.1, 0.2, 0.4 and
187 0.6 mg/ml) of quercetin were prepared. From each standard concentration solution,
188 1.0 ml was added to test tubes containing 1.0 ml of 0.3 mM DPPH solution and 1.0ml
189 of methanol. The tubes were incubated for 10 minutes at room temperature in the dark
190 for the reaction to take place. The absorbance of the sample extract, blank, standard
191 solutions were read at 517 nm using a UV/VIS spectrophotometer. The radical
192 scavenging activity was calculated as percentage inhibition of DPPH discolouration
193 according to the following equation:

$$194 \quad \% \text{ inhibition} = \frac{A_0 - A_s}{A_0} \times 100$$

195 Where A_s is the absorbance of the sample extract or standard and A_0 is the
196 absorbance of the negative control, which is the blank. Quercetin was used as a
197 standard for comparison in antioxidant activities of all infusions

198 **Data Analysis**

199 All determinations were done in triplicates. The results were reported as mean ±
200 standard deviation that was analysed using SPSS software version 21.

201 **Results and discussion**

202 Studies have shown the presence of various phytochemicals in *Hibiscus sabdariffa*
203 calyces (Formaggio et al., 2015; Ghodke and Mane, 2017; Okereke et al., 2016). To
204 evaluate its potential in developing functional foods, this study, determined the
205 phytochemical contents and antioxidant activity of products (powder, jam, and yoghurt)
206 made from dried *Hibiscus sabdariffa* calyces. The results are summarised in Table 1.

207 **Total phenolic compounds**

208 The total phenolic content varied between the three hibiscus products. Comparatively,
209 the hibiscus powder showed the highest total phenolic content of 35.42 ± 0.28 mg
210 GAE. g⁻¹ (Table 1). Hibiscus jam and yoghurt showed lower total phenolic content of
211 6.44 ± 0.20 mg GAE. g⁻¹ and 4.81 ± 0.57 mg GAE. g⁻¹ respectively. The total phenolic
212 content of calyces powder in this study is slightly lower than those obtained by Hassan,
213 (2014) and Oloumi, Shakeri, & Behzadi,(2016) of 41.07 mg GAE. g⁻¹ and 59 mg GAE.
214 g⁻¹ in hibiscus extract, respectively. The yoghurt results (4.81 ± 0.57 mg GAE. g⁻¹)
215 corresponds with findings from Dabija et al., (2018)'s study. In their study, the total
216 phenolic content was 5.12 mg GAE/g and 4.17 mg GAE/g for yoghurt samples
217 inoculated with hawthorn (*Crataegus monogyna*) and sage (*Salvia officinalis L.*)
218 respectively. The variation in total phenolic content among the products is possibly
219 due to processing. High temperature has been reported to negatively affect the
220 phenolic compounds present in medicinal plants and herbs such as hibiscus (2). In
221 this study, the hibiscus powder was minimally processed using pestle and mortar,
222 while the processing of jam and yoghurt involved thermal treatment. The impact of
223 temperature might have caused degradation of phenolic compounds which resulted in
224 lower phenolic content observed in hibiscus jam and yoghurt (Table1). The presence

225 of phenolic compound is attributable to the therapeutic properties of the plant.
226 Phenolics present in plants have been popular mainly because of their potential
227 antioxidant activity (16). They may therefore be responsible for the protective effect
228 against the risk of many disease processes, such as cancer, cardiovascular and
229 circulatory diseases (10).

230 **Total flavonoid compounds**

231 The flavonoid content of the products differed and ranged between 0.19 ± 0.01 QE. g⁻¹
232 and 0.91 ± 0.03 QE. g⁻¹. Hibiscus powder showed the highest flavonoid content, while
233 the hibiscus jam had the lowest flavonoid content. The total flavonoid content of
234 calyces powder observed in this study is comparable to those observed by other
235 researchers. Oloumi et al., (2016) reported flavonoid content of 0.97 ± 0.05 QE. g⁻¹ in
236 hibiscus extract. Formagio, ASN.a*, Ramos, DD.a, Vieira, MC.a, Ramalho et al.,
237 (2015) reported flavonoid content of 1.18 ± 2.51 QE. g⁻¹ in hibiscus calyces. The total
238 flavonoid content of jam in this study was higher compared to jam from other products.
239 Farida, (2018) observed flavonoid content 0.08 QE. g⁻¹ in melon jam. This illustrates
240 that *Hibiscus sabdariffa* is a good source of flavonoid content. Flavonoids are
241 produced as natural secondary metabolites in plants and contain high antioxidant
242 properties (16). These compounds are capable of scavenging free radicals. They can
243 therefore be effective against many human disorders (16). The flavonoids in *hibiscus*
244 *sabdariffa* showed desirable effects on peroxidase and protease activity in human
245 blood (19), which confirmed its potential as an antioxidant and anti-aging plant.
246 Flavonoids have also been shown to protect against coronary heart disease (8).

247

248

249 ***Condensed Tannins***

250 The condensed tannins content of the hibiscus products followed the trend observed
251 with total phenolic content (Table 1). Again, the calyces powder showed the highest
252 condensed tannins of 2.85 ± 0.06 mg CAE. g⁻¹ followed by hibiscus jam with flavonoid
253 content of 1.40 ± 0.05 mg CAE. g⁻¹. Despite the small amounts of tannins present in
254 hibiscus jam and yoghurt, tannins found in *hibiscus sabdariffa* extract have shown
255 antioxidant effects (19). The presence of tannins in food product is advantageous
256 because they have cardioprotective actions (3), may reduce the risk of cancer and
257 may prevent menopausal symptoms (20). Tannins also have anti-microbial activity by
258 precipitating protein content of the outer wall of the microbes, thereby forming a
259 complex with the proteins and stop their activities (10).

260 ***DPPH scavenging activity***

261 The results of the antioxidant activity expressed as DPPH scavenging activity
262 correlates with that of total flavonoid content (Table 1). The highest antioxidant activity
263 of $48.2 \pm 0.2\%$ was observed in calyces' powder. The lowest antioxidant activity of
264 $26.2 \pm 0.3\%$ was observed in jam. Although, hibiscus jam had the lowest antioxidant
265 activity, this percentage was higher than the antioxidant activities observed in other
266 fruit jams. Farida, (2018) found antioxidant activity of 4.95% in melon jam. While
267 Rababah *et al.* (2011) found an antioxidant activity of 10.06%, 9.95% and 8.96%, for
268 cherry, apricot and fig jams respectively. These suggest that hibiscus Jam has good
269 antioxidant activity, with potential health benefits. The higher antioxidant activity in the
270 calyces' powder may be attributed to presence of higher polyphenolic compounds
271 (total phenolic, total flavonoid and tannins) in the product (Table 1).

272 **Table 1** Phytochemical content of *Hibiscus sabdariffa* products

	Hibiscus products		
	Calyces		
Phytochemical compound	powder	Jam	Yogurt
TPC (mg GAE/g)	35.42 ± 0.28	6.44 ± 0.20	4.81 ± 0.57
TFC (mg QE/g)	0.91 ± 0.03	0.19 ± 0.01	0.24 ± 0.03
CT (mg CAE/g)	2.85 ± 0.06	1.40 ± 0.05	0.66 ± 0.06
DPPH scavenging activity (%) inhibition	48.2 ± 0.2	26.2 ± 0.3	39.3 ± 0.2

273

274 **CONCLUSIONS**

275 The results showed the presence of phenolic, flavonoids, tannins and antioxidant
 276 activity in all three hibiscus samples (powder, jam and yoghurt). This indicate that
 277 *Hibiscus Sabdariffa* is a good source of phytochemical compounds and provides high
 278 antioxidant activity based on scavenging the DPPH radicals. This demonstrates its
 279 potential as a nutraceutical food, providing many health benefits. The results also
 280 show higher phytochemical compounds and antioxidant activity in hibiscus calyces
 281 compared to hibiscus jam and yoghurt. This suggests that, when higher phytochemical
 282 content is required, minimally processed hibiscus product such as hibiscus extract or
 283 powder should be used. Likewise, when thermally processed hibiscus products such
 284 as hibiscus jam and yoghurt are consumed, it can be expected to result in less
 285 phytochemical compound acquired. The presence of phytochemical compounds and
 286 antioxidant activity in hibiscus jam and yoghurt, however, implies that hibiscus calyces'
 287 powder can be successfully processed or incorporated into products such as yoghurt
 288 obtain a product with health benefits.

289 **DECLARATIONS**

290 **Ethical approval and consent to participate**

291 Not applicable

292 **Consent for publication**

293 Not applicable

294 **Availability of data and materials**

295 The datasets used and/or analysed during the current study are available from the
296 corresponding author on reasonable request.

297 **Competing interests**

298 The authors declare that they have no competing interests

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301 Research Centre, Ottawa, Canada provided through the African Centre for
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304 **Author's contributions**

305 Conception design: P.H. , C.S.

306 Data acquisition and Analysis: H.S., K.H., P.H.

307 Interpretation of Data: K.H., E.S., C.S.

308 Drafting: E.S., C.S.

309 Reading and approving final manuscript: All authors

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Figures

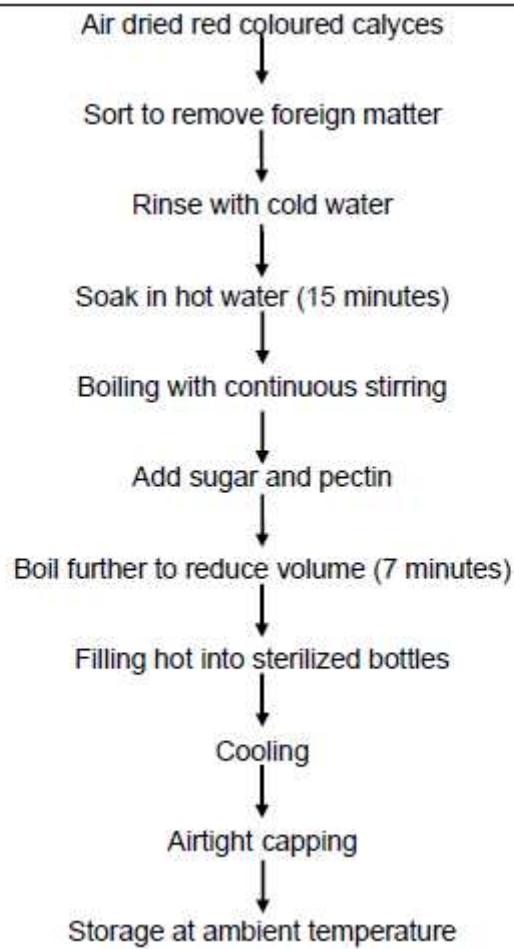


Figure 1

Preparation of Hibiscus jam