

Porous Cu-MOF Nanostructures with Anticancer Properties Prepared by a Controllable Ultrasound-Assisted Reverse Micelle Method

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Abstract

The ultrasonic assisted reverse micelle method was used to create Cu-MOF from $\text{Cu}(\text{NO}_3)_2 \cdot 3\text{H}_2\text{O}$ and 2,6-pyridine dicarboxylic acid in a 1:1 molar proportion. It has been characterized using FT-IR, XRD, nitrogen adsorption analysis SEM and TEM-EDX. Cu-MOF has anticancer properties against MCF-7 breast cancer cells. Cytotoxicity testing was performed on MCF-7 breast cancer cells using the MTT cell viability assay, and cell proliferation and viability were found to be approximately 24 % higher than the control.

1. Introduction

Cancer is, unfortunately, one of the leading causes of death for millions of people worldwide. Cancer biology has made great strides in the last decade, but cancer deaths remain high. [1] Traditional drugs for cancer treatment have limitations such as poor pharmacokinetics, poor biological distribution, and adverse side effects. [2] Chemotherapy is the most commonly used cancer treatment, but it has several drawbacks, the most significant of which are low therapeutic efficiency in the treatment process and side effects on normal cells. [1] Since the 1970s, drug release control in drug delivery has been expanding. [2] Drug delivery systems based on nanoparticles are one of the newer methods in drug delivery systems. Drug delivery systems based on nanoparticles can avoid these issues while also increasing efficiency through targeted drug delivery, controlled release, and drug degradation protection. Metal-organic frameworks (MOFs), layered double hydroxides (LDHs), graphene oxide (GO), and magnetite are some examples of popular drug delivery systems that use nanoparticles today. [1] Some cases that have been investigated in the use of MOF (organic metal frameworks) as drug delivery systems include excellent surface, thermal and chemical stability, high pore volumes, regular porosity and easy operation. [1, 2] Numerous biological reports have been reported from organic metal frameworks, including inhibition of human glioma cell growth [3], anticancer activity [4–6], and antimicrobial properties [7–9]. One of the MOFs with high biological properties is Cu-MOF.[10–12] It is worth noting that the true nature of the active sites in many MOFs containing metal ions is saturated with the coordination of organic ligands. [13] Cu-MOF has superior antibacterial and anticancer properties. [14, 15] The bactericidal mechanism of Cu-MOF is due to the diffusion of Cu^{2+} ions. In addition, the negative charges of lipoproteins are absorbed into the cell wall, they enter the cell and damage the cell wall, alter the function of the enzymes of the cell wall, or create cell wall holes. [16]

In this paper, we have reported synthesis of Cu-MOF using ultrasonic assisted reverse micelle method and characterized by FT-IR, XRD, SEM and nitrogen adsorption. Anticancer activities against MCF-7 breast cancer cells are also reported.

2. Experimental Section

2.1 Materials

$\text{Cu}(\text{NO}_3)_2 \cdot 3\text{H}_2\text{O}$ and 2, 6 pyridine dicarboxylic acid were purchased from Merck. $\text{C}_6\text{H}_8\text{N}_2\text{NaSO}_4$ was purchased from Sigma-Aldrich. Double distilled water (DDW) was used in each experiment.

2.2 Method

$\text{Cu}(\text{NO}_3)_2 \cdot 3\text{H}_2\text{O}$ (Merck, 98%) and 2, 6 pyridine dicarboxylic acid (Merck, 99%) are mixed with 1:1 mmol dissolved in 25 mL of DDW during the preparation of the samples using the ultrasonic assisted reverse micelle method. The resulting solution was added to a mixture of 0.077 mmol of $\text{C}_6\text{H}_8\text{N}_2\text{NaSO}_4$ (Sigma, 99%) and 8 mL of C_6H_{14} . The resulting mixture was then stirred for 1 h at 85°C. The resulting solution was placed in the ultrasonic device and exposed to ultrasonic irradiation under optimal conditions, which included an ultrasonic duration of 21 minutes, a power of 175 W, and an ultrasonic temperature of 40°C. Cu-MOF crystals form after 30 minutes and are separated by centrifugation and washed with DMF.

2.3 Characterization

The X-ray diffraction (XRD) employed for characterization and determination of the crystalline structure and phases during the synthesis of Cu-MOF. To achieve this aim, a powder X-ray diffractometer (Expert MPD, pananalytical, $\text{CuK}\alpha = 0.154.6 \text{ nm}$) were used in the range of $2\theta = 4 - 30$ degree with the step width of 0.05 degree. Scanning electron microscope (SEM, model Em 3200, china KyKy corporation) utilized for investigation of the surface morphology. Fourier transform infrared (FT-IR; SHIMADZU FT8400 spectrometer) with a Bruker- tensor 27 series was utilized for determination of vibrational frequency of the prepared samples in the range of 500 and 4000 cm^{-1} . Porosities, surface area and pore textural characteristics of samples were determined by adsorption/desorption measures (BET, Belsorp mini II) at 77K in N_2 atmosphere. The absorbance was read by spectrophotometer (BioTek Instruments, Inc., Bad Friedrichshall, Germany).

2.4 Anticancer activities: Anticancer activity of Cu-MOF was evaluated using MCF-7 breast cancer cells and the MTT cell viability assay according previously reported methods [17, 18]. MCF-7 cells were isolated from the pleural effusion of a 58-year-old woman with metastatic disease. In pellet culture system consisted of control medium including RPMI 1640, 10% FBS and 100 μL of penicillin G/streptomycin mixtures, cells were cultured for a period of two weeks. During cell culture, cell passaging was performed by trypsinization and washing was by phosphate-buffered saline. The cells with density of 1.2×10^4 (cells/well) were seeded in 96-well plates and for 24 h incubated in condition of 37°C and 5% CO_2 . With concentrations of 5, 10, 20, 40, 80, 120 and 200 $\mu\text{g}/\text{mL}$ of Cu-MOF, cells were treated for 24 and 48 hours. Then, the medium was then removed and 50 $\mu\text{L}/\text{well}$ of MTT solutions (2 mg/mL in PBS) and 150 $\mu\text{L}/\text{well}$ of fresh medium were added and incubated for 4 h. Finally, after removing the MTT solutions, to solubilize the formazan crystals, 200 μL of DMSO was added and at 570 nm, the absorbance was read using a spectrophotometer.

3. Results And Discussion

SEM and TEM analysis were used to investigate the green Cu-MOF morphology, as shown in Fig. 1 and 2. Cu-MOF morphology was spherical or octahedral, and the average particle size distribution is less than 100 nm.[19] As a result, the anticancer properties of the sample were investigated. Fig. 3 depicts the phase formation and purity of the samples as determined by X-ray diffraction. The sample pattern belongs to Cu-MOF, the peaks marked with a circle at 31°, 37°, 43°, 52°, 59°, and 77° for Cu-MOF (JCPDS01-072-0075) [20, 21].

The FTIR spectrum of Cu-MOF is depicted in Fig. 4. Surface water exhibits the structure and hydroxyl bond of the Cu-MOF structure in the 3358 cm⁻¹ band. [19, 22] The C=O vibration in Cu-MOF is responsible for the peaks ranging from 2355 cm⁻¹ to 2099 cm⁻¹ and 1632 cm⁻¹. [16] and COO as CA vibration in Cu-MOF [23], respectively. The strong absorption peaks at 1095, 1152 and 1632 cm⁻¹ corresponds to the asymmetric and symmetric C-O stretching [24, 25]. Absorption bands between 872 and 995 cm⁻¹ can react with symmetric and asymmetric stretching vibrations of C-H [26, 27]. The peaks observed between 451 and 616 are attributed to Cu-O stretching in Cu-MOF [27, 28]. According to the FTIR spectrum, the final structures of Cu-MOF nanostructures are suggested in Fig. 5.

The porous structure of Cu-MOF at 77K is shown in Fig. 5 by nitrogen adsorption/desorption results. Fig. 6 depicts isotherms of the type (I) in the IUPAC classification, which is an example of microporous materials. [29] The early isotherm's dramatic increase and high N₂ uptake indicate a high proportion of microporous. Furthermore, the amount of micro/mesoporous is very low because the samples' isotherms in the high pressure region show no obvious hysteresis and tail. [30] In addition, the calculated N₂ adsorption/desorption isotherms were related to textural parameters. The surface area of Cu-MOF measured using BET analysis was 284.94 m²/g. As a result, the high surface area and porosity of this sample are determined by N₂ uptake by Cu-MOF.

Cu-MOF cytotoxicity was tested on MCF-7 breast cancer cells and the results are shown in Fig. 6. Based on the results of Fig. 7, after 48 h with high concentrations of Cu-MOF (200 µg/mL), cell proliferation and viability were observed to be approximately 24 % higher than the control. The IC₅₀ value for exposure of Cu-MOF at 24 h, 131 µg/mL was obtained and following at 48 h, 109 µg/mL was obtained. Therefore, breast cancer cell survival was dependent on concentrations of Cu-MOF and time of incubation.

Conclusion

Cu-MOF was prepared using UARM method and it was further characterized by FT-IR, XRD, SEM and nitrogen adsorption/desorption analysis. It has shown reasonably good anticancer activities against MCF-7 breast cancer cells. In summary, the synthesized Cu-MOF exhibited anti-cancer properties against MCF-7 breast cancer with an IC₅₀ value of 109 µg/mL in 48 hours and viability about 24% with the highest test concentration (200 µg/mL) was obtained at 48 h.

Declarations

Author Declarations section: The manuscript is in compliance with ethical standards.

o Ethics approval and consent to participate: We confirm all relevant ethical guidelines have been followed, and any necessary IRB and/or ethics committee approvals have been obtained.

All methods were used in accordance with relevant guidelines and regulations. Also, we confirmed that all experimental protocols were approved by the Ethics Licensing Committee of the Bam University of Medical Sciences (no. 06 on 17/03/2021). In addition, informed consent was obtained from all study participants.

o Consent for publication: In accordance with the copyright transfer or open access rules.

o Availability of data and materials: The datasets used and/or analyzed during the current study available from the corresponding author on reasonable request.

o Competing interests: There is no competing interests.

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o Authors' contributions

Malihe Zeraati: Methodology, Software, Writing- Original Visualization, Writing-

Mohammadreza Moghaddam-Manesh: Methodology, Software, Visualization,

Sara Hosseinzadegan: Writing- Reviewing and Editing Investigation.

Parya Kazemzadeh: Writing- Reviewing and Editing Investigation

Narendra Pal Singh Chauhan: Conceptualization, Methodology, Software, Writing- Original, Visualization, Writing- Reviewing and Editing Investigation, Data curation, Validation, Supervision.

Ghasem Sargazi: Conceptualization, Methodology, Software, Writing- Original, Visualization, Writing- Reviewing and Editing Investigation, Data curation, Validation, Supervision.

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o Research involving Human Participants and/or Animals: Not applicable.

o Informed consent: Not applicable.

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Figures

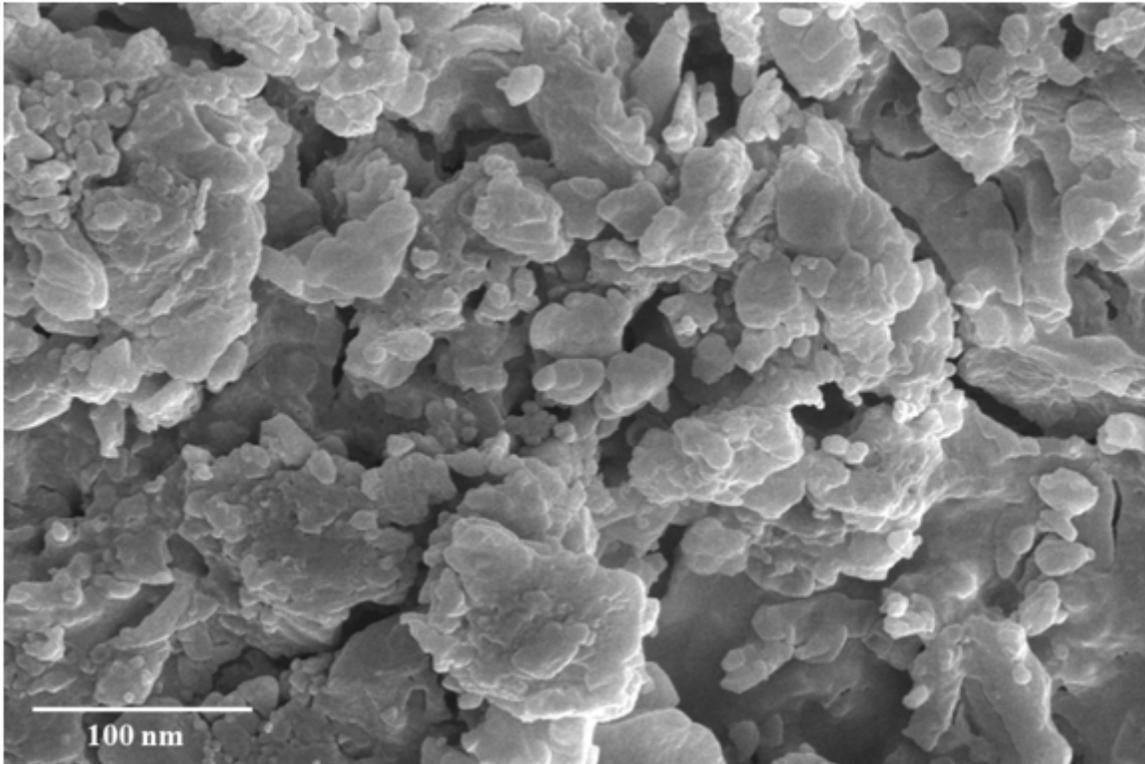


Figure 1

SEM image of the Cu-MOF.

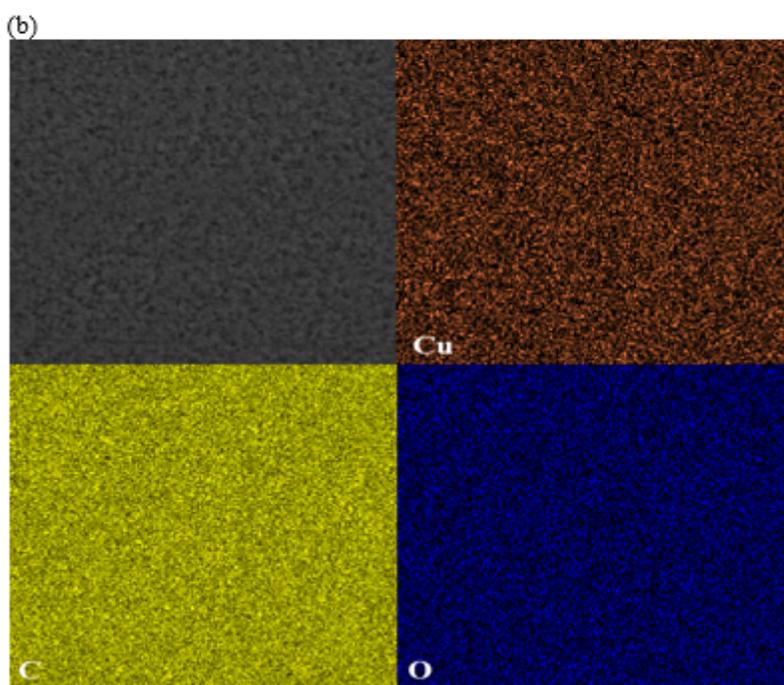
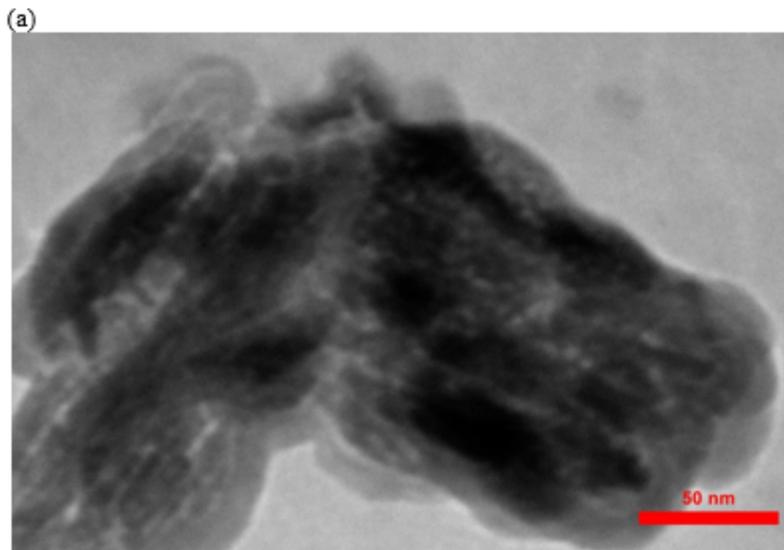


Figure 2

(a)TEM image of the Cu-MOF and (b)its mapping.

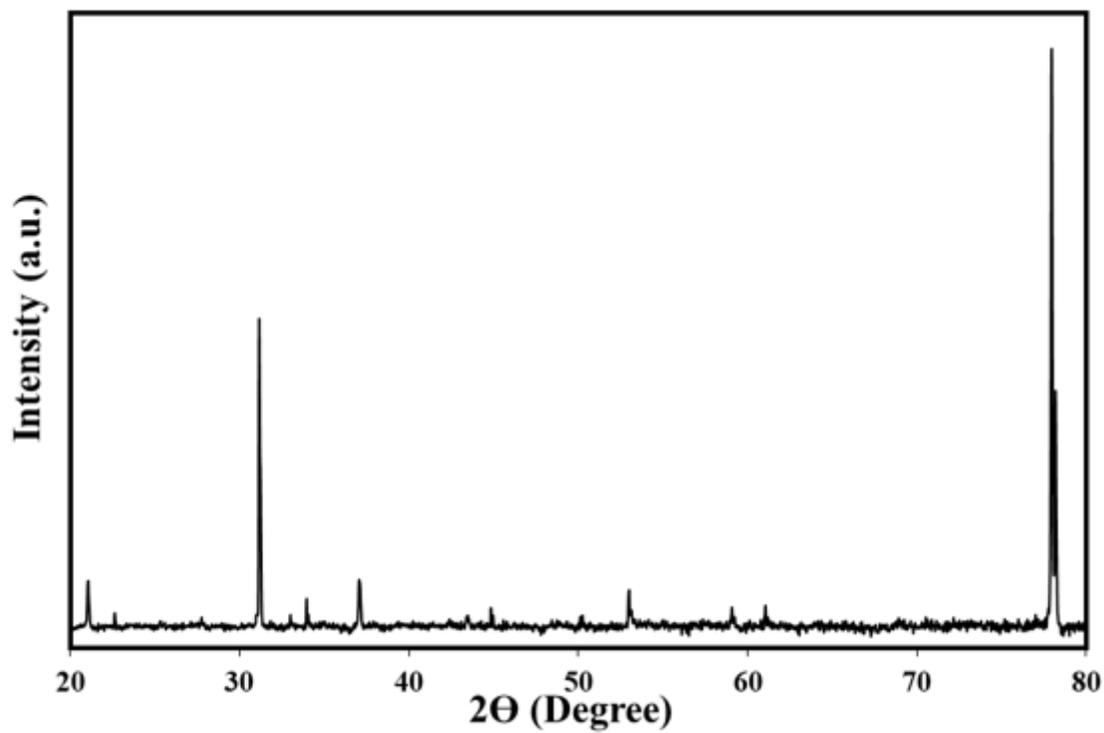


Figure 3

The X-ray diffraction pattern of Cu-MOF.

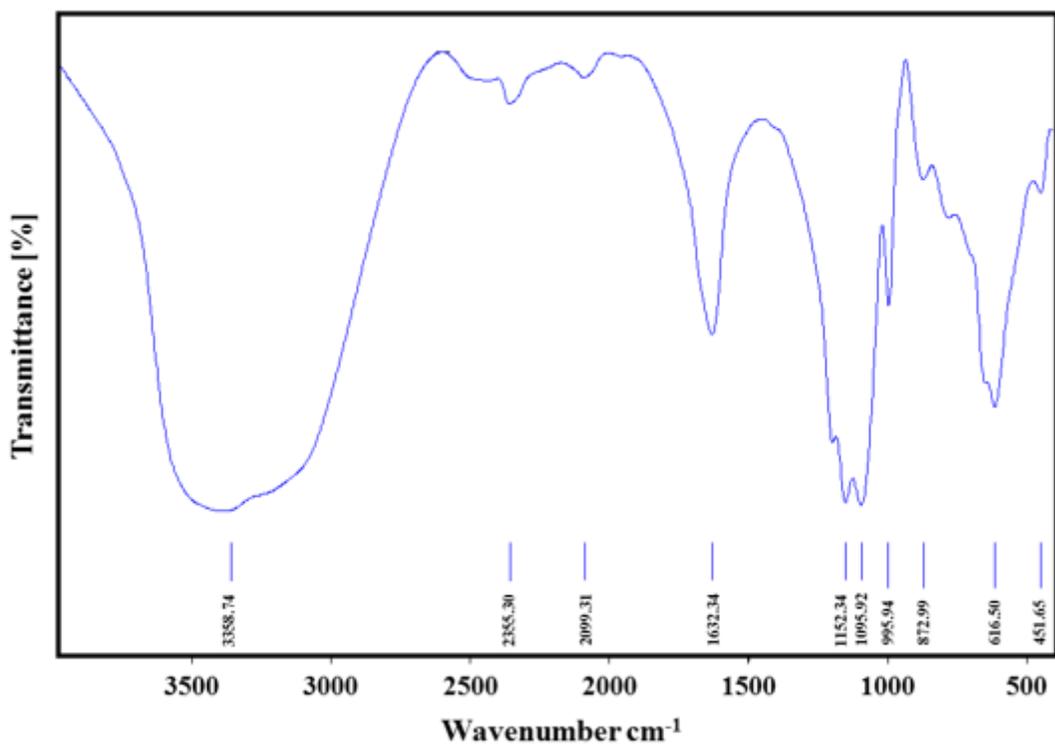


Figure 4

The FTIR spectrum of Cu-MOF.

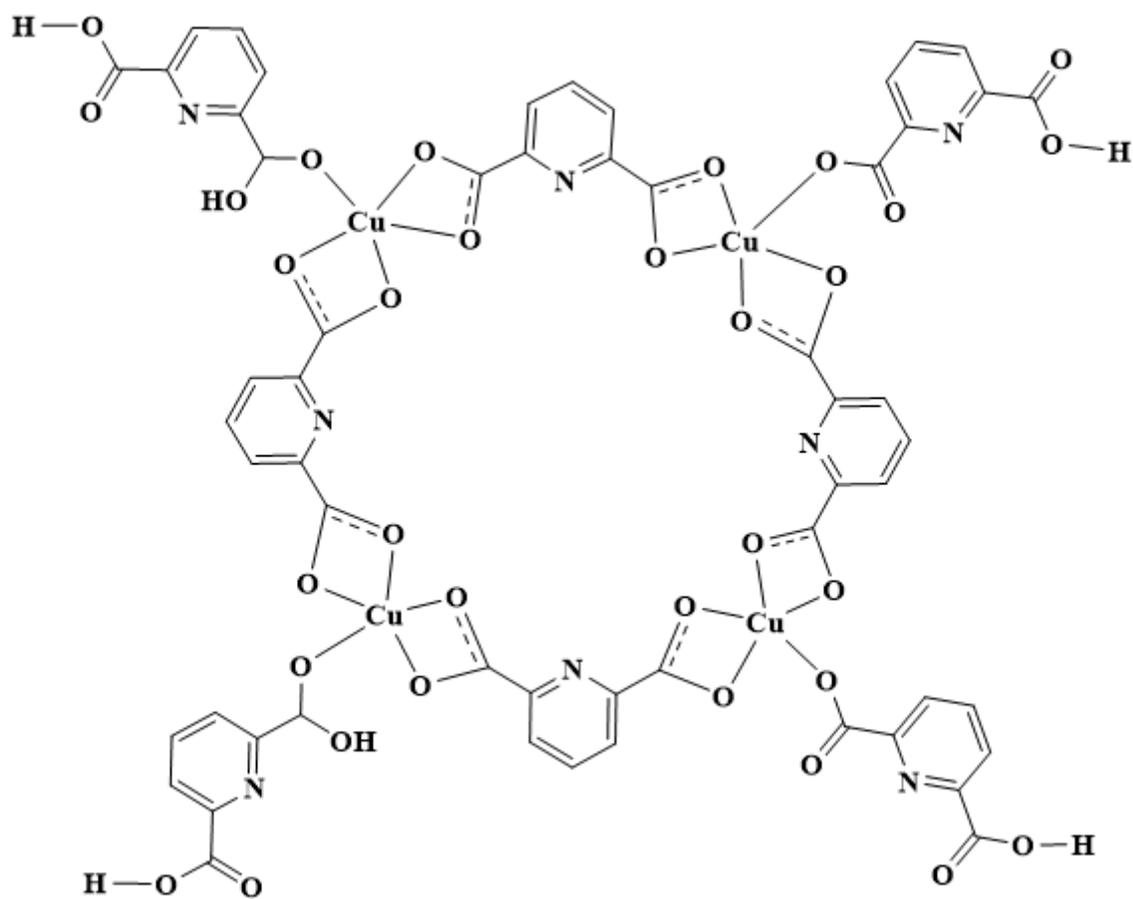


Figure 5

The proposed structures of Cu-MOF.

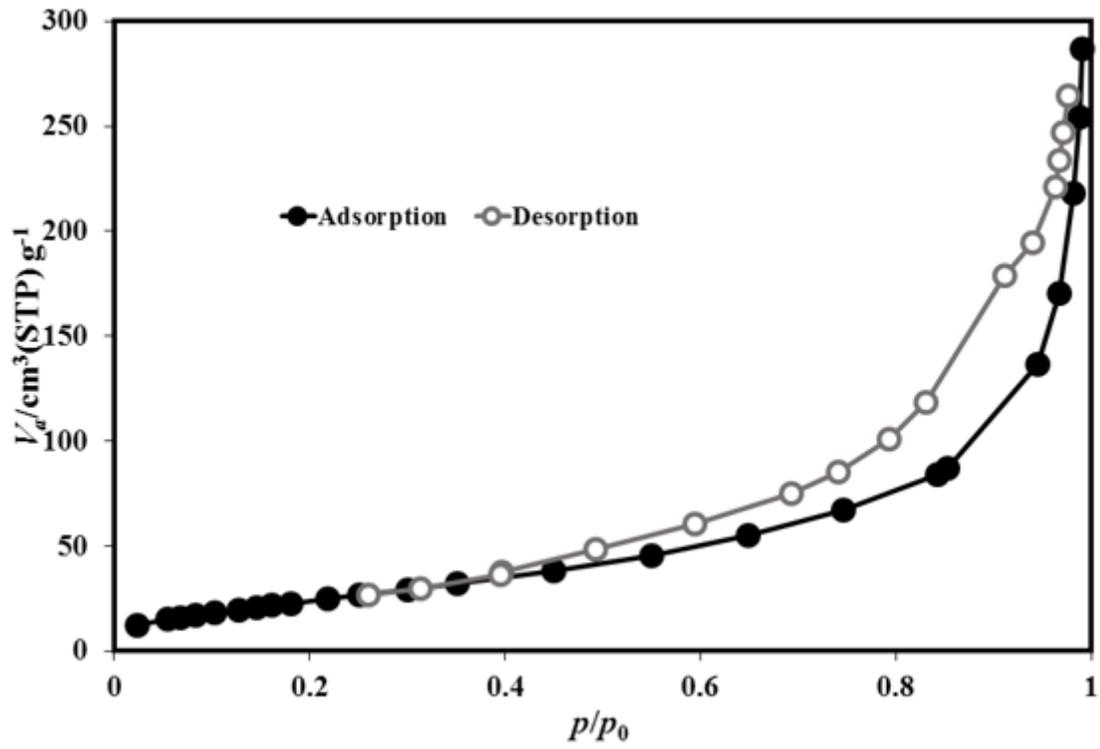


Figure 6

Nitrogen adsorption-desorption curve for Cu-MOF at 77K.

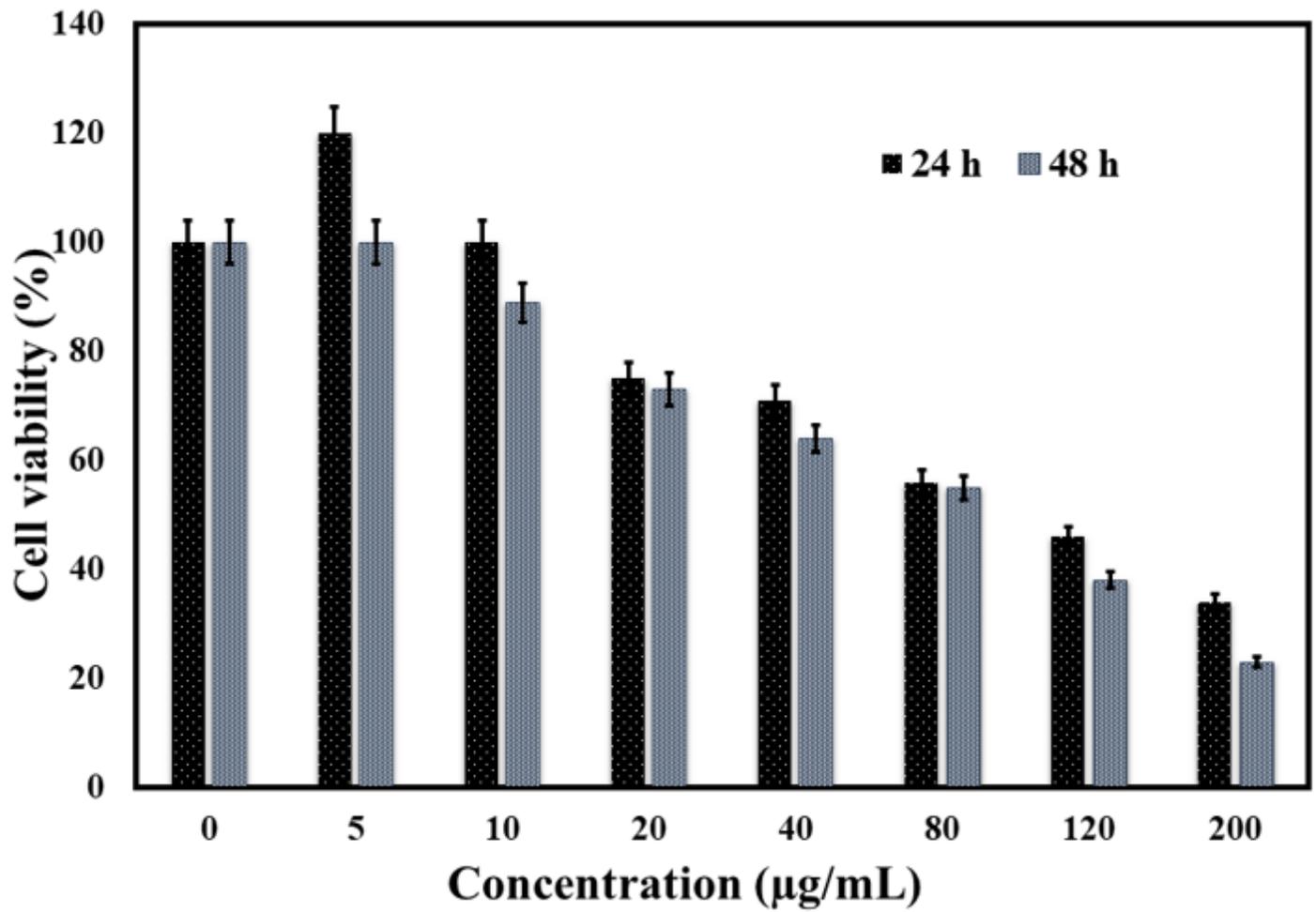


Figure 7

Anticancer result of Cu-MOF in MCF-7 breast cancer cell for 24 and 48 h. Data represents mean ($n = 3$) \pm SD.